

<Electronic Supplementary Information>

**Construction and catalytic effects of solvent- and metal(II)-dependent crystal structures:
transformation of process of 0D into 3D structures**

Gyeongmin Kim, Jihun Han, Dongwon Kim, and Ok-Sang Jung*

Department of Chemistry, Pusan National University, Busan 46241, Republic of Korea Fax: (+82) 51-5163522;

Tel: (+82) 51-5103240; E-mail: oksjung@pusan.ac.kr

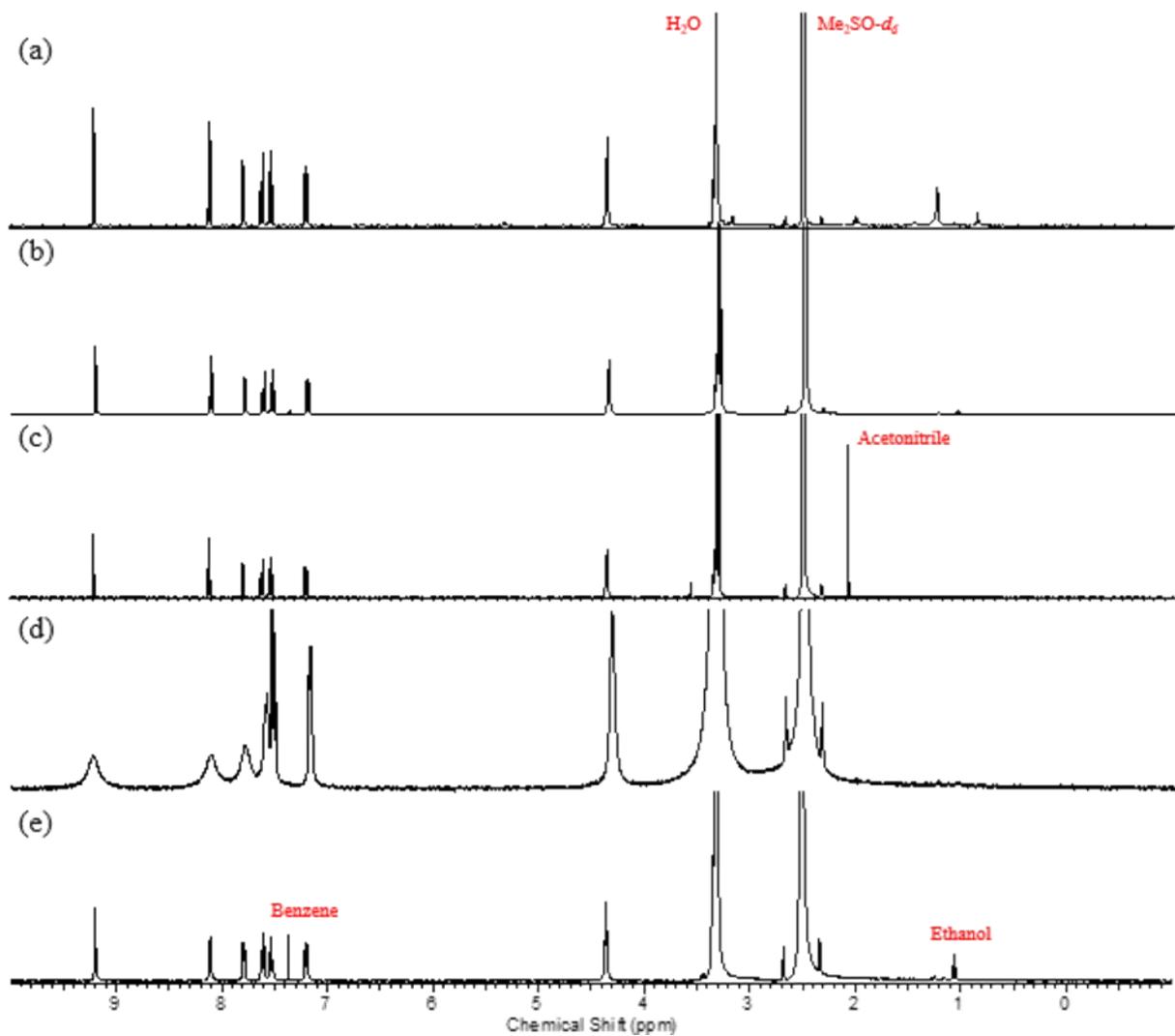


Fig. S1 ^1H NMR spectra of L (a), $[\text{Zn}(\text{NO}_3)\text{L}(\text{H}_2\text{O})](\text{NO}_3)\cdot\text{H}_2\text{O}$ (b), $[\text{NO}_3@\text{Zn}_3(\text{NO}_3)_5\text{L}_2]\cdot 2\text{CH}_3\text{CN}$ (c), $[\text{Co}(\text{NO}_3)\text{L}(\text{H}_2\text{O})](\text{NO}_3)\cdot\text{H}_2\text{O}$ (d), and $[\text{NO}_3@\text{Ni}_3(\text{NO}_3)_3\text{L}_2(\text{H}_2\text{O})_6]2\text{NO}_3\cdot\text{C}_2\text{H}_5\text{OH}\cdot\text{C}_6\text{H}_6$ (e) in $\text{Me}_2\text{SO}-d_6$.

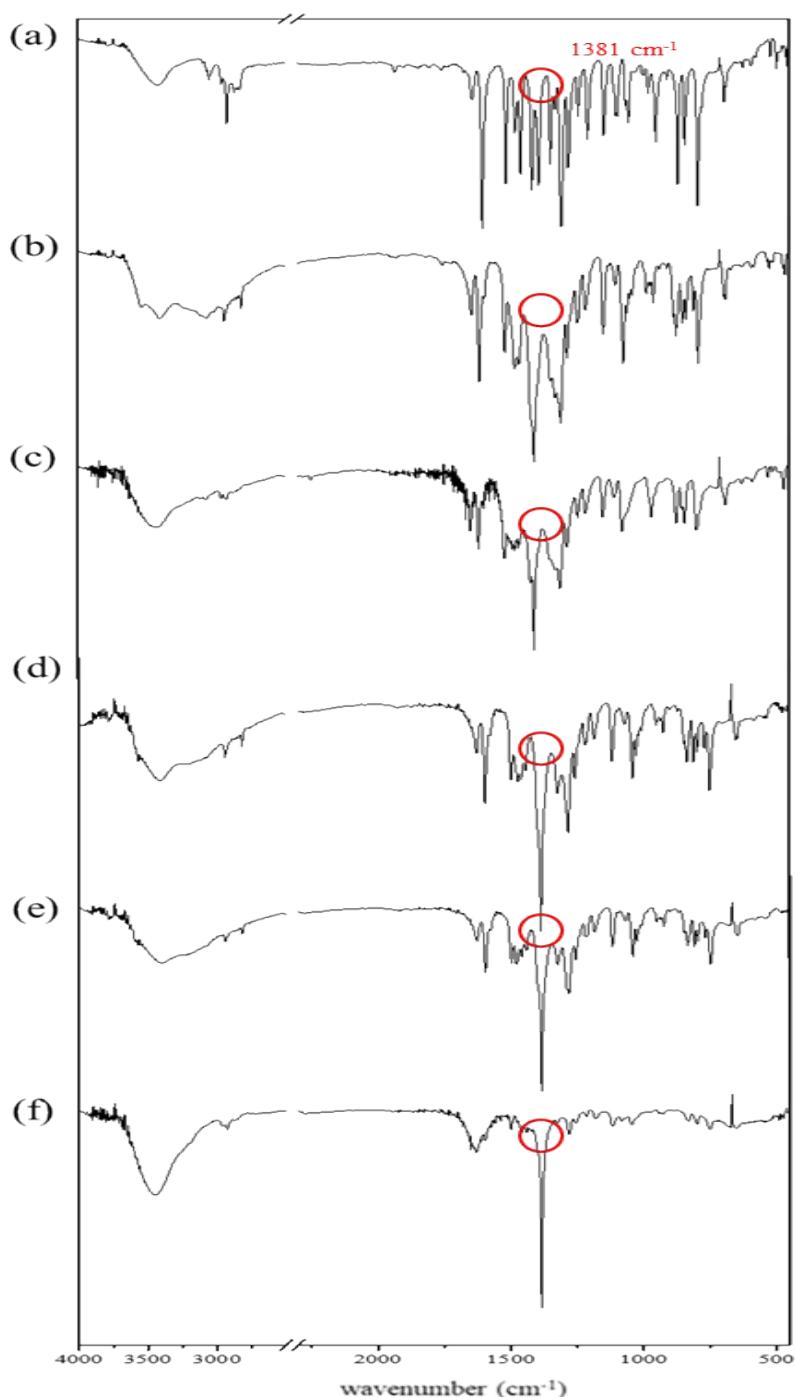


Fig. S2 IR spectra of L (a), $[\text{Zn}(\text{NO}_3)\text{L}(\text{H}_2\text{O})](\text{NO}_3)\cdot\text{H}_2\text{O}$ (b), $[\text{NO}_3@\text{Zn}_3(\text{NO}_3)_5\text{L}_2]\cdot 2\text{CH}_3\text{CN}$ (c), $[\text{Co}(\text{NO}_3)\text{L}(\text{H}_2\text{O})](\text{NO}_3)\cdot\text{H}_2\text{O}$ (d), $[\text{NO}_3@\text{Ni}_3(\text{NO}_3)_3\text{L}_2(\text{H}_2\text{O})_6]2\text{NO}_3\cdot\text{C}_2\text{H}_5\text{OH}\cdot\text{C}_6\text{H}_6$ (e), and $[\text{NO}_3@\text{Cu}_3(\text{NO}_3)_5\text{L}_2(\text{C}_2\text{H}_5\text{OH})]\cdot\text{C}_2\text{H}_5\text{OH}\cdot 2\text{C}_6\text{H}_6$ (f).

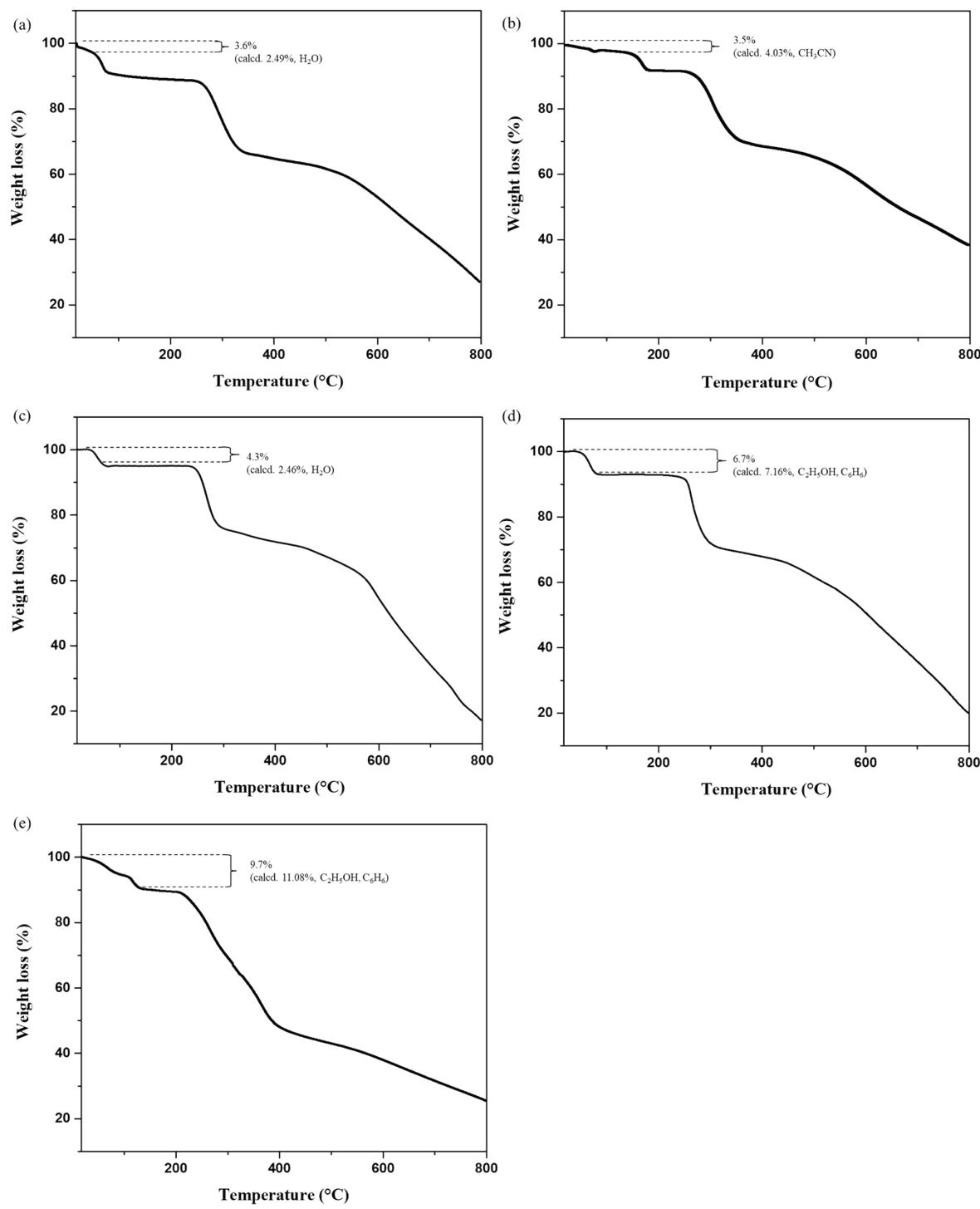


Fig. S3 TGA curves for $[\text{Zn}(\text{NO}_3)\text{L}(\text{H}_2\text{O})](\text{NO}_3)\cdot\text{H}_2\text{O}$ (a), $[\text{NO}_3@\text{Zn}_3(\text{NO}_3)_5\text{L}_2]\cdot 2\text{CH}_3\text{CN}$ (b), $[\text{Co}(\text{NO}_3)\text{L}(\text{H}_2\text{O})](\text{NO}_3)\cdot\text{H}_2\text{O}$ (c), $[\text{NO}_3@\text{Ni}_3(\text{NO}_3)_3\text{L}_2(\text{H}_2\text{O})_6]2\text{NO}_3\cdot\text{C}_2\text{H}_5\text{OH}\cdot\text{C}_6\text{H}_6$ (d), and $[\text{NO}_3@\text{Cu}_3(\text{NO}_3)_5\text{L}_2(\text{C}_2\text{H}_5\text{OH})]\cdot\text{C}_2\text{H}_5\text{OH}\cdot 2\text{C}_6\text{H}_6$ (e).

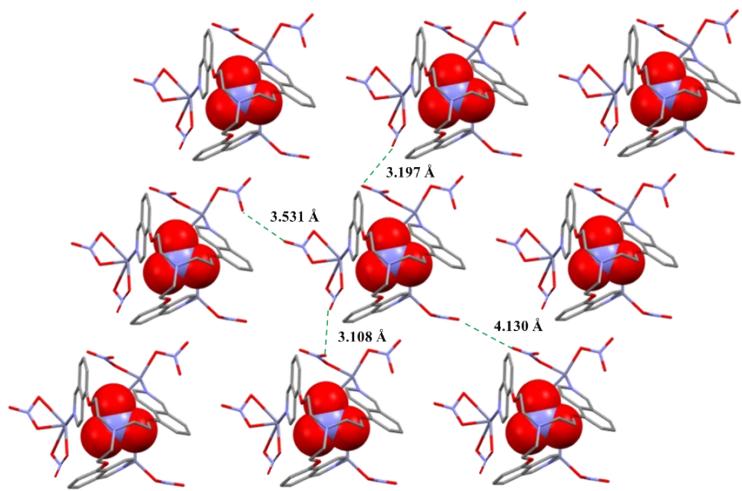


Fig. S4 Packing mode of $[\text{NO}_3@\text{Zn}_3(\text{NO}_3)_5\text{L}_2]\cdot 2\text{CH}_3\text{CN}$.

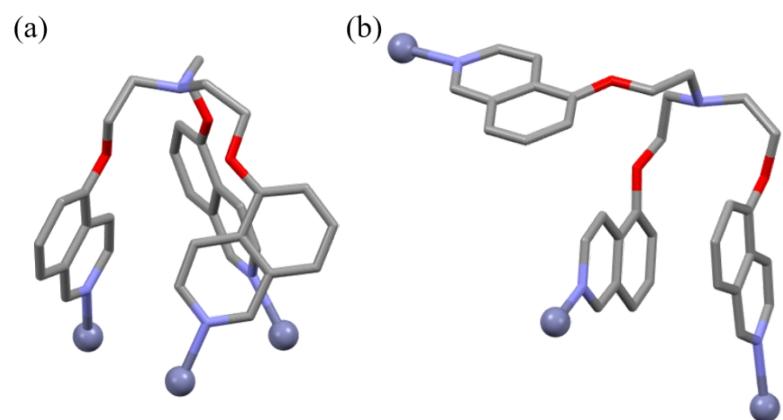


Fig. S5 L conformation of the cage and the 3D networks.

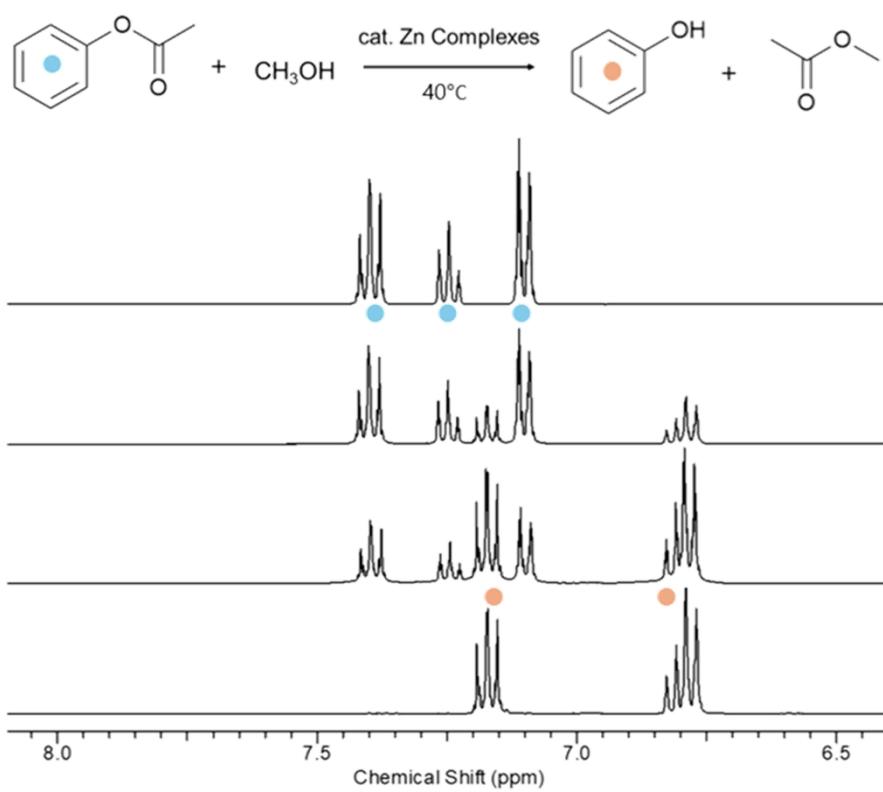


Fig. S6 ^1H -NMR spectra of $[\text{Zn}(\text{NO}_3)_2 \text{L}(\text{H}_2\text{O})](\text{NO}_3)_2 \cdot \text{H}_2\text{O}$ transesterification catalysis at 40°C in MeOD.

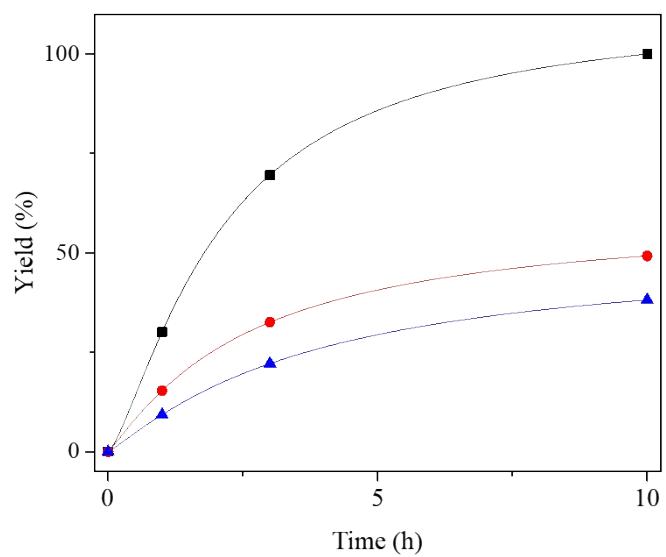


Fig. S7 Plot showing transesterification catalysis of phenyl acetate with dried methanol using $[Zn(NO_3)L(H_2O)](NO_3)\cdot H_2O$ (black line), mixture of $Zn(NO_3)_2 : L$ (1 : 1) (red line), and $[NO_3@Zn_3(NO_3)_5L_2]\cdot 2CH_3CN$ (blue) at 40°C.

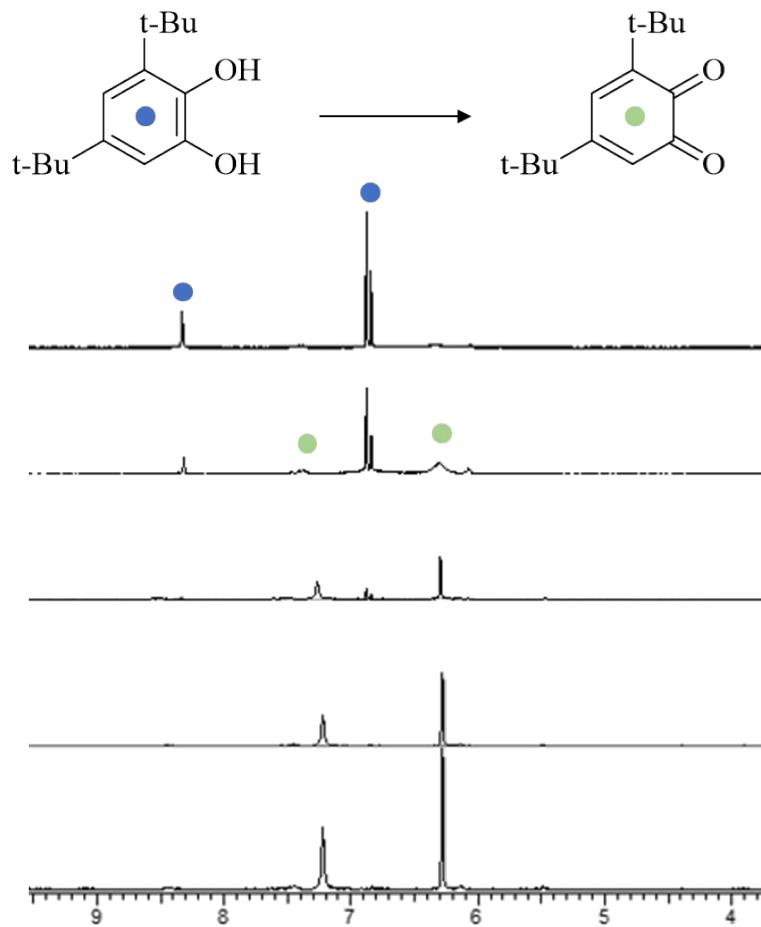


Fig. S8 ¹H NMR spectral change (in Me₂CO-*d*₆) on catalytic oxidation of 3,5-di-*tert*-butylcatechol to 3,5-di-*tert*-butylorthoquinone using [NO₃@Cu₃(NO₃)₅L₂(C₂H₅OH)]·C₂H₅OH·2C₆H₆ in acetone.

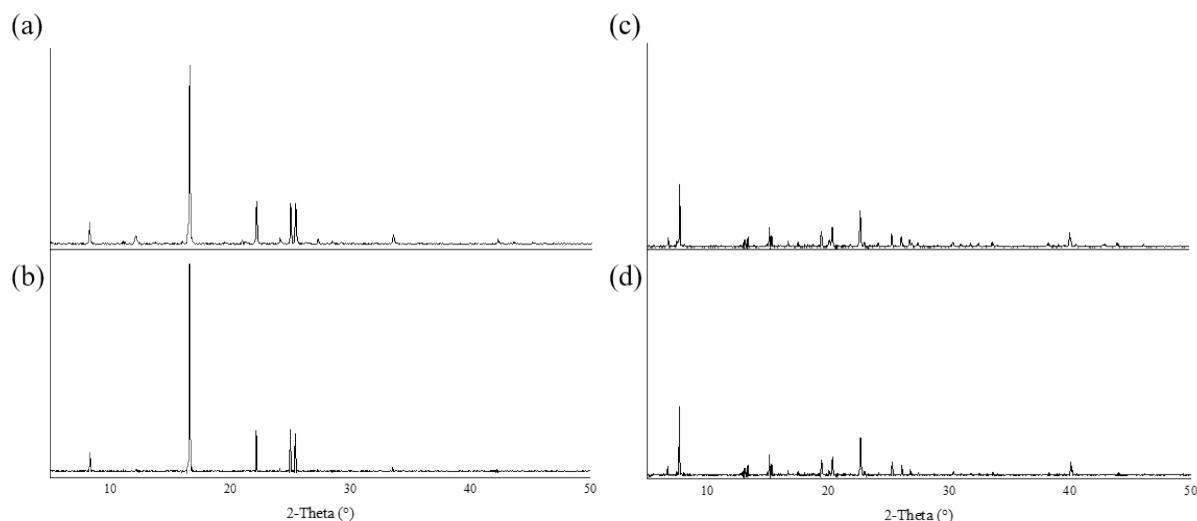


Fig. S9 Powder X-ray diffraction patterns of catalysis -before (a) and -after (b) of $[\text{Zn}(\text{NO}_3)_2\text{L}(\text{H}_2\text{O})](\text{NO}_3)_2\cdot\text{H}_2\text{O}$ and catalysis -before (c) and -after (d) of $[\text{NO}_3@\text{Cu}_3(\text{NO}_3)_5\text{L}_2(\text{C}_2\text{H}_5\text{OH})]\cdot\text{C}_2\text{H}_5\text{OH}\cdot 2\text{C}_6\text{H}_6$.

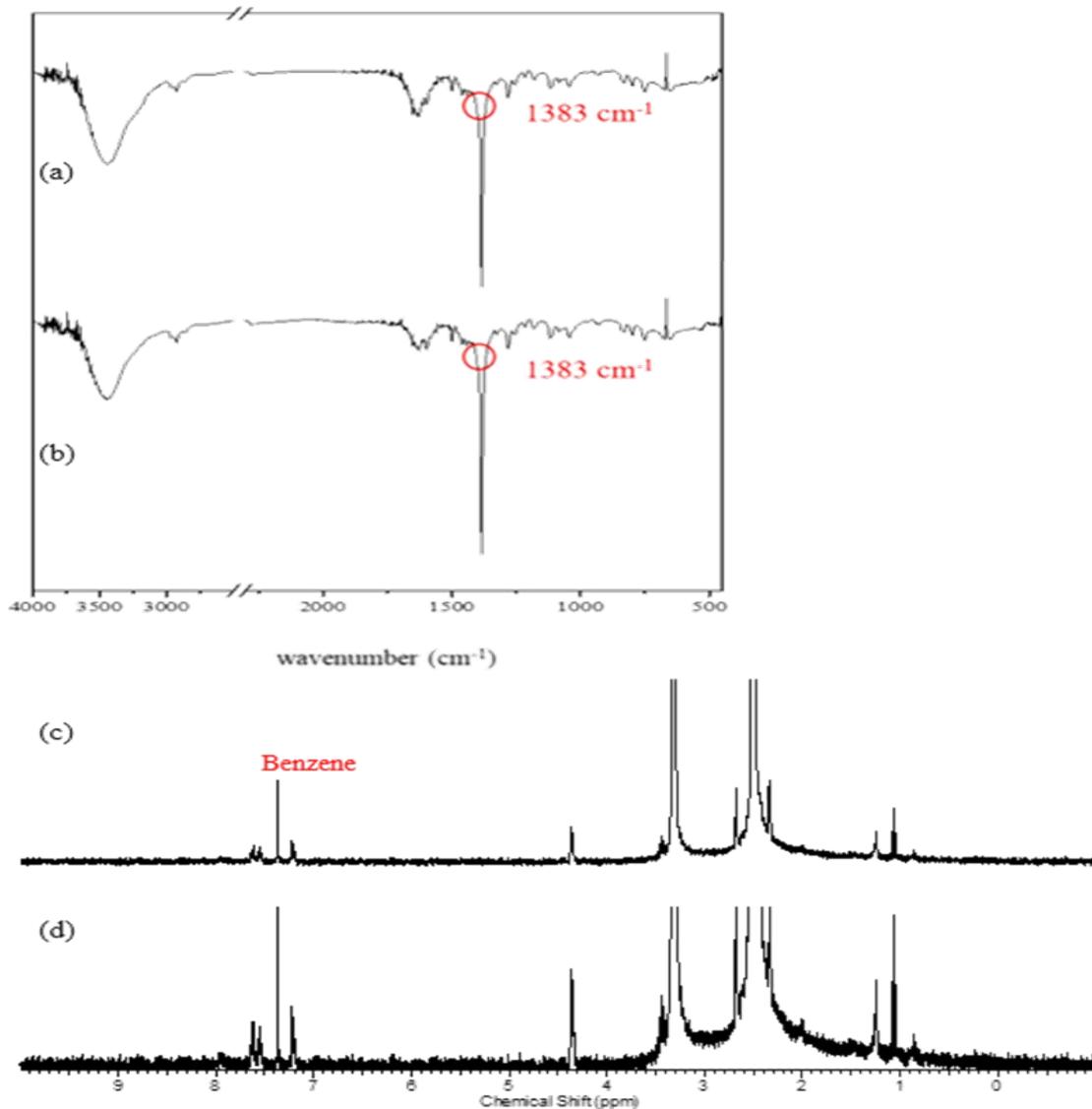


Fig. S10 IR spectra of original $[\text{NO}_3@\text{Cu}_3(\text{NO}_3)_5\text{L}_2(\text{C}_2\text{H}_5\text{OH})]\cdot\text{C}_2\text{H}_5\text{OH}\cdot 2\text{C}_6\text{H}_6$ (a) and the recycled crystals after catechol oxidation (b). ^1H NMR spectra of original $[\text{NO}_3@\text{Cu}_3(\text{NO}_3)_5\text{L}_2(\text{C}_2\text{H}_5\text{OH})]\cdot\text{C}_2\text{H}_5\text{OH}\cdot 2\text{C}_6\text{H}_6$ (c) and the recycled crystals after catechol oxidation (d).

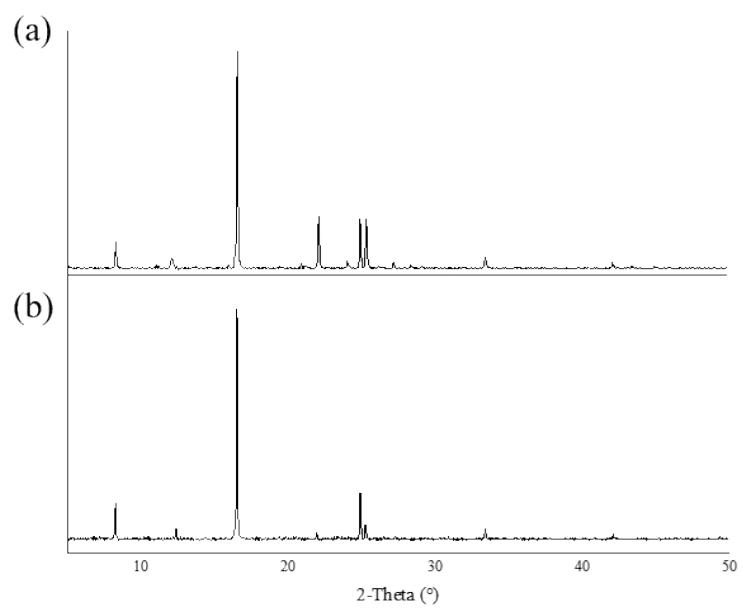


Fig. S11 Powder X-ray diffraction patterns of 3D networks $[\text{Zn}(\text{NO}_3)\text{L}(\text{H}_2\text{O})](\text{NO}_3)\cdot\text{H}_2\text{O}$ (a) and $[\text{Co}(\text{NO}_3)\text{L}(\text{H}_2\text{O})](\text{NO}_3)\cdot\text{H}_2\text{O}$ (b).

Table S1. Crystallographic data

	[Zn(NO ₃)L(H ₂ O)](NO ₃)·H ₂ O	[NO ₃ @Zn ₃ (NO ₃) ₅ L ₂]·2CH ₃ CN	[Co(NO ₃)L(H ₂ O)](NO ₃)·H ₂ O	[NO ₃ @Ni ₃ (NO ₃) ₃ L ₂ (H ₂ O) ₆]2NO ₃ ·C ₂ H ₅ OH·C ₆ H ₆ ·H ₅ OH)	[NO ₃ @Cu ₃ (NO ₃) ₅ L ₂ (C ₂ H ₅ OH) ₂]C ₂ H ₅ OH·2C ₆ H ₆
Formula	C ₃₃ H ₃₄ ZnN ₆ O ₁₁ Zn	C ₇₀ H ₆₆ N ₁₆ O ₂₄ Zn ₃	C ₃₃ H ₃₄ CoN ₆ O ₁₁	C ₇₄ H ₈₄ N ₁₄ Ni ₃ O ₃₁	C ₈₂ H ₈₄ Cu ₃ N ₁₄ O ₂₆
M _w	756.03	1711.49	749.59	1841.68	1872.25
Cryst. sys.	Monoclinic	Monoclinic	Monoclinic	Trigonal	Monoclinic
Space group	<i>Cc</i>	<i>P2₁/m</i>	<i>Cc</i>	<i>P-3</i>	<i>P2₁/m</i>
<i>a</i> (Å)	17.513(5))	11.732(2)	17.551(4)	14.608(2)	21.195(2)
<i>b</i> (Å)	14.247(3)	23.592(5)	14.217(3)	14.608(2)	18.926(2)
<i>c</i> (Å)	14.249(3)	14.243(3)	14.235(3)	24.244(5)	22.497(2)
α (°)	-	-	-	-	-
β (°)	112.70(3)	111.45(3)	112.49(3)	-	105.855(2)
γ (°)	-	-	-	120	-
<i>V</i> (Å ³)	3280(1)	3669(2)	3282(1)	4480(2)	8681(1)
<i>Z</i>	4	2	4	2	4
ρ (g cm ⁻³)	1.531	1.549	1.517	1.365	1.433
μ (mm ⁻¹)	0.595	1.020	0.430	0.679	0.813
<i>R</i> _{int}	0.0223	0.0696	0.1035	0.1116	0.1769
GoF on <i>F</i> ²	1.077	1.072	0.851	1.389	1.075
<i>R</i> ₁ [<i>I</i> >2σ(<i>I</i>) ^a]	0.0332	0.0766	0.0631	0.1458	0.1061
<i>wR</i> ₂ (all data) ^b	0.0853	0.2679	0.1256	0.4034	0.3305

^a*R*₁ = Σ||*F*_o|| - ||*F*_c||/Σ|*F*_o|, ^b*wR*₂ = (Σ[w(*F*_o² - *F*_c²)²]/Σ[w(*F*_o²)²])^{1/2}

Table S2. Selected bond lengths (\AA) and angles ($^\circ$)

[Zn(NO ₃)L(H ₂ O)](NO ₃)·H ₂ O	[NO ₃ @Zn ₃ (NO ₃) ₅ L ₂]·2CH ₃ CN	[Co(NO ₃)L(H ₂ O)](NO ₃)·H ₂ O	[NO ₃ @Ni ₃ (NO ₃) ₅ L ₂ (H ₂ O) ₆]2NO ₃ ·C ₂ H ₅ OH·C ₆ H ₆	[NO ₃ @Cu ₃ (NO ₃) ₅ L ₂ (C ₂ H ₅ OH)]·C ₂ H ₅ OH·2C ₆ H ₆					
Zn(1)-O(1C)	2.031(3)	Zn(1)-O(9')	1.89(3)	N(1)-Co(2) ^{#1}	2.157(4)	Ni(1)-O(3)	2.036(8)	Cu(1)-O(1)	1.983(16)
Zn(1)-N(13)	2.091(3)	Zn(1)-N(3A) ^{#1}	2.044(3)	N(2)-Co(2)	2.103(3)	Ni(1)-O(5)	2.055(10)	Cu(1)-N(2B)	1.992(7)
Zn(1)-N(39) ^{#1}	2.168(3)	Zn(1)-N(3A)	2.044(3)	Co(2)-O(3)	2.056(3)	Ni(1)-O(4)	2.073(10)	Cu(1)-N(2A)	2.002(7)
Zn(1)-N(26) ^{#2}	2.179(3)	Zn(2)-N(4A) ^{#1}	2.001(4)	Co(2)-N(4) ^{#2}	2.175(3)	Ni(1)-N(3)	2.114(10)	Cu(1)-O(4)	2.061(10)
Zn(1)-O(2A)	2.206(3)	Zn(2)-N(4A)	2.001(4)	Co(2)-O(7)	2.191(3)	Ni(1)-N(1)	2.131(4)	Cu(1)-O(3C)	2.256(9)
Zn(1)-O(3A)	2.339(3)	Zn(2)-O(3)	2.063(4)	Co(2)-O(6)	2.197(3)	Ni(1)-O(7)	2.158(11)	Cu(1)-O(3)	2.40(3)
		Zn(2)-O(10')	2.18(4)						
O(1C)-Zn(1)-N(13)	104.17(14)	Zn(2)-O(10') ^{#1}	2.18(4)	C(2)-N(1)-Co(2) ^{#1}	122.6(3)	O(3)-Ni(1)-O(5)	93.7(4)	O(1)-Cu(1)-N(2B)	89.3(4)
O(1C)-Zn(1)-N(39) ^{#1}	94.01(10)	Zn(2)-N(4') ^{#1}	2.45(4)	C(3)-N(1)-Co(2) ^{#1}	121.0(3)	O(3)-Ni(1)-O(4)	178.0(4)	O(1)-Cu(1)-N(2A)	87.8(4)
N(13)-Zn(1)-N(39) ^{#1}	93.47(12)			O(3)-Co(2)-N(2)	98.78(16)	O(5)-Ni(1)-O(4)	87.7(5)	N(2B)-Cu(1)-N(2A)	175.4(3)
O(1C)-Zn(1)-N(26) ^{#2}	93.03(12)	N(3)-O(7)-Zn(1)	116.1(5)	O(3)-Co(2)-N(1) ^{#3}	93.88(12)	O(3)-Ni(1)-N(3)	91.0(3)	O(1)-Cu(1)-O(4)	176.2(4)
N(13)-Zn(1)-N(26) ^{#2}	88.43(11)	O(9')-Zn(1)-N(3A)	109.7(3)	N(2)-Co(2)-N(1) ^{#3}	93.06(14)	O(5)-Ni(1)-N(3)	88.5(4)	N(2B)-Cu(1)-O(4)	93.5(3)
N(39) ^{#1} -Zn(1)-N(26) ^{#2}	172.02(12)	O(16)-Zn(1)-N(3A)	124.9(2)	O(3)-Co(2)-N(4) ^{#2}	91.70(14)	O(4)-Ni(1)-N(3)	90.4(4)	N(2A)-Cu(1)-O(4)	89.6(3)
O(1C)-Zn(1)-O(2A)	101.85(13)	N(3A) ^{#1} -Zn(1)-N(3A)	122.76(17)	N(2)-Co(2)-N(4) ^{#2}	87.85(13)	O(3)-Ni(1)-N(1)	87.3(3)	O(1)-Cu(1)-O(3C)	93.0(4)
N(13)-Zn(1)-O(2A)	153.73(11)	N(3A) ^{#1} -Zn(1)-O(7)	105.9(2)	N(1) ^{#3} -Co(2)-N(4) ^{#2}	174.14(14)	O(5)-Ni(1)-N(1)	89.5(3)	N(2B)-Cu(1)-O(3C)	90.0(3)
N(39) ^{#1} -Zn(1)-O(2A)	88.25(12)	N(3A)-Zn(1)-O(7)	85.8(2)	O(3)-Co(2)-O(7)	160.72(14)	O(4)-Ni(1)-N(1)	91.3(3)	N(2A)-Cu(1)-O(3C)	93.7(3)
N(26) ^{#2} -Zn(1)-O(2A)	86.64(11)	O(11)-Zn(2)-N(4A) ^{#1}	129.6(3)	N(2)-Co(2)-O(7)	100.44(13)	N(3)-Ni(1)-N(1)	177.3(3)	O(4)-Cu(1)-O(3C)	84.5(4)
O(1C)-Zn(1)-O(3A)	158.33(12)	O(11)-Zn(2)-N(4A)	85.1(2)	N(1) ^{#3} -Co(2)-O(7)	86.55(12)	O(3)-Ni(1)-O(7)	86.0(5)	O(1)-Cu(1)-O(3)	55.6(4)
N(13)-Zn(1)-O(3A)	97.50(11)	N(4A) ^{#1} -Zn(2)-N(4A)	128.50(19)	N(4) ^{#2} -Co(2)-O(7)	87.59(12)	O(5)-Ni(1)-O(7)	179.7(5)	N(2B)-Cu(1)-O(3)	85.7(5)
N(39) ^{#1} -Zn(1)-O(3A)	84.37(11)	O(11)-Zn(2)-O(3)	107.5(3)	O(3)-Co(2)-O(6)	102.07(15)	O(4)-Ni(1)-O(7)	92.6(6)	N(2A)-Cu(1)-O(3)	89.8(5)
N(26) ^{#2} -Zn(1)-O(3A)	87.71(10)	N(4A) ^{#1} -Zn(2)-O(3)	101.38(10)	N(2)-Co(2)-O(6)	158.98(13)	N(3)-Ni(1)-O(7)	91.5(4)	O(4)-Cu(1)-O(3)	127.1(5)
O(2A)-Zn(1)-O(3A)	56.55(10)	N(4A)-Zn(2)-O(3)	101.38(10)	N(1) ^{#3} -Co(2)-O(6)	88.18(14)	N(1)-Ni(1)-O(7)	90.6(4)	O(3C)-Cu(1)-O(3)	148.2(4)
		N(4A) ^{#1} -Zn(2)-O(10')	124.8(11)	N(4) ^{#2} -Co(2)-O(6)	88.90(13)	N(5)-O(3)-Ni(1)	129.0(6)		
		N(4A)-Zn(2)-O(10')	71.2(8)	O(7)-Co(2)-O(6)	58.66(12)	N(6)-O(7)-Ni(1)	163.9(10)		
		O(3)-Zn(2)-O(10')	126.9(16)			C(2)-N(1)-Ni(1)	119.0(3)		
		N(4A) ^{#1} -Zn(2)-O(10') ^{#1}	71.2(8)			C(1)-N(1)-Ni(1)	120.4(3)		
		N(4A)-Zn(2)-O(10') ^{#1}	124.8(11)						
		O(3)-Zn(2)-O(10') ^{#1}	126.9(16)						
		N(4A) ^{#1} -Zn(2)-N(4') ^{#1}	101.4(5)						
		N(4A)-Zn(2)-N(4') ^{#1}	101.4(5)						
		O(3)-Zn(2)-N(4') ^{#1}	126.0(12)						
		O(10')-Zn(2)-N(4') ^{#1}	30.7(6)						
		O(10') ^{#1} -Zn(2)-N(4') ^{#1}	30.7(6)						

^{#1} x-1/2, y-1/2, z-1^{#1} x, -y+1/2, z^{#1} x-1/2, y+1/2, z-1^{#2} x, -y+1, z+1/2^{#3} x+1/2, y-1/2, z+1