

Supporting Information

Alkali Metal Doped Ruthenium Dioxide Nanosheets with
Lattice Distortion as High Active Oxygen Evolution
Electrocatalysts in Acidic Media

Hai Tang^{a,#}, Ming Kong,^{a,#} Wenwen Cao,^{b,#} Wenguang Ma^b, Jianjun Ding^b, Chuanxin
Shi,^{b,*} Xiaodong Yang,^{b,*} Qi Shen,^{b,*} Yiqiang Sun^{b,c}

Dr. H. Tang, M. Kong

Suzhou Vocational Institute of Industrial Technology, Suzhou, Jiangsu, 215104, China

Dr. W. W. Cao, M. G. Ma, J. J. Ding, C. X. Shi, X. D. Yang, Q. Shen, Prof. Y. Q. Sun
School of Chemistry and Chemical Engineering, University of Jinan, Jinan 250022, P.
R. China

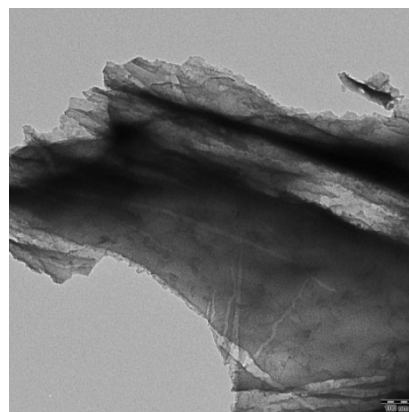
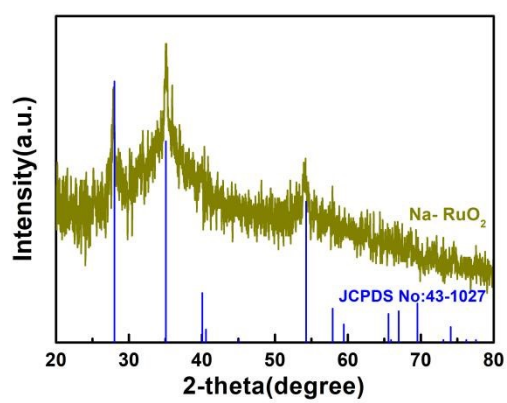


Figure S1: (a) XRD pattern and (b) TEM image of Na-RuO₂ nanosheets.

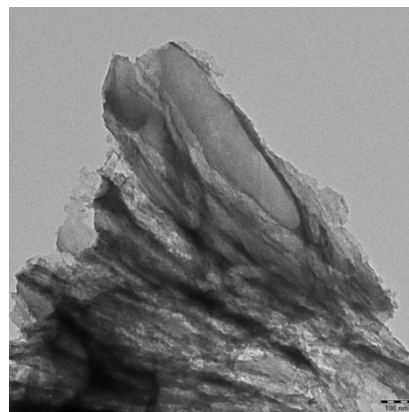
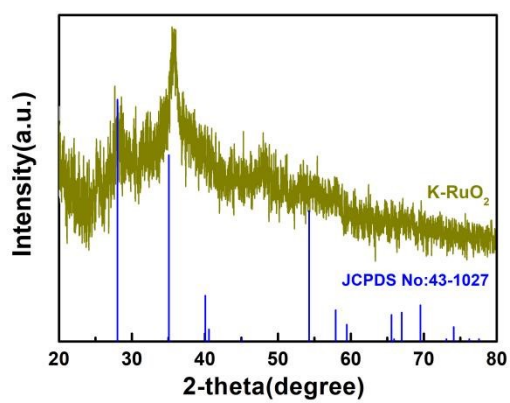


Figure S2: (a) XRD pattern and (b) TEM image of K-RuO₂ nanosheets.

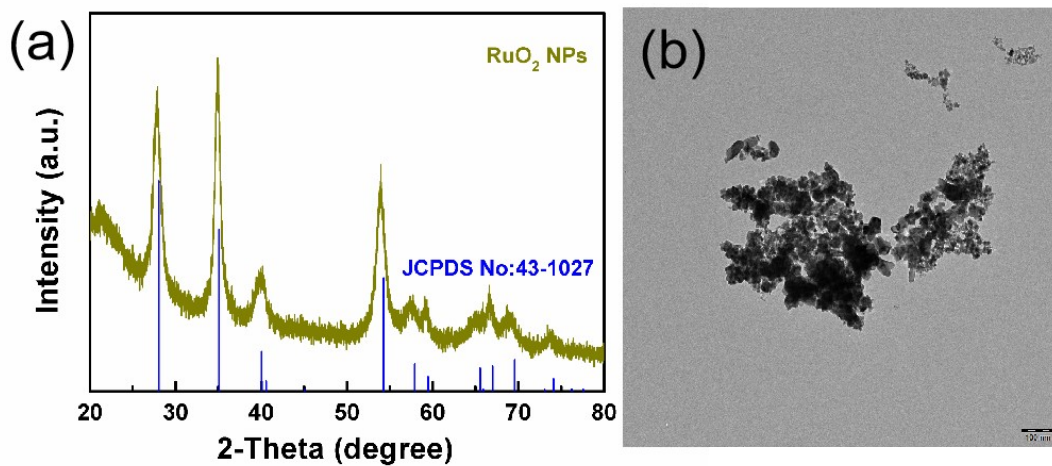


Figure S3: (a) XRD pattern and (b) TEM image of RuO₂ NPs.

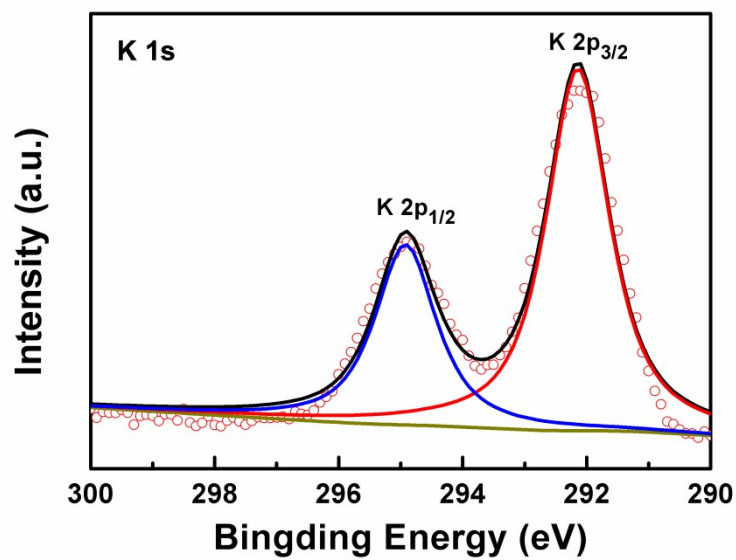


Figure S4: The high-resolution XPS spectra of K 2p of K doped RuO₂ NSs.

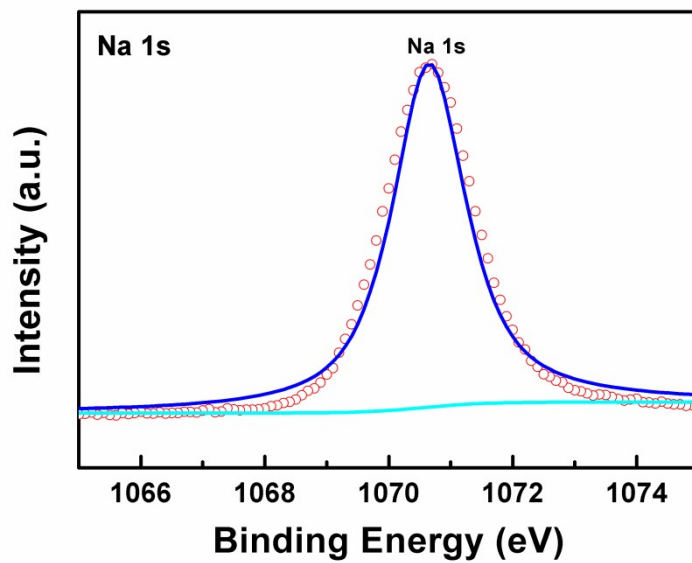


Figure S5: The high-resolution XPS spectra of Na 1s of Na doped RuO₂ NSs.

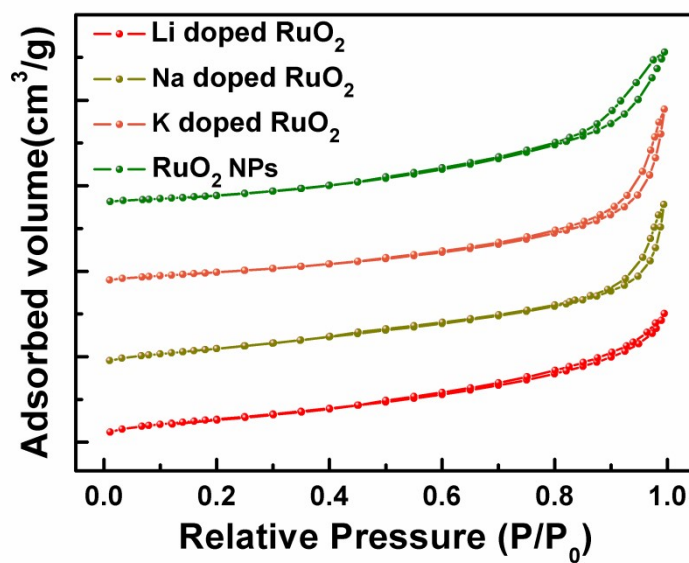


Figure S6: N₂ adsorption-desorption isotherms of Li doped RuO₂ NSs, K doped RuO₂ NSs, Na doped RuO₂ NSs and pristine RuO₂ NPs.

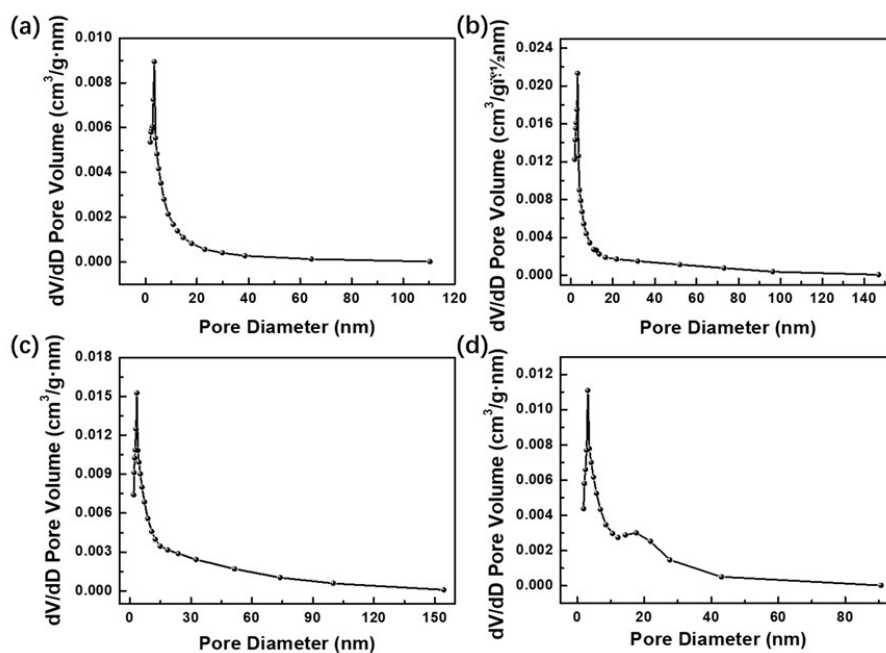


Figure S7: Pore size distribution curves of (a) Li doped RuO₂ NSs, (b) K doped RuO₂ NSs, (c) Na doped RuO₂ NSs, (d) pristine RuO₂ NPs.

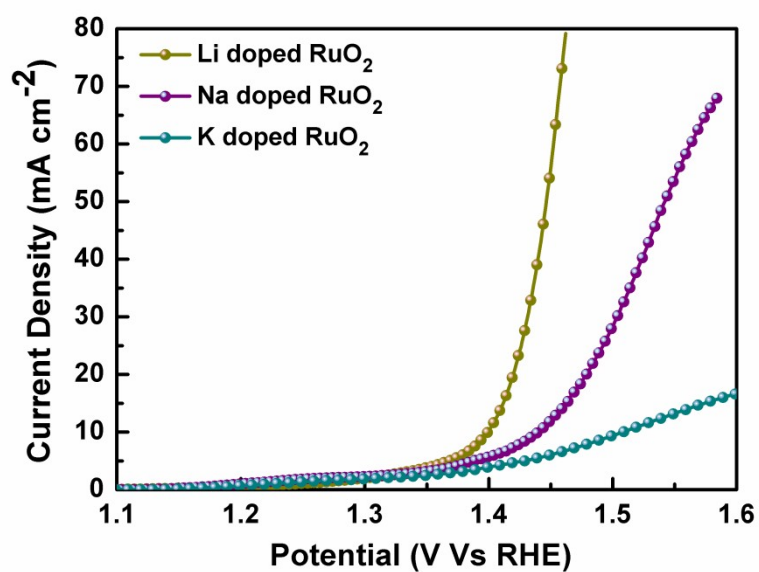


Figure S8. LSV curve of Li doped RuO₂ NSs, K doped RuO₂ NSs, Na doped RuO₂ NSs.

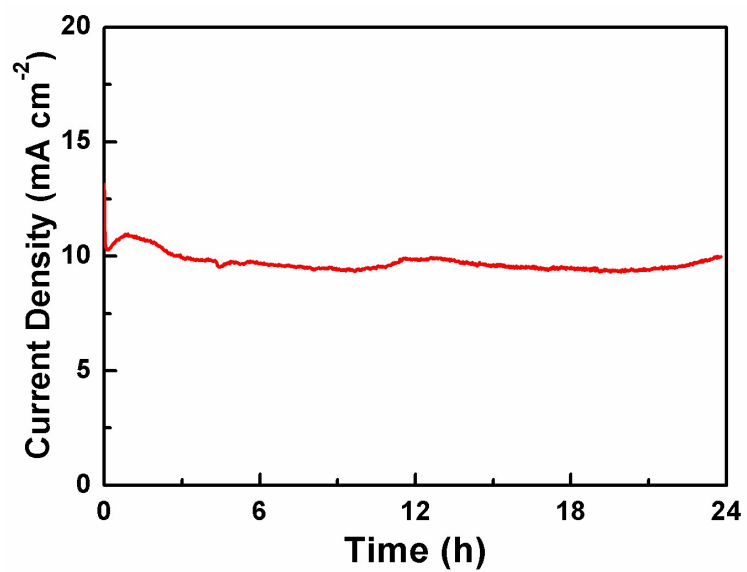


Figure S9. The chronoamperometry curves of the Li-doped RuO₂ nanosheets electrode.

Table S1. Atom% of Li in RuO₂ sample.

Elements	EDS results
Li	6.18

Table S2. Comparisons of OER performance for some Ru based electrocatalysts in acid condition.

Catalyst	Electrolyte	Overpotential[mV] at 10 mA cm ⁻²	Ref.
RuO ₂ /(CoMn) ₃ O ₄ /CC	0.5 M H ₂ SO ₄	270 mV	1
RuNi ₂ @G-250	0.5 M H ₂ SO ₄	227 mV	2
Nd _{0.1} RuO _x /CC	0.5 M H ₂ SO ₄	211 mV	3
Ru/Mo ₂ C	0.5 M H ₂ SO ₄	215 mV	4
Ru@V-RuO ₂ /C HMS	0.5 M H ₂ SO ₄	176 mV	5
Si-RuO _x @C	0.5 M H ₂ SO ₄	220 mV	6
Co-RuIr nanocrystals	0.1 M HClO ₄	235 mV	7
a-RuTe ₂ PNRs	0.5 M H ₂ SO ₄	245 mV	8
RuCu nanosheet	0.5 M H ₂ SO ₄	236 mV	9
Ru nanosheet	0.5 M H ₂ SO ₄	260 mV	10
Cr _{0.6} Ru _{0.4} O ₂	0.5 M H ₂ SO ₄	178 mV	11
Ultrafine defective RuO ₂	0.5 M H ₂ SO ₄	179 mV	12
Cu-doped RuO ₂	0.5 M H ₂ SO ₄	188 mV	13
Co-doped RuO ₂ NWs	0.5 M H ₂ SO ₄	200 mV	14
NaRuO ₂ NSs	0.1 M HClO ₄	255 mV	15
Ru@FLC	0.5 M H ₂ SO ₄	258 mV	16
RuO ₂ NWs	0.5 M H ₂ SO ₄	234 mV	17
Li-RuO ₂ NSs	0.5 M H ₂ SO ₄	169 mV	This work

Reference:

- 1、 S. Q. Niu, X. P. Kong, S. W. Li, Y. Y. Zhang, J. Wu, W. W. Zhao and P. Xu, *Applied Catalysis B: Environmental*, 2021, **297**, 120442.
- 2、 X. J. Cui, P. J. Ren, C. Ma, J. Zhao, R. X. Chen, S. M. Chen, N. P. Rajan, H. B. Li, L. Yu and Z. Q. Tian, *Adv. Mater.*, 2020, **32**, 1908126.
- 3、 L. Li, G. W. Zhang, J. W. Xu, H. J. He, B. Wang, Z. M. Yang and S. C. Yang, *Adv. Funct. Mater.*, 2023, 2213304.
- 4、 Z. Liu, Y. X. Liu, H. W. He, H. B. Shao, Y. Zhang, J. Li and W. W. Cai, *Electrochim. Acta*, 2023, 141920.
- 5、 Y. P. Li, W. T. Wang, M. Y. Cheng, Y. F. Feng, X. Han, Q. Z. Qian, Y. Zhu and G. Q. Zhang, *Adv. Mater.*, 2023, 2206351.
- 6、 C. X. Liu, Y. B. Jiang, T. Wang, Q. S. Li and Y. Z. Liu, *Adv. Sci.*, 2023, 2207429.
- 7、 J. Q. Shan, T. Ling, K. Davey, Y. Zheng and S. Z. Qiao, *Adv. Mater.*, 2019, **31**, 1900510.
- 8、 J. Wang, L. L. Han, B. L. Huang, Q. Shao, H. L. Xin and X. Q. Huang, *Nature commun.*, 2019, **10**, 5692.
- 9、 Q. Yao, B. L. Huang, N. Zhang, M. Z. Sun, Q. Shao and X. Q. Huang, *Angew. Chem.*, 2019, **131**, 14121-14126.
- 10、 X. K. Kong, K. Xu, C. L. Zhang, J. Dai, S. Norooz Oliaee, L. Y. Li, X. C. Zeng, C. Z. Wu and Z. M. Peng, *Acs Catalysis*, 2016, **6**, 1487-1492.
- 11、 Y. C. Lin, Z. Q. Tian, L. J. Zhang, J. Y. Ma, Z. Jiang, B. J. Deibert, R. X. Ge and L. Chen, *Nature commun.*, 2019, **10**, 162.
- 12、 R. X. Ge, L. Li, J. W. Su, Y. C. Lin, Z. Q. Tian and L. Chen, *Adv. Energy Mater.*, 2019, **9**, 1901313.
- 13、 J. W. Su, R. X. Ge, K. M. Jiang, Y. Dong, F. Hao, Z. Q. Tian, G. X. Chen and L. Chen, *Adv. Mater.*, 2018, **30**, 1801351.
- 14、 J. Wang, Y. J. Ji, R. G. Yin, Y. Y. Li, Q. Shao and X. Q. Huang, *J. Mater. Chem. A*, 2019, **7**, 6411-6416.
- 15、 S. Laha, Y. Lee, F. Podjaski, D. Weber, V. Duppel, L. M. Schoop, F. Pielhofer,

C. Scheurer, K. Müller and U. Starke, *Adv. Energy Mater.*, 2019, **9**, 1803795.

16、 C. X. Shi, Y. Yuan, Q. Shen, X. D. Yang, B. Q. Cao, B. Xu, B. T. Kang, Y. Q. Sun and C. C. Li, *J. Colloid Interface Sci.*, 2022, **612**, 488-495.

17、 J. Yang, Y. J. Ji, Q. Shao, N. Zhang, Y. Y. Li and X. Q. Huang, *Adv. Funct. Mater.*, 2018, **28**, 1803722.