# Micro-flow heteroatom alkylation via TfOH-mediated rapid in situ generation of carbocations and subsequent nucleophile addition 

Yuma Matsuura, Shinichiro Fuse*

Affiliations: Department of Basic Medicinal Sciences, Graduate School of Pharmaceutical Sciences, Nagoya University, Nagoya 464-8601, Japan.<br>E-mail: fuse.shinichiro.z3@f.mail.nagoya-u.ac.jp

## 6 February 2024

Note added after publication: This file replaces the supporting information originally published on 29th January 2024, as there was an error within the mass spectrometry data for compound 2s on page S19. This does not affect the results and conclusions of the paper.

## Table of Contents

1. General techniques ..... S-2
2. Micro-flow reactor setup ..... S-2
3. General procedure for the synthesis of alcohols ..... S-5
4. Optimization of reaction conditions ..... S-7
5. Typical procedure for a micro-flow nucleophilic substitution ..... S-11
6. References ..... S-20
7. NMR spectra ..... S-21

## 1. General techniques

NMR spectra were recorded on a JEOL-ECS400 ( 400 MHz for ${ }^{1} \mathrm{H}, 100 \mathrm{MHz}$ for ${ }^{13} \mathrm{C}$ ) or JEOLECZ400 ( 400 MHz for ${ }^{1} \mathrm{H}, 100 \mathrm{MHz}$ for ${ }^{13} \mathrm{C}$ ) instrument in the indicated solvent. Chemical shifts were reported in units of parts per million ( ppm ) relative to tetramethylsilane ( 0.000 ppm ) in $\mathrm{CDCl}_{3}$ for ${ }^{1} \mathrm{H}$ NMR and $\mathrm{CDCl}_{3}(77.16 \mathrm{ppm})$ for ${ }^{13} \mathrm{C}$ NMR. Multiplicities were reported by using the following abbreviations: s ; singlet, d; doublet, dd; double doublet, ddd; double double doublet, t ; triplet, tt ; triple triplet, q ; quartet, sep; septet, m ; multiplet, br; broad $J$; coupling constants in Hertz (Hz). IR spectra were recorded on a JASCO FT/IR-4100 Fourier Transform Infrared Spectrophotometer. Only the strongest and/or structurally important peaks were reported as the IR data are given in $\mathrm{cm}^{-1}$. High-resolution mass spectra (HRMS) were obtained on a Bruker Daltonics Compact in the electrospray ionization-time of light (ESI-TOF) method. Gel permeation chromatography (GPC) for purification was performed on Japan Analytical Industry Model LaboACE LC-5060 (recycling preparative HPLC) on a Japan Analytical Industry Model UV-2564 LA ultraviolet detector and RI-700 LA refractive index detector with a polystyrene gel column (JAIGEL-2HR, $20 \mathrm{~mm} \times 600 \mathrm{~mm}$ ), using chloroform as a solvent $(10 \mathrm{~mL} / \mathrm{min})$. Column chromatography was performed on Silica Gel PSQ 60B purchased from Fuji Silysia Chemical LTD. Reactions were monitored by thin-layer chromatography carried out on 0.25 mm E. Merck silica gel plates ( $60 \mathrm{~F}-254$ ) with UV light, visualized by $p$-anisaldehyde, ceric sulfate solution, $10 \%$ ethanolic phosphomolybdic acid. THF was dried by a Glass Contour Solvent dispensing system (Nikko Hansen \& Co., Ltd.). $\mathrm{CH}_{3} \mathrm{CN}$ was dried by molecular sieves 3A. Other solvents and reagents were purchased from commercial suppliers (FUJIFILM Wako Pure Chemical, Kanto Chemical, Sigma-Aldrich, and Tokyo Chemical Industry) and used without further purification.

## 2. Micro-flow reactor setup

Stainless steel T-shape (inner diameter: $0.250 \mathrm{~mm}, 0.500 \mathrm{~mm}$, or 1.00 mm ) and V-shape mixers (inner diameter: 0.250 mm ) were purchased from Sanko Seiki Co. Ltd. The front and side view of the T-shape and V-shape mixer is shown in Figure S-1. Teflon ${ }^{\circledR}$ tubes (inner diameter: 0.800 or 0.500 mm were purchased from Senshu Scientific Co., Ltd. PEEK fittings, PEEK unions, stainless steel tubes, stainless steel fittings, and stainless steel unions (inner diameter: 0.800 mm ) were purchased from GL Science Inc. Solutions were introduced to a micro-flow system with syringe pumps (Harvard PHD ULTRA) equipped gastight syringes (SGE 10 mL or 50 mL ). The gastight syringes and the Teflon tubes were connected with joints purchased from Flon Industry Co., Ltd.


Figure S-1. T- and V-shape mixers used in this study.

The employed micro-flow system is shown in Figure S-2. The gastight syringes and the 1st and 2nd V-shape mixers were connected with the Teflon tubes and stainless-steel tubes (for controlling the temperature of solutions). The 1st and 2nd V-shape mixers were connected with reaction tube 1 (Teflon tube). The 2 nd $V$-shape mixer was connected with the reaction tube 2 (Teflon tube). These mixers and reaction tubes were immersed in a water bath.


Figure S-2. Micro-flow reactor setup

## 3. General procedure for the synthesis of alcohols ${ }^{[\mathrm{S} 1]}$



A suspension of Mg turnings ( $11.3 \mathrm{mmol}, 1.56$ equiv.) and several pieces of iodine in dry THF $(32 \mathrm{~mL})$ were stirred at room temperature for 10 min . To the resultant mixture, 4-bromoanisole ( $9.00 \mathrm{mmol}, 1.25$ equiv.) was added slowly at room temperature, and the resultant mixture was stirred at $40^{\circ} \mathrm{C}$ for 1 h . After cooling to $0^{\circ} \mathrm{C}$, aldehyde $\mathbf{S} 1$ ( $7.20 \mathrm{mmol}, 1.00$ equiv.) was added slowly to the reaction mixture, and it was stirred at room temperature for 12 h . The reaction was quenched by addition of sat. $\mathrm{NH}_{4} \mathrm{Cl}$ aq. and extracted with EtOAc twice. The combined organic extracts were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated in vacuo.

## $\alpha$-(4-Methoxyphenyl)-1,3-benzodioxole-5-methanol (1b)

According to the general procedure for the synthesis of alcohols using 3,4methylenedioxybenzaldehyde S1a ( $1.08 \mathrm{~g}, 7.20 \mathrm{mmol}, 1.00$ equiv.), the crude product was purified by column chromatography on silica gel (hexane/EtOAc $=10: 1$ to $5: 1$ ) to give $\alpha$-(4-methoxyphenyl)-1,3-benzodioxole-5-methanol (1b) (1.17 g, $4.53 \mathrm{mmol}, 63 \%$ ).

colorless oil; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.30-7.27(\mathrm{~m}, 2 \mathrm{H}), 6.89-6.83(\mathrm{~m}, 4 \mathrm{H}), 6.76(\mathrm{dd}, J=$ $2.8,8.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.93(\mathrm{~s}, 2 \mathrm{H}), 5.73(\mathrm{~d}, J=3.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 2.08(\mathrm{~d}, J=3.2 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 159.2,147.9,147.0,138.4,136.3,127.8,120.0,114.0,108.2,107.2$, 101.1, 75.7, 55.4.

Spectral data of ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR were well consistent with those reported in previous literature ${ }^{[\mathrm{S} 2]}$.

## (4-Methoxyphenyl)(3,4,5-trimethoxyphenyl)methanol (1c)

According to the general procedure for the synthesis of alcohols using 3,4,5trimethoxybenzaldehyde $\mathbf{S 1 b}$ ( $1.77 \mathrm{~g}, 7.20 \mathrm{mmol}, 1.00$ equiv.), the crude product was purified by recrystallization to give (4-methoxyphenyl)(3,4,5-trimethoxyphenyl)methanol (1c) (1.27 g, 5.10 mmol, 71\%).

white solid; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.30-7.28(\mathrm{~m}, 2 \mathrm{H}), 6.89-6.87(\mathrm{~m}, 2 \mathrm{H}), 6.60(\mathrm{~s}, 2 \mathrm{H})$, $5.73(\mathrm{~d}, J=3.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.83(\mathrm{~s}, 9 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 2.23(\mathrm{~d}, J=3.6 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 159.3,153.4,139.8,137.2,136.0,128.0,114.0,103.5,76.0,61.0,56.2,55.4$.

Spectral data of ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR were well consistent with those reported in previous literature ${ }^{[\mathrm{S} 3]}$.

## 4. Optimization of reaction conditions

### 4.1. Examination of Bronsted acids, bases, solvents, and reaction times in micro-flow nucleophilic substitution

A solution of 4,4'-dimethoxybenzhydrol (1a) (C M, 1.00 equiv.) in solvent (flow rate: 2.40 $\mathrm{mL} / \mathrm{min}$ ) and a solution of Brønsted acid ( $0.188 \mathrm{M}, 1.50$ equiv.) in $\mathrm{CH}_{3} \mathrm{CN}$ (flow rate: 4.80 $\mathrm{mL} / \mathrm{min}$ ) were introduced to 1 st V -shape mixer at $\mathbf{D}^{\circ} \mathrm{C}$ with syringe pumps. The resultant mixture was passed through reaction tube 1 (inner diameter: 0.500 mm (entries 1-14, 16-24), or 0.800 mm (entry 15), reaction time: A s) at $\mathbf{D}{ }^{\circ} \mathrm{C}$. The resultant mixture and a solution of piperidine ( 0.0625 $\mathrm{M}, 1.00$ equiv.) and base ( $0.0625 \mathrm{M}, 1.00$ equiv.) in solvent (flow rate: $9.60 \mathrm{~mL} / \mathrm{min}$ ) were introduced to the 2 nd V -shape mixer at $\mathbf{D}{ }^{\circ} \mathrm{C}$ with syringe pumps. The resultant mixture was passed through reaction tube 2 (inner diameter: 0.500 mm (entries 1-16, 18-24), or 0.800 mm (entry 17), reaction time: Bs) at $\mathbf{D}{ }^{\circ} \mathrm{C}$. After being eluted for $c a .25 \mathrm{~s}$ to reach a steady state, the resultant mixture was poured into a test tube containing sat. $\mathrm{NaHCO}_{3}$ aq. ( 5 mL ) and EtOAc (5 mL ) for 30 s at room temperature. The reaction mixture was extracted with EtOAc ( 5 mL ) twice. The organic layer was washed with brine $(10 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated in vacuo. Yields were determined by ${ }^{1} \mathrm{H}$ NMR analysis using dimethyl sulfone as an internal standard.


Table S-1

|  |  |  |  |  | Bronsted |  |  | yield [\%] ${ }^{\text {c }}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| entry | $\begin{gathered} \mathbf{A} \\ {[\mathrm{s}]} \end{gathered}$ | $\begin{gathered} \mathbf{B} \\ {[\mathrm{s}]} \end{gathered}$ | [M] | $\begin{gathered} \mathbf{D} \\ {\left[{ }^{\circ} \mathrm{C}\right]} \end{gathered}$ | $\begin{gathered} \text { acid } \\ \left(\mathrm{p} K_{\mathrm{a}}{ }^{a}\right) \\ \hline \end{gathered}$ | $\begin{gathered} \text { base } \\ \left(\mathrm{p} K_{\mathrm{a}} \mathrm{H}^{b}\right) \end{gathered}$ | solvent | 2 a | 1 a | 3a |
| 1 | 0.10 | 1.0 | 0.250 | 25 | $\begin{gathered} \text { TFA } \\ (12.7)^{\mathrm{S4}} \end{gathered}$ | $\begin{gathered} \text { DBU } \\ (24.3)^{\mathrm{S7}} \end{gathered}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | $<1$ | 93 | 2 |


| 2 | 0.10 | 1.0 | 0.250 | 25 | $\begin{gathered} \text { TCA } \\ (10.6)^{\text {S4 }} \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 4 | 92 | 2 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | 0.10 | 1.0 | 0.250 | 25 | $\begin{gathered} \mathrm{HC} f^{f} \\ (10.3)^{\mathrm{SS}} \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 44 | 48 | 2 |
| 4 | 0.10 | 1.0 | 0.250 | 25 | $\begin{aligned} & \mathrm{H}_{2} \mathrm{SO}_{4} \\ & (8.7)^{\mathrm{S} 6} \end{aligned}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 5 | 32 | 56 |
| 5 | 0.10 | 1.0 | 0.250 | 25 | $\begin{gathered} \mathrm{HBr}^{\mathrm{g}} \\ (5.5)^{\mathrm{S5}} \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 70 | 22 | 2 |
| 6 | 0.10 | 1.0 | 0.250 | 25 | $\begin{gathered} \mathrm{HI}^{h} \\ (2.8)^{\mathrm{S5}} \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 74 | 13 | 2 |
| $7^{d}$ | 0.10 | 1.0 | 0.250 | 25 | $\begin{aligned} & \text { TfOH } \\ & (0.7)^{\mathrm{S} 5} \end{aligned}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 85 | 7 | 4 |
| $8^{\text {d }}$ | 0.10 | 1.0 | 0.250 | 25 | $\begin{gathered} \text { TfOH } \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { TMG } \\ & (23.4)^{\mathrm{s} 7} \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 84 | 7 | 3 |
| $9^{d}$ | 0.10 | 1.0 | 0.250 | 25 | $\begin{gathered} \text { TfOH } \\ (0.7) \end{gathered}$ | $\begin{gathered} i-\mathrm{Pr}_{2} \mathrm{NEt} \\ (18.1)^{\mathrm{s} 8} \end{gathered}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 81 | 8 | 11 |
| $10^{d}$ | 0.10 | 1.0 | 0.250 | 25 | $\begin{gathered} \text { TfOH } \\ (0.7) \end{gathered}$ | $\mathrm{NMe}_{3}$ $(17.6)^{\mathrm{S} 7}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | $<1$ | 21 | 19 |
| $11^{d}$ | 0.10 | 1.0 | 0.250 | 25 | $\begin{gathered} \mathrm{TfOH} \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 85 | 7 | 4 |
| $12^{\text {d, } e}$ | 0.50 | 1.0 | 0.250 | 25 | $\begin{gathered} \text { TfOH } \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ |  | 6 | $3 \pm$ 1 |
| $13^{d}$ | 1.0 | 1.0 | 0.250 | 25 | $\begin{gathered} \text { TfOH } \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 84 | 6 | 3 |
| $14^{d}$ | 5.0 | 1.0 | 0.250 | 25 | $\begin{gathered} \text { TfOH } \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 81 | 5 | 1 |
| $15^{d}$ | 10 | 1.0 | 0.250 | 25 | $\begin{gathered} \text { TfOH } \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 77 | 5 | 1 |
| $16^{d}$ | 0.50 | 5.0 | 0.250 | 25 | $\begin{gathered} \text { TfOH } \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 90 | 6 | 4 |
| $17^{d}$ | 0.50 | 10 | 0.250 | 25 | $\begin{gathered} \mathrm{TfOH} \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 91 | 7 | 4 |
| $18^{d}$ | 0.5 | 1.0 | 0.125 | 25 | $\begin{gathered} \text { TfOH } \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 88 | 8 | 4 |
| $19^{d}$ | 0.5 | 1.0 | 0.500 | 25 | $\begin{gathered} \mathrm{TfOH} \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 77 | 11 | 9 |
| $20^{d}$ | 0.5 | 1.0 | 0.250 | 0 | $\begin{gathered} \text { TfOH } \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 80 | 5 | 6 |
| $21^{d}$ | 0.5 | 1.0 | 0.250 | 40 | $\begin{gathered} \mathrm{TfOH} \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | 81 | 8 | 2 |
| $22^{d}$ | 0.5 | 1.0 | 0.250 | 25 | $\begin{gathered} \mathrm{TfOH} \\ (0.7) \end{gathered}$ | $\begin{aligned} & \text { DBU } \\ & (24.3) \end{aligned}$ | acetone | 9 | 10 | 73 |


| $23^{d}$ | 0.5 | 1.0 | 0.250 | 25 | TfOH <br> $(0.7)$ | DBU <br> $(24.3)$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 28 | 14 | 55 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $24^{d}$ | 0.5 | 1.0 | 0.250 | 25 | TfOH <br> $(0.7)$ | DBU <br> $(24.3)$ | $\mathrm{THF}^{n}$ | 2 | 4 | 84 |

${ }^{a}$ The $\mathrm{p} K_{\mathrm{a}}$ in $\mathrm{CH}_{3} \mathrm{CN}$. ${ }^{b}$ The $\mathrm{p} K_{\mathrm{a}}$ of conjugated acids in $\mathrm{CH}_{3} \mathrm{CN}$. ${ }^{C}$ Yields were determined by ${ }^{1} \mathrm{H}$ NMR analysis using dimethyl sulfone as an internal standard. ${ }^{d}$ A trace amounts of 4,4'dimethoxybenzophenone (6a) and 4,4'-dimethoxydiphenylmethane (7a) were generated. ${ }^{e}$ Three independent experiments were performed. ${ }^{f}$ Aqueous solution of $\mathrm{HCl}(36 \mathrm{w} / \mathrm{w} \%)$ was used. ${ }^{8}$ Aqueous solution of $\mathrm{HBr}(47 \mathrm{w} / \mathrm{w} \%)$ was used. ${ }^{h}$ Aqueous solution of $\mathrm{HI}(57 \mathrm{w} / \mathrm{w} \%)$ was used. TFA $=$ Trifluoroacetic acid. TCA $=$ Trichloroacetic acid. TfOH $=$ trifluoromethanesulfonic acid. DBU $=1,8$-diazabicyclo[5.4.0]undec-7-ene. $\mathrm{TMG}=1,1,3,3$-Tetramethylguanidine. $\mathrm{THF}=$ Tetrahydrofuran.

### 4.2. Examination of 1st mixer and 2nd mixer for micro-flow nucleophilic substitution

A solution of 4,4'-dimethoxybenzhydrol (1a) ( $0.250 \mathrm{M}, 1.00$ equiv.) in $\mathrm{CH}_{3} \mathrm{CN}$ (flow rate: 2.40 $\mathrm{mL} / \mathrm{min}$ ) and a solution of $\mathrm{TfOH}\left(0.188 \mathrm{M}, 1.50\right.$ equiv.) in $\mathrm{CH}_{3} \mathrm{CN}$ (flow rate: $4.80 \mathrm{~mL} / \mathrm{min}$ ) were introduced to 1st mixer at $25^{\circ} \mathrm{C}$ with syringe pumps. The resultant mixture was passed through reaction tube 1 (inner diameter: 0.500 mm , length: 306 mm , volume: $60 \mu \mathrm{~L}$, reaction time: 0.50 s) at $25^{\circ} \mathrm{C}$. The resultant mixture and a solution of piperidine $(0.0625 \mathrm{M}, 1.00$ equiv.) and DBU ( $0.0625 \mathrm{M}, 1.00$ equiv.) in $\mathrm{CH}_{3} \mathrm{CN}$ (flow rate: $9.60 \mathrm{~mL} / \mathrm{min}$ ) were introduced to the 2nd mixer at $25{ }^{\circ} \mathrm{C}$ with syringe pumps. The resultant mixture was passed through reaction tube 2 (inner diameter: 0.500 mm , length: $1,427 \mathrm{~mm}$, volume: $280 \mu \mathrm{~L}$, reaction time: 1.0 s ) at $25^{\circ} \mathrm{C}$. After being eluted for $c a .25$ s to reach a steady state, the resultant mixture was poured into a test tube containing sat. $\mathrm{NaHCO}_{3}$ aq. $(5 \mathrm{~mL})$ and $\operatorname{EtOAc}(5 \mathrm{~mL})$ for 30 s at room temperature. The reaction mixture was extracted with EtOAc ( 5 mL ) twice. The organic layer was washed with brine ( 10 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated in vacuo. Yields were determined by ${ }^{1} \mathrm{H}$ NMR analysis using dimethyl sulfone as an internal standard.

According to a report, the mixing efficiency of the mixer in the microflow synthesis is superior with a V-shape compared to a T-shape ${ }^{59}$, and a smaller inner diameter is also preferable ${ }^{\text {S10 }}$. Our examination of the shape of a mixer and the inner diameter revealed that as the mixing efficiency decreased, the yield of the target product 2 a also decreased (entries 1-5).


Table S-2

| entry | $\underset{\text { mixer }}{\text { 1st }}$ | inner <br> diameter (mm) | 2nd mixer | inner diameter (mm) | NMR yield [\%] |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  | $\begin{gathered} \mathbf{2 a} \\ \text { (desired) } \end{gathered}$ | $\begin{gathered} \mathbf{1 a} \\ \text { (sub) } \end{gathered}$ | $\begin{gathered} \mathbf{3 a} \\ \text { (undesired) } \end{gathered}$ |
| 1 | V shape | 0.25 | V shape | 0.25 | 91 | 6 | 3 |
| 2 | $V$ shape | 0.25 | T shape | 0.25 | 81 | 9 | 4 |
| 3 | V shape | 0.25 | T shape | 0.50 | 76 | 19 | 5 |
| 4 | V shape | 0.25 | T shape | 1.0 | 61 | 25 | 13 |
| 5 | T shape | 1.0 | V shape | 0.25 | 83 | 6 | 3 |

### 4.3. Procedure for synthesis of 2a using a batch reactor

(Quantities of compounds, solvents, and temperature were identical to those of flow condition.) To a vigorously stirred (magnetic stirrer, $1,000 \mathrm{rpm}$.) solution of 4,4'-dimethoxybenzhydrol (1a) ( $0.250 \mathrm{M}, 1.00$ equiv.) in $\mathrm{CH}_{3} \mathrm{CN}\left(1.20 \mathrm{~mL}\right.$ ), a solution of TfOH ( $0.188 \mathrm{M}, 1.50$ equiv.) in $\mathrm{CH}_{3} \mathrm{CN}$ $(2.40 \mathrm{~mL})$ was added to the reaction mixture in one portion at $25^{\circ} \mathrm{C}$ under an argon atmosphere. After being stirred for 10 s at $25^{\circ} \mathrm{C}$, the resultant mixture was added to the reaction mixture in one portion at $25^{\circ} \mathrm{C}$ to a vigorously stirred (magnetic stirrer, $1,000 \mathrm{rpm}$.) solution of piperidine ( $0.0625 \mathrm{M}, 1.00$ equiv.) and DBU ( $0.0625 \mathrm{M}, 1.00$ equiv.) in $\mathrm{CH}_{3} \mathrm{CN}(4.80 \mathrm{~mL})$. After being stirred for 10 s at $25^{\circ} \mathrm{C}$, sat. $\mathrm{NaHCO}_{3}$ aq. $(5 \mathrm{~mL})$ and $\operatorname{EtOAc}(5 \mathrm{~mL})$ was added to the reaction mixture in one portion at $25^{\circ} \mathrm{C}$ (Under the flow condition, activation and nucleophilic substitution were carried out at 0.50 s and 1.0 s respectively. However, under batch conditions, it was impossible to operate the reaction within 10 s . Thus, the reaction time was extended to 10 s .). The reaction mixture was extracted with EtOAc ( 5 mL ) twice. The organic layer was washed with brine ( 10 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated in vacuo. Yields were determined
by ${ }^{1} \mathrm{H}$ NMR analysis using dimethyl sulfone as an internal standard. Three independent experiments were performed.

## 5. Typical procedure for a micro-flow nucleophilic substitution



A solution of alcohol 1 ( $0.250 \mathrm{M}, 1.00$ equiv.) in $\mathrm{CH}_{3} \mathrm{CN}$ (flow rate: $2.40 \mathrm{~mL} / \mathrm{min}$ ) and a solution of TfOH ( $0.188 \mathrm{M}, 1.50$ equiv.) in $\mathrm{CH}_{3} \mathrm{CN}$ (flow rate: $4.80 \mathrm{~mL} / \mathrm{min}$ ) were introduced to 1st V-shape mixer at $25^{\circ} \mathrm{C}$ with syringe pumps. The resultant mixture was passed through reaction tube 1 (inner diameter: 0.500 mm , length: 306 mm , volume: $60 \mu \mathrm{~L}$, reaction time: 0.50 s) at $25^{\circ} \mathrm{C}$. The resultant mixture and a solution of nucleophile ( $0.0625 \mathrm{M}, 1.00$ equiv.) and DBU ( $0.0625 \mathrm{M}, 1.00$ equiv.) in $\mathrm{CH}_{3} \mathrm{CN}$ (flow rate: $9.60 \mathrm{~mL} / \mathrm{min}$ ) were introduced to the 2nd V-shape mixer at $25^{\circ} \mathrm{C}$ with syringe pumps. The resultant mixture was passed through reaction tube 2 (inner diameter: 0.500 mm , length: $1,427 \mathrm{~mm}$, volume: $280 \mu \mathrm{~L}$, reaction time: 1.0 s ) at $25^{\circ} \mathrm{C}$. After being eluted for $c a .25 \mathrm{~s}$ to reach a steady state, the resultant mixture was poured into a test tube containing sat. $\mathrm{NaHCO}_{3}$ aq. ( 5 mL ) and $\mathrm{EtOAc}(5 \mathrm{~mL}$ ) for 30 s at room temperature. The reaction mixture was extracted with EtOAc $(5 \mathrm{~mL})$ twice. The organic layer was washed with brine $(10 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated in vacuo.

## 1-(Bis(4-methoxyphenyl)methyl)piperidine (2a)



Purification method: PTLC $\left(1 \% \mathrm{NEt}_{3}\right.$ in EtOAc: hexane $\left.=1: 2\right) /$ column chromatography on silica gel (hexane: $\mathrm{EtOAc}=10: 1$ to 5:1) (scaled-up synthesis)
$78.6 \mathrm{mg}, 0.252 \mathrm{mmol}, 84 \%$
$1.11 \mathrm{~g}, 3.54 \mathrm{mmol}, 70 \%$ (scaled-up synthesis, the resultant mixture was collected for 255 s )
yellow oil; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.30-7.26(\mathrm{~m}, 4 \mathrm{H}), 6.82-6.79(\mathrm{~m}, 4 \mathrm{H}), 4.14(\mathrm{~s}, 1 \mathrm{H})$, $3.75(\mathrm{~s}, 6 \mathrm{H}), 2.29(\mathrm{brs}, 4 \mathrm{H}), 1.55(\mathrm{tt}, J=5.2,5.2 \mathrm{~Hz}, 4 \mathrm{H}), 1.44-1.38(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 158.4,135.8,129.0,113.8,75.3,55.3,53.2,26.4,24.9$.
Spectral data of ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR were well consistent with those reported in previous literature ${ }^{[59]}$.

## $N$-Benzyl- $N$-(bis(4-methoxyphenyl)methyl)-1,1-bis(4-methoxyphenyl)methanamine (2b)



Purification method: PTLC $\left(5 \% \mathrm{NEt}_{3}\right.$ in EtOAc: hexane $\left.=1: 4\right)$
$78.3 \mathrm{mg}, 0.140 \mathrm{mmol}, 93 \%$
colorless oil; IR (neat): $2835,1608,1508,1459,1300,1246,1173,1034,824,734,567 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 7.23-7.19(\mathrm{~m}, 8 \mathrm{H}), 6.95-6.93(\mathrm{~m}, 3 \mathrm{H}), 6.83-6.81(\mathrm{~m}, 2 \mathrm{H}), 6.76-6.73$ $(\mathrm{m}, 8 \mathrm{H}), 4.96(\mathrm{~s}, 2 \mathrm{H}), 3.83(\mathrm{~s}, 2 \mathrm{H}), 3.76(\mathrm{~s}, 12 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 158.5,142.9$,
134.1, 130.5, 127.7, 127.3, 125.3, 113.4, 69.0, 55.3, 52.8; HRMS (ESI-TOF): calcd. for $\mathrm{C}_{37} \mathrm{H}_{37} \mathrm{NO}_{4}\left([\mathrm{M}+\mathrm{Na}]^{+}\right): 582.2615$, found 582.2615 .

## $N$-(Bis(4-methoxyphenyl)methyl)aniline (2c)



Purification method: PTLC (EtOAc: hexane $=1: 4$, twice)
$69.2 \mathrm{mg}, 0.217 \mathrm{mmol}, 72 \%$
colorless oil; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.25-7.23(\mathrm{~m}, 4 \mathrm{H}), 7.13-7.09(\mathrm{~m}, 2 \mathrm{H}), 6.87-6.83(\mathrm{~m}$, $4 \mathrm{H}), 6.68$ (dd, $J=7.2,7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.54-6.52(\mathrm{~m}, 2 \mathrm{H}), 5.41$ (s, 1H), 4.15 (brs, 1 H ), 3.78 ( $\mathrm{s}, 6 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 158.9,147.6,135.5,129.2,128.6,117.6,114.1,113.6,61.8,55.4$. Spectral data of ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR were well consistent with those reported in previous literature ${ }^{[59]}$.

## $N$-(Bis(4-methoxyphenyl)methyl)-2-bromo-aniline (2d)



Purification method: PTLC (EtOAc: hexane $=1: 4$, twice)
$81.6 \mathrm{mg}, 0.205 \mathrm{mmol}, 68 \%$
yellow oil; IR (neat): $3410,2828,1591,1458,1300,1173,1033,818,743,574 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.42(\mathrm{dd}, J=2.0,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.25-7.21(\mathrm{~m}, 4 \mathrm{H}), 7.03-6.99(\mathrm{~m}, 1 \mathrm{H}), 6.87-$ 6.83 (m, 4H), 6.53 (ddd, $J=1.2,8.0,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.45(\mathrm{dd}, J=1.2,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.46(\mathrm{~d}, J=4.4$ $\mathrm{Hz}, 1 \mathrm{H}), 4.84(\mathrm{~d}, J=4.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.77(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 159.0,144.2$, 134.8, 132.3, 128.5, 128.4, 118.1, 114.2, 112.9, 109.9, 61.6, 55.4; HRMS (ESI-TOF): calcd. for
$\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{NO}_{2} \mathrm{Br}\left([\mathrm{M}+\mathrm{Na}]^{+}\right): 420.0570$, found 420.0568 .

## 3-(Bis(bis)(4-mehoxyphenyl)methyl)amino)propan-1-ol (2e)



Purification method: PTLC $\left(5 \% \mathrm{NEt}_{3}\right.$ in EtOAc: hexane $\left.=1: 4\right)$
$32.1 \mathrm{mg}, 0.0608 \mathrm{mmol}, 41 \%$
colorless oil; IR (neat): 2952, 1608, 1508, 1460, 1300, 1245, 1173, 1033, 822, $569 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.30-7.26(\mathrm{~m}, 8 \mathrm{H}), 6.85-6.82(\mathrm{~m}, 8 \mathrm{H}), 4.95(\mathrm{~s}, 2 \mathrm{H}), 3.78(\mathrm{~s}, 12 \mathrm{H}), 3.10(\mathrm{brs}$, $2 \mathrm{H}), 2.82-2.78(\mathrm{~m}, 2 \mathrm{H}), 0.99-0.89(\mathrm{~m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 158.6,134.8,130.1$, $113.7,68.2,61.6,55.3,45.1,32.5$; HRMS (ESI-TOF): calcd. for $\mathrm{C}_{33} \mathrm{H}_{37} \mathrm{NO}_{5}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$: 550.2564 , found 550.2564.

## 1-(Bis(4-methoxyphenyl)methyl)diethylamine (2g)



Purification method: PTLC $\left(5 \% \mathrm{NEt}_{3}\right.$ in EtOAc: hexane $\left.=1: 1\right)$
$73.5 \mathrm{mg}, 0.245 \mathrm{mmol}, 82 \%$
colorless oil; IR (neat): 2967, 1608, 1508, 1457, 1244, 1173, 1036, $819 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 7.31-7.29(\mathrm{~m}, 4 \mathrm{H}), 6.81-6.79(\mathrm{~m}, 4 \mathrm{H}), 4.62(\mathrm{~s}, 1 \mathrm{H}), 3.76(\mathrm{~s}, 6 \mathrm{H}), 2.53(\mathrm{q}, J=6.8 \mathrm{~Hz}$, $4 \mathrm{H}), 0.96(\mathrm{t}, J=6.8 \mathrm{~Hz}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 158.3,136.0,129.1,113.7,69.8$,
55.3, 42.9, 11.3; HRMS (ESI-TOF): calcd. for $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{NO}_{2}\left([\mathrm{M}+\mathrm{Na}]^{+}\right): 322.1777$, found 322.1777.

## tert-Butyl (bis(4-methoxyphenyl)methyl)prolinate (2h)



Purification method: PTLC $\left(5 \% \mathrm{NEt}_{3}\right.$ in EtOAc: hexane $\left.=1: 4\right)$
$81.9 \mathrm{mg}, 0.206 \mathrm{mmol}, 69 \%$ ( 2.00 eq . of DBU was used.)
yellow oil; IR (neat): 2972, 1723, 1608, 1509, 1245, 1146, 1035, $821 \mathrm{~cm}^{-1} ;[\alpha]^{23}{ }_{\mathrm{D}}=-41.4(\mathrm{c} 0.79$, $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.35(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.29(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.80-$ $6.79(\mathrm{~m}, 2 \mathrm{H}), 6.78-6.77(\mathrm{~m}, 2 \mathrm{H}), 4.81(\mathrm{~s}, 1 \mathrm{H}), 3.75(\mathrm{~s}, 3 \mathrm{H}), 3.74(\mathrm{~s}, 3 \mathrm{H}), 3.38-3.36(\mathrm{~m}, 1 \mathrm{H}), 2.95-$ $2.90(\mathrm{~m}, 1 \mathrm{H}), 2.66-2.60(\mathrm{~m}, 1 \mathrm{H}), 2.12-2.06(\mathrm{~m}, 1 \mathrm{H}), 1.93-1.79(\mathrm{~m}, 3 \mathrm{H}), 1.35(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 174.3,158.5,158.4,136.6,135.8,129.1,128.8,113.73,113.72,80.0,70.3$, 63.3, 55.3, 51.4, 30.2, 28.1, 23.5; HRMS (ESI-TOF): calcd. for $\mathrm{C}_{24} \mathrm{H}_{31} \mathrm{NO}_{4}\left([\mathrm{M}+\mathrm{Na}]^{+}\right): 420.2145$, found 420.2145 .

## N -(Bis(4-methoxyphenyl)methyl)butan-2-amine (2i)



Purification method: PTLC $\left(5 \% \mathrm{NEt}_{3}\right.$ in EtOAc: hexane $\left.=1: 4\right)$
$53.9 \mathrm{mg}, 0.165 \mathrm{mmol}, 55 \%$
yellow oil; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.25-7.22(\mathrm{~m}, 4 \mathrm{H}), 6.80-6.78(\mathrm{~m}, 4 \mathrm{H}), 5.16(\mathrm{~s}, 1 \mathrm{H})$, $3.78(\mathrm{~s}, 6 \mathrm{H}), 3.27(\mathrm{sep}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 0.98(\mathrm{~d}, J=6.4 \mathrm{~Hz}, 12 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 158.0,137.9,130.2,113.2,62.4,55.3,46.4,22.7$.
Spectral data of ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR were well consistent with those reported in previous
literature ${ }^{[99]}$.

## (Bis(4-methoxyphenyl)methyl)(phenyl)sulfane (2j)



Purification method: PTLC (EtOAc: hexane =1:2)
$87.4 \mathrm{mg}, 0.260 \mathrm{mmol}, 87 \%$
white solid; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.33-7.30(\mathrm{~m}, 4 \mathrm{H}), 7.23-7.12(\mathrm{~m}, 5 \mathrm{H}), 6.84-6.81(\mathrm{~m}$, $4 \mathrm{H}), 5.49(\mathrm{~s}, 1 \mathrm{H}), 3.78(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 158.7$, 136.6, 133.4, 130.3, 129.5, 128.8, 126.4, 114.0, 56.1, 55.3.

Spectral data of ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR were well consistent with those reported in previous literature ${ }^{[59]}$.

## (Bis(4-methoxyphenyl)methyl)decanthiolate (2k)



Purification method: PTLC (EtOAc: hexane =1:2)
$75.1 \mathrm{mg}, 0.187 \mathrm{mmol}, 62 \%$
yellow oil; IR (neat): 2925, 2853, 1608, 1508,1457, 1248, 1173,1036 $\mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 7.34-7.30(\mathrm{~m}, 4 \mathrm{H}), 6.85-6.81(\mathrm{~m}, 4 \mathrm{H}), 5.07(\mathrm{~s}, 1 \mathrm{H}), 3.78(\mathrm{~s}, 6 \mathrm{H}), 2.35(\mathrm{t}, J=7.2 \mathrm{~Hz}$, $2 \mathrm{H}), 1.53(\mathrm{tt}, J=7.2,7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.33-1.23(\mathrm{~m}, 14 \mathrm{H}), 0.88(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 158.6,134.1,129.4,113.9,55.3,52.9,32.4,32.0,29.7,29.6,29.4,29.3,29.2$, 29.0, 22.8, 14.3; HRMS (ESI-TOF): calcd. for $\mathrm{C}_{25} \mathrm{H}_{36} \mathrm{O}_{2} \mathrm{~S}\left([\mathrm{M}+\mathrm{Na}]^{+}\right)$: 423.2328 , found 423.2326 .

## (Bis(4-methoxyphenyl)methyl)phenol (21)



Purification method: PTLC (EtOAc: hexane $=1: 4$, twice)
$40.6 \mathrm{mg}, 0.127 \mathrm{mmol}, 42 \%$ ( 2.00 eq . of DBU was used.)
yellow oil; IR (neat): 1608, 1300, 1173, 1032, 816, 753, 690, $573 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right): ~ \delta 7.31-7.29(\mathrm{~m}, 4 \mathrm{H}), 7.22-7.18(\mathrm{~m}, 2 \mathrm{H}), 6.94-6.85(\mathrm{~m}, 7 \mathrm{H}), 6.14(\mathrm{~s}, 1 \mathrm{H}), 3.77(\mathrm{~s}, 6 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 159.2,158.3,133.8,129.4,128.3,121.0,116.3,114.1,81.0,55.4 ;$ HRMS (ESI-TOF): calcd. for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{O}_{3}$ ([M+Na] ${ }^{+}$): 343.1305, found 343.1305.

## 5-(Bis(4-methoxyphenyl)methyl)-2,2-dimethyl-1,3-dioxane-4,6-dione (2n)



Purification method: GPC
$53.4 \mathrm{mg}, 0.144 \mathrm{mmol}, 48 \%$ ( 2.00 eq. of DBU was used. The resultant mixture was poured into a test tube containing 1 M HCl aq. $(5 \mathrm{~mL})$ and $\mathrm{EtOAc}(5 \mathrm{~mL})$ instead of sat. $\mathrm{NaHCO}_{3}$ aq. $(5 \mathrm{~mL})$ and EtOAc ( 5 mL ).)
yellow solid; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.24-7.21(\mathrm{~m}, 4 \mathrm{H}), 6.85-6.81(\mathrm{~m}, 4 \mathrm{H}), 5.28(\mathrm{~d}, J=$ $2.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.25(\mathrm{~d}, J=2.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.78(\mathrm{~s}, 6 \mathrm{H}), 1.73(\mathrm{~s}, 3 \mathrm{H}), 1.52(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 165.0,158.6,132.5,130.4,113.8,105.2,55.3,51.5,48.0,28.4,27.8$.

Spectral data of ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR were well consistent with those reported in previous literature ${ }^{[S 10]}$.

## 4,4'-(Phenylmethylene)bis(methoxybenzene) (20)



Purification method: PTLC (EtOAc: hexane $=1: 4)$

## $70.2 \mathrm{mg}, 0.231 \mathrm{mmol}, 77 \%$

white solid; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.28-7.24(\mathrm{~m}, 2 \mathrm{H}, \mathrm{f}), 7.18(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.11-$ $7.09(\mathrm{~m}, 2 \mathrm{H}), 7.02-6.99(\mathrm{~m}, 4 \mathrm{H}), 6.83-6.80(\mathrm{~m}, 4 \mathrm{H}), 5.44(\mathrm{~s}, 1 \mathrm{H}), 3.76(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 158.1,144.7,136.6,130.4,129.4,128.4,126.3,113.8,55.34,55.30$.

Spectral data of ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR were well consistent with those reported in previous literature ${ }^{[811]}$.

## 1-Phenyl-(4-methoxyphenyl)-methyl-piperidine (2r)



Purification method: PTLC $\left(5 \% \mathrm{NEt}_{3}\right.$ in EtOAc: hexane $\left.=1: 4\right)$
$22.6 \mathrm{mg}, 0.0803 \mathrm{mmol}, 27 \%$
white solid; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.39-7.37(\mathrm{~m}, 2 \mathrm{H}), 7.30-7.23(\mathrm{~m}, 4 \mathrm{H}), 7.17-7.13(\mathrm{~m}$, $1 \mathrm{H}), 6.81-6.79(\mathrm{~m}, 2 \mathrm{H}), 4.18(\mathrm{~s}, 1 \mathrm{H}), 3.75(\mathrm{~s}, 3 \mathrm{H}), 2.30(\mathrm{brs}, 4 \mathrm{H}), 1.55(\mathrm{tt}, J=5.6,5.6 \mathrm{~Hz}, 4 \mathrm{H})$, 1.44-1.39 (m, 2H); ${ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 158.5,143.8,135.5,129.2,128.4,128.0$, 126.7, 113.8, 76.1, 55.3, 53.2, 26.4, 24.9.

Spectral data of ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR were well consistent with those reported in previous literature ${ }^{[512]}$.

## 1-((3,4-Methylenedioxyphenyl)-(4-methoxyphenyl)methylene)piperidine (2s)



Purification method: PTLC $\left(5 \% \mathrm{NEt}_{3}\right.$ in EtOAc: hexane $\left.=1: 4\right)$
$81.9 \mathrm{mg}, 0.252 \mathrm{mmol}, 84 \%$
colorless oil; IR (neat): 2930, 2361, 1506, 1483, 1440, 1244, 1038, $934 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 7.28-7.25(\mathrm{~m}, 2 \mathrm{H}), 6.93(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.82-6.79(\mathrm{~m}, 3 \mathrm{H}), 6.68(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, $1 \mathrm{H}), 5.89-5.87(\mathrm{~m}, 2 \mathrm{H}), 4.08(\mathrm{~s}, 1 \mathrm{H}), 3.75(\mathrm{~s}, 3 \mathrm{H}), 2.28(\mathrm{brs}, 4 \mathrm{H}), 1.54(\mathrm{tt}, J=6.0,6.0 \mathrm{~Hz}, 4 \mathrm{H})$, 1.44-1.41 (m, 2H); ${ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 158.4,147.8,146.2,138.0,135.6,128.9$, 121.0, 113.8, 108.1, 108.0, 100.9, 75.7, 55.3, 53.2, 26.4, 24.9; HRMS (ESI-TOF): calcd. for $\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{NO}_{3}\left([\mathrm{M}+\mathrm{Na}]^{+}\right): 348.1570$, found 348.1570 .

## 1-((4-Methoxyphenyl)(3,4,5-trimethoxyphenyl)methyl)piperidine (2t)



Purification method: PTLC $\left(5 \% \mathrm{NEt}_{3}\right.$ in EtOAc: hexane $\left.=1: 4\right)$
$81.9 \mathrm{mg}, 0.219 \mathrm{mmol}, 73 \%$
colorless oil; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.30-7.29(\mathrm{~m}, 2 \mathrm{H}, \mathrm{c}), 6.83-6.81(\mathrm{~m}, 2 \mathrm{H}), 6.65(\mathrm{~s}$, $2 \mathrm{H}), 4.08(\mathrm{~s}, 1 \mathrm{H}), 3.83(\mathrm{~s}, 6 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 3.76(\mathrm{~s}, 3 \mathrm{H}), 2.30-2.29(\mathrm{~m}, 4 \mathrm{H}), 1.55(\mathrm{tt}, J=6.0,6.0$ $\mathrm{Hz}, 4 \mathrm{H}), 1.44-1.43(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta 158.5,153.1,139.6,136.5,135.1$, $129.1,113.7,104.6,76.2,60.9,56.2,55.3,53.2,26.4,24.8$.

Spectral data of ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR were well consistent with those reported in previous literature ${ }^{[\mathrm{S} 3]}$.

## 6. References

S1) Czyz, M. L.; Taylor, M. S.; Horngren, T. H.; Polyzos, A. ACS Catal. 2021, 11, 5472-5480.
S2) Reddel, J. C. T.; Wang, W.; Koukounas, K.; Thomson, R. J. Chem. Sci. 2017, 8, 2156-2160.
S3) Ana, G.; Kelly, P. M.; Malebari, A. M.; Noorani, S.; Nathwani, S. M.; Twamley, B.; Fayne. D.; O’Boyle, N. M.; Zisterer, D. M.; Pimentel, E. F.; Endringer, D. C.; Meegan, M. J. Pharmaceuticals 2021, 14, 169.

S4) Hayakawa, Y.; Iwase, T.; Nurminen, E. J.; Tsukamoto, M.; Kataoka, M. Tetrahedron 2005, 61, 2203-2209.
S5) Raamata, E.; Kaupmeesa, K.; Ovsjannikova, G.; Trummalb, G.; Kütta, A.; Saamea, J.; Koppela, I.; Kaljuranda, I.; Lippinga, L.; Rodimaa, T.; Pihla, V.; Koppela, I. A.; Leitoa, I. Phys. Org. Chem. 2013, 26, 162-170.
S6) Kutt, A.; Rodima, T.; Saame, J.; Raamat, E.; Mäemets, V.; Kaljurand, I.; Koppel, L. A.; Garlyauskayte, R. Y.; Yagupolskii, Y. L.; Yagupolskii, L. M.; Bernhardt, E.; Willner, H.; Leito, I. J. Org. Chem. 2011, 76, 391-395.

S7) Tshepelevitsh, S.; Kütt, A.; Lõkov, M.; Kaljurand, I.; Saame, J.; Heering, A.; Plieger, P. G.; Vianello, R.; Leito. I. Eur. J. Org. Chem. 2019, 6735-6748.
S8) Kelly-Rowley, A. M.; Lynch, V. M.; Anslyn. E. V. J. Am. Chem. Soc. 1995, 117, 3438-3447.
S9) Otake, Y.; Nakamura, H.; Fuse, S. Angew. Chem. Int. Ed. 2018, 57, 11389-11393.
S10) Zhao, S.; Hu, R.; Nie, Y.; Sheng, L.; He, W.; Zhu, N.; Li, Y.; Ji, D.; Guo, K. Chem. Eng. Process. 2022, 178, 109034-109048.
S11) Babu, K. N.; Massarwe, F.; Shioukhi, I.; Masarwa, A. Angew. Chem. Int. Ed. 2021, 60, 26199-26209.

S12) Wilsily, A.; Nguyen, Y.; Fillion, E. J. Am. Chem. Soc. 2009, 131, 15606-15607.
S13) Too, P. C.; Chan, G. H.; Tnay, Y. L.; Hirao, H.; Chiba, S. Angew. Chem. Int. Ed. 2016, 55, 3719-3723.

S14) Borys, A. M.; Gil-Negrete, J. M.; Hevia, E. Chem. Commun. 2021, 57, 8905-8908.

## 7. NMR chart

## $\alpha$-(4-Methoxyphenyl)-1,3-benzodioxole-5-methanol (1b)

( ${ }^{1} \mathrm{H}$ NMR, $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C} \mathrm{NMR}, 100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


## (4-Methoxyphenyl)(3,4,5-trimethoxyphenyl)methanol (1c)

( ${ }^{1} \mathrm{H}$ NMR, $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

$\left({ }^{13} \mathrm{C} \mathrm{NMR}, 100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$


1-(Bis(4-methoxyphenyl)methyl)piperidine (2a)
( ${ }^{1} \mathrm{H}$ NMR, $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

$\left({ }^{13} \mathrm{C} \mathrm{NMR}, 100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$

$N$-Benzyl- $N$-(bis(4-methoxyphenyl)methyl)-1,1-bis(4-methoxyphenyl)methanamine (2b) ( ${ }^{1} \mathrm{H}$ NMR, $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

$N$-(Bis(4-methoxyphenyl)methyl)aniline (2c)
( ${ }^{1} \mathrm{H} \mathbf{N M R}, 400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


N -(Bis(4-methoxyphenyl)methyl)-2-bromo-aniline (2d)
( ${ }^{1} \mathrm{H}$ NMR, $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


## 3-(Bis(bis)(4-mehoxyphenyl)methyl)amino)propan-1-ol (2e)

( ${ }^{1} \mathrm{H}$ NMR, $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


1-(Bis(4-methoxyphenyl)methyl)diethylamine (2g)
( ${ }^{1} \mathrm{H} \mathbf{N M R}, 400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

tert-Butyl (bis(4-methoxyphenyl)methyl)prolinate (2h)
( ${ }^{\mathbf{1}} \mathrm{H} \mathrm{NMR}, 400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

$N$-(Bis(4-methoxyphenyl)methyl)butan-2-amine (2i)
( ${ }^{\mathbf{1}} \mathrm{H} \mathbf{N M R}, 400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

(Bis(4-methoxyphenyl)methyl)(phenyl)sulfane (2j)
( ${ }^{1} \mathrm{H}$ NMR, $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

(Bis(4-methoxyphenyl)methyl)decanthiolate (2k)
$\left({ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}, 400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

(Bis(4-methoxyphenyl)methyl)phenol (21)
( ${ }^{1} \mathrm{H}$ NMR, $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


5-(Bis(4-methoxyphenyl)methyl)-2,2-dimethyl-1,3-dioxane-4,6-dione (2n)
( ${ }^{1} \mathrm{H} \mathbf{N M R}, 400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


4,4'-(Phenylmethylene)bis(methoxybenzene) (20)
( ${ }^{1} \mathrm{H}$ NMR, $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


1-Phenyl-(4-methoxyphenyl)-methyl-piperidine (2r)
$\left({ }^{1} \mathrm{H} \mathbf{N M R}, 400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


1-((3,4-Methylenedioxyphenyl)-(4-methoxyphenyl)methylene)piperidine (2s)
$\left({ }^{\mathbf{1}} \mathrm{H} \mathbf{N M R}, 400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


1-((4-Methoxyphenyl)(3,4,5-trimethoxyphenyl)methyl)piperidine (2t)
( ${ }^{1} \mathrm{H}$ NMR, $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )

( ${ }^{13} \mathrm{C}$ NMR, $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ )


