## Cage-like structures based on constrained cyclic arylopeptoids

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## I-Chemicals and general experimental information

Chemicals: Rink acid resin ( $0.62 \mathrm{mmol} / \mathrm{g}$ loading) and 2-chlorotrityl resin ( $1.06 \mathrm{mmol} / \mathrm{g}$ loading) was obtained from Novabiochem, dichloromethane from Carlo Erba; 3-chloromethylbenzoylchloride, 2chloromethylbenzoylchloride, TFA, DIPEA, 4-DMAP isopropyl and propargyl amines from TCI; DMSO from Acros. 1,4-bis(azidomethyl)benzene was synthetized according to literature procedures. 10 mL jacketed reactors were purchased from Kamush and thermo-regulated using a Lauda thermostat. Purification was performed on a Buchi Pure Chromatography system.

NMR was recorded on a Bruker AC-400 spectrometer, operating at 400 MHz for ${ }^{1} \mathrm{H}$ and 100 MHz for ${ }^{13} \mathrm{C} \mathrm{J}$-modulation or a Bruker AC-500 spectrometer operating at 500 MHz for ${ }^{1} \mathrm{H}$ and 125 MHz for ${ }^{13} \mathrm{C}$.

High-resolution mass spectra (HRMS) are recorded using electrospray ionization in positive mode (ESI+) on or a Q Exactive Quadrupole-Orbitrap Mass Spectrometer or on a Qtof-micro WATERS (3000V) spectrometer. Liquid chromatography mass spectroscopy (LC-MS) were recorded on a Q Exactive Quadrupole-Orbitrap Mass Spectrometer coupled to a UPLC Ultimate 3000 (Kinetex EVO C18; 1,7 $\mu$ m; $100 \mathrm{~mm} \times 2,1 \mathrm{~mm}$ column with a flow rate of $0.45 \mathrm{ml} . \mathrm{min}-1$ with the following gradient: a linear gradient of solvent B from $5 \%$ to $95 \%$ over 7.5 min (solvent $A=H 2 O+0.1 \%$ formic acid, solvent $B=$ acetonitrile $+0.1 \%$ formic acid) equipped with a DAD UV/VIS 3000 RS detector) or on a Qtof-micro WATERS (3000V) with electrospray ionization coupled to a HPLC ALLIANCE WATERS system with a diode array detector (DAD) using a reverse-phase C18 Nucleosil column ( $100 \mathrm{~mm} \times 2.1 \mathrm{~mm}, 5 \mu \mathrm{~m}$ pore size) with an $\mathrm{H} 2 \mathrm{O} /$ acetonitrile gradient and a flow rate of $0.2 \mathrm{~mL} / \mathrm{min}$.

Analytical HPLC was recorded on a Hitachi liquid chromatograph (Oven 5310, $30^{\circ} \mathrm{C}$; Pump 5160; DAD detector 5430) equipped with a C18 Acclaim column ( $4.6 \mathrm{~mm} \times 250 \mathrm{~mm}, 5 \mu \mathrm{~m}, 120 \AA$ ). Detection wavelength was 240 nm or 280 nm and flow rate $0.5 \mathrm{~mL} / \mathrm{min}$. Gradient elution used (A) water $/ 0.1 \%$ TFA; (B) methanol according Method A: (Solvents A/B: 0 to 5 minutes isocratic at 95/5; 5 to 25 minutes gradient to $5 / 95 ; 25$ to 35 minutes isocratic at $5 / 95 ; 35$ to 45 minutes gradient to $95 / 5 ; 45$ to 50 minutes 95/5) or Method B (Solvents A/B: 0 to 5 minutes isocratic at 95/5; 5 to 10 minutes gradient to $75 / 25 ; 10$ to 50 minutes gradient to $40 / 60 ; 50$ to 65 minutes gradient to $5 / 95$; 65 to 70 minutes isocratic at 5/95; 70 to 80 minutes gradient to $95 / 5$ ).

## II. General Procedures and compounds characterization

## II.1. Linear arylopeptoids

## II.1.a. Linear ortho-hexamer la


a) Swelling, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$; b) 2-chloromethyl benzoylchloride, DIPEA, DMAP; $\mathrm{CH}_{2} \mathrm{Cl}_{2} ;$ c) Isopropyl amine or propargyl amine, dmso; d) 2-chloromethyl benzoylchloride, DIPEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2} ; \mathrm{e}$ ) Cleavage: TFA/CH2 $\mathrm{Cl}_{2}$ (20:80).

Scheme S1. Solid-phase submonomer synthesis of la on Rink acid resin
1-For 100 mg of resin, swelling: 2 ml of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at RT for 10 min
2- 2-chloromethylbenzoylchloride (3 equiv. per mmol loading), 4-DMAP (3 equiv. per mmol loading) DIPEA ( 3 equiv. per mmol loading) dissolved in $1 \mathrm{mLCH}_{2} \mathrm{Cl}_{2}$ were added at RT , shaken 10 minutes. This step is repeated. The liquid was drained off, then the resin was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \times 2 \mathrm{ml})$, then with DMSO ( $5 \times 2 \mathrm{ml}$ ).

3- Isopropyl amine ( 20 equiv per mmol loading) dissolved in 0.5 mL of DMSO were added. The reaction was shaken for 1 h at RT. The liquid was drained off, then the resin was washed with DMSO ( $5 \times 2 \mathrm{ml}$ ), then with $\mathrm{CH}_{2} \mathrm{CL}_{2}(5 \times 2 \mathrm{ml})$.

4- 2-chloromethylbenzoylchloride (3 equiv. per mmol loading), DIPEA (6 equiv. per mmol loading) dissolved in $1 \mathrm{mLCH} \mathrm{Cl}_{2}$ were added at RT , shaken 10 minutes. The liquid was drained off, then the resin was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \times 2 \mathrm{ml})$, then with DMSO ( $5 \times 2 \mathrm{ml}$ ).

5- Isopropyl amine or propargyl amine ( 20 equiv per mmol loading, 1 M ) dissolved in 0.5 mL of DMSO were added. The reaction was shaken for 1 h at RT. The liquid was drained off, then the resin was washed with DMSO ( $5 \times 2 \mathrm{ml}$ ), then with $\mathrm{CH}_{2} \mathrm{CL}_{2}(5 \times 2 \mathrm{ml})$.

Steps 4 and 5 were repeated to grow the targeted arylopeptoid oligomer until the expected sequence length.

6-The arylopeptoid was cleaved from the resin by the addition of 1 ml of $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{TFA}$ (8:2) for 30 min at RT. The solution was collected and then the resin was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \times 2 \mathrm{ml})$. The solution was evaporated under reduced pressure. The product was purified on Buchi LC auctomatic C18 column (Water $+0.1 \%$ TFA/MeOH) affording $90 \%$ of product la (HPLC purity 96\%).

HRMS (TOF MS ES+): $m / z$ calcd for $\mathrm{C}_{66} \mathrm{H}_{73} \mathrm{~N}_{6} \mathrm{O}_{7}[\mathrm{M}+\mathrm{H}]^{+}: 1061.55352$; found: 1061.5520 (-1.43 ppm).


Figure S1. HPLC chromatogram of the pure la.


Figure S2. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ in $\mathrm{CDCl}_{3}$ of the pure la.


Figure S3. LCMS spectrum of the crude la.


Figure S4. IR spectra of la.

## II.1.b. Linear meta-hexamer.Ib


a) Swelling, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$; b) 3-chloromethyl benzoic acid, DIPEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$; c) Isopropyl amine or propargyl amine, dmso, $50^{\circ} \mathrm{C}$; d) 3-chloromethyl benzoiylchloride, DIPEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2} ;$ e) Cleavage: TFA/CH2Cl $\mathrm{Cl}_{2}(2: 8)$.

Scheme S2. Synthesis of linear meta-arylopeptoid Ib
1-For 100 mg of resin, swelling: 2 ml of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at RT for 10 min
2- 3-chloromethylbenzoic acid (1.2 equiv. per mmol loading), DIPEA ( 5 equiv. per mmol loading) dissolved in $1 \mathrm{mLCH} \mathrm{Cl}_{2}$ were added at RT and shaken 40 minutes. This step is repeated. The liquid was drained off, then the resin was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \times 2 \mathrm{ml})$, then with DMSO ( $5 \times 2 \mathrm{ml}$ ).

3- Isopropyl amine or propargyl amine ( 20 equiv per mmol loading, 1 M ) dissolved in 0.5 mL of DMSO were added. The reaction was shaken for 1 h at $50^{\circ} \mathrm{C}$. The liquid was drained off, then the resin was washed with DMSO ( $5 \times 2 \mathrm{ml}$ ), then with $\mathrm{CH}_{2} \mathrm{CL}_{2}(5 \times 2 \mathrm{ml})$.

4- 3-chloromethylbenzoylchloride (3 equiv. per mmol loading), DIPEA (6 equiv. per mmol loading) dissolved in $1 \mathrm{mLCH} \mathrm{Cl}_{2}$ were added at RT , shaken 10 minutes. The liquid was drawn down, then the resin was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \times 2 \mathrm{ml})$, then with DMSO $(5 \times 2 \mathrm{ml})$.

5- Isopropyl amine or propargyl amine (20 equiv per mmol loading) dissolved in 0.5 mL of DMSO were added. The reaction was shaken for 1 h at $50^{\circ} \mathrm{C}$. The liquid was drained off, then the resin was washed with DMSO ( $5 \times 2 \mathrm{ml}$ ), then with $\mathrm{CH}_{2} \mathrm{CL}_{2}(5 \times 2 \mathrm{ml})$.

Steps 4 and 5 were repeated to grow the targeted arylopeptoid oligomer until the expected sequence length.

6-The arylopeptoid was cleaved from the resin by the addition of 1 ml of $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{TFA}$ (8:2) for 30 min at RT. The solution was collected and then the resin was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \times 2 \mathrm{ml})$. The solution was evaporated under reduced pressure. The product was purified on LC Buchi C18 column (water $+0.1 \% \mathrm{TFA} / \mathrm{MeOH}$ ) affording 93\% of product lb (HPLC purity 96\%).

HRMS (TOF MS ES+): $m / z$ calcd for $\mathrm{C}_{66} \mathrm{H}_{73} \mathrm{~N}_{6} \mathrm{O}_{7}[\mathrm{M}+\mathrm{H}]^{+}: 1061.55352$; found: 1061.5536 (0.06 ppm).


Figure S5. HPLC chromatogram of pure lb.


Figure $\mathbf{S 6} \mathbf{I}^{1} \mathrm{H}-\mathrm{NMR}$ spectrum in $\mathrm{CDCl}_{3}$ of pure $\mathbf{l b}$.


Figure S7. LCMS spectrum of the crude lb.


Figure S8. IR spectra of pure Ib.

## II.2. Cyclisation

II.2.a. ortho-cyclohexamer IIa.


The ortho-hexamer la was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{mmol} / \mathrm{L})$, then 5 equiv. of DIPEA was added followed by 1.2 equiv. of HATU. The reaction was stirred overnight at RT. The solvent was evaporated under reduced pressure and then the residue was evaporated with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 20 \mathrm{ml})$. The residue is then
dissolved in EtOAc ( 20 ml ), extracted with $\mathrm{NaHCO}_{3}(2 x 10 \mathrm{ml})$ then brine ( $1 \times 10 \mathrm{ml}$ ). The organic layer was dried over $\mathrm{MgSO}_{4}$, filtered then evaporated under reduced pressure. The product was purified on C 18 column (Water $+0.1 \% \mathrm{TFA} / \mathrm{MeOH}$ ) affording macrocycle lla in $70 \%$ yield (HPLC purity $95 \%$ ).

HRMS (TOF MS ES+): $m / z$ calcd for $\mathrm{C}_{66} \mathrm{H}_{71} \mathrm{~N}_{6} \mathrm{O}_{6}[\mathrm{M}+\mathrm{H}]^{+}: 1043.54296$; found: 1043.5422 (-0.69 ppm).


Figure S9. HPLC chromatogram of the pure ortho-macrocycle (IIa).


Figure S10. ${ }^{1} \mathrm{H}$-NMR spectrum in $\mathrm{CDCl}_{3}$ of pure ortho-macrocycle (IIa).


Figure S11. COSY spectrum in $\mathrm{CDCl}_{3}$ of pure ortho-macrocycle (Ila).


Figure S12. LCMS spectra of ortho-macrocycle (Ila).


Figure S13. IR spectra of ortho-macrocycle (Ila).
II.2.b. meta-cyclohexamer IIb.


The meta-hexamer lb was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{mmol} / \mathrm{L})$, then 5 equiv. of DIPEA was added followed by 1.2 equiv. of HATU. The reaction was stirred overnight at RT. The solvent was evaporated under reduced pressure and then the residue was evaporated with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 20 \mathrm{ml})$. The residue is then
dissolved in EtOAc ( 20 ml ), extracted with $\mathrm{NaHCO}_{3}(2 \times 10 \mathrm{ml})$ then brine ( $1 \times 10 \mathrm{ml}$ ). The organic layer was dried over $\mathrm{MgSO}_{4}$, filtered then evaporated under reduced pressure. The product was purified on C18 column (Water $+0.1 \% \mathrm{TFA} / \mathrm{MeOH}$ ) affording macrocycle llb in $81 \%$ yield (HPLC purity $95 \%$ ).

HRMS (TOF MS ES+): $m / z$ calcd for $\mathrm{C}_{66} \mathrm{H}_{71} \mathrm{~N}_{6} \mathrm{O}_{6}[\mathrm{M}+\mathrm{H}]^{+}: 1043.54296$; found: 1043.5427 ( -0.23 ppm ).


Figure S14. HPLC of pure meta-macrocycle (IIb).


Figure S15. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ in $\mathrm{CDCl}_{3}$ of pure meta-macrocycle (IIb).


Figure S16. COSY spectra in $\mathrm{CDCl}_{3}$ of pure meta-macrocycle (IIb).


Figure S17. LCMS spectrum of the crude meta-macrocycle (IIb).


Figure S18. IR spectra of meta-macrocycle (IIb).

## II.3. CuAAC reactions

High dilution procedure: Macrocycle Ila or IIb ( 1.0 equiv.) was dissolved in methanol ( $\mathrm{C}=1.0 \mathrm{mM}$ ). 1,4bis(azidomethyl) benzene ( 1.1 equiv.) and catalyst III ( $10 \mathrm{~mol}-\%$ per alkyne) were added. The reaction was stirred for 2 days at room temperature and the solvent was evaporated under reduced pressure.

High concentration procedure for ortho series: Macrocycle lla (1.0 equiv.) was dissolved in methanol $/ \mathrm{CH}_{2} \mathrm{Cl}_{2}(1 / 1)$ at concentration of 0.1 M . 1,4-bis(azidomethyl) benzene (1.1 equiv.) and catalyst III ( $10 \mathrm{~mol}-\%$ per alkyne) were added. The reaction was stirred for 2 days at room temperature and the solvent was evaporated under reduced pressure.

High concentration procedure for meta series: Macrocycle IIb was dissolved in methanol ( $C=0.1 \mathrm{M}$ ). 1,4-bis(azidomethyl) benzene ( 1.1 equiv.) and catalyst III ( $10 \mathrm{~mol} \%$ per alkyne) were added. The reaction was stirred for 2 days at room temperature and the solvent was evaporated under reduced pressure.

## II.3.a. Compound IVa.



High dilution procedure performed on 95 mg ( 0.09 mmol ) of macrocycle lla. HRMS of the crude indicates the presence of crown-like IVa as major compound and tube-like Va as traces. The product IVa was purified by column chromatography ( $\mathrm{EtOAc} / \mathrm{MeOH} 90: 10$ then $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95: 5$ to 70:30) followed by C 18 purification ( $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}+0.1 \% \mathrm{TFA}$ ) furnishing IVa ( $86 \mathrm{mg}, 78 \%, 98 \% \mathrm{HPLC}$ purity).

High concentration procedure performed on $50 \mathrm{mg}(0.045 \mathrm{mmol})$ of la. HRMS of the crude also indicates the presence of crown-like IVa as major compound and tube-like Va as traces. The product IVa was purified by column chromatography (EtOAc/MeOH 90:10 then $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95: 5$ to 70:30) followed by C 18 purification ( $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}+0.1 \% \mathrm{TFA}$ ) furnishing IVa ( $40 \mathrm{mg}, 76 \%, 98 \% \mathrm{HPLC}$ purity).

HRMS (TOF MS ES+): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{74} \mathrm{H}_{79} \mathrm{~N}_{12} \mathrm{O}_{6}[\mathrm{M}+\mathrm{H}]^{+}: 1231.624$; found: 1231.6248 ( 0.61 ppm )


Figure S19. HPLC of the pure ortho compound IVa.


Figure S20. LC-MS spectra of the pure ortho compound IVa.


Figure S21. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ in $\mathrm{CDCl}_{3}$ of the pure ortho compound IVa.


Figure S22. COSY spectra in $\mathrm{CDCl}_{3}$ of the pure ortho compound IVa.


Figure S23. IR spectra of ortho compound IVa.
II.3.b. Compounds IVb and Vb.



High dilution procedure performed on $200 \mathrm{mg}(0.19 \mathrm{mmol})$ of macrocycle llb furnishing 103 mg of cupola-like compound IVb ( $44 \%$ yield, $98 \%$ HPLC purity) and 88 mg of tube-like compound $\mathbf{V b}$ ( $37 \%$ yield, $98.5 \%$ HPLC purity).

High concentration procedure performed on 100 mg ( 0.1 mmol ) of llb furnishing 48 mg of cupola-like compound $\mathbf{I V b}$ ( $39 \%$ yield, $98 \%$ purity) and 50 mg of tube-like compound $\mathbf{V b}$ ( $41 \%$ yield, $98 \%$ purity).

Purification procedure: The products were purified by column chromatography (EtOAc/MeOH 90:10 then $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95: 5$ to 70:30) affording mixture of cupola-like IVb and tube-like Vb. Further C18 purification ( $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}+0.1 \% \mathrm{TFA}$ ) was performed for separation of $\mathbf{I V b}$ and $\mathbf{V b}$.

Compound IVb:
HRMS (TOF MS ES+): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{74} \mathrm{H}_{79} \mathrm{~N}_{12} \mathrm{O}_{6}[\mathrm{M}+\mathrm{H}]^{+}$: 1231.624 ; found: 1231.6257 (1.4 ppm).


Figure S24. HPLC chromatogram of the pure meta compound IVb.


Figure S25. LCMS spectrum of crude meta compound IVb.


Figure S26. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectra in $\mathrm{CDCl}_{3}$ of the pure meta compound IVb.


Figure $\mathbf{S 2 7 .}$ COSY spectra in $\mathrm{CDCl}_{3}$ of the pure meta compound IVb.


Figure S28. Infra-red of meta compound IVb.
Compound Vb:
HRMS (TOF MS ES + ): $m / z$ calcd for $\mathrm{C}_{148} \mathrm{H}_{158} \mathrm{~N}_{24} \mathrm{O}_{12}[\mathrm{M}+2 \mathrm{H}]^{2+}$ : 1231.624 ; found: 1231.6252 (1.01 ppm).


Figure S29. LCMS spectra of the pure meta compound Vb.


Figure S30. ${ }^{1} \mathrm{H}$-NMR spectra in $\mathrm{CDCl}_{3}$ of the pure meta compound $\mathbf{V b}$.


Figure S31. COSY in $\mathrm{CDCl}_{3}$ of the pure meta compound $\mathbf{V b}$.


Figure S32. IR spectra of pure meta compound Vb.

## II.4. Triazolium formation

The methylation of cupola-like compounds IVa and IVb was performed in ace pressure tube in pure Mel at $70^{\circ} \mathrm{C}$ for 24 hours and produced the double methylated products VIa and VIb in quantitative yields.
II.4.a. ortho cupola-like bis-triazolium VIa.


HRMS (TOF MS ES+): $m / z$ calcd for $\mathrm{C}_{76} \mathrm{H}_{84} \mathrm{~N}_{12} \mathrm{O}_{6}[\mathrm{M}]^{2+}: 630.33129$; found: 630.3325 (1.95 ppm).


Figure S33. HPLC spectra of the crude ortho bis-triazolium compound VIa.


Figure S34. ${ }^{1} \mathrm{H}$-NMR spectra in $\mathrm{CDCl}_{3}$ of the crude ortho bis-triazolium compound VIa.


Figure S35. COSY spectra in $\mathrm{CDCl}_{3}$ of the crude ortho bis-triazolium compound VIa.


Figure S36. LCMS of the crude ortho bis-triazolium compound VIa.
II.4.b. Meta cupola-like triazolium VIb.

$m_{\text {crude }}=40 \mathrm{mg}$ (HPLC purity 96\%), isolated yield 98\%
HRMS (TOF MS ES + ): $m / z$ calcd for $\mathrm{C}_{82} \mathrm{H}_{90} \mathrm{~N}_{12} \mathrm{O}_{6}[\mathrm{M}]^{2+}$ : 669.35477; found: 669.3538 (-1.41 ppm).


Figure S37. HPLC chromatogram of meta bis-triazolium compound VIb.


Figure S38. LCMS spectrum of meta bis-triazolium compound VIb.


Figure S39. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ in $\mathrm{CDCl}_{3}$ of the pure meta bis-triazolium compound VIb.


Figure S40. IR spectra of meta bis-triazolium compound VIb.

III-Conformational studies
III.1. NMR study


Figure S41. Comparison of ${ }^{1} \mathrm{H}$ NMR spectra of Ila (red curve) and Ilb (blue curve) in $\mathrm{CDCl}_{3}$ at 298 K .


Figure S42. Variable temperature study of macrocycle Ila in DMSO ( 2.7 mg in 0.6 mL ): 298 K (blue curve), 338 K (red curve), 358 K (green curve), 388 K (purple curve), 398 K (yellow curve), back to 298 $K$ (orange curve).


Figure S43. VT study of macrocycle Ila in $\mathrm{CDCl}_{3}$ ( 5 mg in 0.6 mL ): 298 K (blue curve), 288 K (red curve), 278 K (green curve), 268 K (purple curve), 265 K (yellow curve), back to 298 K (orange curve).


Figure S44. COSY experiment of ortho bicyclic compound IVa in $\mathrm{CDCl}_{3}(8-10 \mathrm{mM})$ at 278 K .


Figure S45. Comparison of ${ }^{1} \mathrm{H}$ NMR spectra of ortho bicyclic compound IVa in $\mathrm{CDCl}_{3}$ at 298 K (green curve) and at 268 K (purple curve) and meta bicyclic compound IVb in $\mathrm{CDCl}_{3}$ at 298 K (blue curve) and at 268 K (red curve).


Figure S46. Comparison of ${ }^{1} \mathrm{H}$ NMR spectra of the ortho bis-triazolium bicyclic compound Vla at 298 K in various solvants: $\mathrm{CDCl}_{3}$ (blue curve), $\mathrm{CD}_{3} \mathrm{CN}$ (red curve), $\mathrm{CD}_{3} \mathrm{OD}$ (green curve) and DMSO- $\mathrm{d}_{6}$ (purple curve).


Figure S47. COSY NMR spectra of the ortho bis-triazolium bicyclic compound VIa ( 2 mM ) in $\mathrm{CD}_{3} \mathrm{CN}$ at 288K.


Figure S48. TOCSY NMR spectra of the ortho bis-triazolium bicyclic compound Vla ( 20 mM ) in $\mathrm{CD}_{3} \mathrm{CN}$ at 298 K .


Figure S49. NOESY NMR spectra of the ortho bis-triazolium bicyclic compound VIa ( 2 mM ) in $\mathrm{CD}_{3} \mathrm{CN}$ at 288 K ( $\mathrm{d} 8=500$ ).


Figure S50. NOESY NMR spectra (3-9.5 ppm region) of the ortho bis-triazolium bicyclic compound VIa ( 2 mM ) in $\mathrm{CD}_{3} \mathrm{CN}$ at 288 K ( $\mathrm{d} 8=500$ ).


Figure S51. HSQC NMR spectra of the ortho bis-triazolium bicyclic compound Vla ( 20 mM ) in $\mathrm{CD}_{3} \mathrm{CN}$ at 298K.


Figure S52. HMBC NMR spectra of the ortho bis-triazolium bicyclic compound Vla ( 20 mM ) in $\mathrm{CD}_{3} \mathrm{CN}$ at 298K.


| Protons number | Ha (CH) | $\mathrm{Hb}\left(\mathrm{CH}_{3}\right)$ | $\mathrm{b}^{\prime}\left(\mathrm{CH}_{3}\right)$ | c, $\mathrm{c}^{\prime}$ ABsystem ( $\mathrm{CH}_{2}$ ) |
| :---: | :---: | :---: | :---: | :---: |
| Residue 1 | 3.87 (hept, J = 7.5 Hz | -0.14 (d, J = 7.5 Hz) | 1.03 (d, J = 7.5Hz) | $\begin{aligned} & 4.12(d, J=15.9 \mathrm{~Hz}) \\ & 4.53(d, J=15.9 \mathrm{~Hz}) \end{aligned}$ |
| Residue 2 | 3.72 (hept, $J=6.5 \mathrm{~Hz}$ ) | 0.15 (d, J = 6.5 Hz) | 0.53 (d, J = 6.5 Hz) | $\begin{aligned} & 4.00(d, J=17.7 \mathrm{~Hz}) \\ & 4.59(d, J=17.7 \mathrm{~Hz}) \end{aligned}$ |
| Residue 3 | 3.66 (hept, $J=6.5 \mathrm{~Hz}$ ) | 0.87 (d, J = 6.5 Hz) | 1.23 (d, J = 6.5 Hz) | $\begin{aligned} & 4.16(d, J=16,0 \mathrm{~Hz}) \\ & 5.34(\mathrm{~d}, J=16.0 \mathrm{~Hz}) \end{aligned}$ |
| Residue 4 | 4.06 (m) | 1.38 (d, J = 6.8 Hz) | 1.52 (d, J = 6.4 Hz) | $\begin{aligned} & 4.23(\mathrm{~d}, J=14.9 \mathrm{~Hz}) \\ & 5.07(\mathrm{~d}, J=14.9 \mathrm{~Hz}) \end{aligned}$ |
|  | d, d' ABsystem ( $\mathrm{CH}_{2}$ ) | e, e' ABsystem ( $\mathrm{CH}_{2}$ ) | f, $\mathrm{f}^{\prime}$ ABsystem ( $\mathrm{CH}_{2}$ ) | $\mathrm{g}, \mathrm{g}^{\prime} \mathrm{ABsystem}\left(\mathrm{CH}_{2}\right)\left(\mathrm{CH}_{2}\right)$ |
| Residue 5 | $\begin{aligned} & 4.07(\mathrm{~d}, J=14 \mathrm{~Hz}) \\ & 6.03(\mathrm{~d}, J=14 \mathrm{~Hz}) \end{aligned}$ | $\begin{aligned} & 4.48(\mathrm{~d}, J=18.6 \mathrm{~Hz}) \\ & 4.97(\mathrm{~d}, J=18.6 \mathrm{~Hz}) \end{aligned}$ |  |  |
| Residue 6 |  |  | $\begin{aligned} & 4.16(d, J=18.5 \mathrm{~Hz}) \\ & 4.82(d, J=18.5 \mathrm{~Hz}) \end{aligned}$ | $\begin{aligned} & 4.33(\mathrm{~d}, J=17.5 \mathrm{~Hz}) \\ & 5.77(\mathrm{~d}, J=17.5 \mathrm{~Hz}) \end{aligned}$ |
| Triazolium | $\mathrm{Hh}(\mathrm{CH})$ | $\mathrm{Hi}\left(\mathrm{CH}_{3}\right)$ | Hj (CH) | $\mathrm{Hk}\left(\mathrm{CH}_{3}\right)$ |
| Triazolium 1 | 8.38 (s) | 3.47 (s) |  |  |
| Triazolium 2 |  |  | 9.21 (s) | 4.14 (s) |
| Linker | I, ${ }^{\prime}$ ABsystem ( $\mathrm{CH}_{2}$ ) | m,m'ABsystem ( $\mathrm{CH}_{2}$ ) |  |  |
|  | $\begin{aligned} & 5.35(\mathrm{~d}, J=14 \mathrm{~Hz}) \\ & 5.69(\mathrm{~d}, J=14 \mathrm{~Hz}) \end{aligned}$ | $\begin{aligned} & 5.58(\mathrm{~d}, J=14.5 \mathrm{~Hz}) \\ & 5.92(\mathrm{~d}, J=14.5 \mathrm{~Hz}) \end{aligned}$ |  |  |

Table S1. $\mathrm{H}^{1}$ NMR signals attribution of bis-triazolium bicyclic compound VIa in $\mathrm{CD}_{3} \mathrm{CN}$ at 298K.


Figure S53. Variable temperature NMR spectra of the ortho bis-triazolium bicyclic compound VIa (2 mM ) in CD3CN: 298K (blue curve), 288K (red curve) and 273K (green curve).


Figure S54. Variable temperature NMR of the ortho bis-triazolium bicyclic compound Vla ( 3 mM ) in DMSO-d ${ }_{6}$ : 298 K (blue curve), 338 K (red curve), 358 K (green curve), 378 K (purple curve), 388 K (Yellow curve) and back to 298K (orange curve).


Figure S55. Variable temperature NMR of the meta bis-triazolium bicyclic compound VIb ( 3 mM ) in DMSO-d ${ }_{6}$ : 298 K (blue curve), 358 K (red curve), 388 K (green curve) and back to 298 K (purple curve).
III.2. Computational study


Figure S56. Model structure of ortho-arylopeptoid macrocycle Ila with all amides in the trans conformation and the two propargyl groups on the same side of the ring.

## Computational study of bicyclic bis-triazolium Vla using Conformer-Rotamer Ensemble Sampling Tool $\left(\right.$ CREST) ${ }^{1}$

We first built a possible structure, without a priori, of the bicyclic system with Chimera software. This structure has been geometrically optimized with Gaussian $16^{2}$ at DFT level of theory, using the B3LYPas a hybrid functional and $6-31 \mathrm{~g}$ as a basis for the description of the electrons and acetonitrile as solvent in SCRF model (Table S2). ${ }^{3,4}$ It was not necessary to do an optimization with a more precise electronic basis because only a geometrically valid starting system was needed. the rest of the conformational research being done by semi-empirical methods. CREST program is used using the conformer search module of ChimeraX software. ${ }^{5}$ The charge is set to +2 to respect the two triazoliums and the six amide bonds are constrained in the trans conformation. The APLB implicit solvent model was used with the dielectric constant of acetonitrile $(37,5)$. Energy threshold for energy and RMSD was set by default to 6 kcal and 0.125 . There is no protonation screening. We used the method GFN2-xTB ${ }^{6}$ with no empirical dispersion and with the default integration grid. 198 solutions proposed by CREST have energy in a 6 kcal window (Fig. S57). Structures of the 10 lowest energy solutions are represented in Figure 58.


| Atom | X | Y | Z |
| :--- | :--- | :--- | :--- |
| N | 5.31900 | -1.61200 | 0.04500 |
| C | 4.43800 | -2.80500 | 0.03800 |
| H | 3.39500 | -2.46700 | 0.04600 |
| H | 4.60100 | -3.34300 | 0.98100 |
| C | 6.20500 | -1.37600 | 1.22800 |
| C | 6.66700 | -0.39400 | 1.06100 |


| H | 6.93800 | -3.41100 | 1.60800 |
| :--- | :--- | :--- | :--- |
| H | 8.01700 | -2.11500 | 2.14100 |
| H | 7.90000 | -2.50400 | 0.42000 |
| C | 5.37600 | -1.29500 | 2.52200 |
| H | 4.87000 | -2.24100 | 2.74500 |
| H | 4.61400 | -0.51200 | 2.45100 |
| H | 6.03200 | -1.06300 | 3.36800 |
| C | 4.66100 | -3.77800 | -1.11500 |
| C | 3.59000 | -4.54000 | -1.63300 |
| C | 3.83500 | -5.53600 | -2.59600 |
| H | 3.00700 | -6.13100 | -2.97400 |
| C | 5.12500 | -5.76100 | -3.08000 |
| H | 5.30000 | -6.53300 | -3.82300 |
| C | 6.18300 | -4.97900 | -2.60300 |
| H | 7.19000 | -5.13200 | -2.98000 |
| C | 5.94700 | -4.00400 | -1.62900 |
| H | 6.77700 | -3.40900 | -1.26300 |
| C | 2.19400 | -4.37400 | -1.10300 |
| O | 1.96000 | -4.59700 | 0.12000 |
| N | 1.17800 | -4.06000 | -1.97000 |
| C | -0.18300 | -3.94100 | -1.42000 |
| H | -0.10200 | -3.50100 | -0.41400 |
| H | -0.74000 | -3.23000 | -2.03900 |
| C | 1.37100 | -3.66100 | -3.40200 |
| H | 2.43400 | -3.82200 | -3.62000 |
| C | 0.53800 | -4.55000 | -4.33400 |
| H | -0.53800 | -4.42300 | -4.16300 |
| H | 0.78400 | -5.60900 | -4.20500 |
| H | 0.74300 | -4.27600 | -5.37500 |
| C | 1.07300 | -2.16500 | -3.60100 |
| H | 0.03000 | -1.91600 | -3.37700 |
| H | 1.25300 | -1.89900 | -4.64900 |
| H | 1.73300 | -1.55500 | -2.97800 |
| C | -0.98300 | -5.23700 | -1.34300 |
| C | -2.36200 | -5.17400 | -1.03200 |
| C | -3.12500 | -6.35200 | -0.96300 |
| H | -4.19100 | -6.29300 | -0.75700 |
| C | -2.52800 | -7.59700 | -1.17800 |
| H | -3.12600 | -8.50100 | -1.12800 |
| C | -1.15900 | -7.66600 | -1.45900 |
| H | -0.68300 | -8.62900 | -1.61900 |
| C | -0.39800 | -6.49400 | -1.53800 |
| H | 0.66100 | -6.56100 | -1.76400 |
| C | -3.03700 | -3.83900 | -0.89000 |
| O | -3.10300 | -3.05600 | -1.86900 |
| N | -4.23400 | -3.46500 | 0.32600 |
| C | -2.13800 | 0.40200 |  |
| H | -1.44000 | -0.19100 |  |
| H | -1.80300 | 1.44700 |  |
| H | -4.28400 | 1.55200 |  |
| H | -5.31400 | 1.27500 |  |
|  | -4.30800 | 2.02500 |  |
|  |  |  |  |


| C | -5.67500 | -2.10000 | -0.09100 |
| :--- | :--- | :--- | :--- |
| C | -6.21700 | -0.88500 | -0.56800 |
| H | -7.95900 | 0.09800 | -1.37900 |
| C | -7.55500 | -0.83400 | -0.99300 |
| C | -8.36500 | -1.97100 | -0.93700 |
| H | -9.39700 | -1.91900 | -1.27000 |
| C | -7.83400 | -3.17400 | -0.46000 |
| H | -8.45200 | -4.06600 | -0.41600 |
| C | -6.50100 | -3.23200 | -0.04000 |
| H | -6.10100 | -4.17700 | 0.31200 |
| C | -5.34800 | 0.32900 | -0.78200 |
| O | -4.60300 | 0.38100 | -1.79900 |
| N | -5.37500 | 1.36300 | 0.12000 |
| C | -4.45600 | 2.48800 | -0.13600 |
| H | -4.23900 | 2.98900 | 0.81600 |
| H | -3.50800 | 2.05800 | -0.49700 |
| C | -6.15400 | 1.33600 | 1.39500 |
| H | -6.71300 | 0.39100 | 1.38000 |
| C | -5.21900 | 1.31300 | 2.61800 |
| H | -4.70800 | 2.27100 | 2.76500 |
| H | -4.46000 | 0.52900 | 2.52200 |
| H | -5.80600 | 1.11300 | 3.52100 |
| C | -7.16100 | 2.49400 | 1.47100 |
| H | -7.88200 | 2.45100 | 0.64900 |
| H | -6.66000 | 3.46800 | 1.44200 |
| H | -7.71600 | 2.43100 | 2.41400 |
| C | -4.93100 | 3.52600 | -1.14600 |
| C | -4.05200 | 4.57600 | -1.49500 |
| C | -4.47000 | 5.58500 | -2.37600 |
| H | -3.79100 | 6.39800 | -2.62200 |
| C | -5.74500 | 5.54000 | -2.94900 |
| H | -6.06000 | 6.32300 | -3.63300 |
| C | -6.60100 | 4.47800 | -2.64400 |
| H | -7.58600 | 4.42200 | -3.09800 |
| C | -6.19300 | 3.48100 | -1.74900 |
| H | -6.86600 | 2.66500 | -1.51200 |
| C | -2.68800 | 4.65200 | -0.85900 |
| O | -2.56800 | 5.12100 | 0.30800 |
| N | -1.59500 | 4.20000 | -1.55500 |
| C | -0.29800 | 4.17400 | -0.84500 |
| H | 0.21200 | 3.24600 | -1.11800 |
| H | -0.52700 | 4.13300 | 0.22900 |
| C | -1.65100 | 3.65400 | -2.95100 |
| H | -2.68200 | 3.80600 | -3.28900 |
| C | -1.36100 | 2.14400 | -2.97600 |
| H | -0.34500 | 1.91800 | -2.63400 |
| H | -2.08100 | -4.00600 |  |
| H | 1.58000 | -2.36500 |  |
| C | 4.42800 | -3.89900 |  |
| H | 4.23000 | -3.93000 |  |
| H | 5.50600 | -3.83600 |  |
|  |  |  |  |


| C | 0.59600 | 5.37500 | -1.11800 |
| :--- | :--- | :--- | :--- |
| C | 1.98100 | 5.21900 | -1.34100 |
| C | 2.78500 | 6.34500 | -1.59800 |
| H | 3.84100 | 6.21700 | -1.82000 |
| C | 2.23000 | 7.62800 | -1.59800 |
| H | 2.85800 | 8.48900 | -1.80400 |
| C | 0.86400 | 7.79100 | -1.33800 |
| H | 0.42600 | 8.78500 | -1.32500 |
| C | -1.05900 | 6.67300 | -1.10300 |
| H | 2.57400 | 6.80200 | -0.89900 |
| C | 2.27400 | 3.84300 | -1.43900 |
| O | 3.42100 | 3.09200 | -2.39900 |
| N | 3.92200 | 3.38700 | -0.44500 |
| C | 3.75100 | 2.00100 | -0.55900 |
| H | 3.29200 | 1.49700 | 0.40200 |
| H | 3.87300 | 1.49200 | -1.29900 |
| C | 4.94400 | 4.19500 | 0.70600 |
| H | 3.79200 | 3.99600 | 0.86000 |
| H | 5.38300 | 5.25900 | 0.46600 |
| C | 5.94500 | 1.85400 | -0.97000 |
| C | 7.69900 | 0.55600 | -1.03900 |
| H | 7.28200 | -0.60800 | -1.51000 |
| C | 8.07200 | 0.39200 | -1.43500 |
| C | 9.10400 | 1.50000 | -1.75700 |
| H | 7.51700 | 1.35900 | -2.06300 |
| C | 8.11500 | 2.78100 | -1.70100 |
| H | 6.18200 | 3.64900 | -1.96200 |
| C | 5.76200 | 2.95200 | -1.31300 |
| H | 5.07100 | 3.95200 | -1.29800 |
| C | 4.13100 | -0.66400 | -0.91200 |
| O | 1.98100 | -0.81700 | -1.74000 |
| C | 1.50900 | -0.09800 | 5.25600 |
| C | 0.12600 | 1.22200 | 5.21500 |
| C | -0.77300 | 1.46300 | 5.21500 |
| C | -0.30000 | 0.39600 | 5.28700 |
| C | 1.08200 | -0.92400 | 5.35800 |
| C | 2.48800 | -1.16600 | 5.32700 |
| C | -1.27000 | 2.37400 | 5.27100 |
| C | -1.56000 | -2.06700 | 5.56000 |
| N | 2.59400 | -2.84300 | 4.32300 |
| N | -2.71300 | 3.13500 | 3.99600 |
| C | -2.53100 | -2.85300 | 3.58700 |
| C | -1.24400 | -3.79700 | 2.58300 |
| N | -0.64900 | -4.27200 | 2.77400 |
| N | 1.96500 | -3.70200 | 3.83600 |
| N | 3.38200 | 4.74100 | 2.63900 |
| N | 4.27400 | 3.84600 |  |
| C | 2.87000 | 2.91000 |  |
| C | 3.91100 | 2.01100 |  |
| H | -0.29500 | 5.26000 |  |
| H | 2.48200 | 5.18800 |  |
|  | 0.59400 | 5.31500 |  |


| H | 1.45900 | -2.18300 | 5.38500 |
| :--- | :--- | :--- | :--- |
| H | 2.19100 | 3.10700 | 6.03200 |
| H | 3.50000 | 2.02700 | 5.51000 |
| H | -2.24100 | -1.70600 | 5.91900 |
| H | -0.88200 | -2.78600 | 6.29200 |
| H | -3.57300 | -2.25700 | 3.84600 |
| H | 4.04200 | 2.01700 | 2.86000 |
| C | -0.49100 | -5.29500 | 2.02400 |
| H | -1.18300 | -5.86500 | 1.40500 |
| H | 0.26600 | -4.81900 | 1.38600 |
| H | -0.01000 | -5.96500 | 2.74100 |
| C | 1.69500 | 6.02800 | 2.19900 |
| H | 2.24300 | 6.39000 | 1.32800 |
| H | 0.64200 | 5.89100 | 1.93900 |
| H | 1.78800 | 6.75000 | 3.01500 |

Table S2. Starting point for CREST: Representation (H atom are removed for clarity) and xyz file of the optimized structure of bicyclic bis-triazolium Vla.


Figure S57. Energies of the 198 conformers proposed by CREST

$1 \mathrm{Eh}=-264.675$

$3 \mathrm{Eh}=-264.67316$

$5 \mathrm{Eh}=-264.67207$


$9 \mathrm{Eh}=-264.67150$

$2 \mathrm{Eh}=-264.67487$

$4 \mathrm{Eh}=-264.67244$



8 Eh = -264.67182

$10 \mathrm{Eh}=-264.67147$

Figure S58. Structures of the 10 lowest energy solutions obtained by CREST


Figure S59. ${ }^{1} \mathrm{H}$ NMR of the ortho bis-triazole bicyclic compound IVa ( 3 mM ) in $\mathrm{CD}_{3} \mathrm{CN}$ at 298 K (blue curve), upon addition of $\mathrm{TBAHSO}_{4}: \approx 0.25$ equiv. (red curve), $\approx 0.70$ equiv. (green curve), $\approx 1.05$ equiv. (purple curve) and $\approx 6.0$ equiv. (yellow curve).


Figure S60. ${ }^{1} \mathrm{H}$ NMR of the ortho bis-triazole bicyclic compound $\mathrm{IVa}\left(3 \mathrm{mM}\right.$ ) in $\mathrm{CDCl}_{3}$ at 298 K (blue curve), upon addition of $\mathrm{TBAHSO}_{4}: \approx 0.5$ equiv. (red curve), $\approx 2.5$ equiv. (green curve) and $\approx 9$ equiv. (purple curve).


Figure S61. ${ }^{1 \mathrm{H}}$ NMR of the ortho bis-triazolium bicyclic compound Vla in $\mathrm{CD}_{3} \mathrm{CN}$ at 298 K (blue curve), upon addition of $\mathrm{TBAHSO}_{4}: \approx 1.5$ equiv. (red curve), $\approx 3$ equiv. (green curve), $\approx 5.3$ equiv. (purple curve), $\approx 6$ equiv. (yellow curve) and $\approx 20$ equiv. (orange curve).


Figure S62. ${ }^{1} \mathrm{H}$ NMR of the ortho bis-triazolium bicyclic compound VIa ( 3 mM ) in $\mathrm{CDCl}_{3}$ at 298 K (blue curve), upon addition of $\mathrm{TBAHSO}_{4}: \approx 1.9$ equiv. (red curve), $\approx 2.4$ equiv. (green curve), $\approx 5$ equiv. (purple curve), $\approx 6.5$ equiv. (yellow curve) and $\approx 8$ equiv. (orange curve).


Figure S63. COSY NMR spectra of the ortho bis-triazolium bicyclic compound VIa + 2.25 equiv. $\mathrm{TBAHSO}_{4}$ in $\mathrm{CDCl}_{3}(3 \mathrm{mM})$ at 298 K .


Figure S64. HSQC NMR spectra of the ortho bis-triazolium bicyclic compound Vla +2.25 equiv. $\mathrm{TBAHSO}_{4}$ in $\mathrm{CDCl}_{3}$ (3 mM) at 298K.


Figure S65. HMBC NMR spectra of the ortho bis-triazolium bicyclic compound Vla +2.25 equiv. $\mathrm{TBAHSO}_{4}$ in $\mathrm{CDCl}_{3}$ (3 mM) at 298K.


| Protons number | Ha (CH) | $\mathrm{Hb}\left(\mathrm{CH}_{3}\right)$ | $\mathrm{b}^{\prime}\left(\mathrm{CH}_{3}\right)$ | c, c' ABsystem ( $\mathrm{CH}_{2}$ ) |
| :---: | :---: | :---: | :---: | :---: |
| Residue 1,3 | 3.55 (hept, J = 6.6 Hz | 0.67 (d, J = 6.6 Hz) | 1.23 (d, J = 6.6Hz) | $\begin{aligned} & 4.63(\mathrm{~d}, J=17.3 \mathrm{~Hz}) \\ & 5.52(\mathrm{~d}, J=17.3 \mathrm{~Hz}) \end{aligned}$ |
| Residue 2,4 | 4.08 (hept, J = 6.6 Hz ) | 0.88 (d, J = 6.6 Hz ) | 1.25 (d, J = 6.6 Hz) | $\begin{aligned} & 3.88(\mathrm{~d}, J=17.1 \mathrm{~Hz}) \\ & 5.45(\mathrm{~d}, J=17.1 \mathrm{~Hz}) \end{aligned}$ |
|  | d, d' ABsystem ( $\mathrm{CH}_{2}$ ) | e,e' ABsystem ( $\mathrm{CH}_{2}$ ) |  |  |
| Residue 5,6 | $\begin{aligned} & 4.85(\mathrm{~d}, J=14.9 \mathrm{~Hz}) \\ & 5.12(\mathrm{~d}, J=14.9 \mathrm{~Hz}) \end{aligned}$ | $\begin{aligned} & 4.39(\mathrm{~d}, J=15.5 \mathrm{~Hz}) \\ & 5.17(\mathrm{~d}, J=15.5 \mathrm{~Hz}) \end{aligned}$ |  |  |
| Triazolium | Hh (CH) | $\mathrm{Hi}\left(\mathrm{CH}_{3}\right)$ | Hj (CH) | $\mathrm{Hk}\left(\mathrm{CH}_{3}\right)$ |
| Triazolium 1,2 | 9.21 (s) | 4.46 (s) |  |  |
| Linker | I, ${ }^{\prime}$ ' ABsystem ( $\mathrm{CH}_{2}$ ) |  |  |  |
|  | $\begin{aligned} & 5.25(\mathrm{~d}, J=15.2 \mathrm{~Hz}) \\ & 5.92(\mathrm{~d}, J=15.2 \mathrm{~Hz}) \end{aligned}$ |  |  |  |

Table S3. $\mathrm{H}^{1} \mathrm{NMR}$ signals attribution of bicyclic bis-triazolium compound $\mathrm{VIa}+\mathrm{TBAHSO}_{4}$ in $\mathrm{CD}_{3} \mathrm{CN}$ at 298K.

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