

Supporting Information

Pressure-dependent structure and electronic properties of energetic NTO crystal dominated by hydrogen-bonding interactions

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Table S1. The fractional atomic coordinates and lattice vectors of NTO at ambient condition.

lattice vectors	9.2413997650	0.0000000000	0.0000000000
	0.0000000000	5.5838999748	0.0000000000
	-1.5979251633	0.0000000000	8.8859687293
atomic coordinates	<i>x</i>	<i>y</i>	<i>z</i>
H ₁	0.569580011	0.615239960	0.196170013
H ₂	0.430420002	0.384760040	0.803830075
H ₃	0.430419994	0.115239981	0.303830024
H ₄	0.569580007	0.884760040	0.696169998
H ₅	0.960030000	0.013049960	0.316989993
H ₆	0.039969989	0.986950021	0.683010029
H ₇	0.039969982	0.513049979	0.183010018
H ₈	0.960029956	0.486950021	0.816990044
C ₁	0.797939992	0.740069991	0.226899999
C ₂	0.202059989	0.259930009	0.773100050
C ₃	0.202059986	0.240069991	0.273100026
C ₄	0.797939995	0.759930051	0.726900023
C ₅	0.736450033	0.047369959	0.356929994
C ₆	0.263549982	0.952630025	0.643070028
C ₇	0.263549968	0.547369975	0.143070017
C ₈	0.736450040	0.452630025	0.856930045
N ₁	0.649829972	0.738939959	0.239230007
N ₂	0.350170029	0.261059998	0.760770015
N ₃	0.350170031	0.238940002	0.260770018
N ₄	0.649829980	0.761059998	0.739230058
N ₅	0.611710010	0.930219985	0.320190034
N ₆	0.388289996	0.069779993	0.679810014
N ₇	0.388289968	0.430220028	0.179809990
N ₈	0.611710013	0.569779972	0.820190059

N ₉	0.851850010	0.946269976	0.304790007
N ₁₀	0.148150025	0.053730013	0.695209988
N ₁₁	0.148150019	0.446269976	0.195210017
N ₁₂	0.851850022	0.553730024	0.804790085
N ₁₃	0.748460038	0.261870034	0.448280021
N ₁₄	0.251540020	0.738129988	0.551720027
N ₁₅	0.251540016	0.761870055	0.051719996
N ₁₆	0.748460040	0.238129966	0.948280046
O ₁	0.862450034	0.589519967	0.162850014
O ₂	0.137550011	0.410480033	0.837150034
O ₃	0.137549997	0.089519977	0.337149983
O ₄	0.862449985	0.910479991	0.662850039
O ₅	0.870559956	0.359689946	0.473120019
O ₆	0.129440012	0.640310054	0.526880029
O ₇	0.129440020	0.859689946	0.026879997
O ₈	0.870560001	0.140310054	0.973119990
O ₉	0.636480025	0.329949983	0.494569986
O ₁₀	0.363520023	0.670050017	0.505430009
O ₁₁	0.363520025	0.829949941	0.005430013
O ₁₂	0.636479995	0.170050027	0.994570118

Table S2. Characteristics of vibrational modes in NTO crystal at ambient pressure. Abbreviation: asym: asymmetric, st: stretching, bend: bending, sci: scissoring, def: deformation, wag: wagging, tw: twisting, tor: torsion, rock: rocking.

Mode	Wavenumber/cm ⁻¹					Assignment
	Expt. ^{S1}	A _g	B _g	A _u	B _u	
ν_1		3140	3160	3139	3175	N-H st.
ν_2		2841	2885	2844	2912	N-H st.
ν_3		1649	1699	1663	1672	C=O st. + N-H bend.
ν_4		1531	1535	1533	1532	N-H bend. + C-N st.
ν_5		1462	1464	1453	1463	NO ₂ asym. st + N-H bend. in plane
ν_6	1395	1451	1452	1440	1442	N-H bend. in plane
ν_7	1323	1399	1403	1396	1405	N-H bend. in plane
ν_8	1271	1292	1303	1299	1298	N-H bend. in plane
ν_9	1238	1265	1267	1265	1265	C-N st. + N-H bend.
ν_{10}		1208	1217	1210	1212	C-N st. + N-H bend.
ν_{11}	1090	1093	1089	1099	1091	C-N st. + N-H bend.
ν_{12}	1000	984	1000	985	1000	C-N st. + N-H bend.
ν_{13}		976	971	975	978	N-H bend. out of plane
ν_{14}	881	894	914	875	887	N-H bend. out of plane
ν_{15}	789	805	810	808	808	NO ₂ sci. + Ring def.
ν_{16}	754	741	759	739	754	N-H wag. out of plane + C-N st.
ν_{17}	724	733	739	736	736	N-H wag. out of plane + C-N st.
ν_{18}	710	723	733	724	733	N-H wag. out of plane
ν_{19}	662	684	691	686	686	N-H wag. out of plane
ν_{20}		602	593	606	589	Ring def. + N-H bend. in plane
ν_{21}	571	574	574	574	575	Ring rock. + N-H bend. in plane
ν_{22}	468	454	473	454	471	Ring def.
ν_{23}	458	404	405	405	403	Ring def. + NO ₂ sci.
ν_{24}	339	335	341	343	335	Ring def. + C=O bend. out of plane
ν_{25}	210	220	238	213	246	Ring def. + C=O bend. out of plane
ν_{26}	186	186	161	176	184	Ring def. + C=O bend. out of plane
ν_{27}	165	165	133	160	141	Ring def. + NO ₂ tw.
ν_{28}	137	151	116	144	125	Ring def. + NO ₂ tw.
ν_{29}		121	100	116	117	Ring def. + NO ₂ tw.
ν_{30}	77	96	82	96	88	Ring def. + NO ₂ tor.
ν_{31}		77	65	86	81	Ring def. + NO ₂ tor.
ν_{32}		29	48	31	38	Ring def. + NO ₂ tor.
ν_{33}		15	34	28	30	Ring def. + NO ₂ tor.

Table S3. The frequencies of low-wavenumber vibrational modes (below 100 cm⁻¹) at ambient and pressure conditions.

Pressure/GPa	Wavenumber/cm ⁻¹							
	Raman modes				IR modes			
	ν_{30}	ν_{31}	ν_{32}	ν_{33}	ν_{30}	ν_{31}	ν_{32}	ν_{33}
0	96	77	29	19	96	86	31	29
1	122	99	83	32	149	104	43	37
2	110	83	48	35	140	69	56	38
3	160	106	87	38	152	140	70	51
4	157	127	110	42	145	136	92	66
5	151	121	102	43	131	123	84	72
6	175	150	89	49	159	148	86	78
7	175	162	92	51	161	150	95	83
8	182	148	76	59	161	145	91	86
9	182	177	106	67	175	147	113	99
10	166	161	104	79	171	147	65	81
11	172	160	108	95	-	-	-	-
12	180	162	115	97	-	-	-	-
13	187	166	120	97	-	-	-	-
14	195	172	126	97	-	-	-	-
15	197	177	132	99	-	-	-	-

Table S4. The charge amount (e) on atoms at hydrostatic compression.

Pressure / GPa	C ₁	C ₂	N ₁	N ₂	N ₃	N ₄	O ₁	O ₂	O ₃
0	0.67	0.81	4.55	6.25	6.69	8.13	6.45	6.46	8.00
2	0.66	0.81	4.56	6.24	6.69	8.12	6.46	6.47	7.99
4	0.65	0.82	4.57	6.23	6.70	8.09	6.47	6.47	7.98
6	0.63	0.83	4.59	6.21	6.71	8.07	6.50	6.50	7.96
8	0.63	0.84	4.59	6.20	6.72	8.06	6.51	6.50	7.95
10	0.61	0.84	4.60	6.19	6.72	8.05	6.53	6.51	7.95

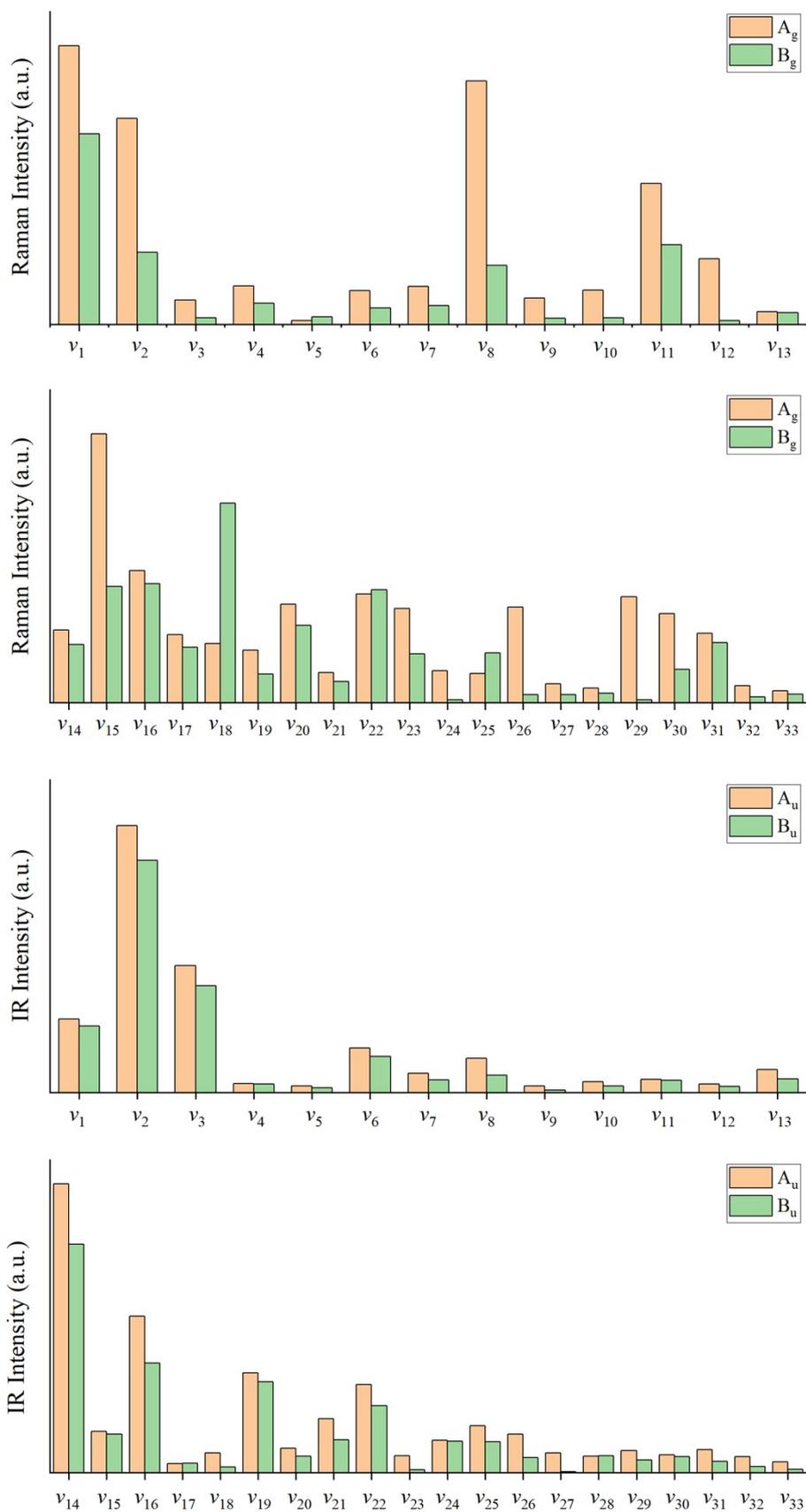


Fig. S1 The intensity of Raman and infrared modes with different symmetries (A_g , B_g , A_u , B_u).

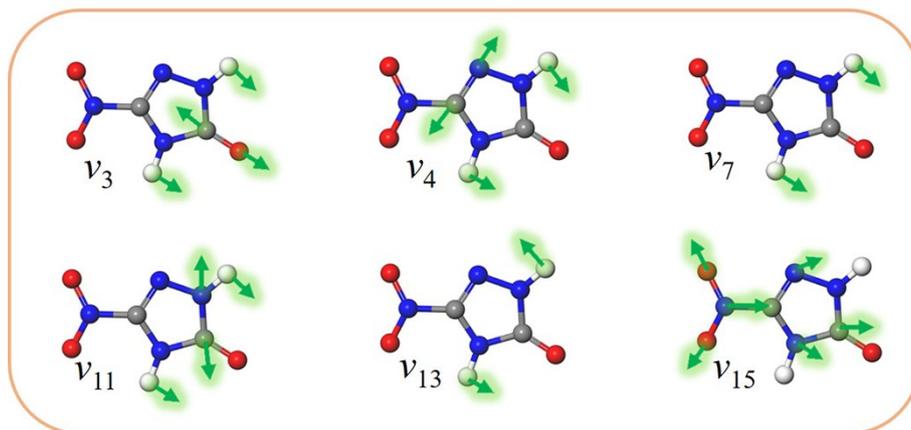


Fig. S2 The visualization of N-H bending modes $v_{7,13}$; and coupled with C=O stretching v_3 , C-N stretching $v_{4,11}$ or O-N-O scissoring modes v_{15} .

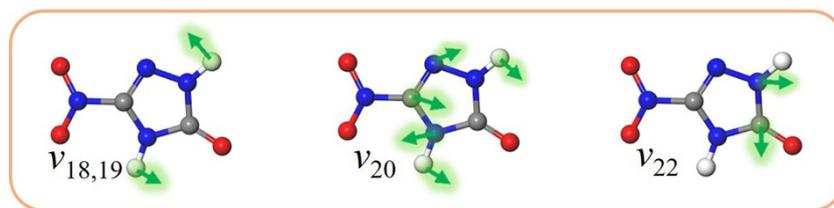


Fig. S3 The visualization of N-H wagging modes $\nu_{18,19}$; and coupled with ring deformation $\nu_{20,22}$.

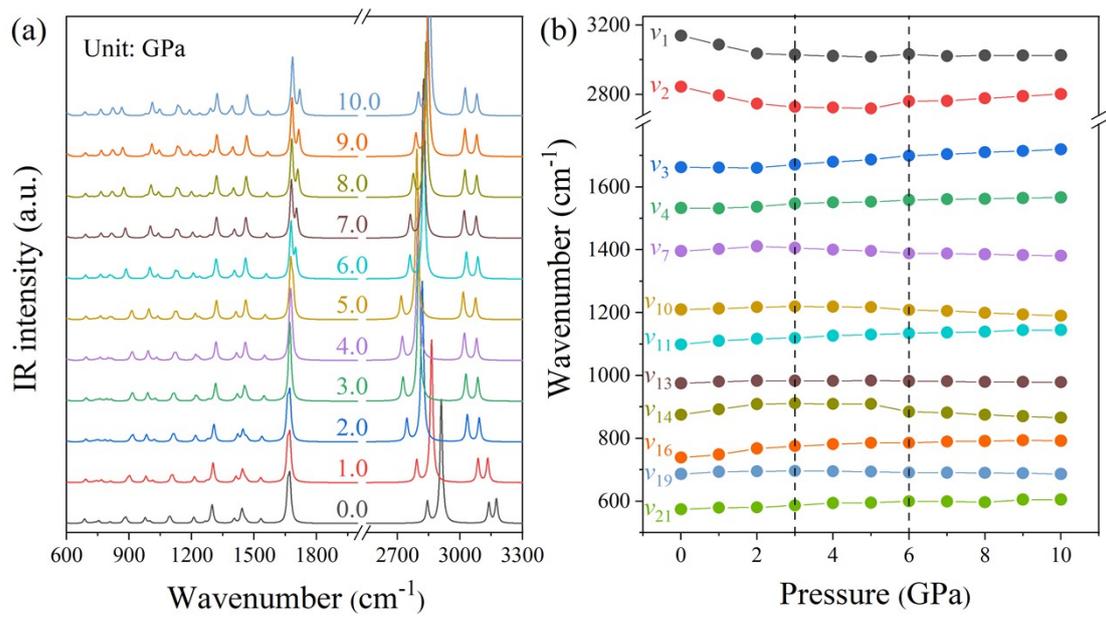


Fig. S4 (a) The variations of infrared spectra under hydrostatic pressure. (b) The wavenumber shifts of selected infrared vibrational modes under pressure.

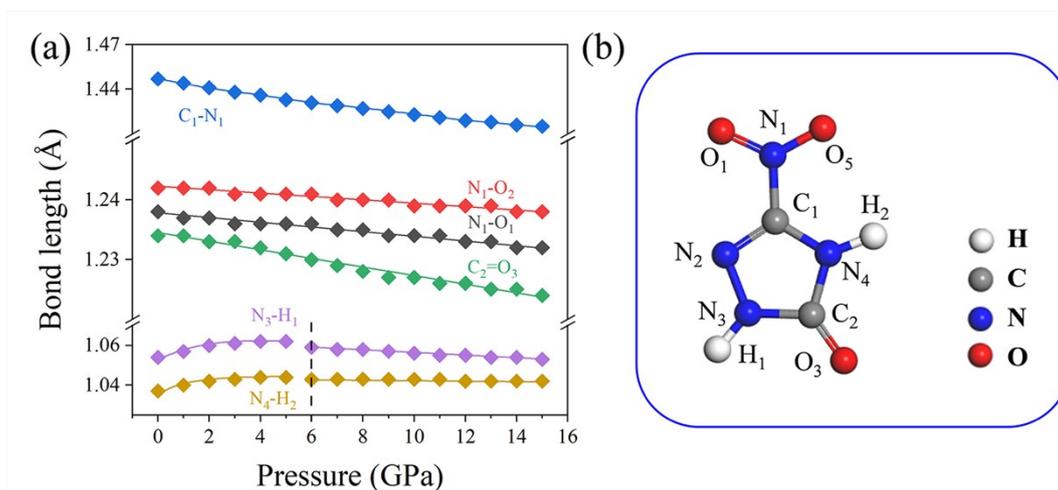


Fig. S5 (a) The evolution of molecular bond lengths under hydrostatic compression; (b) The atom serial numbers of NTO molecule are labeled.

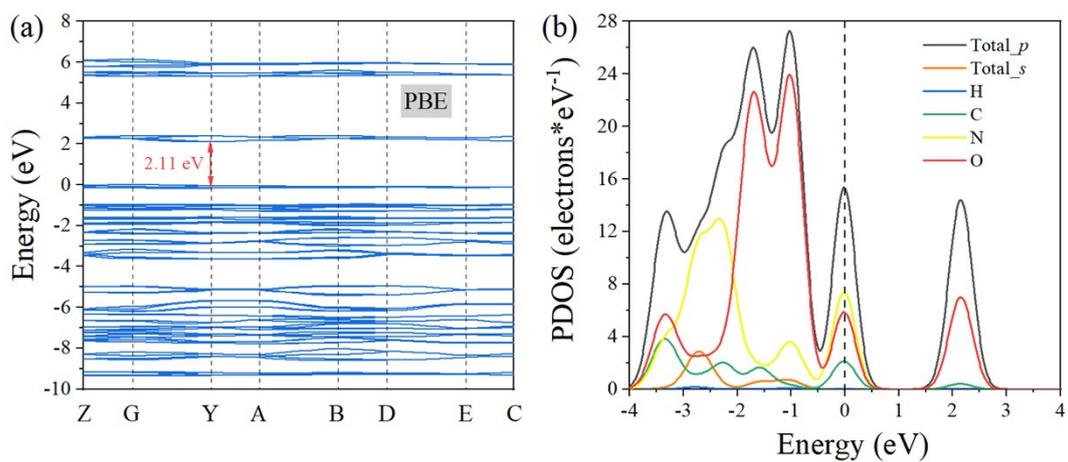


Fig. S6 The Energy band structure (a) and partial density of states (b) from PBE functional.

References

[S1] J. A. Ciezak and S. F. Trevino, *J. Mol. Struct.*, 2005, **732**, 211-218.