

Supporting information

Linker length-dependent hydrogen peroxide photosynthesis performance over crystalline covalent organic frameworks

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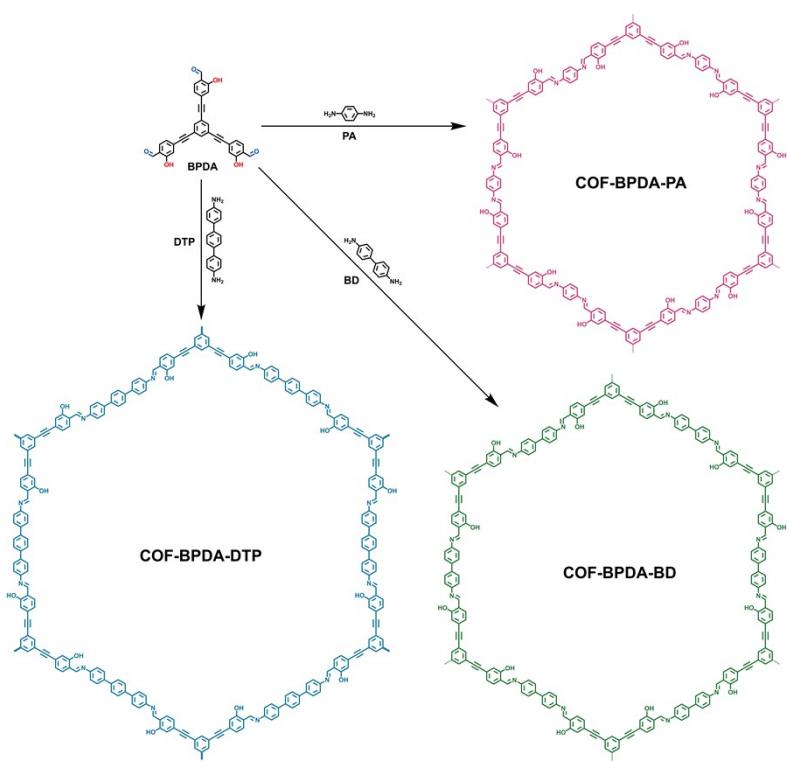


Figure S1. Synthetic route of COF-BPDA-DTP, COF-BPDA-BD and COF-BPDA-PA.

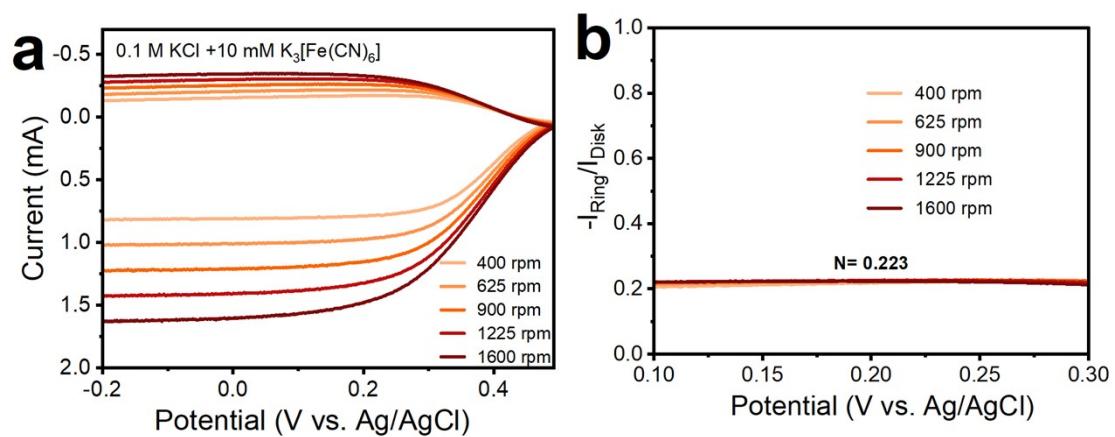


Figure S2. (a) The LSV curves at different rotate speeds. (b) The calculated collection efficiency (N) for RRDE

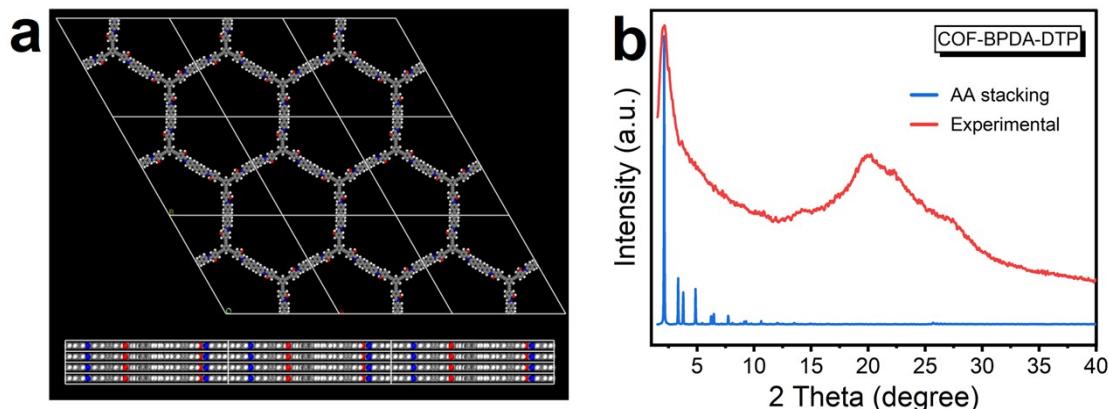


Figure S3. (a) Eclipsed (AA) stacking mode of COF-BPDA-DTP. (b) Comparison between experimental PXRD pattern and simulated pattern. White, gray, blue, and red spheres represent H, C, N, and O atoms, respectively.

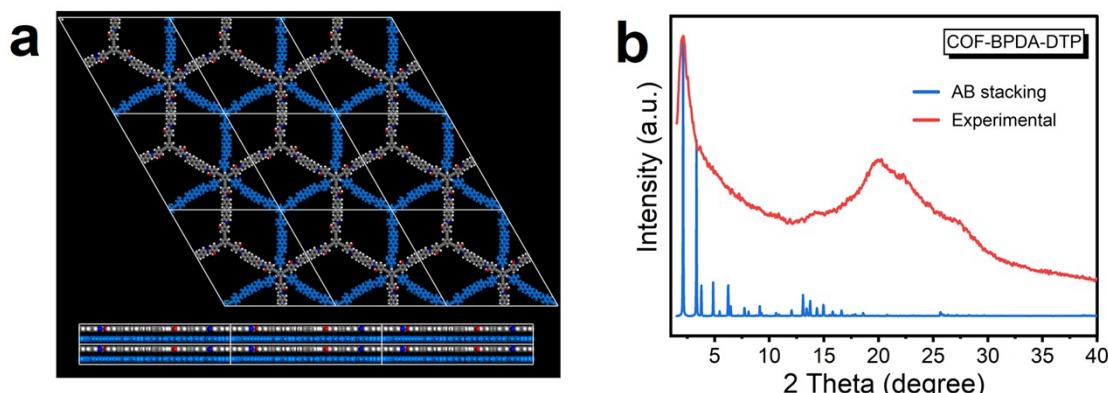


Figure S4. (a) Staggered (AB) stacking mode of COF-BPDA-DTP. (b) Comparison between experimental PXRD pattern and simulated pattern. White, gray, blue, and red spheres represent H, C, N, and O atoms, respectively; the second layer is highlighted in blue for clarity.

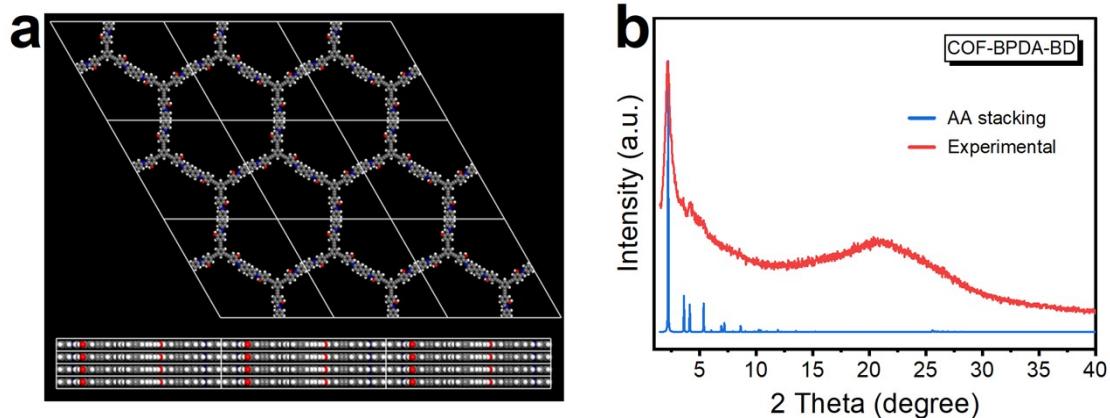


Figure S5. (a) Eclipsed (AA) stacking mode of COF-BPDA-BD. (b) Comparison between experimental PXRD pattern and simulated pattern. White, gray, blue, and red spheres represent H, C, N, and O atoms, respectively.

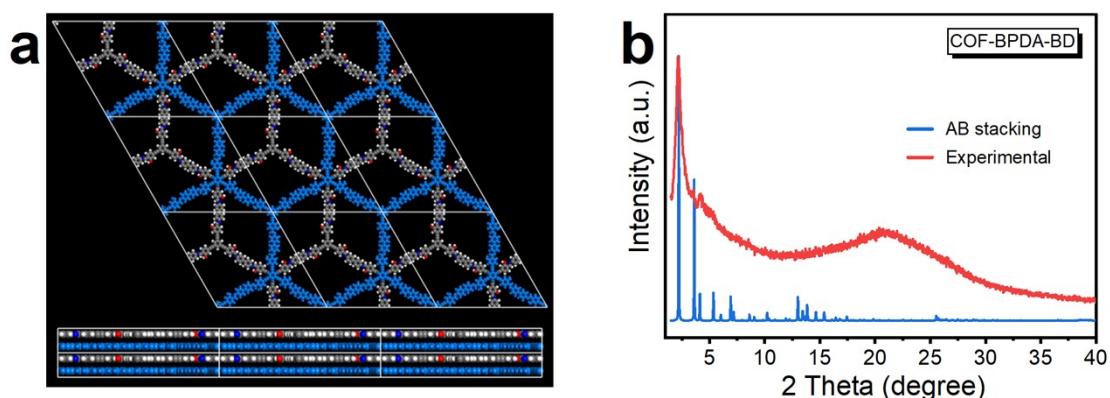


Figure S6. (a) Staggered (AB) stacking mode of COF-BPDA-BD. (b) Comparison between experimental PXRD pattern and simulated pattern. White, gray, blue, and red spheres represent H, C, N, and O atoms, respectively; the second layer is highlighted in blue for clarity.

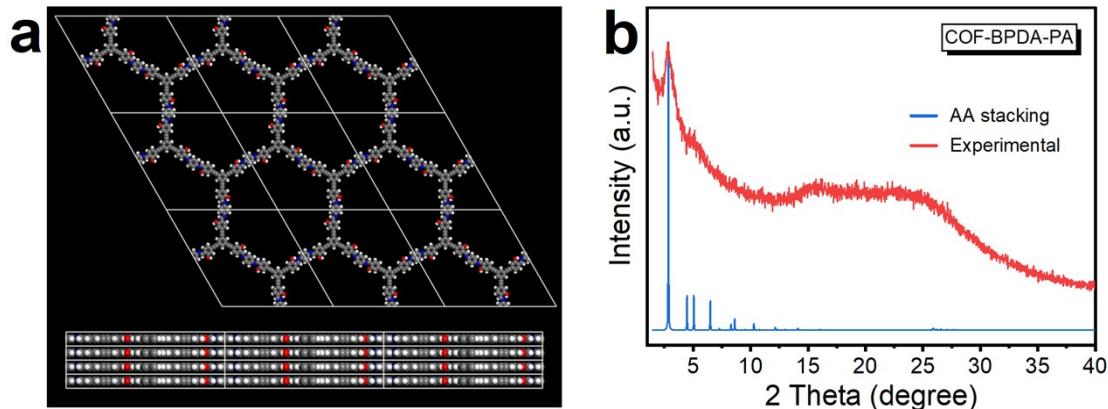


Figure S7. (a) Eclipsed (AA) stacking mode of COF-BPDA-PA. (b) Comparison between experimental PXRD pattern and simulated pattern. White, gray, blue, and red spheres represent H, C, N, and O atoms, respectively.

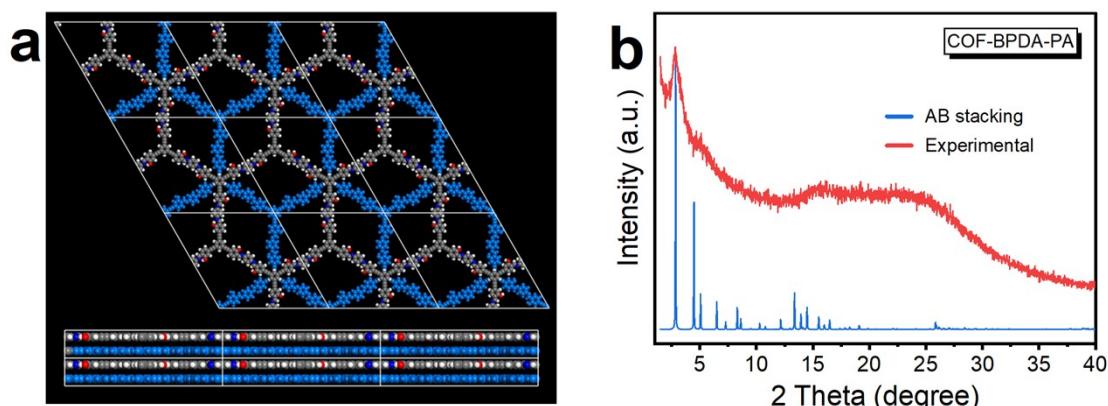


Figure S8. (a) Staggered (AB) stacking mode of COF-BPDA-PA. (b) Comparison between experimental PXRD pattern and simulated pattern. White, gray, blue, and red spheres represent H, C, N, and O atoms, respectively; the second layer is highlighted in blue for clarity.

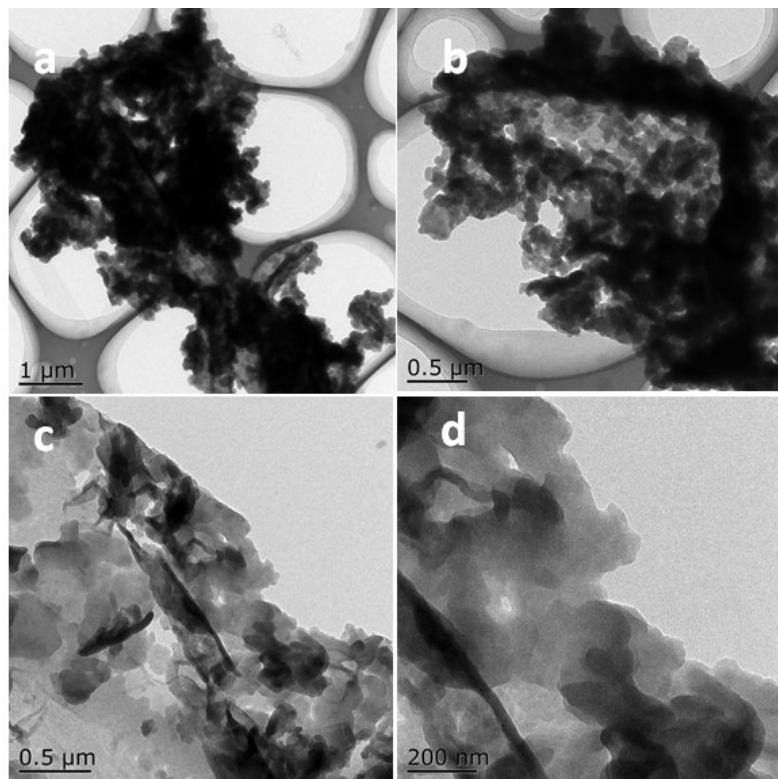


Figure S9. TEM images of COF-BPDA-BD (a and b) and COF-BPDA-PA (c and d)

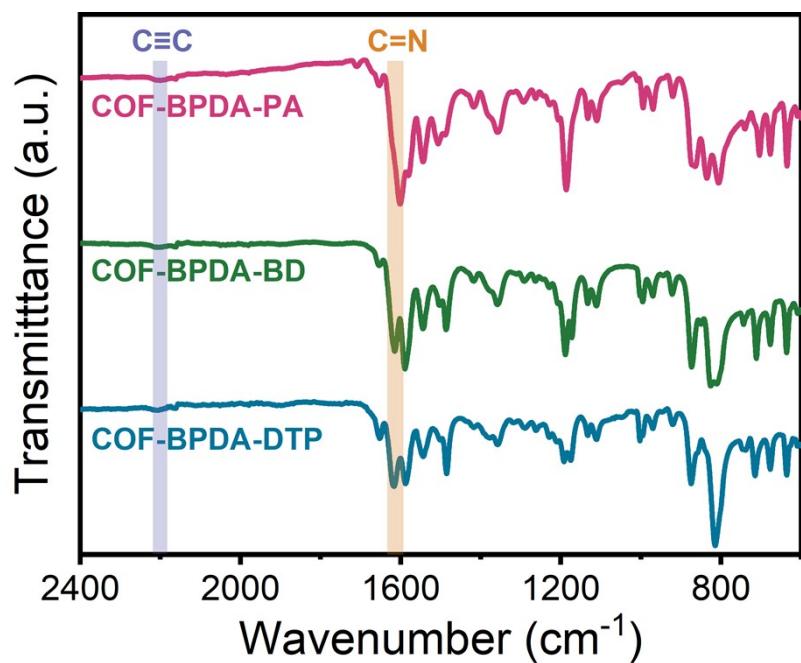
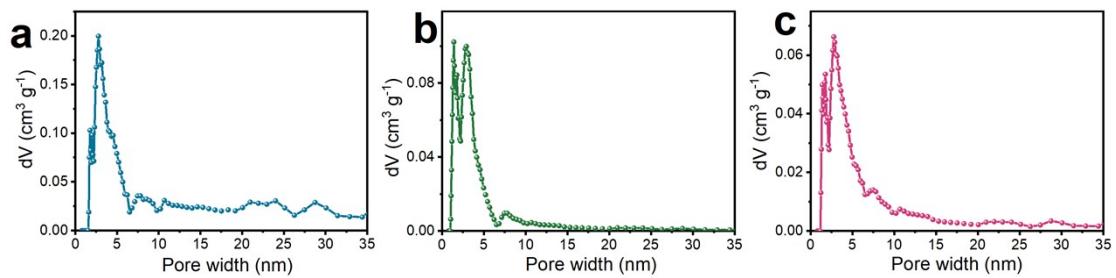


Figure S11. FT-IR spectra of COF-BPDA-DTP, COF-BPDA-BD and COF-BPDA-PA.

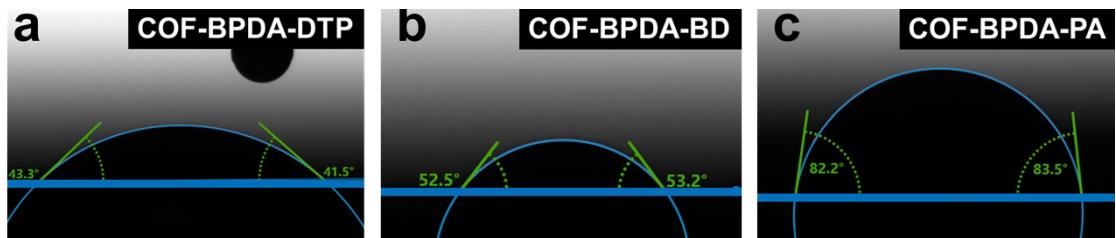


Figure S12. The water contact angle of (a) COF-BPDA-DTP, (b) COF-BPDA-BD and (C) COF-BPDA-PA.

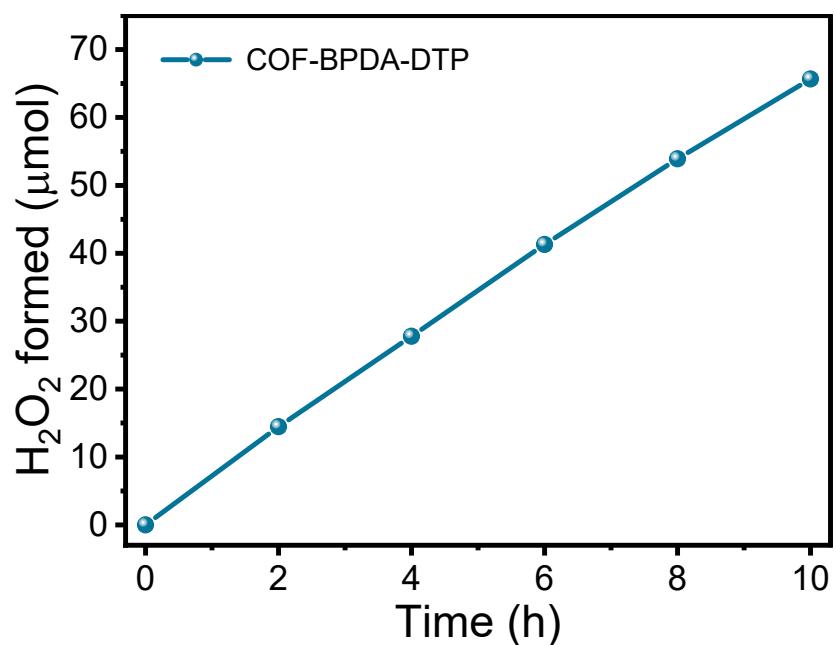


Figure S13. The time-amount curves for photocatalytic H_2O_2 production over COF-BPDA-DTP (35mg, 40 mL pure water).

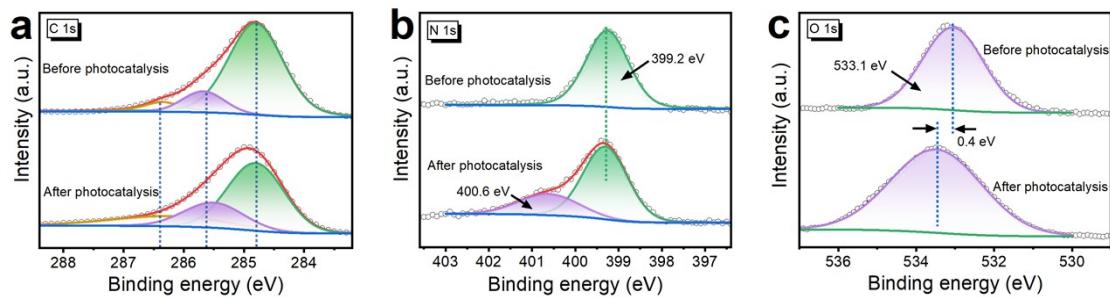


Figure S14. (a) C 1s, (b) N 1s and (c) O 1s XPS spectra of COF-BPDA-DTP before and after undergoing photocatalysis.

Table S1. The comparison of H₂O₂ production rate with other reported photocatalysts without sacrificial reagents.

Samples	H ₂ O ₂ yield rate ($\mu\text{mol h}^{-1} \text{g}_{\text{cat}}^{-1}$)	Test conditions	Irradiated conditions	Solvent	Ref.
CTF-BDDBN	97	30 mg catalysts and 50 mL water	$\lambda > 420 \text{ nm}$	H ₂ O	S1
COF-TfpBpy	695	5 mg catalysts and 10 mL water	$\lambda > 420 \text{ nm}$	H ₂ O	S2
MRF-250	582	50 mg catalysts and 30 mL water	$\lambda > 420 \text{ nm}$	H ₂ O	S3
N ₀ -COF	1570	10 mg catalysts and 20 mL water	$\lambda = 495 \text{ nm}$	H ₂ O	S4
1H-COF	700	30 mg catalysts and 30 mL water	$\lambda > 420 \text{ nm}$	H ₂ O	S5
DE7-M	266	30 mg catalysts and 50 mL water	$\lambda > 420 \text{ nm}$	H ₂ O	S6

SonoCOF-F2	1244	3 mg catalysts and 5 mL water	$\lambda > 420 \text{ nm}$	H ₂ O	S7
HEP-TAPT-COF	1750	50 mg catalysts and 100 mL water	$\lambda > 420 \text{ nm}$	H ₂ O	S8
FS-COFs	3904	5 mg catalysts and 20 mL water	$\lambda > 420 \text{ nm}$	H ₂ O	S9
TTF-BT-COF	2760	5 mg catalysts and 10 mL water	$\lambda > 420 \text{ nm}$	H ₂ O	S10
TTF-pT-COF	996				
TPE-BT-COF	592				
COF-BPDA-PA	450	5 mg catalysts and 40 mL water	$\lambda > 420 \text{ nm}$	H ₂ O	This work
COF-BPDA-BD	1040			H ₂ O	
COF-BPDA-DTP	1164			H ₂ O	

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