SUPPORTING MATERIAL

Synthesis and Characterisation of New Coordination Polymers by Combining 2-Pyridyl Oximes or Alcohols with Functionalised Terephthalic Acid Analogues

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Complex	1·3DMF	2 ·2DMF	3
Empirical formula	$C_{29}H_{37}N_9O_{11}Zn$	C ₂₃ H ₂₅ N ₇ O ₉ Mn	$C_{50}H_{42}N_{12}O_{12}Co_3$
Formula weight	753.04	598.44	1179.74
Crystal system	Triclinic	Triclinic	Triclinic
Space group	Pī	Pī	Pī
<i>a</i> (Å)	9.3796 (7)	9.5541(6)	11.1458(8)
<i>b</i> (Å)	12.5378 (8)	10.0432(6)	14.7658(8)
<i>c</i> (Å)	15.3451 (10)	15.8933(8)	18.7690(9)
α (°)	87.967 (5)	76.089(5)	81.504(4)
β (°)	73.076 (6)	72.979(5)	89.225(5)
γ (°)	84.839 (6)	76.732(5)	86.463(5)
V (Å ³)	1719.3 (2)	1394.49(15)	3049.2(3)
Ζ	2	2	2
ρ_{calc} (g cm ⁻³)	1.455	1.406	1.285
Radiation, λ (Å)	1.54184	1.54184	1.54184
µ (mm ⁻¹)	1.608	4.392	6.851
Measd/independent reflns (<i>R</i> _{int})	11682 /6767 (0.060)	9478/ 6313 (0.0588)	21718/11529 (0.0992)
Parameters refined	483	722	694
GoF (on <i>P</i> ²)	1.130	1.135	1.020
R_1^{a} (<i>I</i> > 2 σ (<i>I</i>))	0.0656	0.0778	0.0764
$wR_{2^{b}}(l > 2\sigma(l))$	0.1816	0.2130	0.2121
$(\Delta ho)_{ m max}/(\Delta ho)_{ m min}$ (e Å ⁻³)	0.919/-0.772	0.895 / -0.929	0.913 /-1.047
Complex	4 DMF	5	6.2DMF
Empirical formula	C ₂₃ H ₂₅ N ₃ O ₉ Ni	$C_{20}H_{16}N_2O_8Zn$	C ₂₆ H ₃₀ N ₈ O ₉ Ni
Formula weight	546.17	461.74	657.29
Crystal system	Triclinic	Monoclinic	Monoclinic
Space group	Pī	P2/n	P2 1/c

Table S1: Crystallographic data for 1-7

a (Å)	9.6616(13)	9.49420(2)	17.2182(10)
b (Å)	10.644(2)	10.0245(3)	10.0832(6)
<i>c</i> (Å)	12.4656(15)	13.4633(4)	18.2479(9)
a (°)	95.832(14)	90	90
β (°)	90.191(11)	93.449(2)°	105.108(5)
γ (°)	97.496(15)	90	90
V (Å ³)	1264.2(4)	1278.75(7)	3058.6(3)
Ζ	2	2	4
ρ_{calc} (g cm ⁻³)	1.435	1.199	1.427
Radiation, λ (Å)	0.71073	1.54184	0.71073
µ (mm ⁻¹)	0.823	1.654	0.698
Measd/independent reflns (<i>R</i> _{int})	8537 /2970 (0.0963)	8508/4741 (0.0332)	21202 / 5571 (0.0911)
Parameters refined	332	145	419
GoF (on <i>F</i> ²)	0.707	1.043	0.901
$R_1^{a} (I > 2\sigma(I))$	0.0815	0.0419	0.0812
wR_2^{b} $(I > 2\sigma(I))$	0.1818	0.1199	0.2268
$(\Delta ho)_{ m max}/(\Delta ho)_{ m min}$ (e Å ⁻³)	1.078/ -0.429	0.361/-0.289	1.147/- 1.242
Complex	7 ·DMF		
Empirical formula	$C_{72}H_{74}N_{14}O_{24}Zn_3$		
Formula weight	1715.56		
Crystal system	Monoclinic		
Space group	la		
a (Å)	18.3768(9)		
b (Å)	10.3212(6)		
<i>c</i> (Å)	45.523(2)		
a (°)	90		
β (°)	92.253(4)		
γ (°)	90		

V (Å ³)	8627.6(8)
Ζ	4
$ ho_{ m calc}$ (g cm ⁻³)	1.321
Radiation, λ (Å)	0.71073
μ (mm ⁻¹)	0.904
Measd/independent reflns (<i>R</i> _{int})	36376 / 18203 (0. 1108)
Parameters refined	985
GoF (on <i>F</i> ²)	0.982
$R_1^{\rm a} \left(l > 2\sigma(l) \right)$	0.1022
$wR_2^{\rm b}\left(l>2\sigma(l)\right)$	0.2450
$(\Delta ho)_{max}/(\Delta ho)_{min}$ (e Å ⁻³)	1.437 /-0.453

Table S2: Hydrogen bonding details for **1** 3DMF.

D – H A	D A	Н А	DHA	Symmetry operator
	(Å)	(Å)	(°)	of A
O(8) – H(8) O(5)	2.570	1.787	140.0	х, <i>у</i> , <i>z</i>
O(2) – H(2) O(4)	2.550	1.603	165.5	x, y, z
O(7) – H(7)O(3)	2.520	1.566	147.6	x, y, z
O(1) – H(1) O(6)	2.549	1.584	175.8	x, y, z
N(3) – H(1N3) O(11)	2.869	2.037	158.1	x, y, z
N(3) -H(2N3)O(10)	2.890	2.160	148.5	x, y, z
N(6)-H(2N6)O(7)	2.957	2.171	150.2	-1+x, y, z
N(6)-H(1N6)O(1)	3.028	2.254	166.2	-1+x, y, z

A=acceptor, D=donor

D – H A	D A	Η Α	DHA	Symmetry operator
	(Å)	(Å)	(°)	of A
O(3) – H(3A) O(2)	2.516	1.809	143.7	1-x, 2-y, 1-z
O(8) – H(8) O(1)	2.609	1.789	177.6	x, y, z
O(7) – H(7A)O(5)	2.605	1.849	152.5	х, у, z
O(6) – H(6) O(5)	2.583	1.867	145.2	x, y, z

Table S3: Hydrogen bonding details for 4 DMF.

A=acceptor, D=donor

Table S4: Hydrogen bonding details for 6 2DMF.

D – H A	D A	Η Α	DHA	Symmetry operator
	(Å)	(Å)	(°)	of A
O(3) – H(3A) O(4)	2.497	2.176	103.4	x, y, z
N(2) – H(2B)O(9A)	2.843	2.033	156.5	x, y, z
O(6) – H(6) … O(5)	2.583	1.867	145.2	x, y, z
N(2)-H(2A)O(6)	2.576	2.276	100.6	x, y, z
N(6)-H(6A)O(7)	2.576	2.280	100.6	x, y, z

A=acceptor, D=donor

D – H A	D A	ΗΑ	DHA	Symmetry operator
	(Å)	(Å)	(°)	of A
O(7) – H(7) O(2)	2.537	1.863	138.6	x, y, z
O(5) -H(5B) O(3)	2.655	2.255	110.3	x, y, z
O(10) – H(10B)O(11)	2.597	3.151	41.8	x, y, z
O(15) – H(15A) O(18)	2.512	1.751	153.4	x, y, z
O(19) – H(19B) O(21)	2.562	1.860	142.9	x, y, z
O(14)-H(14A)O(9)	2.519	1.778	149.3	x, y, z
O(6)-H(6)O(3)	2.634	1.890	150.1	0.5+x, 2-y, z
O(13)-H(13A)O(12)	2.556	1.755	164.3	0.5+x, 1-y, z
O(16)-H(16)O(20)	2.503	1.685	175.2	-0.5+x, -y, z

Table S5: Hydrogen bonding details for 7 DMF.

A=acceptor, D=Donor



Fig. S1: TGA plot for 4-DMF.



Fig. S2: TGA plot for 6-2DMF.



Fig. S3: The 2-c network of **1** along [100] plane $({Zn(Hhmp)_2}^{2+}: orange and H_2DHBDC: yellow node).$





Fig. S4: The 2-c network of **5** along [100] plane $({Zn(H_2pyaox)_2}^{2+}: orange and H_2DHBDC: yellow node).$

Fig. S5: The PXRD patterns of **1** 3DMF as synthesized (below) and theoretical (above).



Fig. S6: The PXRD patterns of 2:2DMF as synthesized (below) and theoretical (above).



Fig. S7: The PXRD patterns of 4 DMF as synthesized (below) and theoretical (above).



Fig. S8: The PXRD patterns of **4** DMF as synthesized (above) and after being immersed and stirred in EtOH for 100 h (below).



Fig. S9: The PXRD patterns of 6 2DMF as synthesized (below) and theoretical (above).



Fig. S10: The PXRD patterns of **7** DMF as synthesized (below) and theoretical (above).



Fig. S11: IR spectrum of 1.3DMF



Fig. S12: IR spectrum of 2-2DMF



Fig. S13: IR spectrum of 3-2DMF



Fig. S14: IR spectrum of 4-DMF



Fig. S15: IR spectrum of 5.DMF



Fig. S16: IR spectrum of 6•DMF



Fig. S17: IR spectrum of 7.DMF











Fig. S20: IR spectrum of the product which has been derived by the reaction blend Hmpko:H₃MHBDC: $Zn(CH_3CO_2)_2.2H_2O$ in molar ratio 2:4:1. (excess of dicarboxylate). The product contains only the dicarboxylic linker.