

Formation of lanthanide luminescent di-metallic helicates in solution using a bis-tridentate (1,2,3-triazol-4-yl)-picolinamide (tzpa) ligand

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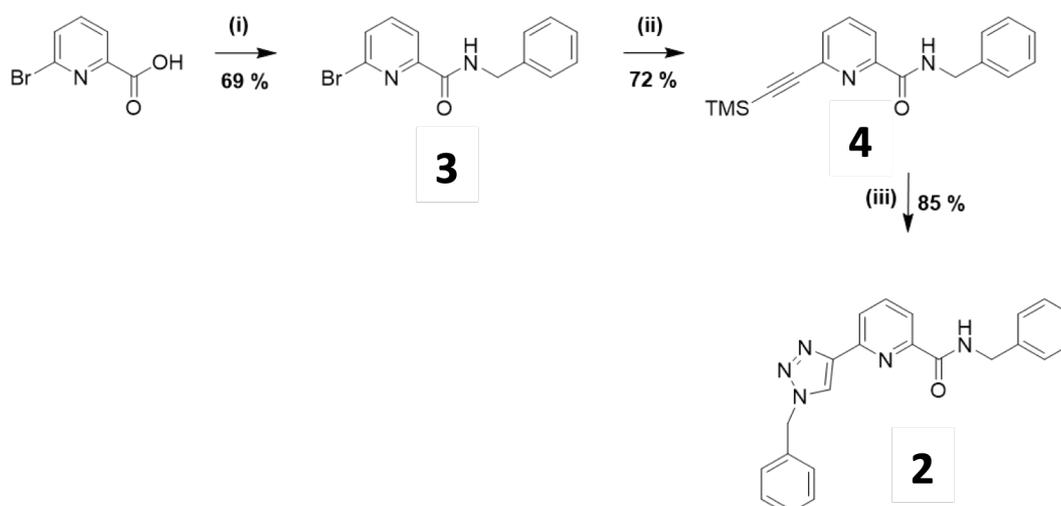
Electronic Supplementary Information (ESI)

Characterisation of Ligand 1.

N,N'-(1,3-phenylenebis(methylene))bis(6-(1-((*S*)-1-phenylethyl)-1H-1,2,3-triazol-4-yl)picolinamide)

Obtained as white solid (48 mg, 0.012 mmol, 62%); m.p. 118 – 123 °C; HRMS (*m/z*) (ESI+): C₄₀H₃₇N₁₀O₂⁺ *m/z* = 689.3101 [M+H]⁺. Found *m/z* = 689.3109; ¹H NMR (600 MHz, DMSO-*d*₆): δ (ppm) = 9.41 (t, 2H, NH), 8.98 (s, 2H, triazolyl H), 8.18 (dd, *J* = 7.8, 1.0 Hz, 2H, pyr CH-triazole), 8.02 (t, 2H, *ortho* pyr CH), 7.91 (dd, *J* = 7.7, 1.0 Hz, 2H, pyr CH-Ar), 7.45 – 7.09 (m, 14H, Ar CH), 6.05 (q, *J* = 7.1 Hz, 2H, CH-CH₃), 4.55 (d, *J* = 6.5 Hz, 4H, CH₂), 1.95 (d, *J* = 7.1 Hz, 6H, CH₃); ¹³C NMR (150 MHz, DMSO-*d*₆): δ (ppm) = 163.75, 149.53, 148.78, 146.79, 140.98, 139.59, 138.79, 128.82, 128.37, 128.05, 126.09, 125.61, 123.03, 121.57, 120.73, 64.90, 59.51, 42.28, 21.29; IR ν_{max} (cm⁻¹): 3326, 2934, 2165, 1664, 1660, 1518, 1444, 1241, 1172, 1076, 995, 830, 765, 697.

Synthesis



Scheme S1. Synthesis of ligand 2.

N-Benzyl-6-bromopicolinamide (3)

Compound **3** was synthesized from 6-bromopyridine-2-carboxylic acid (2.0 g, 9.9 mmol) and benzylamine (1.1 mL, 10.8 mmol) in a DMF:DCM (4:1) solvent mixture. Under argon, then to this solution was added HOBT (1 equivalent), NEt₃ (1.2 equivalents), and the reaction mixture was left to stir for 30 mins, after which it was cooled to 0 °C. EDC·HCl (1.5 equivalents) was then added to the suspension and the resulting mixture stirred at 0 °C for a further 30 mins. The mixture was then allowed to warm to room temperature and stirred for a further 48 hrs. The product was dried under reduced pressure and taken up in CH₂Cl₂, washed with 1M HCl, sat. NaHCO₃ and H₂O dried over magnesium sulfate and concentrated under reduced pressure to yield orange oil. This resulting oil was dried to a solid by repeated evaporation under reduced pressure with toluene to give compound **3** in 69% yield. m.p. 167 - 174 °C; HRMS (*m/z*) (ESI+): C₁₃H₁₁BrN₂O⁺ *m/z* = 291.0133 [M+H]⁺. Found *m/z* = 291.0127; ¹H NMR (600 MHz, CDCl₃): δ (ppm) = 8.20 (d, *J* = 7.7 Hz, 1H, Ha), 8.14 (s, *J* = 6.2 Hz, 1H, NH), 7.72 (t, *J* = 7.7 Hz, 1H, Hb), 7.61 (d, *J* = 7.7 Hz, 1H, Hc), 7.36 (dd, *J* = 8.3, 5.5 Hz, 4H, Hd & He), 7.33 - 7.27 (m, 1H, *J* = 8.3, 5.5 Hz, Hf), 4.67 (d, *J* = 6.2 Hz, 2H, CH₂); ¹³C NMR (150 MHz, CDCl₃): δ (ppm) = 162.9, 151.1, 140.7, 139.8, 138.1, 130.9, 128.9, 128.1, 127.8, 121.6, 43.7; IR ν_{max} (cm⁻¹): 3354, 3318, 3166, 3033, 2942, 1736, 1639, 1603, 1568, 1527, 1502, 1451, 1374, 1278, 1247, 1222, 1161, 1089, 1049, 972, 865, 819, 764, 743, 728, 687, 631, 606.

N-benzyl-6-((trimethylsilyl)ethynyl)picolinamide (4)

To a solution of **3** (1 g, 3.43 mmol) in THF/NEt₃ stirring at 0 °C (4:1, 40 mL), CuI (0.2 mmol) and Pd(PPh₃)₄ (0.2 mmol) were added under an argon atmosphere. Ethynyltrimethylsilane (0.5 mL, 3.43 mmol) was added dropwise and left to stir at room temperature for 48 hrs. The resulting solution was concentrated under reduced pressure and filtered through a plug of celite to yield a brown solid which was purified by flash column chromatography (RediSep® 24g, gradient elution 0 → 20 % EtOAc in Hexane) to yield **4** as a brown solid in 63% yield. m.p. 187 - 191 °C; HRMS (*m/z*) (ESI+): C₁₈H₂₁N₂OSi⁺ *m/z* = 309.1418 [M+H]⁺. Found *m/z* = 309.1418; ¹H NMR (600 MHz, DMSO-*d*₆): δ (ppm) = 9.22 (t, 1H, *J* = 6.4 Hz, NH), 8.02 (m, *J* = 7.5, 1.1 Hz, 2H, pyr H), 7.74 (dd, *J* = 7.5, 1.1 Hz, 1H, pyr H), 7.37 - 7.25 (m, 4H, Ar H), 7.27 - 7.18 (m, 1H, Ar H), 4.49 (d, *J* = 6.4 Hz, 2H, CH₂), 0.26 (s, 9H, CH₃-Si); ¹³C NMR (150 MHz, DMSO-*d*₆): δ (ppm) = 163.8, 151.0, 141.2, 139.9, 139.0, 130.6, 128.7, 127.9, 127.3, 122.5, 104.0, 95.7, 43.0, 0.1; IR ν_{max} (cm⁻¹): 3371.0, 1674.1, 1520.7, 1442.9, 1248.4, 841.6, 762.4, 645.7, 610.3

N-benzyl-6-(1-benzyl-1H-1,2,3-triazol-4-yl)picolinamide (2)

Ligand **2** was synthesised from benzylbromide (0.13 mL, 1.135 mmol) and sodium azide (1.135 mmol) which were added to a mixture of DMF/H₂O (4:1)

(15 mL) and stirred for an hour. To the mixture CuSO₄·5H₂O sodium ascorbate (0.9 mmol), and K₂CO₃ (1.135 mmol) were added and the mixture was degassed. Finally **4** (0.35 g, 1.13 mmol) in DMF (2 ml) was added. The reaction was stirred at r.t. under Ar for 48 hrs and dried under reduced pressure. A solution of EDTA/NH₄ (15 mL) was added and the product was extracted into DCM, washed with H₂O (3x30 mL) and dried over MgSO₄. The solution was concentrated under reduced pressure to yield a white solid (85% yield) HRMS (*m/z*) (ESI⁺): C₂₂H₂₀N₅O⁺ *m/z* = 370.1662 [M+H]⁺. Found *m/z* = 370.1662; ¹H NMR (600 MHz, DMSO-*d*₆): δ (ppm) = 9.47 (t, 1H, *J* = 6.5 Hz, NH), 8.98 (s, 1H, triazole H), 8.21 (d, *J* = 7.8, 0.6 Hz, 1H, pyr H), 8.09 (t, 1H, *J* = 7.8 Hz, pyr H), 7.99 (d, 1H, *J* = 7.8, 0.6 Hz, Pyr H), 7.39 – 7.23 (m, 10H, Ar H), 5.73 (s, 2H, CH₂-triazole), 4.56 (d, *J* = 6.5 Hz, 2H, CH₂-NH); ¹³C NMR (150 MHz, DMSO-*d*₆): δ (ppm) = 163.8, 149.7, 148.8, 147.2, 139.5, 138.9, 135.9, 128.9, 128.3, 128.2, 127.7, 127.1, 126.8, 124.5, 121.7, 120.9, 53.2, 42.3; IR ν_{max} (cm⁻¹): 3322.8, 1654.9, 1601.8, 1571.5, 1521.9, 1545.6, 1420.3, 1354.1, 1257.3, 1232.6, 1168.8, 1042.9, 995.4, 977.0, 820.2, 750.7, 716.7, 696.3, 582.9.

Figures

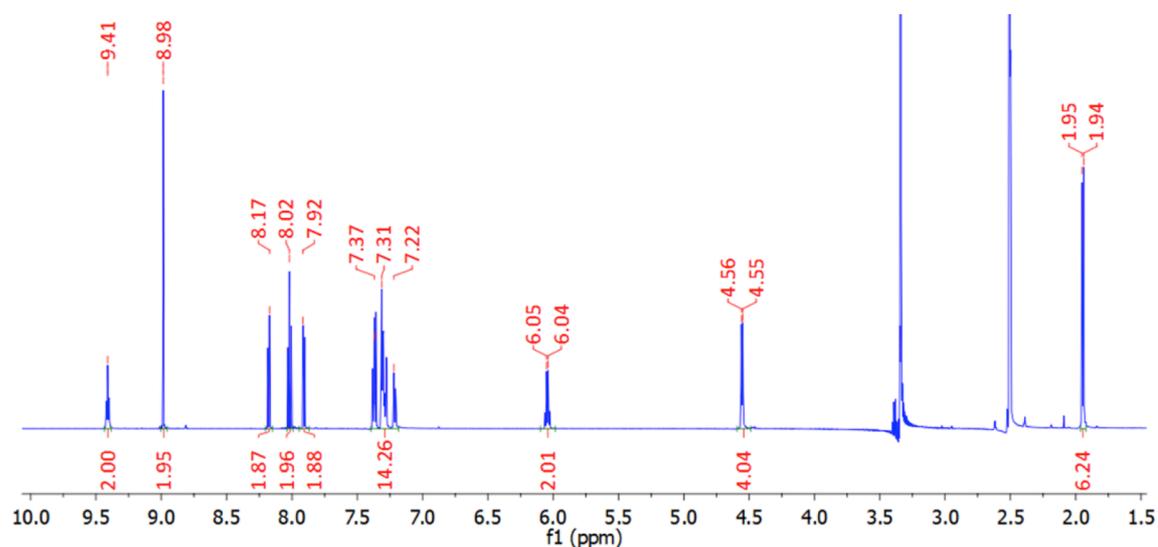


Figure 1S. The ¹H NMR spectrum (600 MHz, DMSO-*d*₆) of **1**.

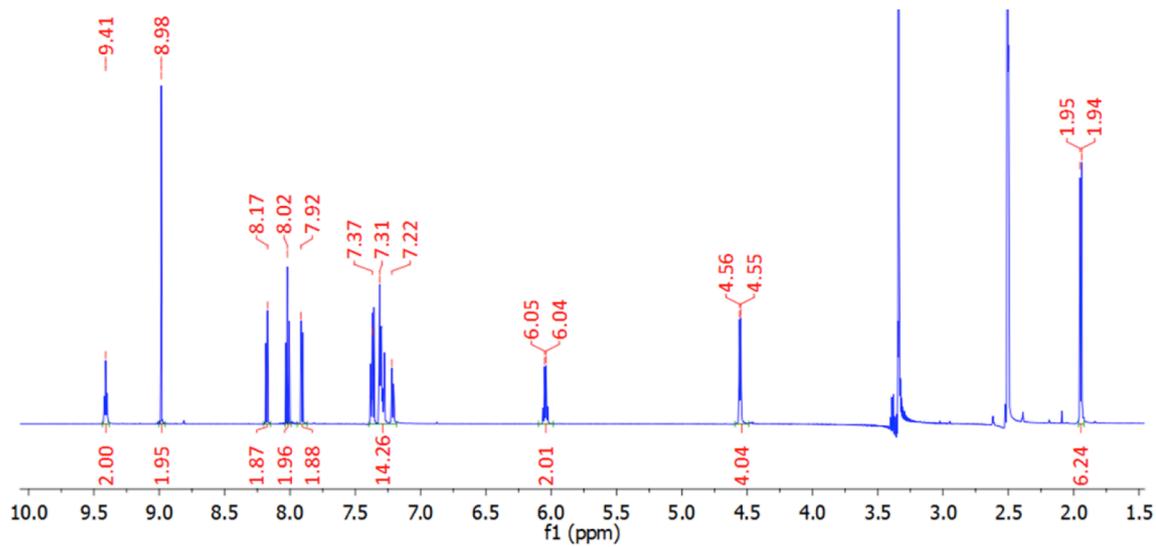


Figure 2S. The ^1H NMR spectrum (600 MHz, DMSO-d_6) of **2**.

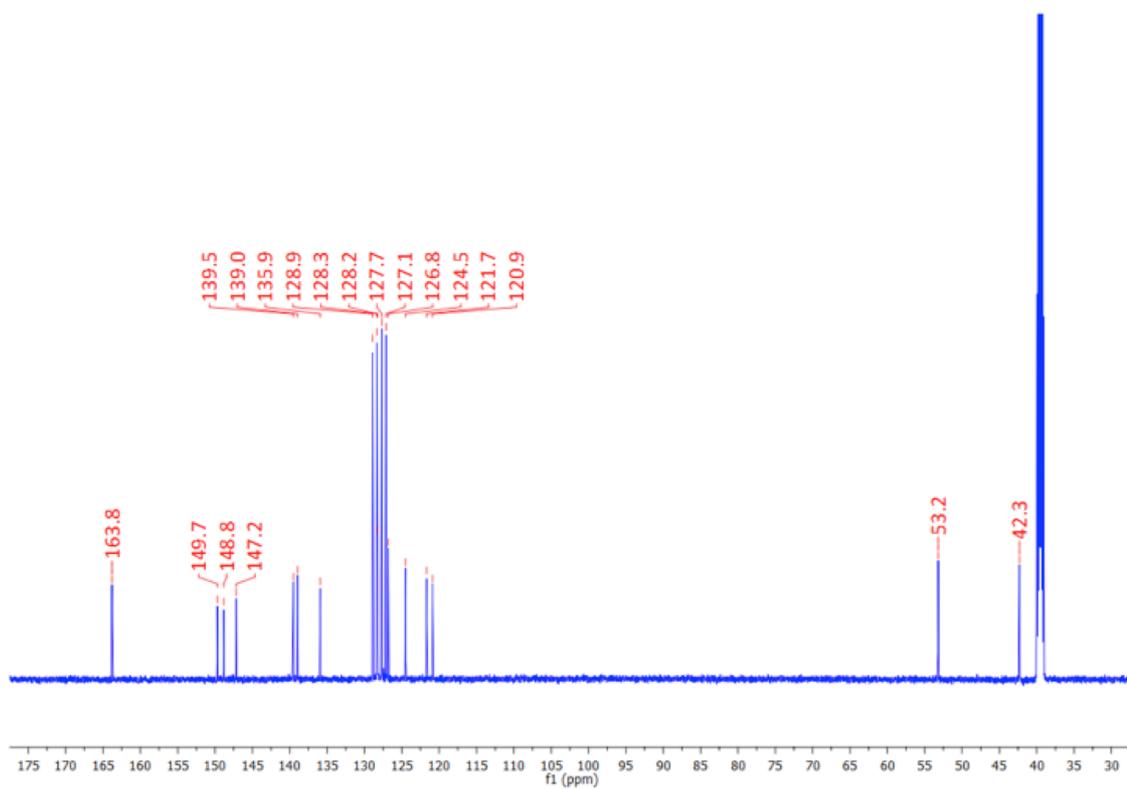


Figure 3S. The ^{13}C NMR spectrum (150 MHz, DMSO) of **2**.

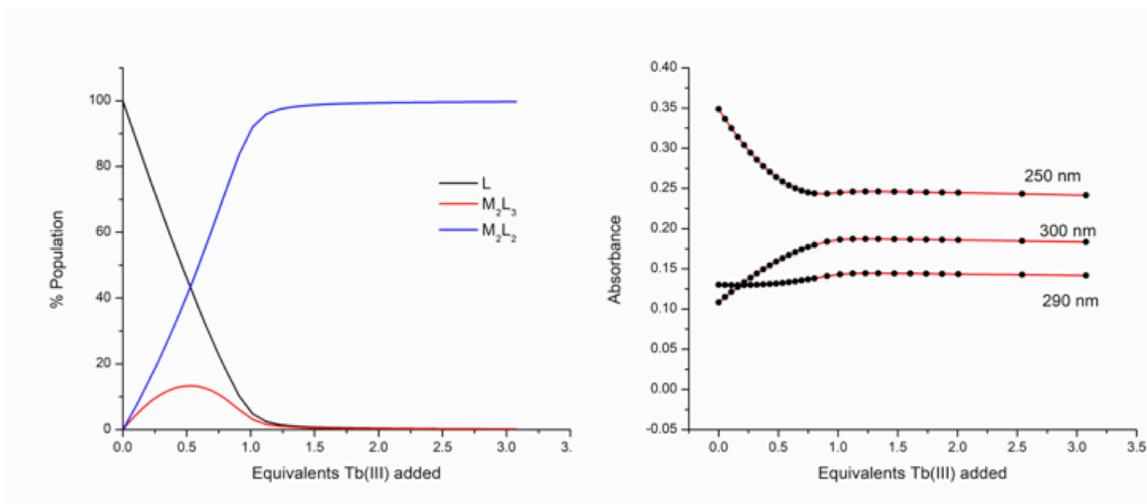


Figure 4S. (left) The speciation distribution diagram obtained from the fit of the UV-visible absorption titration data of ligand **1** against $Tb(CF_3SO_3)_3$ in CH_3CN . (right) The fit of the experimental binding isotherms using non-linear regression analysis software ReactLab.

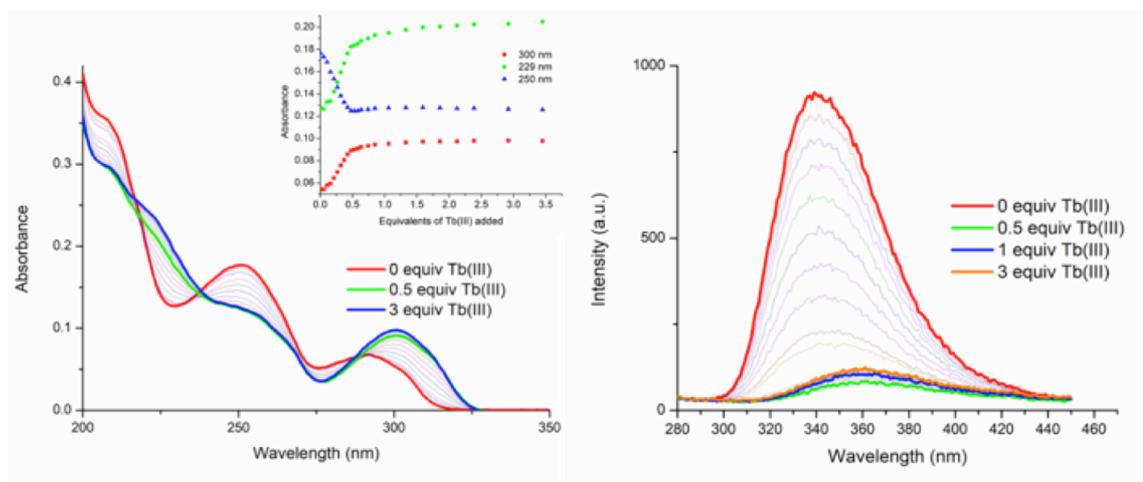


Figure 5S. The overall changes in the (left) UV-visible absorption spectra and (right) fluorescence emission spectra (excitation wavelength $\lambda = 250$ nm) upon titrating **2** (1×10^{-5} M) against $Tb(CF_3SO_3)_3$ (0 \rightarrow 3 equiv.) in CH_3CN at RT. **Inset:** Corresponding experimental binding isotherms of absorbance at $\lambda = 300, 230$ and 250 nm.

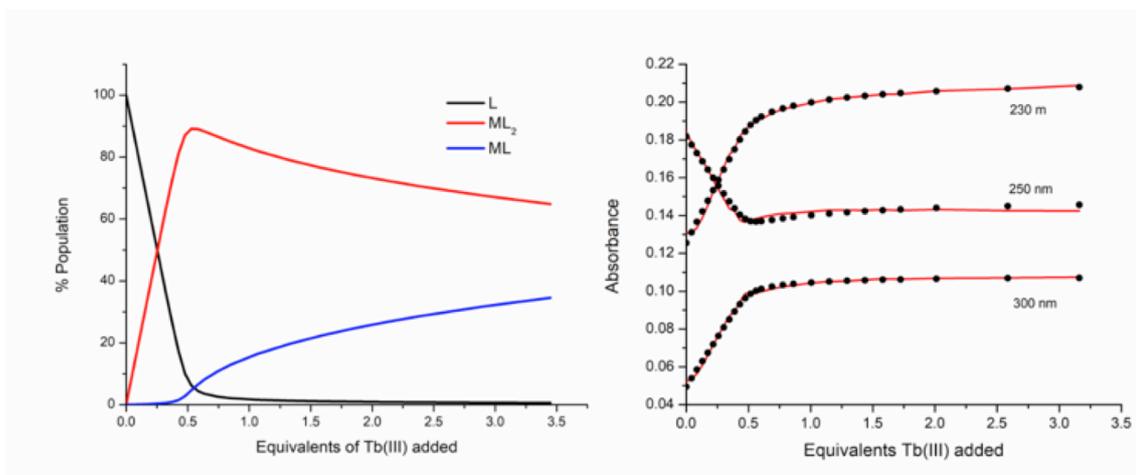


Figure 6S. (left) The speciation distribution diagram obtained from the fit of the UV-visible absorption titration data of ligand **2** against Tb(CF₃SO₃)₃ in CH₃CN. (right) The fit of the experimental binding isotherms using non-linear regression analysis software ReactLab.

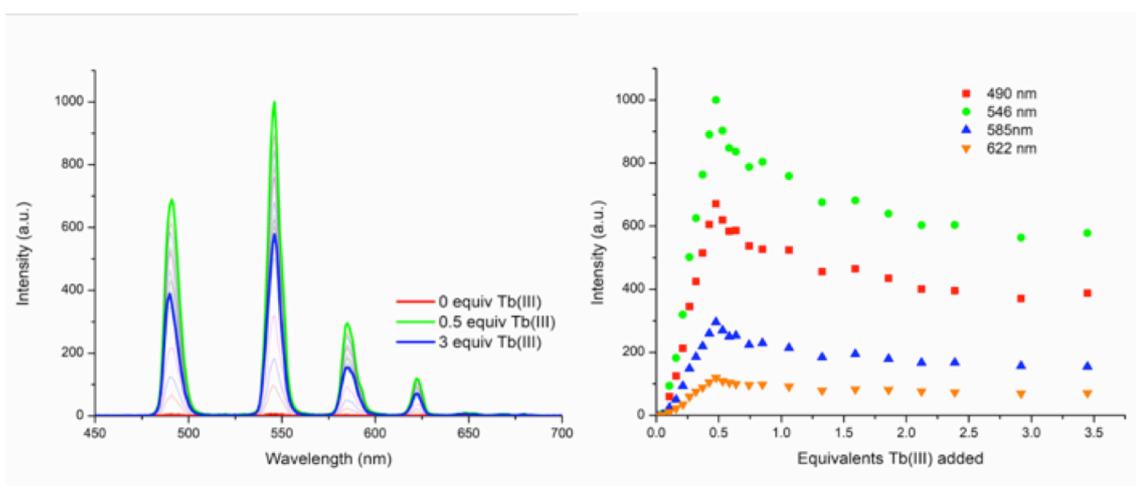


Figure 7S. The overall changes in the (left) UV-visible absorption spectra and (right) fluorescence emission spectra (excitation wavelength $\lambda = 250$ nm) upon titrating **2** (1×10^{-5} M) against Tb(CF₃SO₃)₃ (0→3 equiv.) in CH₃CN at RT. **Inset:** corresponding experimental binding isotherms of absorbance at $\lambda = 300$, 230 and 250 nm.

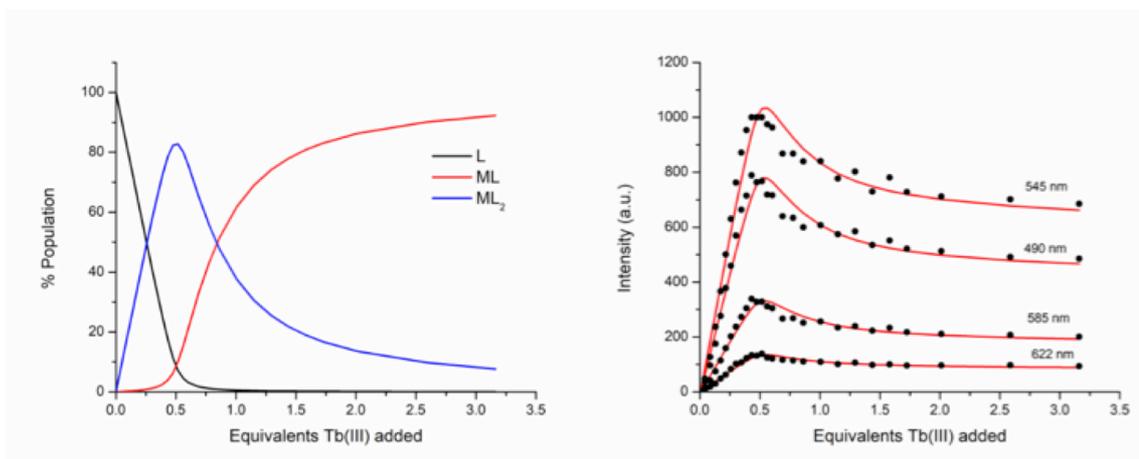


Figure 8S. (left) The speciation distribution diagram obtained from the fit of the time-gated luminescence titration data of ligand **2** against Tb(CF₃SO₃)₃ in CH₃CN. (right) The fit of the experimental binding isotherms using non-linear regression analysis software ReactLab.

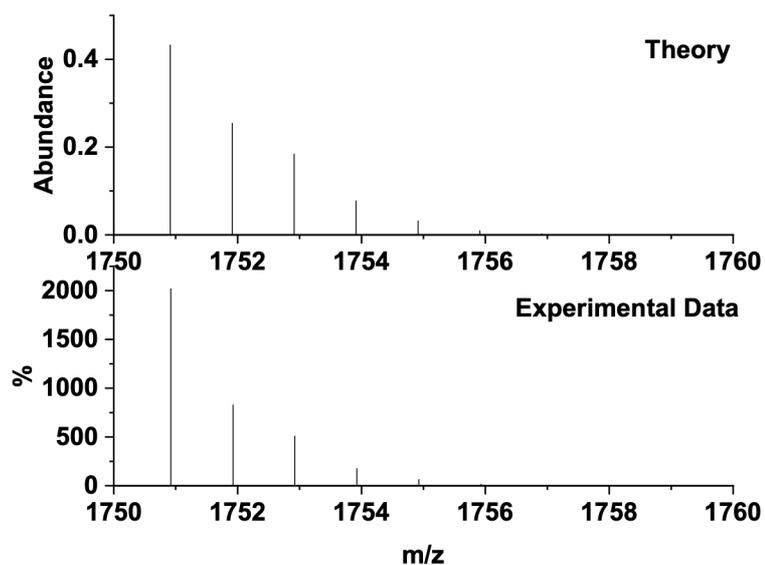


Figure 9S. The HRMS of **1** in the presence of Tb(III) showing the **2M:L**: C₄₅H₃₆F₁₅N₁₀O₁₇S₅Tb₂ m/z = 1750.9125, found 1750.9208.

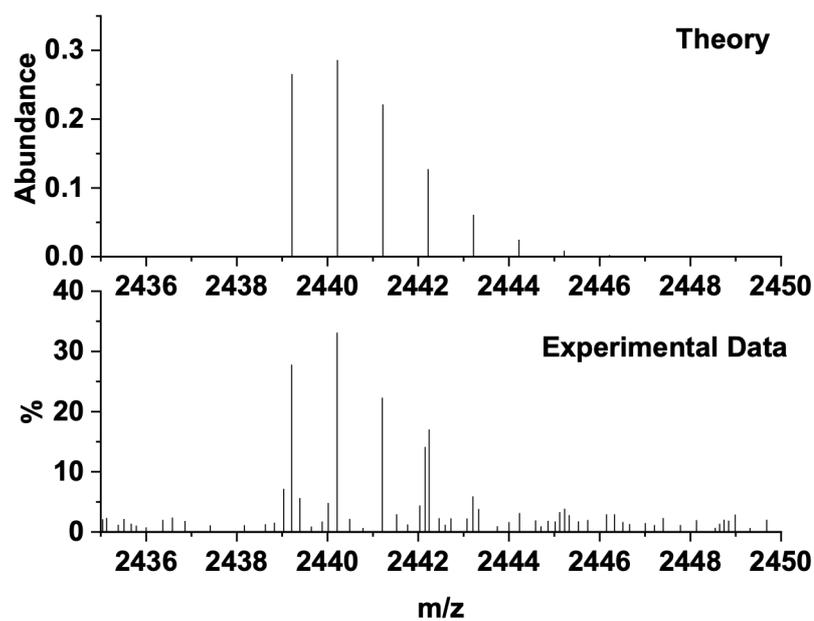


Figure 10S. The HRMS of **1** in the presence of Tb(III) showing the **2M:2L:** $C_{85}H_{72}F_{15}N_{20}O_{19}S_5Tb_2$ $m/z = 2439.2148$, found 2439.2107 .