

Electronic Supplementary Information

for **Molecular heterometallic clusters** $[\text{Re}_{6-x}\text{Mo}_x\text{S}_8\text{L}_6]$ ($x=2-3$, $L = \text{py}, \text{etpy}, \text{tbp}$):
negligible influence of the x value on the crystal structures

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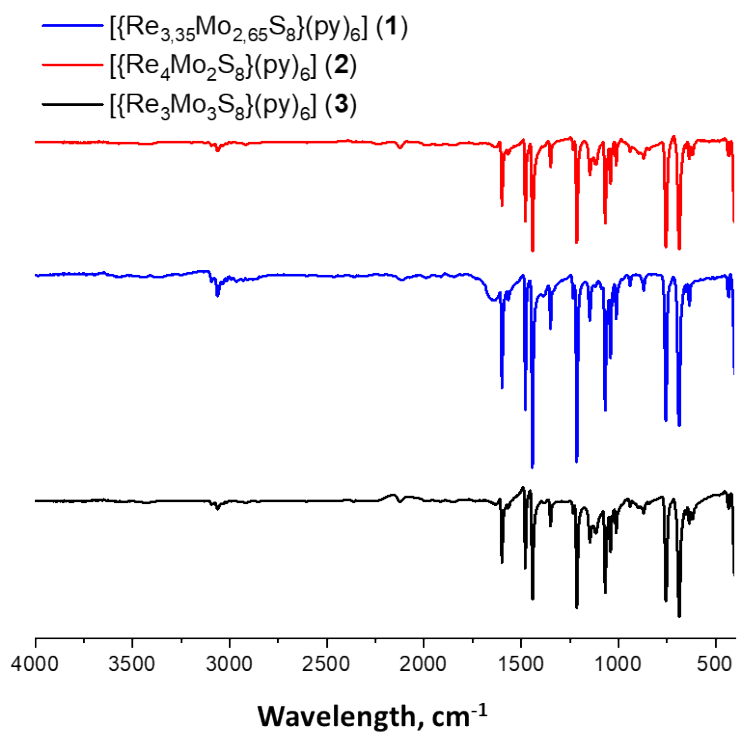


Figure S1. IR spectra for compounds 1-3.

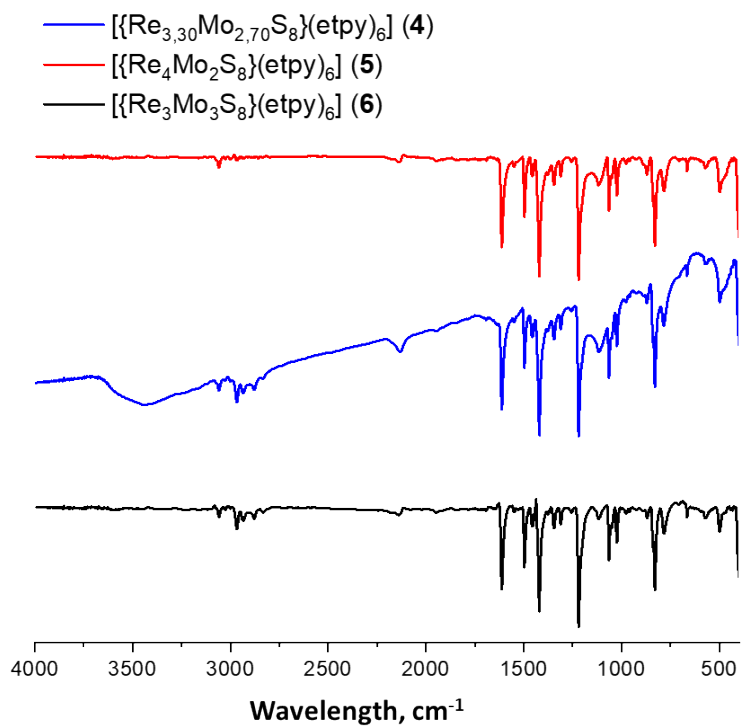


Figure S2. IR spectra for compounds (4-6).

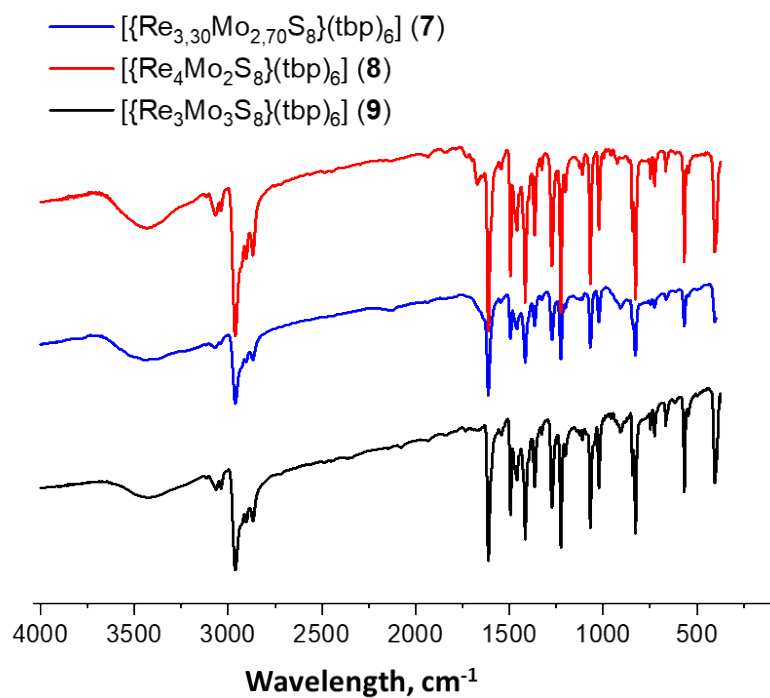


Figure S3. IR spectra for compounds (7-9).

Table S1. Crystallographic data and details of the structure refinements for the compounds **1–3**.

| | 1 | 2 | 3 |
|-----------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------|
| Chemical formula | C ₃₀ H ₃₀ N ₆ Re _{3.35} Mo _{2.65} S ₈ | C ₃₀ H ₃₀ N ₆ Re ₄ Mo ₂ S ₈ | C ₃₀ H ₃₀ N ₆ Re ₃ Mo ₃ S ₈ |
| Molecular weight, g/mol | 1609.09 | 1667.76 | 1577.50 |
| Space group | <i>P</i> $\bar{1}$ | <i>P</i> $\bar{1}$ | <i>P</i> $\bar{1}$ |
| <i>a</i> , Å | 9.3458(4) | 9.3333(2) | 9.3362(8) |
| <i>b</i> , Å | 10.5903(4) | 10.5520(2) | 10.5758(8) |
| <i>c</i> , Å | 11.8441(5) | 11.8701(2) | 11.8771(9) |
| α , ° | 114.4320(10) | 114.4530(10) | 114.524(2) |
| β , ° | 90.7490(10) | 91.1010(10) | 91.007(3) |
| γ , ° | 109.1920(10) | 108.9960(10) | 109.100(3) |
| <i>V</i> , Å ³ | 992.72(7) | 989.99(3) | 992.01(14) |
| <i>Z</i> | 1 | 1 | 1 |
| $\rho_{\text{calcd.}}$, g·cm ⁻³ | 2.692 | 2.797 | 2.641 |
| μ , mm ⁻¹ | 11.436 | 13.246 | 10.488 |
| θ range, ° | 2.23 – 26.39 | 2.28 – 30.49 | 2.60 – 29.55 |
| Miller indices range | -11 ≤ <i>h</i> ≤ 9 -12 ≤ <i>k</i> ≤ 13 -14 ≤ <i>l</i> ≤ 14 | -12 ≤ <i>h</i> ≤ 13 -15 ≤ <i>k</i> ≤ 15 -16 ≤ <i>l</i> ≤ 16 | -12 ≤ <i>h</i> ≤ 12 -14 ≤ <i>k</i> ≤ 14 -16 ≤ <i>l</i> ≤ 14 |
| Reflections collected/unique | 10546/4061 | 15877/5997 | 15390/5504 |
| Reflections (<i>I</i> >2σ(<i>I</i>)) | 3615 | 5315 | 5133 |
| <i>R</i> _{int} | 0.0218 | 0.0297 | 0.0273 |
| GoF | 1.044 | 1.023 | 1.059 |
| <i>R</i> ₁ / <i>wR</i> ₂ (<i>I</i> > 2σ(<i>I</i>)) | <i>R</i> ₁ = 0.0195 <i>wR</i> ₂ = 0.0402 | <i>R</i> ₁ = 0.0274 <i>wR</i> ₂ = 0.0493 | <i>R</i> ₁ = 0.0217 <i>wR</i> ₂ = 0.0479 |
| <i>R</i> ₁ / <i>wR</i> ₁ (all data) | <i>R</i> ₁ = 0.0240 <i>wR</i> ₂ = 0.0418 | <i>R</i> ₁ = 0.0324 <i>wR</i> ₂ = 0.0515 | <i>R</i> ₁ = 0.0239 <i>wR</i> ₂ = 0.0488 |
| $\Delta\rho_{\text{max}}/\Delta\rho_{\text{min}}$ (e·Å ³) | 0.842/-0.655 | 1.077/-1.110 | 1.115/-1.063 |

Table S2. Crystallographic data and details of the structure refinements for the compounds **4–6**.

| | 4 | 5 | 6 |
|-----------------------------------------------------------------------------|---------------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------|
| Chemical formula | C ₄₂ H ₅₄ N ₆ Re _{3.3} Mo _{2.7} S ₈ | C ₄₂ H ₅₄ N ₆ Re ₄ Mo ₂ S ₈ | C ₄₂ H ₅₄ N ₆ Re ₃ Mo ₃ S ₈ |
| Molecular weight, g/mol | 1772.89 | 1836.07 | 1745.81 |
| Space group | <i>I</i> 4 ₁ / <i>a</i> | <i>I</i> 4 ₁ / <i>a</i> | <i>I</i> 4 ₁ / <i>a</i> |
| <i>a</i> , Å | 14.2868(2) | 14.2615(4) | 14.2677(11) |
| <i>b</i> , Å | 14.2868(2) | 14.2615(4) | 14.2677(11) |
| <i>c</i> , Å | 24.4236(8) | 24.4091(10) | 24.4726(4) |
| <i>α</i> , ° | 90 | 90 | 90 |
| <i>β</i> , ° | 90 | 90 | 90 |
| <i>γ</i> , ° | 90 | 90 | 90 |
| <i>V</i> , Å ³ | 4985.2(2) | 4964.6(3) | 4981.8(10) |
| <i>Z</i> | 4 | 4 | 4 |
| $\rho_{\text{calcd.}}$, g·cm ⁻³ | 2.362 | 2.456 | 2.328 |
| μ , mm ⁻¹ | 9.012 | 10.578 | 8.366 |
| θ range, ° | 2.62 – 28.68 | 2.62 – 31.42 | 2.62 – 29.50 |
| Miller indices range | -8 ≤ <i>h</i> ≤ 19 -18 ≤ <i>k</i> ≤ 7 -30 ≤ <i>l</i> ≤ 20 | -20 ≤ <i>h</i> ≤ 20 -20 ≤ <i>k</i> ≤ 20 -35 ≤ <i>l</i> ≤ 35 | -19 ≤ <i>h</i> ≤ 19 -19 ≤ <i>k</i> ≤ 18 -22 ≤ <i>l</i> ≤ 33 |
| Reflections collected/unique | 6112/3043 | 120331/4144 | 16324/3483 |
| Reflections (<i>I</i> > 2σ(<i>I</i>)) | 2304 | 2933 | 2378 |
| <i>R</i> _{int} | 0.017 | 0.0757 | 0.0445 |
| GoF | 1.060 | 1.055 | 1.033 |
| <i>R</i> ₁ / <i>wR</i> ₂ (<i>I</i> > 2σ(<i>I</i>)) | <i>R</i> ₁ = 0.0189 <i>wR</i> ₂ = 0.0363 | <i>R</i> ₁ = 0.0218 <i>wR</i> ₂ = 0.0400 | <i>R</i> ₁ = 0.0245 <i>wR</i> ₂ = 0.0456 |
| <i>R</i> ₁ / <i>wR</i> ₁ (all data) | <i>R</i> ₁ = 0.0301 <i>wR</i> ₂ = 0.0390 | <i>R</i> ₁ = 0.0414 <i>wR</i> ₂ = 0.0445 | <i>R</i> ₁ = 0.0417 <i>wR</i> ₂ = 0.0516 |
| $\Delta\rho_{\text{max}}/\Delta\rho_{\text{min}}$ (e·Å ³) | 0.735/-0.635 | 0.968/-1.603 | 0.529/-0.522 |

Table S3. Crystallographic data and details of the structure refinements for the compounds 7–9.

| | 7 | 8 | 9 |
|-----------------------------------------------------------------------|----------------------------------------------|----------------------------------------------|----------------------------------------------|
| Chemical formula | $C_{54}H_{78}N_6Re_{3.30}Mo_{2.70}S_8$ | $C_{59.188}H_{85.782}N_{8.594}Re_4Mo_2S_8$ | $C_{59.168}H_{85.752}N_{8.584}Re_3Mo_3S_8$ |
| Molecular weight, g/mol | 1941.20 | 2110.91 | 2020.18 |
| Space group | $P4_2/n$ | $P\bar{1}$ | $P\bar{1}$ |
| a , Å | 26.7541(6) | 14.3716(4) | 14.3459(11) |
| b , Å | 26.7541(6) | 15.5898(4) | 15.5318(13) |
| c , Å | 9.3946(3) | 18.3061(5) | 18.3357(14) |
| α , ° | 90 | 105.3240(10) | 105.271(3) |
| β , ° | 90 | 104.1010(10) | 103.906(3) |
| γ , ° | 90 | 100.1840(10) | 100.383(3) |
| V , Å ³ | 6724.5(4) | 3706.73(17) | 3694.1(5) |
| Z | 4 | 2 | 2 |
| $\rho_{\text{calcd.}}$, g·cm ⁻³ | 1.917 | 1.891 | 1.816 |
| μ , mm ⁻¹ | 6.690 | 7.098 | 5.655 |
| θ range, ° | 2.42 – 25.64 | 2.42 – 30.58 | 2.32 – 29.56 |
| Miller indices range | -32 ≤ h ≤ 32 -32 ≤ k ≤ 31 -11 ≤ l ≤ 11 | -19 ≤ h ≤ 17 -20 ≤ k ≤ 21 -24 ≤ l ≤ 24 | -19 ≤ h ≤ 19 -19 ≤ k ≤ 21 -25 ≤ l ≤ 25 |
| Reflections collected/unique | 54716/6401 | 64715/18377 | 61226/20697 |
| Reflections ($I > 2\sigma(I)$) | 5616 | 16088 | 14973 |
| R_{int} | 0.061 | 0.037 | 0.054 |
| GoF | 1.161 | 1.027 | 1.045 |
| $R_1 / wR_2(I > 2\sigma(I))$ | $R_1 = 0.0369$ $wR_2 = 0.0660$ | $R_1 = 0.0272$ $wR_2 = 0.0580$ | $R_1 = 0.0408$ $wR_2 = 0.0825$ |
| R_1 / wR_1 (all data) | $R_1 = 0.0473$ $wR_2 = 0.0691$ | $R_1 = 0.0339$ $wR_2 = 0.0272$ | $R_1 = 0.0650$ $wR_2 = 0.0939$ |
| $\Delta\rho_{\text{max}}/\Delta\rho_{\text{min}}$ (e·Å ³) | 1.195/-1.120 | 1.553/-1.415 | 1.144/-0.971 |

Table S4. Occupancy of metal atom crystallographic positions in the structures 1–9.

| | $[\{\text{Re}_{3.35}\text{Mo}_{2.65}\text{S}_8\}(\text{py})_6]$ (1) | $[\{\text{Re}_4\text{Mo}_2\text{S}_8\}(\text{py})_6]$ (2) | $[\{\text{Re}_3\text{Mo}_3\text{S}_8\}(\text{py})_6]$ (3) |
|----------------------------------|--------------------------------------------------------------------------|--------------------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------------------|
| Re ¹ /Mo ¹ | 0.61/0.39 | 0.66/0.34 | 0.52/0.48 |
| Re ² /Mo ² | 0.46/0.54 | 0.63/0.37 | 0.50/0.50 |
| Re ³ /Mo ³ | 0.59/0.41 | 0.54/0.46 | 0.43/0.57 |
| | | | |
| | $[\{\text{Re}_{3.30}\text{Mo}_{2.70}\text{S}_8\}(\text{etpy})_6]$ (4) | $[\{\text{Re}_4\text{Mo}_2\text{S}_8\}(\text{etpy})_6]$ (5) | $[\{\text{Re}_3\text{Mo}_3\text{S}_8\}(\text{etpy})_6]$ (6) |
| Re ¹ /Mo ¹ | 0.52/0.48 | 0.70/0.30 | 0.47/0.53 |
| Re ² /Mo ² | 0.59/0.41 | 0.76/0.24 | 0.49/0.51 |
| | | | |
| | $[\{\text{Re}_{3.30}\text{Mo}_{2.70}\text{S}_8\}(\text{tbp})_6]$ (7) | $[\{\text{Re}_4\text{Mo}_2\text{S}_8\}(\text{tbp})_6]$ $\cdot(\text{CH}_3\text{CN})_2$ (8) | $[\{\text{Re}_3\text{Mo}_3\text{S}_8\}(\text{tbp})_6]$ $\cdot(\text{CH}_3\text{CN})_2$ (9) |
| Re ¹ /Mo ¹ | 0.45/0.55 | 0.68/0.32 | 0.52/0.48 |
| Re ² /Mo ² | 0.60/0.40 | 0.66/0.34 | 0.51/0.49 |
| Re ³ /Mo ³ | 0.60/0.40 | 0.65/0.35 | 0.49/0.51 |
| Re ⁴ /Mo ⁴ | – | 0.66/0.34 | 0.51/0.49 |
| Re ⁵ /Mo ⁵ | – | 0.57/0.43 | 0.53/0.47 |
| Re ⁶ /Mo ⁶ | – | 0.60/0.40 | 0.48/0.52 |

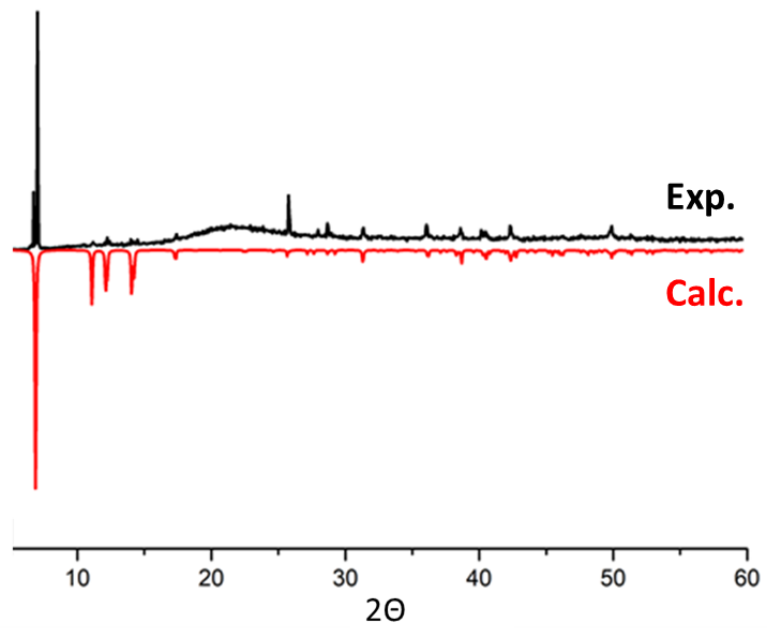


Figure S4. Powder XRD patterns for compound 4.

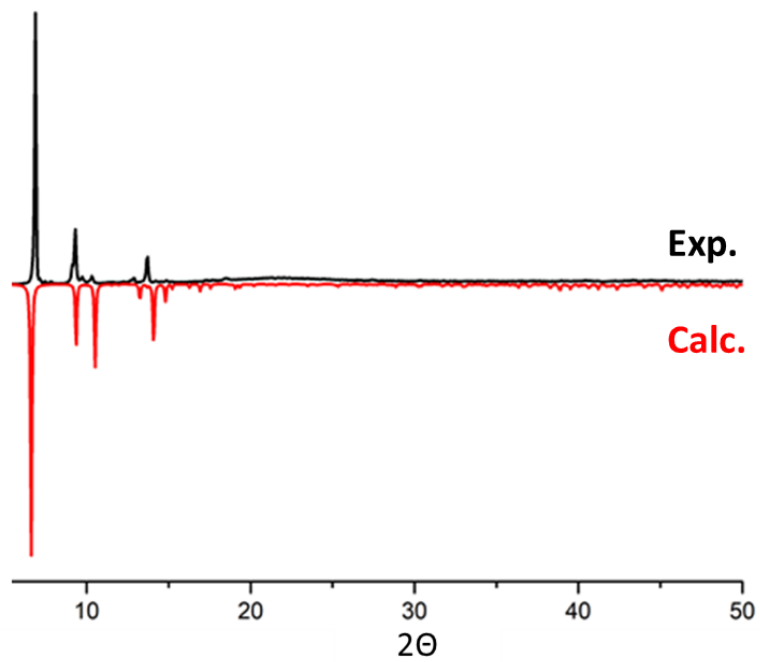


Figure S5. Powder XRD patterns for compound 7.

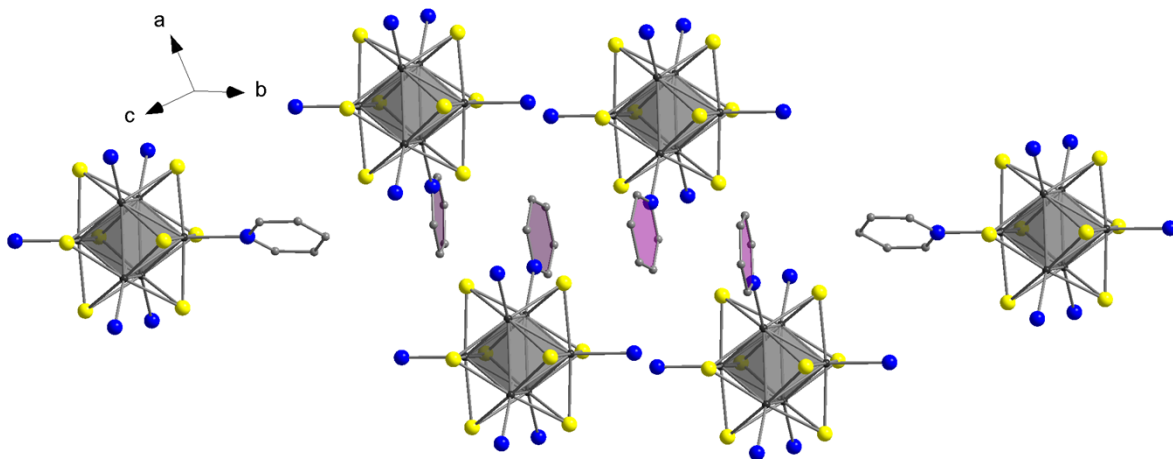


Figure S6. Formation of the isolated tetrameric fragments through the π - π interactions in the structures of compounds **1–3**.

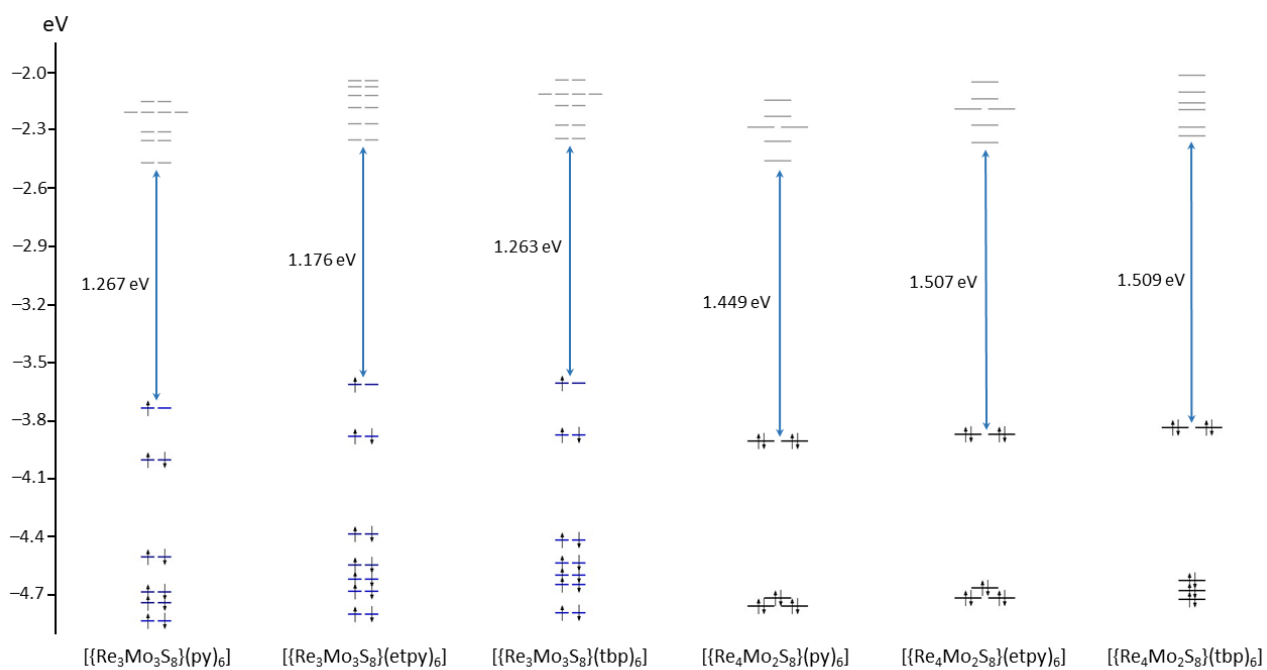


Figure S7. Calculated energy level diagrams and HOMO-LUMO gap values for *mer*- $[[\text{Re}_3\text{Mo}_3\text{S}_8]\text{L}_6]$ and *trans*- $[[\text{Re}_4\text{Mo}_2\text{S}_8]\text{L}_6]$ clusters (L = py, etpy, tbp)