Supporting Information

A mitochondria-targeted chemiluminescent probe for detection of hydrogen sulfide in cancer cells, human serum and *in vivo*

Hande Gunduz,^{ab1} Toghrul Almammadov,^{b1} Musa Dirak,^{b1} Alperen Acari,^{c1} Berkan Bozkurt^{cd} and Safacan Kolemen*^{bce}

^aNanofabrication and Nanocharacterization Centre for Scientific and Technological Advanced Research, Koç University, Istanbul 34450 Türkiye

^bDepartment of Chemistry, Koç University, Rumelifeneri Yolu, Istanbul 34450, Türkiye

^cKoç University Research Center for Translational Medicine (KUTTAM), Istanbul 34450, Türkiye

^dGraduate School of Health Sciences, Koç University, Rumelifeneri Yolu, Istanbul 34450, Türkiye

^eKoç University Surface Science and Technology Center (KUYTAM), Istanbul 34450, Türkiye ¹These authors contributed equally.

E-mail: skolemen@ku.edu.tr

1. Materials and Instruments

All reagents were commercially available and used without further purification unless otherwise noted. All dry solvents were obtained by using standard procedures and dry THF was obtained from Inert PureSolv solvent drying system. Reactions needed to be done under inert atmosphere were done in a Schlenk line and Argon was used as an inert gas. For Thin layer chromatography (TLC Merck Silica Gel 60 F254) commercially prepared 0.25 mm silica gel plates were used, and visualization of the compounds were done by UV light. Silica gel 60 (Merck 230-400 mesh) was used for column chromatography experiments. The relative proportions of solvents in chromatography solvent mixtures refer to the volume:volume ratio. The reaction monitoring and purification of the target probe were accomplished by reversephase high pressure liquid chromatography (RP-HPLC): C18 4µm, 150x4.6mm, eluent given in parentheses; preparative RP-HPLC: C18 5 µm, 250x21.2 mm, eluent given in parentheses. ¹H-NMR and ¹³C-NMR analyses were recorded by a 500 MHz Bruker Ascend magnet equipped with an Avance NEO console spectrometer using CDCl₃ as the solvent. The chemical shifts are reported in parts per million (ppm) downfield from an internal TMS (trimethylsilane) reference. Coupling constants (J) are reported in hertz (Hz), and the spin multiplicities were specified by the following symbols: s (singlet), d (doublet), t (triplet), and m (multiplet). NMR spectras were processed with MestReNova program. Mass spectra were recorded on Waters Vion QTOF mass spectrometer. HPLC analyses were performed on an Agilent 1260 series. fluorescence spectrums were acquired on UV-Vis and Shimadzu UV-Vis-NIR spectrophotometer Eclipse spectrophotometer, and Agilent Cary respectively. Chemiluminescence signal was detected by Biotek Synergy H1 MF microplate reader. Confocal images were taken by using Leica DMI8 SP8 Inverted Confocal Microscope. Perkin Elmer IVIS Lumina Series III was used for in vivo imaging studies.

2. Photophysical Characterization

Photophysical Characterization Details:

Absorbance and fluorescence spectra of the **MCH** (10 μ M), before and after treatment with H₂S (20 μ M) in DMSO (2% PBS, pH 7.4), were recorded every minute during a total 10-minute time span (λ_{ex} = 440 nm). Fluorescence spectra of MC-benzoate were captured with similar conditions and superimposed with the spectra of **MCH** (10 μ M) treated with H₂S (20 μ M).

Chemiluminescence Signal:

Chemiluminescence signal of **MCH** (10 μ M) was measured in the absence or presence of varying concentrations of H₂S (1-30 μ M) in DMSO (2% PBS, pH 7.4). Chemiluminescence response of **MCH** was captured every minute for a total of 60 minutes and no filters were used.

Selectivity:

Chemiluminescence response of the **MCH** (10 μ M) against other biologically relevant or competing analytes was measured to prove its selectivity towards H₂S. **MCH** (10 μ M) treated with GSH (5 mM), L-Cys (5 mM), H₂O₂ (1 mM), Na₂S₂O₃ (1 mM), SO₄²⁻ (1 mM), KSCN (1 mM), NaF (1 mM), OAc⁻ (1 mM) and NaNO₂ (1 mM) and the peak luminescence signal of each analyte was recorded.

Limit of Detection:

The limit of detection (LOD) was calculated by the following formula:

$$LOD = \frac{3 * \sigma}{m}$$

where σ stands for the standard deviation of the total chemiluminescence intensity of seven separate replicates of **MCH** (10 µM), and m refers to the slope of the total chemiluminescence signal versus the concentration of H₂S (0-1 eq.).

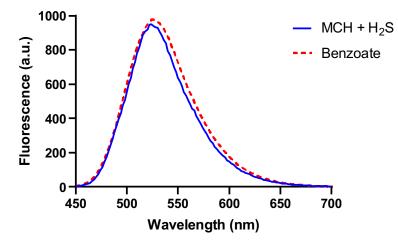
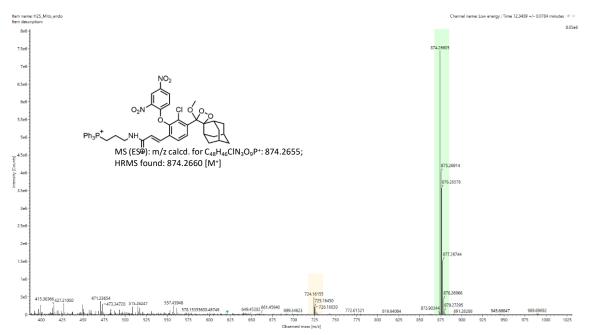


Figure S1. Fluorescence spectra of H₂S-treated (2 equivalents) **MCH** and **MC-benzoate** ester in DMSO (2% PBS, pH 7.4). λ_{ex} = 440 nm.



3. HR-MS Analyses and Activation Mechanism

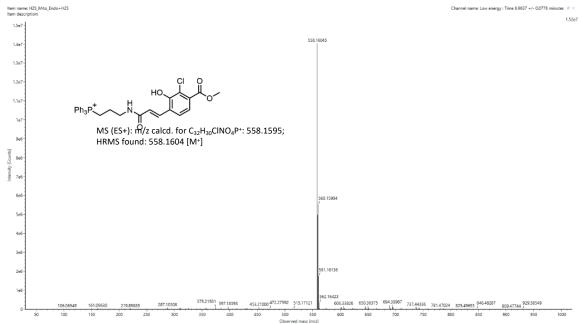


Figure S2. The HRMS spectrum of **MCH** (top) and resulting **MC-benzoate** (bottom) ester after H_2S treatment.

4. Selectivity

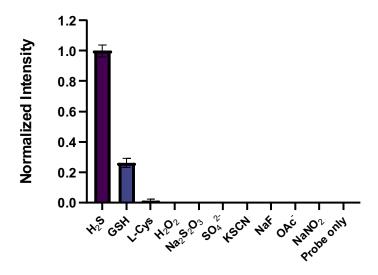


Figure S3. Luminescence intensity of MCH (10 μ M) with various analytes. (n=3)

5. Detection of the Chemiluminescence Signal

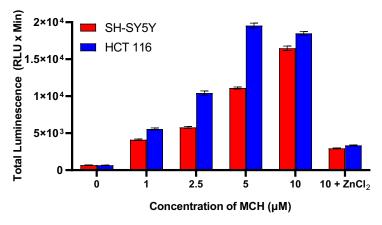
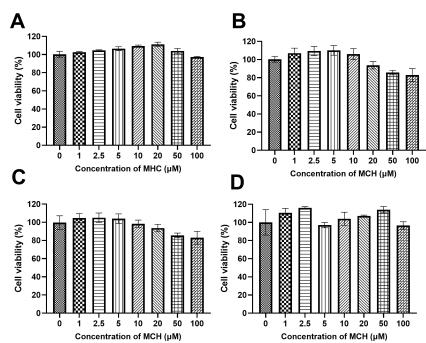


Figure S4. Total luminescent signal of MCH in HCT116 and SH-SY5Y (2 h). (n=3)



6. Cell Culture Studies

Figure S5. Cell viability of A) Vero, B) HCT116, C) SH-SY5Y and D) HGrC1 cells treated with various concentration of MCH (0–100 μ M). (n=3)

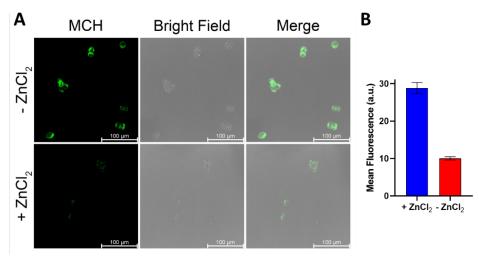


Figure S6. A) Confocal microscopy images of SH-SY5Y cells treated with and without ZnCl₂ (300 μ M) for 10 minutes. Data was collected after 30 minutes **MCH** (10 μ M) treatment. **B)** Mean fluorescence. (n=4) (λ_{ex} = 405 nm / λ_{em} = 500-600 nm). (n=3)

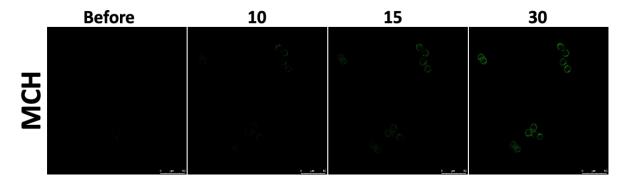


Figure S7. Confocal images of MCH before and after treatment of HCT116 cells captured over the course of 30 minutes. Scale bar: 50 μ M.

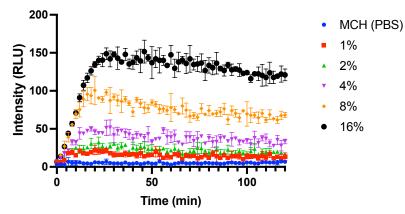


Figure S8. Time dependent luminescence response of **MCH** (10 μ M) in the various (1%-16%) percentages of human serum in PBS (1% DMSO, pH 7.4, n=3).

7. Serum Experiments

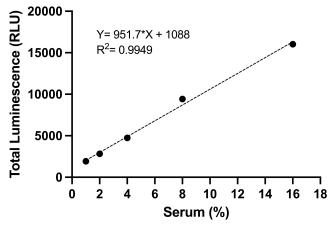


Figure S9. Calibration curve of MCH (10 μ M) in the various (1%-16%) percentage of human serum. (n=3)

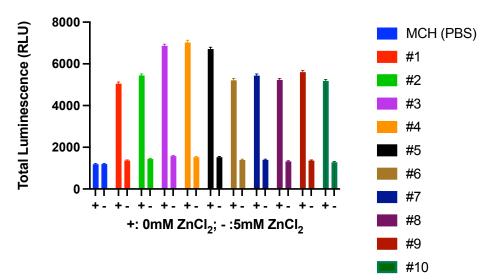


Figure S10. Total luminescence response of MCH (10 μ M) in 10 different human serum (%4). (n=3)

8. In vivo Imaging

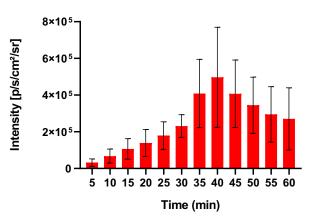


Figure S11. Time dependent luminescence intensity in the tumor region. (n=4)

9. NMR and HR-MS Spectra

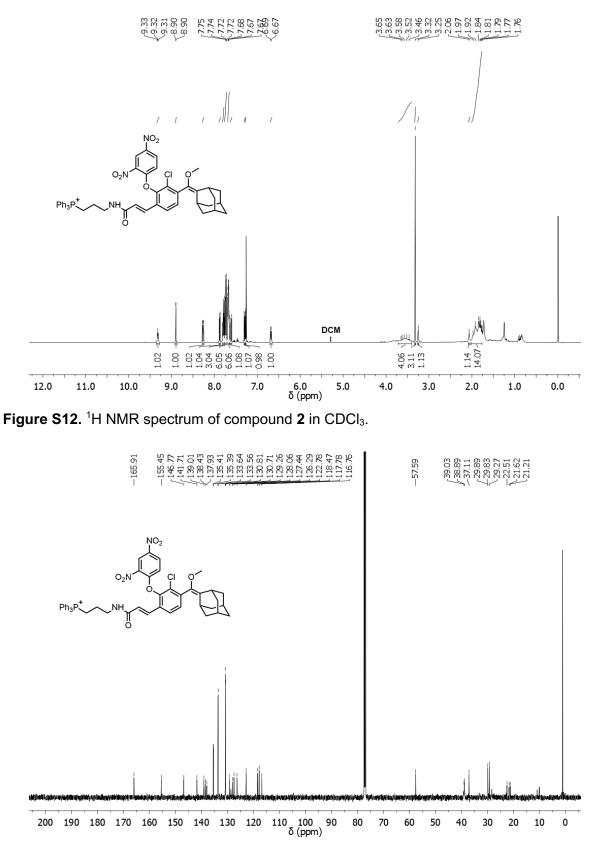


Figure S13. ¹³C NMR spectrum of compound 2 in CDCl₃.

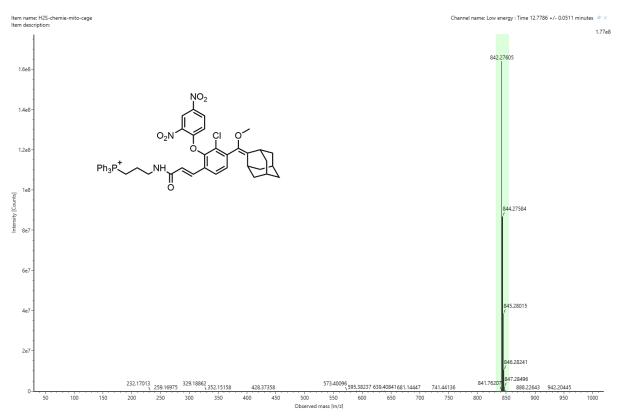


Figure S14. HRMS spectrum of compound 2.

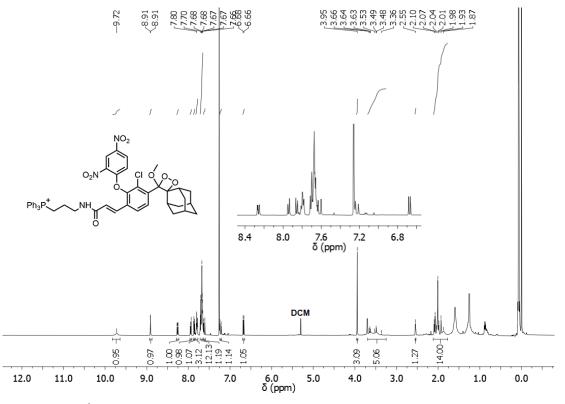


Figure S15. ¹H NMR spectrum of compound MCH in CDCl₃.

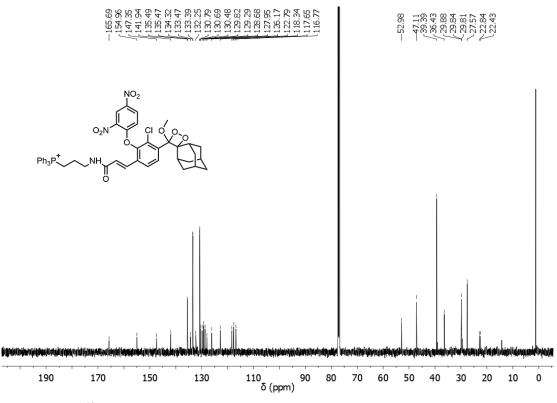


Figure S16. ¹³C NMR spectrum of compound MCH in CDCl₃.