

Supporting Information

A Case of Alkylidyne-Imine Metathesis

Rinku Yadav, Ion Ghiviriga, Khalil A. Abboud, Adam S. Veige*

Department of Chemistry, University of Florida, Center for Catalysis, P.O. Box 117200,
Gainesville, Florida 32611, United States.

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EXPERIMENTAL SECTION

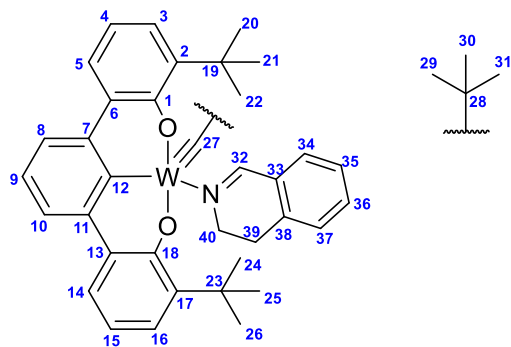
1. General Considerations

Unless otherwise specified, all manipulations were performed under an inert atmosphere using standard Schlenk or glove-box techniques. Glassware was oven-dried before use. Pentane, hexane, toluene, diethyl ether (Et₂O), tetrahydrofuran (THF), benzene (C₆H₆) were dried using a GlassContour drying column and stored over 3 Å molecular sieves. Benzene-*d*₆ (Cambridge Isotopes) was dried over sodium benzophenone distilled, and stored over 3 Å molecular sieves. The tungsten-alkylidyne [tBuOCO]W≡CtBu(THF)₂ (**1**)¹ was prepared according to published procedures. *N*-methyl-1-phenylmethanimine and *N*-benzyl-1-phenylmethanimine were bought from sigma –Aldrich, *N*,1-diphenylmethanimine and *N*-(trimethylsilyl)-1-phenylmethanimine were bought from Oakwood chemicals, and other imines were synthesized according to reported literature.²⁻⁴ Liquid imines were dried over CaH₂ and stored under a nitrogen atmosphere, solid imines were dried under vacuum overnight before use. ¹H NMR, ¹³C{¹H} NMR and 2D NMR spectra were obtained on Varian INOVA (500 MHz) and Bruker (400 MHz) spectrometers. The chemical shifts are reported in δ (ppm), referenced to the lock signal on solvent for ¹H and ¹³C{¹H} spectra. The assignments are primarily based on the cross-peaks observed in the ¹H-¹³C gradient Heteronuclear Multiple Bond Correlation (gHMBC), gradient Heteronuclear Single Quantum Correlation (gHSQC), ¹H-¹H Homonuclear Correlation Spectroscopy (COSY) and Nuclear Overhauser Effect Spectroscopy (NOESY) spectra. The spectra were recorded at 25 °C unless noted otherwise. Elemental analyses were performed at the CENTC Elemental Analysis Facility, Department of Chemistry at University of Rochester.

2. Synthetic procedures

2.1. Synthesis of complex 2

Complex **1** (101 mg, 13.1 x 10⁻² mmol, 1.00 eq) was dissolved in 5 mL of benzene in a vial equipped with a magnetic stir bar and 3,4-dihydroisoquinoline (17.0 mg, 13.1 x 10⁻² mmol, 1.00 eq) was added to the solution at room temperature and stirred for 1 h. The color of the solution immediately changed from red to dark brown. Volatiles were removed under vacuum yielding **2** as a dark red solid, it was further washed with 5 mL of cold pentane to yield **2** (93.8 mg, 95% yield). Maroon colored crystals of **2** can be obtained from a concentrated solution of **2** in either pentane or toluene at -30 °C within a 2 d.

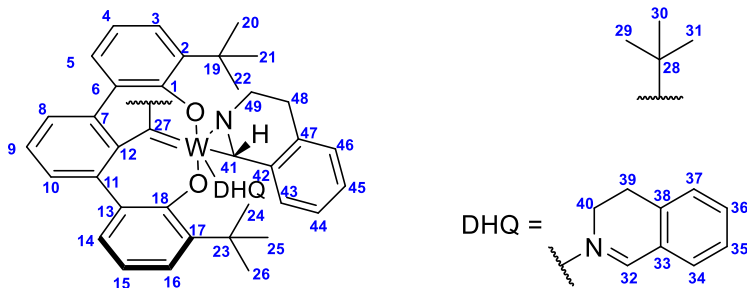


^1H NMR (C_6D_6 , 400 MHz) δ (ppm): 8.49 (s, 1H, H_{32}), 8.10 (d, 2H, $J = 7.8$ Hz, Ar- $H_{8,10}$), 7.93 (dd, 2H, $J_1 = 7.9$ Hz, $J_2 = 1.6$ Hz, Ar- $H_{5,14}$), 7.44 (m, 3H, Ar- $H_{3,9,16}$), 7.08 (t, 2H, $J = 1.6$ Hz, Ar- $H_{4,15}$), 6.98 (m, 1H, Ar- H_{37}), 6.84 (m, 2H, Ar- $H_{36,35}$), 6.65 (d, 1H, $J = 7.4$ Hz, Ar- H_{34}), 3.69 (t, 2H, $J = 7.7$ Hz, Ar- H_{40}), 2.44 (t, 2H, $J = 7.7$ Hz, Ar- H_{39}), 1.57 (s, 18H, Ar- t Bu, $H_{20-22,24-26}$), 0.70 (s, 9H, $\text{W}\equiv\text{C}^t\text{Bu}$, H_{29-31}).

$^{13}\text{C}\{^1\text{H}\}$ NMR (C_6D_6 , 400 MHz) δ (ppm): 320.8 (C_{27}), 193.1 (C_{12}), 171.8 (C_{32}), 163.2 ($\text{C}_{1,18}$), 140.2 ($\text{C}_{7,11}$), 136.5 (C_{38}), 135.2 ($\text{C}_{6,13}$), 135.0 (C_{37}), 134.7 ($\text{C}_{2,17}$), 130.7 (C_{36}), 127.9 ($\text{C}_{5,14}$), 127.8 (C_{34}), 127.3 (C_{35}), 127.0 ($\text{C}_{8,10}$), 126.6 (C_{33}), 124.7 ($\text{C}_{3,16,9}$), 120.8 ($\text{C}_{4,15}$), 50.0 (C_{28}), 49.3 (C_{40}), 35.2 ($\text{C}_{19,23}$), 32.7 (C_{29-31}), 30.4 ($\text{C}_{20-22,24-26}$), 25.1 (C_{39}).

2.2. Synthesis of complex 3

Complex **1** (100 mg, 13.0×10^{-2} mmol, 1.00 eq) was dissolved in 5 mL of benzene in a vial equipped with a magnetic stir bar and 3,4-dihydroisoquinoline (85.2 mg, 65.2×10^{-2} mmol, 5.00 eq) was added to the solution at room temperature and stirred for 1 h. NMR spectral data of the reaction mixture reveals that complexes **2** and **3** coexist in the solution in ratio 1:2.5, respectively under these conditions. Volatiles were removed under vacuum. A brown precipitate (120.4 mg) was obtained after 3-4 triturations using pentane. Separation of **3** via crystallization was not possible. In situ solution NMR characterization of **3** is provided below.



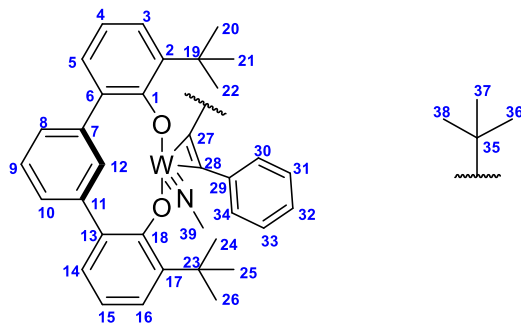
^1H NMR (C_6D_6 , 500 MHz) δ (ppm): 8.50 (t, 1H, $J = 1.6$ Hz, H_{32}), 7.40 (dd, 1H, $J_1 = 6.3$ Hz, $J_2 = 1.2$ Hz, Ar- H_{10}), 7.38 (dd, 1H, $J_1 = 6.5$ Hz, $J_2 = 1.5$ Hz, Ar- H_{16}), 7.33 (m, 1H, $J_1 = 6.2$ Hz, $J_2 = 1.3$ Hz, Ar- H_{14}), 7.30 (dd, 1H, $J_1 = 6.4$ Hz, $J_2 = 1.5$ Hz, Ar- H_3), 7.28 (dd, 1H, $J_1 = 6.3$ Hz, $J_2 = 1.4$ Hz, Ar- H_8), 7.26 (dd,

1H, $J_1 = 6.2$ Hz, $J_2 = 1.5$ Hz, Ar- H_5), 7.21 (m, 1H, Ar- H_{43}), 7.16 (m, 1H, Ar- H_{44}), 7.09 (t, 1H, $J = 6.3$ Hz, Ar- H_9), 7.06 (m, 1H, Ar- H_{46}), 7.00 (m, 1H, H_{45}), 6.91 (m, 1H, Ar- H_{36}), 6.86 (t, 1H, $J = 6.3$ Hz, Ar- H_{15}), 6.81 (t, 1H $J = 6.1$ Hz, Ar- H_4), 6.77 (m, 1H, Ar- H_{35}), 6.71 (m, 1H, Ar- H_{34}), 6.54 (d, 1H, $J = 6.1$ Hz, Ar- H_{37}), 5.15 (ddd, 1H, $J_1 = 10.4$ Hz, $J_2 = 8.1$ Hz, $J_3 = 4.9$ Hz, Ar- $H_{49'}$), 5.06 (ddd, 1H, $J_1 = 10.4$ Hz, $J_2 = 6.1$ Hz, $J_3 = 2.6$ Hz, Ar- $H_{49''}$), 3.91 (s, 1H, H_{41}), 3.66 (td, 2H, $J_1 = 6.4$ Hz, $J_2 = 1.6$ Hz, Ar- H_{40}), 3.13 (m, 1H, Ar- H_{48a}), 2.90 (ddd, 1H, $J_1 = 13.4$ Hz, $J_2 = 5.1$ Hz, $J_3 = 2.6$ Hz, Ar- H_{48b}), 2.10 (m, 2H, H_{39}), 1.50 (s, 9H, Ar- t Bu, H_{24-26}), 1.04 (s, 9H, Ar- t Bu, H_{20-22}), 0.97 (s, 9H, H_{29-31}).

$^{13}\text{C}\{^1\text{H}\}$ NMR (C_6D_6 , 500 MHz) δ (ppm): 256.3 (C_{27}), 169.7 (s, C_{18}), 169.6 (s, C_1), 167.5 (s, C_{32}), 161.4 (s, C_7), 159.5 (s, C_{11}), 143.7 (s, C_{42}), 139.3 (s, C_2), 139.1 (s, C_{17}), 136.9 (s, C_{38}), 135.9 (s, C_{47}), 135.1 (s, C_9), 131.5 (s, C_{13}), 131.5 (s, C_6), 131.0 (s, C_8), 130.0 (s, C_{10}), 129.6 (s, C_{34}), 129.5 (s, C_{46}), 129.0 (s, C_5), 128.7 (s, C_{43}), 128.4 (s, C_{33}), 128.3 (s, C_{14}), 127.4 (s, C_{35}), 127.2 (s, C_{37}), 127.1 (s, C_{36}), 126.7 (s, C_{16}), 126.8 (s, C_3), 125.9 (s, C_{44}), 124.3 (s, C_{45}), 118.6 (s, C_4), 118.4 (s, C_{15}), 118.1 (s, C_{12}), 57.1 (s, C_{49}), 48.8 (s, C_{40}), 45.0 (s, C_{28}), 44.7 (s, C_{41}), 35.1 (s, C_{29-31}), 35.0 (s, C_{23}), 34.7 (s, C_{19}), 31.0 (s, C_{48}), 30.6 (s, C_{20-22}), 30.3 (s, C_{24-26}), 25.9 (s, C_{39}).

2.3. Synthesis of complex 4-R (R = Me, TMS, Ph, Bn)

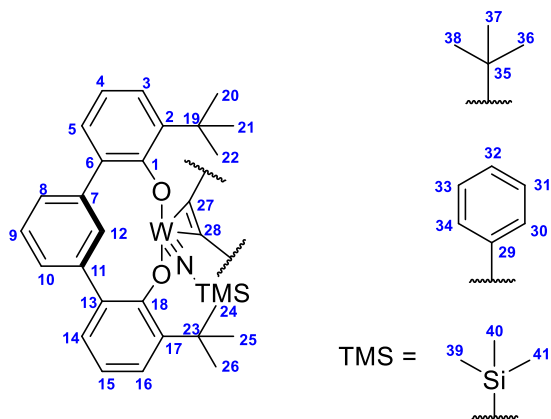
Complex **1** (100 mg, 13.0×10^{-2} mmol, 1.00 eq) was dissolved in 5 mL of benzene in a vial equipped with a magnetic stir bar. Imine (13.0×10^{-2} mmol, 1.00 eq) was added to the above solution at room temperature resulting in a dark red solution. The resultant solution was then transferred to a Schlenk tube equipped with a magnetic stir bar and heated at 80 °C in an oil bath for 2 d for *N*-methyl-1-phenylmethanimine (color changed to transparent yellow) and *N*-benzyl-1-phenylmethanimine (color changed to transparent amber), and 4d for *N*,1-diphenylmethanimine (color changed to transparent yellow) and *N*-(trimethylsilyl)-1-phenylmethanimine (color changed to transparent yellow). Volatiles were removed under vacuum precipitating **4-R** in 95-98% yields. **4-Me** and **4-Bn** were crystallized from concentrated pentane solutions as yellow crystals at -30 °C. **4-Ph** and **4-TMS** were crystallized from concentrated Et₂O:THF (9:1) solutions as yellow crystals at -30 °C.



4-Me, ^1H NMR (C_6D_6 , 500 MHz) δ (ppm): 7.99 (t, 1H, $J = 1.9$ Hz, Ar- H_{12}), 7.34 (d, 1H, $J = 7.4$ Hz, Ar- H_{16}), 7.28 (d, 1H, $J = 7.7$ Hz, Ar- H_3), 7.18 (d, 1H, $J = 1.7$ Hz, Ar- H_{14}), 7.16 (d, 2H, $J = 1.6$ Hz, Ar- $H_{30,34}$), 7.13 (t, 1H, $J = 9.6$ Hz, Ar- H_9), 7.02 (m, 2H, Ar- $H_{31,33}$), 6.99 (m, 1H, Ar- H_{10}), 6.93 (m, 2H, Ar- $H_{5,15}$), 6.85 (tt, 1H, $J_1 = 9.2$ Hz, $J_2 = 1.7$ Hz, Ar- H_{37}), 6.75 (t, 1H, $J = 9.7$ Hz, Ar- H_4), 6.64 (d, 1H, $J = 9.5$ Hz, Ar- H_8), 4.00 (s, 3H, H_{39}), 1.69 (s, 9H, Ar- t Bu, H_{24-26}), 1.68 (s, 9H, Ar- t Bu, H_{20-22}), 1.10 (s, 9H, H_{36-38}).

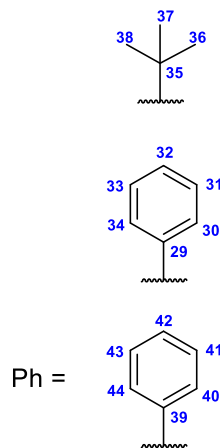
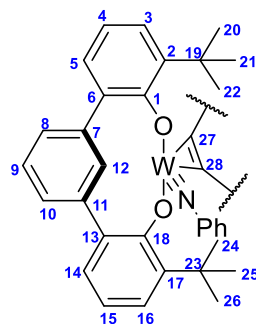
$^{13}\text{C}\{^1\text{H}\}$ NMR (C_6D_6 , 500 MHz) δ (ppm): 198.3 (C_{27}), 180.3 (C_{28}), 162.0 (C_{18}), 161.4 (C_1), 142.6 (C_{11}), 141.7 (C_7), 141.1 (C_{17}), 138.6 (C_{29}), 133.6 (C_{13}), 132.5 (C_6), 130.9 ($\text{C}_{10,30,34}$), 129.7 (C_8), 127.6 (C_{12}), 127.3 ($\text{C}_{31,33}$), 127.1 (C_{32}), 127.0 (C_{14}), 126.9 (C_5), 126.9 (C_{16}), 126.9 (C_3), 122.9 (C_{15}), 122.4 (C_4), 51.6 (C_{39}), 42.3 (s, C_{35}), 35.7 (s, $\text{C}_{19,23}$), 31.9 (s, C_{36-38}), 30.2 (s, $\text{C}_{20-22,24-26}$).

Elemental Analysis calcd. (%) for $\text{C}_{39}\text{H}_{45}\text{NO}_2\text{W}$ (743.63 g/mol): C, 62.990; H, 6.100; N, 1.884; Found: C, 62.605; H, 5.904; N, 1.795.



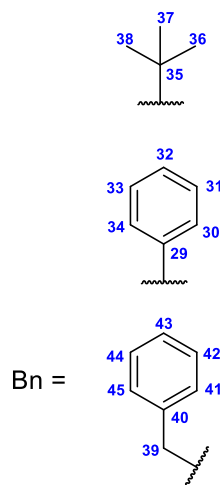
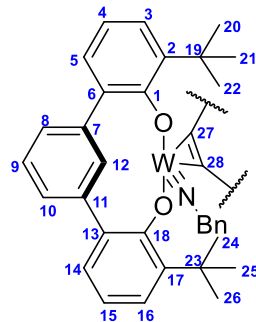
4-TMS, ^1H NMR (C_6D_6 , 400 MHz) δ (ppm): 7.90 (t, 1H, $J = 7.6$ Hz, Ar- H_{12}), 7.33 (dd, 1H, $J_1 = 1.8$ Hz, $J_2 = 7.8$ Hz, Ar- H_{16}), 7.26 (dd, 1H, $J_1 = 1.8$ Hz, $J_2 = 7.8$ Hz, Ar- H_3), 7.21 (m, 2H, Ar- $H_{30,34}$), 7.19 (dd, 1H, $J_1 = 1.8$ Hz, $J_2 = 7.8$ Hz, Ar- H_{14}), 7.13 (t, 1H, $J = 7.6$ Hz, Ar- H_9), 7.02 (m, 2H, Ar- $H_{31,33}$), 7.00 (m, 1H, Ar- H_{10}), 6.93 (dd, 1H, $J_1 = 1.8$ Hz, $J_2 = 7.8$ Hz, Ar- H_5), 6.91 (t, 1H, $J = 7.8$ Hz, Ar- H_{15}), 6.83 (tt, 1H, $J_1 = 1.3$ Hz, $J_2 = 7.4$ Hz, Ar- H_{32}), 6.74 (t, 1H, $J = 7.8$ Hz, Ar- H_4), 6.70 (ddd, 1H, $J_1 = 1.1$ Hz, $J_2 = 1.8$ Hz, $J_3 = 7.6$ Hz, Ar- H_8), 1.71 (s, 9H, Ar- t Bu, H_{24-26}), 1.69 (s, 9H, Ar- t Bu, H_{20-22}), 1.12 (s, 9H, H_{36-38}), 0.25 (s, 9H, TMS, H_{39-41}).

$^{13}\text{C}\{^1\text{H}\}$ NMR (C_6D_6 , 400 MHz) δ (ppm): 196.7 (C_{27}), 180.0 (C_{28}), 161.7 (C_{18}), 161.0 (C_1), 141.5 (C_{11}), 140.9 (C_{17}), 140.8 (C_2), 140.7 (C_7), 138.4 (C_{29}), 133.8 (C_{13}), 132.8 (C_6), 131.1 (C_9), 130.9 ($\text{C}_{30,34}$), 129.3 (C_{10}), 129.1 (C_{12}), 128.1 (C_8), 127.4 ($\text{C}_{31,34}$), 127.3 (C_{32}), 127.1 (C_{14}), 126.9 (C_5), 126.8 ($\text{C}_{3,16}$), 123.2 (C_{15}), 122.8 (C_4), 41.6 (C_{35}), 35.8 (C_{23}), 35.7 (C_{19}), 31.7 (C_{36-38}), 30.5 (C_{24-26}), 30.4 (C_{20-22}), 2.71 (C_{39-41}).



4-Ph, $^1\text{H NMR}$ (C_6D_6 , 400 MHz) δ (ppm): 8.00 (t, 1H, $J = 1.8$ Hz, Ar- H_{12}), 7.34 (dd, 1H, $J_1 = 1.7$ Hz, $J_2 = 7.8$ Hz, Ar- H_{16}), 7.32 (dd, 2H, $J_1 = 1.2$ Hz, $J_2 = 8.2$ Hz, Ar- $H_{30,34}$), 7.27 (dd, 1H, $J_1 = 1.7$ Hz, $J_2 = 7.8$ Hz, Ar- H_3), 7.18 (dd, 1H, $J_1 = 1.7$ Hz, $J_2 = 7.4$ Hz, Ar- H_{14}), 7.14 (dd, 2H, $J_1 = 1.3$ Hz, $J_2 = 8.6$ Hz, Ar- $H_{40,44}$), 7.12 (t, 1H, $J = 7.7$ Hz, Ar- H_9), 7.05 (m, 4H, Ar- $H_{31,33,41,43}$), 6.99 (ddd, 1H, $J_1 = 1.2$ Hz, $J_2 = 1.8$ Hz, $J_3 = 7.7$ Hz, Ar- H_{10}), 6.92 (t, 1H, $J = 7.8$ Hz, Ar- H_{15}), 6.91 (dd, 1H, $J_1 = 1.7$ Hz, $J_2 = 7.8$ Hz, Ar- H_5), 6.87 (tt, 1H, $J_1 = 1.3$ Hz, $J_2 = 7.4$ Hz, Ar- H_{42}), 6.75 (t, 1H, $J = 7.8$ Hz, Ar- H_4), 6.73 (tt, 1H, $J_1 = 1.2$ Hz, $J_2 = 7.2$ Hz, Ar- H_{32}), 6.63 (ddd, 1H, $J_1 = 1.2$ Hz, $J_2 = 1.8$ Hz, $J_3 = 7.7$ Hz, Ar- H_8), 1.73 (s, 9H, Ar- $t\text{Bu}$, H_{24-26}), 1.64 (s, 9H, Ar- $t\text{Bu}$, H_{20-22}), 1.07 (s, 9H, H_{36-38}).

$^{13}\text{C}\{^1\text{H}\}\text{NMR}$ (C_6D_6 , 400 MHz) δ (ppm): 203.3 (C_{27}), 183.9 (C_{28}), 161.8 (C_{18}), 161.0 (C_1), 157.2 (C_{39}), 142.7 (C_{11}), 141.7 (C_7), 141.3 (C_{17}), 141.2 (C_2), 137.9 (C_{29}), 133.7 (C_{13}), 132.5 (C_6), 131.7 (C_9), 131.3 ($\text{C}_{41,43}$), 129.8 (C_{10}), 128.5 (C_8), 127.7 ($\text{C}_{31,33}$), 127.6 (C_{14}), 127.5 (C_{16}), 127.3 (C_{12}), 127.1 (C_{42}), 126.8 (C_3), 126.7 (C_5), 125.4 ($\text{C}_{40,44}$), 124.8 (C_{32}), 123.3 (C_{15}), 122.8 (C_4), 42.8 (C_{35}), 35.8 (C_{23}), 35.7 (C_{19}), 31.9 (C_{36-38}), 30.3 (C_{24-26}), 30.2 (C_{20-22}).



4-Bn, ^1H NMR (C_6D_6 , 400 MHz) δ (ppm): 8.01 (t, 1H, $J = 1.8$ Hz, Ar- H_{12}), 7.32 (dd, 1H, $J_1 = 1.7$ Hz, $J_2 = 7.9$ Hz, Ar- H_{16}), 7.28 (m, 3H, Ar- $H_{3,41,45}$), 7.17 (m, 4H, Ar- $H_{9,14,30,34}$), 7.12 (m, 2H, Ar- $H_{42,44}$), 7.02 (m, 3H, Ar- $H_{31,33,43}$), 6.98 (ddd, 1H, $J_1 = 1.2$ Hz, $J_2 = 1.8$ Hz, $J_3 = 8.2$ Hz, Ar- H_{10}), 6.94 (dd, 1H, $J_1 = 1.7$ Hz, $J_2 = 7.9$ Hz, Ar- H_5), 6.91 (t, 1H, $J = 7.9$ Hz, Ar- H_{15}), 6.86 (tt, 1H, $J_1 = 1.2$ Hz, $J_2 = 7.5$ Hz, Ar- H_{32}), 6.76 (t, 1H, $J = 7.9$ Hz, Ar- H_4), 6.64 (ddd, 1H, $J_1 = 1.2$ Hz, $J_2 = 1.8$ Hz, $J_3 = 8.2$ Hz, Ar- H_8), 5.39 (s, 2H, H_{39}), 1.62 (s, 18H, Ar- $t\text{Bu}$, $H_{20-22,24-26}$), 1.04 (s, 9H, H_{36-38}).

$^{13}\text{C}\{^1\text{H}\}$ NMR (C_6D_6 , 400 MHz) δ (ppm): 198.9 (C_{27}), 180.9 (C_{28}), 161.9 (C_{18}), 161.3 (C_1), 142.4 (C_{11}), 141.5 (C_7), 141.2 (C_{17}), 141.1 (C_2), 140.1 (C_{40}), 138.4 (C_{29}), 133.7 (C_{13}), 132.7 (C_6), 131.5 (C_9), 131.1 ($\text{C}_{30,34,43}$), 129.6 (C_{10}), 128.6 ($\text{C}_{12,42,44}$), 128.2 ($\text{C}_{41,45}$), 128.0 (C_8), 127.3 ($\text{C}_{14,16}$), 127.2 (C_3), 126.9 (C_5), 126.8 ($\text{C}_{31,32,33}$), 122.9 (C_{15}), 122.5 (C_4), 68.3 (C_{39}), 42.2 (C_{35}), 35.7 ($\text{C}_{19,23}$), 31.8 (C_{36-38}), 30.1 ($\text{C}_{20-22,24-26}$).

3. Experimental NMR data

3.1. NMR spectra of 2

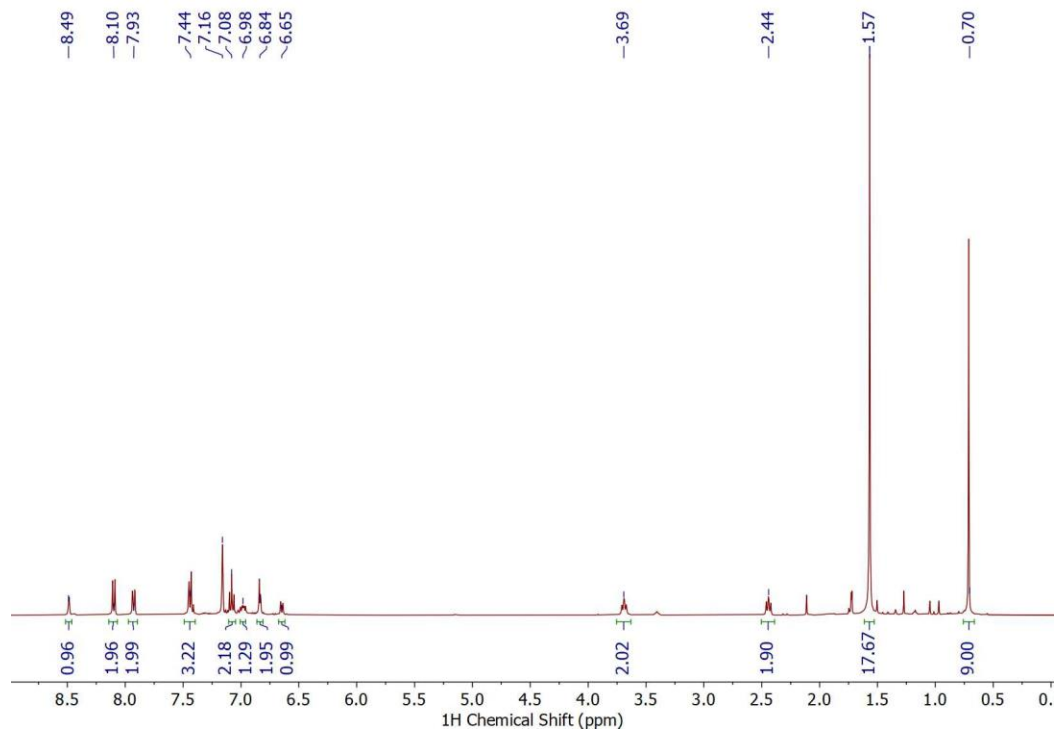


Figure S1. ^1H NMR spectrum of **2** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

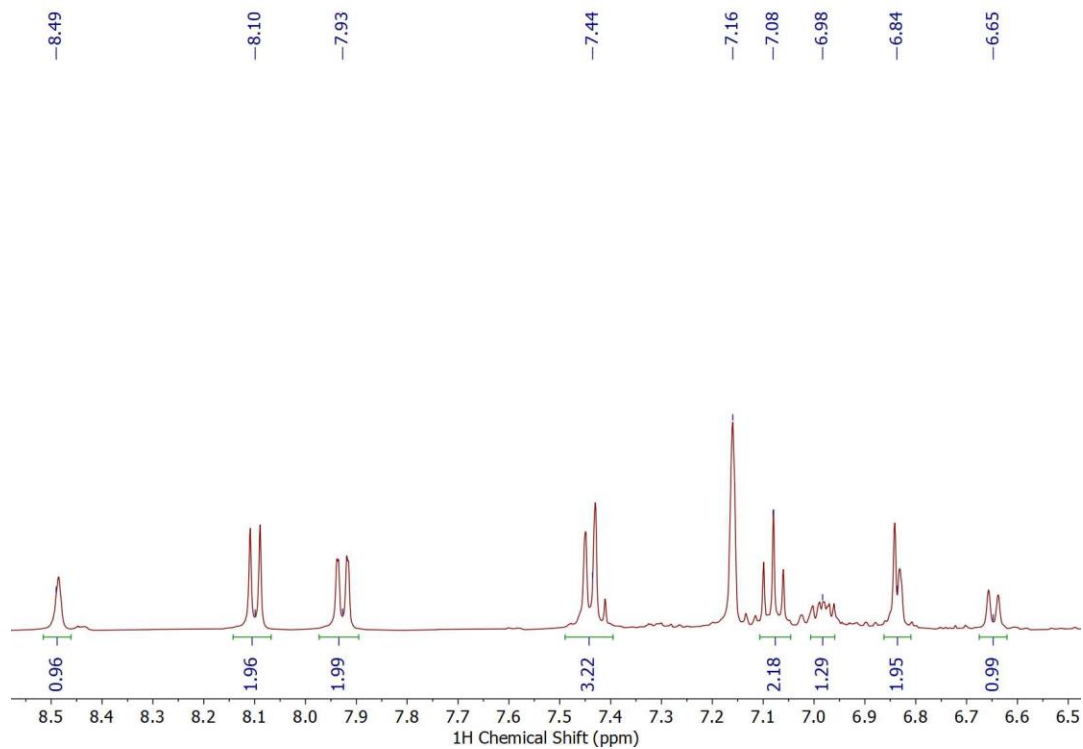


Figure S2. ^1H NMR spectrum-expansion of **2** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

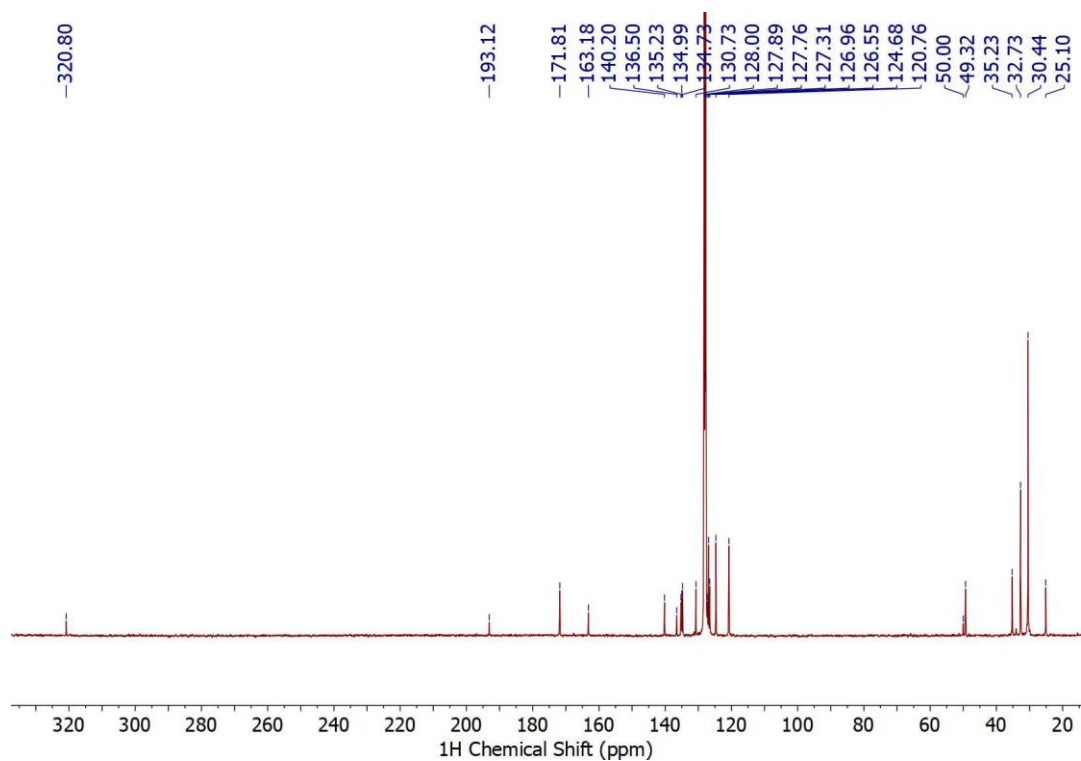


Figure S3. ^{13}C NMR spectrum of **2** (C_6D_6 , 400 MHz, 25 °C).

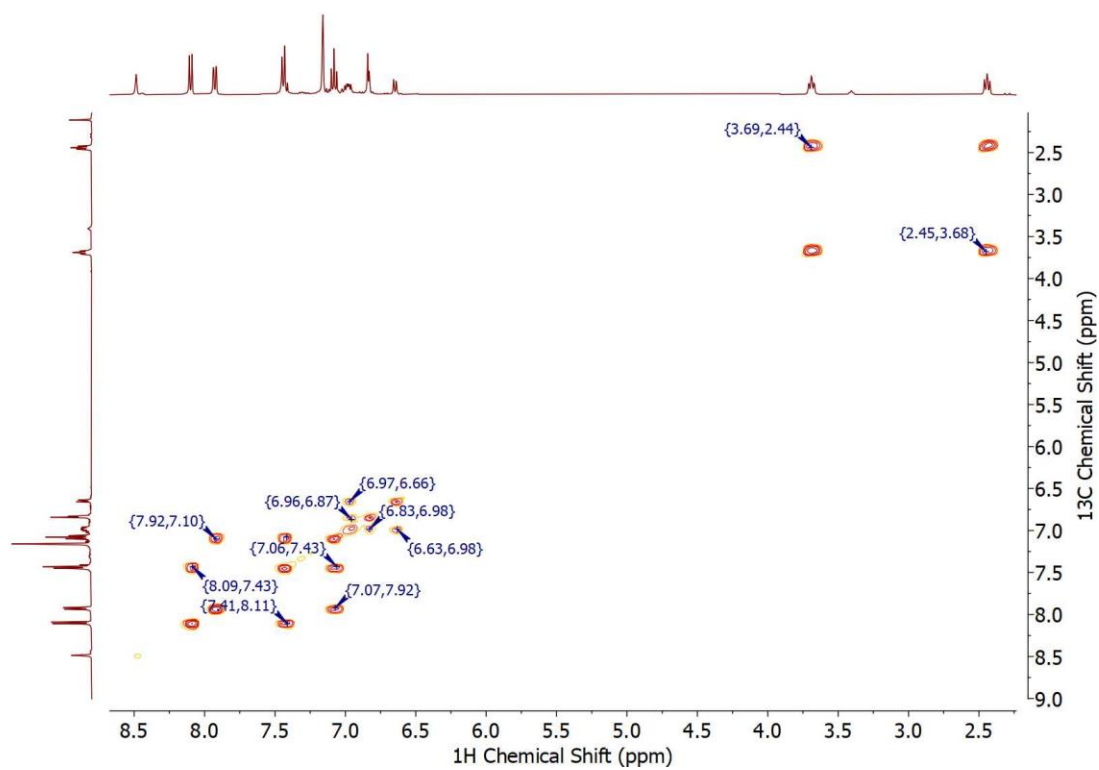


Figure S4. ^1H - ^1H COSY NMR spectrum-expansion of **2** (C_6D_6 , 400 MHz, 25 °C).

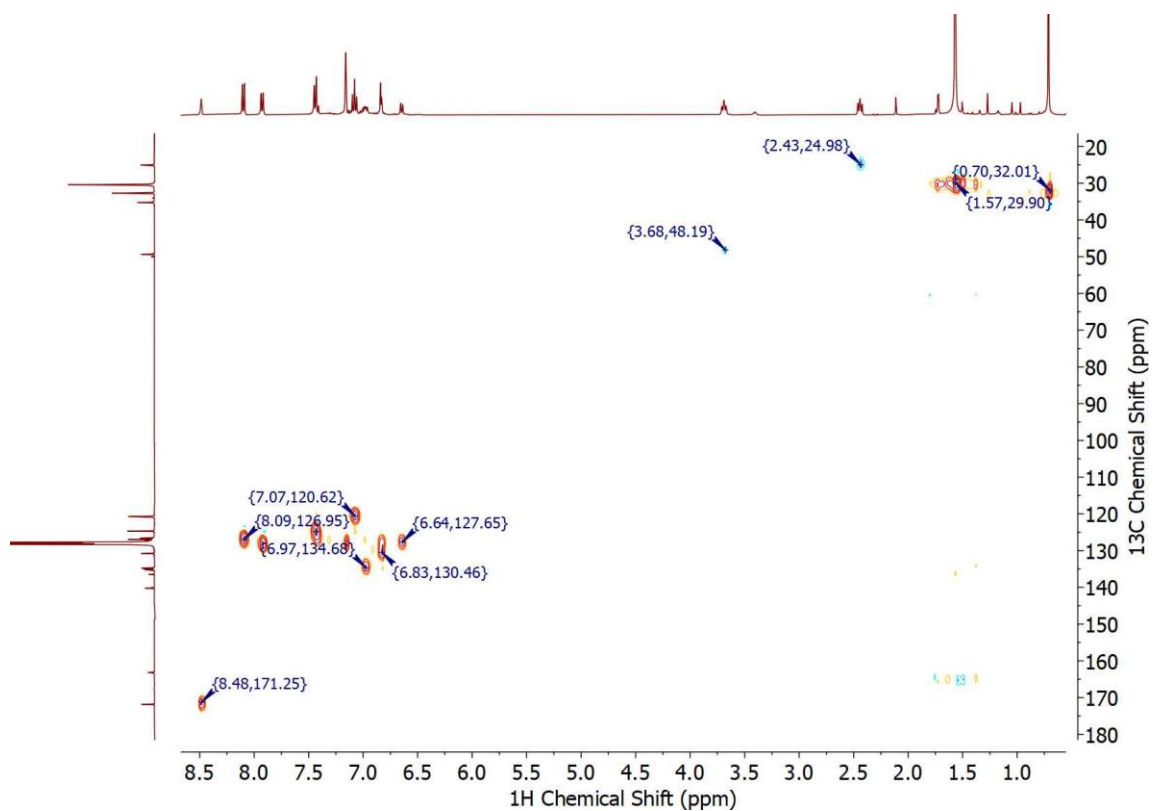


Figure S5. ^1H - ^{13}C HSQC NMR spectrum of **2** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

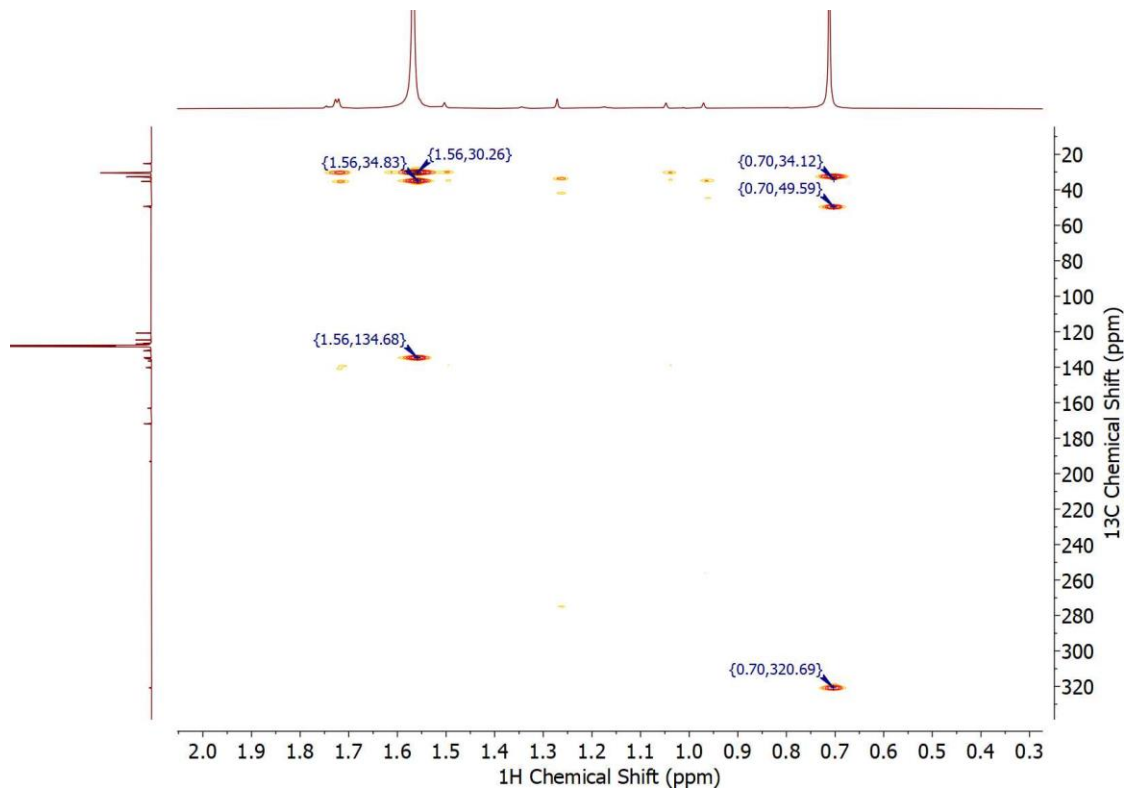


Figure S6. ^1H - ^{13}C HMBC NMR spectrum-expansion of **2** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

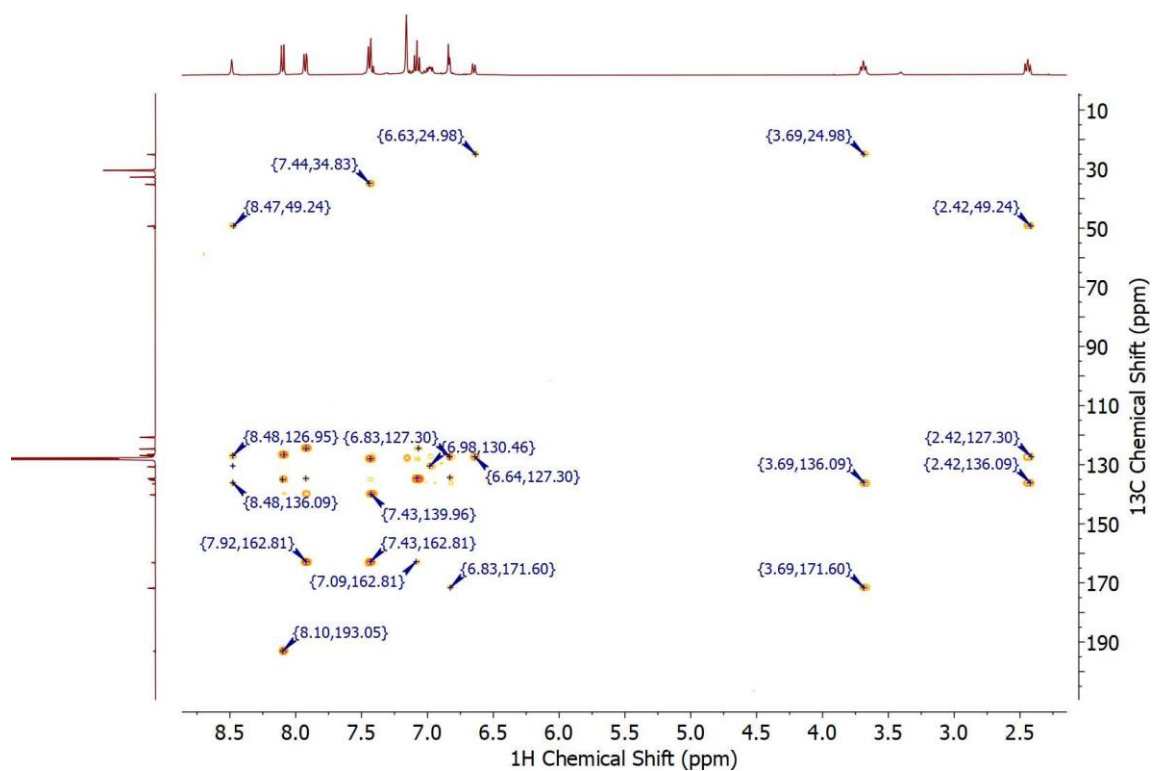


Figure S7. ^1H - ^{13}C HMBC NMR spectrum-expansion of **2** (C_6D_6 , 400 MHz, 25 °C).

3.2. NMR spectra of **3**

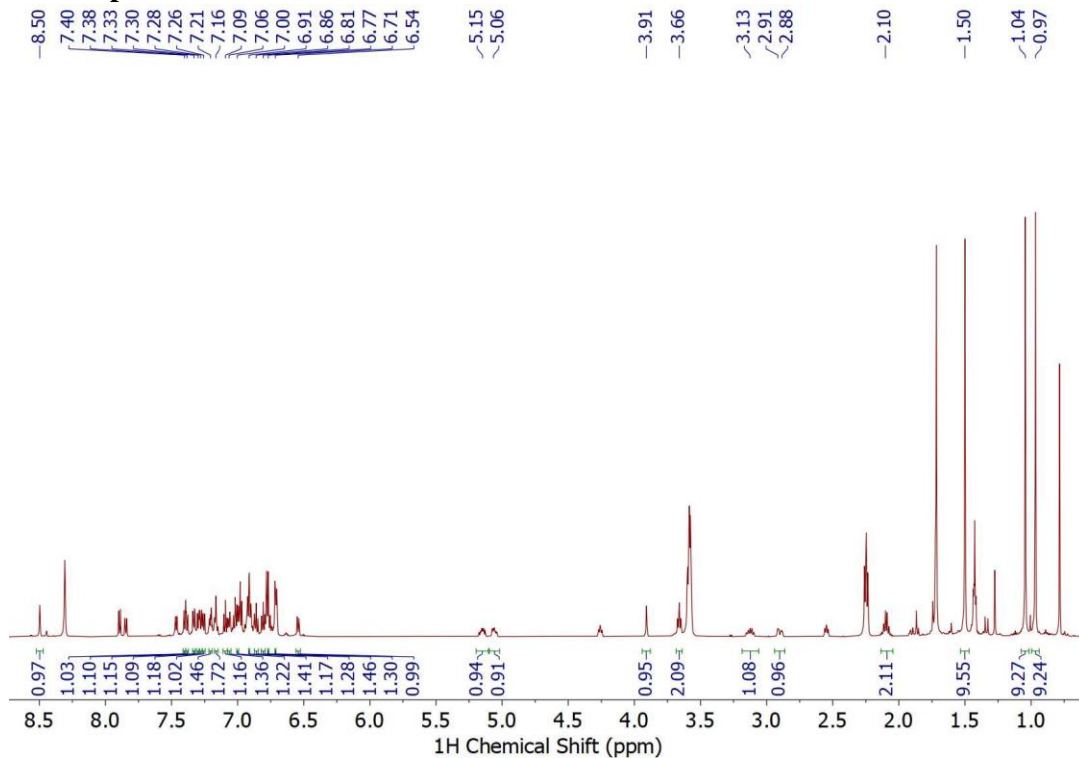


Figure S8. ^1H NMR spectrum of **3** (and **2** + 2,4-dihydroisoquinoline) (C_6D_6 , 500 MHz, 25 °C).

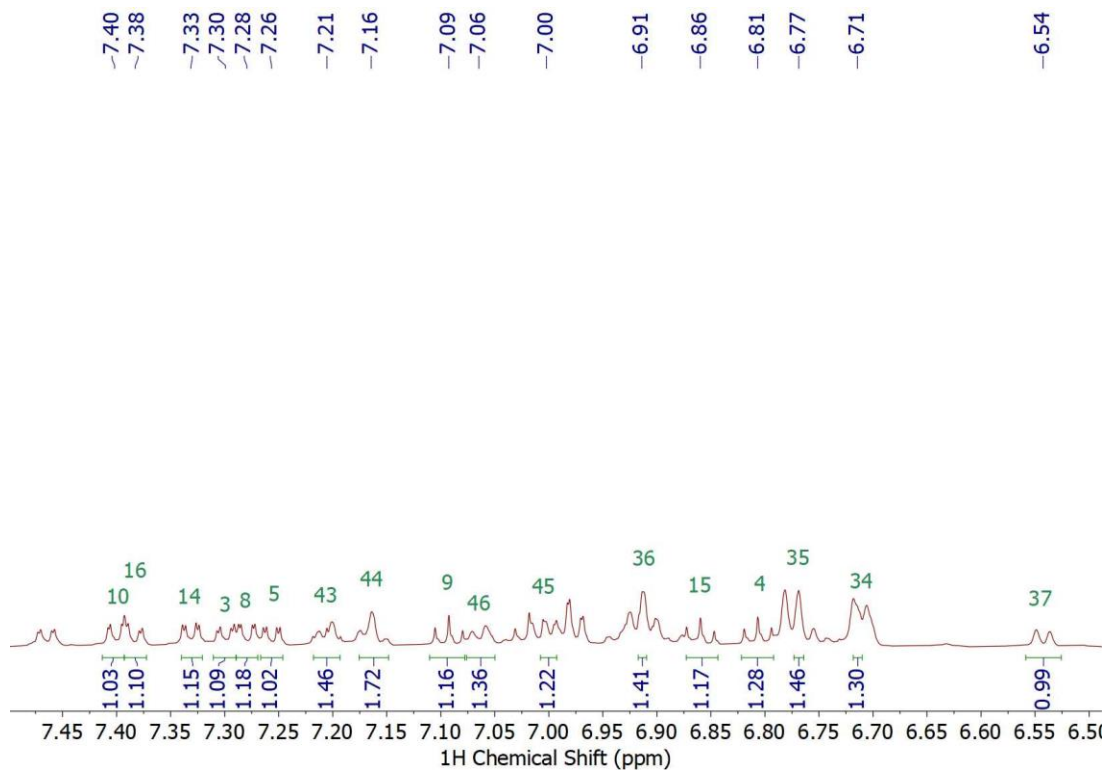


Figure S9. ^1H NMR spectrum-expansion of **3** (and **2** + 2,4-dihydroisoquinoline) (C_6D_6 , 500 MHz, 25 $^\circ\text{C}$).

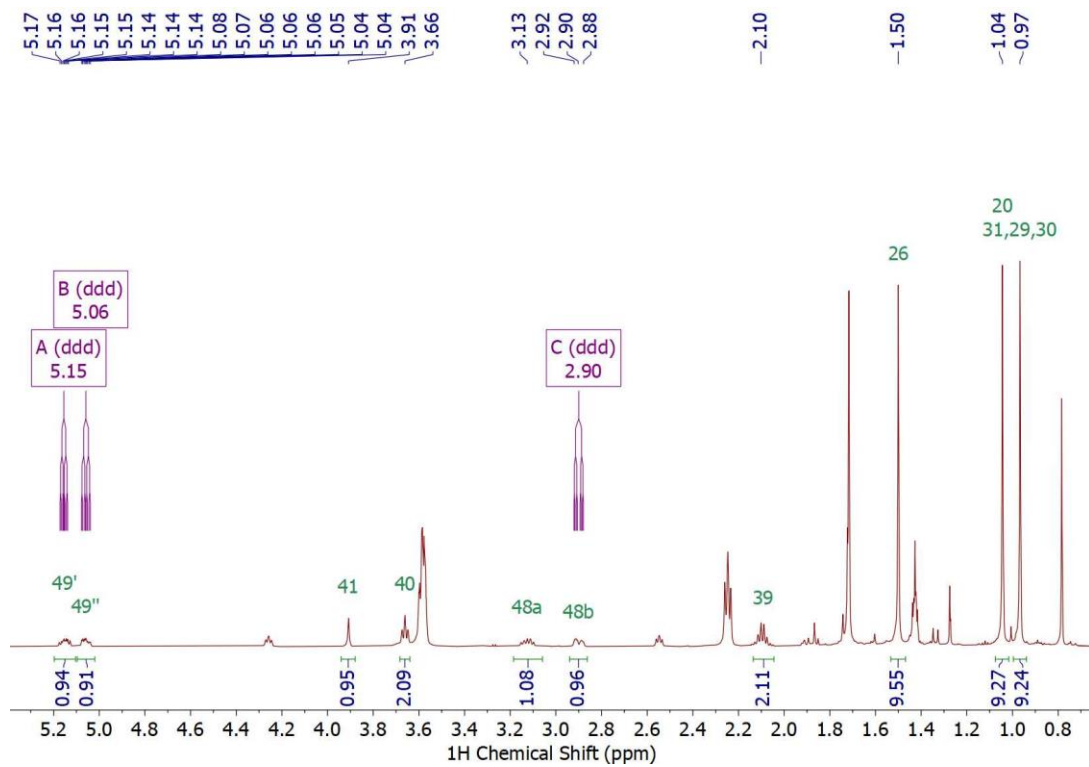


Figure S10. ^1H NMR spectrum of **3** (and **2** + 2,4-dihydroisoquinoline) (C_6D_6 , 500 MHz, 25 $^\circ\text{C}$).

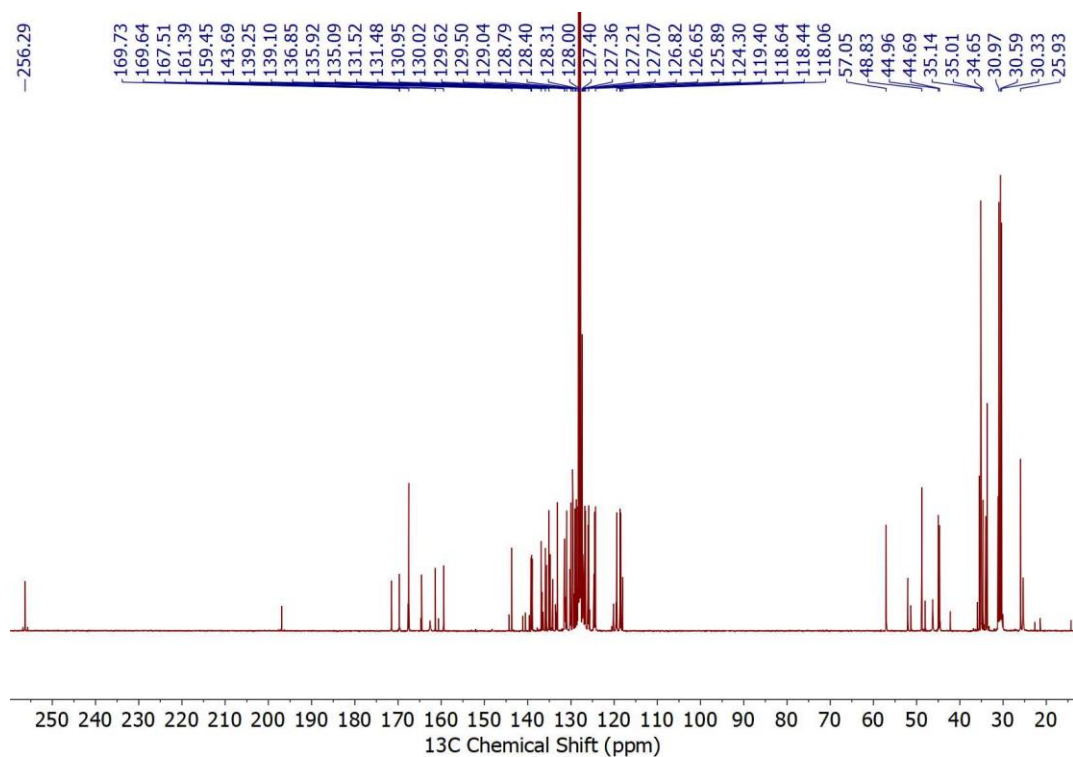


Figure S11. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **3** (and **2** + 2,4-dihydroisoquinoline) (C_6D_6 , 500 MHz, 25 °C).-

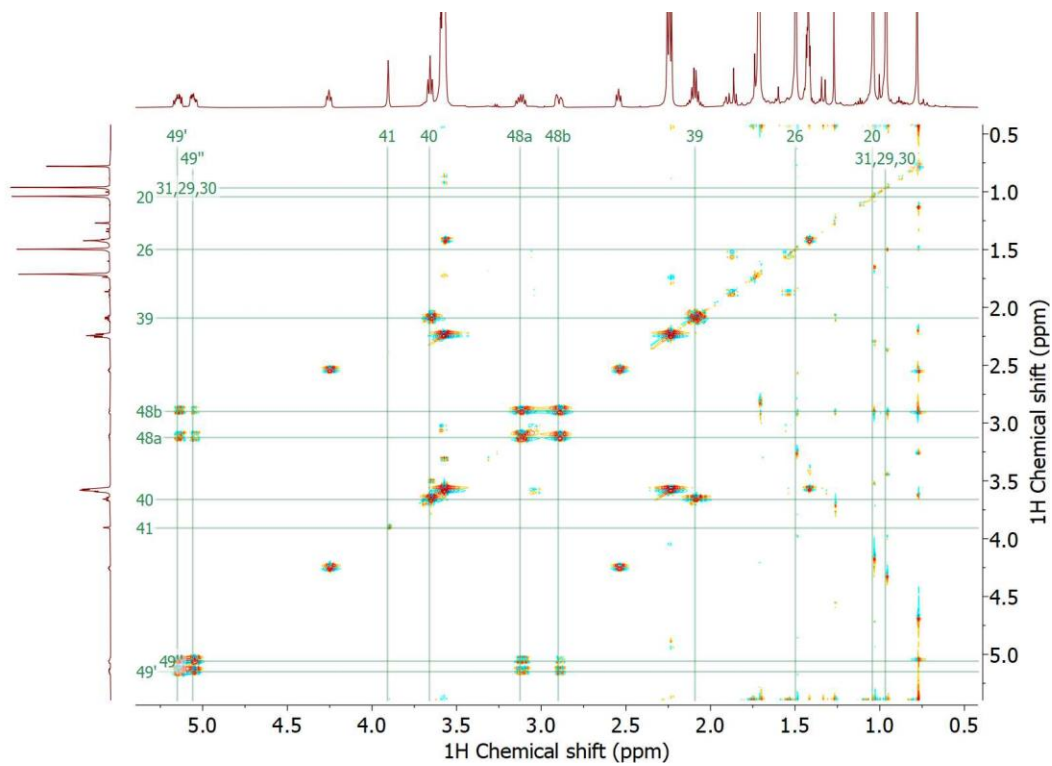


Figure S12. ^1H - ^1H COSY NMR spectrum-expansion of **3** (and **2** + 2,4-dihydroisoquinoline) (C_6D_6 , 500 MHz, 25 °C).

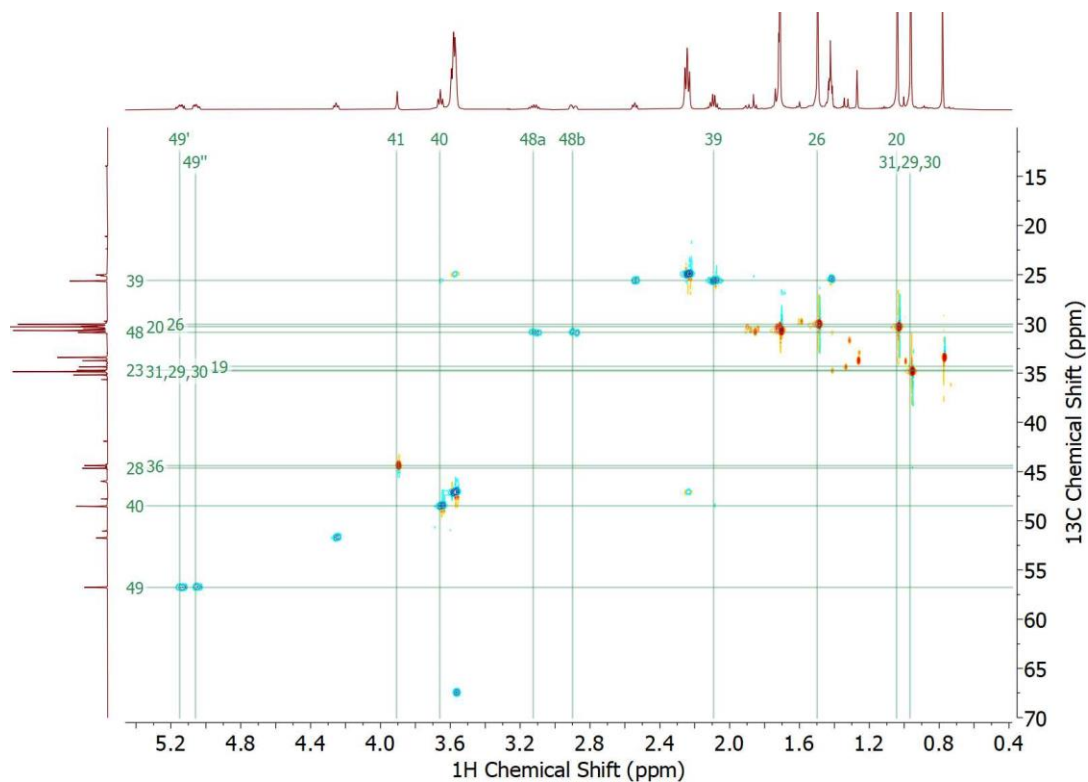


Figure S15. ^1H - ^{13}C HSQC NMR spectrum-expansion of **3** (and **2** + 2,4-dihydroisoquinoline) (C_6D_6 , 500 MHz, 25 $^\circ\text{C}$).

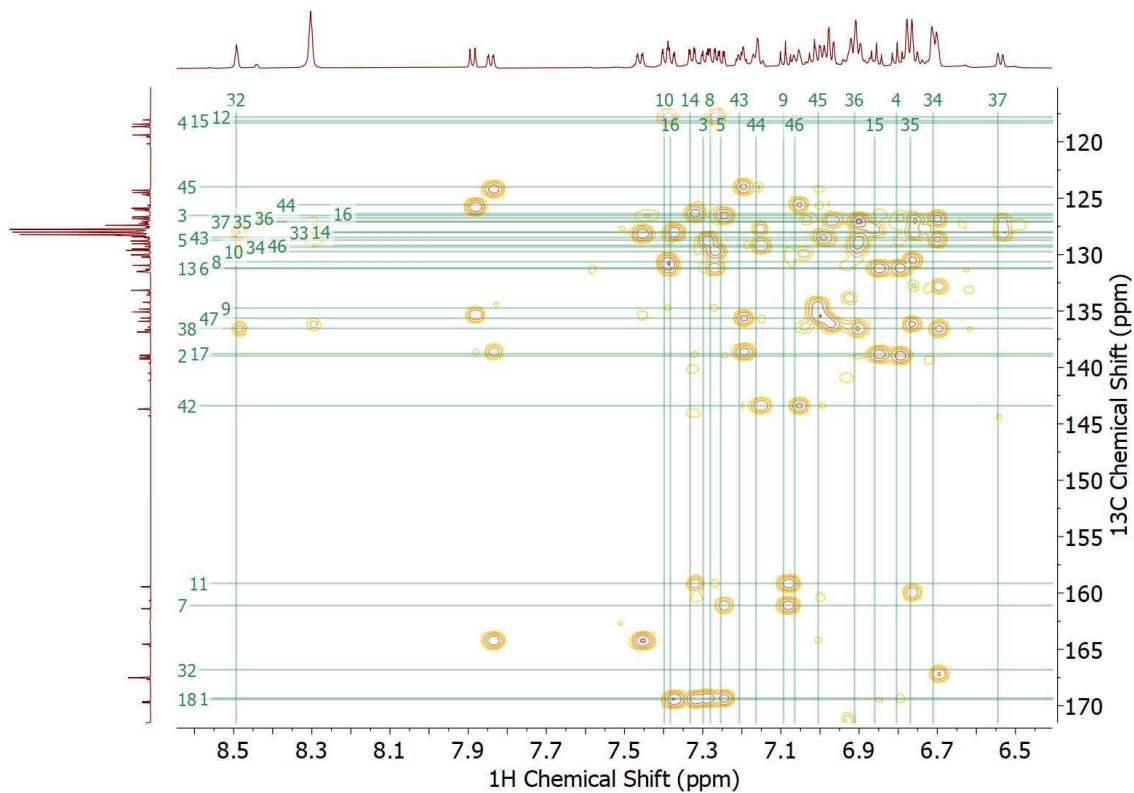


Figure S16. ^1H - ^{13}C HMBC NMR spectrum-expansion of **3** (and **2** + 2,4-dihydroisoquinoline) (C_6D_6 , 500 MHz, 25 $^\circ\text{C}$).

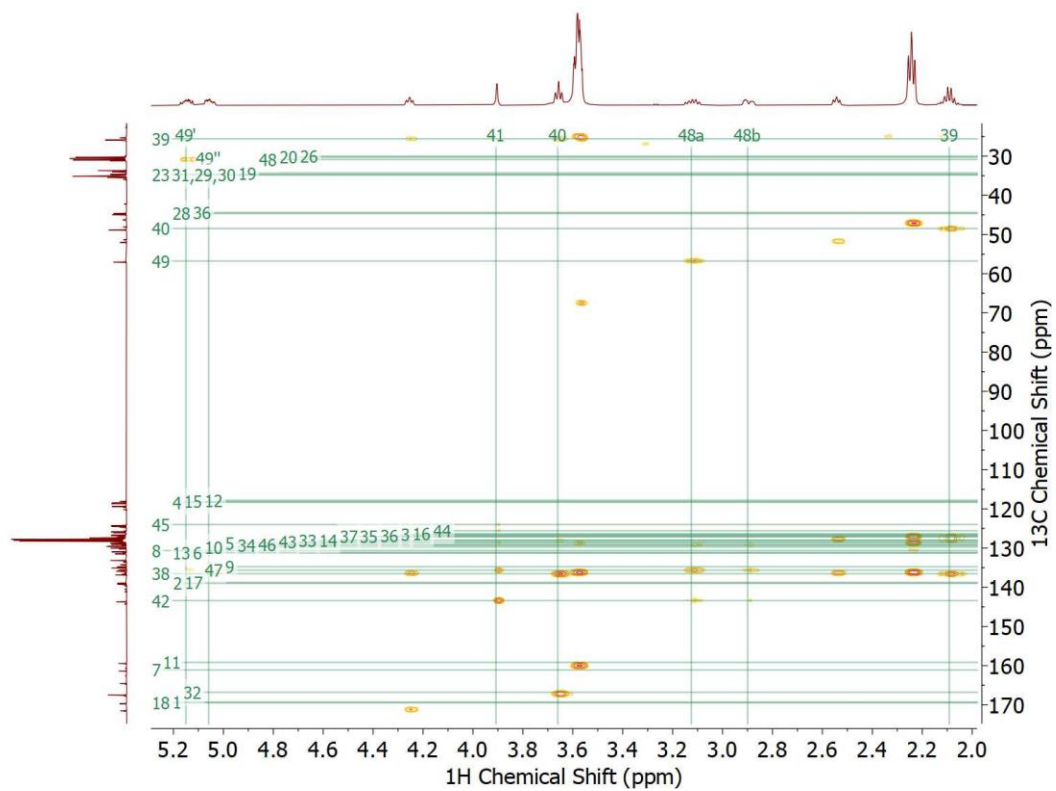


Figure S17. ^1H - ^{13}C HMBC NMR spectrum-expansion of **3** (and **2** + 2,4-dihydroisoquinoline) (C_6D_6 , 500 MHz, 25 $^\circ\text{C}$).

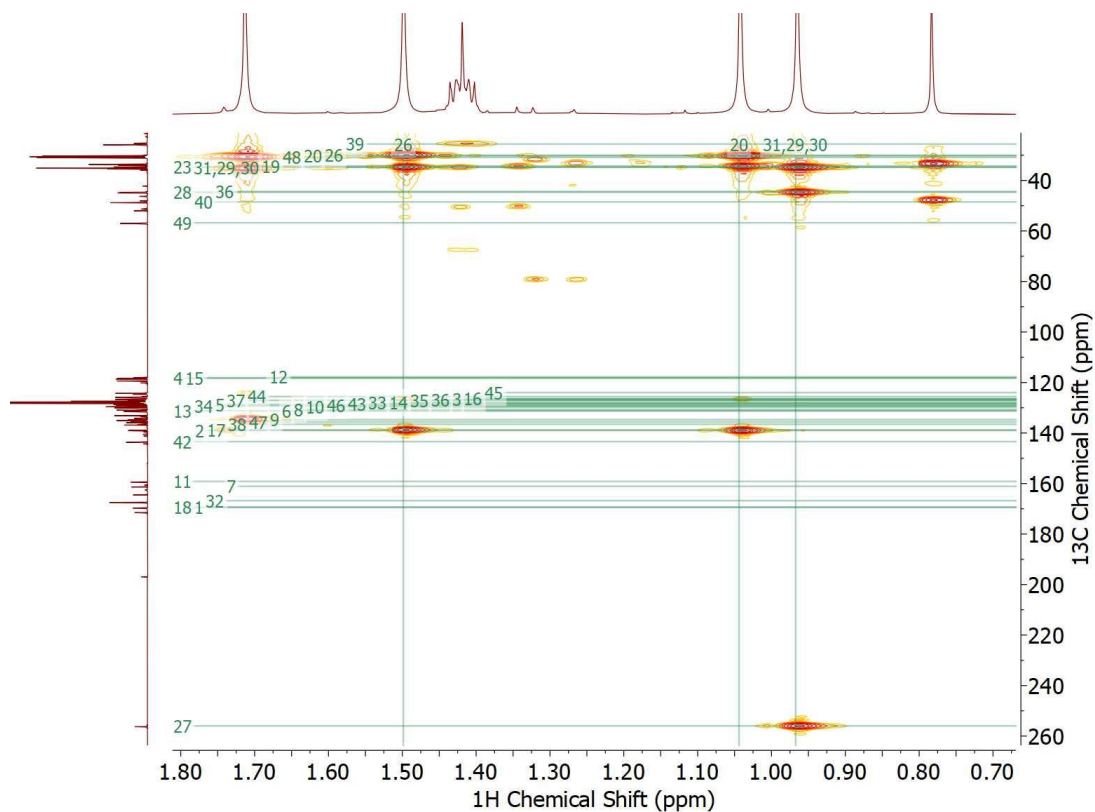


Figure S18. ^1H - ^{13}C HMBC NMR spectrum-expansion of **3** (and **2** + 2,4-dihydroisoquinoline) (C_6D_6 , 500 MHz, 25 $^\circ\text{C}$).

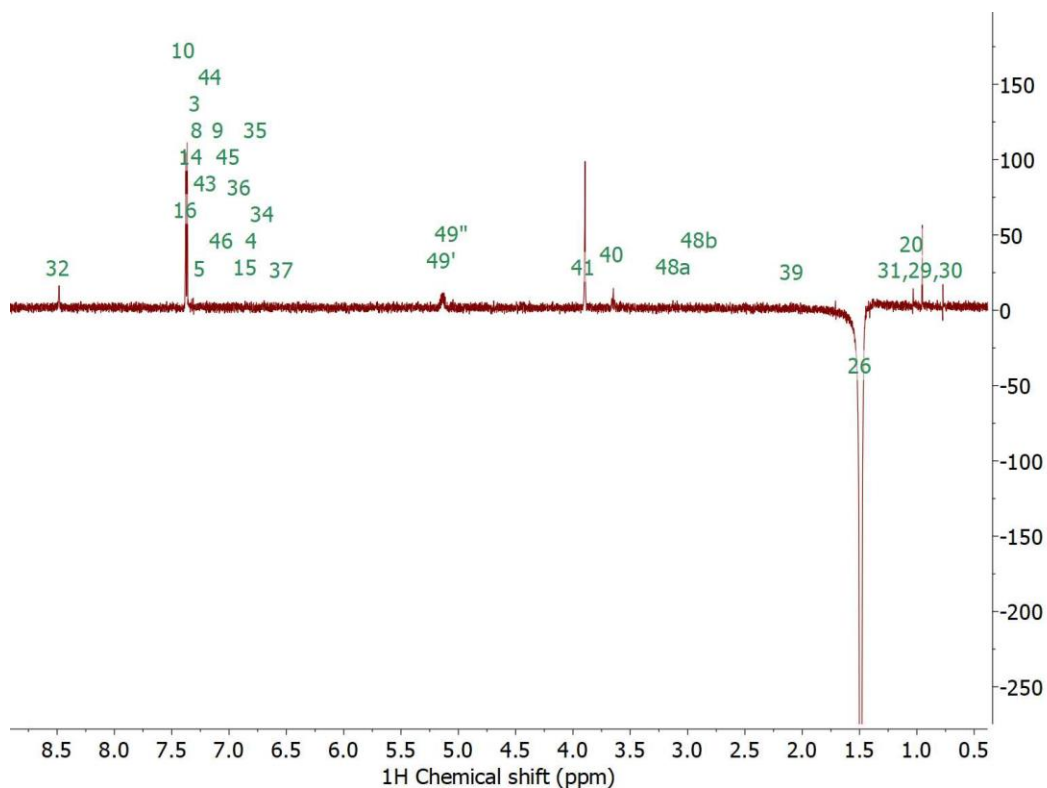


Figure S19. 1D gNOESY spectrum of **3** (and **2** + 2,4-dihydroisoquinoline), collected on H₂₄₋₂₆ (C₆D₆, 500 MHz, 25 °C).

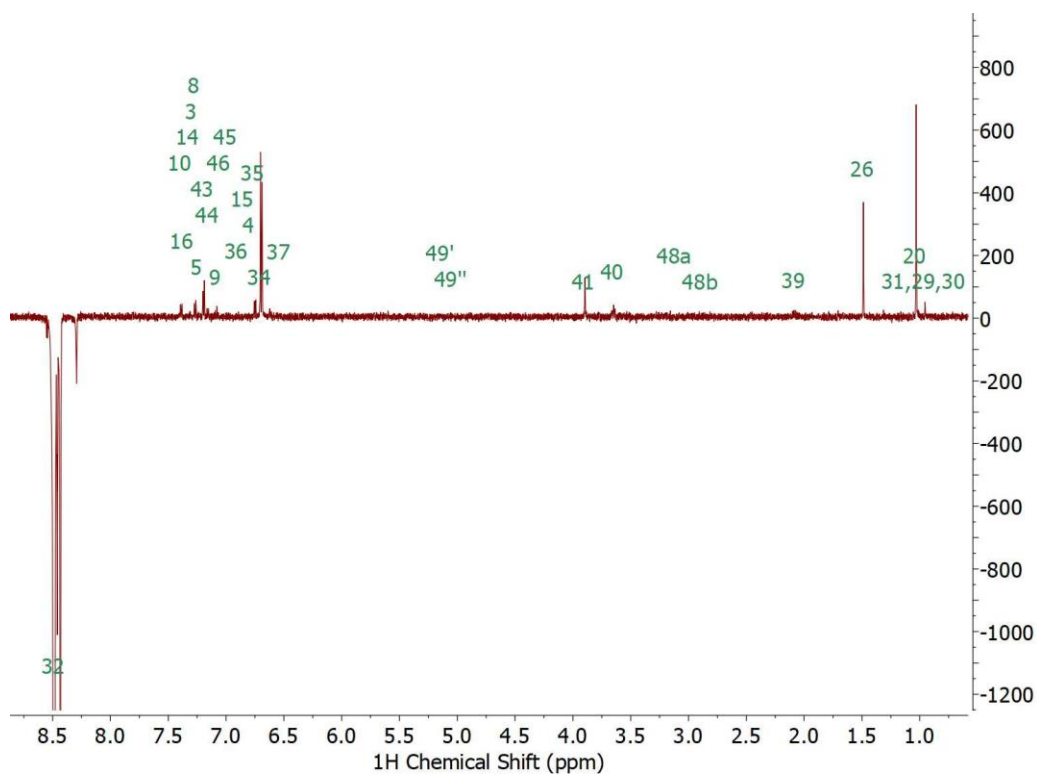


Figure S20. 1D gNOESY spectrum of **3** (and **2** + 2,4-dihydroisoquinoline), collected on H₃₂ (C₆D₆, 500 MHz, 25 °C).

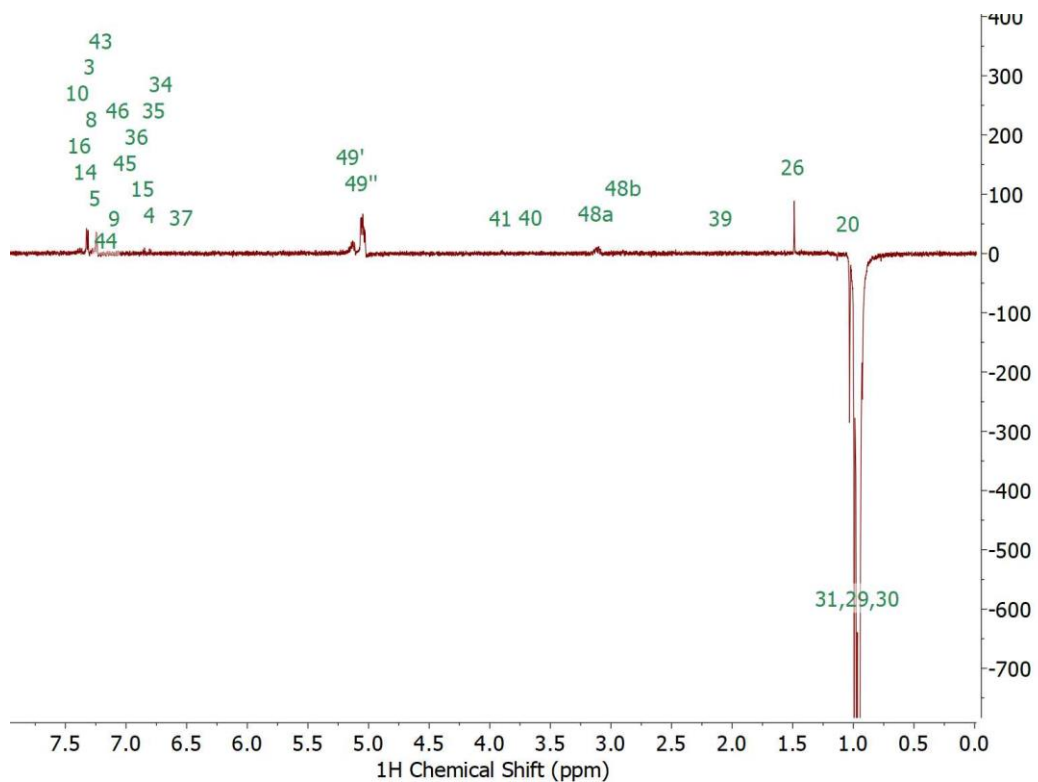


Figure S21. 1D gNOESY spectrum of **3** (and **2** + 2,4-dihydroisoquinoline), collected on H_{29-31} (C_6D_6 , 500 MHz, 25 °C).

3.3. NMR spectra of 4-Me

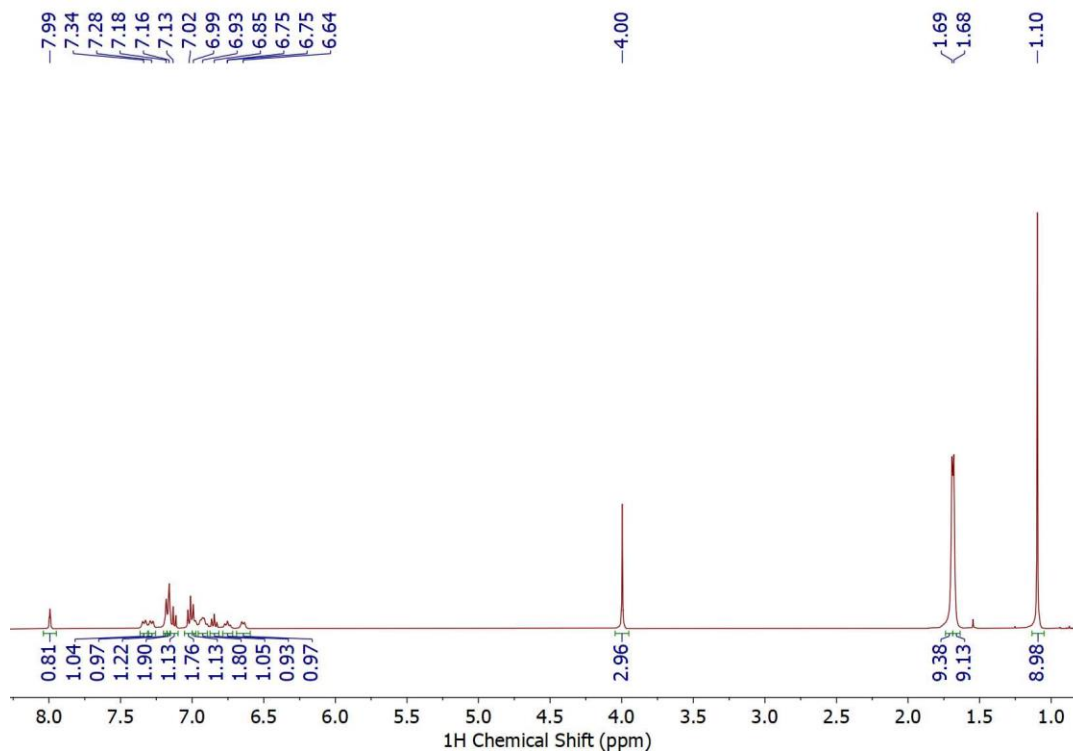


Figure S22. ^1H NMR spectrum of **4-Me** (C_6D_6 , 500 MHz, 25 °C).

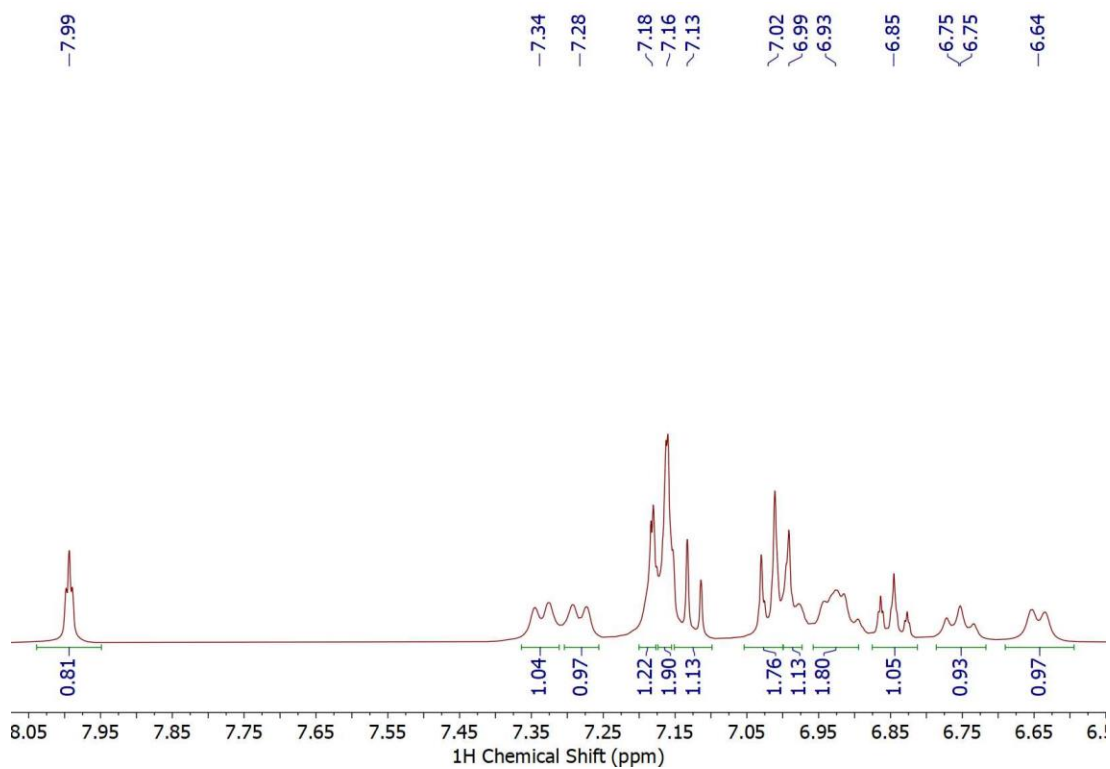


Figure S23. 1H NMR spectrum-expansion of **4-Me** (C_6D_6 , 500 MHz, 25 °C).

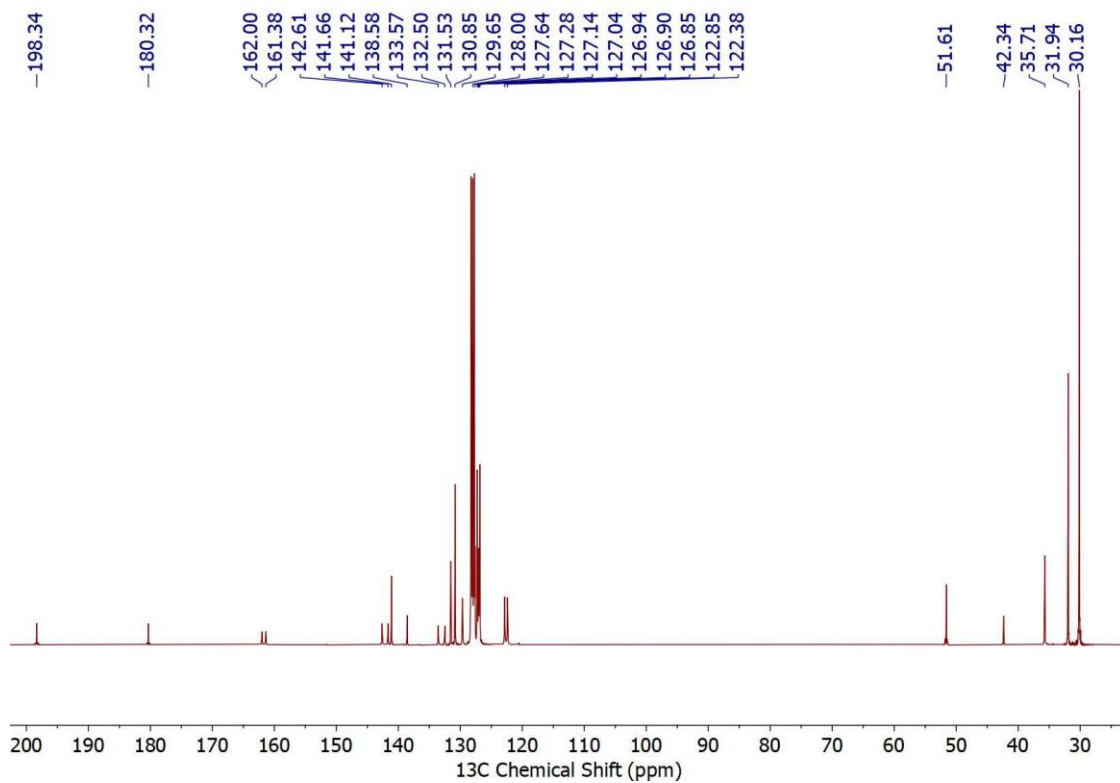


Figure S24. $^{13}C\{^1H\}$ NMR spectrum of **4-Me** (C_6D_6 , 500 MHz, 25 °C).

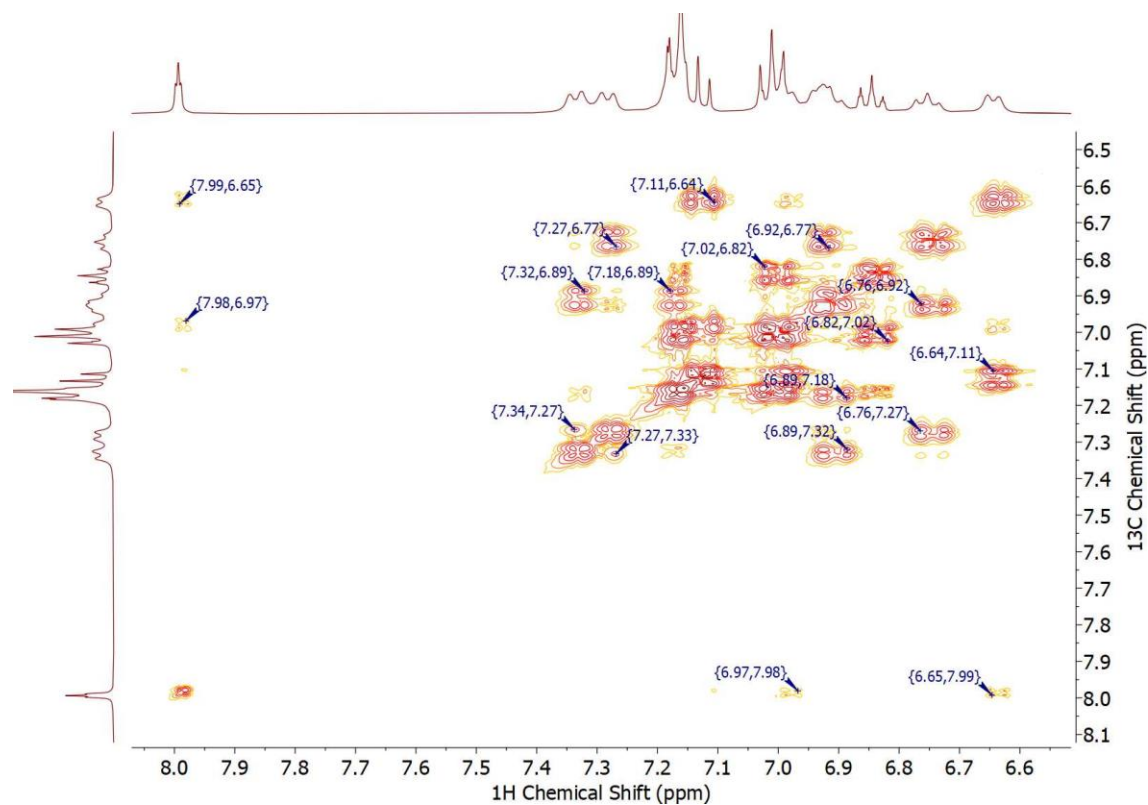


Figure S25. ^1H - ^1H COSY NMR spectrum-expansion of **4-Me** (C_6D_6 , 500 MHz, 25 °C).

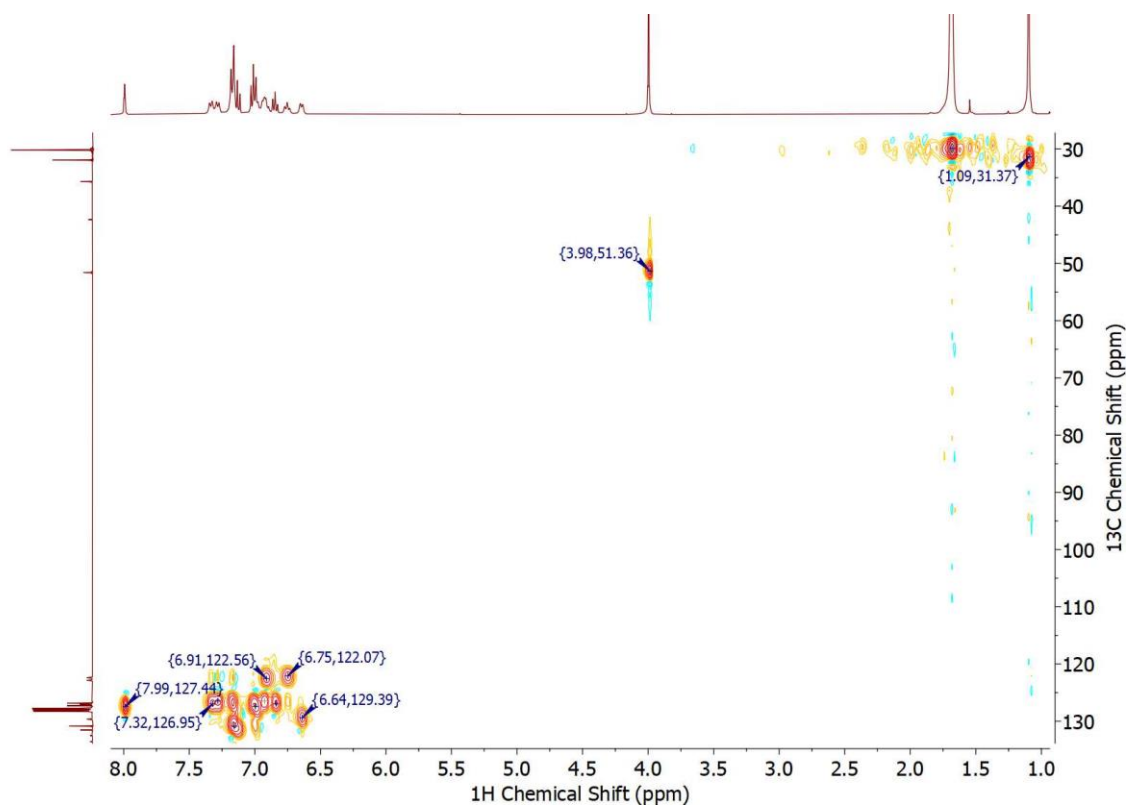


Figure S26. ^1H - ^{13}C HSQC NMR spectrum of **4-Me** (C_6D_6 , 500 MHz, 25 °C).

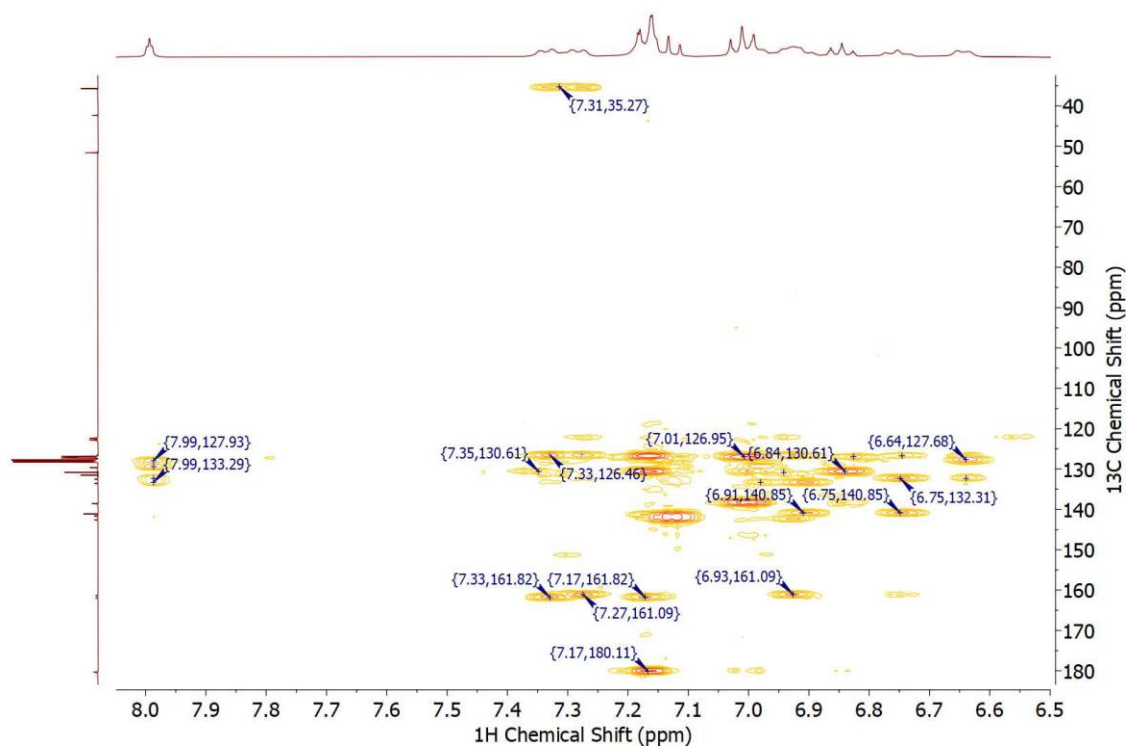


Figure S27. ^1H - ^{13}C HMBC NMR spectrum-expansion of **4-Me** (C_6D_6 , 500 MHz, 25 $^\circ\text{C}$).

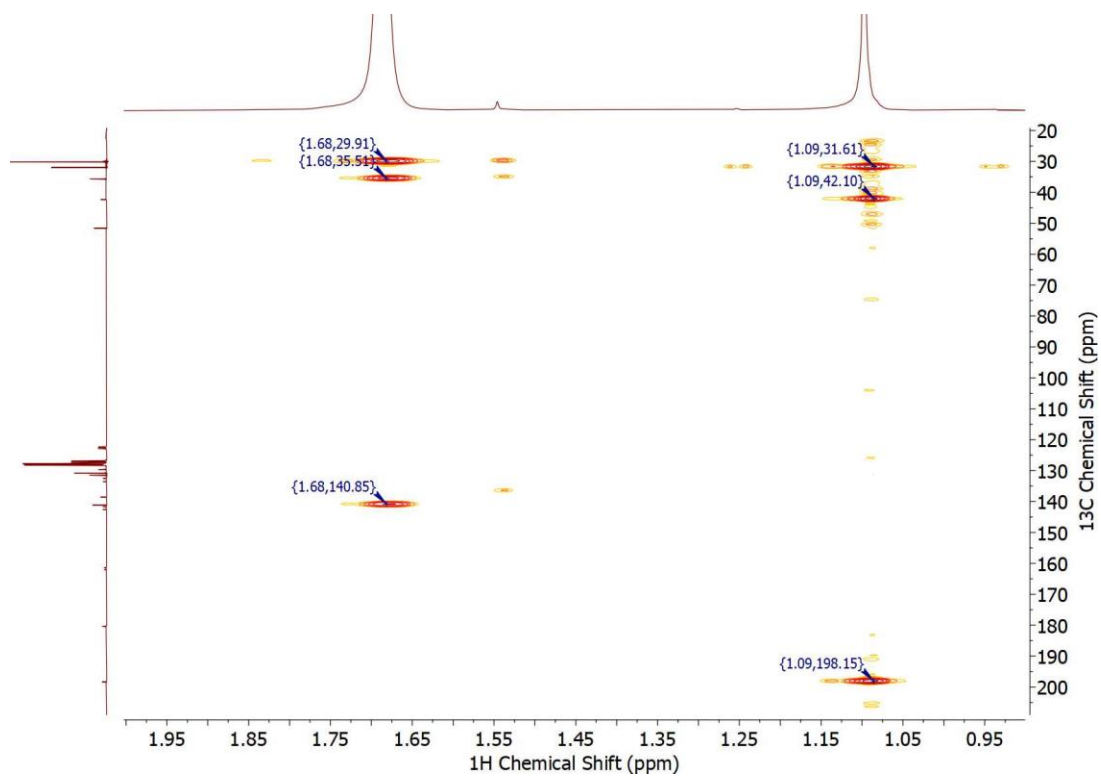


Figure S28. ^1H - ^{13}C HMBC NMR spectrum-expansion of **4-Me** (C_6D_6 , 500 MHz, 25 $^\circ\text{C}$).

3.4. NMR spectra of 4-TMS

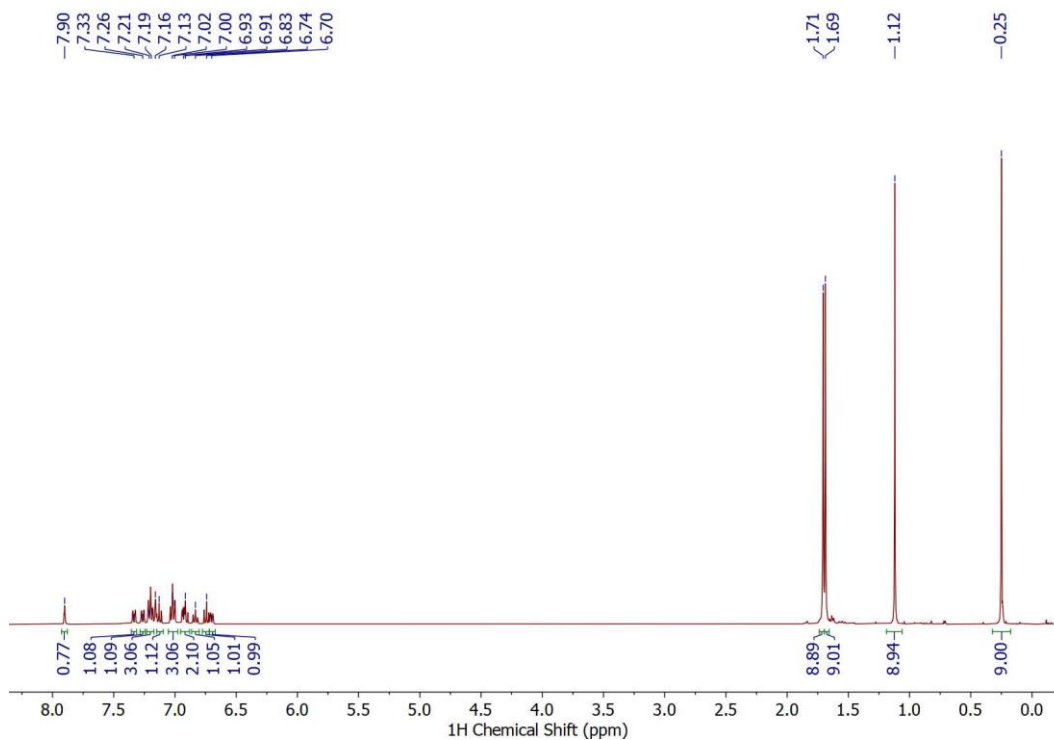


Figure S29. ^1H NMR spectrum of **4-TMS** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

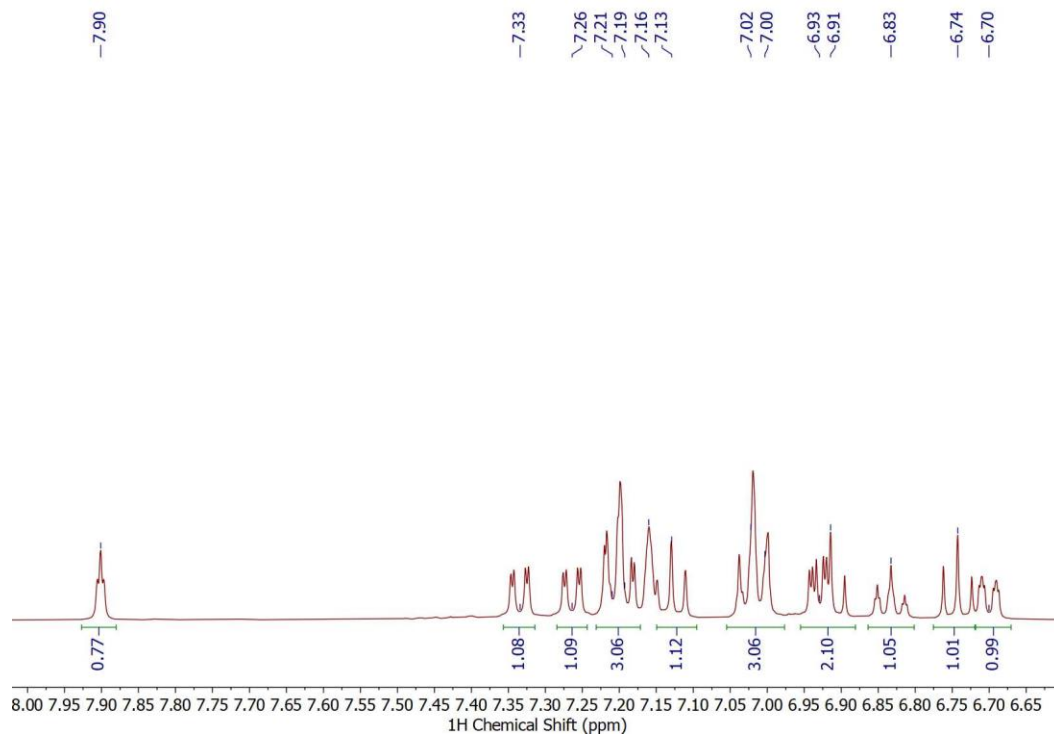


Figure S30. ^1H NMR spectrum-expansion of **4-Me** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

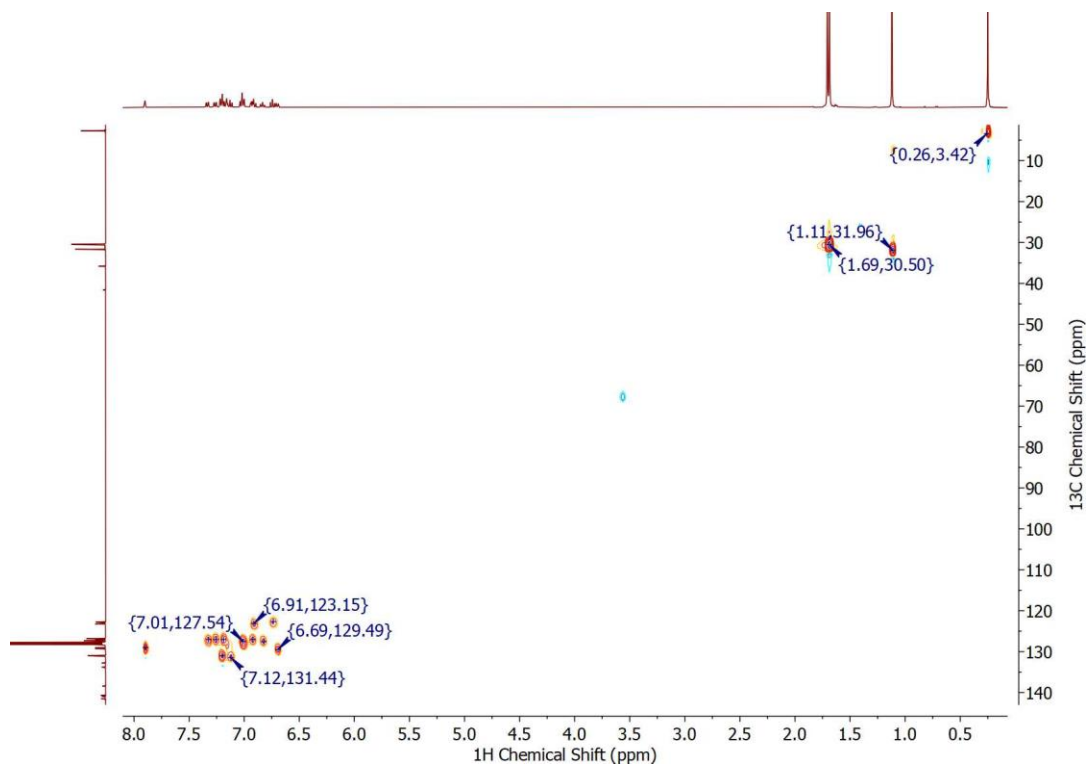


Figure S33. ^1H - ^{13}C HSQC NMR spectrum of **4-Me** (C_6D_6 , 400 MHz, 25 °C).

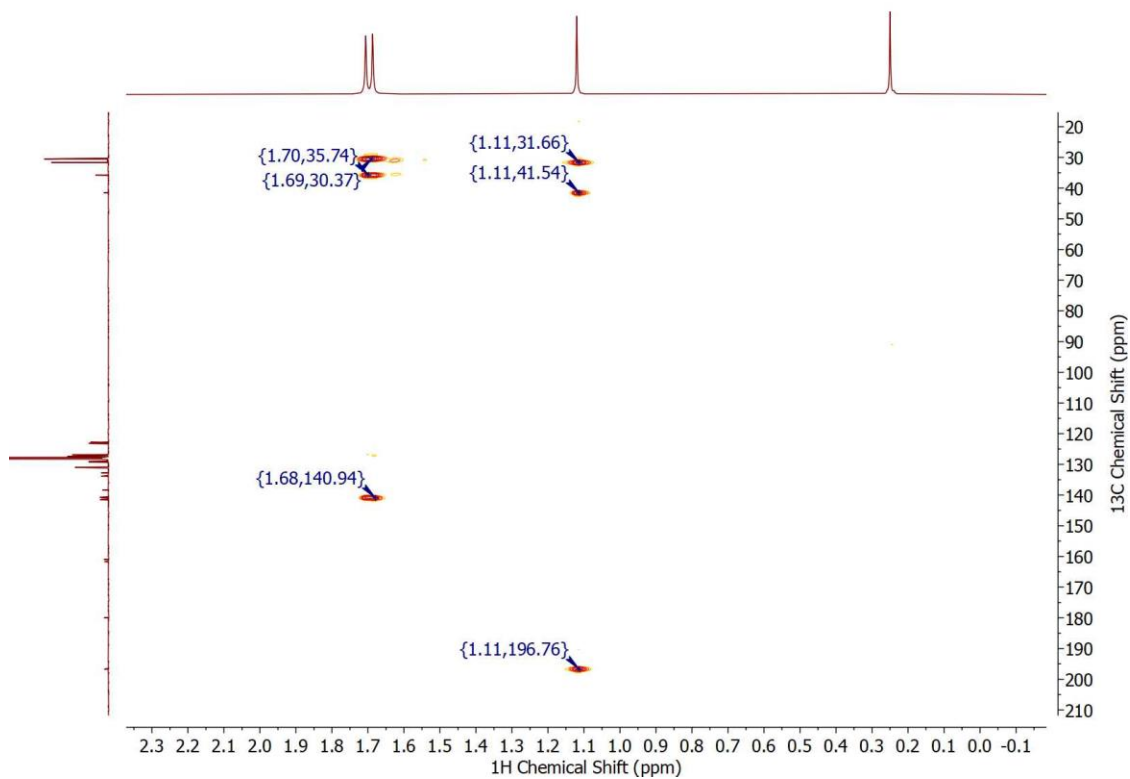


Figure S34. ^1H - ^{13}C HMBC NMR spectrum-expansion of **4-Me** (C_6D_6 , 400 MHz, 25 °C).

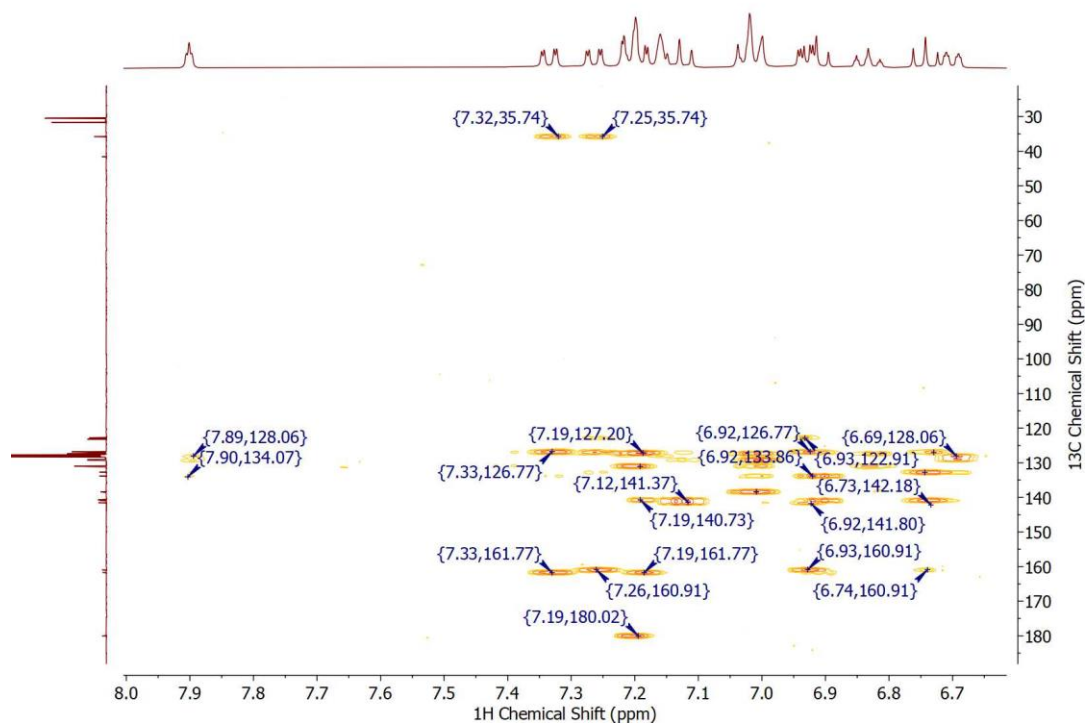


Figure S35. ^1H - ^{13}C HMBC NMR spectrum-expansion of **4-Me** (C_6D_6 , 400 MHz, 25 °C).

3.5. NMR spectra of **4-Ph**

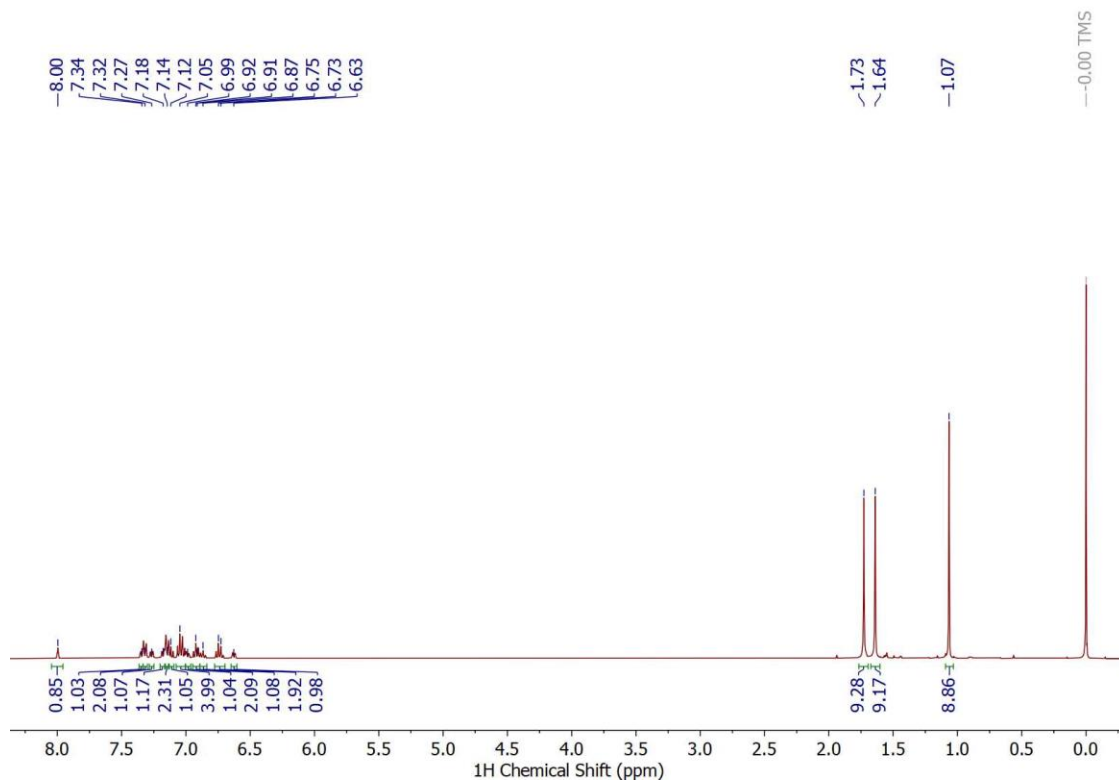


Figure S36. ^1H NMR spectrum of **4-Ph** (C_6D_6 , 400 MHz, 25 °C).

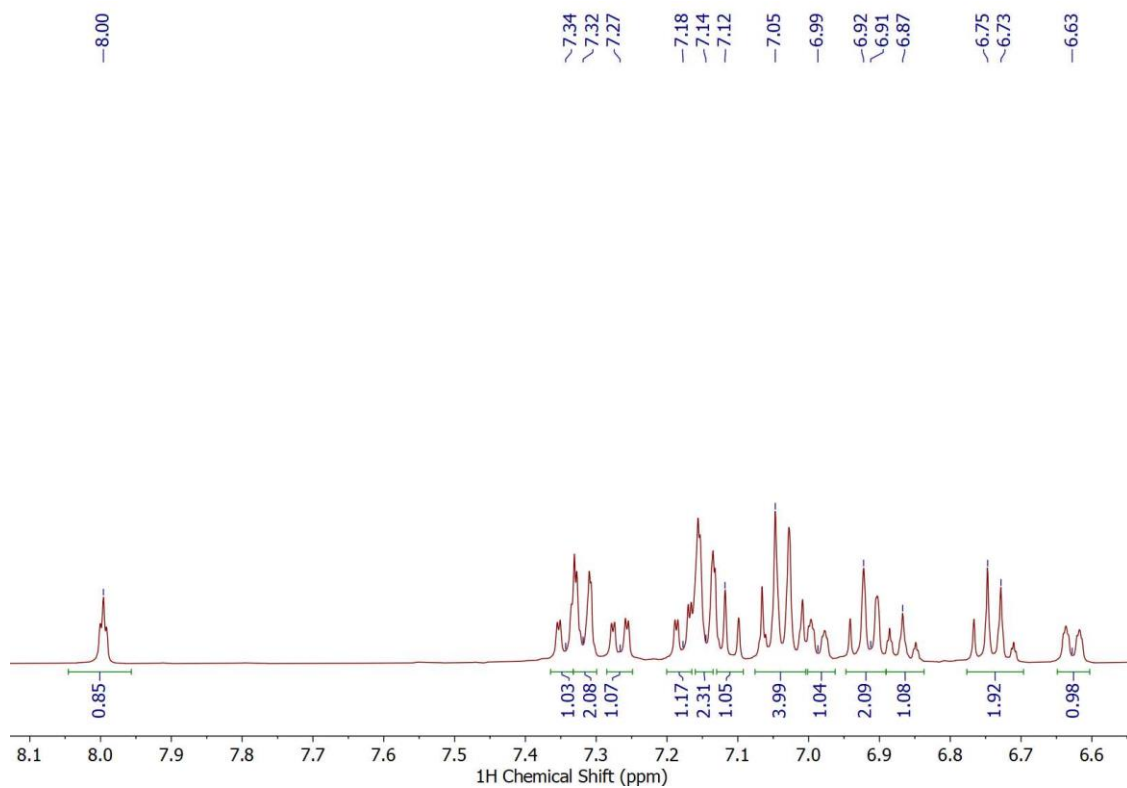


Figure S37. ^1H NMR spectrum-expansion of **4-Ph** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

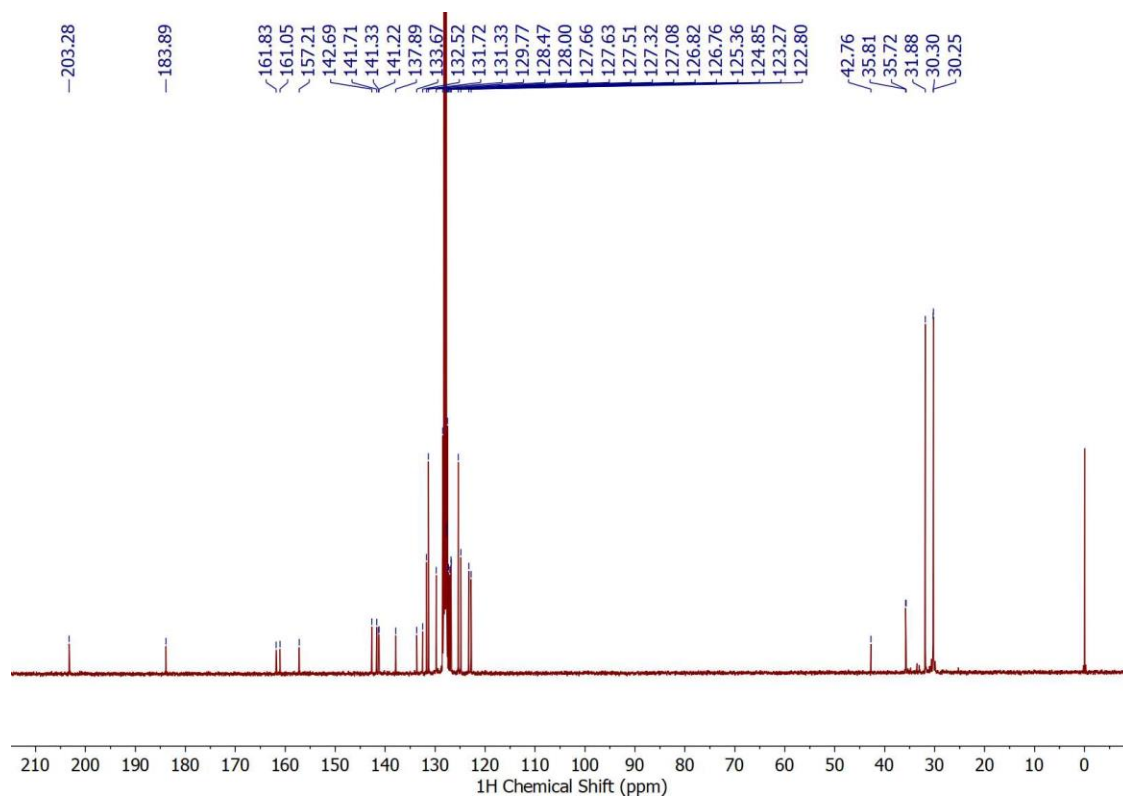


Figure S38. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **4-Ph** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

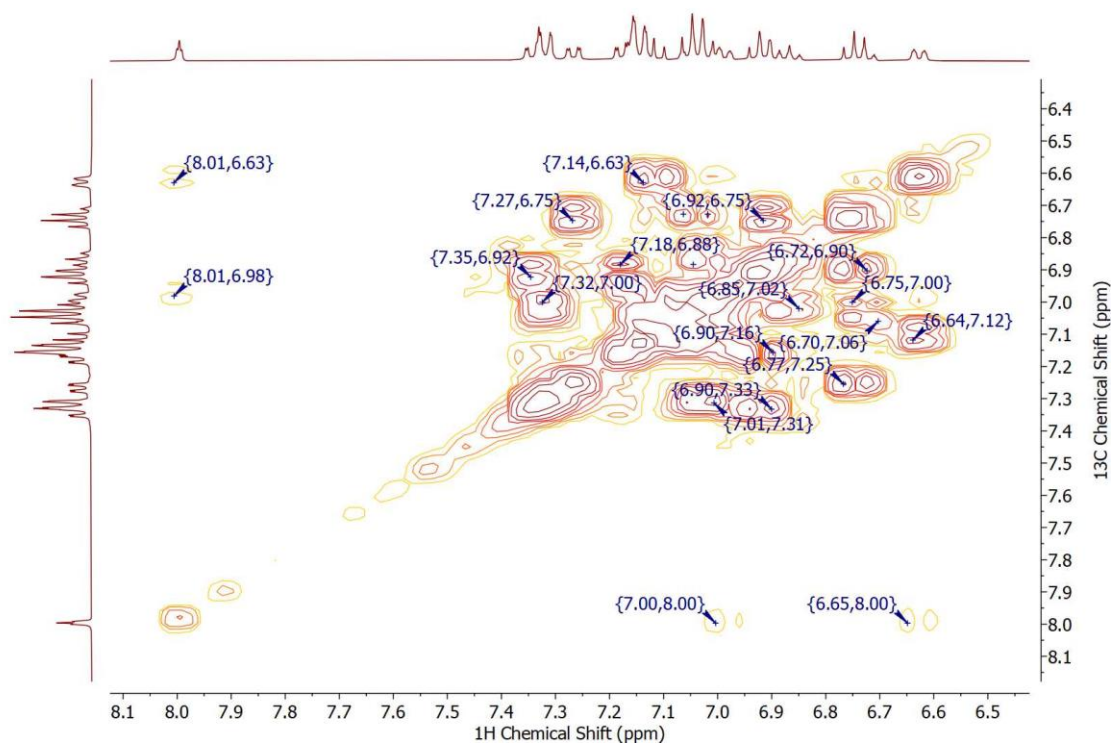


Figure S39. ^1H - ^1H COSY NMR spectrum-expansion of **4-Ph** (C_6D_6 , 400 MHz, 25 °C).

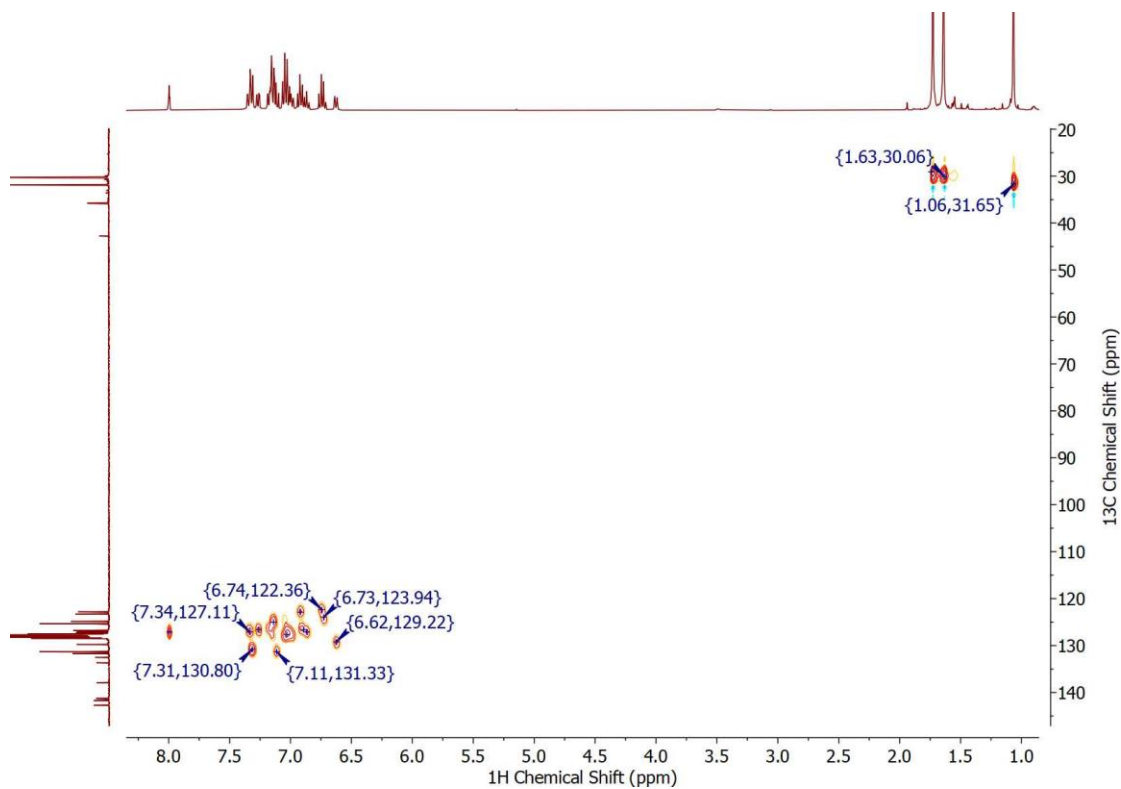


Figure S40. ^1H - ^{13}C HSQC NMR spectrum of **4-Ph** (C_6D_6 , 400 MHz, 25 °C).

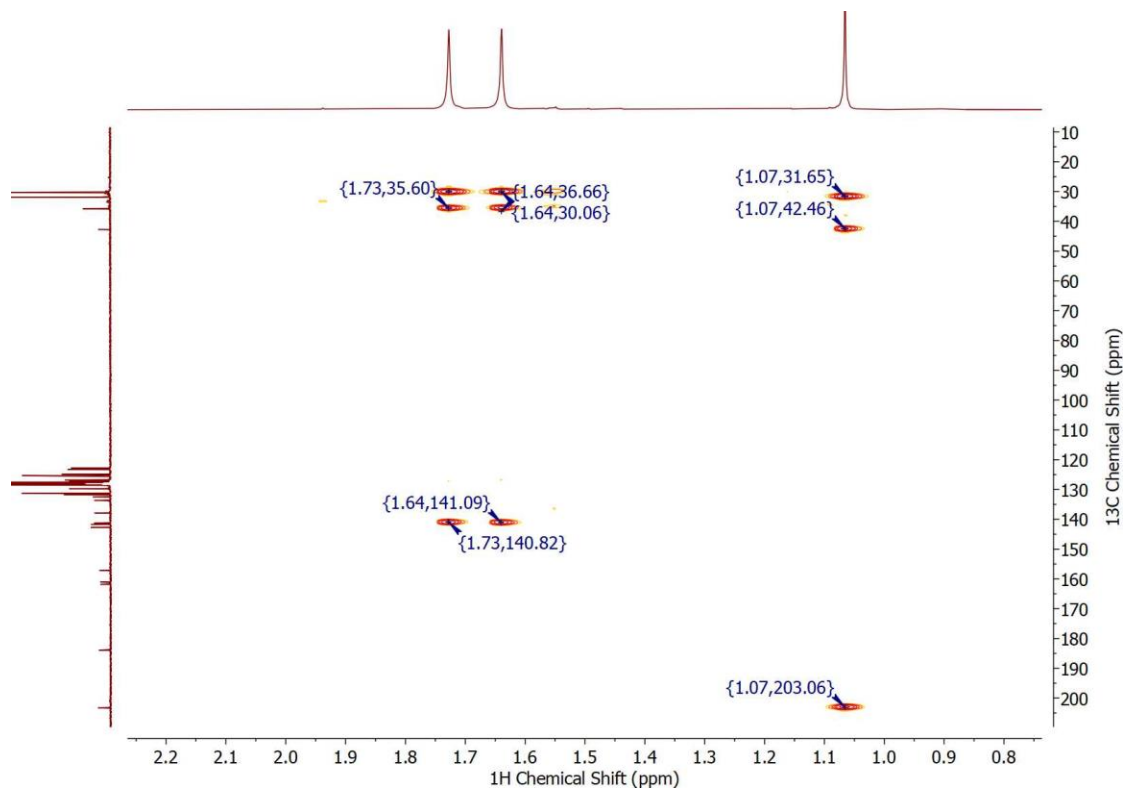


Figure S41. ^1H - ^{13}C HMBC NMR spectrum-expansion of **4-Ph** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

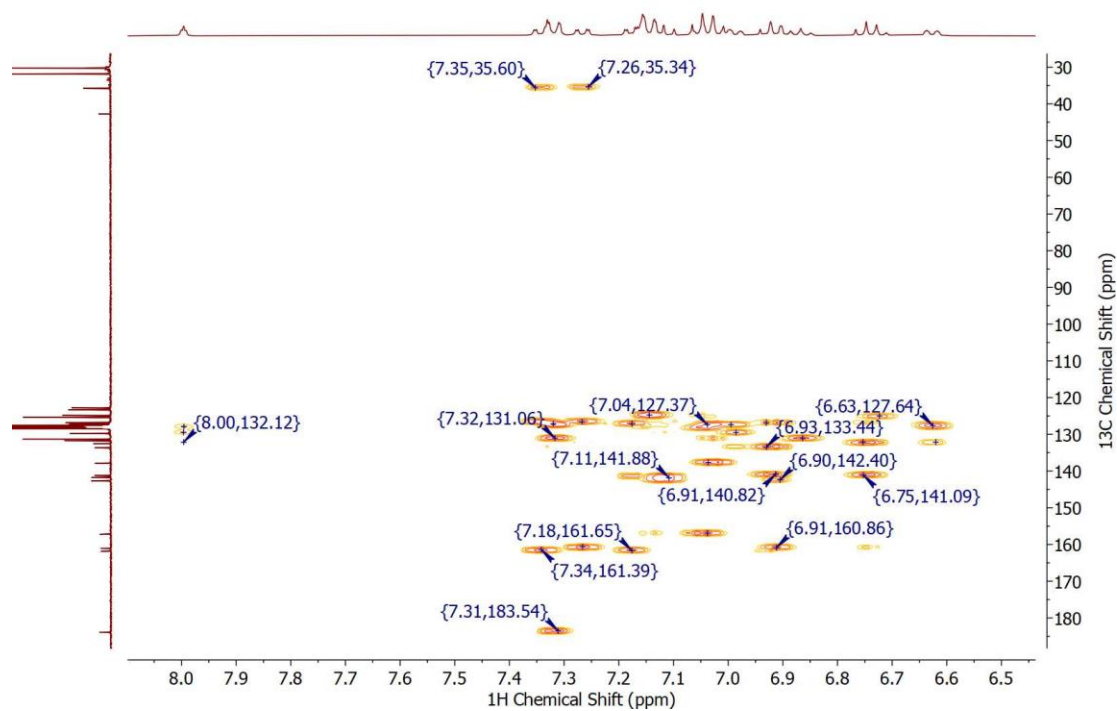


Figure S42. ^1H - ^{13}C NMR HMBC spectrum-expansion of **4-Ph** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

3.6. NMR spectra of 4-Bn

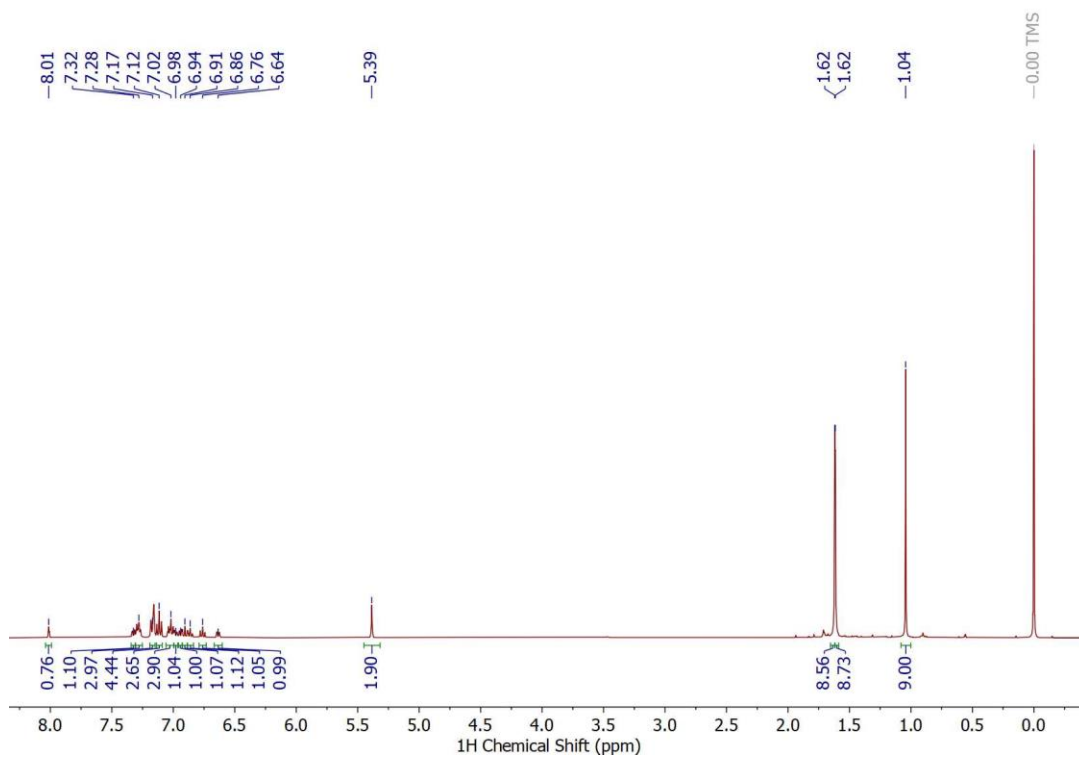


Figure S43. ^1H NMR spectrum of **4-Bn** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

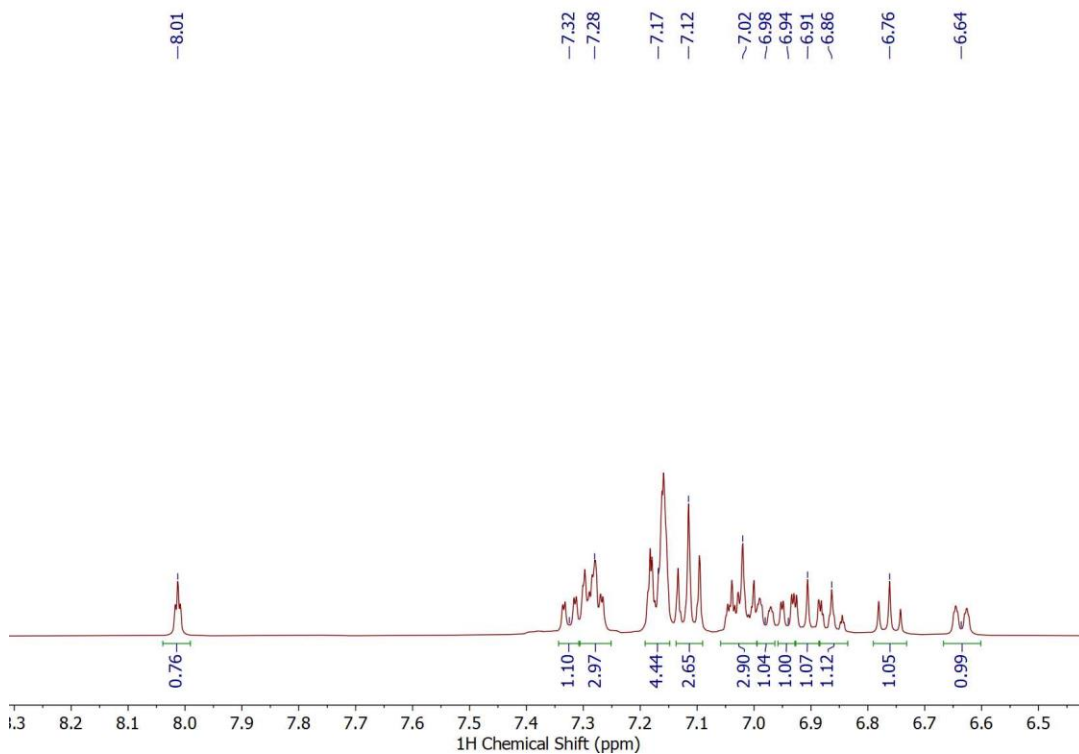


Figure S44. ^1H NMR spectrum-expansion of **4-Bn** (C_6D_6 , 400 MHz, 25 $^\circ\text{C}$).

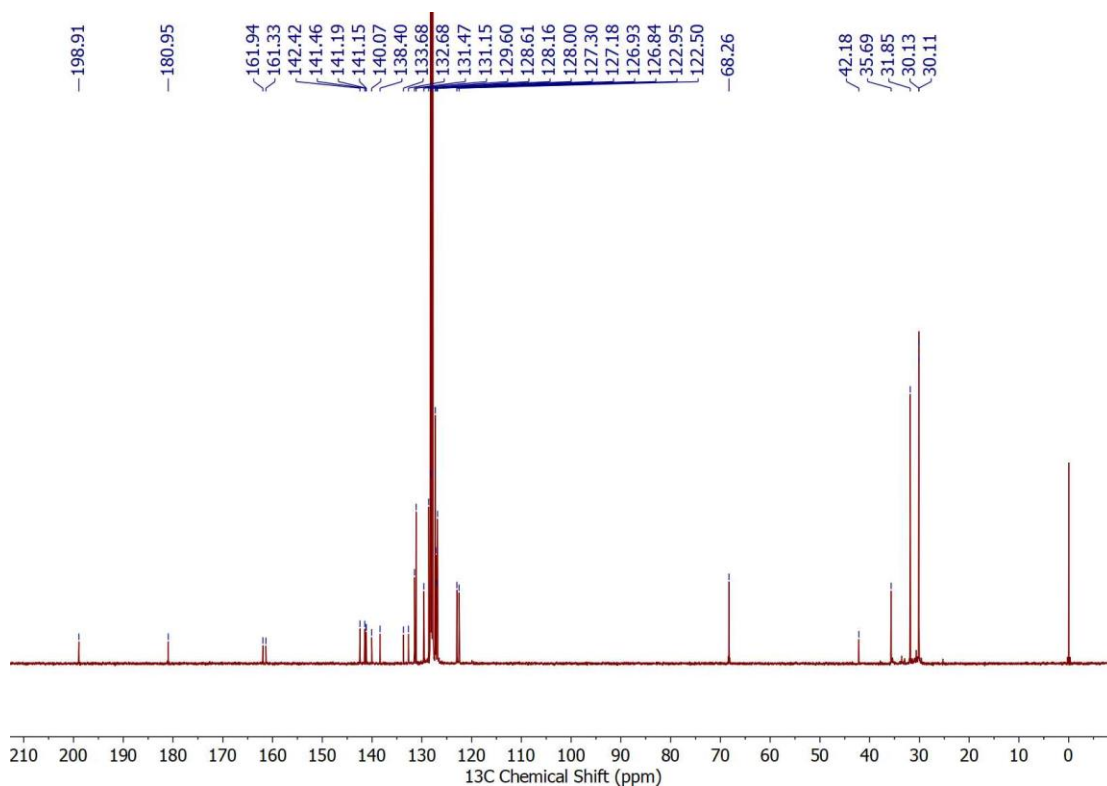


Figure S45. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **4-Bn** (C_6D_6 , 400 MHz, 25 °C).

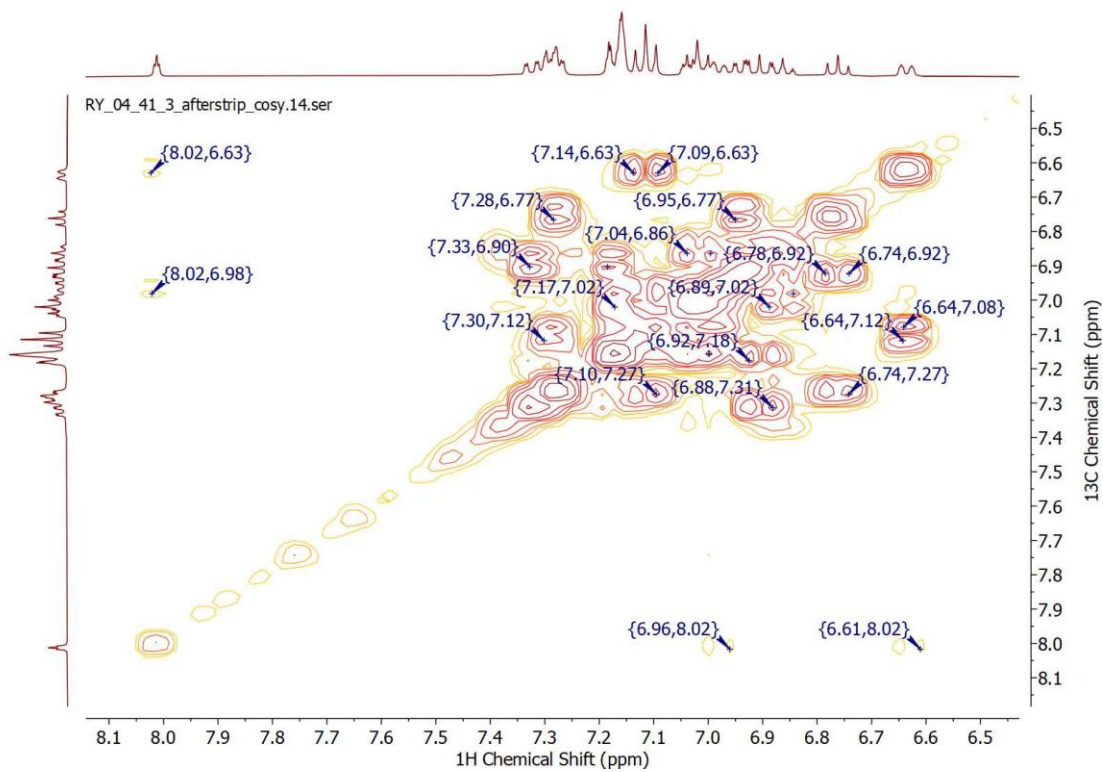


Figure S46. ^1H - ^{13}C COSY NMR spectrum-expansion of **4-Bn** (C_6D_6 , 400 MHz, 25 °C).

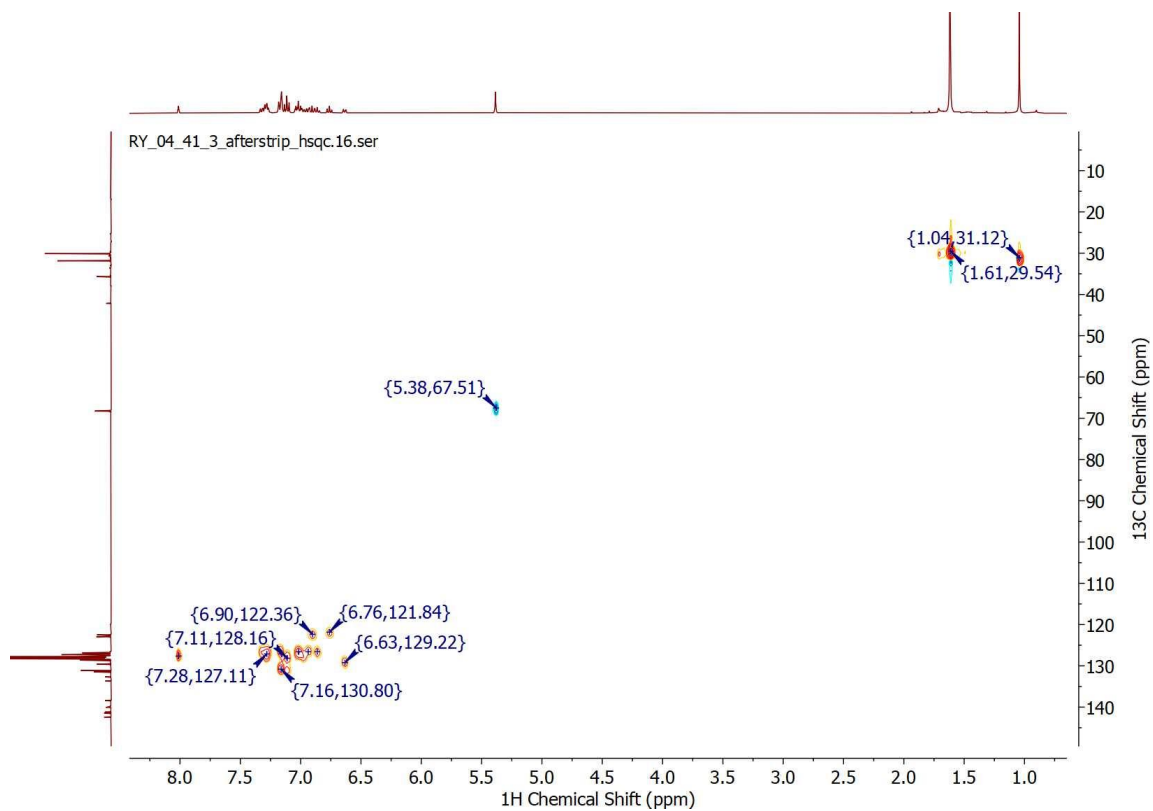


Figure S47. ^1H - ^{13}C HSQC NMR spectrum of **4-Bn** (C_6D_6 , 400 MHz, 25 °C).

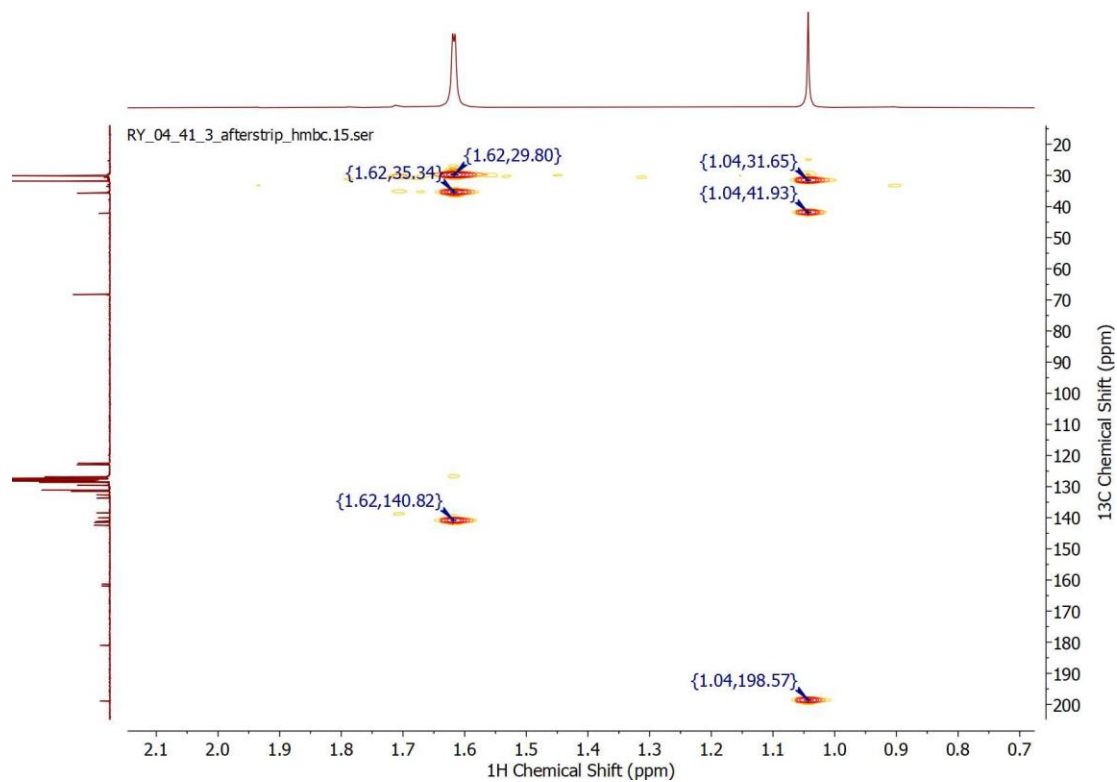


Figure S48. ^1H - ^{13}C HMBC NMR spectrum-expansion of **4-Bn** (C_6D_6 , 400 MHz, 25 °C).

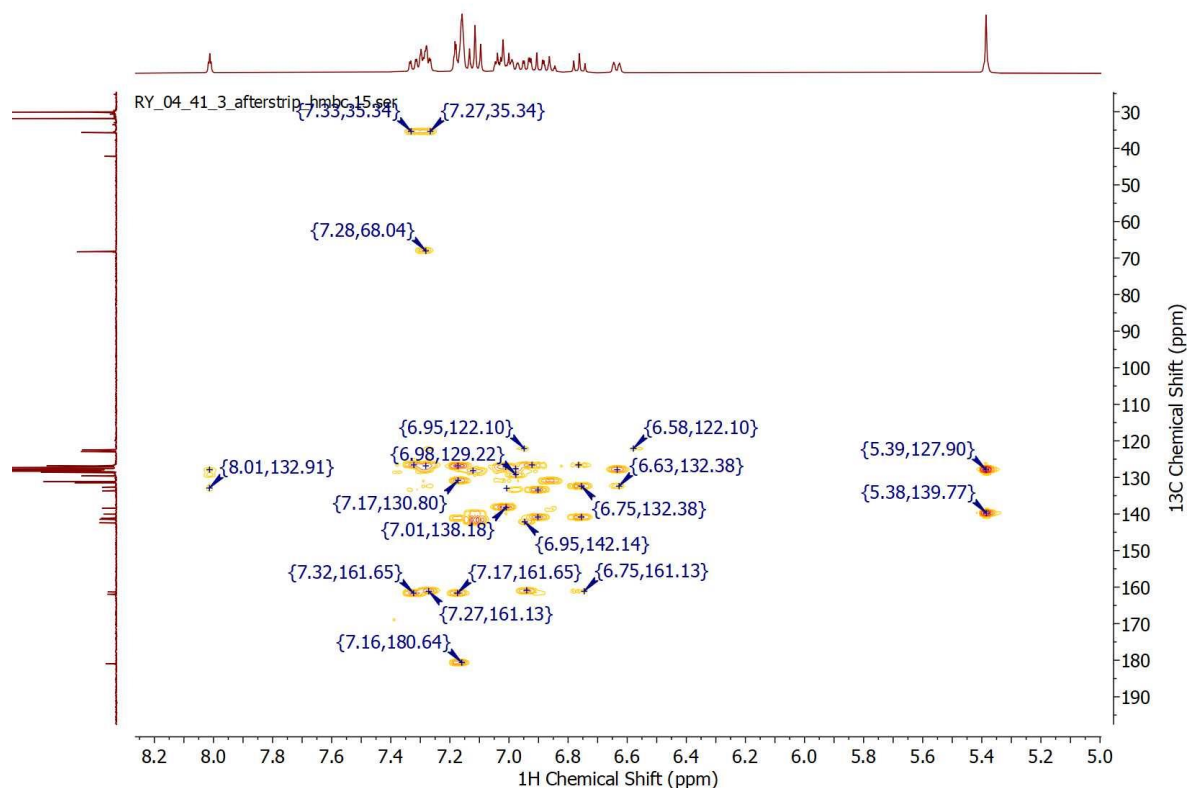


Figure S49. ^1H - ^{13}C HMBC NMR spectrum-expansion of **4-Bn** (C_6D_6 , 400 MHz, 25 °C).

4. X-ray crystallographic data

4.1. Procedure for data collection

X-Ray Intensity data were collected at 100 K on a Bruker Dual micro source D8 Venture diffractometer and PHOTON III detector running APEX3 software package of programs and using $\text{MoK}\alpha$ radiation ($\lambda = 0.71073 \text{ \AA}$). The data frames were integrated and multi-scan scaling was applied in APEX3. Intrinsic phasing structure solution provided all of the non-H atoms. The structure was refined using full-matrix least-squares refinement (SHELXL, Sheldrick G.M. 2015). The non-H atoms were refined with anisotropic displacement parameters and all of the H atoms were calculated in idealized positions and refined riding on their parent atoms. The refinement was carried out by minimizing the wR_2 function using F^2 rather than F values. R_1 is calculated to provide a reference to the conventional R value but its function is not minimized.

4.1.1. Crystallographic data of **2**

In the final cycle of refinement, 11557 reflections (of which 10619 are observed with $I > 2\sigma(I)$) were used to refine 406 parameters and the resulting R_1 , wR_2 and S (goodness of fit) were 2.20%, 5.74% and 1.317, respectively.

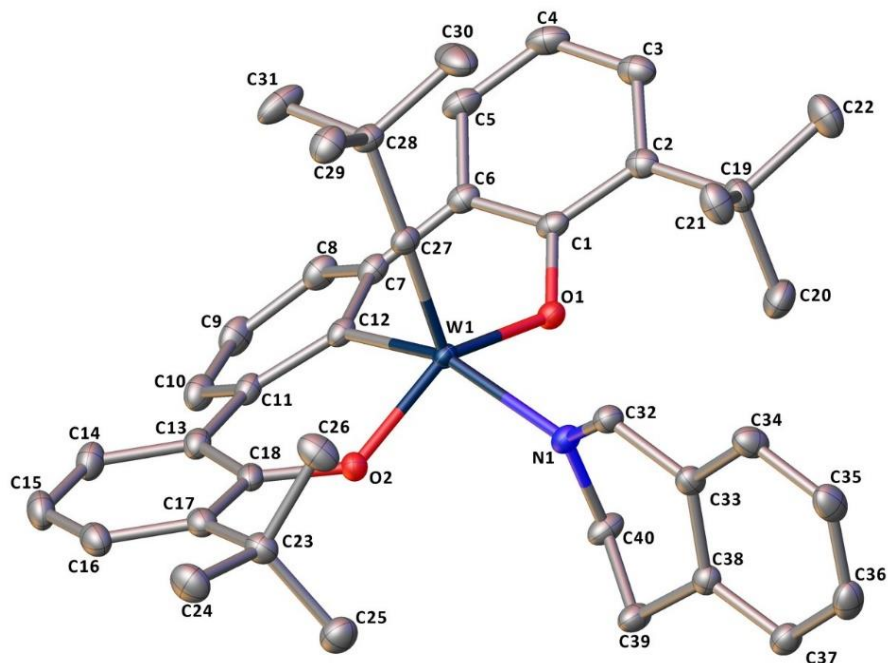


Figure S50. Solid-state molecular structure of **2**. Ellipsoids drawn at 50% probability. Hydrogen atoms are removed for clarity.

Table S1. Crystal data and structure refinement for **2**.

Identification code	rinku18	
Empirical formula	C ₄₀ H ₄₅ NO ₂ W	
Formula weight	755.62	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P2 ₁ /n	
Unit cell dimensions	a = 13.2327(4) Å	a = 90°.
	b = 18.0672(5) Å	b = 97.2920(10)°.
	c = 14.0890(4) Å	g = 90°.
Volume	3341.12(17) Å ³	
Z	4	
Density (calculated)	1.502 Mg/m ³	
Absorption coefficient	3.493 mm ⁻¹	
F(000)	1528	
Crystal size	0.279 x 0.158 x 0.143 mm ³	
Theta range for data collection	1.918 to 32.836°.	
Index ranges	-19 ≤ h ≤ 19, -27 ≤ k ≤ 26, -21 ≤ l ≤ 21	
Reflections collected	94414	
Independent reflections	11557 [R(int) = 0.0328]	

Completeness to theta = 25.242°	99.9 %
Absorption correction	multi-scan
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	11557 / 0 / 406
Goodness-of-fit on F ²	1.317
Final R indices [I>2sigma(I)]	R ₁ = 0.0220, wR ₂ = 0.0574 [10619]
R indices (all data)	R ₁ = 0.0250, wR ₂ = 0.0585
Extinction coefficient	n/a
Largest diff. peak and hole	0.844 and -1.807 e.Å ⁻³

$$R_1 = \frac{\sum(|F_o| - |F_c|)}{\sum F_o}$$

$$wR_2 = \left[\frac{\sum[w(F_o^2 - F_c^2)^2]}{\sum[w(F_o^2)^2]} \right]^{1/2}$$

$$S = \left[\frac{\sum[w(F_o^2 - F_c^2)^2]}{(n-p)} \right]^{1/2}$$

$$w = 1/[\sigma^2(F_o^2) + (m^*p)^2 + n^*p], p = [\max(F_o^2, 0) + 2 * F_c^2]/3, m \text{ \& } n \text{ are constants.}$$

4.1.2. Crystallographic data of 4-Me

The complex has two disordered regions in the methyl group C40 and the t-butyl group on C36. In the final cycle of refinement, 16702 reflections (of which 14213 are observed with $I > 2\sigma(I)$) were used to refine 385 parameters and the resulting R₁, wR₂ and S (goodness of fit) were 2.62%, 5.10 % and 1.028, respectively.

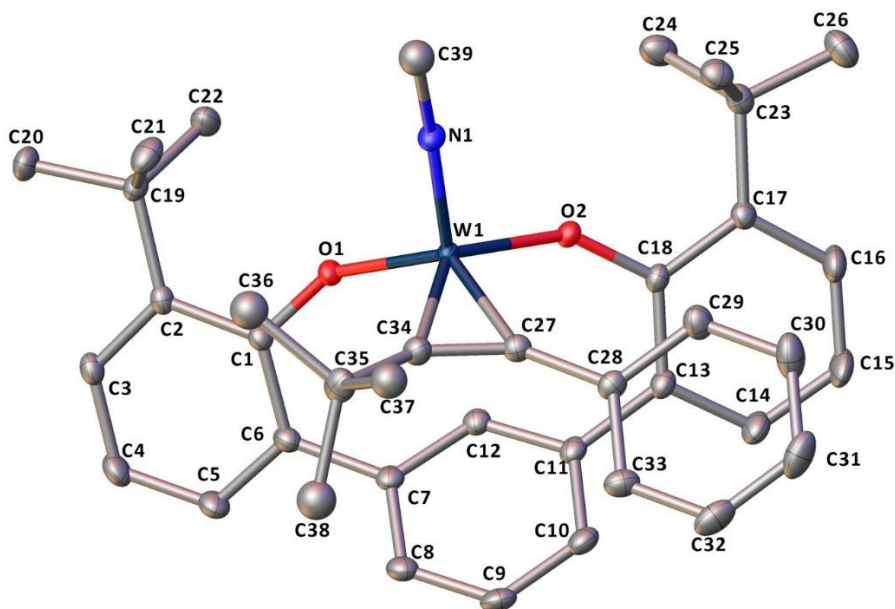


Figure S51. Solid-state molecular structure of 4-Me. Ellipsoids drawn at 50% probability. Hydrogen atoms are removed for clarity.

Table S2. Crystal data and structure refinement for **4-Me**.

Identification code	rinku2	
Empirical formula	C ₃₉ H ₄₅ NO ₂ W	
Formula weight	743.61	
Temperature	293(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	C2/c	
Unit cell dimensions	a = 23.9026(19) Å	a = 90°.
	b = 11.1201(5) Å	b = 97.854(3)°.
	c = 26.2556(14) Å	g = 90°.
Volume	6913.3(7) Å ³	
Z	8	
Density (calculated)	1.429 Mg/m ³	
Absorption coefficient	3.375 mm ⁻¹	
F(000)	3008	
Crystal size	0.152 x 0.114 x 0.024 mm ³	
Theta range for data collection	2.023 to 36.334°	
Index ranges	-39 ≤ h ≤ 39, -18 ≤ k ≤ 18, -43 ≤ l ≤ 43	
Reflections collected	157067	
Independent reflections	16702 [R(int) = 0.0635]	
Completeness to theta = 25.242°	99.4 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.9278 and 0.7247	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	16702 / 0 / 385	
Goodness-of-fit on F ²	1.028	
Final R indices [I > 2σ(I)]	R ₁ = 0.0262, wR ₂ = 0.0510 [14213]	
R indices (all data)	R ₁ = 0.0342, wR ₂ = 0.0545	
Extinction coefficient	n/a	
Largest diff. peak and hole	1.361 and -1.348 e.Å ⁻³	

$$R_1 = \frac{\sum(|F_o| - |F_c|)}{\sum F_o}$$

$$wR_2 = \left[\frac{\sum[w(F_o^2 - F_c^2)^2]}{\sum[w(F_o^2)^2]} \right]^{1/2}$$

$$S = \left[\frac{\sum[w(F_o^2 - F_c^2)^2]}{(n-p)} \right]^{1/2}$$

$$w = 1/[\sigma^2(F_o^2) + (m^*p)^2 + n^*p], p = [\max(F_o^2, 0) + 2 * F_c^2]/3, m \text{ \& } n \text{ are constants.}$$

4.1.3. Crystallographic data of 4-Ph

In the final cycle of refinement, 9050 reflections (of which 8223 are observed with $I > 2\sigma(I)$) were used to refine 442 parameters and the resulting R_1 , wR_2 and S (goodness of fit) were 2.08%, 4.52% and 1.035, respectively.

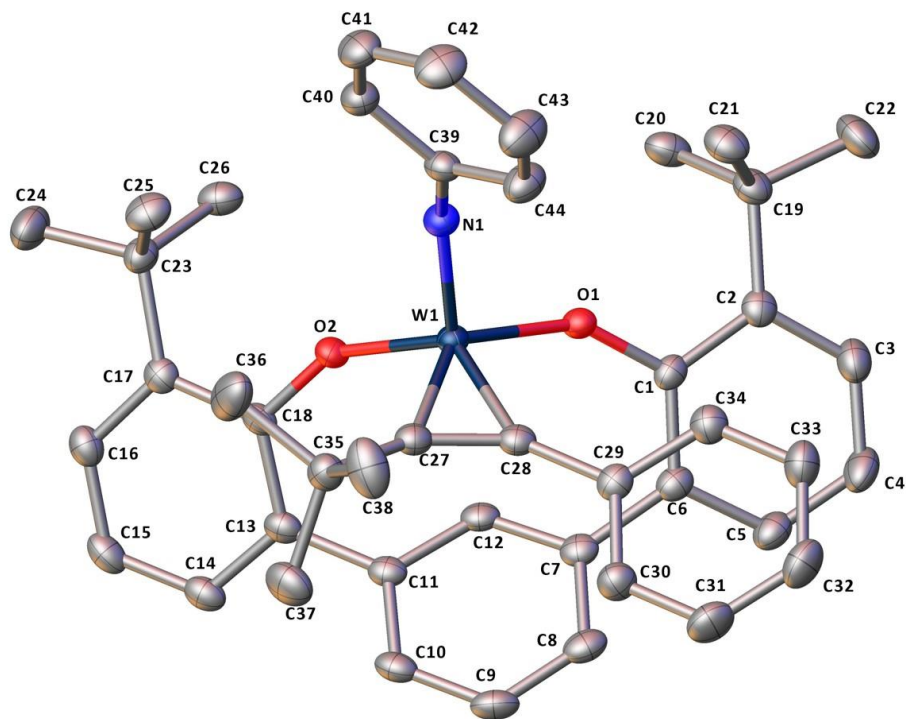


Figure S52. Solid-state molecular structure of **4-Ph**. Ellipsoids drawn at 50% probability. Hydrogen atoms are removed for clarity.

Table S3. Crystal data and structure refinement for **4-Ph**.

Identification code	rinku33	
Empirical formula	$C_{44}H_{47}NO_2W$	
Formula weight	805.67	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P-1	
Unit cell dimensions	$a = 9.3557(5)$ Å	$a = 101.459(10)^\circ$.
	$b = 11.8178(5)$ Å	$b = 91.875(10)^\circ$.
	$c = 16.8483(9)$ Å	$g = 99.739(2)^\circ$.
Volume	$1795.12(16)$ Å ³	
Z	2	
Density (calculated)	1.491 Mg/m ³	
Absorption coefficient	3.255 mm ⁻¹	

F(000)	816
Crystal size	0.650 x 0.084 x 0.075 mm ³
Theta range for data collection	1.951 to 28.538°
Index ranges	-12≤h≤12, -15≤k≤15, -22≤l≤22
Reflections collected	96677
Independent reflections	16702 [R(int) = 0.0635]
Completeness to theta = 25.242°	99.9 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.9278 and 0.7247
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	16702 / 0 / 442
Goodness-of-fit on F ²	1.035
Final R indices [I>2sigma(I)]	R ₁ = 0.0208, wR ₂ = 0.0434
R indices (all data)	R ₁ = 0.0258, wR ₂ = 0.0452
Extinction coefficient	n/a
Largest diff. peak and hole	0.875 and -0.477 e.Å ⁻³

$$R_1 = \frac{\sum(|F_o| - |F_c|)}{\sum|F_o}$$

$$wR_2 = \frac{[\sum[w(F_o^2 - F_c^2)^2]]}{[\sum[w(F_o^2)^2]]}^{1/2}$$

$$S = \frac{[\sum[w(F_o^2 - F_c^2)^2]]}{(n-p)}^{1/2}$$

$$w = 1/[\sigma^2(F_o^2) + (m \cdot p)^2 + n \cdot p], p = [\max(F_o^2, 0) + 2 \cdot F_c^2]/3, m \text{ \& } n \text{ are constants.}$$

4.1.4. Crystallographic data of 4-Bn

The complex has five disordered regions in the Ph ring on C40, C41, C42, C43, and C44. It also has a high electron density peak at 0.76 Å from tungsten arising from anisotropic modelling of heavy metal atom. In the final cycle of refinement, 9366 reflections (of which 8548 are observed with I > 2σ (I)) were used to refine 436 parameters and the resulting R₁, wR₂ and S (goodness of fit) were 3.13%, 7.18% and 1.087, respectively.

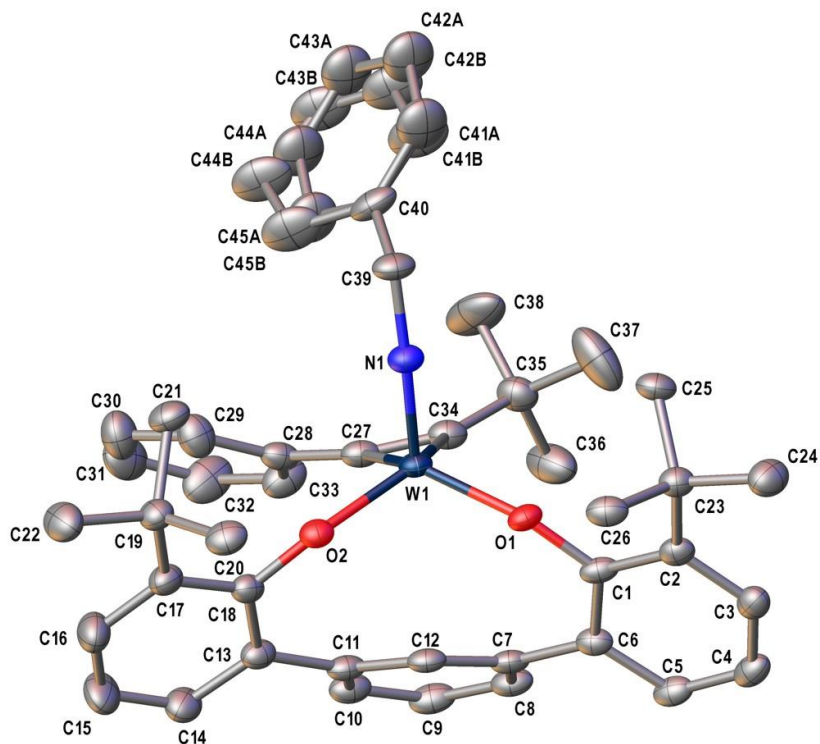


Figure S53. Solid-state molecular structure of **4-Bn**. Ellipsoids drawn at 50% probability. Hydrogen atoms are removed for clarity.

Table S4. Crystal data and structure refinement for **4-Bn**.

Identification code	rinku36	
Empirical formula	$C_{45}H_{49}NO_2W$	
Formula weight	819.70	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P-1	
Unit cell dimensions	$a = 9.425(2)$ Å	$a = 95.897(5)^\circ$.
	$b = 12.400(3)$ Å	$b = 92.628(7)^\circ$.
	$c = 16.608(3)$ Å	$g = 106.096(6)^\circ$.
Volume	$1875.8(7)$ Å ³	
Z	2	
Density (calculated)	1.451 Mg/m ³	
Absorption coefficient	3.117 mm ⁻¹	
F(000)	832	
Crystal size	0.650 x 0.084 x 0.075 mm ³	
Theta range for data collection	1.980 to 28.348°	
Index ranges	$-12 \leq h \leq 12$, $-16 \leq k \leq 16$, $-22 \leq l \leq 22$	
Reflections collected	90763	

Independent reflections	9366 [R(int) = 0.0553]
Completeness to theta = 25.242°	99.9 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	n/a
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	9366 / 4 / 436
Goodness-of-fit on F ²	1.087
Final R indices [I > 2σ(I)]	R ₁ = 0.0313, wR ₂ = 0.0718
R indices (all data)	R ₁ = 0.0360, wR ₂ = 0.0697
Extinction coefficient	n/a
Largest diff. peak and hole	4.021 and -1.546 e.Å ⁻³

$$R_1 = \Sigma(|F_o| - |F_c|) / \Sigma F_o$$

$$wR_2 = [\Sigma[w(F_o^2 - F_c^2)^2] / \Sigma[w(F_o^2)^2]]^{1/2}$$

$$S = [\Sigma[w(F_o^2 - F_c^2)^2] / (n-p)]^{1/2}$$

$$w = 1/[\sigma^2(F_o^2) + (m \cdot p)^2 + n \cdot p], p = [\max(F_o^2, 0) + 2 \cdot F_c^2] / 3, m \text{ \& } n \text{ are constants.}$$

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