

Supporting Information

Doped TiO₂-Supported IrO₂ Electrocatalyst with High Activity and Durability toward the Acidic Oxygen Evolution Reaction

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Characterization data

Figure S1. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of TiO₂@IrO₂.

Figure S2. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of V-TiO₂@IrO₂

Figure S3. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of Mn-TiO₂@IrO₂.

Figure S4. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of Fe-TiO₂@IrO₂.

Figure S5. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of Ni-TiO₂@IrO₂.

Figure S6. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of Cu-TiO₂@IrO₂.

Figure S7. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of Nb-TiO₂@IrO₂.

Figure S8. Electrocatalysis OER stability test of unsupported IrO₂, TiO₂@IrO₂, Fe-TiO₂@IrO₂, and W-TiO₂@IrO₂ (lasting for 3.1 h).

Table S1. Performance table of Ir-based electrocatalysts.

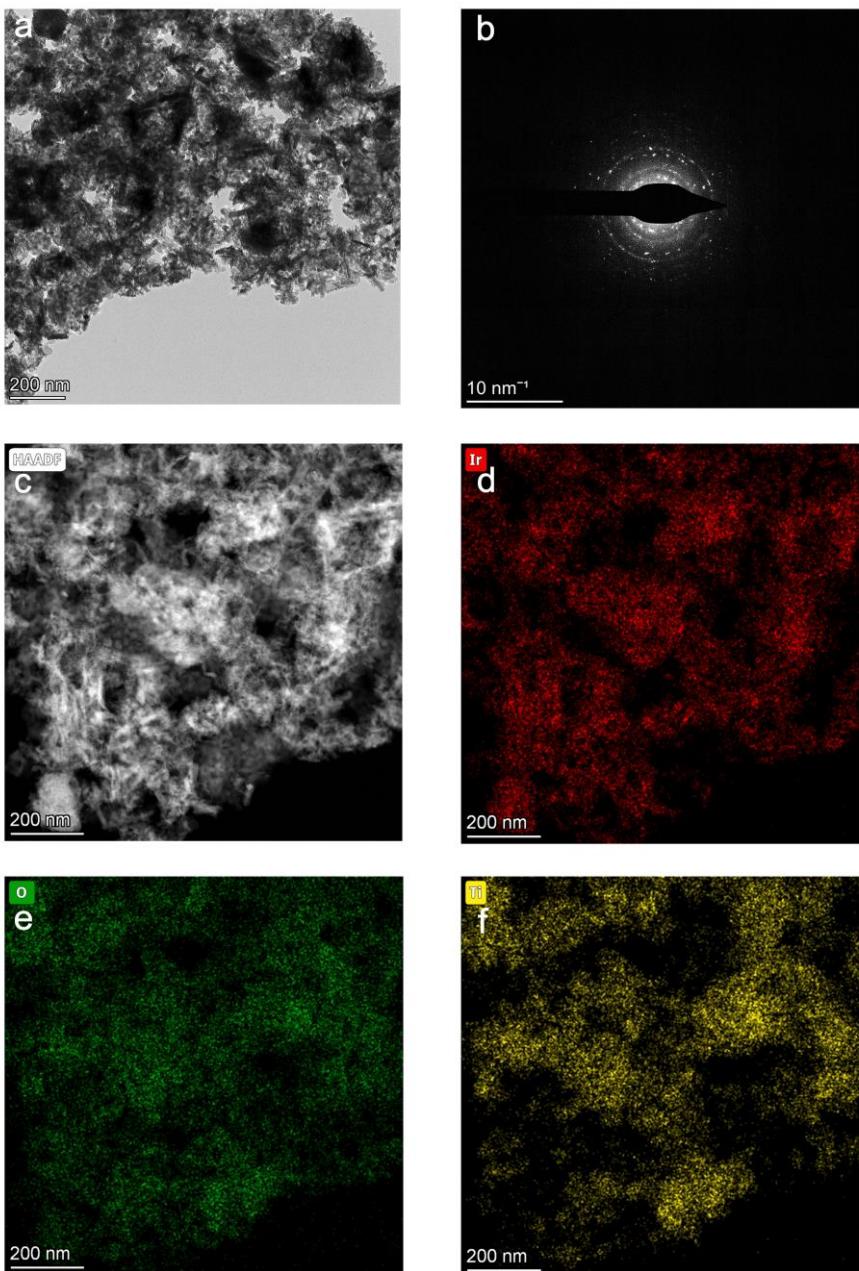


Figure S1. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of TiO_2 @ IrO_2 .

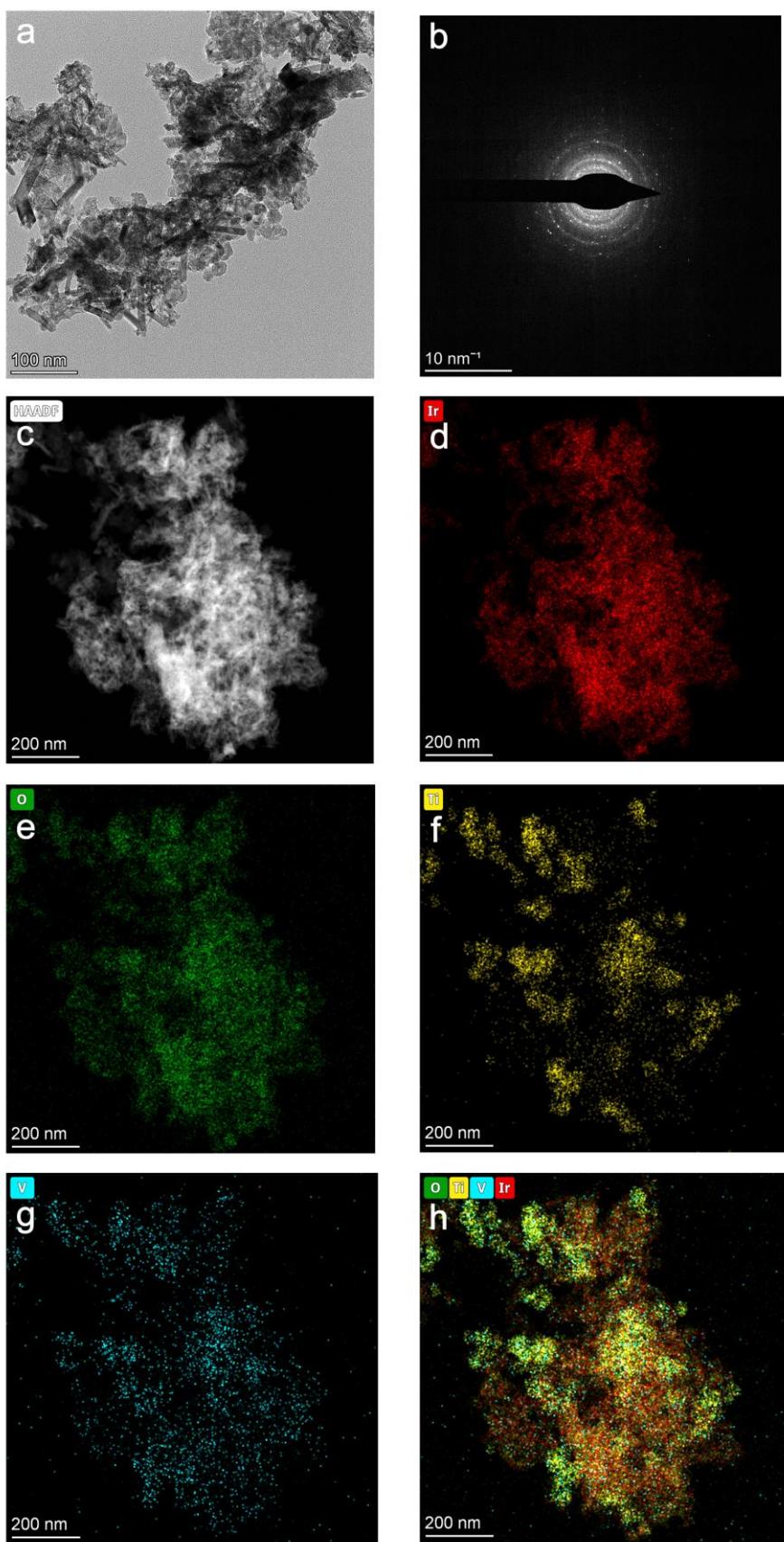


Figure S2. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of V-TiO₂@IrO₂.

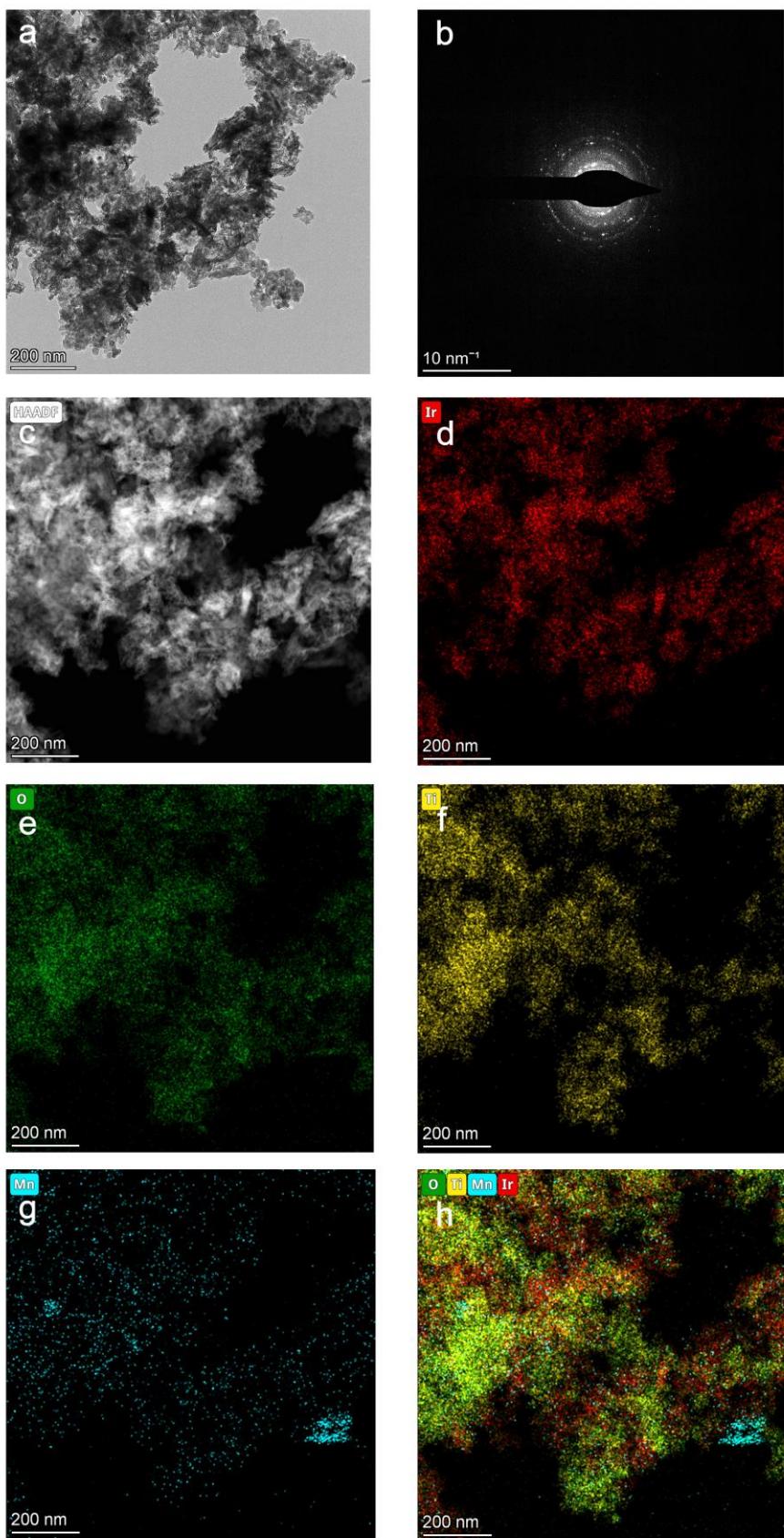


Figure S3. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of Mn-TiO₂@IrO₂.

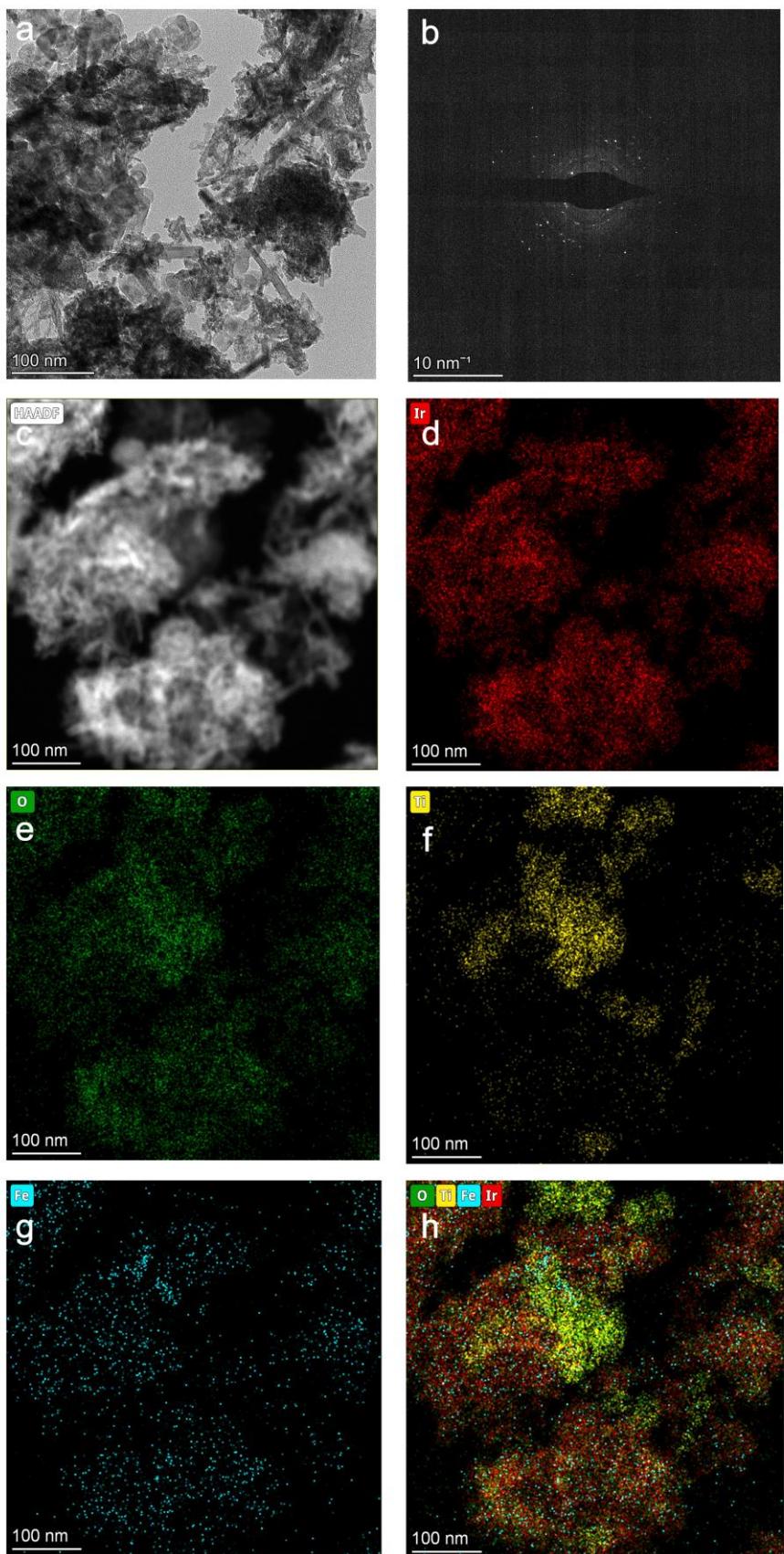


Figure S4. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of Fe-TiO₂@IrO₂.

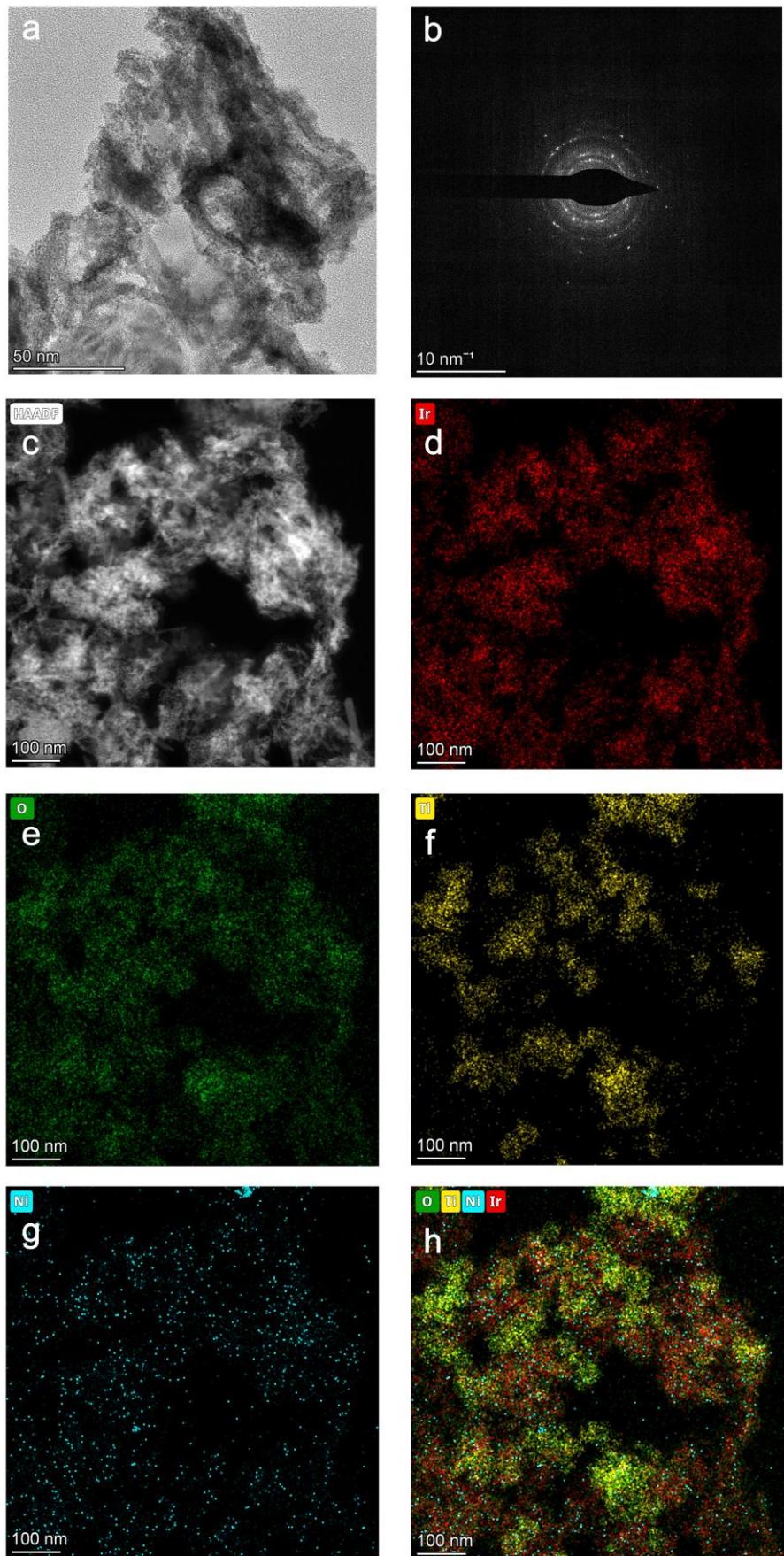


Figure S5. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of Ni-TiO₂@IrO₂.

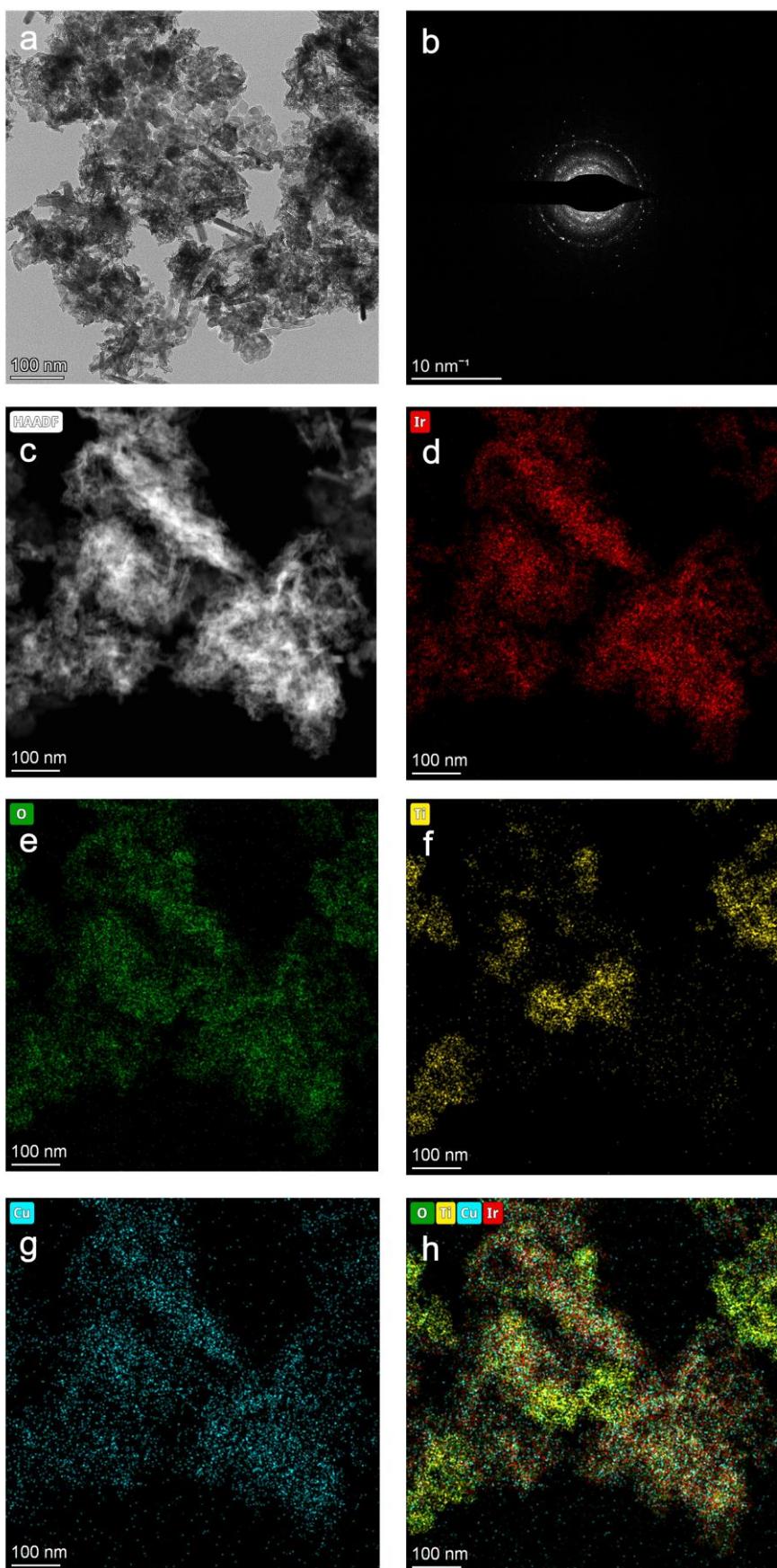


Figure S6. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of Cu-TiO₂@IrO₂.

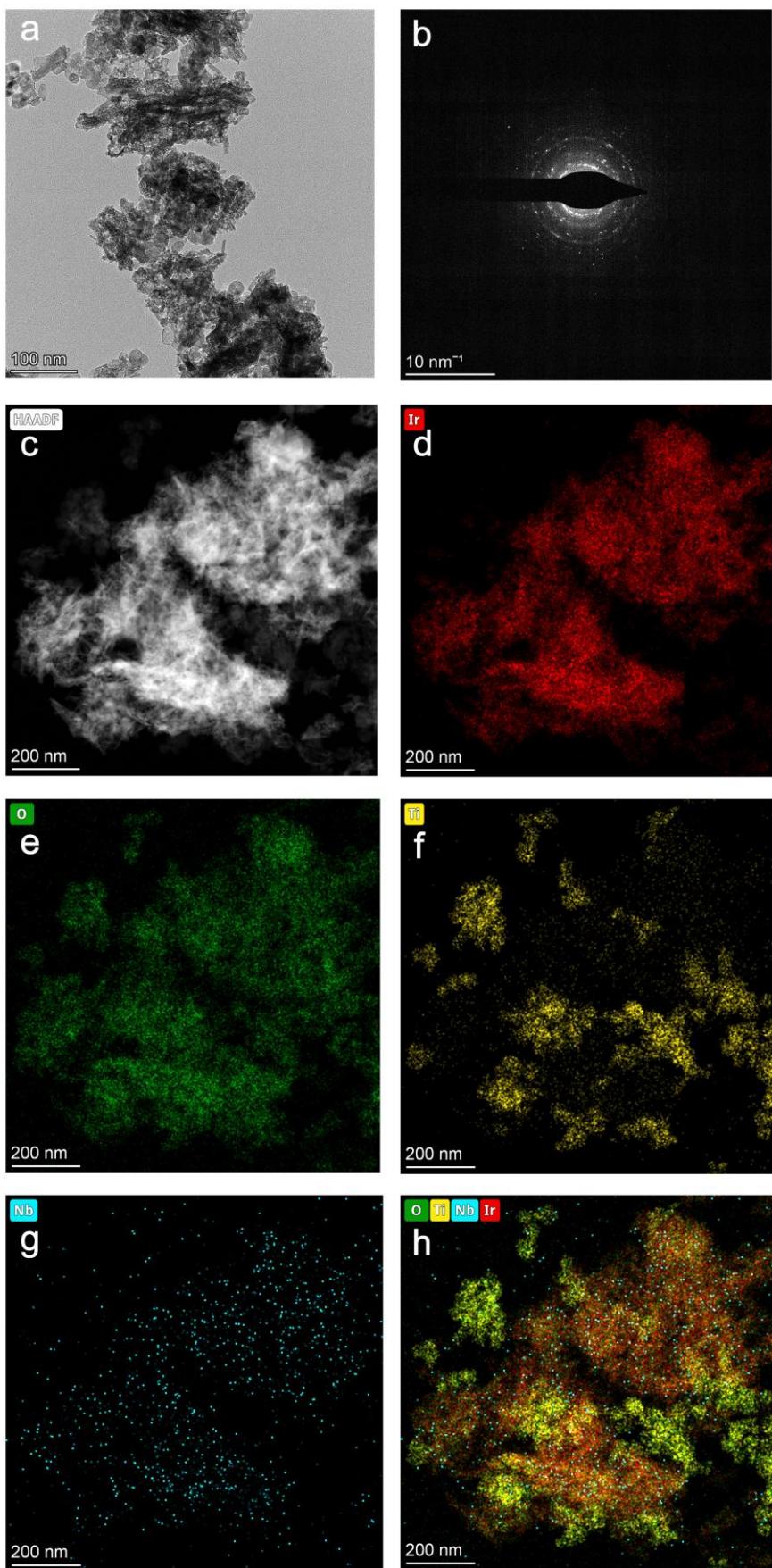


Figure S7. TEM images, SAED patterns, HAADF-STEM images and EDS elemental mappings of Nb-TiO₂@IrO₂.

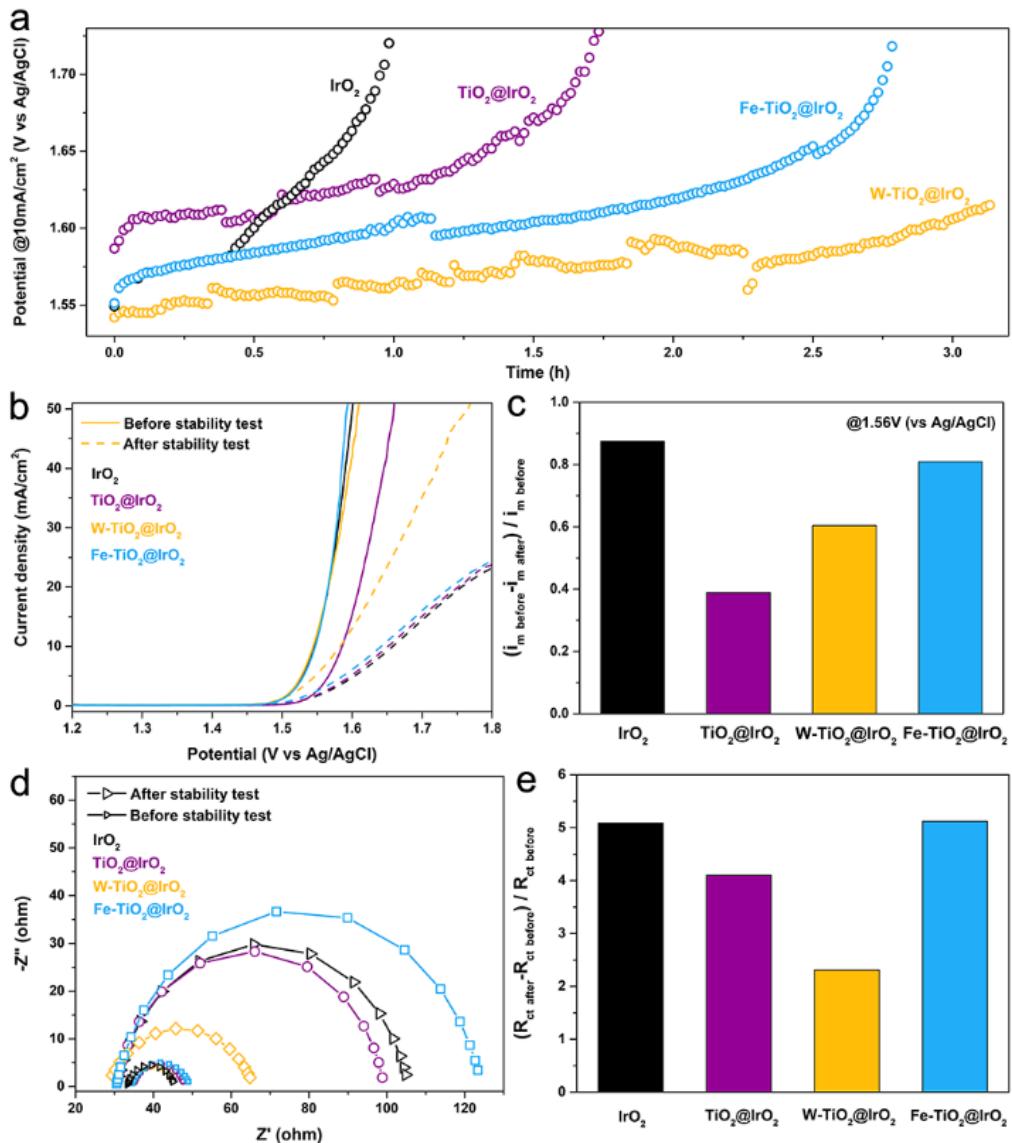


Figure S8. Electrocatalysis OER stability test of unsupported IrO_2 , $\text{TiO}_2@\text{IrO}_2$, $\text{Fe}-\text{TiO}_2@\text{IrO}_2$, and $\text{W}-\text{TiO}_2@\text{IrO}_2$ (test lasting for 3.1h). (a) Chronopotentiometry test at $10\text{mA}/\text{cm}^2_{\text{geo}}$. (b) OER activity before and after the chronopotentiometry test. (c) The attenuation degree of mass activity (i_m) before and after the chronopotentiometry test at 1.56 V (vs Ag/AgCl). (d) Nyquist plots before and after chronopotentiometry test at the potential of $10\text{ mA}/\text{cm}^2_{\text{geo}}$. (e) The increased degree of R_{ct} before and after the chronopotentiometry test corresponding to the Nyquist plots.

Table S1. Performance table of Ir-based electrocatalysts.

Catalyst	Electrolyte	Overpotential @10mA/cm ² (mV)	Tafel slope (mV/dec)	Mass activity (A/mg _{Ir}) @Overpotential (v)	Ref.
W-TiO ₂ /IrO ₂	0.1M HClO ₄	308	42.36	0.72@330	This work
IrO ₂	0.1M HClO ₄	373	112	0.0126@300	[1]
IrO ₂ ns	0.5M H ₂ SO ₄	350	57	0.437@xx	[2]
Ni&Co-IrO ₂	0.1M HClO ₄	~280	53	0.055@270	[3]
TiO ₂ /IrO ₂	0.1M HClO ₄	255@1mA/cm ²	42	0.07@295	[4]
Nb-TiO ₂ /IrO ₂	0.1M HClO ₄	310	/	/	[5]
Nb _{0.05} TiO _{0.95} O ₂ /IrO ₂	0.5M H ₂ SO ₄	270@1mA/cm ²	282	0.471@370	[6]
W _x Ti _{1-x} O ₂ /Ir	0.1M HClO ₄	~300	/	~0.77@570	[7]
TiN/IrO ₂	0.5M H ₂ SO ₄	313	65.5	0.874@370	[8]
IrNiCu DNF/C	0.1M HClO ₄	307	48	0.053@300	[9]
IrCoNi PHNCs	0.1M HClO ₄	303	53.8	0.7@300	[10]
P-IrCu _{1.4} NCs	0.05M H ₂ SO ₄	311	53.9	0.213@320	[11]
SrCo _{0.9} Ir _{0.1} O _{3-δ}	0.1M HClO ₄	310	/	/	[12]
RuIr	0.1M HClO ₄	344	111.5	/	[13]
IrOOH NSs	0.1M HClO ₄	344	58	/	[14]
Sputtered IrO _x films	0.1M HClO ₄	490	100	/	[15]

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