

Electronic Supplementary Information

**Metal-dependent activity control of a compact-sized 8-17
DNAzyme based on metal-mediated unnatural base pairing**

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1. Experimental methods

Materials and equipment. All the natural DNA strands and 6-carboxyfluorescein (FAM)-labeled substrates containing a riboadenosine (rA) were purchased from Japan Bio Service Co., Ltd. (Saitama, Japan) at HPLC purification grade. A DNAzyme strand (**DzIm**) containing a 4-carboxyimidazole (**Im^C**) nucleotides was synthesized on an automated DNA synthesizer and purified by reverse-phase HPLC according to the reported procedure.^[S1] The **DzIm** strand was characterized by matrix assisted laser desorption ionization time-of-flight (MALDI-TOF) mass spectrometry on Autoflex III (Bruker) using a mixture of 3-hydroxypicolinic acid (3-HPA) and ammonium citrate as a matrix. Metal sources (CuSO₄·5H₂O (99.5% purity), Hg(ClO₄)₂·3H₂O (99%), AgNO₃ (99.8%), Pb(OAc)₂·3H₂O (99.9%)) were purchased from FUJIFILM Wako Pure Chemical Industries. Denaturing polyacrylamide gel electrophoresis (PAGE) was carried out with 20% polyacrylamide gel containing 7 M urea. The gels were imaged and analyzed using Gel Doc EZ Imager and Image Lab software (Bio-Rad).

RNA-cleaving reaction by Im^C-modified DNAzyme (DzIm). DNAzyme strands were annealed (85 °C → 25 °C, 1.0 °C/min) in a reaction buffer (50 mM HEPES buffer (pH 7.5), 100 mM NaCl, 10 μM Pb(OAc)₂) in the absence or presence of CuSO₄ (1.0 equiv). The concentrations of Pb^{II} ions were set to 10 μM to reduce unintended interactions between Pb^{II} and **Im^C** nucleobases. The DNAzyme reactions were initiated by the addition of a FAM-labeled substrate strand. The final concentration of each component was as follows: [DNAzyme] = 1.0 μM, [substrate] = 10 μM, [CuSO₄] = 0 or 1.0 μM. After incubation at 25 °C, an aliquot of the sample solution was taken at the designated time points. The reaction was stopped by the addition of a loading buffer (5.25 M urea, 1.25 mM EDTA, 7.5% glycerol), and the mixtures were stored at -28 °C until the time course was completed. The cleavage of the substrate was analyzed by denaturing PAGE. The fractions of the cleaved substrate (*F*) were calculated as follows:

$$F (\%) = I_c / (I_c + I_u) \times 100,$$

where *I_c* and *I_u* are the band intensities of the cleaved product and the uncleaved substrate, respectively. The apparent first-order rate constants (*k_{obs}*) were calculated from the initial reaction rates, which were determined from the time points when *F* was less than 20%.

RNA-cleaving reactions by DNAzymes DzT and DzC. DNAzyme strands were annealed (85 °C → 25 °C, 1.0 °C/min) in a reaction buffer (50 mM HEPES buffer (pH 7.5), 100 mM NaNO₃, 10 μM Pb(OAc)₂) in the absence or presence of Hg(ClO₄)₂ or AgNO₃ (1.0 equiv). Note that NaCl

was replaced with NaNO_3 in the standard reaction buffer to prevent the precipitation of HgCl_2 and AgCl . The following procedures are the same as that for **DzIm**. The final concentration of each component was as follows: $[\text{DNAzyme}] = 1.0 \mu\text{M}$, $[\text{substrate}] = 10 \mu\text{M}$, $[\text{Hg}(\text{ClO}_4)_2] = [\text{AgNO}_3] = 0$ or $1.0 \mu\text{M}$.

Cu^{II} titration experiment with DzIm DNAzyme. The overall procedures are the same as the DNAzyme reactions described above. The final concentration of each component was as follows: $[\text{DNAzyme}] = 1.0 \mu\text{M}$, $[\text{substrate}] = 10 \mu\text{M}$, $[\text{CuSO}_4] = 0, 0.5, 1.0, 2.0,$ or $3.0 \mu\text{M}$.

CD spectroscopy. The samples were prepared by combining an equimolar amount of the DNAzyme (8-17 DNAzyme or **DzIm**) and an uncleavable DNA substrate that contains a dA nucleotide instead of the scissile rA, and annealed prior to the measurements ($85 \text{ }^\circ\text{C} \rightarrow 4 \text{ }^\circ\text{C}$, $1.0 \text{ }^\circ\text{C}/\text{min}$). The concentrations of each component were as follows: $[\text{DNAzyme}] = 1.0 \mu\text{M}$, $[\text{substrate}] = 1.0 \mu\text{M}$, $[\text{CuSO}_4] = 0$ or $3.0 \mu\text{M}$. The CD spectra were recorded on a JASCO J-820 spectropolarimeter with 10-time accumulation using a path length of 0.5 cm at $4 \text{ }^\circ\text{C}$. The spectra were smoothed using a simple moving average smoothing program.

2. Supporting figures

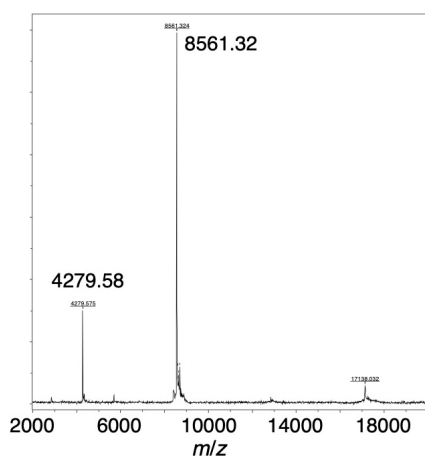


Figure S1. MALDI-TOF mass spectrum of the synthesized Im^{C} -modified 8-17 DNAzyme (**DzIm**). Negative mode. Observed: 8561.32; calcd for $[\text{M} - \text{H}]^-$: 8560.33.

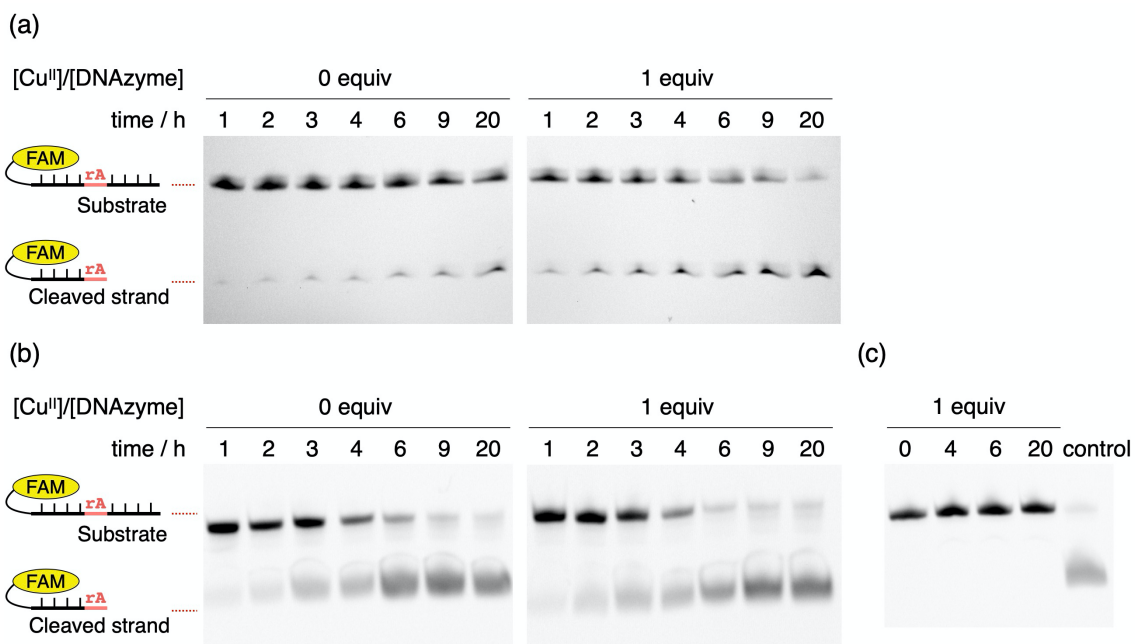


Figure S2. Denaturing polyacrylamide gel electrophoresis (PAGE) of the RNA-cleaving reactions (a) catalyzed by the **Im^C**-modified 8-17 DNAzyme (**DzIm**), (b) catalyzed by the original 8-17 DNAzyme, and (c) without DNAzymes. [DNAzyme] = 1.0 μ M, [substrate] = 10 μ M in 50 mM HEPES (pH 7.5), 100 mM NaCl, 0.01 mM Pb(OAc)₂, 25 °C. 20% polyacrylamide gel containing 7 M urea, detected by FAM fluorescence. The results are summarized in Fig. 3 in the main text.

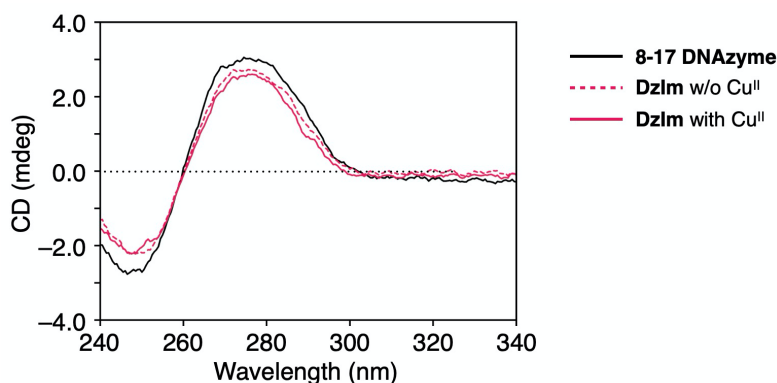


Figure S3. CD spectra of 8-17 DNAzyme and **DzIm** in the presence of an uncleavable DNA substrate containing a dA instead of the scissile rA. [DNAzyme] = 1.0 μ M, [DNA substrate] = 1.0 μ M, [Cu^{II}] = 0 or 1.0 μ M in 50 mM HEPES buffer (pH 7.5), 100 mM NaCl, 0.01 mM Pb(OAc)₂, *l* = 0.5 cm, 4 °C.

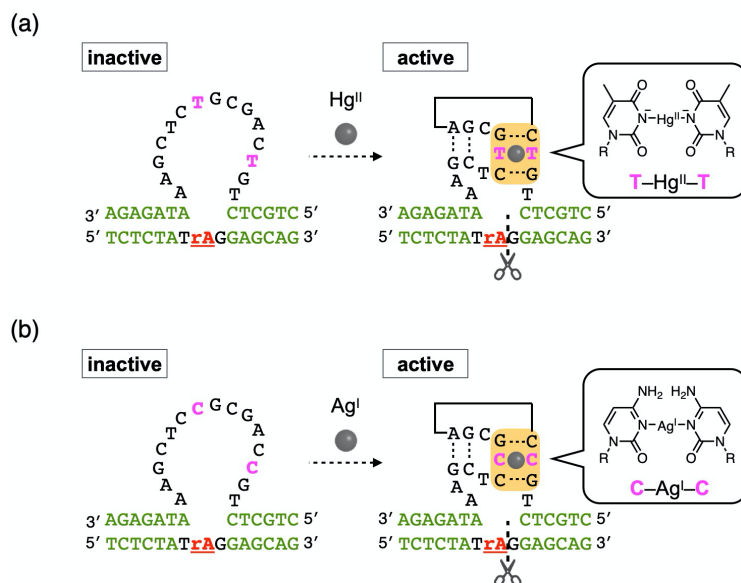


Figure S4. Molecular design of Hg^{II} - and Ag^{I} -responsive 8-17 DNazymes containing a $\text{T-Hg}^{\text{II}}\text{-T}$ or a $\text{C-Ag}^{\text{I}}\text{-C}$ base pair (**DzT** and **TzC**, respectively). **rA** = adenine ribonucleotide at the cleavage site. Plausible secondary structures for both active and inactive states are shown.

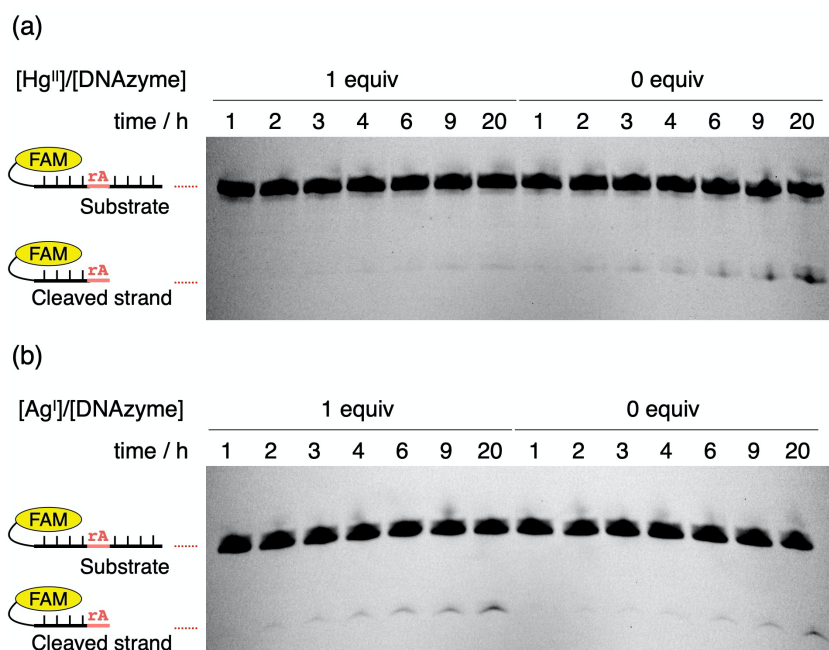


Figure S5. Denaturing polyacrylamide gel electrophoresis (PAGE) of the RNA-cleaving reactions catalyzed by the DNazymes modified (a) with a T-T base pair (**DzT**) and (b) with a C-C base pair (**DzC**). [DNAzyme] = 1.0 μM , [substrate] = 10 μM in 50 mM HEPES (pH 7.5), 100 mM NaNO_3 , 0.01 mM $\text{Pb}(\text{OAc})_2$, 25 $^\circ\text{C}$. 20% polyacrylamide gel containing 7 M urea, detected by FAM fluorescence. The results are summarized in Fig. 4b in the main text.

3. Reference

S1. Y. Takezawa, L. Hu, T. Nakama and M. Shionoya, *Angew. Chem., Int. Ed.*, 2020, **59**, 21488–21492.