

Supplementary Information

Structural aspects of dehydration and rehydration in adenosine 5'-diphosphate (ADP)–potassium–water system

Oskar Kaszubowski, Katarzyna Ślepokura

University of Wrocław, Faculty of Chemistry, 14. F. Joliot-Curie, 50-383 Wrocław, Poland

Corresponding author Email: oskar.kaszubowski@uwr.edu.pl

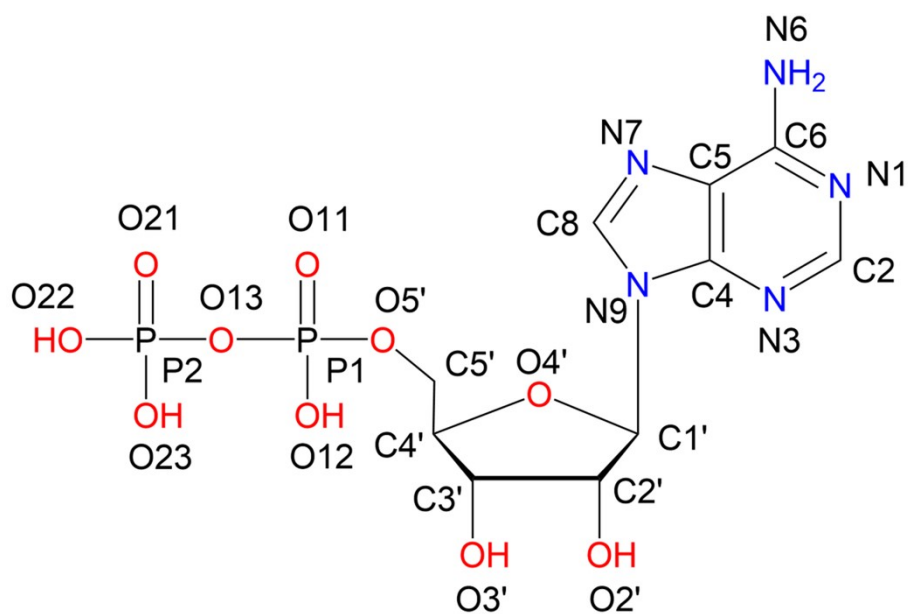


Fig. S1 Atom numbering scheme for reported structures. ADP in neutral form is shown.

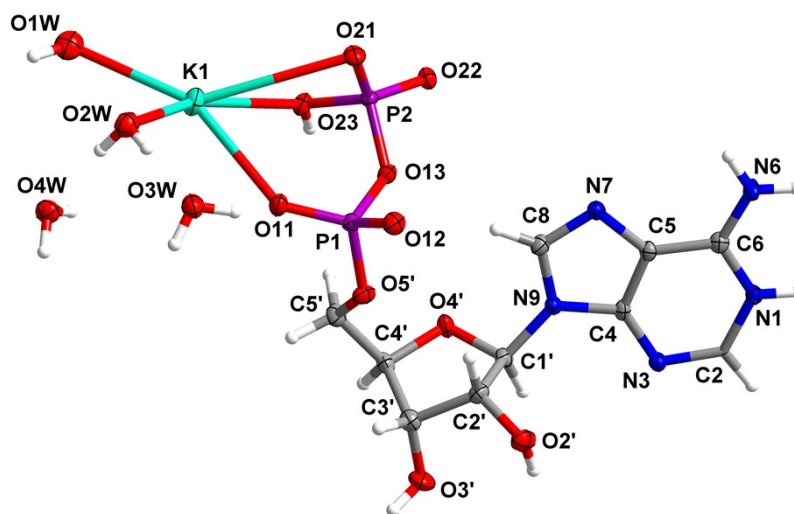


Fig. S2 Asymmetric unit of crystal $K(ADP) \cdot 4H_2O$ (I). Displacement ellipsoids are shown at the 50% probability level.

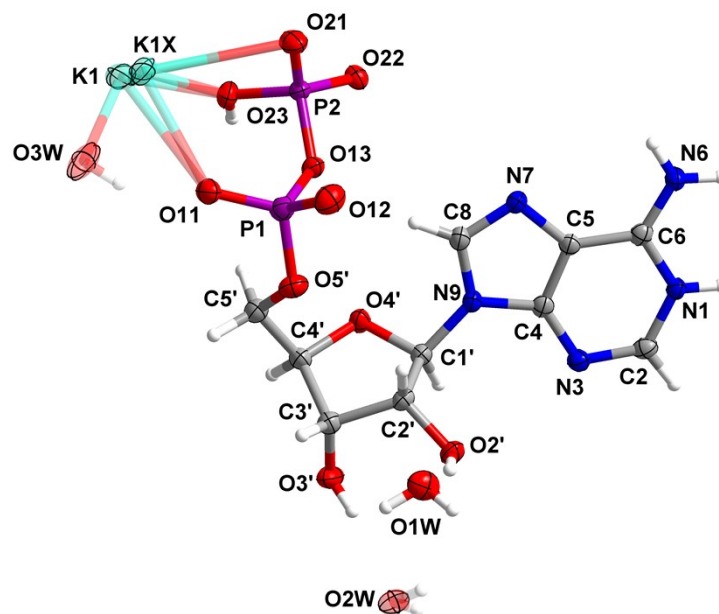


Fig. S3 Asymmetric unit of crystal $\text{K(ADP)} \cdot 2\text{H}_2\text{O}$ (II). Transparent ellipsoids and bonds represent atoms with <1 site occupation factors values. Displacement ellipsoids are shown at the 50% probability level.

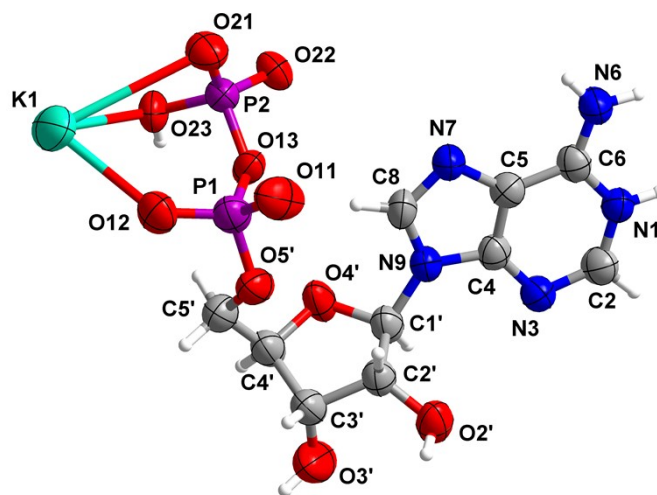


Fig. S4 Asymmetric unit of crystal K(ADP) (III). Displacement ellipsoids are shown at the 50% probability level.

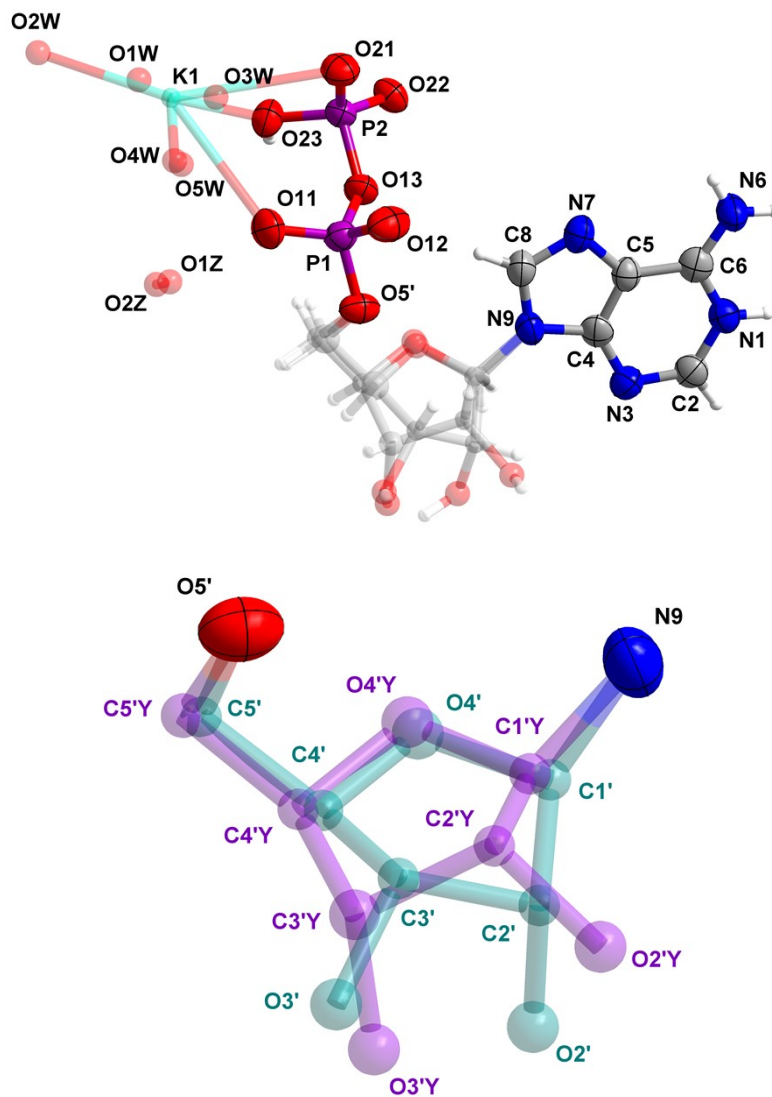


Fig. S5 Asymmetric unit of crystal (H₅O₂)K(ADP)₂·4.25H₂O (IVa). Transparent spheres, ellipsoids and bonds represent atoms with <1 site occupation factors values. Displacement ellipsoids are shown at the 50% probability level. Disordered ribose fragment with atom labels is shown below.

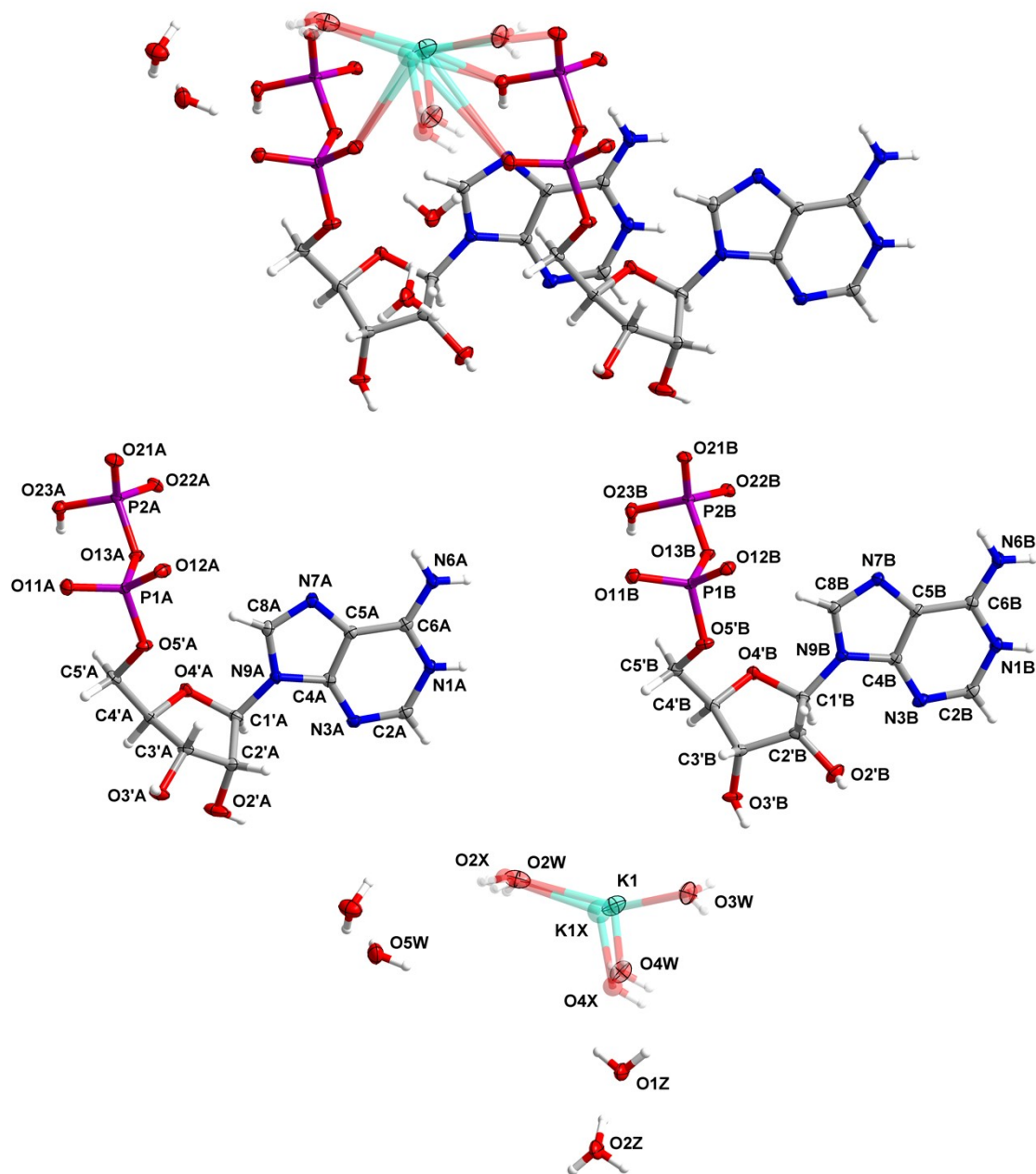


Fig. S6 Asymmetric unit of crystal $(\text{H}_5\text{O}_2)\text{K}(\text{ADP})_2 \cdot 4.25\text{H}_2\text{O}$ (IVb). Transparent spheres, ellipsoids and bonds represent atoms with <1 site occupation factors values. Displacement ellipsoids are shown at the 50% probability level. Two ADP^- anions and K^+ cations, Zundel cation and water molecules with atom labels are shown below.

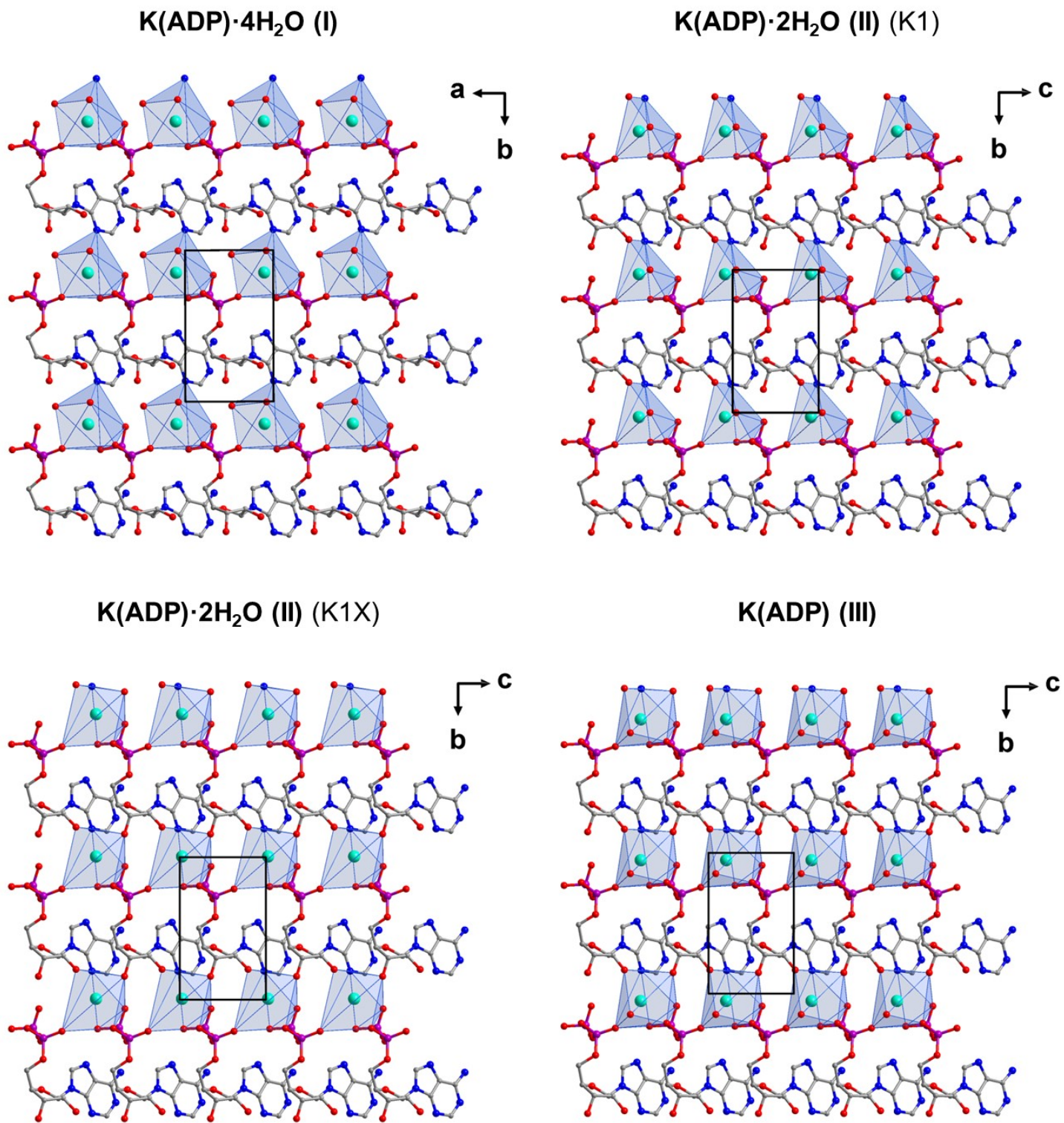


Fig. S7 Comparison of packing diagrams of structures I, II and III. For II, two diagrams with different potassium ions are shown (K1 and K1X; SOF = 0.5). Hydrogen atoms and selected water molecules are omitted for clarity.

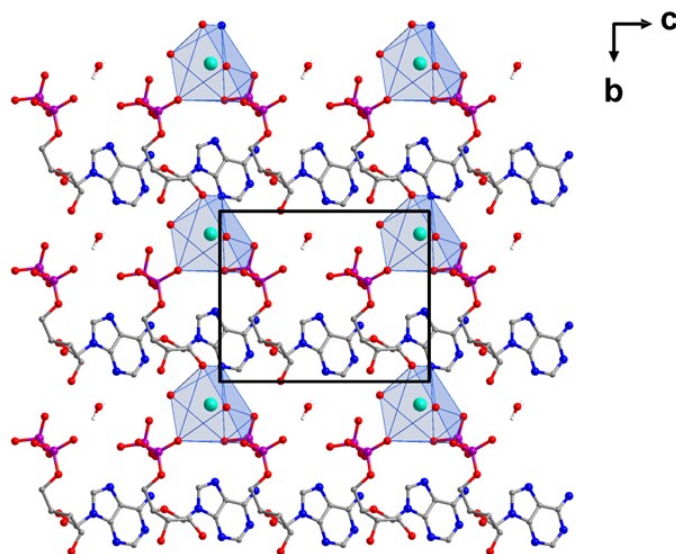
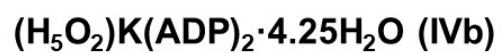
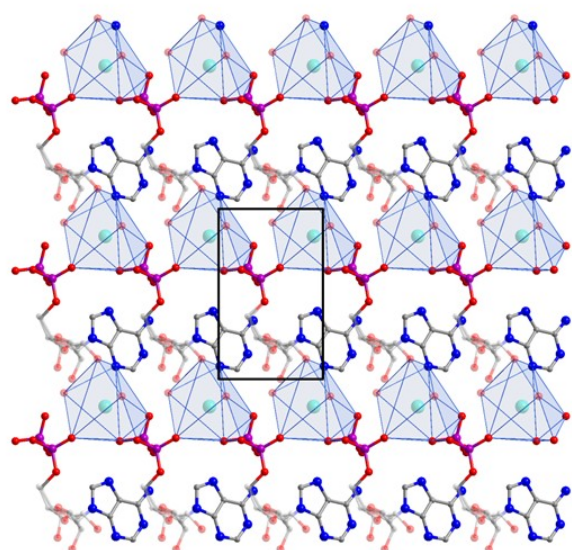
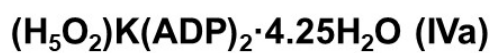


Fig. S8 Comparison of packing diagrams of structures **IVa** and **IVb**. For **IVb**, disorder of potassium cations and water molecules is omitted and diagram for only one potassium ion is shown (K1; SOF = 0.744(7)). Disordered parts in **IVa** are presented with transparent atoms, bonds and polyhedra. Selected water molecules, oxygen atoms of Zundel cation and hydrogen atoms are omitted for clarity.

Table S1 P–O bonds lengths [Å] and selected valence angles and torsion angles values [°] for K(ADP)·4H₂O (**I**), K(ADP)·2H₂O (**II**), K(ADP) (**III**) and (H₅O₂)K(ADP)₂·4.25H₂O (**IVa** and **IVb**).

	I	II	III	IVa		IVb	
				–	Y	A	B
P1–O5'	1.594(3)	1.602(3)	1.600(7)	1.577(6)		1.588(2)	1.585(2)
P1–O11	1.481(3)	1.480(3)	1.469(7)	1.472(6)		1.486(2)	1.490(2)
P1–O12	1.487(3)	1.483(3)	1.490(8)	1.464(6)		1.493(2)	1.493(2)
P1–O13	1.620(3)	1.617(3)	1.601(6)	1.600(6)		1.605(2)	1.6047(19)
P2–O13	1.609(3)	1.604(3)	1.589(6)	1.611(5)		1.617(2)	1.620(2)
P2–O21	1.487(3)	1.482(3)	1.470(7)	1.471(6)		1.486(2)	1.485(2)
P2–O22	1.497(3)	1.499(3)	1.499(6)	1.511(5)		1.505(2)	1.503(2)
P2–O23	1.570(3)	1.575(3)	1.572(6)	1.569(5)		1.575(2)	1.563(2)
O5'–P1–O11	112.81(17)	111.99(18)	107.0(4)	112.9(4)		112.56(12)	112.29(12)
O5'–P1–O12	103.80(16)	106.47(19)	111.5(4)	105.2(3)		105.34(12)	105.95(12)
O5'–P1–O13	100.27(15)	99.26(15)	99.0(4)	100.5(3)		99.84(11)	100.58(11)
O11–P1–O12	119.71(16)	119.16(19)	118.8(5)	117.9(4)		118.64(13)	117.43(12)
O11–P1–O13	109.48(15)	108.96(16)	109.5(4)	109.8(3)		109.75(12)	109.55(12)
O12–P1–O13	108.94(16)	109.15(18)	109.3(4)	109.1(4)		109.00(12)	109.68(12)
O13–P2–O21	110.36(16)	110.11(17)	111.0(4)	110.5(3)		111.11(12)	110.27(11)
O13–P2–O22	103.87(15)	104.93(16)	104.8(3)	103.1(3)		102.92(11)	103.08(11)
O13–P2–O23	105.51(16)	104.32(16)	104.0(4)	105.0(3)		104.47(11)	105.33(11)
O21–P2–O22	117.84(16)	117.43(17)	117.8(4)	117.2(3)		117.55(12)	116.58(12)
O21–P2–O23	108.00(16)	108.71(18)	108.5(4)	110.2(3)		109.17(13)	109.43(12)
O22–P2–O23	110.57(16)	110.52(17)	109.9(4)	109.9(3)		110.73(12)	111.41(12)
P1–O13–P2	130.09(17)	131.56(18)	133.3(4)	131.6(4)		131.07(13)	131.52(13)
O4'–C1'–N9–C4	–144.4(3)	–149.6(4)	–147.8(8)	–154.5(8)	–150.6(9)	–157.8(2)	–152.7(2)
O5'–C5'–C4'–O4'	–70.2(4)	–63.7(4)	–62.3(11)	–82.6(13)	–61.4(11)	–68.9(3)	–61.7(3)
O5'–C5'–C4'–C3'	48.8(4)	57.1(5)	57.4(11)	43.9(14)	65.6(12)	49.3(3)	60.3(3)
P1–O5'–C5'–C4'	148.2(3)	143.7(3)	142.3(7)	151.3(6)	150.4(6)	157.68(18)	143.62(18)
P2–O13–P1–O5'	148.6(2)	156.6(2)	157.3(6)	162.2(5)	162.2(5)	166.06(17)	155.39(17)
O13–P1–O5'–C5'	–62.9(3)	–64.2(3)	–61.9(8)	–64.5(10)	–71.5(8)	–73.8(2)	–64.9(2)

Table S2 Pseudorotation parameters (P , τ_m) and Cremer & Pople puckering parameters (q_2 , ϕ_2) for K(ADP)·4H₂O (I), K(ADP)·2H₂O (II), K(ADP) (III) and (H₅O₂)K(ADP)₂·4.25H₂O (IVa and IVb).

	I	II	III	IVa		IVb	
				–	Y	A	B
P [°]	158.3(4)	165.1(4)	165.9(11)	1.8(11)	159.2(12)	18.4(3)	147.7(3)
τ_m [°]	40.9(3)	38.0(3)	31.9(6)	37.4(7)	37.9(8)	36.0(2)	33.2(2)
q_2 [Å]	0.397(4)	0.371(4)	0.305(10)	0.362(11)	0.363(12)	0.353(3)	0.315(3)
ϕ_2 [°]	70.2(6)	77.0(6)	77.4(18)	271.8(18)	71.0(18)	286.9(5)	60.6(5)

Table S3 Geometric parameters of the hydrogen bonds for K(ADP)·4H₂O (I).

$D-H\cdots A$	$D-H$ [Å]	$H\cdots A$ [°]	$D\cdots A$ [Å]	$D-H\cdots A$ [°]
O23–H23 ⁱ …N7 ⁱ	0.84	1.85	2.688(4)	175
O2'–H2O ^{iv} …O2W ^{iv}	0.84	1.94	2.730(4)	157
O3'–H3O ^v …O2 ^v	0.84	2.53	3.201(4)	137
O3'–H3O ^{vi} …O3W ^{vi}	0.84	2.21	2.805(4)	128
N1–H1 ^{vii} …O22 ^{vii}	0.88	1.80	2.658(4)	165
N6–H61 ^{vii} …O21 ^{vii}	0.88	2.03	2.899(4)	171
N6–H62 ^{viii} …O22 ^{viii}	0.88	2.07	2.832(4)	144
C1'–H1' ^{iv} …O1W ^{iv}	1.00	2.21	3.149(5)	156
C2'–H2' ^{ix} …O3W ^{ix}	1.00	2.52	3.285(5)	133
C3'–H3' ^v …O2' ^v	1.00	2.48	3.227(5)	131
C4'–H4' ⁱ …O2' ⁱ	1.00	2.46	3.225(5)	133
C2–H2 ^{iv} …O21 ^{iv}	0.95	2.34	3.276(5)	169
C8–H8 ^{iv} …O13	0.95	2.43	3.367(5)	169
O1W–H1W ⁱ …O21 ⁱ	0.84	1.99	2.778(4)	157
O1W–H2W ⁱⁱ …O3' ⁱⁱ	0.84	2.17	2.924(5)	149
O2W–H3W ^{iv} …O3W	0.84	1.96	2.802(5)	177
O3W–H5W ^{viii} …O4W ^{viii}	0.84	1.90	2.721(4)	165
O3W–H6W ^x …O12 ^x	0.84	2.07	2.895(4)	166
O4W–H7W ⁱ …O12 ⁱ	0.84	1.94	2.764(4)	168
O4W–H8W ^x …O11 ^x	0.84	1.95	2.753(4)	161

Symmetry codes: ⁽ⁱ⁾ $x+1, y, z$; ⁽ⁱⁱ⁾ $x+1, y-1, z$; ^(iv) $x-1, y+1, z$; ^(v) $x+1/2, -y+3/2, -z+1$; ^(vi) $x, y+1, z$; ^(vii) $-x, y+1/2, -z+1/2$; ^(viii) $x-1, y, z$; ^(ix) $x-1/2, -y+1/2, -z+1$; ^(x) $x+1/2, -y+1/2, -z+1$.

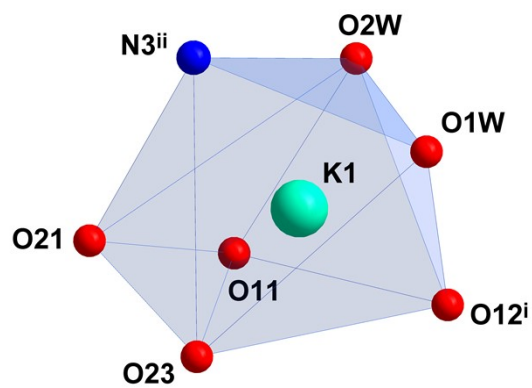


Fig. S9 Coordination polyhedron of potassium ion in K(ADP)·4H₂O (I). Symmetry codes are given in Table S4.

Table S4 K–O and K–N bonds lengths [Å] in K(ADP)·4H₂O (I).

K1–O11	2.731(3)
K1–O12 ⁱ	2.805(3)
K1–O21	3.305(3)
K1–O23	2.816(3)
K1–N3 ⁱⁱ	3.593(4)
K1–O1W	2.731(4)
K1–O2W	2.753(3)

Symmetry codes: ⁽ⁱ⁾ $x+1, y, z$; ⁽ⁱⁱ⁾ $x+1, y-1, z$.

Table S5 Geometric parameters of the hydrogen bonds for K(ADP)·2H₂O (II).

<i>D</i> -H... <i>A</i>	<i>D</i> -H [Å]	H... <i>A</i> [°]	<i>D</i> ... <i>A</i> [Å]	<i>D</i> -H... <i>A</i> [°]
O23-H23...N7 ⁱ	0.84	1.96	2.794(5)	169
O2'-H2O...O1W	0.84	1.80	2.610(4)	162
O3'-H3O...O2W	0.84	1.95	2.770(8)	164
O3'-H3O...O2W ^{iv}	0.84	1.99	2.803(8)	161
N1-H1...O22 ^v	0.88	1.81	2.662(4)	161
N6-H61...O21 ^v	0.88	2.07	2.921(5)	163
N6-H62...O22 ^{vi}	0.88	2.00	2.758(5)	143
C4'-H4'...O2' ⁱ	1.00	2.58	3.519(5)	157
C2-H2...O21 ^{viii}	0.95	2.21	3.063(5)	149
C8-H8...O13	0.95	2.25	3.186(5)	168
O1W-H1W...O11 ^{ix}	0.84	1.91	2.709(4)	159
O1W-H2W...O12 ^x	0.84	1.90	2.725(5)	168
O2W-H4W...O12 ^{xi}	0.84	1.98	2.814(7)	176
O2W-H3W...O3W ^{viii}	0.84	2.08	2.845(8)	151
O2W-H3W...O3W ^{ix}	0.84	2.08	2.845(8)	151
O3W-H5W...O11	0.84	2.36	3.082(3)	145

Symmetry codes: ⁽ⁱ⁾ $x, y, z-1$; ^(iv) $-x+1, -y+2, z$; ^(v) $-x+1/2, y+1/2, -z+2$; ^(vi) $x, y, z+1$; ^(viii) $x, y+1, z+1$; ^(ix) $-x+1, -y+1, z+1$; ^(x) $-x+1, -y+1, z$; ^(xi) $x, y+1, z$.

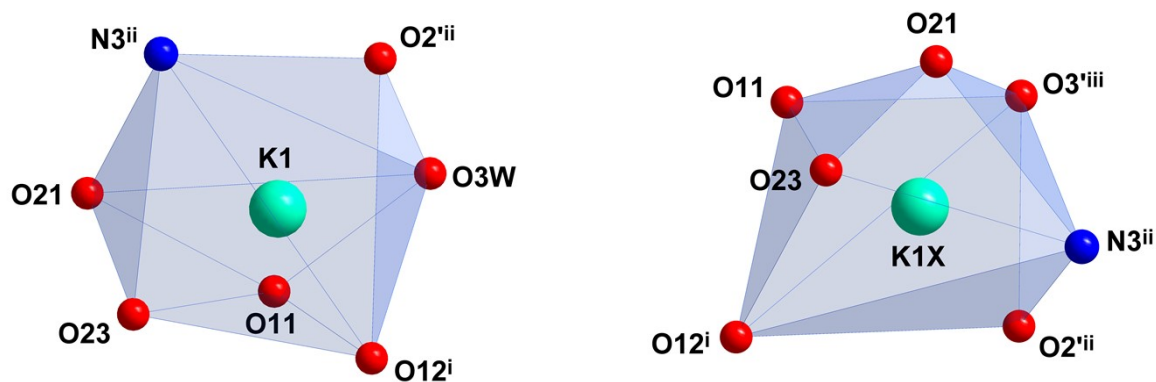
**Fig. S10** Coordination polyhedra of potassium ions in K(ADP)·2H₂O (II). Symmetry codes are given in Table S6.

Table S6 K–O and K–N bonds lengths [Å] in K(ADP)·2H₂O (II).

K1–O11	2.919(4)	K1X–O11	2.654(4)
K1–O12 ⁱ	2.821(5)	K1X–O12 ⁱ	3.506(5)
K1–O21	3.397(4)	K1X–O21	3.068(4)
K1–O23	2.789(4)	K1X–O23	3.122(4)
K1–O2 ⁱⁱ	2.881(4)	K1X–O2 ⁱⁱ	2.751(4)
K1–N3 ⁱⁱ	3.206(4)	K1X–O3 ⁱⁱⁱ	2.915(5)
K1–O3W	3.226(4)	K1X–N3 ⁱⁱ	3.122(4)

Symmetry codes: ⁽ⁱ⁾ $x, y, z-1$; ⁽ⁱⁱ⁾ $x, y-1, z-1$; ⁽ⁱⁱⁱ⁾ $x, y-1, z$.

Table S7 Geometric parameters of the hydrogen bonds for K(ADP) (III).

$D-H\cdots A$	$D-H$ [Å]	$H\cdots A$ [°]	$D\cdots A$ [Å]	$D-H\cdots A$ [°]
O23–H23 \cdots N7 ⁱ	0.82	2.01	2.811(11)	167
O2 ^l –H2O \cdots O12 ^v	0.82	1.91	2.692(10)	158
O3 ^l –H3O \cdots O11 ⁱⁱⁱ	0.82	2.21	2.918(13)	145
N1–H1 \cdots O22 ^{vi}	0.86	1.86	2.686(9)	161
N6–H61 \cdots O21 ^{vi}	0.86	2.09	2.912(10)	159
N6–H61 \cdots O22 ^{vi}	0.86	2.61	3.257(9)	133
N6–H62 \cdots O22 ^{vii}	0.86	2.00	2.744(10)	144
C2–H2 \cdots O21 ^{viii}	0.93	2.21	3.012(12)	145
C8–H8 \cdots O13	0.93	2.25	3.156(11)	165

Symmetry codes: ⁽ⁱ⁾ $x, y, z-1$; ⁽ⁱⁱⁱ⁾ $-x+3/2, -y+1, z-1/2$; ^(v) $-x+3/2, -y+1, z+1/2$; ^(vi) $-x+1, y+1/2, -z+5/2$; ^(vii) $x, y, z+1$; ^(viii) $x, y+1, z+1$.

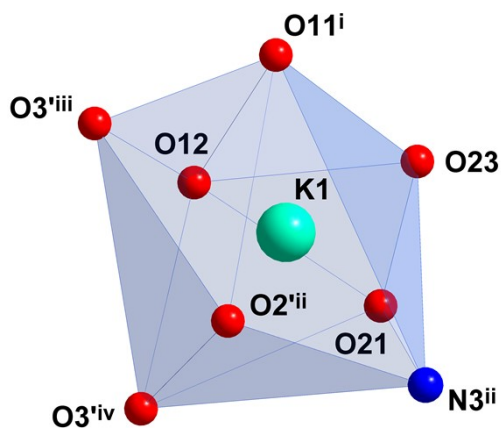
**Fig. S11** Coordination polyhedron of potassium ion in K(ADP) (III). Symmetry codes are given in Table S8.

Table S8 K–O and K–N bonds lengths [Å] in K(ADP) (III).

K1–O11 ⁱ	2.798(8)
K1–O12	2.694(8)
K1–O21	3.365(8)
K1–O23	2.886(7)
K1–O2 ⁱⁱ	2.792(8)
K1–O3 ⁱⁱⁱ	3.413(10)
K1–O3 ^{iv}	3.525(11)
K1–N3 ⁱⁱ	3.165(8)

Symmetry codes: ⁽ⁱ⁾ $x, y, z-1$; ⁽ⁱⁱ⁾ $x, y-1, z-1$; ⁽ⁱⁱⁱ⁾ $-x+3/2, -y+1, z-1/2$; ^(iv) $x, y-1, z$.

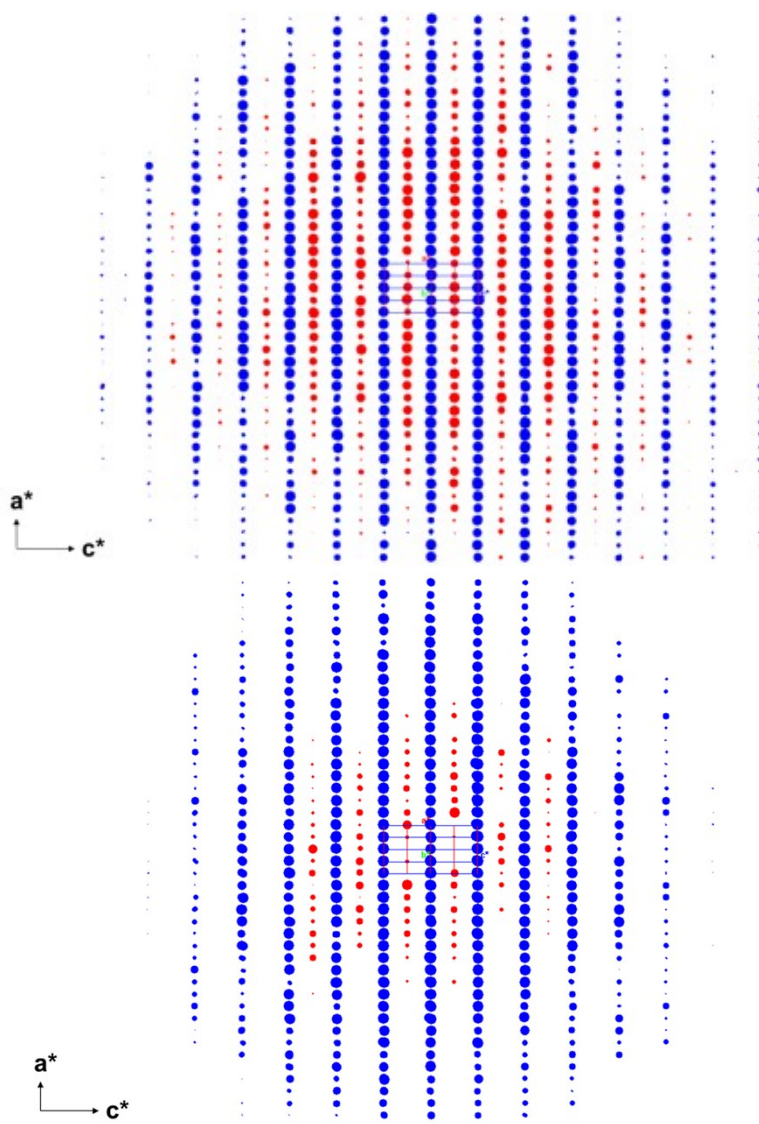


Fig. S12 Diffraction pattern of $(\text{H}_5\text{O}_2)\text{K}(\text{ADP})_2 \cdot 4.25\text{H}_2\text{O}$ at 100 K (top) and 280 K (bottom). Projection of the Ewald sphere along the b^* axis. Visible superstructure reflections (red) in the direction of the c^* axis. Visible reduction in the number of weak reflections with increasing temperature.

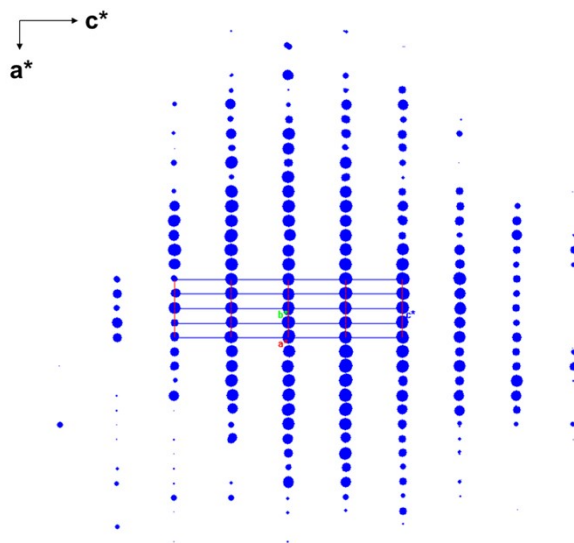


Fig. S13 Diffraction pattern of $(\text{H}_5\text{O}_2)\text{K}(\text{ADP})_2 \cdot 4.25\text{H}_2\text{O}$ at 300 K without superstructure reflections. Projection of the Ewald sphere along the b^* axis.

Table S9 Geometric parameters of the hydrogen bonds for $(\text{H}_5\text{O}_2)\text{K}(\text{ADP})_2 \cdot 4.25\text{H}_2\text{O}$ (**IVb**).

$D\cdots H\cdots A$	$D\cdots H$ [Å]	$H\cdots A$ [°]	$D\cdots A$ [Å]	$D\cdots H\cdots A$ [°]
O23A–H23A \cdots N7B	0.85(4)	1.88(4)	2.721(3)	173(3)
O2'A–H2OA \cdots O1W ^{iv}	0.84	1.92	2.726(4)	160
O3'A–H3OA \cdots O12B ^v	0.84	1.87	2.708(3)	176
N1A–H1A \cdots O22B ^{vii}	0.81(4)	1.86(4)	2.675(3)	178(3)
N6A–H61A \cdots O21B ^{vii}	0.88	1.97	2.848(3)	172
N6A–H62A \cdots O22B ^{viii}	0.88	1.95	2.763(3)	152
C2'A–H2'A \cdots O4X ^v	1.00	2.46	3.259(17)	136
C2A–H2A \cdots O21B ^{iv}	0.95	2.24	3.170(4)	167
O23B–H23B \cdots N7A ⁱⁱⁱ	0.78(4)	1.93(4)	2.699(3)	169(4)
O2'B–H2OB \cdots O5W ^v	0.84	1.85	2.676(3)	169
O3'B–H3OB \cdots O2'A ^{vi}	0.84	2.06	2.831(3)	153
O3'B–H3OB \cdots O3'A ^{vi}	0.84	2.26	2.784(3)	120
N1B–H1B \cdots O22A ^{vii}	0.84(4)	1.84(4)	2.670(3)	172(4)
N6B–H61B \cdots O21A ^{vii}	0.88	1.99	2.860(3)	171
N6B–H62B \cdots O22A	0.88	2.00	2.816(3)	155
C1'B–H1'B \cdots O2W ^{ix}	1.00	2.18	3.082(9)	149
C1'B–H1'B \cdots O2X ^{ix}	1.00	2.20	3.04(3)	141
C2B–H2B \cdots O21A ^{ix}	0.95	2.36	3.268(4)	160
C8B–H8B \cdots O13B	0.95	2.57	3.486(4)	163

O1W-H1W...O12A ⁱⁱⁱ	0.84	2.08	2.782(3)	140
O1W-H2W...N3A ^x	0.84	2.37	3.060(3)	140
O2W-H3W...O3'B ⁱ	0.84	2.07	2.867(8)	158
O2W-H4W...O21B	0.84	1.91	2.693(8)	154
O2X-H3X...O3'B ⁱ	0.84	1.83	2.64(3)	161
O2X-H4X...O21B	0.84	2.03	2.75(3)	143
O3W-H5W...O21A	0.84	1.85	2.618(9)	151
O3W-H6W...O2'A ⁱ	0.84	1.83	2.500(10)	136
O4W-H7W...O3'B ^v	0.84	2.40	3.051(6)	135
O4W-H7W...O5W ^{xii}	0.84	2.41	3.121(6)	143
O4W-H8W...O11A	0.84	2.06	2.814(6)	150
O4X-H7X...O5W ^{xii}	0.84	2.42	3.244(18)	166
O4X-H8X...O11A	0.84	2.30	2.868(18)	125
O5W-H9W...O1W	0.84	2.00	2.808(4)	162
O5W-H10W...O3'A ^{xi}	0.84	2.23	3.040(3)	163
O1Z-H1Z...O11A	0.84	1.84	2.673(3)	173
O1Z-H2Z...O12B	0.84	1.85	2.669(3)	166
O2Z-H12Z...O1Z	1.01(6)	1.49(6)	2.487(4)	168(6)
O2Z-H3Z...O11B ^v	0.84	1.67	2.507(3)	172
O2Z-H4Z...O12A ^{xi}	0.84	1.68	2.520(3)	176

Symmetry codes: (i) $x, y-1, z$; (iii) $x, y, z-1$; (iv) $x, y+1, z+1$; (v) $-x+3/2, -y+1, z+1/2$; (vi) $-x+3/2, -y+2, z-1/2$; (vii) $-x+1, y+1/2, -z+1/2$; (viii) $x, y, z+1$; (ix) $x, y+1, z$; (x) $x, y-1, z-1$; (xi) $-x+3/2, -y+1, z-1/2$

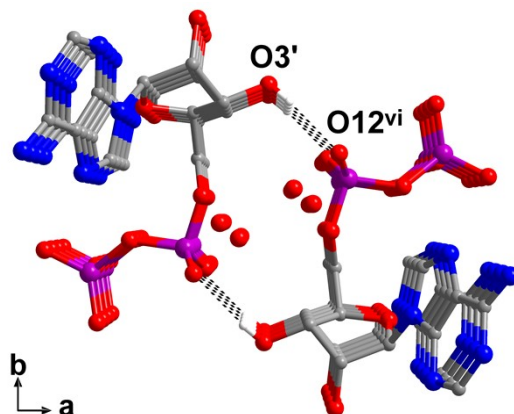


Fig. S14 Location of the disordered Zundel cation oxygen atoms between the bilayers in crystal **IVa**. For clarity, the atoms of one of the positions of the disordered sugar ring (Y) and most of the hydrogen atoms are omitted. Selected hydrogen bonds are marked with black dotted lines. Symmetry codes are given in Table S11.

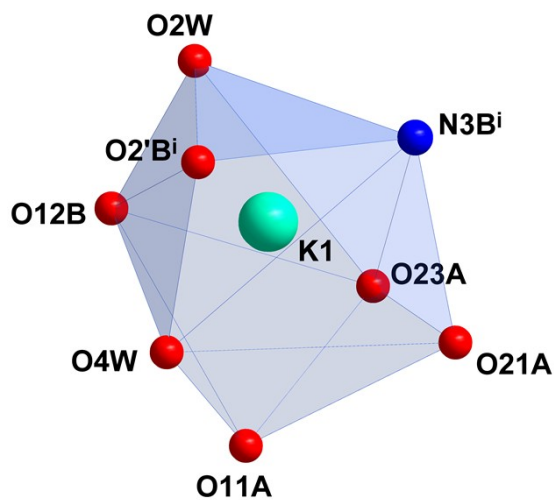


Fig. S15 Coordination polyhedron of potassium ion in $(\text{H}_5\text{O}_2)\text{K}(\text{ADP})_2 \cdot 4.25\text{H}_2\text{O}$ (**IVb**). Symmetry codes are given in Table S10.

Table S10 K–O and K–N bonds lengths [Å] in (H₅O₂)K(ADP)₂·4.25H₂O (**IVb**).

K1–O11A	3.540(3)
K1–O21A	3.373(4)
K1–O23A	3.103(3)
K1–O12B	3.308(3)
K1–O2'B ⁱ	2.949(3)
K1–N3B ⁱ	3.077(3)
K1–O2W	2.671(9)
K1–O4W	2.691(5)

Symmetry codes: ⁽ⁱ⁾ $x, y-1, z$.**Table S11** Geometric parameters of the hydrogen bonds for (H₅O₂)K(ADP)₂·4.25H₂O (**IVa**).

<i>D</i> –H⋯ <i>A</i>	<i>D</i> –H [Å]	H⋯ <i>A</i> [°]	<i>D</i> ⋯ <i>A</i> [Å]	<i>D</i> –H⋯ <i>A</i> [°]
O23–H23⋯N7 ⁱ	0.82	1.92	2.720(9)	164
O2'–H2O⋯O3'Y ^{iv}	0.82	2.19	2.94(18)	152
O3'–H3O⋯O12 ^{vi}	0.82	2.04	2.815(13)	159
N1–H1⋯O22 ^{viii}	0.86	1.86	2.694(8)	163
N6–H61⋯O21 ^{viii}	0.86	2.01	2.868(9)	172
N6–H62⋯O22 ^{ix}	0.86	2.02	2.795(9)	150
C2–H2⋯O21 ^{vii}	0.93	2.36	3.262(10)	164
O2'Y–H2OY⋯O4W ^v	0.82	1.72	2.43(4)	144
O2'Y–H2OY⋯O5W ^v	0.82	1.91	2.68(4)	158
O3'Y–H3OY⋯O2W ^{vii}	0.82	2.17	2.80(3)	133
C1'Y–H1'Y⋯O2W ^{vii}	0.98	2.23	3.16(3)	157

Symmetry codes: ⁽ⁱ⁾ $x, y, z-1$; ^(iv) $-x+1, -y+2, z$; ^(v) $-x+1, -y+1, z+1$; ^(vi) $-x+1, -y+1, z$; ^(vii) $x, y+1, z+1$; ^(viii) $-x+1/2, y+1/2, -z+2$; ^(ix) $x, y, z+1$.

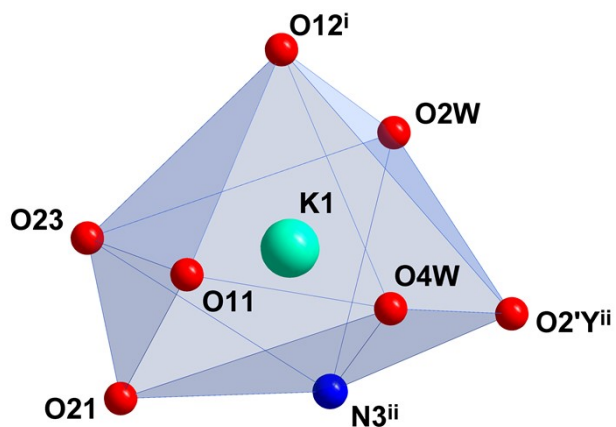


Fig. S16 Coordination polyhedron of potassium ion in $(\text{H}_5\text{O}_2)\text{K}(\text{ADP})_2 \cdot 4.25\text{H}_2\text{O}$ (**IVa**). Symmetry codes are given in Table S12.

Table S12 K–O and K–N bonds lengths [\AA] in $(\text{H}_5\text{O}_2)\text{K}(\text{ADP})_2 \cdot 4.25\text{H}_2\text{O}$ (**IVa**).

K1–O11	3.225(12)
K1–O12 ⁱ	3.111(12)
K1–O21	3.44(12)
K1–O23	3.087(11)
K1–O2' ^{Yii}	3.53(19)
K1–N3 ⁱⁱ	3.178(12)
K1–O2W	2.81(2)
K1–O4W	2.71(4)

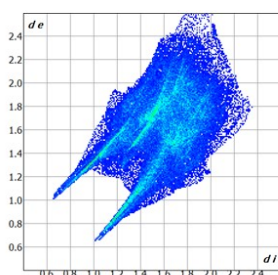
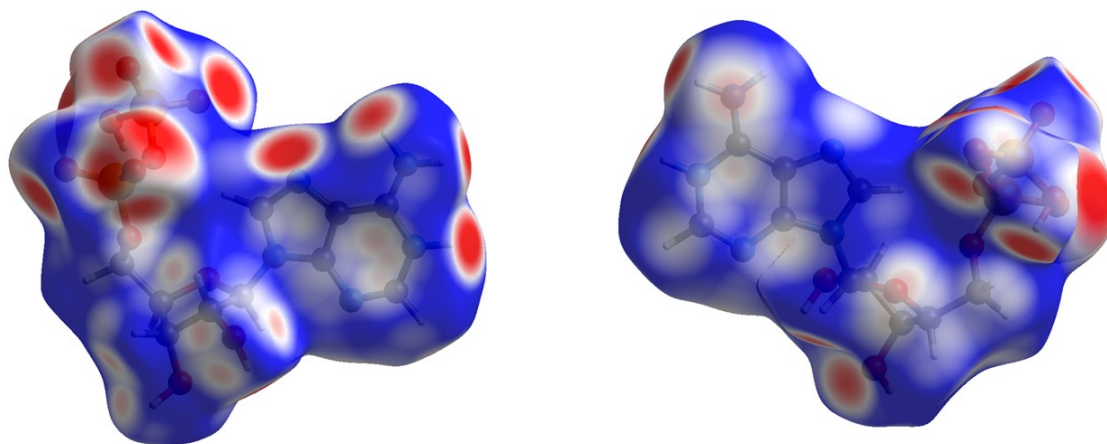
Symmetry codes: ⁽ⁱ⁾ $x, y, z-1$; ⁽ⁱⁱ⁾ $x, y-1, z-1$.

Table S13 Values of deformation parameters of the analyzed coordination polyhedra from idealized shapes calculated in the *Shape* program for crystals K(ADP)·4H₂O (**I**), K(ADP)·2H₂O (**II**), K(ADP) (**III**) and (H₅O₂)K(ADP)₂·4.25H₂O (**IVa** and **IVb**).

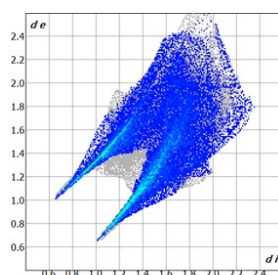
Idealized shape and symmetry	HP-7 ^[a] (D _{7h})	HPY-7 ^[b] (C _{6v})	PBPY-7 ^[c] (D _{5h})	COC-7 ^[d] (C _{3v})	CTPR-7 ^[e] (C _{2v})	JPBPY-7 ^[f] (D _{5h})	JETPY-7 ^[g] (C _{3v})	
K(ADP)·4H ₂ O (I)								
K1	25.317	17.918	7.628	7.823	5.492	9.193	14.940	
K(ADP)·2H ₂ O (II)								
K1	23.025	15.798	12.868	9.094	8.029	15.258	11.426	
K1X	31.098	20.394	8.360	4.623	3.154	10.580	17.189	
Idealized shape and symmetry	JETBPY-8 ^[h] (D _{3h})	JGBF-8 ^[i] (D _{2d})	TDD-8 ^[j] (D _{2d})	SAPR-8 ^[k] (D _{4d})	CU-8 ^[l] (O _h)	HBPY-8 ^[m] (D _{6h})	HPY-8 ^[n] (C _{7v})	OP-8 ^[o] (D _{8h})
K(ADP) (III)								
K1	23.406	12.655	5.429	7.930	14.348	16.059	19.718	26.577
(H ₅ O ₂)K(ADP) ₂ ·4.25H ₂ O (IVa)								
K1	19.401	12.832	9.272	9.578	16.271	16.753	16.320	30.287
(H ₅ O ₂)K(ADP) ₂ ·4.25H ₂ O (IVb)								
K1	17.463	11.945	9.440	11.647	18.950	18.038	18.537	30.375
Idealized shapes:				^[h] JETBPY-8 – Johnson elongated triangular bipyramid J14				
^[a] HP-7 – Heptagon				^[i] JGBF-8 – Johnson gyrobifastigium J26				
^[b] HPY-7 – Hexagonal pyramid				^[j] TDD-8 – Triangular dodecahedron				
^[c] PBPY-7 – Pentagonal bipyramid				^[k] SAPR-8 – Square antiprism				
^[d] COC-7 – Capped octahedron				^[l] CU-8 – Cube				
^[e] CTPR-7 – Capped trigonal prism				^[m] HBPY-8 – Hexagonal bipyramid				
^[f] JPBPY-7 – Johnson pentagonal bipyramid J13				^[n] HPY-8 – Heptagonal pyramid				
^[g] JETPY-7 – Johnson elongated triangular pyramid J7				^[o] OP-8 – Octagon				

Table S14 Lengths of $\text{Ip}\cdots\pi$ ($\text{Ip}\cdots\text{centroid}$ and $\text{Ip}\cdots\text{plane}$) interactions of oxygen atoms and adenine six- and five-membered rings.

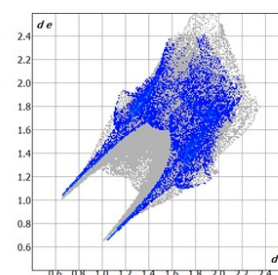
	Six-membered ring		Five-membered ring	
	$\text{Ip}\cdots\text{centroid}$ [Å]	$\text{Ip}\cdots\text{plane}$ [Å]	$\text{Ip}\cdots\text{centroid}$ [Å]	$\text{Ip}\cdots\text{plane}$ [Å]
K(ADP)·4H₂O (I)				
O4 ⁱ	2.947(3)	2.885(3)	–	–
O22 ⁱⁱ	3.473(3)	3.117(3)	3.511(3)	3.118(3)
O23 ⁱⁱ	3.207(3)	3.048(3)	–	–
K(ADP)·2H₂O (II)				
O4 ^{riii}	2.864(3)	2.826(3)	–	–
O22 ^{iv}	3.422(3)	3.174(3)	3.541(3)	3.142(3)
O23 ^{iv}	3.378(3)	3.133(3)	–	–
K(ADP) (III)				
O4 ^{riii}	2.950(6)	2.896(5)	–	–
O22 ^v	3.499(6)	3.310(6)	3.718(6)	3.299(6)
O23 ^v	3.475(8)	3.188(7)	–	–
(H₅O₂)K(ADP)₂·4.25H₂O (IVa)				
O4 ^{riii}	3.014(7)	2.878(8)	–	–
O4 ^{Yiii}	3.161(8)	3.029(8)	–	–
O22 ^{iv}	3.612(5)	3.115(5)	3.553(5)	3.092(5)
O23 ^{iv}	3.185(5)	3.108(5)	–	–
(H₅O₂)K(ADP)₂·4.25H₂O (IVb) (A)				
O4 ^{Biii}	3.151(2)	2.904(2)	–	–
O22A ⁱⁱ	3.473(2)	2.945(2)	3.274(2)	2.937(2)
O23A ⁱⁱ	3.193(2)	3.098(2)	–	–
(H₅O₂)K(ADP)₂·4.25H₂O (IVb) (B)				
O4 ^A	3.071(2)	2.944(2)	–	–
O21B ^{vi}	–	–	3.791(2)	3.762(2)
O22B ^{vi}	3.677(2)	3.020(2)	3.711(2)	2.993(2)
O23B ^{vi}	3.167(2)	3.115(2)	–	–
Symmetry codes: ⁽ⁱ⁾ $x-1, y, z$; ⁽ⁱⁱ⁾ $-x+1, y+1/2, -z+1/2$; ⁽ⁱⁱⁱ⁾ $x, y, z+1$; ^(iv) $-x+1/2, y+1/2, -z+1$; ^(v) $-x+1, y+1/2, -z+3/2$; ^(vi) $-x+1, y+1/2, -z-1/2$				



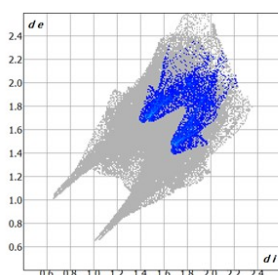
all contacts



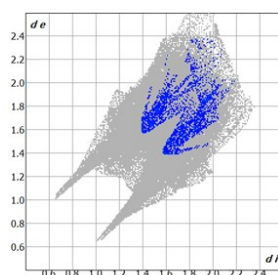
**H \cdots O (15.9%) / O \cdots H (25.2%)
(41.1% in total)**



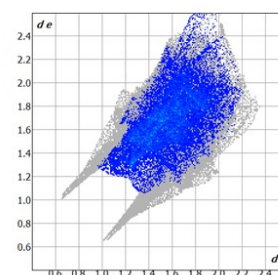
**H \cdots N (5.6%) / N \cdots H (7.2%)
(12.8% in total)**



O \cdots C / C \cdots O (7.0%)

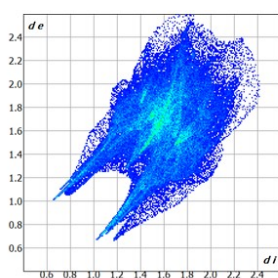
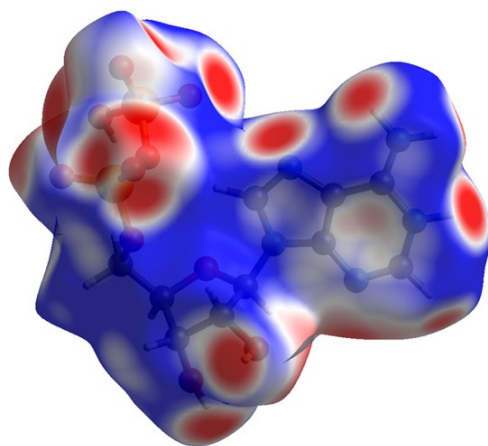


O \cdots N / N \cdots O (3.4%)

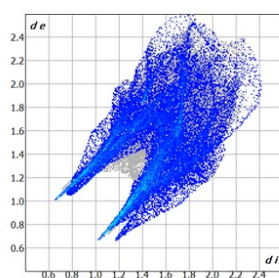


H \cdots H (23.1%)

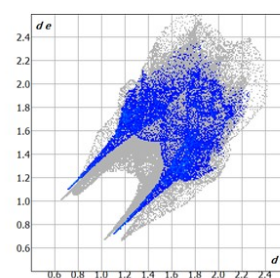
Fig. S17 Hirshfeld surface of the ADP^- anion (front and back side) in $\text{K}(\text{ADP})\cdot 4\text{H}_2\text{O}$ (I): d_{norm} [from -0.5 (blue) to 0.5 Å (red)] mapped on the surface (top), and 2D fingerprint plots with selected types of contacts and their contribution to the surface area (bottom).



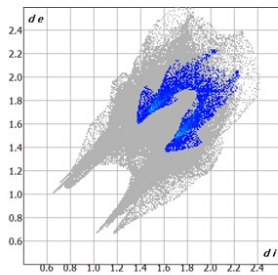
all contacts



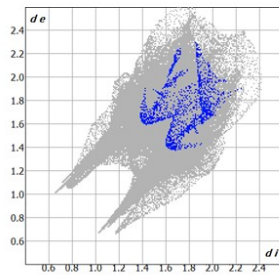
**H...O (14.6%) / O...H (22.2%)
(36.7% in total)**



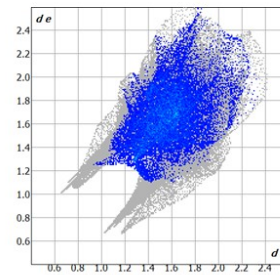
**H...N (6.3%) / N...H (7.5%)
(13.8% in total)**



O...C / C...O (6.5%)

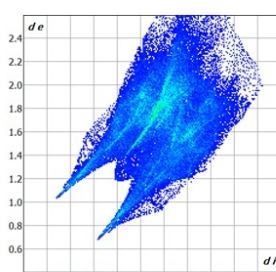
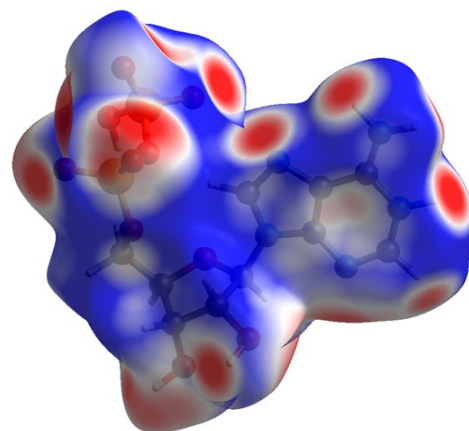


O...N / N...O (2.0%)

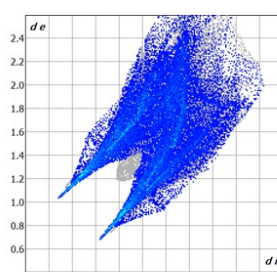


H...H (23.0%)

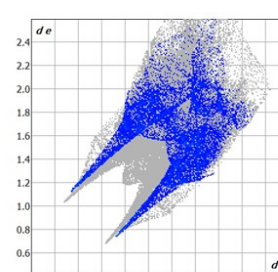
Fig. S18 Hirshfeld surface of the ADP^- anion in $\text{K}(\text{ADP})\cdot 2\text{H}_2\text{O}$ (II): d_{norm} [from -0.5 (blue) to 0.5 Å (red)] mapped on the surface (top), and 2D fingerprint plots with selected types of contacts and their contribution to the surface area (bottom).



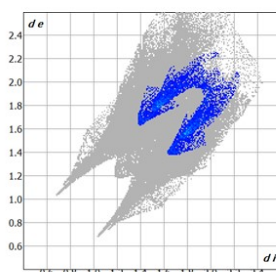
all contacts



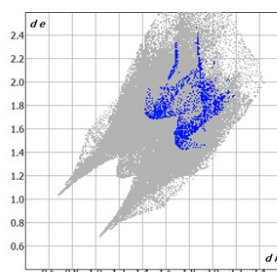
**H...O (18.5%) / O...H (22.9%)
(41.4% in total)**



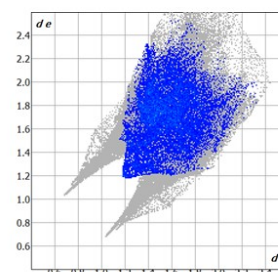
**H...N (6.0%) / N...H (7.3%)
(13.3% in total)**



O...C / C...O (6.3%)

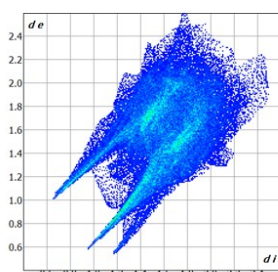
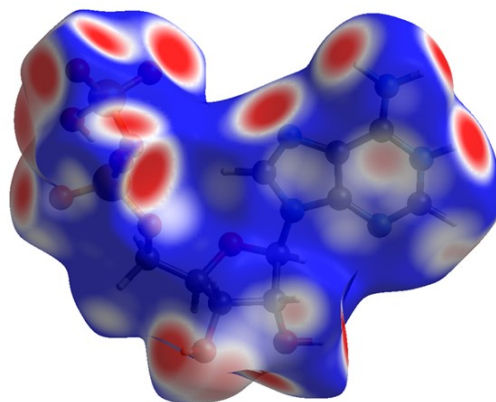


O...N / N...O (2.4%)

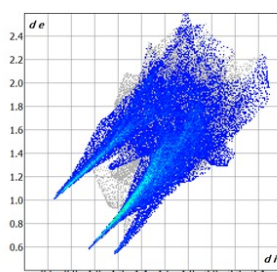


H...H (18.7%)

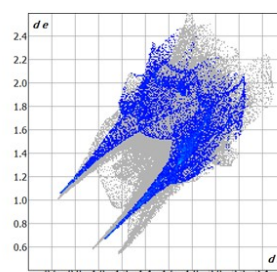
Fig. S19 Hirshfeld surface of the ADP^- anion in K(ADP) (III) : d_{norm} [from -0.5 (blue) to 0.5 Å (red)] mapped on the surface (top), and 2D fingerprint plots with selected types of contacts and their contribution to the surface area (bottom).



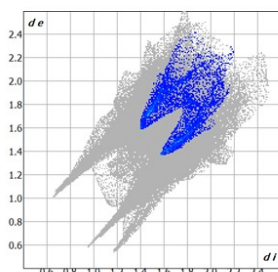
all contacts



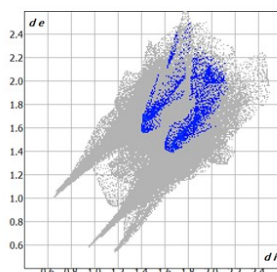
**H...O (16.8%) / O...H (27.4%)
(44.2% in total)**



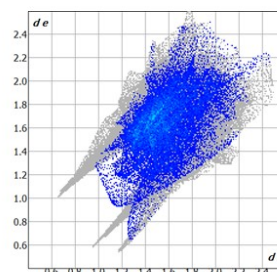
**H...N (4.9%) / N...H (7.9%)
(12.8% in total)**



O...C / C...O (6.1%)

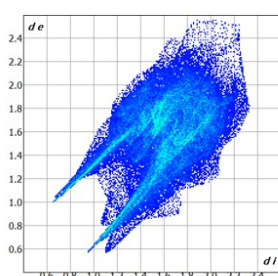
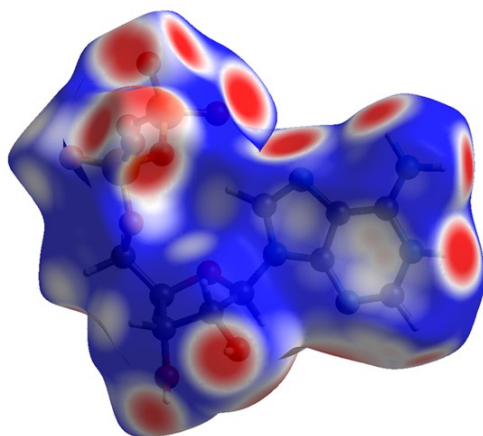


O...N / N...O (3.0%)

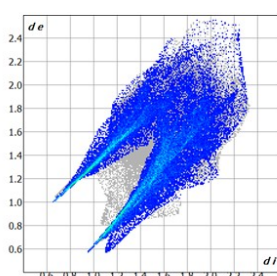


H...H (24.7%)

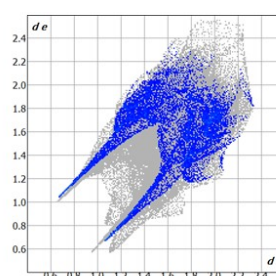
Fig. S20 Hirshfeld surface of the ADP^- anion **A** in low-temperature phase of $(\text{H}_5\text{O}_2)\text{K}(\text{ADP})_2 \cdot 4.25\text{H}_2\text{O}$ (**IVb**): d_{norm} [from -0.5 (blue) to 0.5 Å (red)] mapped on the surface (top), and 2D fingerprint plots with selected types of contacts and their contribution to the surface area (bottom).



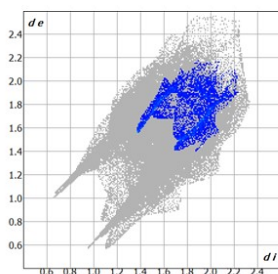
all contacts



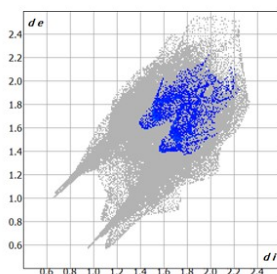
**H...O (15.8%) / O...H (27.6%)
(43.4% in total)**



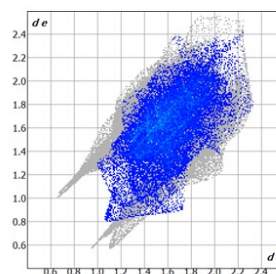
**H...N (4.8%) / N...H (6.6%)
(12.8% in total)**



O...C / C...O (5.5%)



O...N / N...O (4.0%)



H...H (25.6%)

Fig. S21 Hirshfeld surface of the ADP^- anion B in low-temperature phase of $(\text{H}_5\text{O}_2)\text{K}(\text{ADP})_2 \cdot 4.25\text{H}_2\text{O}$ (**IVb**): d_{norm} [from -0.5 (blue) to 0.5 Å (red)] mapped on the surface (top), and 2D fingerprint plots with selected types of contacts and their contribution to the surface area (bottom).

Table S15 Estimated percentage content of K(ADP)·2H₂O and K(ADP) form in the sample during the dehydration based on series of micro-PXRD measurements.

Temperature [K]	K(ADP)·2H ₂ O	K(ADP)
100	100.0	0.0
250	100.0	0.0
255	100.0	0.0
260	100.0	0.0
265	100.0	0.0
270	100.0	0.0
275	100.0	0.0
280	100.0	0.0
285	100.0	0.0
290	100.0	0.0
295	100.0	0.0
300	96.5	3.5
305	93.1	6.9
310	91.0	9.0
315	86.4	13.6
320	80.4	19.6
325	72.5	27.5
330	66.1	33.9
335	56.9	43.1
340	47.3	52.7
345	35.2	64.8
350	19.7	80.3
355	8.2	91.8
330	4.1	95.9
330	3.8	96.2
330	0.5	99.5
100*	4.8	95.2

(*) Based on diffraction pattern of the sample after dehydration after two days.

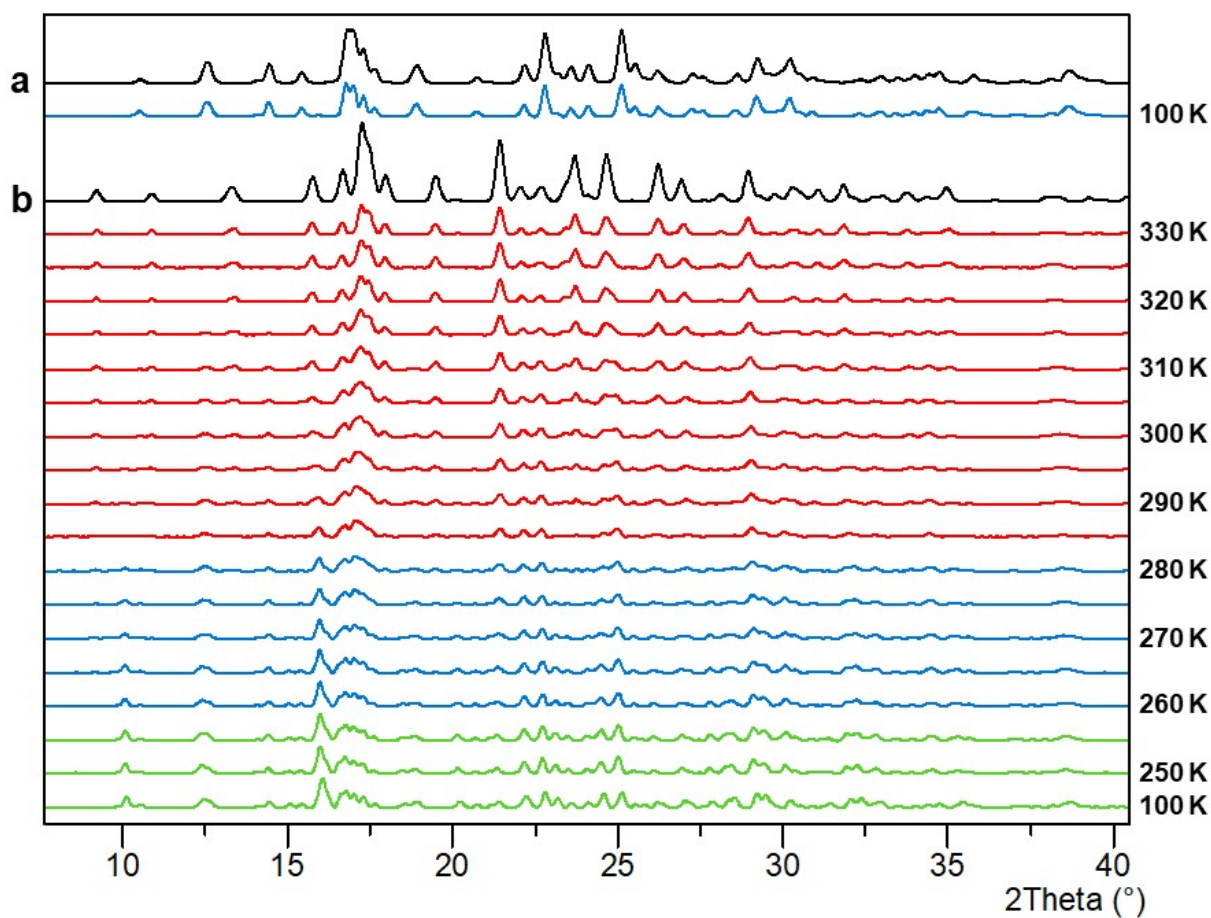


Fig. S22 The variable temperature micro-powder X-ray diffraction patterns (2θ range $7.5\text{--}40.5^\circ$) for mixture of I and II (55:45) recorded at 100–330 K. Patterns for which the tetrahydrate phase was the dominant form are marked in green, for dihydrate dominant – in blue and anhydrous phase – in red. Theoretical (calculated) diffraction patterns for II (a) and III (b) are marked in black.

Table S16 Estimated percentage content of K(ADP)·4H₂O, K(ADP)·2H₂O and K(ADP) form in the sample during the dehydration based on series of micro-PXRD measurements.

Temperature [K]	K(ADP)·4H ₂ O	K(ADP)·2H ₂ O	K(ADP)
100	55.9	44.1	0.00
250	52.2	47.8	0.00
255	50.9	49.1	0.00
260	49.3	50.7	0.00
265	44.8	47.9	7.3
270	39.1	46.8	14.1
275	31.8	46.0	22.2
280	24.0	43.9	32.1
285	15.6	41.8	42.6
290	8.1	40.2	51.7
295	3.6	38.0	58.4
300	1.2	35.3	63.5
305	0.0	30.4	69.6
310	0.0	22.9	77.1
315	0.0	11.4	88.6
320	0.0	5.8	94.2
325	0.0	1.6	98.4
330	0.0	0.1	99.9
100*	0.0	0.0	100.0

(*) Based on diffraction pattern of the sample after dehydration after two days.

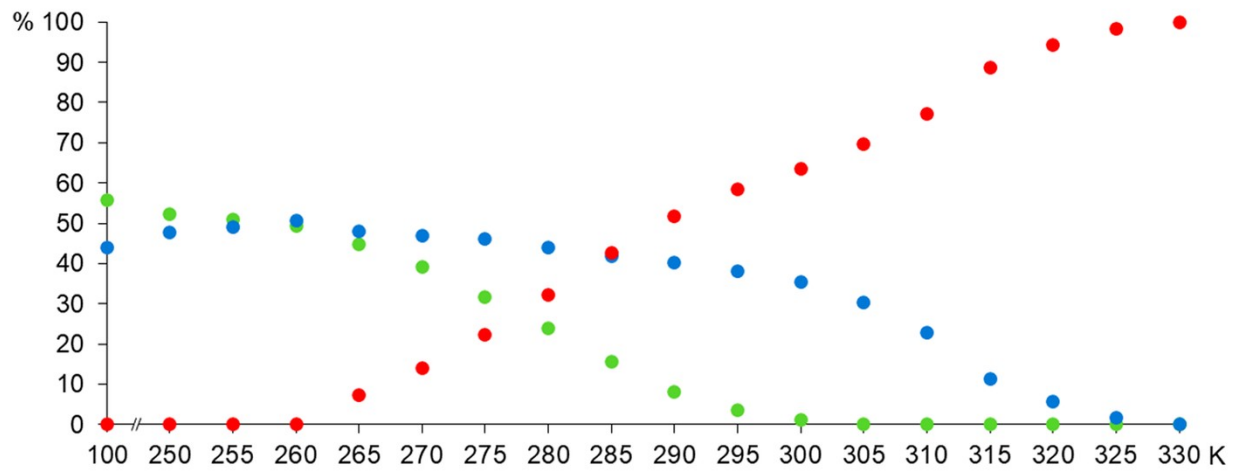


Fig. S23 Estimated percentage content of K(ADP)·4H₂O (marked in green), K(ADP)·2H₂O (marked in blue) and K(ADP) form (marked in red) in the sample during the dehydration based on series of micro-PXRD measurements.

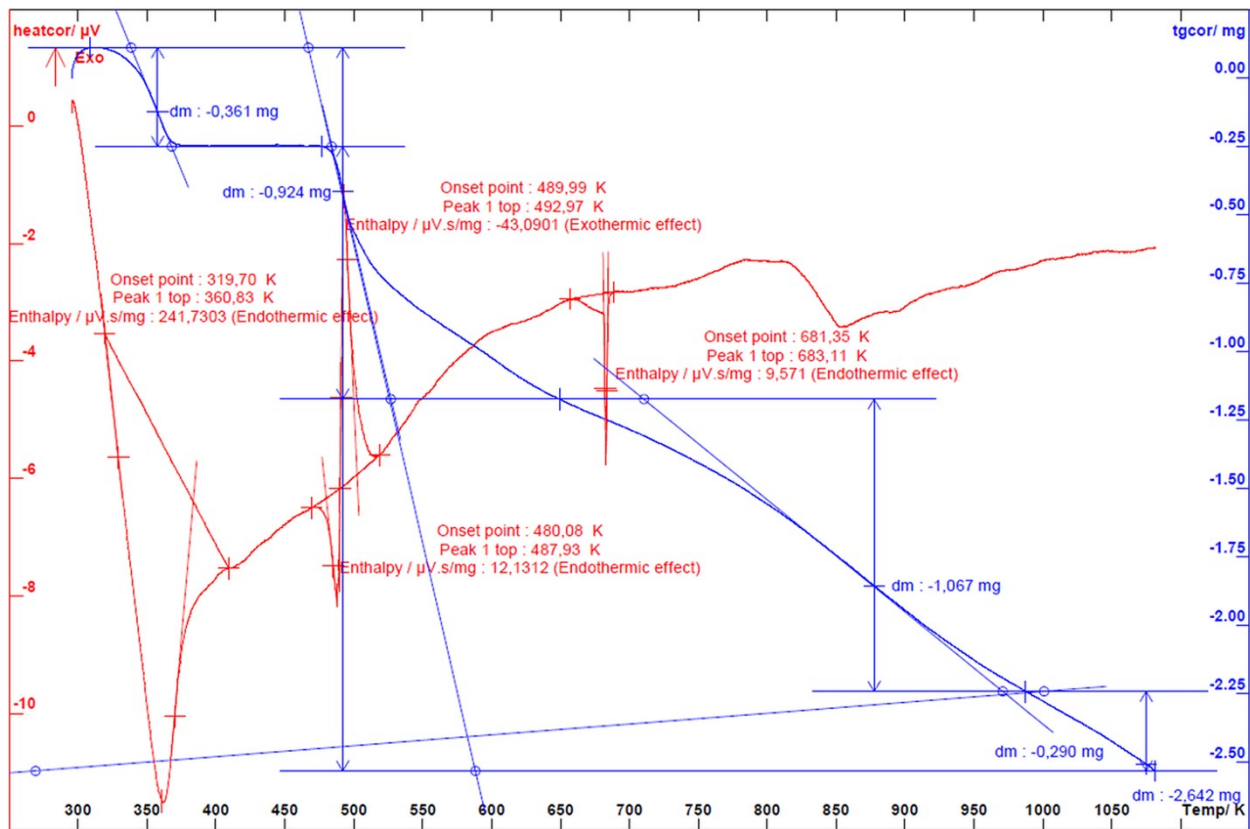


Fig. S24 TGA/DTA curves for II (mass of sample = 5.192 mg).

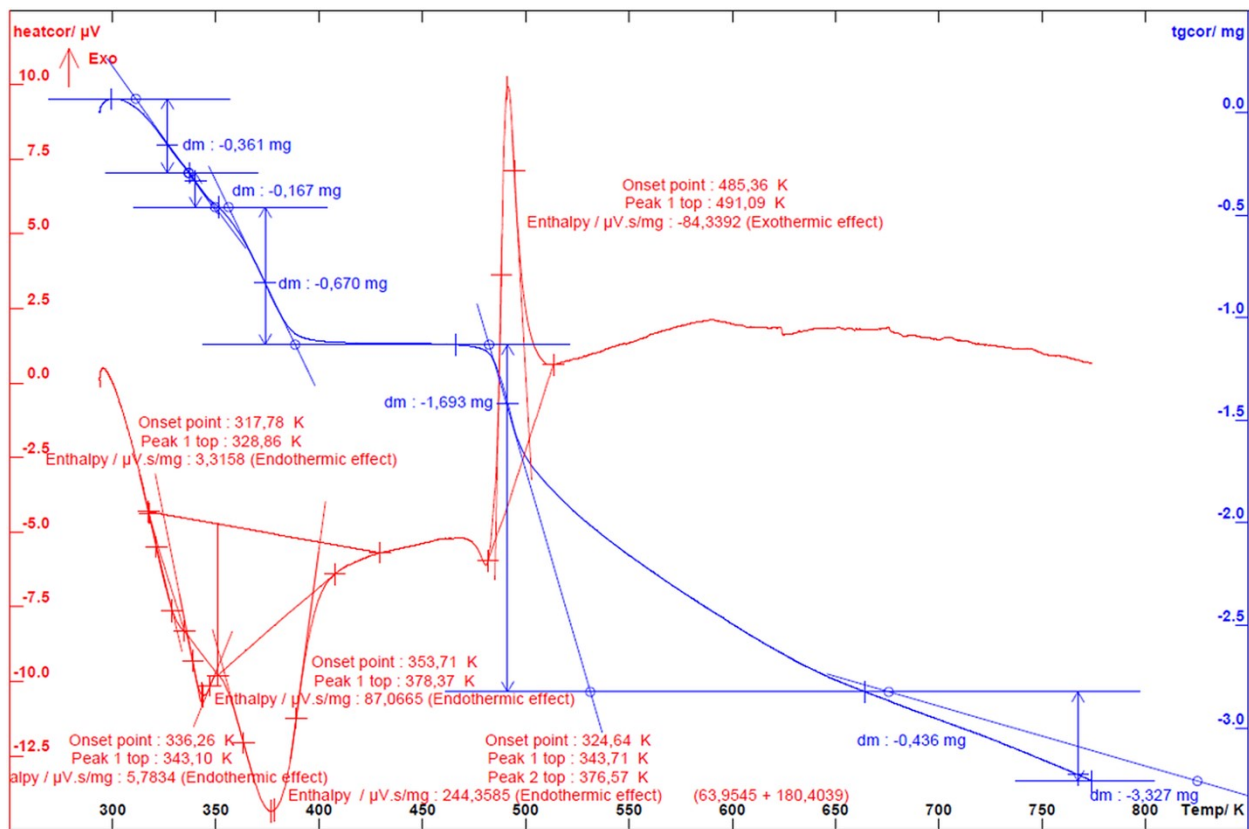


Fig. S25 TGA/DTA curves for mixture of I and II (mass of sample = 9.712 mg).