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# Ferromagnetism induced by in-plane strain in a bulk VS2 based superlattice: (LiOH)0.1VS2

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## Experimental and Methods Section Sample synthesis.

 $K_x VS_2 Crystal$ : V and S powder were ground with a reactive flux  $K_2S_2$ . The molar ration of  $K_2S_2$ :V:S was kept constant at about 0.5: 1: 1. After all the starting materials were thoroughly mixed, they were loaded into a Al<sub>2</sub>O<sub>3</sub> crucible and then sealed in a quartz tube. The quartz tube was placed in a muffle furnace, the furnace was heated at a rate of 20 degree/min to 1050 °C, then kept at 1050 °C for 5 hours, and then cooled to 550 °C at a rate of 3-10 °C/hour Following natural cooling to room temperature, silver needle-like single crystal  $K_x VS_2$  was obtained.

**1T-VS<sub>2</sub> Crystal**:  $K_xVS_2$  crystals and deionized water were placed in a 25-mL Teflon-lined autoclave with 60% of its capacity. The autoclave was then tightly closed and heated at 140 °C for 7 days, the sliver 1T-VS<sub>2</sub> crystals were filtered after washing with deionized water.

 $(LiOH)_{0.1}VS_2$  Crystal: K<sub>x</sub>VS<sub>2</sub> crystals and 2g LiOH•H<sub>2</sub>O were placed in a 25-mL Teflon-lined autoclave filled 60% of its capacity with deionized water. The autoclave was then tightly closed and heated at 140 °C for 7 days, the sliver  $(LiOH)_{0.1}VS_2$  crystals were filtered after washing with deionized water.

#### Characterization

Powder X-ray diffraction (PXRD) patterns were collected at room temperature on a Rigaku smart Lab X-ray diffractometer operated at 40 kV voltage and 40 mA current using Cu Ka radiation ( $\lambda$ =1.5406Å). The 20 range was 10–80° with a step size of 0.01. Indexing and Rietveld refinement were performed using the DICVOL91, Fullprof, and MDI Jade programs. Single crystal X-ray diffraction (SCXRD) patterns at 295 K were collected using a Bruker D8 VENTURE PHOTO II diffractometer with multilayer mirror monochromatized Mo K $\alpha$  ( $\lambda$  = 0.71073 Å) radiation. Unit cell refinement and data merging were performed using the SAINT program, and an absorption correction was applied using Multi-Scans scanning. Structural solutions were obtained by intrinsic phasing methods using the program APEX3, and the final refinement was completed with the Jana 2020 suite of programs. The electron density maps of the two samples were firstly constructed by the charge flipping method implemented in the Jana2020 software. Scanning electron microscopy (SEM) images were taken on a Phenom pro XL microscope equipped with an electron microprobe analyzer for the semiquantitative elemental analysis in the energy-dispersive X-ray spectroscopy (EDS) mode.

#### The density functional theory (DFT) calculations

The density functional theory (DFT) calculations were performed within the Vienna ab initio simulation package<sup>i</sup>. We adopted the generalized gradient approximation (GGA) in the form of Perdew-Burke-Ernzerhof (PBE) for the exchange-correlation potentials<sup>ii</sup>. The projector augmented-wave (PAW) pseudopotentials were used with a plane wave energy of 500 eV;  $3p^63d^44s^1$  for V, and  $3s^23p^4$  for S electron configuration were treated as valence electrons. The van der Waals interaction was considered using DFT-D3 method of Grimme. The GGA+U method with Dudarev's approach was applied to V-d orbitals, where the effective U parameter  $U_{eff} = U$ -J was set as 1 eV. A Monkhorst-Pack<sup>iii</sup> Brillouin zone sampling grid with a resolution of  $0.02 \times 2\pi$  Å<sup>-1</sup> was used. The self-consistent field procedure was considered convergent when the energy difference between two consecutive cycles was lower than 10<sup>-6</sup> eV. The lattice constants and atomic coordinates were relaxed until all the forces on the ions were less than 0.01 eV/Å. For (LiOH)<sub>0.1</sub>VS<sub>2</sub>, LiOH were omitted, and the

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lattice constant were kept constant during the relaxation, which were derived from the experimental results. In calculating magnetic anisotropy energy calculation, a cutoff energy of 700 eV and a denser kspacing of  $0.01 \times 2\pi$  Å<sup>-1</sup> were adopted. And the energy difference between two consecutive cycles in the self-consistent calculation was improved to  $10^{-8}$  eV.

#### **Figure Section**



Fig. S1 SEM image and EDS compositional mapping of K<sub>0.6</sub>VS<sub>2</sub>.



Fig. S2 FC curves obtained with an out-of-plane magnetic field of 1 T for  $(LiOH)_{0.1}VS_2$ .



Fig. S3 FC curves obtained with an out-of-plane magnetic field of 1 T for  $K_{0.6}VS_2$ .

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 $\label{eq:1.08} \begin{array}{ll} \mbox{meV/atom} & 11.51\mbox{ meV/atom} & 10.18\mbox{ meV/atom} & 8.62\mbox{ meV/atom} \\ \mbox{Fig. S4 The formation energy of different magnetic states in bulk $1T-VS_2$.} \end{array}$ 



Fig. S5 The formation energy of different magnetic states in (LiOH)<sub>0.1</sub>VS<sub>2</sub>.



Fig. S6 The definition of the structures used in DFT calculations.

#### **Table Section**

Table S1. Molar ratio for Li, V, S via ICP-AES method.

Table 51: Woral Table for El, V, 5 Via Tel 74E5 metriod.						
Element	Li	V	S			
Molar ration	0.11	0.96	2			

Table S2. Crystallographic data of (LiOH)<sub>0.1</sub>VS<sub>2</sub>.

Formula		S10 V4.92 O 0.484 Li 0.476					
Formula	Weight	575.95					
Space gro	oup	P - 1					
a, c(Å)	-	5.6557(6), 8.6279(2), 9.0551(7)					
Ζ		1					
Crystal si	ze(µm <sup>3</sup> )	87 * 64 * 23					
Tempera	ture(K)	298					
Radiation	ı(Å)	Mo-Ka $\lambda = 0.71073$					
$R_1$ , w $R_2$ , S	S	0.0671, 0.1544, 1.1000					
Atomic positions and equivalent displacement parameters							
Atom	Muti.	x	y	Z	Occ.	U <sub>iso</sub>	
V1	2	0.2013(4)	0.6001(3)	0.4997(3)	0.99(2)	0.0238(3)	
V2	2	-0.3989(3)	0.8002(2)	0.49931(19)	0.98(7)	0.0240(5)	
V3	1	0	1	0.5	0.98(3)	0.0240(6)	
<b>S1</b>	2	0.4782(4)	0.4326(3)	0.3362(3)	1	0.0181(5)	
<b>S2</b>	2	-0.3217(4)	1.0331(3)	0.3359(3)	1	0.0184(5)	
<b>S3</b>	2	-0.0775(4)	0.7665(3)	0.6644(3)	1	0.0177(5)	
<b>S4</b>	2	-0.1217(4)	0.6330(3)	0.3356(3)	1	0.0163(5)	
<b>S</b> 5	2	0.2785(4)	0.8329(3)	0.3362(3)	1	0.0183(5)	
01	2	0.2519(2)	0.7004(8)	-0.0233(5)	0.242(3)	0.059(3)	
Li	2	0.7116(3)	0.6118(6)	0.0023(1)	0.238(6)	0.068(2)	

Table S3. Selected bond length and bond angle for  $(LiOH)_{0.1}VS_2$ ,  $1T-VS_2$  (PDF card- 04-003-2308) and LiOH (PDF card- 00-032-0564).

00 032 030 1).	(LiOH)01VS2	1T-VS <sub>2</sub>	LiOH
Bond length (Å	(11011)0.1 + 22		21011
V-S	2.375(3)	2.34986(1)	
Li-O	2.481(3)		2.01296(6)
Bond angle (de	eg)		
S-V-S	86.39(9)	86.4274(0)	

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<sup>&</sup>lt;sup>i</sup>. Kresse, G., Furthmuller, J., Efficiency of ab-initio total energy calculations for metals and semiconductors using a plane-wave basis set. *Comput. Mater. Sci.* 6 (1), 15-50 (1996).

<sup>&</sup>lt;sup>ii</sup>. Perdew, JP., Burke, K., Ernzerhof, M., Generalized gradient approximation made simple. *Phys. Rev. Lett.* **77**, 3865-3868 (1996).

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