Supporting Information

Linear π -conjugated units of the D_{∞ h} point group as superior ultraviolet birefringent units

Rong Yang,^{ab} Abudukadi Tudi,^{*ab} Yu Dang,^{ab} Kewang Zhang,^{ab} and Hongsheng Shi^{*ab}

 a. Research Center for Crystal Materials, Key Laboratory of Functional Materials and Devices for Special Environments, Xinjiang Technical Institute of Physics & Chemistry, Chinese Academy of Sciences; Xinjiang Key Laboratory of Electronic Information Materials and Devices, 40–1 South Beijing Road, Urumqi 830011, China.

b. Center of Materials Science and Optoelectronics Engineering

 University of Chinese Academy of Sciences, Beijing 100049, China.

* Corresponding author: shihs@ms.xjb.ac.cn; abdkd@ms.xjb.ac.cn

Calculations detials

The electronic structures and optical properties were obtained on the basis of ab initio calculations implemented in the CASTEP package through density functional theory (DFT).¹ The generalized gradient approximation (GGA)² was adopted, and Perdew-Burke-Ernzerhof (PBE)³ functional was chosen to calculate the exchange-correlation potential, with an energy cutoff of 750 eV. The numerical integration of the Brillouin zone was performed using a $3 \times 3 \times 2$ Monkhorst-Pack *k*-point sampling. Norm-conserving pseudopotentials were used with the following valence electron configurations: Li-2s¹,Na-2s²2p⁶3s¹, Si-3s²3p², C-2s²2p², N-2s²2p³, B-2s²2p¹, O-2s²2p⁴ and F-2s²2p⁵. In addition, B3LYP (Becke, three-parameter, Lee-Yang-Parr) exchange-correlation functional with the Lee-Yang-Parr correlation functional at the 6-311G basis set in Gaussian was employed to calculate the groups.

	BO ₂	BO ₃
Polarizability anisotropy	9.2	8.3
HOMO-LUMO gap	14.4	7
Type of hybrid orbitals of B atom in ground state	sp	sp ²
Theoretical bond angle of O–B–O (°)	180	120

Table S1 Performances of (BO₂)- and (BO₃)³⁻anions



Figure S1 Calculated phonon dispersion curves of NaBO₂.



Figure S2 Calculated birefringence dispersion curves of NaBO₂



Figure S3 Crystal structure of (a) Si_2CN_4 , (b) BF_4NO_2 , and (c) Li_3BN_2

References

- 1 S. J. Clark, M. D. Segall, C. J. Pickard, P. J. Hasnip, M. I. J. Probert, K. Refson, M. C. Payne, Z. Kristallogr. Cryst. Mater. 2005, **220**, 567-572.
- 2 Perdew, J. P.; Burke, K.; Ernzerhof, M. Phys. Rev. Lett. 1996, 77, 3865.
- 3 Rappe, A.; Rabe, K.; Kaxiras, E.; Joannopoulos, J. D. Phys. Rev. B 1990, 41, 1227.