

Visualizing the Alignment of Lone Pair Electrons in $\text{La}_3\text{AsS}_5\text{Br}_2$ and $\text{La}_5\text{As}_2\text{S}_9\text{Cl}_3$ to Form an
Acentric or Centrosymmetric Structure

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Supporting Information

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Table S1. Refined atomic coordinates and isotropic displacement parameters for $\text{La}_3\text{AsS}_5\text{Br}_2$ and $\text{La}_5\text{As}_2\text{S}_9\text{Cl}_3$

Atoms	Wyckoff	<i>x</i>	<i>y</i>	<i>z</i>	Occupancy	U_{eq} (\AA^2)
$\text{La}_3\text{AsS}_5\text{Br}_2$						
La1	<i>4a</i>	0.1227(5)	0.1977(12)	0.3695(16)	1	0.019(8)
La2	<i>4a</i>	-0.0278(5)	0.1974(13)	-0.1259(18)	1	0.018(8)
La3	<i>4a</i>	0.2744(6)	0.2081(15)	0.013(2)	1	0.019(8)
As1	<i>4a</i>	0.3189(10)	0.309(2)	-0.425(4)	1	0.023(10)
Br1	<i>4a</i>	0.1161(9)	0.379(3)	-0.020(3)	1	0.019(10)
Br2	<i>4a</i>	-0.0165(7)	0.387(3)	0.257(3)	1	0.034(10)
S1	<i>4a</i>	-0.148(2)	0.408(7)	-0.115(8)	1	0.020(18)
S2	<i>4a</i>	0.201(2)	0.055(6)	0.693(9)	1	0.002(16)
S3	<i>4a</i>	-0.116(2)	-0.046(6)	0.049(8)	1	0.01(2)
S4	<i>4a</i>	0.2431(18)	0.422(5)	0.354(7)	1	0.00(2)
S5	<i>4a</i>	0.047(3)	0.052(5)	0.615(8)	1	0.024(18)
$\text{La}_5\text{As}_2\text{S}_9\text{Cl}_3$						
La1	<i>4d</i>	0.79282(15)	0.19329(15)	0.75	1	0.0030(4)
La2	<i>8e</i>	1.20520(11)	0.27459(11)	0.83927(2)	1	0.0044(4)
La3	<i>8e</i>	0.71453(11)	0.04165(10)	0.95753(2)	1	0.0036(4)
As1	<i>8e</i>	0.6919(2)	0.3080(2)	0.86781(4)	1	0.0064(5)
S1	<i>8e</i>	0.9173(5)	0.1059(4)	0.88787(8)	1	0.0059(8)
S2	<i>8e</i>	1.4525(5)	0.2479(4)	0.90716(8)	1	0.0052(8)
S3	<i>8e</i>	0.5801(4)	0.1247(5)	0.67708(8)	1	0.0060(8)
S4	<i>8e</i>	0.6137(7)	-0.1684(7)	0.75	1	0.0114(12)
S5	<i>4d</i>	1.1120(5)	-0.0903(5)	0.96471(8)	1	0.0087(8)
Cl1	<i>4c</i>	0.5538(7)	-0.25	1	1	0.0067(11)
Cl2	<i>8e</i>	1.0579(5)	-0.0207(4)	0.70333(8)	1	0.0078(8)

Table S2. Selected interatomic distances for La₃AsS₅Br₂ and La₅As₂S₉Cl₃

Atom Pairs		Distances (Å)		Atom Pairs		Distances (Å)	
La ₃ AsS ₅ Br ₂							
La1	S5	2.81(7)		La3	S2	2.83(6)	
	S2	2.88(7)			S2	2.92(5)	
	S5	2.91(6)			S4	2.92(4)	
	S2	2.92(5)			S1	2.98(6)	
	S4	3.14(4)			S3	2.98(5)	
	Br1	3.06(2)			S4	3.05(5)	
	Br1	3.13(2)			S1	3.08(5)	
	Br2	3.37(2)		As1	S3	2.25(5)	
La2	S5	2.84(5)			S1	2.26(5)	
	S5	2.86(7)			S4	2.28(5)	
	S3	3.02(6)					
	S3	3.04(4)					
	S1	3.08(5)					
	Br2	3.04(3)					
	Br2	3.10(2)					
	Br1	3.44(3)					
La ₅ As ₂ S ₉ Cl ₃							
La1	S4	2.867(5)		La3	Cl1	2.841(2)	
	S4	3.029(5)			Cl1	2.874(3)	
	S3	3.132(3)			S5	2.902(4)	
	S3	3.132(3)			S5	2.966(4)	
	Cl2	2.872(3)			S1	2.989(3)	
	Cl2	2.872(3)			S2	3.010(3)	
	Cl2	2.968(3)			S2	3.041(3)	
	Cl2	2.968(3)			S5	3.153(3)	
La2	Cl2	2.827(3)		As1	S3	2.258(3)	
	Cl2	2.839(3)			S1	2.269(3)	

S3	2.913(3)	S2	2.272(4)
S1	2.968(3)		
S3	2.977(3)		
S2	3.069(3)		
S1	3.091(3)		
As1	3.5592(16)		

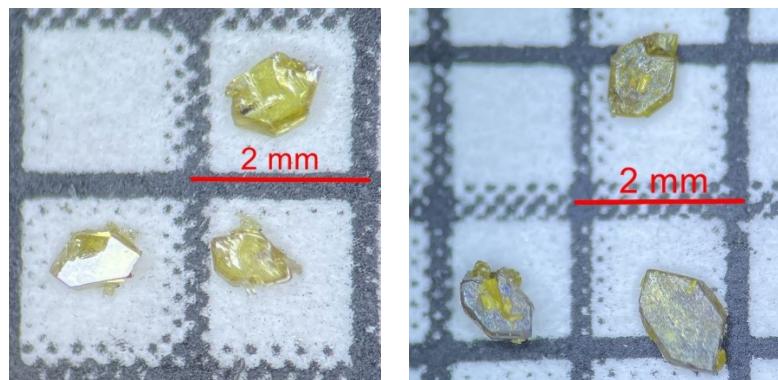


Figure S1. Optical microscope images of selected crystals of $\text{La}_3\text{AsS}_5\text{Br}_2$ (left) and $\text{La}_5\text{As}_2\text{S}_9\text{Cl}_3$ (right).

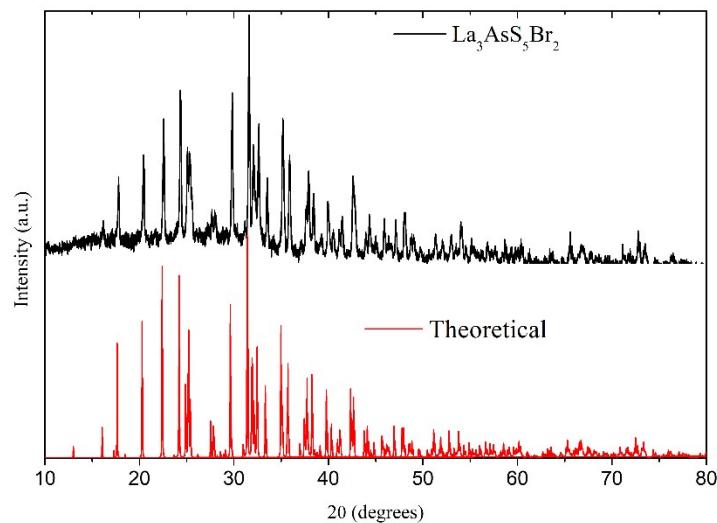


Figure S2. The calculated and experimental X-ray diffraction patterns of $\text{La}_3\text{AsS}_5\text{Br}_2$.

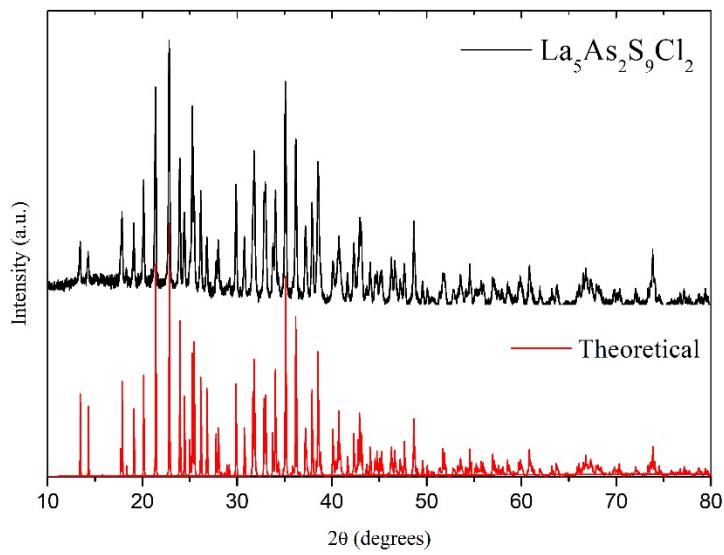


Figure S3. The calculated and experimental X-ray diffraction patterns of $\text{La}_5\text{As}_2\text{S}_9\text{Cl}_2$.

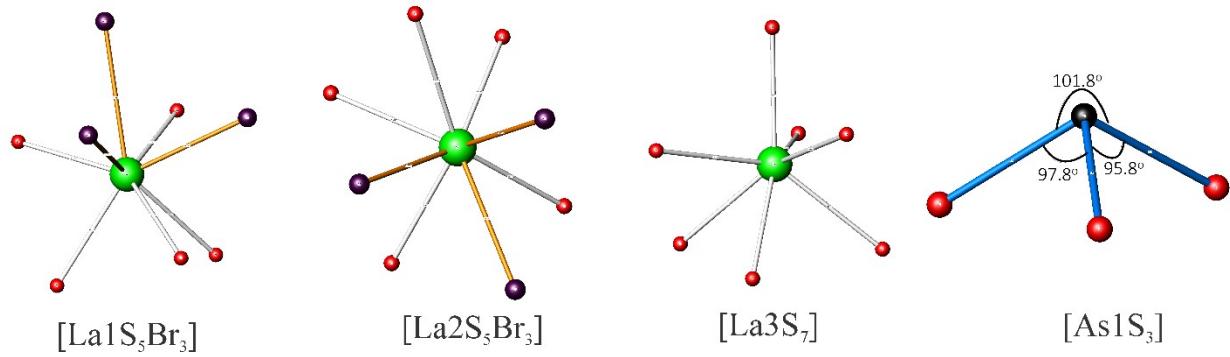


Figure S4. Basic building units (BBU) of $\text{La}_3\text{AsS}_5\text{Br}_2$.

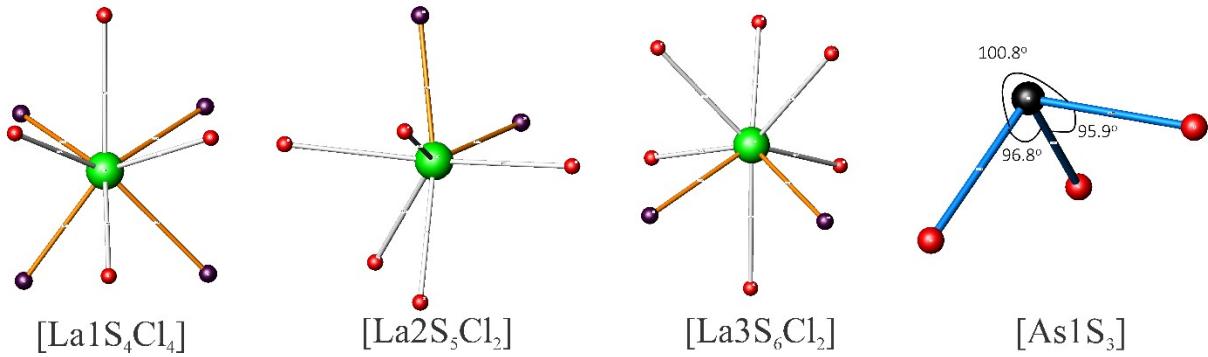


Figure S5. Basic building units (BBU) of $\text{La}_5\text{As}_2\text{S}_9\text{Cl}_3$.

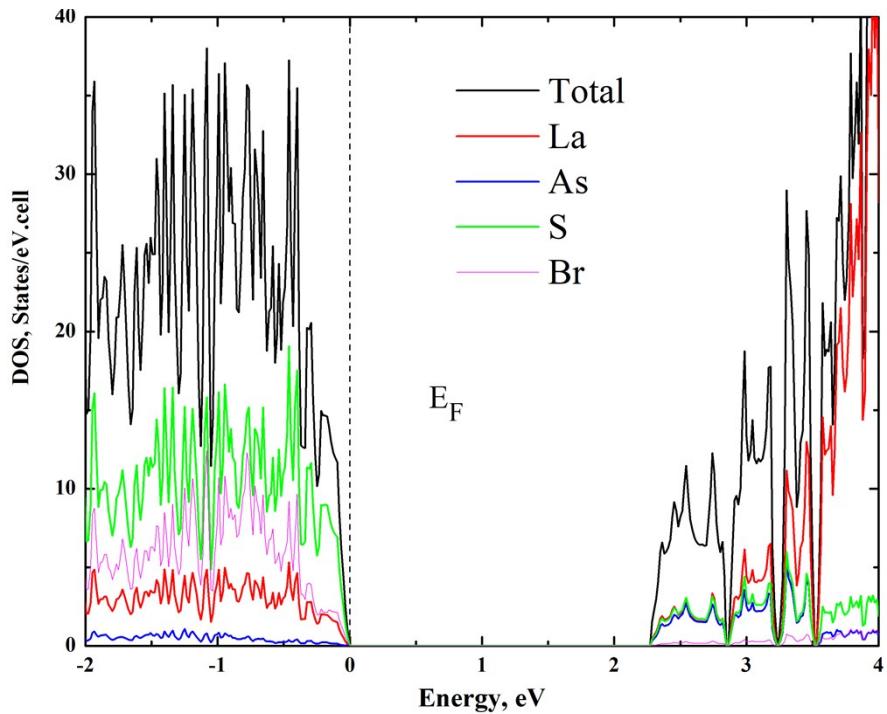


Figure S6. Density of states (DOS) of $\text{La}_3\text{AsS}_5\text{Br}_2$.