

Supporting information

Conformational preferences of the flexible galactofuranose sugar in gas-phase

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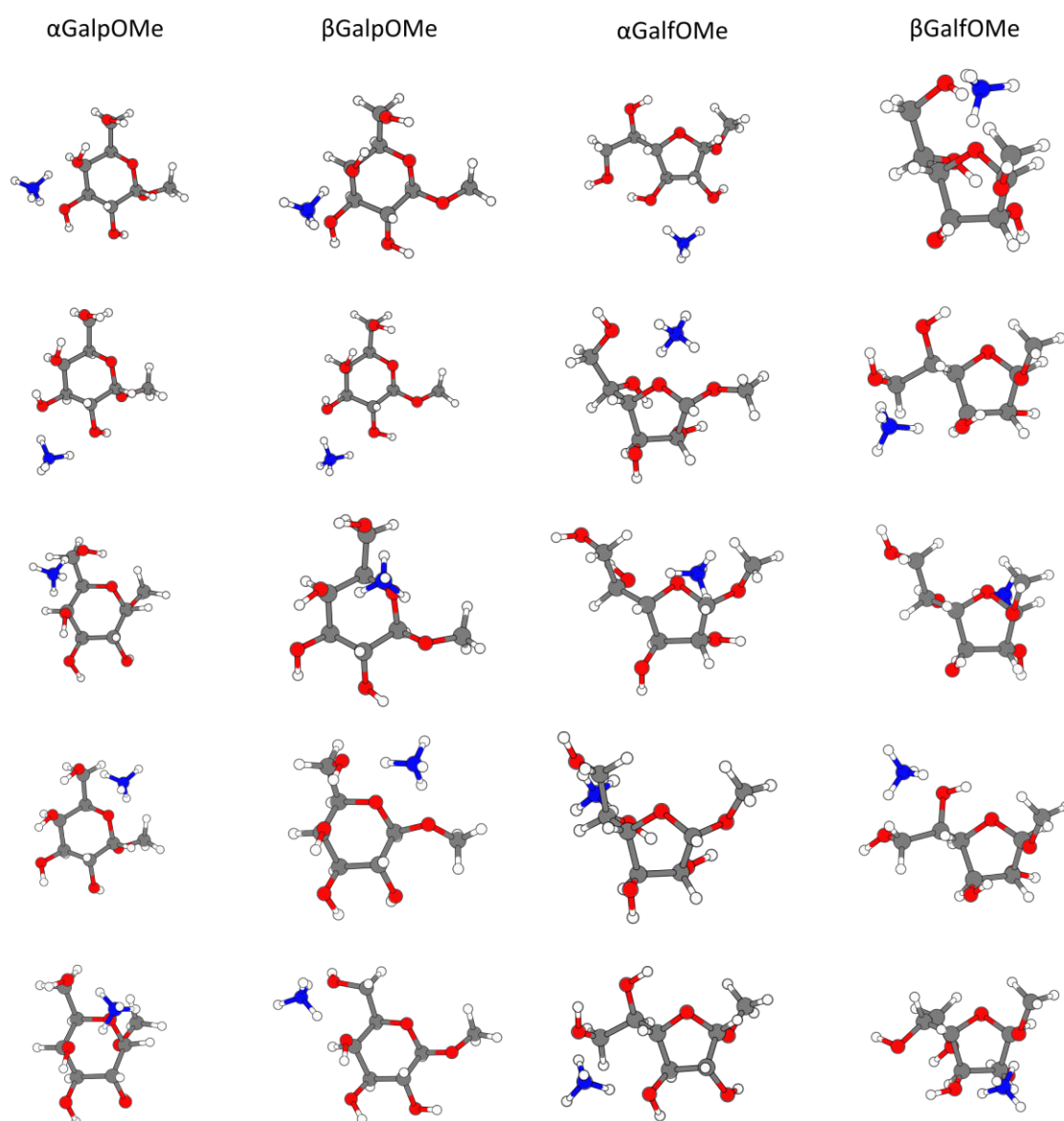


Figure S1 Examples of conformers of α GalpOMe, β GalpOMe, α GalfOMe and β GalfOMe with different NH_4^+ position.

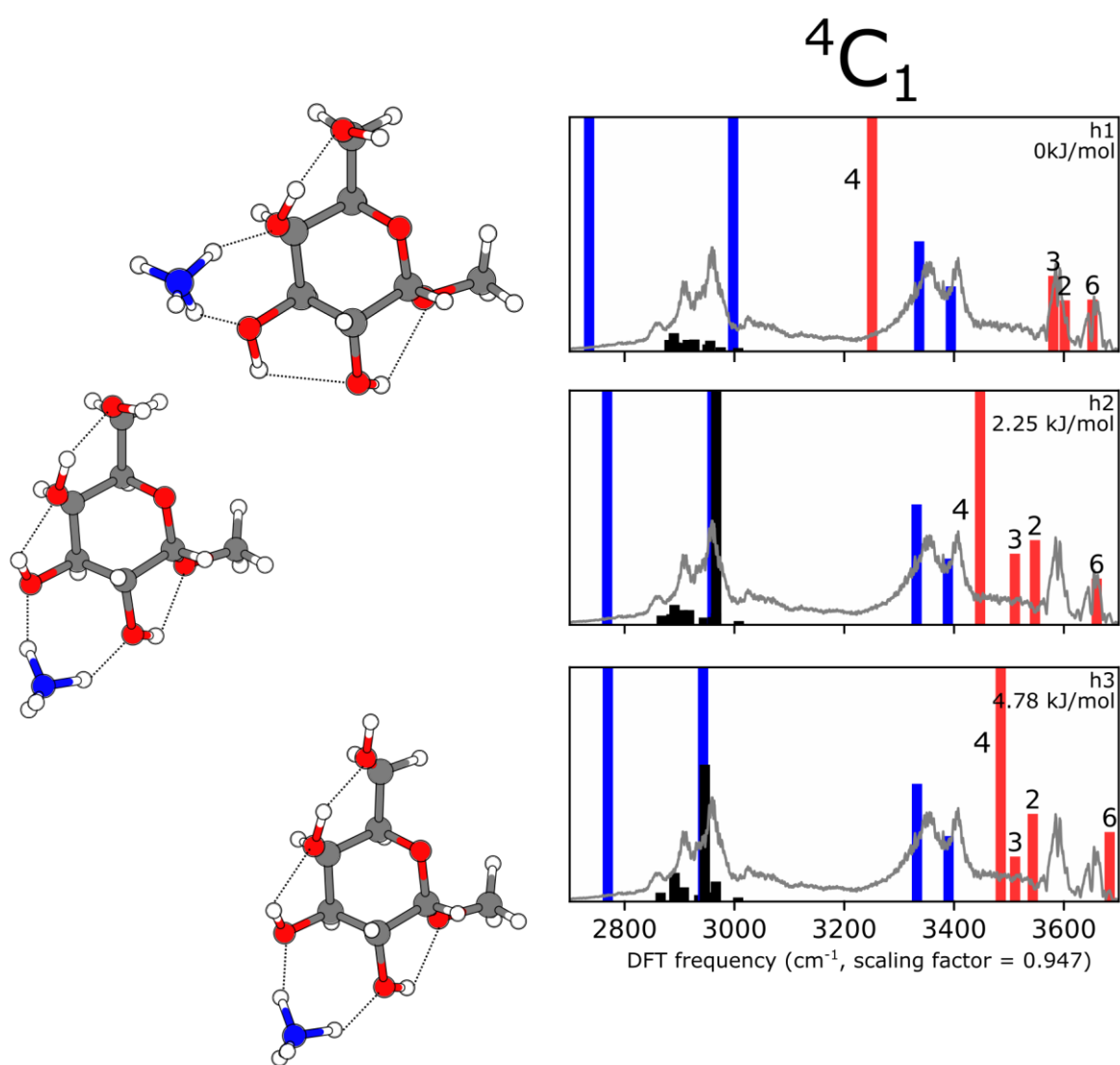


Figure S2 Structures of the most stable conformations for $\alpha\text{GalpOMe}$ and their OH (red), NH (blue) and CH (black) vibrational modes compared to the experimental spectrum (grey). The labels indicate the ring OH position. The relative energies are given in reference to the most stable form h1 (only conformers with relative energy under 8.3 kJ/mol are represented). Hydrogen bonds are represented by dotted lines.

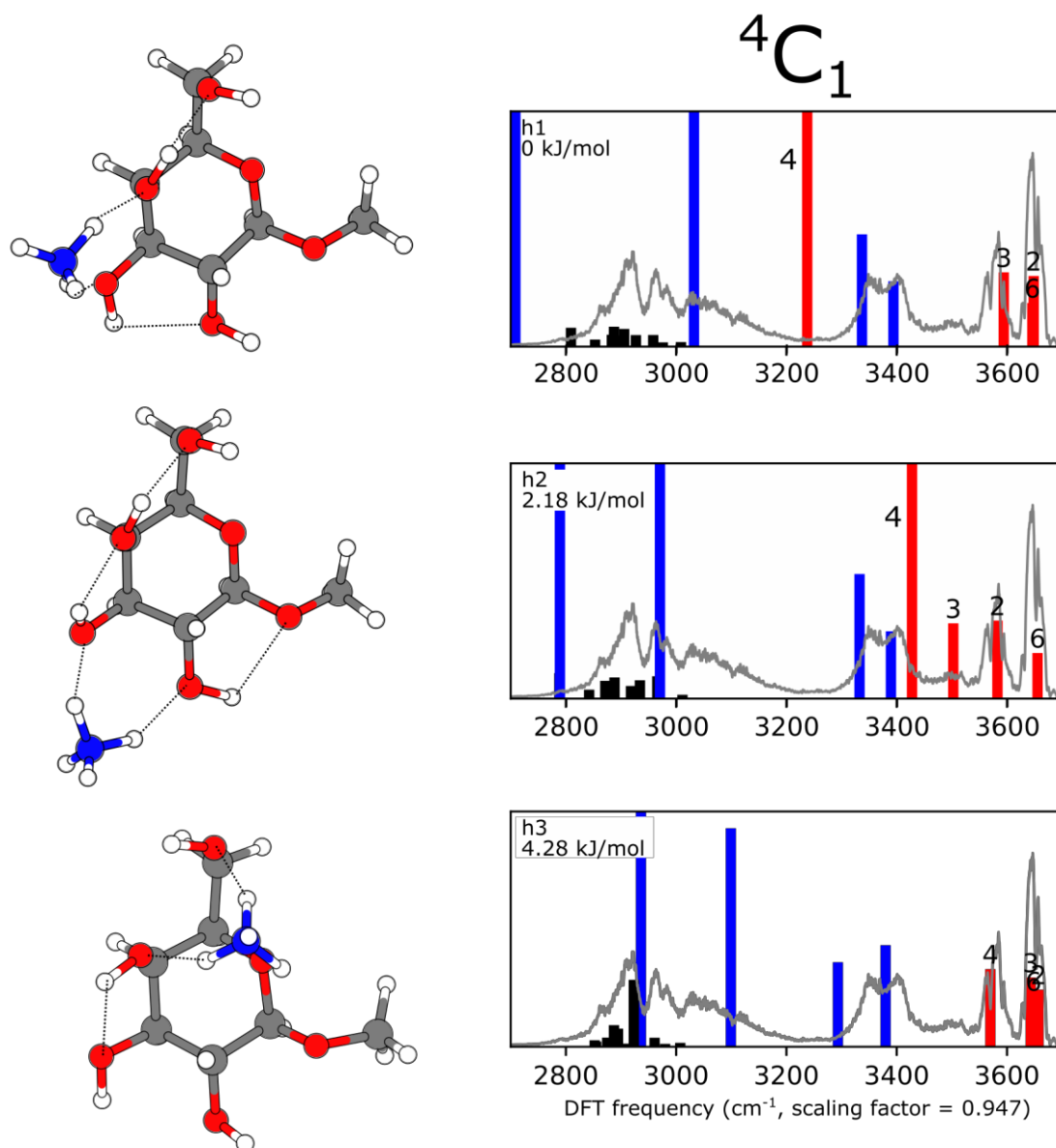
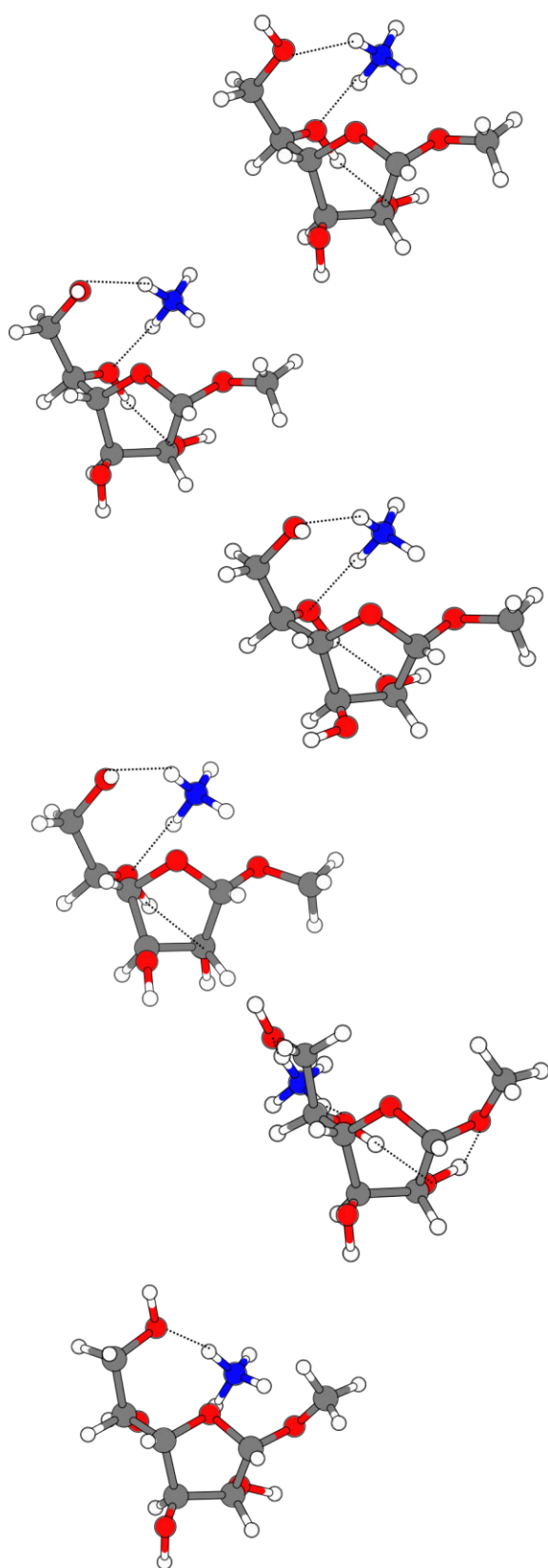
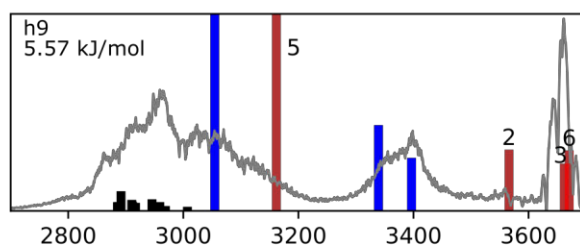
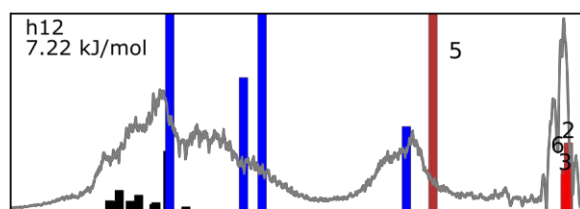
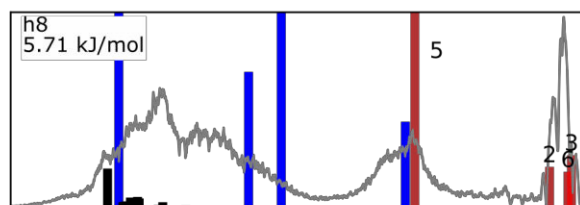
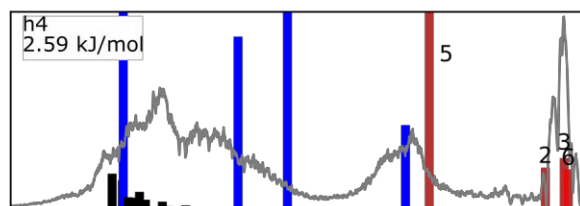
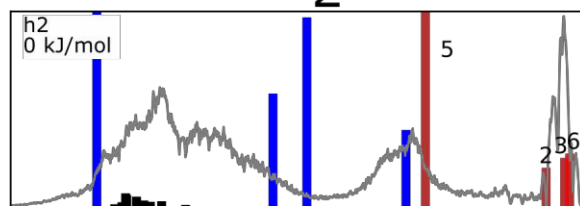


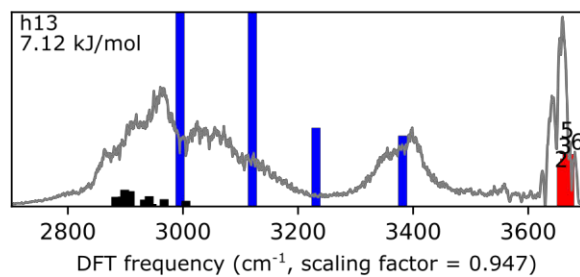
Figure S3 Structures of the most stable conformations for β GalpOMe and their OH (red), NH (blue) and CH (black) vibrational modes compared to the experimental spectrum (grey). The labels indicate the ring OH position. The relative energies are given in reference to the most stable form h1 (only conformers with relative energy under 8.3 kJ/mol are represented). Hydrogen bonds are represented by dotted lines.



E_2



1E



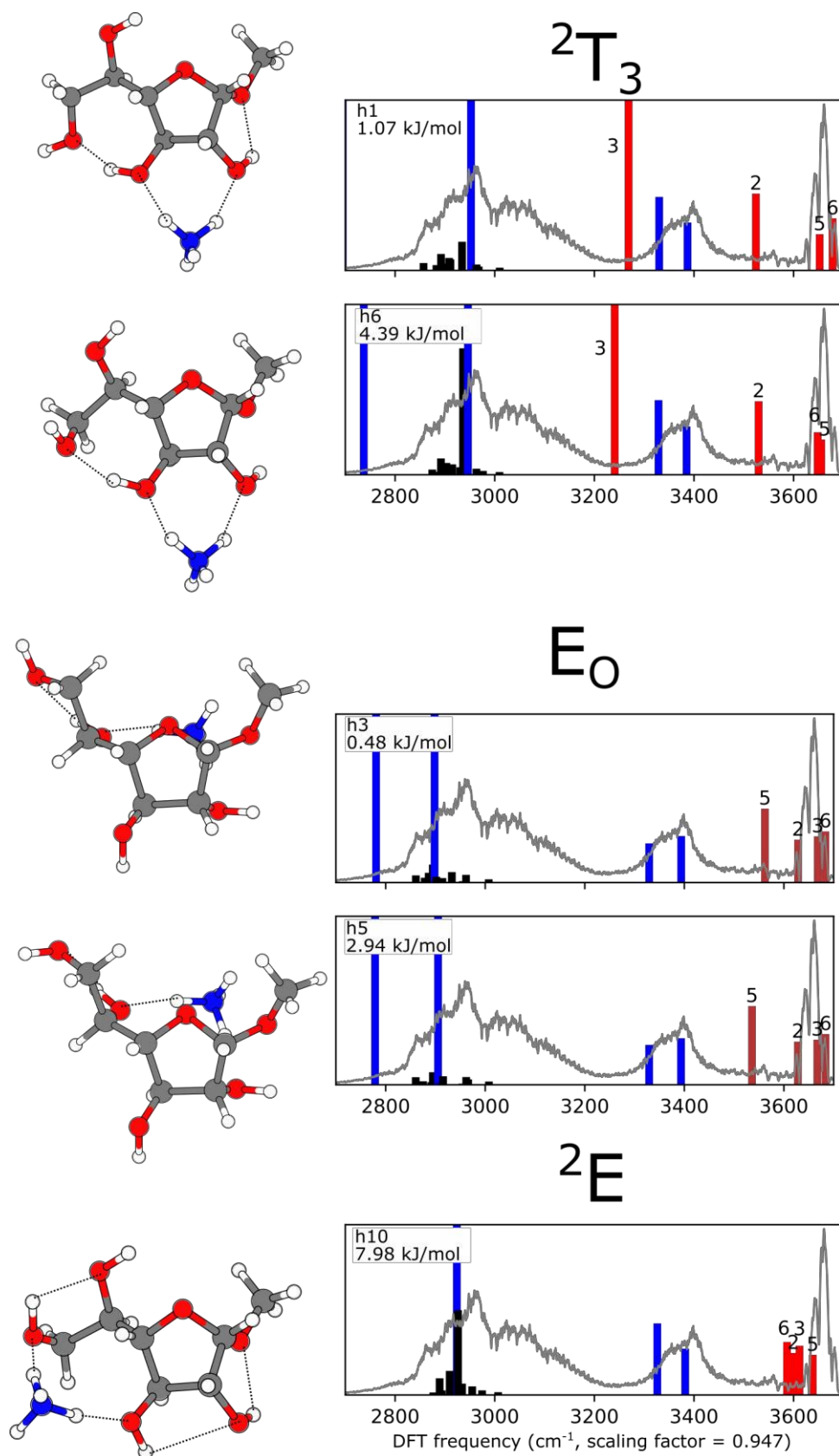
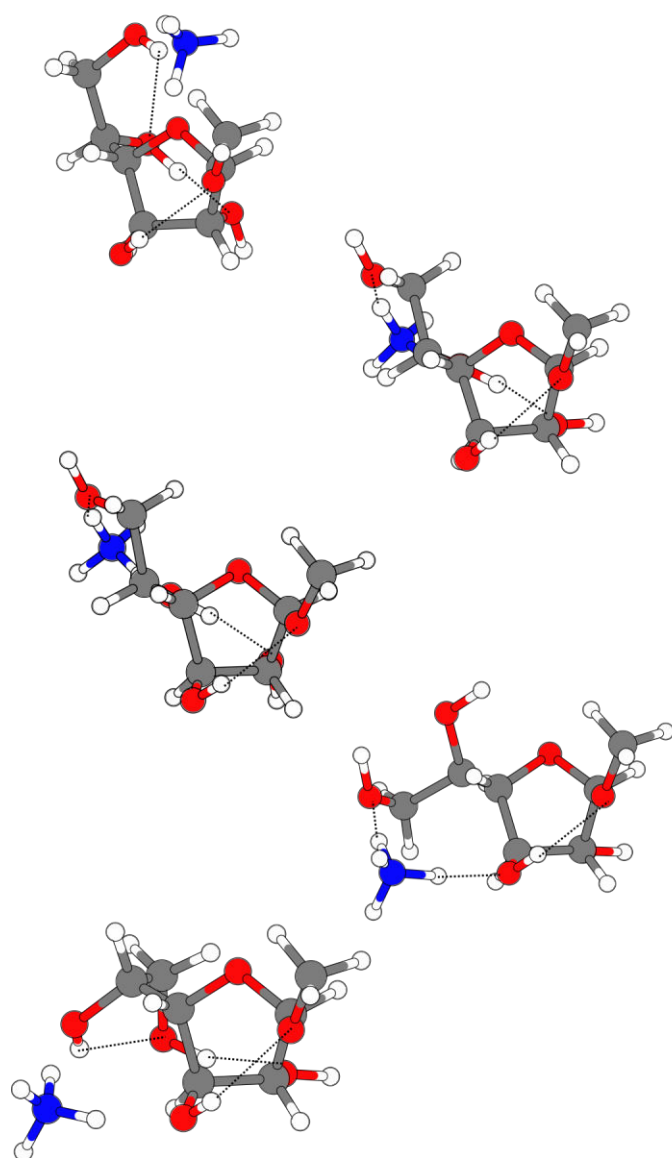
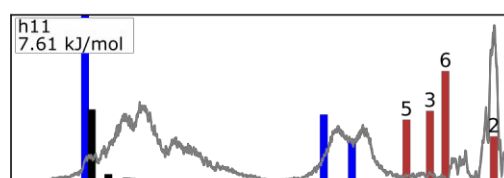
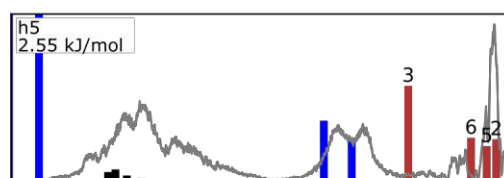
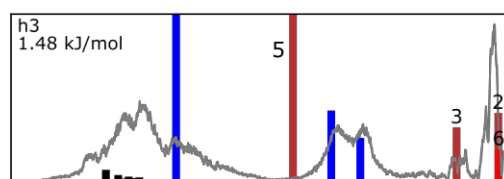
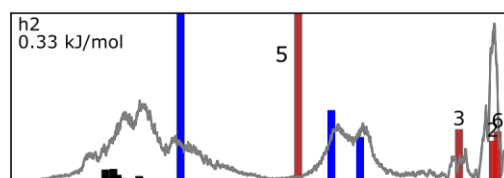
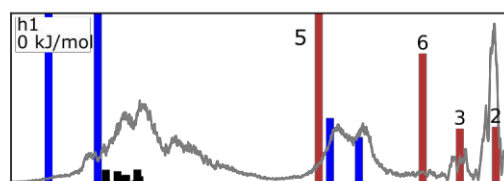
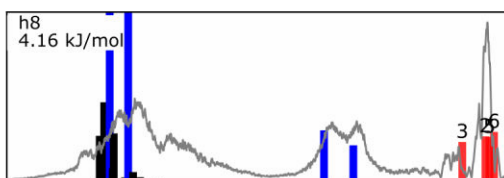
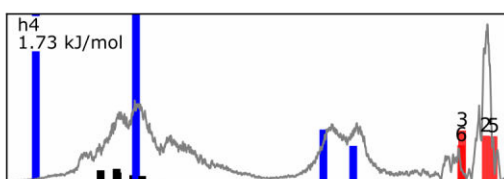


Figure S4 Structures of the most stable conformations for $\alpha\text{GalfOMe}$, gathered according to their geometry type, and their OH (red), NH (blue) and CH (black) vibrational modes compared to the experimental spectrum (grey). The labels indicate the ring OH position. The relative energies are given in reference to the most stable form h1 (only conformers with relative energy under 8.3 kJ/mol are represented). Hydrogen bonds are represented by dotted lines.


 1T_2

 1E


DFT frequency (cm^{-1} , scaling factor = 0.947)

1T_0

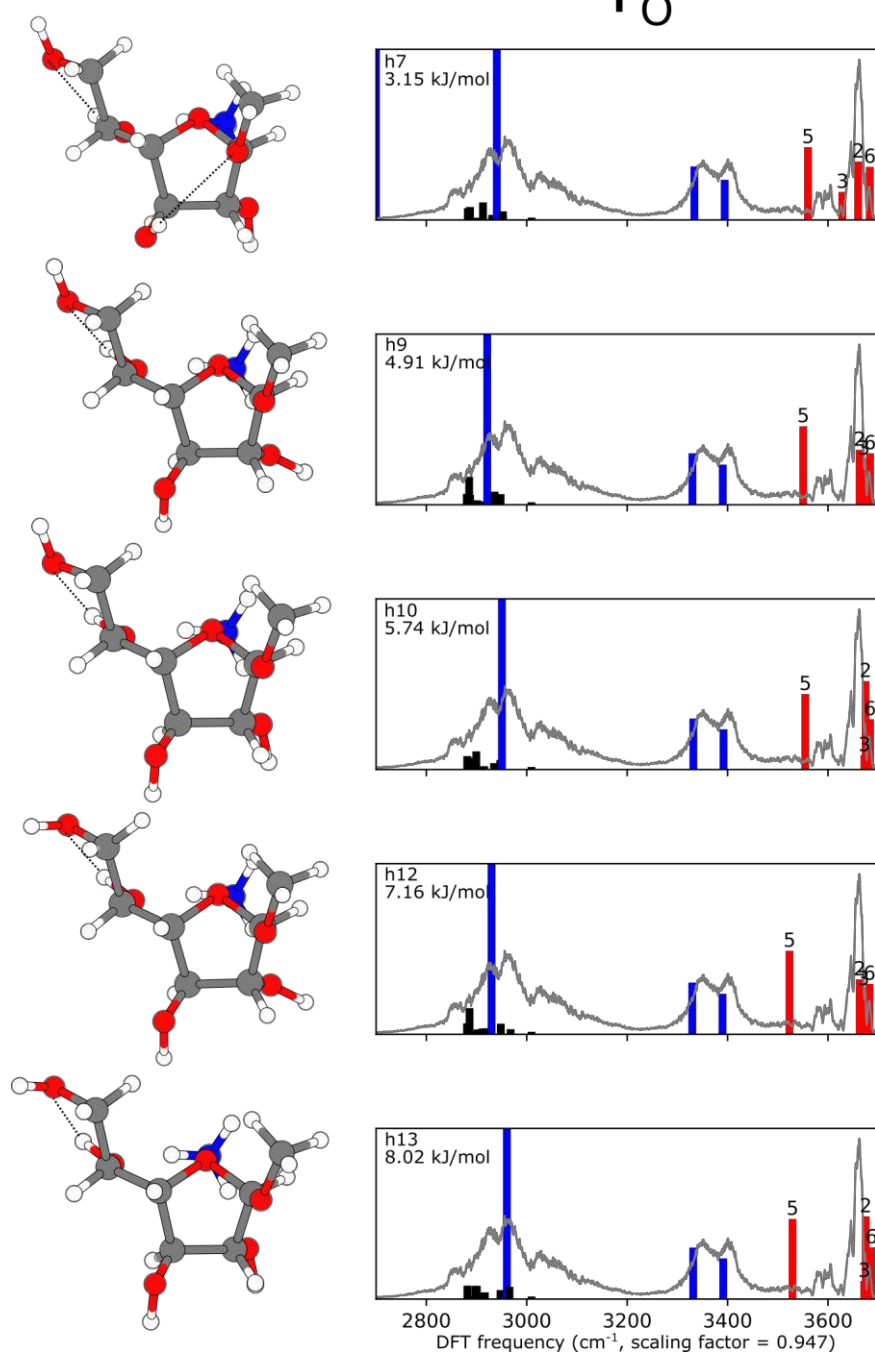


Figure S5 Structures of the most stable conformations for β GalfOMe, gathered according to their geometry type, and their OH (red), NH (blue) and CH (black) vibrational modes compared to the experimental spectrum (grey). The labels indicate the ring OH position. The relative energies are given in reference to the most stable form h1 (only conformers with relative energy under 8.3 kJ/mol are represented). Hydrogen bonds are represented by dotted lines.