

Electronic Supplementary Information

MOF-derived transition metal-based catalyst for the electrochemical reduction of CO₂ to CO: A mini review

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Table S1 Molecular formula for MOF precursors in the MOF-derived materials.

MOF	Molecular formula	Ref.
ZIF-8	$\text{Zn}(\text{C}_4\text{H}_5\text{N}_2)_2$	1-12
ZIF-67	$\text{CoC}_4\text{H}_6\text{N}_2$	1
PCN-222/MOF-545	$\text{Zr}_6\text{C}_{96}\text{H}_{64}\text{Cl}_2\text{Fe}_2\text{N}_8\text{O}_{32}$	13-15
PCN-224	$\text{Zr}_{15}\text{C}_{144}\text{H}_{72}\text{N}_{12}\text{O}_{64}$	13
MOF-74	$\text{Zn}_2\text{C}_8\text{H}_4\text{O}_8$	16, 17
Fe-BTT	$\text{Fe}_3[(\text{Fe}_4\text{Cl})_3(\text{BTT})_8(\text{MeOH})_4]_2$	18
Cu-BTT	$[\text{Cu}(\text{DMF})_6][(\text{Cu}_4\text{Cl})_3(\text{BTT})_8(\text{H}_2\text{O})_{12}]$	19, 20
HKUST-1/Cu-BTC	$\text{Cu}_3(\text{C}_9\text{H}_3\text{O}_6)_2$	21

Table S2. The performance of MOF catalysts for the electrochemical reduction of CO₂ to CO.

Metal center	MOF	Electrolyte	Potential	Main product	Max FE/%	Catalytic stability/h	Ref.
Co	CoCp ₂ @MOF-545-Co	0.5 M KHCO ₃	-0.7 V vs. RHE	CO	97.0	/	22
Co	CoPc-Cu-O	0.2 M KHCO ₃	-0.74 V vs. RHE	CO	85.0	10	23
Co	MOF-1992	0.1 M KHCO ₃	-0.63 V vs. RHE	CO	78.0	2	24
Co	PPy@MOF-545-Co	0.1 M KHCO ₃	-0.8 V vs. RHE	CO	98.0	10	25
Co	Al ₂ (OH) ₂ TCPP-Co	0.5 M KHCO ₃	-0.7 V vs. RHE	CO	76.0	7	26
Zn	ZIF-A-LD	0.1 M KHCO ₃	-1.1 V vs. RHE	CO	90.6	10	27
Zn	ZIF-8	0.25 M K ₂ SO ₄	-1.1 V vs. RHE	CO	81.0	/	28
Zn	ZIF-8	0.5 M NaCl	-1.8 V vs. SCE	CO	65.0	4	29
Zn	PcCu-O ₈ -Zn	0.1 M KHCO ₃	-0.7 V vs. RHE	CO	88.0	11	30
Ni	Ni(Im) ₂ ZIF	0.5 M KHCO ₃	-0.85 V vs. RHE	CO	78.8	14	31
Ni	NiPc-Ni(NH) ₄	0.5 M KHCO ₃	-0.7 V vs. RHE	CO	96.4	10	32
Ni	NiPc-NiO ₄	0.5 M KHCO ₃	-0.85 V vs. RHE	CO	98.4	10	33
Fe	PCN-222(Fe)/C	0.5 M KHCO ₃	-1.35 V vs. SCE	CO	91.0	10	34
	PCN-223-BA				76.8		
Zr	PCN-223-FA	0.5 M KHCO ₃	-0.77 V vs. RHE	CO	81.0	20	35
	PCN-223-AA				90.7		

Table S3. The performance of MOF-derived catalysts for the electrochemical reduction of CO₂ to CO.

Catalyst	MOF precursor	Electrolyte	Potential	Main product	Max FE/%	Catalytic stability/h	Ref.
Fe ₁ N ₂ O ₂ /NC	Fe/Zn-MOF 74	0.1 M KHCO ₃	-0.5 V vs. RHE	CO	99.7	12	16
Zn _{0.5} Fe _{0.5} -N-C-222	MOF PCN-222	0.1 M KHCO ₃	-1.3 V vs. Ag/AgCl	CO	96.0	12	13
Zn _{0.5} Fe _{0.5} -N-C-224	MOF PCN-224		-1.2 V vs. Ag/AgCl		92.0		
FeSAs/CNF-900	ZnFe-ZIF	0.5 M KHCO ₃	-0.47 V vs. RHE	CO	86.9	12	2
Fe-N ₄	Fe-ZIF	0.1 M KHCO ₃	-0.58 V vs. RHE	CO	93.0	20	3
Ni ₁ -N-C	Ni-PCN-222	0.5 M KHCO ₃	-0.8 V vs. RHE	CO	96.8	10	14
C-Zn ₁ Ni ₄ ZIF-8	ZnNi-ZIF-8	0.5 M KHCO ₃	-0.73 V vs. RHE	CO	98.0	12	5
Ni _{SA} -N ₂ -C	MgNi-MOF-74	0.5 M KHCO ₃	-0.8 V vs. RHE	CO	98.0	10	17
Ni-N ₄ -O/C	Mn,Ni-MOFs	0.5 M KHCO ₃	-0.9 V vs. RHE	CO	100.0	20	36
Cu-In-NC	CuIn-ZIF-8	0.1 M KHCO ₃	-0.7 V vs. RHE	CO	96.0	12	10
NiNPIC-4	Zn-Ni MOFs	0.5 M KHCO ₃	-0.65 V vs. RHE	CO	95.1	24	37
Fe-N-C ₉₀₀	Fe-BTT	0.1 M KHCO ₃	-1.2 V vs. Ag/AgCl	CO	86.8	10	18
PCN-222(Fe)/C	PCN-222(Fe)	0.5 M KHCO ₃	-0.6 V vs. RHE	CO	91.0	10	34
PCN-222(Fe)/CNTs	PCN-222(Fe)	0.5 M KHCO ₃	-0.6 V vs. RHE	CO	95.5	10	38
PCN-222(Mn)/CNT	PCN-222(Mn)		-0.6 V vs. RHE		88.5		
PCN-222(Co)/CNT	PCN-222(Co)	0.5 M KHCO ₃	-0.65 V vs. RHE	CO	89.3	20	39
PCN-222(Zn)/CNT	PCN-222(Zn)		-0.7 V vs. RHE		92.5		

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