

Supporting Information

Understanding the Polaron Behavior in $\text{Cs}_2\text{CuSbCl}_6$ Halide Double Perovskite

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Supplementary Figures



Fig. S1. Schematic picture of the solvothermal method to obtain $\text{Cs}_2\text{CuSbCl}_6$ single crystals.

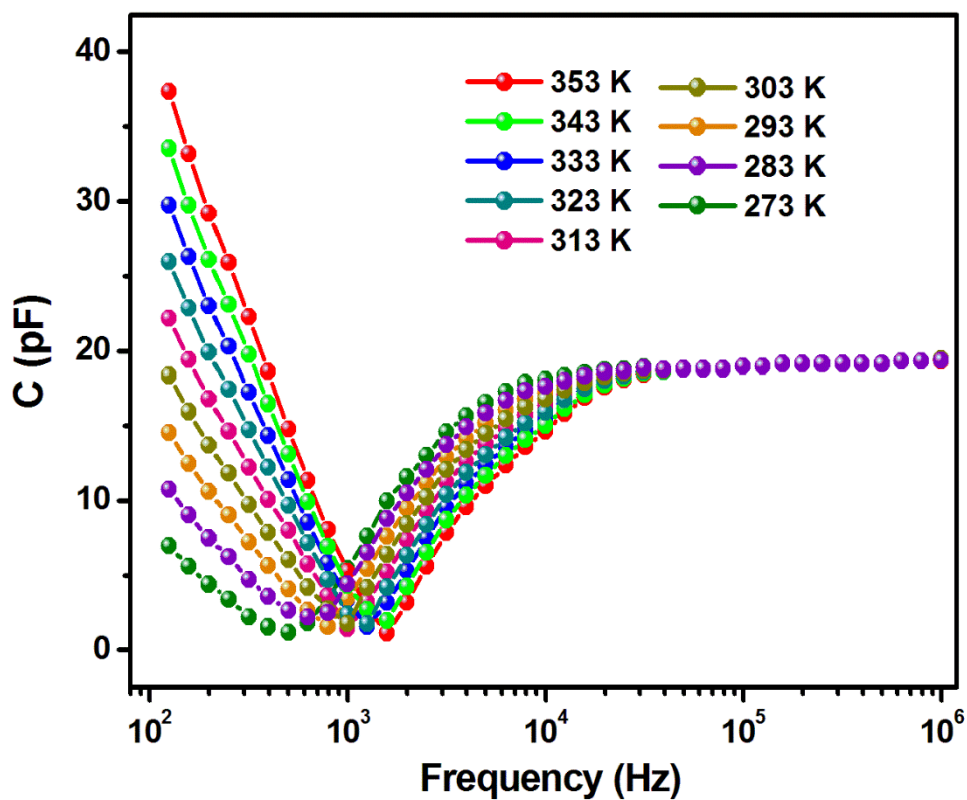


Fig. S2. Frequency dependence of capacitance for $\text{Cs}_2\text{CuSbCl}_6$ at various temperature.

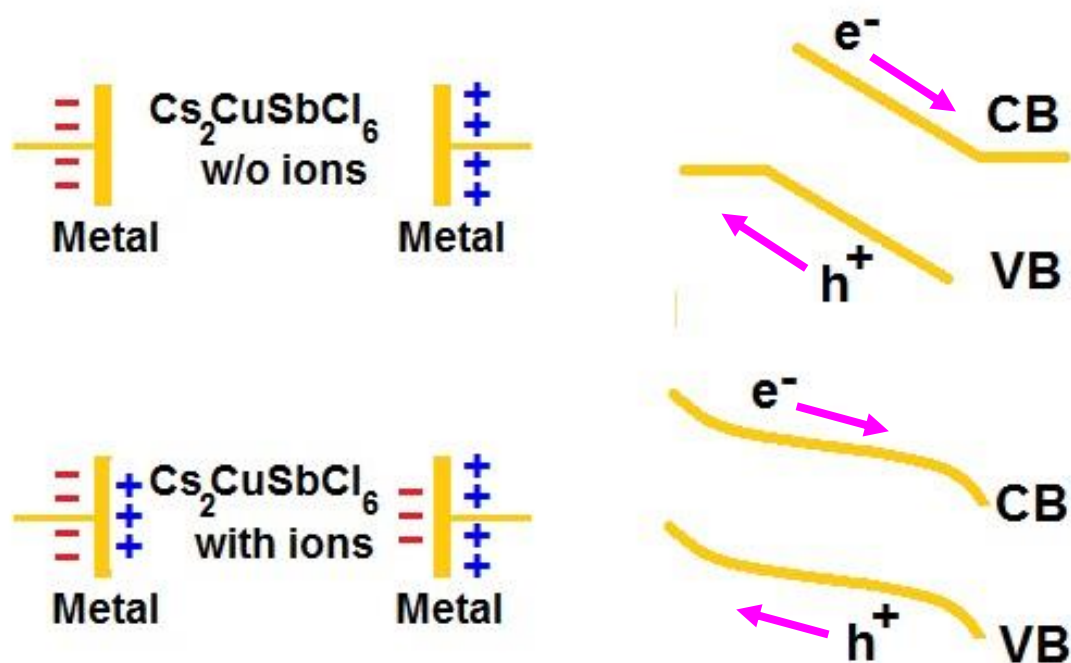


Fig. S3. Metal/ $\text{Cs}_2\text{CuSbCl}_6$ /metal band diagram demonstrating ion accumulation.

Supplementary Tables

Table S1. Proportions of Chemical Elements in $\text{Cs}_2\text{CuSbCl}_6$ SCs.

<i>Chemical element</i>	<i>% Atomic</i>
Caesium (Cs)	22
Copper (Cu)	9
Antimony (Sb)	11
Chloride (Cl)	58

Table S2. Thermal variation of parameters n_1 , n_2 and n_3 .

<i>Temperature (K)</i>	n_1	n_2	n_3
273	0.95	0.27	0.62
283	0.93	0.18	0.60
293	0.90	0.22	0.59
303	0.87	0.21	0.56
313	0.82	0.23	0.53
323	0.76	0.19	0.47
333	0.70	0.21	0.45
343	0.64	0.23	0.43
353	0.57	0.24	0.42

Table S3. The values of 'm' were calculated based on the fitted plots of $\epsilon''(\omega)$ obtained at various temperatures

Temperature	Value of 'm'
273	-0.51
283	-0.54
293	-0.59
303	-0.63
313	-0.66
323	-0.71
333	-0.75
343	-0.79
353	-0.82