Supporting information

Fabrication of Shish-kebab Structure of MnO₂@Co₃O₄ with Remarkable Enhanced Oxidase-mimicking Activity for Detection of L-cysteine

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Figure. S1.EDX spectrum of MnO₂@Co₃O₄



Figure. S2. C 1s spectra of the $MnO_2@Co_3O_4$.



Figure. S3. SEM and Particle size distributions images of MnO₂@ZIF-67, the mixture was stirred for 2 minutes(A), 10 minutes(B), and 30 minutes(C) respectively, without overnight standing. SEM and Particle size distributions images of MnO₂@ZIF-67, the mixture was stirred for 2 minutes(D), 10 minutes(E), and 30 minutes(F) respectively, with overnight standing.



Figure. S4. The SEM images of MnO₂@ZIF-67 without PVP(A), with the addition of 100mg PVP(B), 200mg PVP(C) and 300mg PVP(D).



Figure. S5. The absorption spectra of the reaction in the air (a) and $N_2(b)$.



Figure. S6. (A)Typical UV-Vis spectra of different reaction systems: (a) $MnO_2@Co_3O_4+TMB+buffer$, (b) $MnO_2@Co_3O_4+TMB+buffer+NaHSO_3$ and (c) $MnO_2@Co_3O_4+TMB+buffer+Vitamin C$. Experiments were performed by adding 2mg.mL⁻¹ of $MnO_2@Co_3O_4$ into pH 3.0 HAc-NaAc buffer solution containing 0.25mM TMB applied in the experiments as substrates at 50°C for 8 min.



Figure. S7. The effect of free radical scavenger on the TMB-MnO₂@Co₃O₄ system. Experiments were performed by adding 2mg.mL-1 of MnO₂@Co₃O₄ into pH 3.0 HAc-NaAc buffer solution containing 0.25mM TMB applied in the experiments as substrates at 25°C for 10min.



Figure. S8. EPR spectrum for detection of O_2^{-} radicals. EPR determination was performed with 30 µl 5,5-dimethyl-1-pyrroline N-oxide (DMPO) and 30 µl of 1mg/ml MnO₂@Co₃O₄ in 100 ml deionized water/methanol as solvent. After the mixture is evenly mixed, a certain amount of the mixture is placed on a quartz tube and put into the EPR sample chamber for superoxide radical test.