

Supplementary Materials

Cd(II) Coordination Polymers Constructed from Tris-pyridyl-tris-amide and Polycarboxylic acid: Synthesis, Structures and Sensing Properties

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Fig. S1. IR spectrum of complex 1.

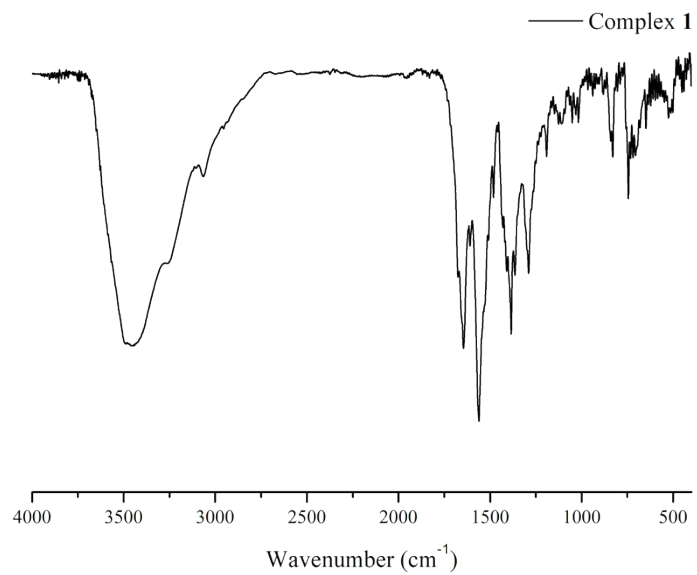


Fig. S2. IR spectrum of complex **2**.

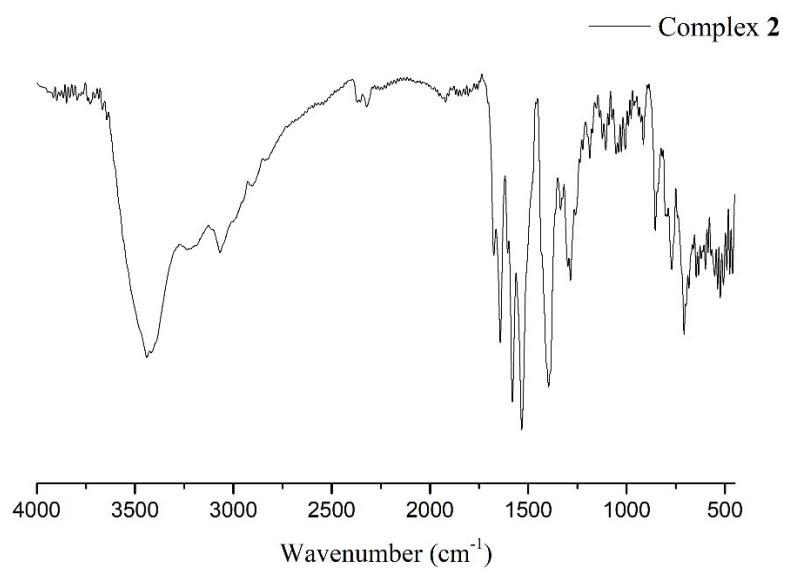


Fig. S3. IR spectrum of complex **3**.

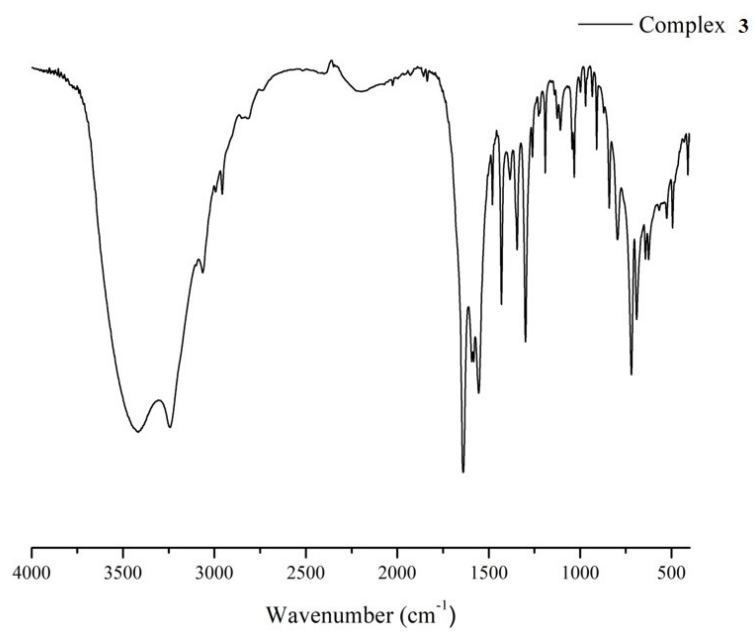


Fig. S4. The PXRD patterns of complex **1**.

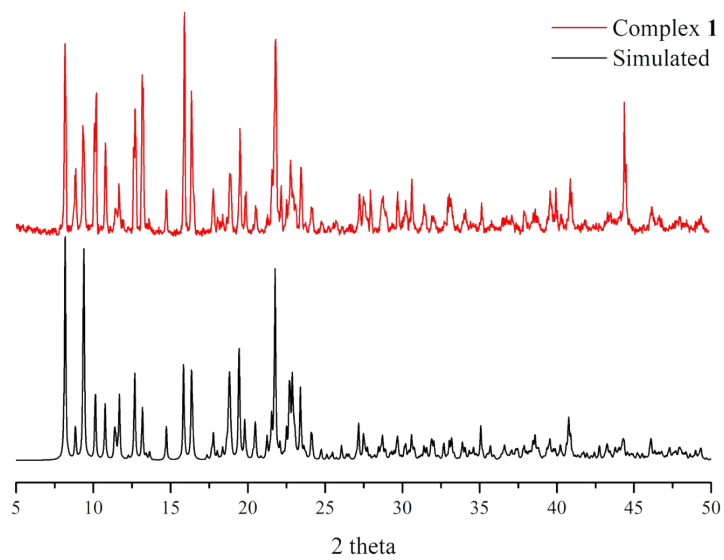


Fig. S5. The PXRD patterns of complex **2**.

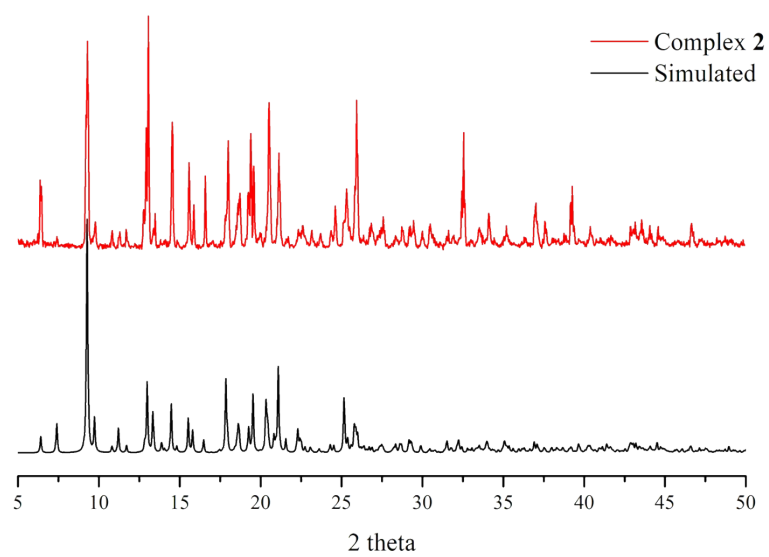


Fig. S6. The PXRD patterns of complex **3**.

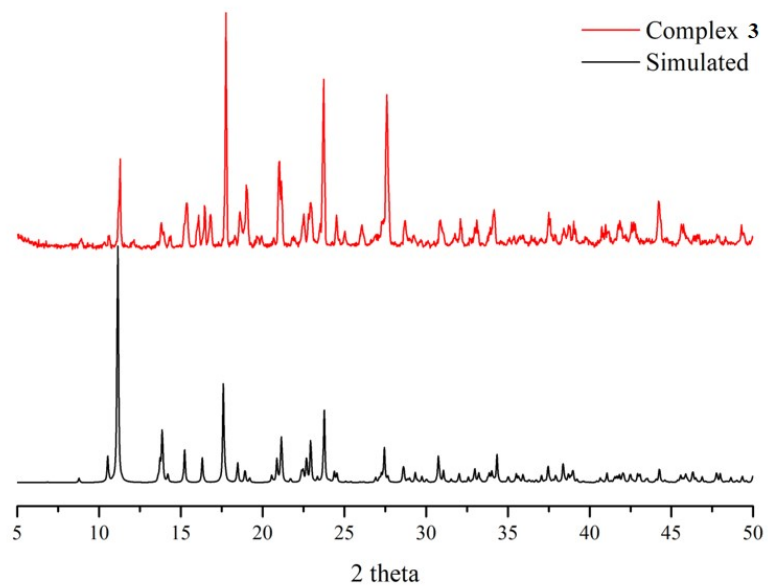


Fig. S7. TGA curve of complex **1**.

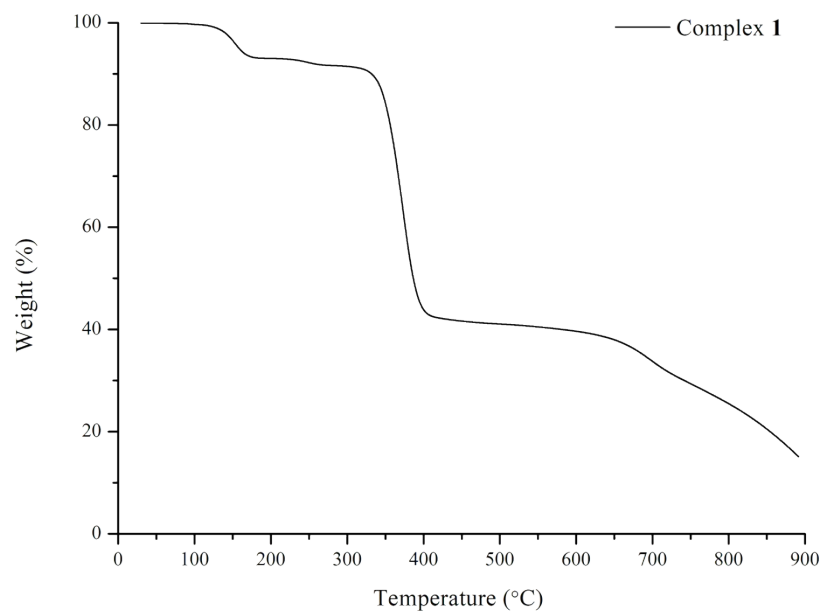


Fig. S8. TGA curve of complex **2**.

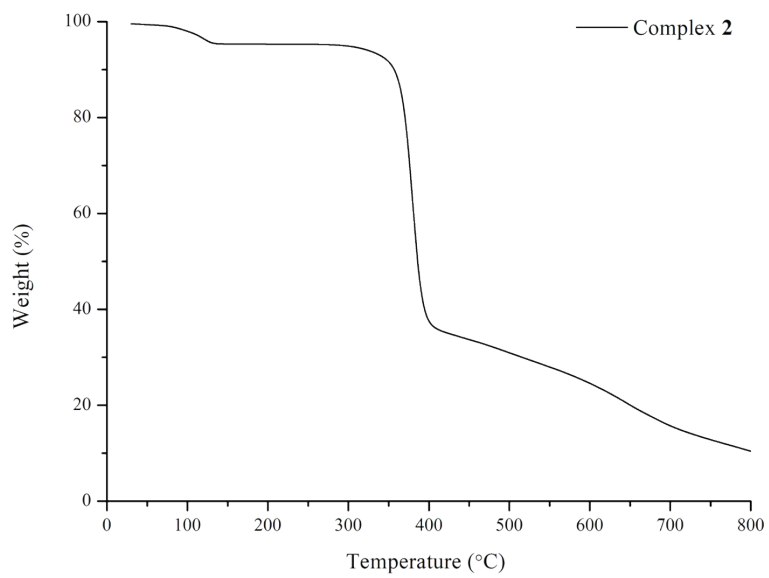


Fig. S9. TGA curve of complex **3**.

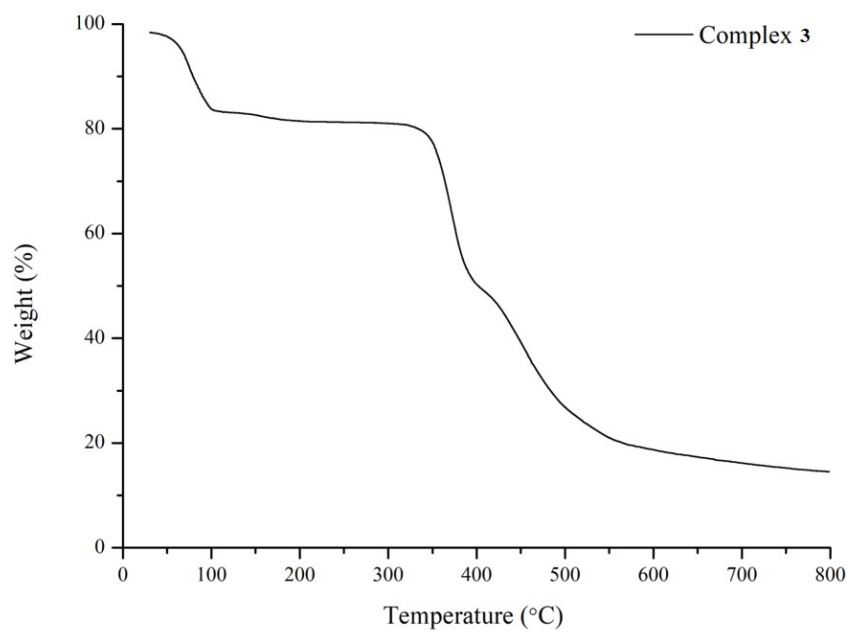


Fig. S10. The PXRD patterns of complex **1** before and after immersion in different solvents.

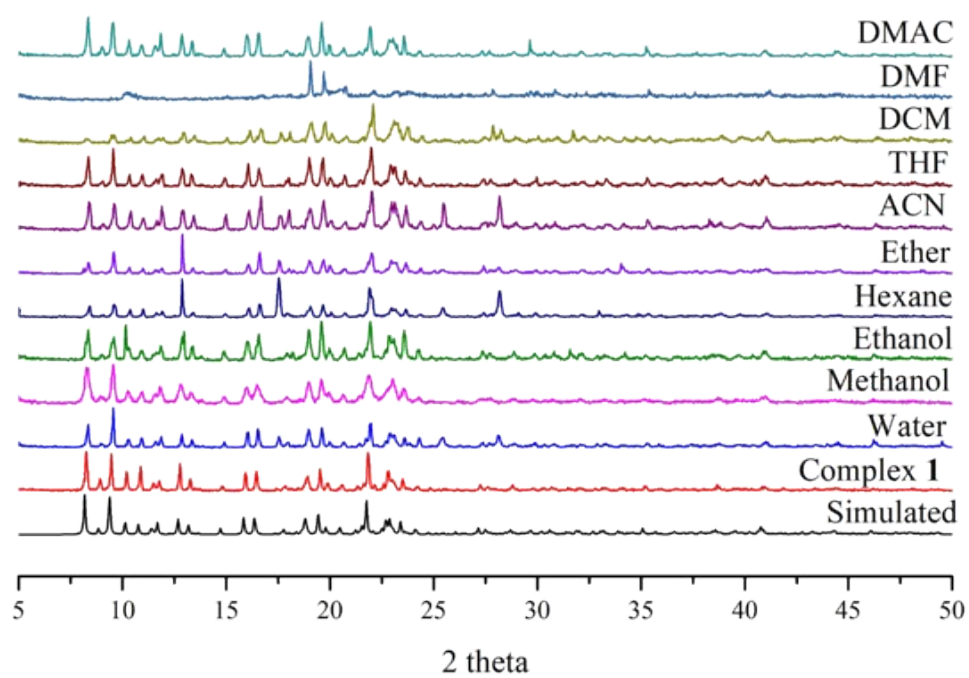


Fig. S11. The PXRD patterns of complex **2** before and after immersion in different solvents.

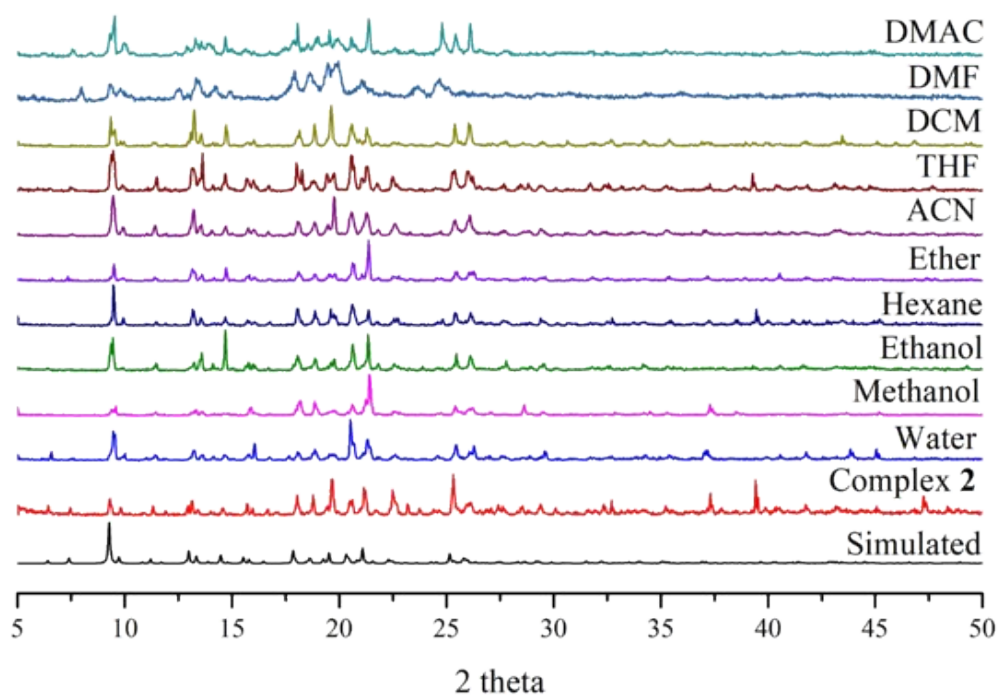


Fig. S12. The photoluminescence spectra of **L**.

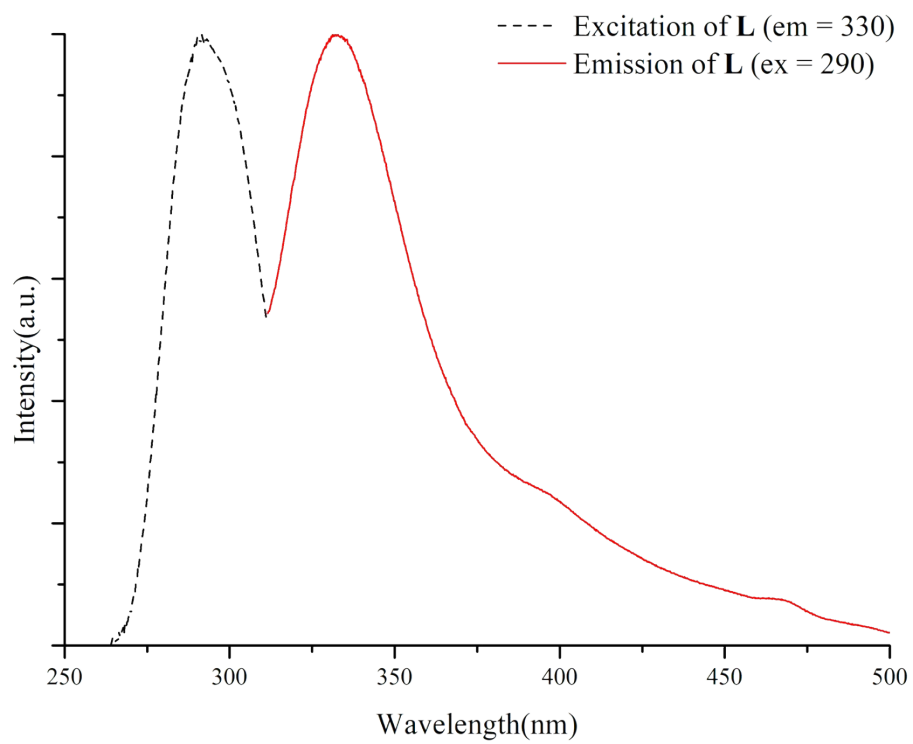


Fig. S13. The photoluminescence spectra of 1,4-H₂BDC.

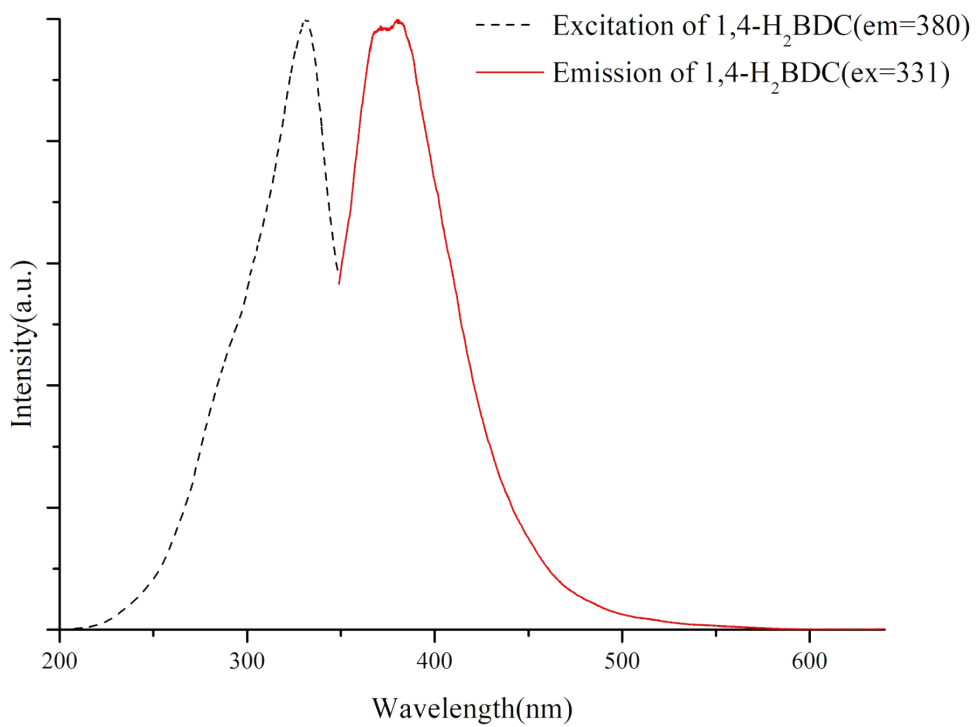


Fig. S14. The photoluminescence spectra of 4,4'-H₂BDC.

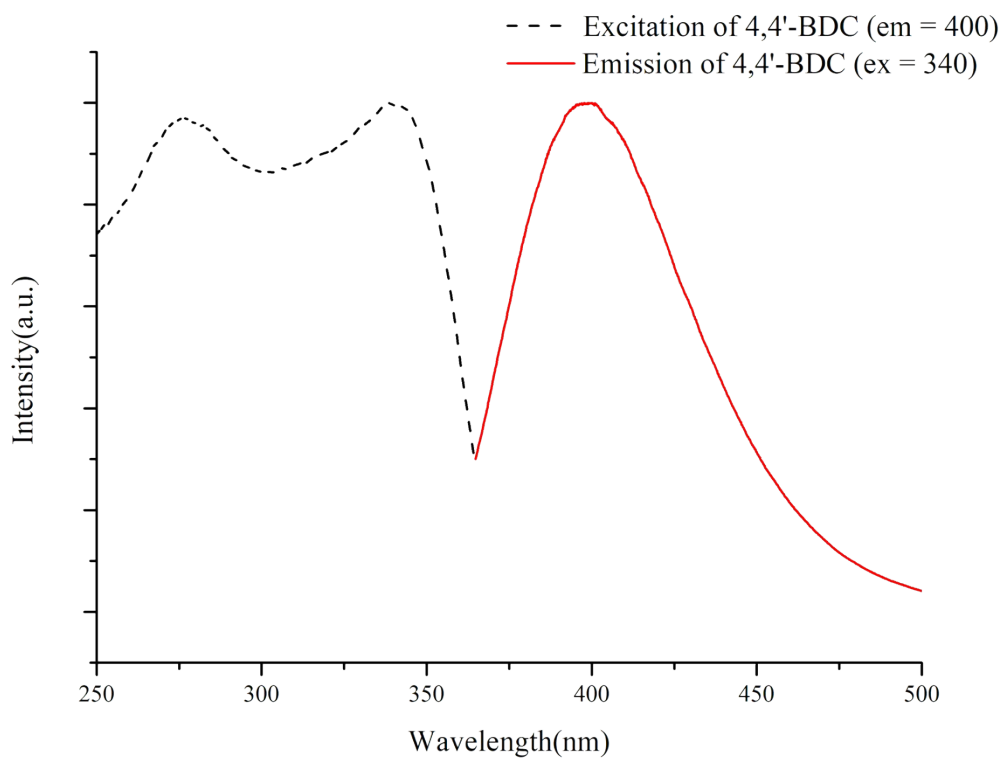


Fig. S15. The photoluminescence spectra of complex **1**.

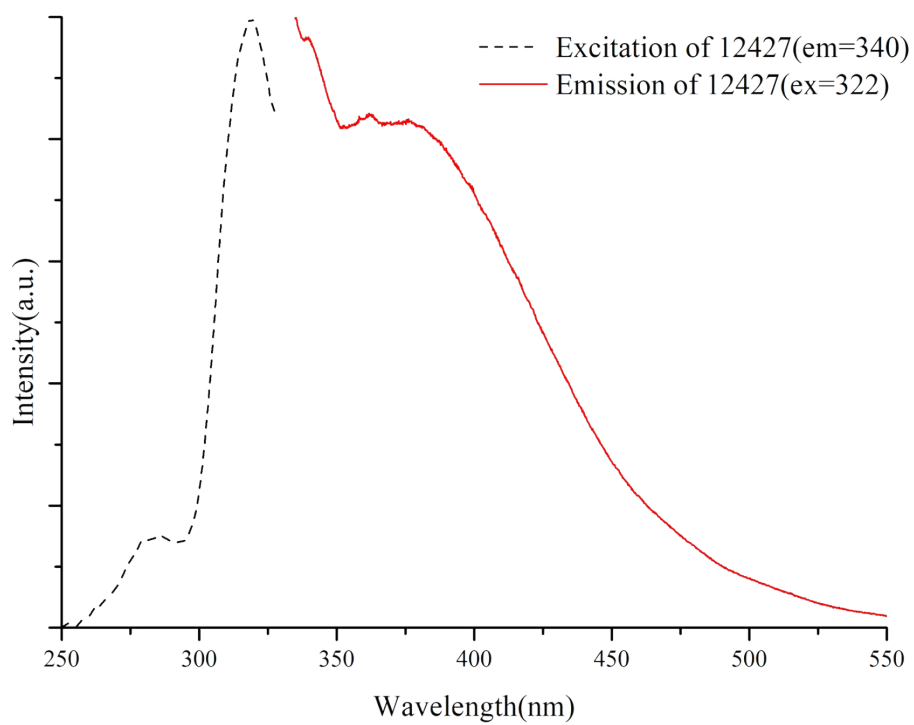


Fig. S16. The photoluminescence spectra of complex **2**.

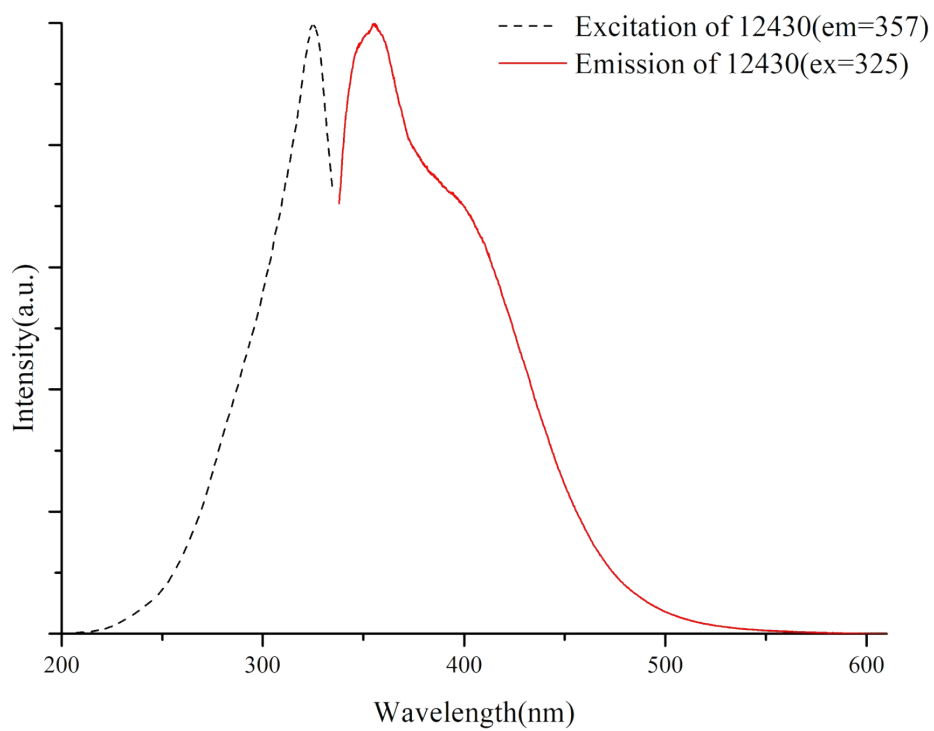


Fig. S17. The solid-state luminescence spectra of complex **1** before and after immersion into different metal cations.

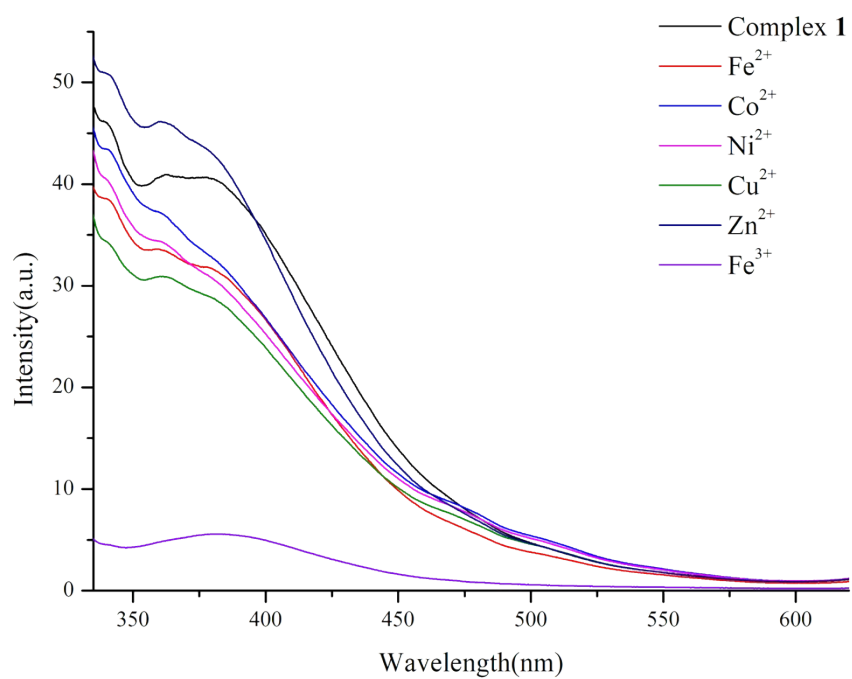


Fig. S18. The solid-state luminescence spectra of complex **2** before and after immersion into different metal cations.

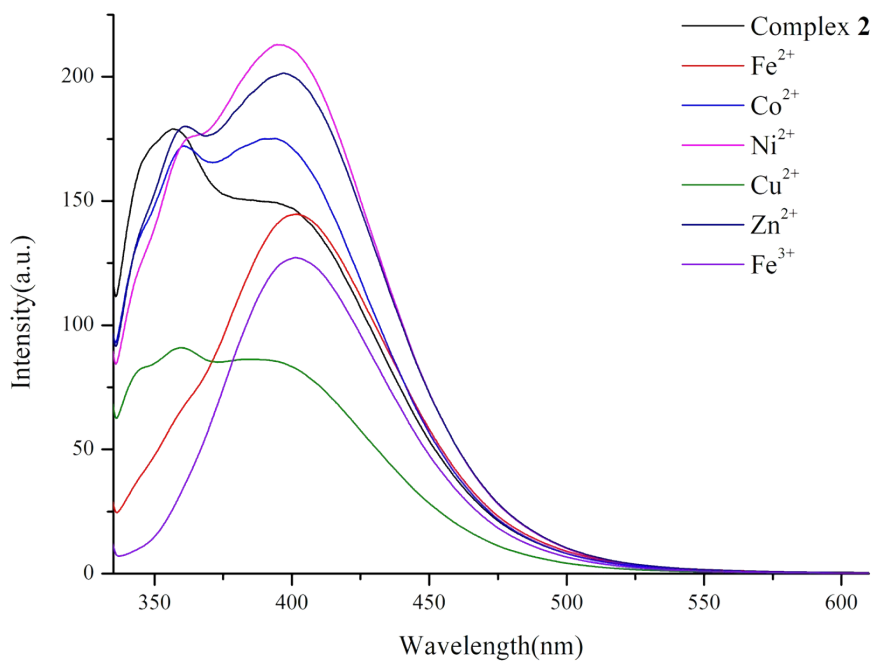


Fig. S19. The emission spectra of complex **1** in different concentrations of Fe^{3+} ion.

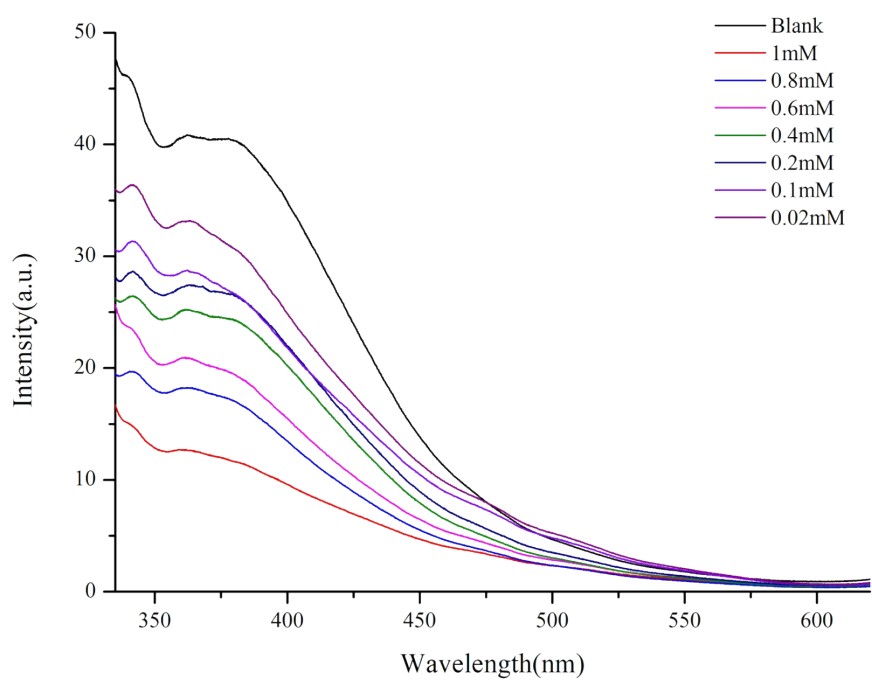


Fig. S21. PXRD patterns for complex **1** after Fe³⁺ sensing for five cycles.

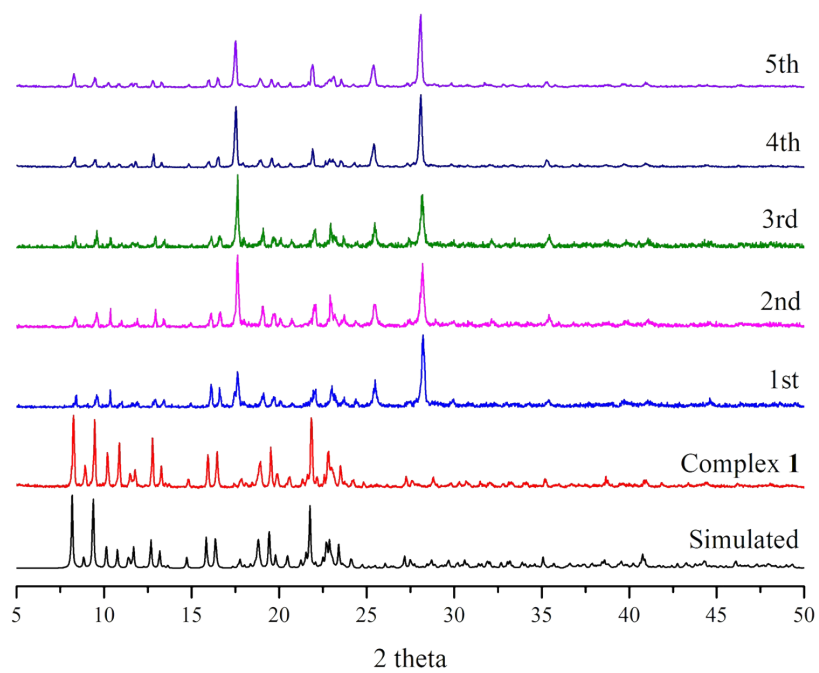


Fig. S22. PXRD patterns for complex **2** after Fe³⁺ sensing for five cycles.

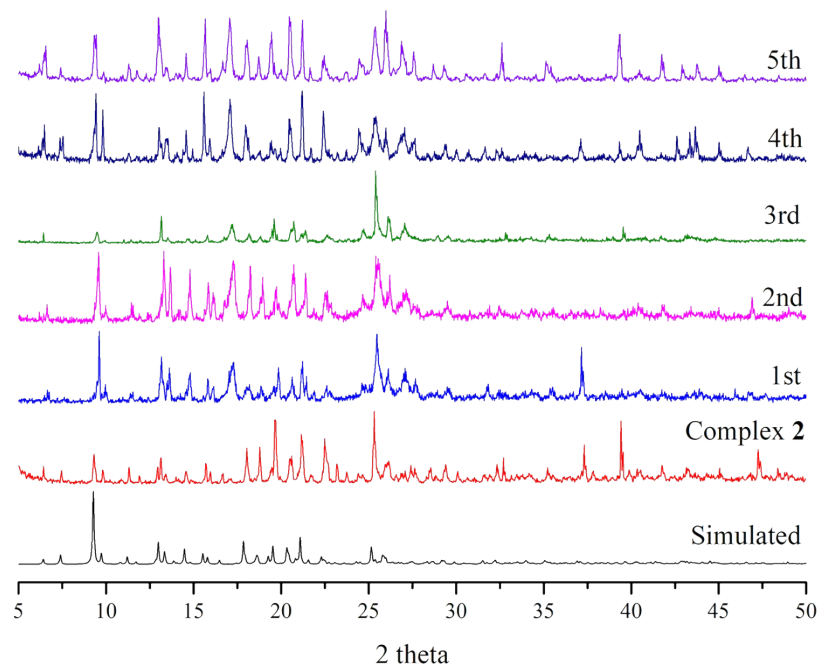


Fig. S23. The PXRD patterns of complex **1** before and after immersion in different metal cations.

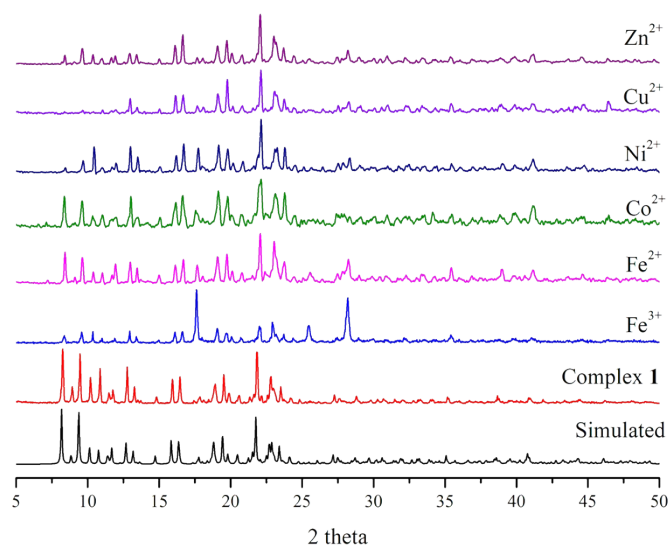


Fig. S24. The PXRD patterns of complex **2** before and after immersion in different metal cations.

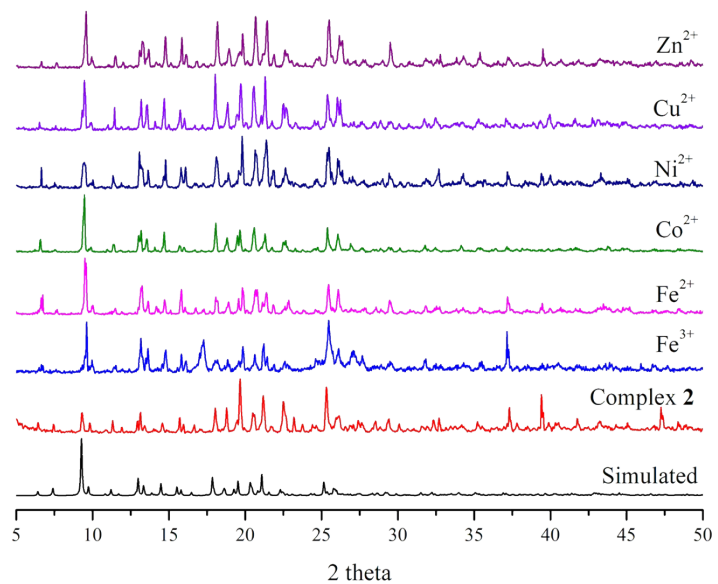
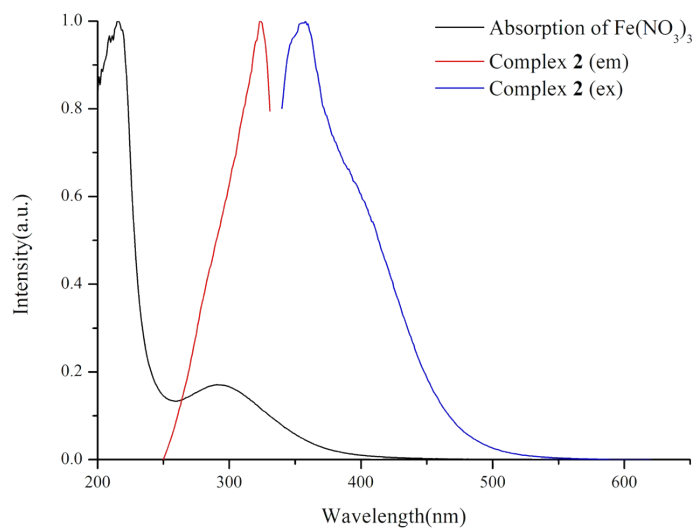
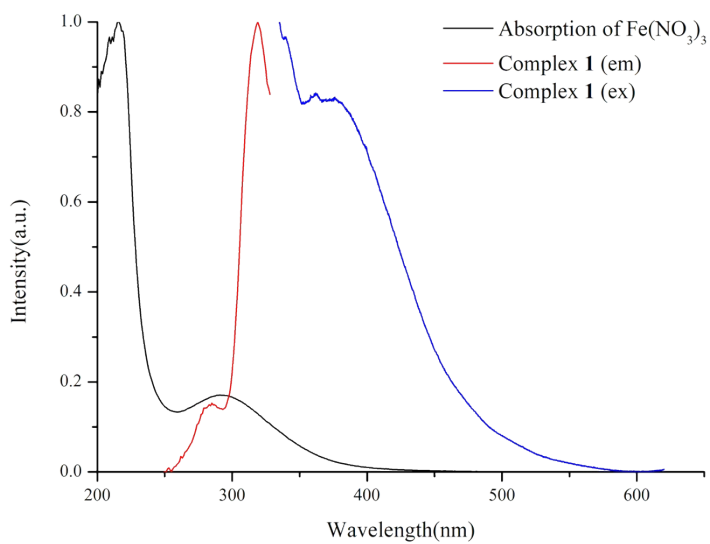


Fig. S25. UV-vis absorption spectra of Fe^{3+} and the excitation and emission spectra of (a) **1** and (b) **2**.



(b)



(a)

Fig. S26. The EDX data of complex **1**. (a) The selected part for point scan. (b) The result of point scan. (c) The element mapping scan of Fe³⁺. (d) The element percentage analysis of point scan.

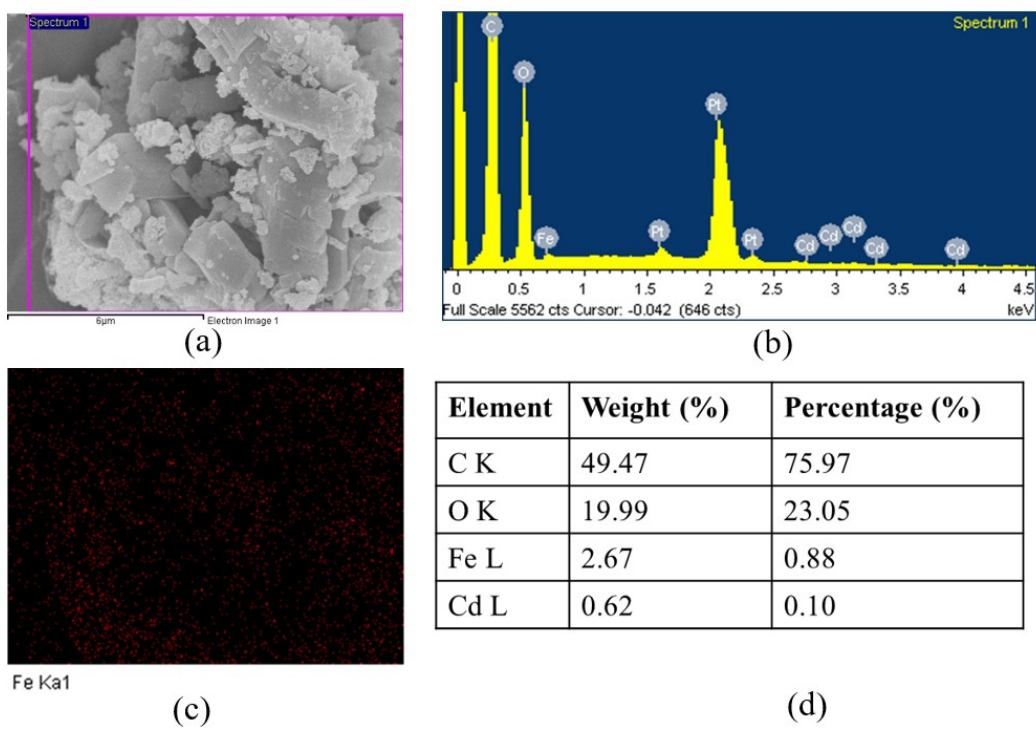
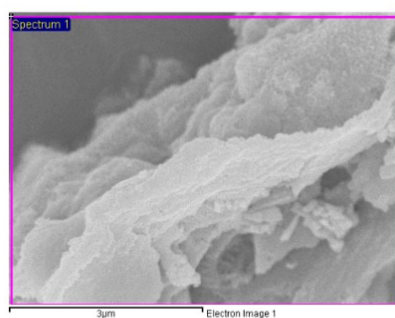
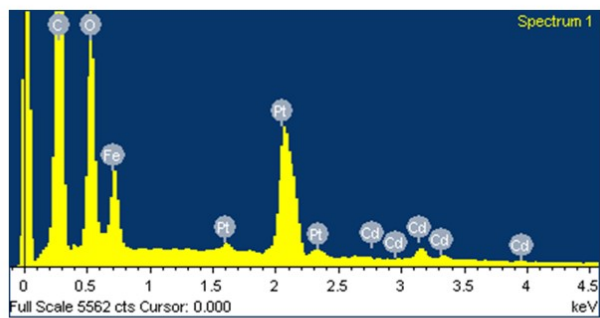


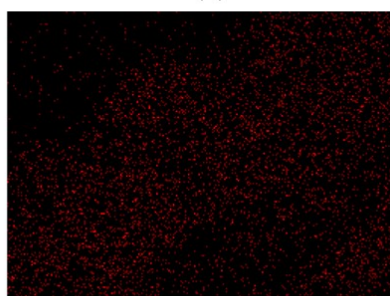
Fig. S27. The EDX data of complex **2**. (a) The selected part for point scan. (b) The result of point scan. (c) The element mapping scan of Fe³⁺. (d) The element percentage analysis of point scan.



(a)



(b)



Fe Kα1

(c)

Element	Weight (%)	Percentage (%)
C K	38.66	67.49
O K	18.02	23.63
Fe L	22.40	8.41
Cd L	2.60	0.49

(d)

Fig. S28. IR spectra of **L**, 1,4-H₂BDC and complex **1** before and after immersion in Fe³⁺.

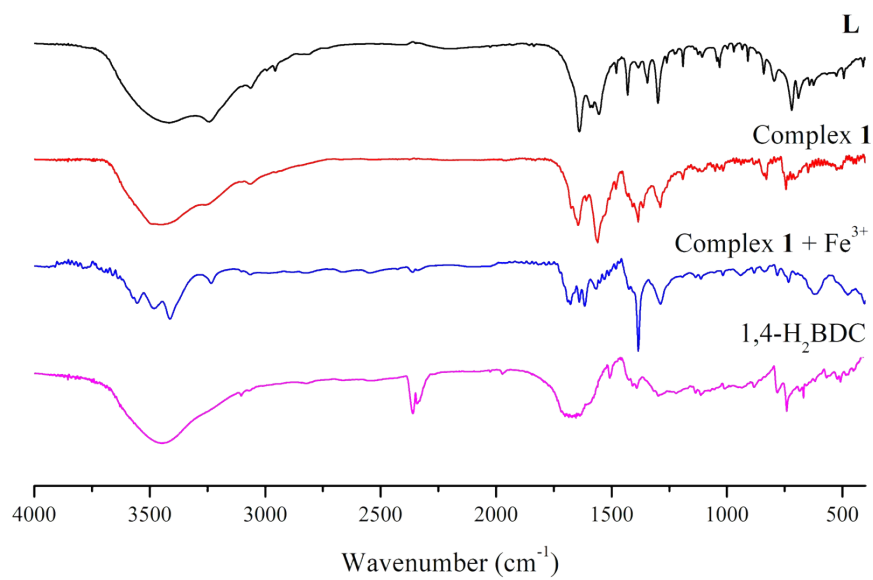


Fig. S29. IR spectra of **L**, 4,4'-H₂BDC and complex **2** before and after immersion in Fe³⁺.

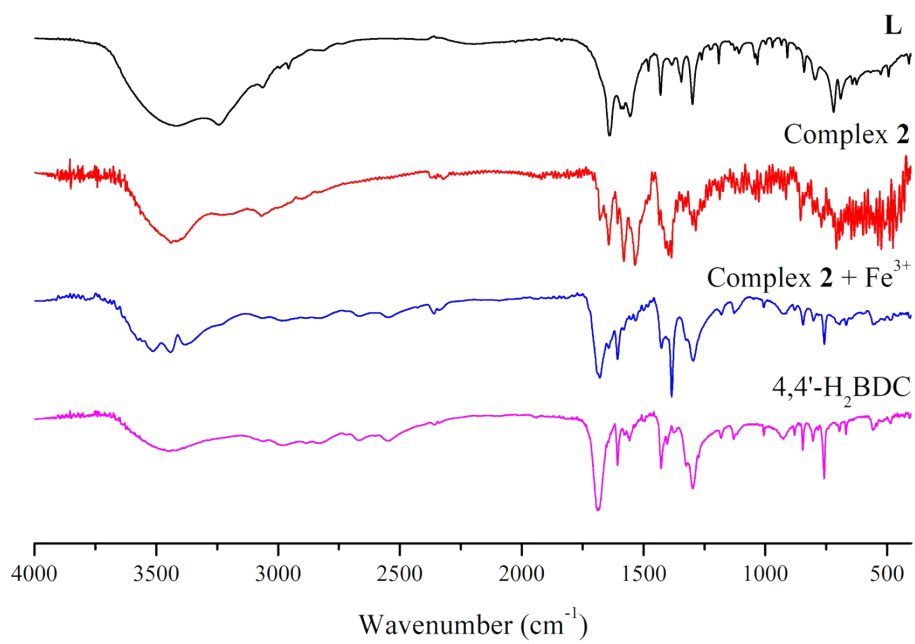


Fig. S30. Selected XPS patterns before and after addition of Fe^{3+} ion to complex **2**.

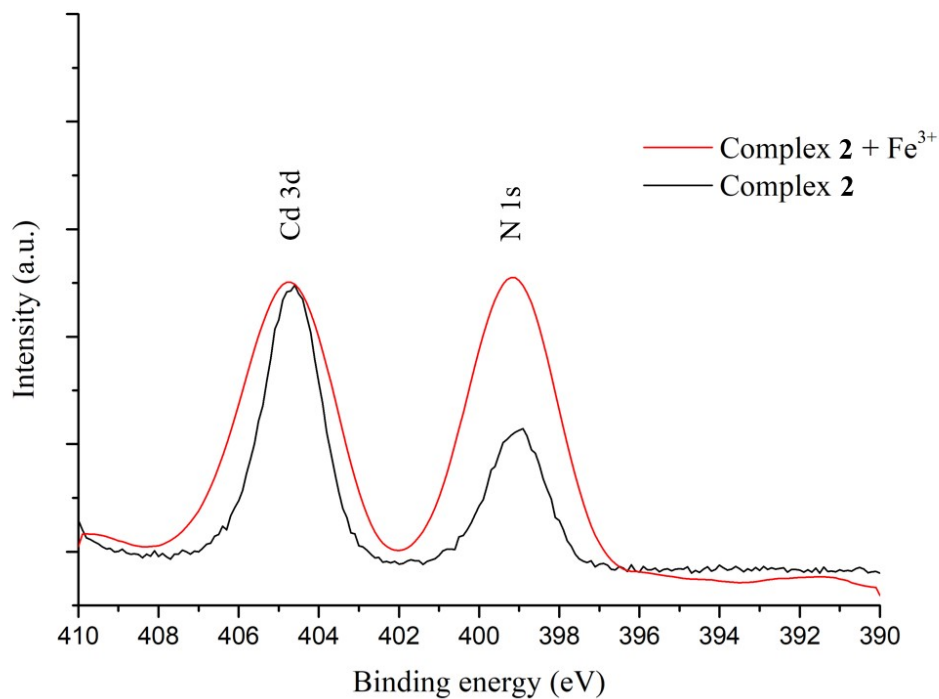
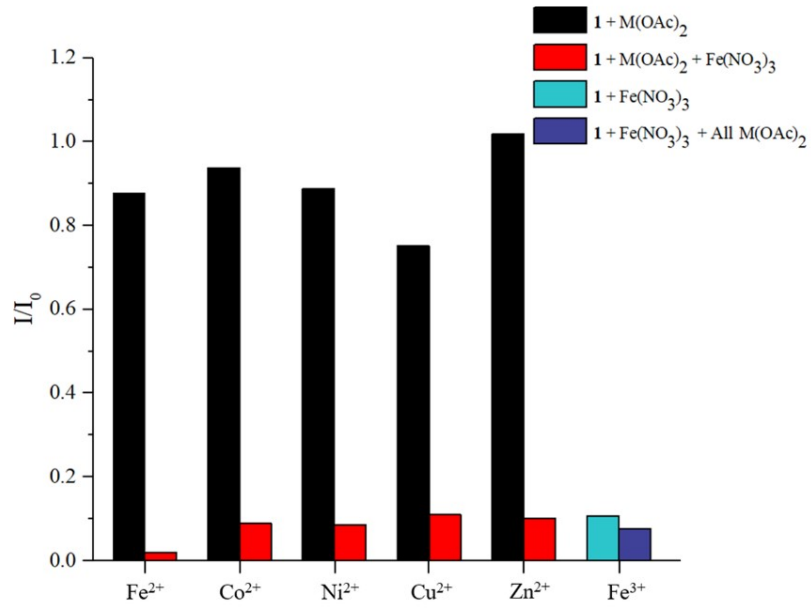
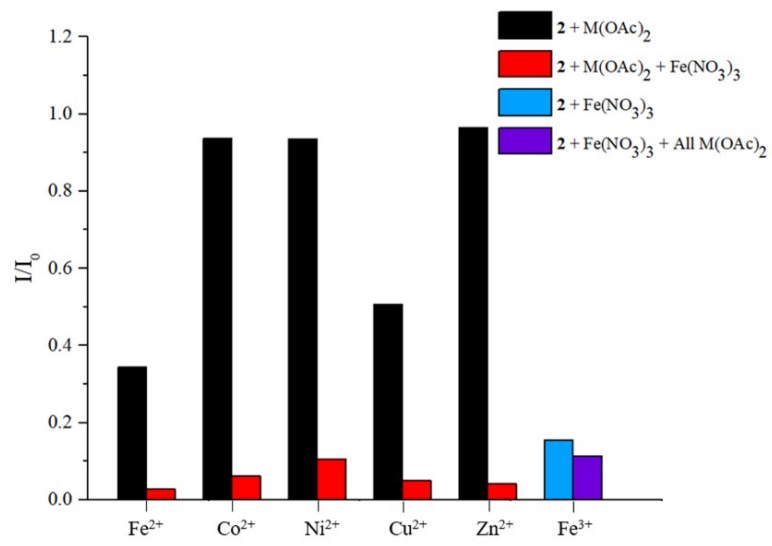


Fig. S31. Competitive sensing experiments in the presence of the other cations for complexes (a) **1** and (b) **2**.



(a)



(b)

Fig. S32. (a) N₂ adsorption–desorption isotherms for complex **1**. (b) Pore-size distribution curve for complex **1**.

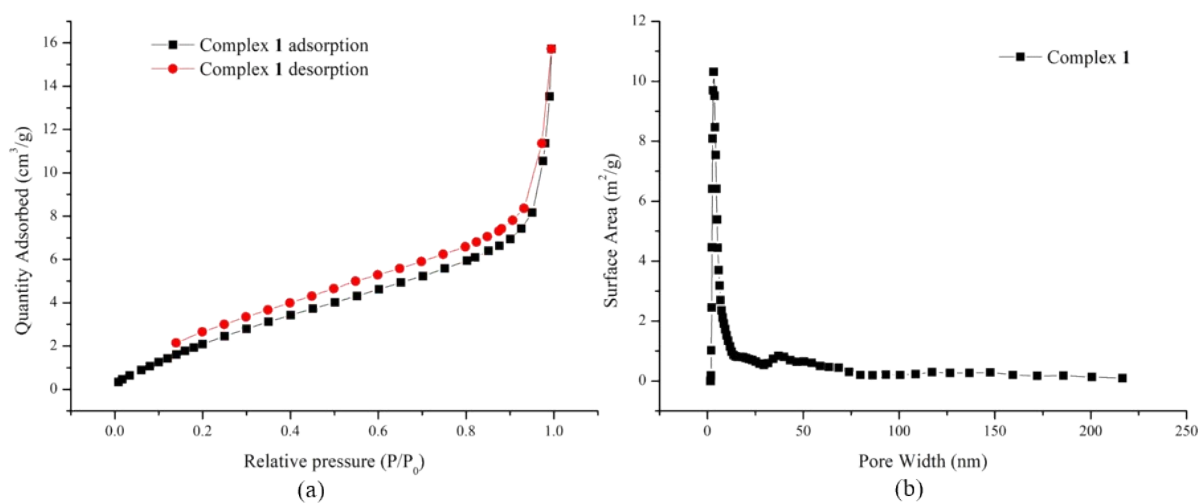


Fig. S33. (a) N₂ adsorption–desorption isotherms for complex **2**. (b) Pore-size distribution curve for complex **2**.

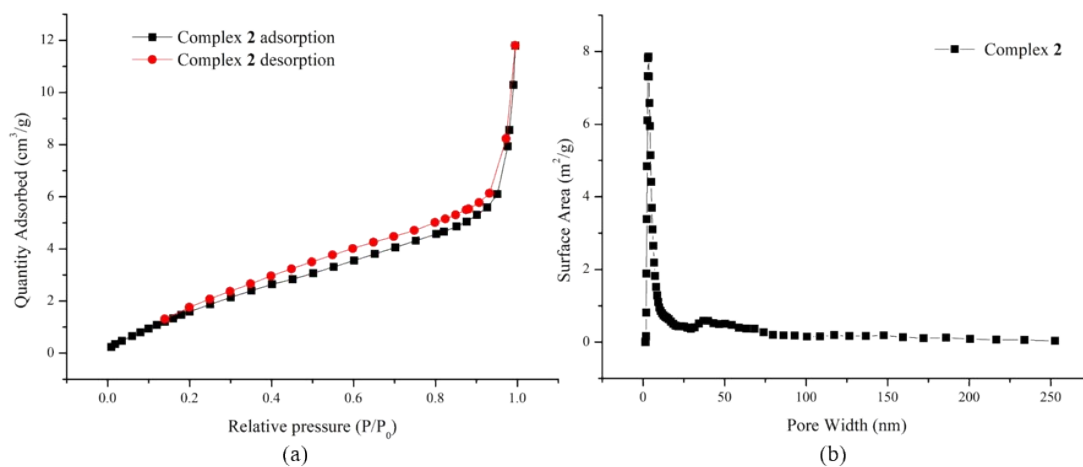


Fig. S34. PXRD patterns for complex **1** before and after N₂ adsorption and desorption.

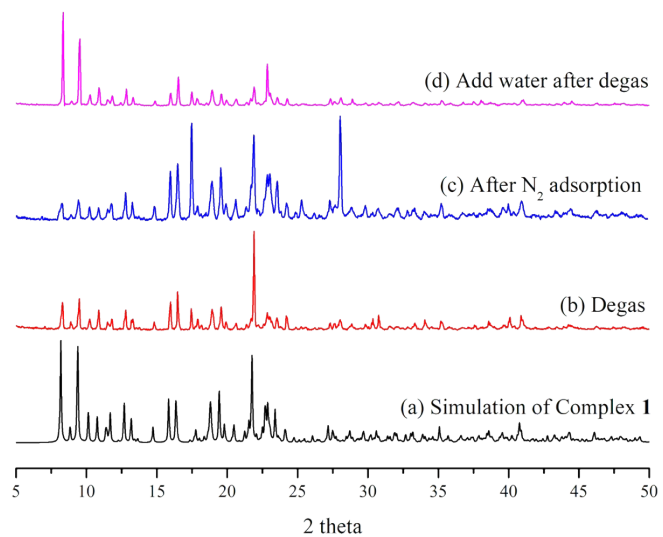


Fig. S35. PXRD patterns for complex **2** before and after N₂ adsorption and desorption.

