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Electronic Supplementary Information for

Sumanene-stacked supramolecular polymers. Dynamic, solvation-directed control

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Experimental Procedures

Instruments. UV/vis spectra were measured using a JASCO V-560 or -650 spectrometer with a temperature controller. Fluorescence spectra were recorded using an FP-8500 spectrometer equipped with a temperature controller. All solutions were measured in a quartz cell (path lengths of 1, 2, 5, or 10 mm) or a high-pressure cell (path length of 2 mm). Atomic force microscopy (AFM) images were obtained using a Shimadzu SPM-9700HT microscope. Hydrostatic pressure measurements were performed using a custom-built high-pressure apparatus, the details of which have been reported previously.¹ Variable-temperature measurements were performed using a UNISOKU CoolSpek cryostat.

Materials. Spectroscopy grade MCH and CH_2Cl_2 were used. A commercially available highly oriented pyrolytic graphite (HOPG) substrate was used without further treatment. **Sumanene** was synthesized according to literature.²

AFM measurements. A solution of **sumanene** was dropped onto an HOPG surface, pre-dried by N_2 gas flow and then fully dried under high vacuum, prior to observation by AFM.

Spectroscopic measurements. For the spectral analysis, a smoothing process was performed using MATLAB. Specifically, a range of 340–400 nm was extracted from the raw data and then converted to applicable extinction coefficients by smoothing, using the "smoothing spline" method (a smoothing parameter of $5.614035861113226 \times 10^{-6}$).

Computational studies. The density functional theory (DFT) calculations were performed using the long-range and dispersion-corrected ω B97X-D functional³ with 6-311G** basis sets.⁴ All calculations were performed using the Gaussian 16.⁵

¹ H. Mizuno, G. Fukuhara, Acc. Chem. Res. 2022, 55, 1748-1762.

² H. Sakurai, T. Daiko, T. Hirao, *Science* **2003**, *301*, 1878.

³ J.-D. Chai, M. Head-Gordon, *Phys. Chem. Chem. Phys.* **2008**, *10*, 6615-6620.

⁴ R. Krishnan, J. S. Binkley, R. Seeger, J. A. Pople, J. Chem. Phys. 1980, 72, 650-654.

⁵ M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A. V. Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian, J. V. Ortiz, A. F. Izmaylov, J. L. Sonnenberg, D. Williams-Young, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. J. Bearpark, J. J. Heyd, E. N. Brothers, K. N. Kudin, V. N. Staroverov, T. A. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. P. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M. Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas, J. B. Foresman, D. J. Fox, Gaussian, Inc. **2016**, Wallingford CT.

SAXS measurements. Small-angle X-ray scattering (SAXS) measurements were performed at the BL-6A beamline^{6,7} at the Photon Factory of the High Energy Accelerator Research Organization (KEK) in Tsukuba, Japan. The **sumanene** solution was loaded into a custom-built aluminum cell sealed with polypropylene films. The optical path length is set to 18 mm. Because of its strong absorption of X-rays, CH₂Cl₂ was not used in the SAXS measurements. X-ray wavelength of 1.5 Å and a sample-detector distance of 523 mm (calibrated using silver behenate) gave a detectable *q* range of the order of 0.2 to 9.7 nm⁻¹. All data were analyzed using the SAngler software package.⁸

Variable temperature measurements. The measured temperature was decreased from 60 °C to - 50 °C at -1 °C/min. To prevent frost, the cell chamber in the spectrometer was replaced with Ar, and a cell containing the **sumanene** solution was inserted and measured. Dry air flowed continuously during the measurements. However, accurate data could not be obtained below 0 °C due to the frost effect.

High-pressure measurements. Since the optical path length is fixed at 2 mm in the high-pressure cell, a 7:3 (v/v) MCH-CH₂Cl₂ mixture was used from the viewpoint of extinction coefficient.

⁶ N. Shimizu, T. Mori, N. Igarashi, H. Ohta, Y. Nagatani, T. Kosuge, K. Ito, *J. Phys.: Conf. Ser.* **2013**, *425*, 202008.

⁷ H. Takagi, N. Igarashi, T. Mori, S. Saijo, H. Ohta, Y. Nagatani, T. Kosuge, N. Shimizu, *AIP Conf. Proc.* **2016**, *1741*, 030018.

⁸ N. Shimizu, K. Yatabe, Y. Nagatani, S. Saijyo, T. Kosuge, N. Igarashi, *AIP Conf. Proc.* **2016**, *1741*, 050017.

AFM measurements



Figure S1. (a)-(f) AFM images and line profiles of **sumanene** (1.95 mM) prepared from the MCH solution on an HOPG surface.



Figure S2. AFM image of sumanene (95.3 μ M) prepared from the MCH solution on an HOPG surface.



Figure S3. AFM images of **sumanene** ((a) 2.08 mM and (b) 34.3 mM) prepared from the CH_2Cl_2 solution on an HOPG surface. (b) width, 286.7 ± 68.4 nm; height, 43.7 ± 30.7 nm.



Figure S4. (a)-(c) AFM images and line profiles of **sumanene** (34.3 mM) prepared from the CH₂Cl₂ solution on an HOPG surface.

DFT calculations



Figure S5. Optimized structures and dipole moments of **sumanene**: (a,b) monomer, (c,d) heptamer, (e) tetramer, and (f) pentamer. Dipole moment: monomer, 2.7 D; octamer, 21.7 D (3.1 D per monomer).

Supramolecular polymerization analysis^{9,10}

Chemical equilibrium and sequential equations based on the (anti-)cooperative model

The (anti-)cooperative model assumes different equilibrium constants for the nucleation and elongation stages, and the chemical equilibrium and sequential equations for the formation of a supramolecular polymer of a certain chemical species "A", are given as follows.

$$A + A \rightleftharpoons A_{2}, [A_{2}] = K_{n}[A]^{2} \quad (1)$$

$$A_{2} + A \rightleftharpoons A_{3}, [A_{3}] = K_{n}K_{e}[A]^{3} \quad (2)$$

$$A_{N-1} + A \rightleftharpoons A_{N}, [A_{N}] = \sigma K_{e}^{-1}(K_{e}[A])^{N}, \sigma = \frac{K_{n}}{K_{e}} \quad (3)$$

where A_X represents the X-mer of A, K_n is the equilibrium constant between the two monomers, K_e is the equilibrium constant between the polymer and monomer, and σ is the cooperative factor. When $\sigma < 1$, it is a cooperative model; when $\sigma > 1$, it is an anti-cooperative model.

Mass balance

From the sequential equation, the material balance can be expressed as follows.

$$K_e C_t = (1 - \sigma) K_e[A] + \frac{\sigma K_e[A]}{(1 - K_e[A])^2}$$
 (4)

where C_t represents the total concentration of "A". In other words,

$$C_t = [A] + \sum_{i=2}^{\infty} i[A_i]$$
 (5)

<u>Degree of polymerization (α_{agg})</u> Degree of polymerization (α_{agg}) can be expressed as follows.

$$\alpha_{agg} = 1 - \frac{K_e[A]}{K_e C_t} \quad (6)$$

On the other hand, α_{agg} is determined by the extinction coefficient as follows.

⁹ D. Zhao, J. S. Moore, Org. Biomol. Chem. 2003, 1, 3471-3491.

¹⁰ P. P. N. Syamala, F. Würthner, *Chem.-Eur. J.* **2020**, *26*, 8426-8434.

$$\alpha_{agg} = 1 - \frac{\overline{\varepsilon_{Ct} - \varepsilon_{agg}}}{\varepsilon_{mon} - \varepsilon_{agg}} \quad (7)$$

where $\bar{\varepsilon}_{Ct}$ represents the apparent extinction coefficient at a given concentration, ε_{agg} the extinction coefficient of the aggregates and ε_{mon} the extinction coefficients of the monomers. The latter two values were determined from the concentration-dependent UV spectra.

(anti-)cooperative model

We analyzed the supramolecular polymerization based on the (anti)cooperative model by plotting and fitting α_{agg} against C_t using Equations 4 and 7.

isodesmic model

The isodesmic model assumes that the equilibrium constants for the nucleation and elongation stages are identical. In other words,

$$K_n = K_e = K, \, \sigma = 1 \quad (8)$$

Substituting Equation 8 into Equations (4) and (6) yields Equations 9 and 10, respectively.

$$KC_{t} = \frac{K[A]}{(1 - K[A])^{2}} \quad (9)$$

$$\alpha_{agg} = 1 - \frac{K[A]}{KC_{t}} \quad (10)$$

Furthermore, substituting equation 9 into equation 10 yields equation 11.

$$\alpha_{agg} = 1 - \frac{2KC_t + 1 - \sqrt{4KC_t + 1}}{2K^2C_t^2} \quad (11)$$

Finally, the supramolecular polymerization was analyzed based on the isodesmic model by plotting and fitting α_{agg} against C_t using Equation 11. The number-average degree of polymerization (DP) in the isodesmic model was calculated based on equation 12.

$$DP = \frac{C_t}{[A] + \sum_{i=2}^{\infty} [A_i]} = \frac{\sqrt{4KC_t + 1} + 1}{2} \quad (12)$$

Concentration-dependent measurements



Figure S6. Plot of the degree of polymerization (α_{agg}) of sumanene (monitored at 356 nm in Figure 2a) as a function of C_t ; the red line represents the fitting result based on the isodesmic model and the blue line represents the anti-cooperative model, respectively.

Table S1. Equilibrium constants of **sumanene** supramolecular polymerization in MCH at 25 °C and 0.1 MPa based on the isodesmic and cooperative models.

fitting model	$K \text{ or } K_{e} / M^{-1}$	σ	correlation coefficient
isodesmic	18600 ± 2200	1	0.990
cooperative	15200 ± 200	0.07 ± 0.02	0.994

SAXS analysis

The equation for the cylinder model is expressed as equation 13¹¹

$$I(q) = \int_{0}^{\frac{\pi}{2}} \frac{\sin^{2}(q\frac{L}{2}\cos\theta) 4J_{1}^{2}(qR\sin\theta)}{q^{2}(\frac{L}{2})^{2}\cos^{2}\theta q^{2}R^{2}\sin^{2}\theta} \sin\theta d\theta \quad (13)$$
$$J_{1} = 3\frac{\sin qR - qR\cos qR}{(qR)^{3}} \quad (14)$$

where *L* represents the length of a cylinder, *R* the radius of a cylinder and J_1 the first-order Bessel function. The *L* and *R* values were obtained from the optimized structure using DFT calculations: *L* = 1.359 (tetramer), 1.727 (pentamer) nm and, *R* = 0.407 nm.

¹¹ A. Guinier, G. Fournet, Small-Angle Scattering of X-rays John Wiley & Sons, 1955.

Variable temperature measurements

To consider the effect of solvent density change with temperature, the temperature dependence data of the MCH density were obtained from the literature^{12,13} and fitted with a quadratic function. Thus, equation 15 was obtained.

$$\rho(T) = -7.38 \times 10^{-7} T^2 - 4.85 \times 10^{-4} T + 0.975$$
 (15)

where $\rho(T)$ represents the density of MCH at a given temperature. The absorbance was corrected by substituting $\rho(T)$ into equation 16.

$$A_{corrected} = A(T) \times \frac{\rho(T)}{\rho(298.15)} = A(T) \times \frac{\rho(T)}{0.76482} \quad (16)$$

Figure S7. Temperature-dependent absorption of sumanene (101 μ M) in MCH monitored at 356 nm (black: raw data and red: corrected data).

¹² J. Jonas, D. Hasha, S. G. Huang, J. Chem. Phys. 1979, 71, 3996-4000.

¹³ M. R. Usman, Z. Shahid, M. S. Akram, R. Aslam, Int. J. Thermophys. 2020, 41, 44.

High pressure measurements



Figure S8. (a) Concentration-dependent UV spectra of **sumanene** (0.02-9.48 mM, from black to blue) in 7:3 (v/v) MCH-CH₂Cl₂ at room temperature and 100 MPa, measured in a high-pressure cell. (b) Plot of the degree of polymerization (α_{agg}) of **sumanene** (monitored at 356 nm in (a)) as a function of C_t ; the red line represents the fitting result based on the isodesmic model and the blue line represents the anti-cooperative model, respectively.

Table S2. Equilibrium constants of **sumanene** supramolecular polymerization in 7:3 (v/v) MCH- CH_2Cl_2 at room temperature and 100 MPa based on the isodesmic and anti-cooperative models.

fitting model	$K \text{ or } K_{\text{e}} / \text{M}^{-1}$	σ	correlation coefficient
isodesmic	2300 ± 260	1	0.987
anti-cooperative	150 ± 90	25.4 ± 17.7	0.999



Figure S9. (a) Concentration-dependent UV spectra of **sumanene** (0.02-8.66 mM, from black to sky blue) in 7:3 (v/v) MCH-CH₂Cl₂ at room temperature and 150 MPa, measured in a high-pressure cell. (b) Plot of the degree of polymerization (α_{agg}) of **sumanene** (monitored at 356 nm in (a)) as a function of C_t ; the red line represents the fitting result based on the isodesmic model and the blue line represents the cooperative model, respectively.

fitting model	$K \text{ or } K_{\text{e}} / \text{M}^{-1}$	σ	correlation coefficient		
isodesmic	2060 ± 220	1	0.989		
cooperative	2000 ± 60	0.18 ± 0.06	0.991		

Table S3. Equilibrium constants of **sumanene** supramolecular polymerization in 7:3 (v/v) MCH- CH_2Cl_2 at room temperature and 150 MPa based on the isodesmic and cooperative models.



Figure S10. (a) Concentration-dependent UV spectra of **sumanene** (0.02-8.13 mM, from black to sky blue) in 7:3 (v/v) MCH-CH₂Cl₂ at room tempearture and 200 MPa, measured in a high-pressure cell. (b) Plot of the degree of polymerization (α_{agg}) of **sumanene** (monitored at 356 nm in (a)) as a function of C_t ; the red line represents the fitting result based on the isodesmic model and the blue line represents the cooperative model, respectively.

Table S4. Equilibrium constants of **sumanene** supramolecular polymerization in 7:3 (v/v) MCH- CH_2Cl_2 at room temperature at 200 MPa based on isodesmic and cooperative models.

fitting model	$K \text{ or } K_{\text{e}} / \text{M}^{-1}$	σ	correlation coefficient
isodesmic	1990 ± 170	1	0.993
cooperative	2370 ± 80	0.28 ± 0.07	0.997



Figure S11. (a) Concentration-dependent UV spectra of **sumanene** (0.02-7.17 mM, from black to sky blue) in 7:3 (v/v) MCH-CH₂Cl₂ at room temperature and 250 MPa, measured in a high-pressure cell. (b) Plot of the degree of polymerization (α_{agg}) of **sumanene** (monitored at 356 nm in (a)) as a function of C_t ; the red line represents the fitting result based on the isodesmic model and the blue line represents the anti-cooperative model, respectively.

Table S5. Equilibrium constants of **sumanene** supramolecular polymerization in 7:3 (v/v) MCH- CH_2Cl_2 at room temperature at 250 MPa based on isodesmic and anti-cooperative models.

fitting model	$K \text{ or } K_{\text{e}} / \text{M}^{-1}$	σ	correlation coefficient
isodesmic	1940 ± 190	1	0.990
anti-cooperative	490 ± 280	8.39 ± 6.82	0.993



Figure S12. Natural logarithm plot of the equilibrium constants upon the **sumanene** supramolecular polymerization in 7:3 (v/v) MCH-CH₂Cl₂ as a function of pressure.

Table S6. Equilibrium constants of **sumanene** supramolecular polymerization in 7:3 (v/v) MCH- CH_2Cl_2 solution based on the isodesmic model under high pressure.

Pressure / MPa	K / M^{-1}	correlation coefficient	DP^{a}
100	2300 ± 260	0.987	5.2
150	2060 ± 220	0.989	4.8
200	1990 ± 170	0.993	4.6
250	1940 ± 190	0.993	4.3

a: number-average degree of polymerization at the maximum measured concentration.

Solvent effect of sumanene supramolecular polymerization



Figure S13. (a) Concentration-dependent UV spectra of **sumanene** (0.07-34.3 mM, from black to sky blue) in CH_2Cl_2 at 25 °C and 0.1 MPa, measured in a 1 mm cell. (b) Plot of the degree of polymerization (a_{agg}) of **sumanene** (monitored at 360 nm in (a)) as a function of C_t ; the red line represents the fitting result based on the isodesmic model and the blue line represents the anti-cooperative model, respectively.

Table S7. Equilibrium constants of **sumanene** supramolecular polymerization in CH_2Cl_2 at room temperature at 0.1 MPa based on isodesmic and anti-cooperative models.

fitting model	$K \text{ or } K_{\text{e}} / \text{M}^{-1}$	σ	correlation coefficient
isodesmic	630 ± 70	1	0.991
anti-cooperative	320 ± 20	2.85 ± 0.59	0.997

Figure S14. (a) Concentration-dependent UV spectra of **sumanene** (0.06-30.7 mM, from black to sky blue) in 1:9 (v/v) MCH-CH₂Cl₂ solution at 25 °C and 0.1 MPa, measured in a 1 mm cell. (b) Plot of the degree of polymerization (a_{agg}) of **sumanene** (monitored at 356 nm in (a)) as a function of C_t ; the red line represents the fitting result based on the isodesmic model and the blue line represents the cooperative model, respectively.

fitting model	$K \text{ or } K_{e} / M^{-1}$	σ	correlation coefficient
isodesmic	670 ± 60	1	0.991
cooperative	490 ± 10	0.99 ± 0.11	0.999
	(a) 1.2 1.2 0.05 0.01 0.8 1.6 1.6 1.3.1 26.2 2.2	(b) 1.0 isodesmic model anti-cooperative model 0.5 0.5 0.5 0.1 0.1 0.1 0.1 0.1 0.1 0.1 0.1	100

Table S8. Equilibrium constants of sumanene supramolecular polymerization in 1:9 (v/v) MCH-CH₂Cl₂ solution at room temperature and 0.1 MPa based on isodesmic and cooperative models.

Figure S15. (a) Concentration-dependent UV spectra of sumanene (0.05-26.2 mM, from black to navy) in 1:3 (v/v) MCH-CH₂Cl₂ solution at 25 °C and 0.1 MPa, measured in a 1 mm cell. (b) Plot of the degree of polymerization (a_{agg}) of sumanene (monitored at 360 nm in (a)) as a function of C_t ; the red line represents the fitting result based on the isodesmic model and the blue line represents the anticooperative model, respectively.

Table S9. Equilibrium constants of sumanene supramolecular polymerization in 1:3 (v/v) MCH-
CH ₂ Cl ₂ solution at room temperature and 0.1 MPa based on the isodesmic and anti-cooperative
models.

fitting model	$K \text{ or } K_{\text{e}} / \text{M}^{-1}$	σ	correlation coefficient
isodesmic	710 ± 70	1	0.994
anti-cooperative	530 ± 10	1.30 ± 0.13	0.999

Figure S16. (a) Concentration-dependent UV spectra of **sumanene** (0.04-21.3 mM, from black to sky blue) in 2:3 (v/v) MCH-CH₂Cl₂ solution at 25 °C and 0.1 MPa, measured in a 1 mm cell. (b) Plot of the degree of polymerization (a_{agg}) of **sumanene** (monitored at 360 nm in (a)) as a function of C_t ; the red line represents the fitting result based on the isodesmic model and the blue line represents the cooperative model, respectively.

Table S10. Equilibrium constants of **sumanene** supramolecular polymerization in 2:3 (v/v) MCH- CH_2Cl_2 solution at room temperature and 0.1 MPa based on the isodesmic and cooperative models.

fitting model	$K \text{ or } K_{\text{e}} / \text{M}^{-1}$	σ	correlation coefficient
isodesmic	1470 ± 140	1	0.992
cooperative	1880 ± 10	0.64 ± 0.04	0.999

Figure S17. (a) Concentration-dependent UV spectra of **sumanene** (0.04-18.0 mM, from black to blue) in 1:1 (v/v) MCH-CH₂Cl₂ solution at 25 °C and 0.1 MPa, measured in a 2 mm cell. (b) Plot of the degree of polymerization (a_{agg}) of **sumanene** (monitored at 360 nm in (a)) as a function of C_t ; the red line represents the fitting result based on the isodesmic model and the blue line represents the cooperative model, respectively.

	1		1
fitting model	$K \text{ or } K_{\text{e}} / \text{M}^{-1}$	σ	correlation coefficient
isodesmic	2030 ± 300	1	0.979
cooperative	1050 ± 40	0.73 ± 0.18	0.997
	(a) 1.2 1.2 0.02 0.04 0.04 0.08 0.3 0.6 1.3 2.5 5.0 10.0 1.0 0.4	(b) 1.0 isodesmic mode anti-cooperative	el 9 model -

0___

0.1

 C_t / mM

10

100

360 380 Wavelength / nm

0 340

Table S11. Equilibrium constants of sumanene supramolecular polymerization in 1:1 (v/v) MCH-CH₂Cl₂ solution at room temperature and 0.1 MPa based on isodesmic and cooperative models.

Table S12. Equilibrium constants of sumanene supramolecular polymerization in 3:1 (v/v) MCH-CH₂Cl₂ solution at room temperature and 0.1 MPa based on the isodesmic and anti-cooperative models.

fitting model	$K \text{ or } K_{\text{e}} / \text{M}^{-1}$	σ	correlation coefficient
isodesmic	3800 ± 570	1	0.985
anti-cooperative	1420 ± 170	3.10 ± 1.04	0.994

Summary of structure optimization

Sumanene monomer

Symbol	Х	Y	Ζ
С	1.4053125	0.0159229	-0.7013613
С	-1.4053125	0.0159229	-0.7013613
С	0.7164459	1.2090749	-0.7013613
С	0.6888666	-1.2249978	-0.7013613
С	-0.6888666	-1.2249978	-0.7013613
С	-0.7164459	1.2090749	-0.7013613
С	1.2057868	2.3481307	-0.0691930
С	2.4939784	2.2613631	0.4521217
Н	2.9320307	3.0964874	0.9893603
С	2.6364343	-0.1298234	-0.0691930
С	3.2053871	1.0291671	0.4521217
Н	4.1476521	0.9909694	0.9893603
С	-1.2057868	2.3481307	-0.0691930
С	-2.4939784	2.2613631	0.4521217
Н	-2.9320307	3.0964874	0.9893603
С	-3.2053871	1.0291671	0.4521217
Н	-4.1476521	0.9909694	0.9893603
С	-2.6364343	-0.1298234	-0.0691930
С	-1.4306475	-2.2183074	-0.0691930
С	-0.7114087	-3.2905302	0.4521217
Н	-1.2156214	-4.0874568	0.9893603
С	0.7114087	-3.2905302	0.4521217
Н	1.2156214	-4.0874568	0.9893603
С	1.4306475	-2.2183074	-0.0691930
С	-2.8409447	-1.6402202	0.1867129
Н	-3.5886715	-2.0719205	-0.4874684
Н	-3.1726400	-1.8317246	1.2096614
С	0.0000000	3.2804404	0.1867129
Н	0.0000000	4.1438410	-0.4874684
Н	0.0000000	3.6634492	1.2096614
С	2.8409447	-1.6402202	0.1867129
Н	3.5886715	-2.0719205	-0.4874684
Н	3.1726400	-1.8317246	1.2096614

Total energy: -807.330745 Hartree

Sumanene tetramer

Symbol	Х	Y	Ζ
С	4.8421800	1.2957290	-0.5337610
С	4.8406430	-1.2855680	0.5616440
С	4.8418650	1.1280030	0.8310180
С	4.8419070	0.1545930	-1.3937490
С	4.8410800	-1.1111710	-0.8566070
С	4.8415630	-0.1874440	1.3891360
С	5.4650270	2.0269380	1.6909840
С	5.9702520	3.1860120	1.1096150
Н	6.4979890	3.9228990	1.7066790
С	5.4654650	2.3761290	-1.1503090
С	5.9703340	3.3599730	-0.3054530
Н	6.4982490	4.2194690	-0.7061360
С	5.4643880	-0.1940450	2.6332860
С	5.9684040	-1.4179410	3.0629970
Н	6.4959960	-1.5009950	4.0078480
С	5.9673650	-2.5566420	2.2050610
Н	6.4942060	-3.4425710	2.5449490
С	5.4623480	-2.4804290	0.9104690
С	5.4631780	-2.1859630	-1.4840110
С	5.9687950	-1.9466260	-2.7582540
Н	6.4959070	-2.7239530	-3.3022050
С	5.9699030	-0.6341240	-3.3151710
Н	6.4978040	-0.4859010	-4.2518280
С	5.4651630	0.4492980	-2.6022690
С	5.7189470	-3.2575860	-0.4003690
Н	5.0429110	-4.1126390	-0.5057280
Н	6.7418290	-3.6379280	-0.4468340
С	5.7231670	1.2798780	3.0188300
Н	5.0501700	1.6172420	3.8140230
Н	6.7472300	1.4280930	3.3687720
С	5.7233240	1.9729440	-2.6197650
Н	5.0490980	2.4926380	-3.3085670
Н	6.7469510	2.2022320	-2.9242550
С	1.1770420	0.1875620	-1.3876900
С	1.1772080	-0.1538370	1.3922480
С	1.1779300	1.2857470	-0.5602780

С	1.1760180	-1.1250380	-0.8305150
С	1.1762210	-1.2926660	0.5342420
С	1.1782970	1.1117690	0.8549740
С	1.7730750	2.4879710	-0.9136060
С	2.2311460	2.5732700	-2.2285740
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Н	2.7267060	1.5296000	-4.0466960
С	1.7733650	2.1927240	1.4888180
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С	2.2300750	0.6461350	3.3422230
Н	2.7275700	0.5050810	4.2967360
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Н	2.7249940	-3.9708040	-1.7112670
С	1.7698550	-2.0327760	-1.6953300
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Н	1.4113790	-2.5055000	3.3291320
Н	3.0912650	-2.1743400	2.8873830
С	2.0540130	3.2558160	0.4001280
Н	1.4150450	4.1381900	0.5088600
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С	2.0515430	-1.2792430	-3.0170440
Н	1.4143590	-1.6262950	-3.8367960
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С	-2.5015690	-1.2860210	0.5592510
С	-2.4991320	1.1248370	0.8311060
С	-2.5007440	0.1551730	-1.3926740
С	-2.5016740	-1.1109340	-0.8561820
С	-2.4999590	-0.1883590	1.3875700
С	-1.9017720	2.0305010	1.6957140
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Н	-0.9381350	3.9640920	1.7122860
С	-1.9027540	2.3823100	-1.1525680
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Н	-0.9386840	4.2617410	-0.6990300
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Н	-0.9411030	-1.5326120	4.0406340
С	-1.4446930	-2.5753130	2.2244740
Н	-0.9443030	-3.4716740	2.5772230
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С	-1.9068000	-2.1917520	-1.4905760
С	-1.4466490	-1.9563450	-2.7861710
Н	-0.9464860	-2.7399000	-3.3466790
С	-1.4456760	-0.6446830	-3.3420000
Н	-0.9453260	-0.5032440	-4.2949010
С	-1.9053080	0.4505590	-2.6103780
С	-1.6271410	-3.2558960	-0.4031050
Н	-2.2672500	-4.1375240	-0.5126710
Н	-0.5868430	-3.5904060	-0.4443640
С	-1.6192220	1.2756810	3.0160680
Н	-2.2539830	1.6241500	3.8372620
Н	-0.5772310	1.4045590	3.3218220
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Н	-2.2590830	2.5087530	-3.3279870
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Н	-4.6047730	3.4595010	-2.5694440
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Н	-4.6068320	1.5204570	-4.0313430
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Н	-4.6045320	2.7344110	3.3328700
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Н	-4.6042070	0.4982450	4.2807400
С	-5.5922540	-0.4490200	2.6091630
С	-5.5945410	-2.3812110	1.1526860
С	-5.1206000	-3.3802120	0.3007110
Н	-4.6097770	-4.2502980	0.7010600
С	-5.1212880	-3.2067310	-1.1099150
Н	-4.6102110	-3.9536640	-1.7092430
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Н	-5.9336060	-2.5096190	3.3334980
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Н	-5.9316980	4.1441540	0.5108300
Н	-4.2602410	3.5741400	0.4400190
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Н	-5.9391940	-1.6268410	-3.8398310
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Symbol	Х	Y	Z
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С	6.6811730	1.3302880	-0.4424100
С	6.6792670	-1.0482830	-0.9326590
С	6.6787490	-1.3406940	0.4109720
С	6.6805960	1.0263240	0.9538660
С	7.3050870	2.5510160	-0.6797890
С	7.8101710	2.7453420	-1.9619550
Н	8.3385590	3.6579040	-2.2188390
С	7.3031330	0.4331400	-2.6058210
С	7.8091820	1.6905850	-2.9211660
Н	8.3368370	1.8593600	-3.8544710
С	7.3042400	2.0377940	1.6776620
С	7.8086100	1.6815120	2.9248220

Н	8.3368310	2.4046370	3.5380320
С	7.8073180	0.3234170	3.3587270
Н	8.3345450	0.0889020	4.2779670
С	7.3016230	-0.6891130	2.5489140
С	7.3005330	-2.4739220	0.9256370
С	7.8050320	-3.3763530	-0.0059620
Н	8.3318270	-4.2695970	0.3142890
С	7.8055840	-3.0731290	-1.3990460
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С	7.3015550	-1.8650040	-1.8715300
С	7.5582790	-2.2081010	2.4259120
Н	6.8840600	-2.7884840	3.0645190
Н	8.5818480	-2.4650150	2.7077520
С	7.5635040	3.2036280	0.6969160
Н	6.8905060	4.0476780	0.8806270
Н	8.5875770	3.5748060	0.7780710
С	7.5600680	-0.9994740	-3.1252660
Н	6.8865960	-1.2620490	-3.9478540
Н	8.5839370	-1.1159510	-3.4875540
С	3.0138260	-1.0263410	-0.9525680
С	3.0140220	1.0458320	0.9322920
С	3.0138050	0.2835770	-1.3708420
С	3.0138990	-1.3298900	0.4407580
С	3.0140100	-0.3126350	1.3659530
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С	3.6087610	0.6899130	-2.5564240
С	4.0673320	-0.3337110	-3.3858410
Н	4.5646850	-0.1055050	-4.3233580
С	3.6087510	-2.0445110	-1.6832810
С	4.0672300	-1.6909910	-2.9524630
Н	4.5645270	-2.4202830	-3.5842870
С	3.6087230	2.4803390	-0.9276760
С	4.0675800	3.4026380	0.0129510
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С	3.8901160	-3.2031370	-0.6970230
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С	-0.6624000	-1.3359530	0.4130870
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С	-0.0655360	2.5614510	-0.6812630
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Н	0.8975440	3.6905460	-2.2514880
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Н	0.8984510	0.1083480	4.3200420
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С	0.3938280	-3.0967200	-1.4025790
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Н	-2.7753540	-0.1050440	-4.3172840
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С	-3.2778400	-1.6907660	-2.9493160
Н	-2.7738800	-2.4186770	-3.5773580
С	-3.7435550	2.4797750	-0.9279670
С	-3.2826560	3.4015640	0.0120630
Н	-2.7802260	4.3102080	-0.3045020
С	-3.2829430	3.0990250	1.4039900
Н	-2.7805740	3.7942080	2.0692910
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С	-3.7432310	-0.4348240	2.6133400
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Н	-2.7781620	-1.8888140	3.8869380
С	-3.2796500	-2.7633300	1.9837990
Н	-2.7758690	-3.6862450	2.2534990
С	-3.7405180	-2.5584690	0.6832560
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Н	-4.0961960	1.2699200	3.9719670
Н	-2.4186440	1.0997700	3.4371760
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Н	-4.0901250	-4.0735510	-0.8845720
Н	-2.4134120	-3.5238260	-0.7644600
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Н	-6.4420030	3.6765800	-2.2430120
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С	-6.9542570	1.7051670	-2.9359980
Н	-6.4400330	1.8793180	-3.8759620
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С	-6.9531480	-3.3958480	-0.0119120
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Н	-6.4373080	-3.7810040	-2.0649180
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С	-7.1461930	3.1987960	0.6937580
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Н	-6.1015100	3.5150600	0.7629710
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Symbol	Х	Y	Ζ
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С	10.3539550	0.2816240	-1.3724980
С	10.3586210	-1.3292580	0.4447830
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С	10.3537260	1.3403750	-0.4128740
С	10.9744710	0.6855910	-2.5504620
С	11.4796280	-0.3279790	-3.3592890
Н	12.0055330	-0.0947220	-4.2795910
С	10.9794770	-2.0399200	-1.6748470
С	11.4821240	-1.6853810	-2.9231980
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С	10.9788270	1.8687410	1.8680120
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Н	12.0185340	-3.6535590	2.2231660
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С	7.2871810	-1.8713970	-1.8739190
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Н	8.2460480	-3.7981020	-2.0661390
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С	7.7473480	-3.4022350	-0.0082620
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С	7.2846030	0.4334740	-2.6129290
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Н	8.2390220	1.8913850	-3.8905010
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Н	8.2368410	3.6909730	-2.2576490
С	7.2820760	2.5593180	-0.6839750
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Н	8.2395840	2.4257690	3.5813940
С	7.7444040	0.3384220	3.3857820
Н	8.2427090	0.1119530	4.3232150
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С	7.5629350	3.2045450	0.6939780
Н	6.9255240	4.0743680	0.8831560
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С	7.5681020	-1.0001220	-3.1212840
Н	6.9327120	-1.2731020	-3.9700010
Н	8.6099590	-1.1006700	-3.4381300
С	7.5690160	-2.2023910	2.4275230
Н	6.9326390	-2.8012350	3.0871170
Н	8.6105260	-2.4260780	2.6748260
С	3.0137530	-1.0290470	-0.9520070
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С	3.0130180	0.2804330	-1.3719940
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С	3.0138560	-0.3124100	1.3658240
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С	3.6089580	0.6853880	-2.5576020
С	4.0708820	-0.3385150	-3.3846080

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С	4.0716010	-1.6950660	-2.9495510
Н	4.5743470	-2.4236020	-3.5778320
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С	3.8926710	0.9997550	3.1200780
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С	3.8905250	2.2009480	-2.4270910
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Н	4.9323630	2.4245470	-2.6728290
С	3.8929470	-3.2043420	-0.6937850
Н	3.2573410	-4.0757550	-0.8822940
Н	4.9347000	-3.5291220	-0.7645080
С	-0.6621430	-1.3396950	0.4133230
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