

Supporting information

Evaluation of synthon influence on ethenzamide-polyphenol pharmaceutical cocrystals

Francisco Javier Acebedo-Martínez,^a Alicia Domínguez-Martín,^b Carolina Alarcón-Payer,^c Antonio Frontera,^d Ángel Ibáñez,^b and Duane Choquesillo-Lazarte^{*a}

^aLaboratorio de Estudios Cristalográficos, IACT, CSIC-Universidad de Granada, Avda. de las Palmeras 4, 18100 Armilla, Spain.

^bDepartment of Inorganic Chemistry, Faculty of Pharmacy, University of Granada, 18071 Granada, Spain.

^cServicio de Farmacia, Hospital Universitario Virgen de las Nieves, 18014 Granada, Spain.

^dDepartment of Chemistry, Universitat de les Illes Balears, Crta de Valldemossa km 7.5, 07122 Palma de Mallorca (Balears), Spain

Email: duane.choquesillo@csic.es

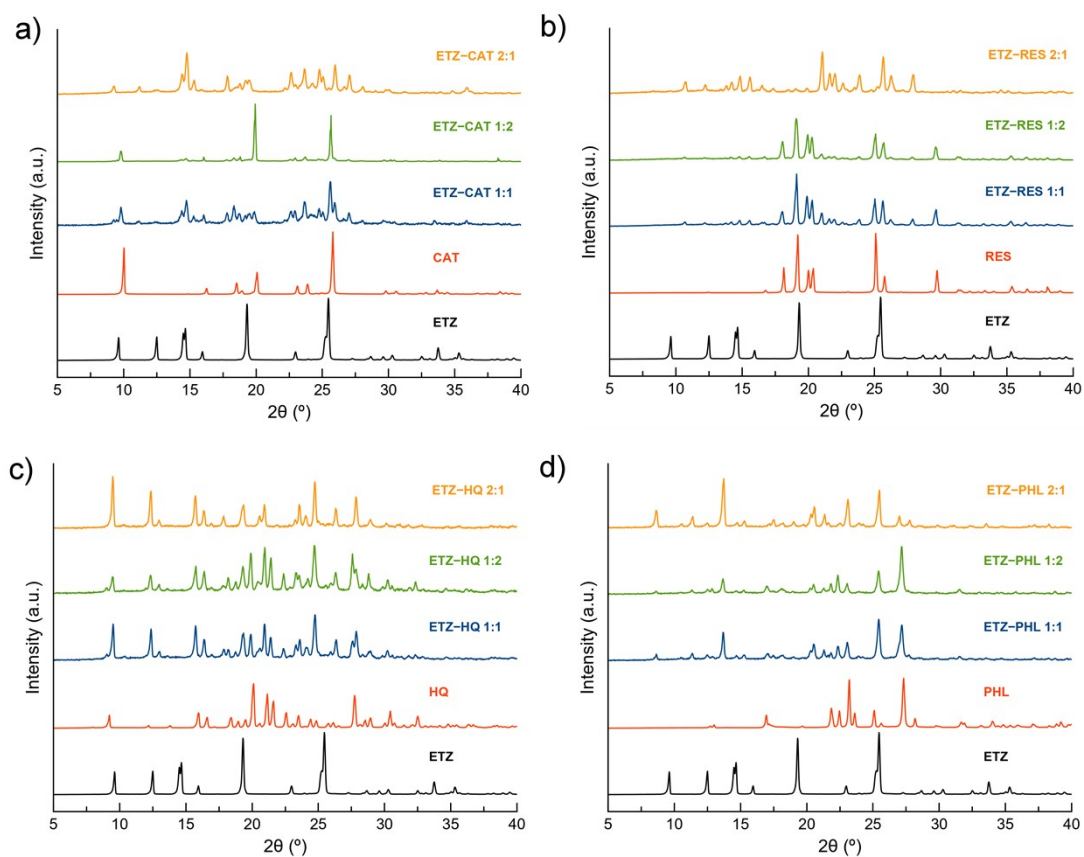
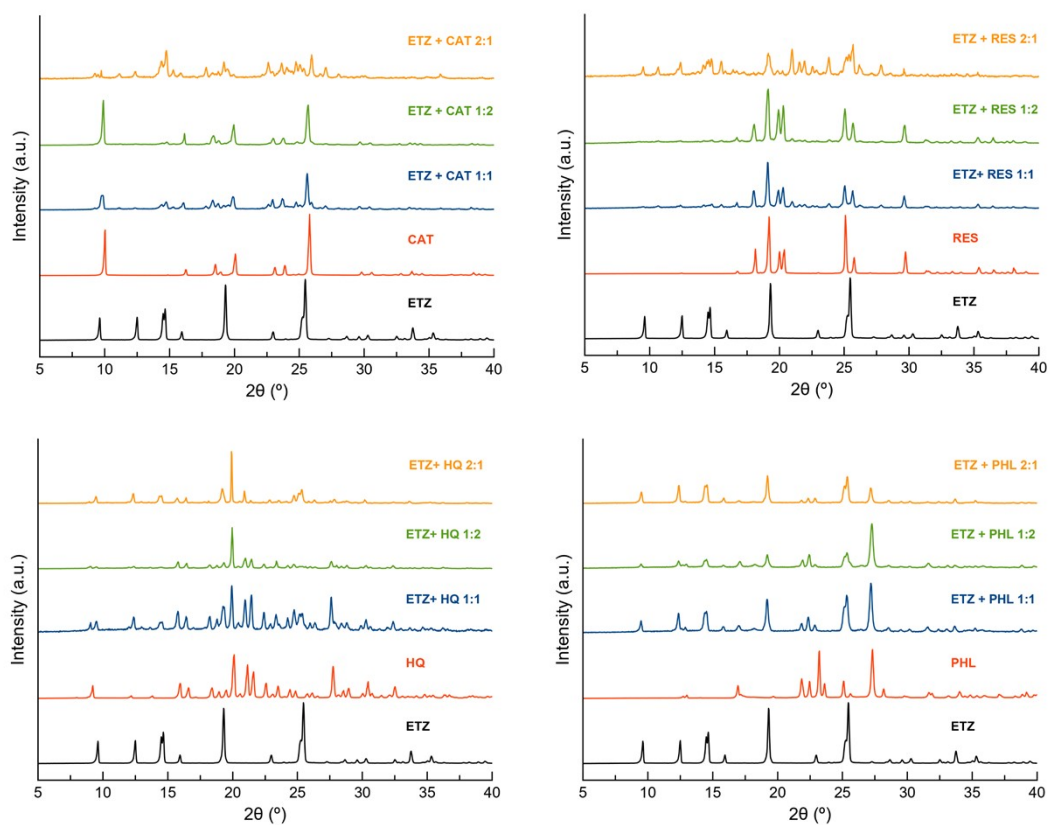


Fig. S1 Powder X-ray diffraction patterns of the product obtained by liquid-assisted grinding (LAG).



ig. S2 Powder X-ray diffraction patterns of the products obtained by neat grinding in different stoichiometric ratios.

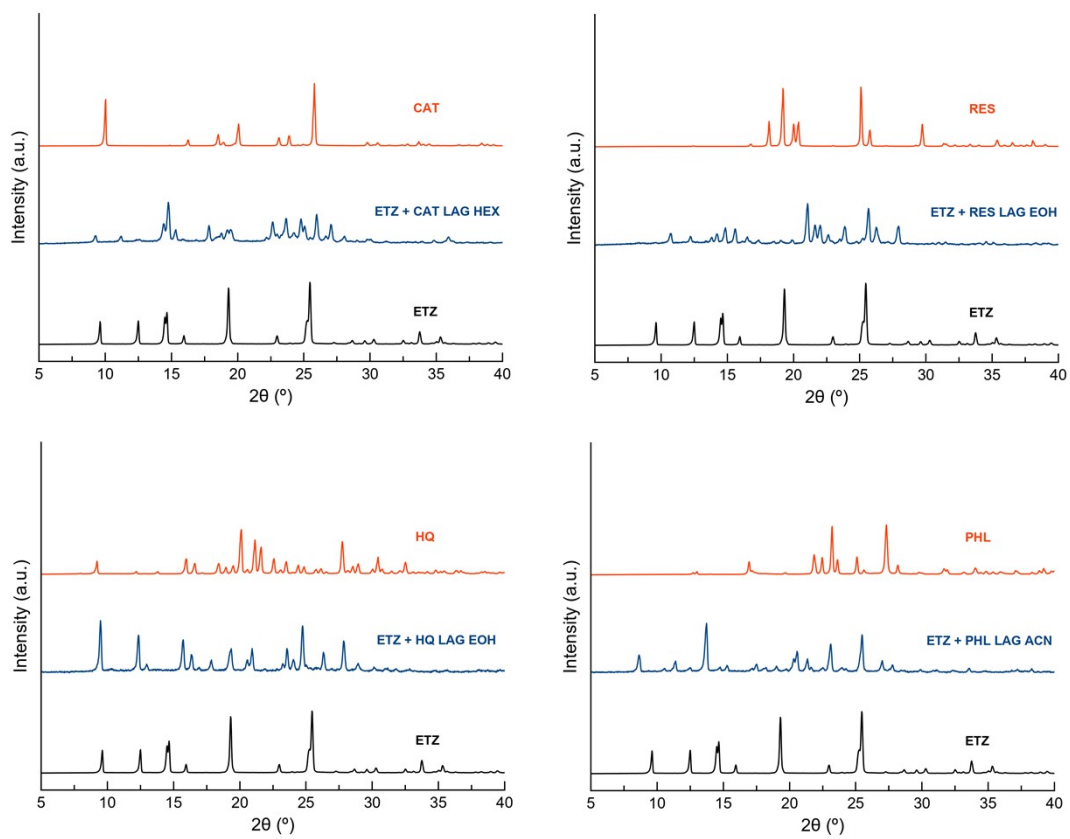


Fig. S3 Powder X-ray diffraction patterns of the products obtained by liquid-assisted grinding (LAG) in selected solvents at 2:1 stoichiometric ratio.

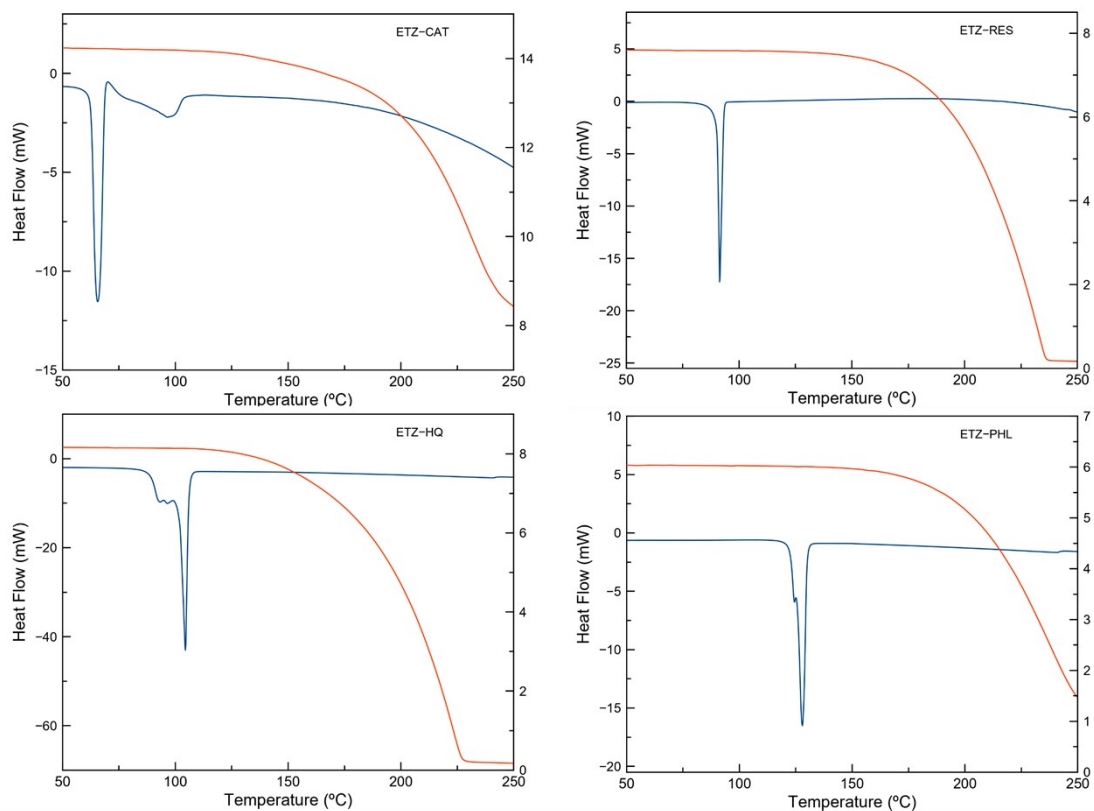


Fig. S4 Differential scanning calorimetry (DSC) and thermogravimetric (TG) traces of the ETZ cocrystals.

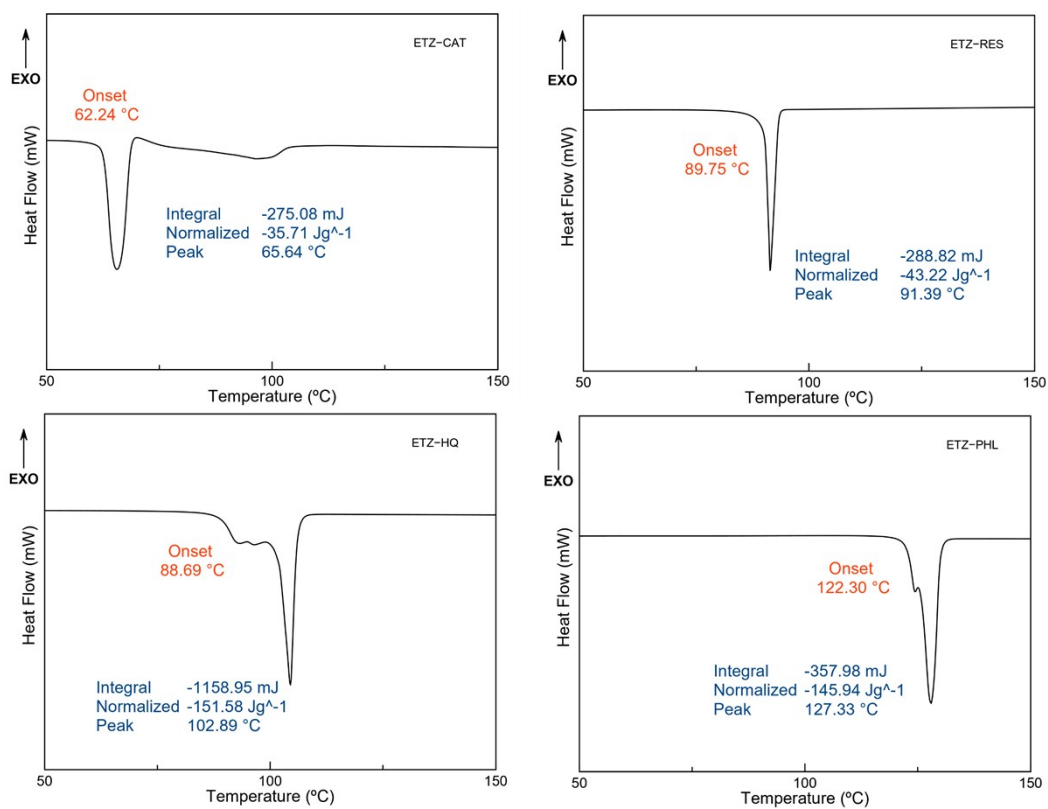


Fig. S5 DSC traces of novel cocrystals with the corresponding onsets and enthalpies values.

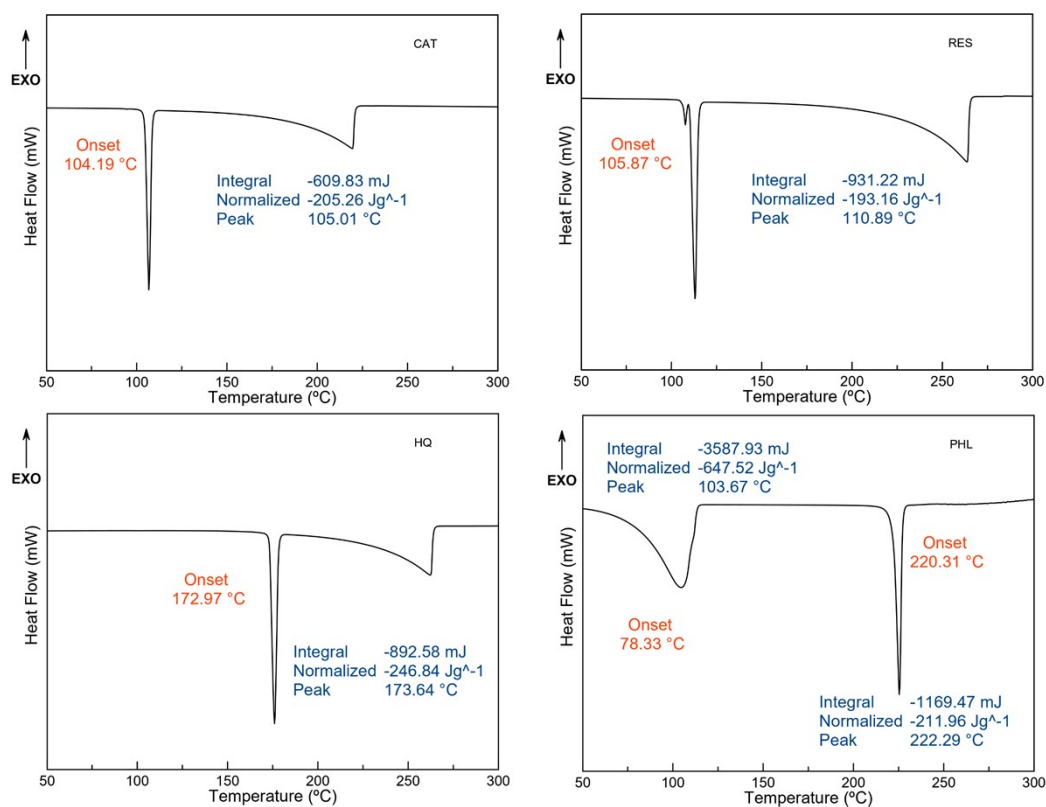


Fig. S6 DSC of isolated coformers CAT, RES, HQ and PHL used in this work with the corresponding onsets and enthalpies values.

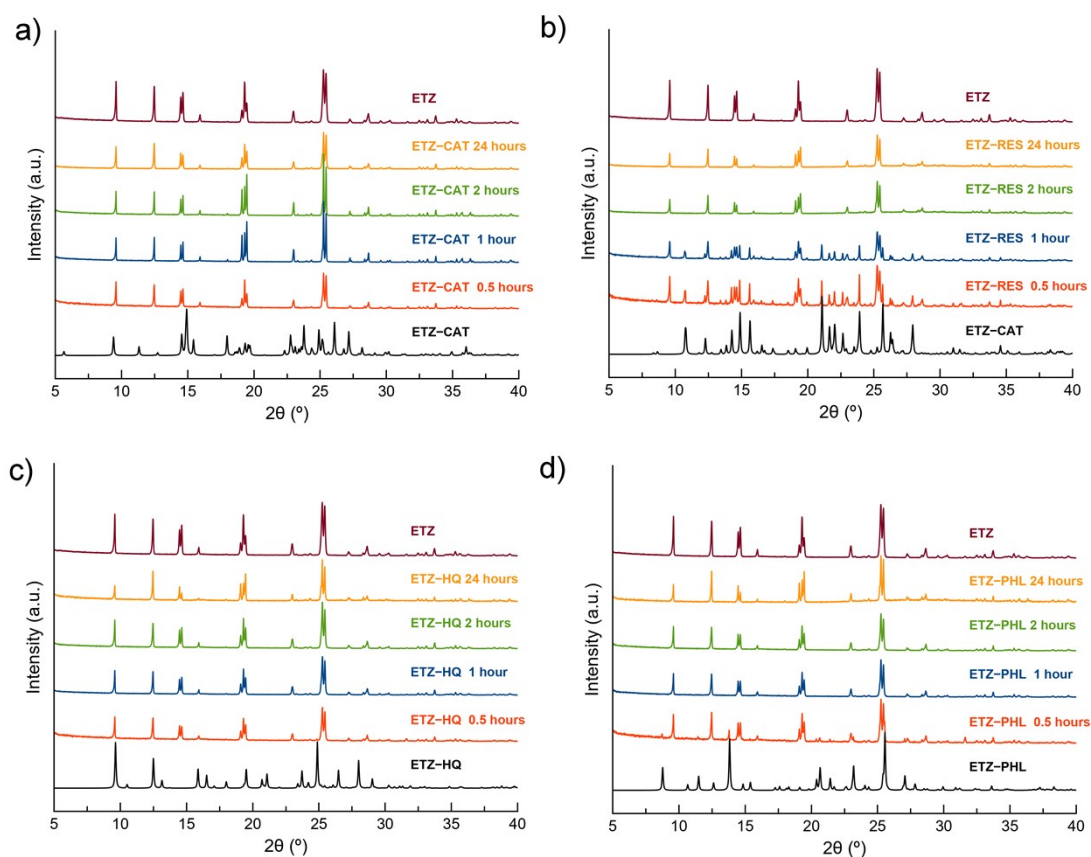


Fig. S7 PXRD patterns of the reported ETZ cocrystals after aqueous slurry experiments (pH 6.8 phosphate buffer medium) for 24 hours.

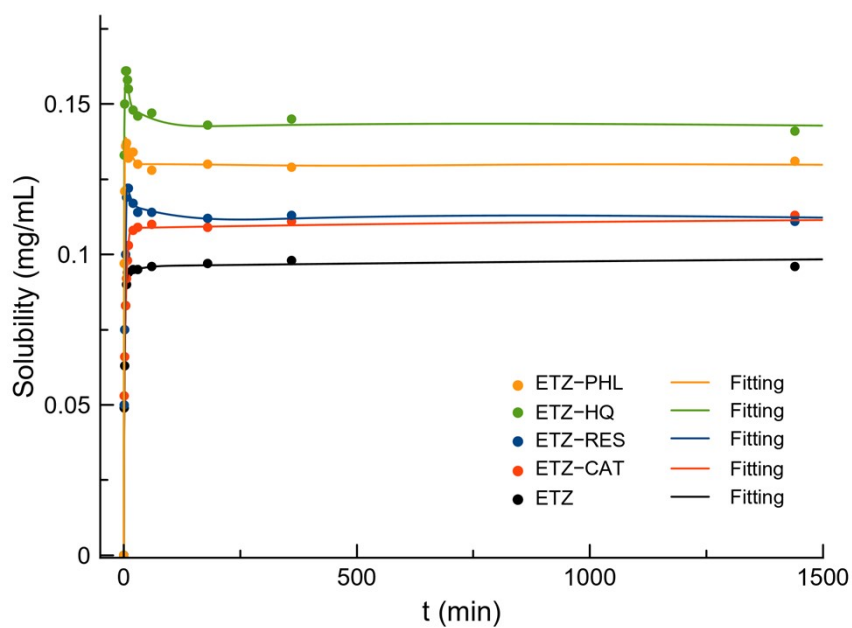


Fig. S8 Dissolution profile of ETZ and cocrystals at 24 h (1440 min).

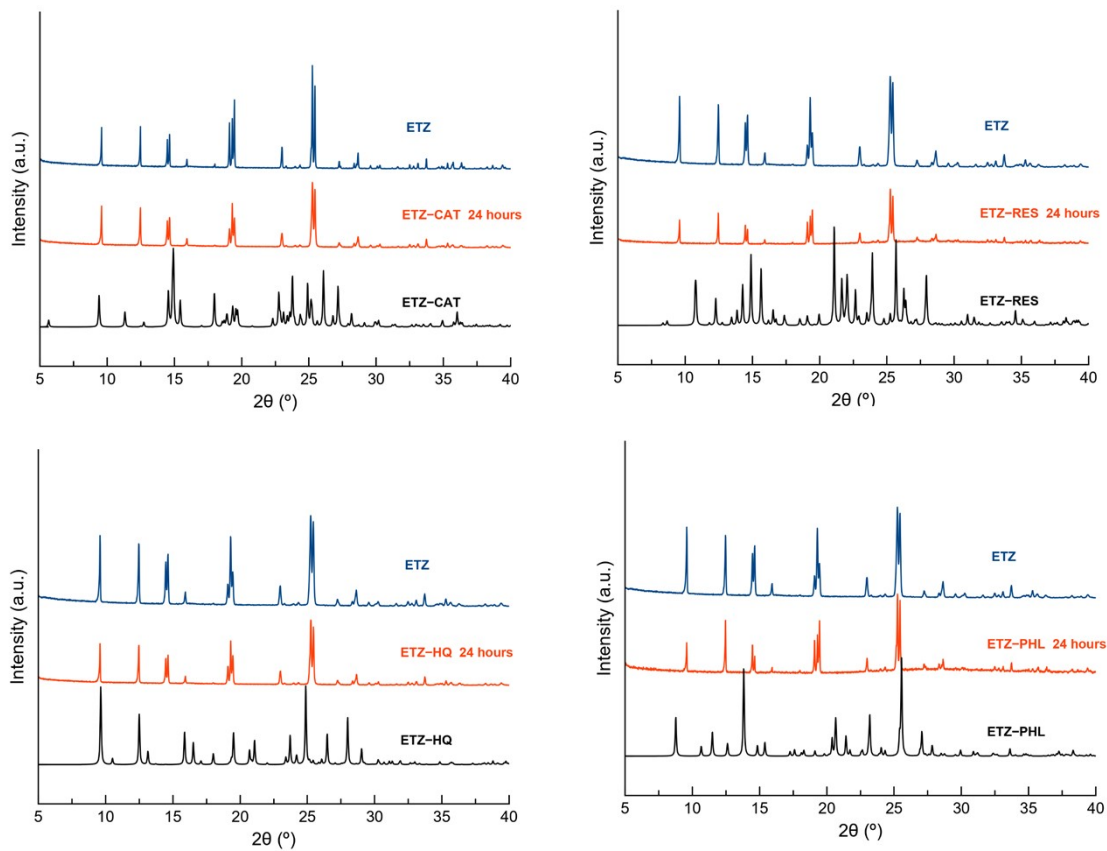


Fig. S9 XRD patterns of the reported ETZ cocrystals after powder dissolution experiments (pH 6.8 phosphate buffer medium, 24 hours).

Table 1. Hydrogen bonds for ETZ cocrystals [\AA and deg.].

ETZ—CAT	D-H \cdots A	d(D-H)	d(H \cdots A)	d(D \cdots A)	<(DHA)
	N(1)-H(1A) \cdots O(3)#1	0.86	2.23	3.084(3)	171.3
	N(1)-H(1B) \cdots O(2)	0.86	1.99	2.649(3)	133.2
	N(2)-H(2A) \cdots O(1)#2	0.86	2.03	2.879(3)	168.1
	N(2)-H(2B) \cdots O(4)	0.86	1.98	2.622(3)	131.0
	O(5)-H(5A) \cdots O(3)	0.82	1.90	2.715(2)	172.1
	O(6)-H(6A) \cdots O(1)	0.82	1.87	2.687(2)	173.0
	Symmetry transformations used to generate equivalent atoms: #1 x-1,y+1,z #2 x+1,y-1,z				
ETZ—RES	D-H \cdots A	d(D-H)	d(H \cdots A)	d(D \cdots A)	<(DHA)
	N(1)-H(1A) \cdots O(3)#1	0.86	2.12	2.9807(15)	176.5
	N(1)-H(1B) \cdots O(2)	0.86	1.97	2.6369(14)	134.0
	N(2)-H(2A) \cdots O(1)#2	0.86	2.17	2.9603(16)	152.8
	N(2)-H(2B) \cdots O(4)	0.86	1.97	2.6418(14)	134.0
	O(5)-H(5A) \cdots O(1)#2	0.82	1.89	2.6886(15)	165.9
	O(6)-H(6A) \cdots O(3)#3	0.82	1.94	2.7550(14)	171.8
	Symmetry transformations used to generate equivalent atoms: #1 x,-y+3/2,z-1/2 #2 x,-y+3/2,z+1/2 #3 -x+2,y-1/2,-z+3/2				
ETZ—HQ	D-H \cdots A	d(D-H)	d(H \cdots A)	d(D \cdots A)	<(DHA)
	N(1)-H(1A) \cdots O(3)#1	0.86	2.18	2.9861(15)	156.2
	N(1)-H(1B) \cdots O(2)	0.86	1.97	2.6488(14)	134.4
	O(3)-H(3A) \cdots O(1)	0.82	1.84	2.6642(13)	178.0
	Symmetry transformations used to generate equivalent atoms: #1 -x,-y+1,-z				
ETZ—PHL	D-H \cdots A	d(D-H)	d(H \cdots A)	d(D \cdots A)	<(DHA)
	N(1)-H(1A) \cdots O(3)#1	0.86	2.08	2.919(4)	165.9
	N(1)-H(1B) \cdots O(2)	0.86	1.94	2.611(4)	134.2
	N(2)-H(2A) \cdots O(1)#2	0.86	2.07	2.933(4)	177.3
	N(2)-H(2B) \cdots O(4)	0.86	1.95	2.623(4)	134.0
	O(5)-H(5A) \cdots O(1)#3	0.82	1.85	2.653(3)	167.7
	O(6)-H(6A) \cdots O(3)#1	0.82	1.86	2.664(3)	168.5
	O(7)-H(7) \cdots O(5)#4	0.82	2.02	2.833(3)	169.4
	Symmetry transformations used to generate equivalent atoms: #1 x+1/2,-y+1,z-1/2 #2 x-1/2,-y+1,z+1/2 #3 x,y-1,z #4 x+1/2,-y,z+1/2				