

# **Exploring the Luminescence Properties and Sensing Mechanism of a Turn-on TADF Probe for Sulfite**

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Table S1. The absorption properties for probe DCF-MPYM-lev and product DCF-MPYM calculated by several functionals.

Function	Molecule	States	Energy(eV)	Wavelength(nm)	Osciallor(a.u.)
B3lyp	DCF-MPYM-lev	$S_4$	3.11	399/370*	1.28
		$S_1$	2.22	558/525*	0.26
		$S_5$	2.87	433/480*	0.64
M062x	DCF-MPYM-lev	$S_2$	3.65	340	2.09
		$S_1$	2.80	443	0.73
		$S_4$	3.64	341	1.29
$\omega b97xd$	DCF-MPYM-lev	$S_2$	3.72	333	2.10
		$S_1$	2.89	429	0.80
		$S_4$	3.71	335	1.34
Pbe0	DCF-MPYM-lev	$S_2$	3.72	333	2.10
		$S_1$	2.89	429	0.80
		$S_5$	3.71	335	1.34

Table S2 Fluorescence properties, including the vertical excitation energies ( $E_{vt}/eV$ ), radiative rates ( $k_r/s^{-1}$ ), non-radiative decay rate ( $k_{nr}/s^{-1}$ ), total reorganization energies ( $\lambda/cm^{-1}$ ) and fluorescence efficiency ( $\Phi/\%$ ) of DCF-MPYM and  $[DCF - MPYM - 2H]^{2-}$  with different conformers are calculated.

	$E_{vt}$	$k_r$	$k_{nr}$	$\lambda$	$\Phi$
DCF-MPYM-A	1.54	$3.57 \times 10^7$	$2.23 \times 10^9$	3177	1.57%
DCF-MPYM-B	1.53	$3.87 \times 10^7$	$3.55 \times 10^9$	3190	1.08%
DCF-MPYM-C	1.54	$2.93 \times 10^7$	$2.03 \times 10^{11}$	4338	0.01%
A	1.94	$4.28 \times 10^7$	$3.44 \times 10^8$	936	11.04%
B	2.01	$8.31 \times 10^7$	$2.99 \times 10^8$	906	21.73%
C	1.90	$452 \times 10^7$	$3.56 \times 10^8$	1369	11.25%

Table S3 One-photon absorption properties including the excitation energy  $E_{opa}$  (eV), the corresponding wavelengths  $\lambda_{opa}$  (nm), the oscillator strengths  $\delta_{opa}$  (a.u.) and transition nature of  $DCF - MPYM - lev$  and  $DCF - MPYM$ .

Mol	Excited state	$E_{opa}$	$\lambda_{opa}$	$\delta_{opa}$	Transition nature	
DCF-MPYM -lev	S4	3.11	399	1.28	H-1-L+1	41.6%
					H-1-L	26.9%
DCF-MPYM	S4	2.71	457	0.30	H-2-L	76.7%
					H-L+2	65.6%

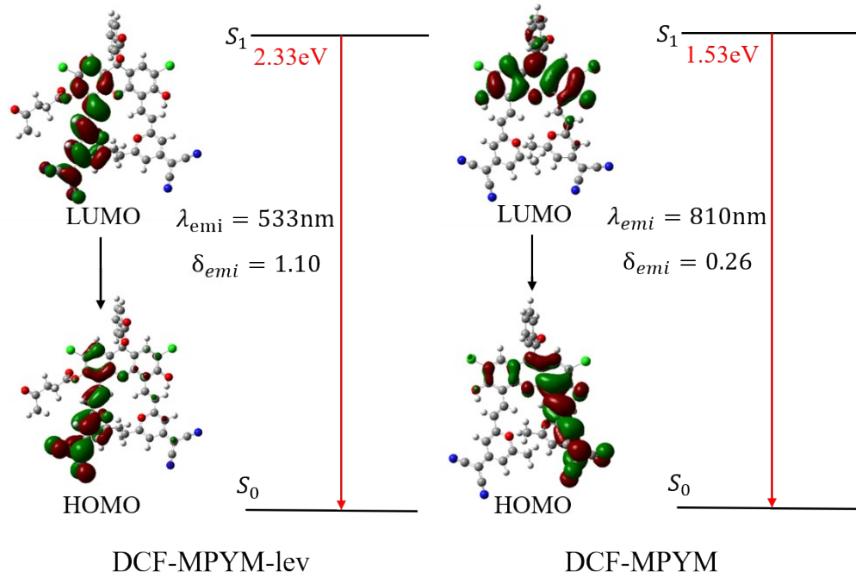


Figure S1. The emission properties of probe *DCF – MPYM – lev* and *DCF – MPYM*. The green and red sections represent the positive and negative phases of the wave function respectively.

Table S4. Fluorescence properties, including the vertical excitation energies ( $E_{vt}/eV$ ), radiative rates ( $k_r/s^{-1}$ ), non-radiative decay rate ( $k_{nr}/s^{-1}$ ), total reorganization energies ( $\lambda/cm^{-1}$ ) and fluorescence efficiency ( $\Phi/\%$ ) are calculated of *DCF – MPYM – lev* and *DCF – MPYM* with different conformers are calculated.

Mol	$E_{vt}$	$k_r$	$k_{nr}$	$\lambda$	$\Phi$
DCF-MPYM-lev-B	2.33	$3.22 \times 10^8$	$7.22 \times 10^7$	2851	81.7%
DCF-MPYM-B	1.53	$3.87 \times 10^7$	$3.55 \times 10^9$	3190	1.08%

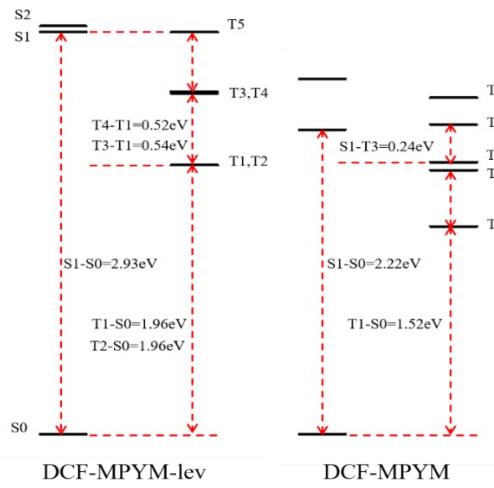


Figure S2. Excitation energies for *DCF – MPYM – lev* and *DCF – MPYM*.

Table S5. Frontier energies (eV) of H<sub>2</sub><sup>O</sup>OMO-3 to LUMO+3 for [DCF - MPYM - lev - H]<sup>-</sup> and [DCF - MPYM - 2H]<sup>-</sup> with different conformers in acetonitrile.

	HOMO-3	HOMO-2	HOMO-1	HOMO	LUMO	LUMO+1	LUMO+2	LUMO+3
lev-A	-6.14	-5.98	-5.73	-4.94	-2.67	-2.22	-1.54	-1.43
lev-B	-6.13	-5.99	-5.70	-4.97	-2.70	-2.19	-1.57	-1.44
lev-C	-6.11	-5.99	-5.70	-4.92	-2.67	-2.19	-1.52	-1.43
A	-5.82	-5.55	-5.32	-4.96	-2.30	-2.24	-1.95	-1.17
B	-5.82	-5.54	-5.32	-4.98	-2.28	-2.24	-1.96	-1.14
C	-5.81	-5.53	-5.27	-4.95	-2.33	-2.21	-1.94	-1.20

Figure S3. The orbitals distribution form HOMO-3 to LUMO+3 of probe  $DCF - MPYM - lev$  and  $DCF - MPYM$ .

