

Structural, Mechanical, Electronic and Optical Properties of Biphenylene Hydrogenation: A First-Principles Study

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Supporting Information

The carbon-carbon band length of α , β and γ configurations numbered in Fig. 1 are shown in Table S1.

Table S1 The carbon-carbon band length of α , β and γ configurations

	Band length(Å)					
Biphenylene	$d_1 = 1.448$	$d_2 = 1.407$		$d_3 = 1.457$	$d_4 = 1.456$	
Configuration α	$d_5 = 1.573$	$d_6 = 1.566$		$d_7 = 1.541$	$d_8 = 1.579$	
Configuration β	$d_9 = 1.574$	$d_{10} = 1.514$	$d_{11} = 1.557$	$d_{12} = 1.532$	$d_{13} = 1.591$	$d_{14} = 1.359$
Configuration γ	$d_{15} = 1.502$	$d_{16} = 1.500$		$d_{17} = 1.519$	$d_{18} = 1.541$	

Refractive index n , extinction coefficient k , absorption coefficient α , transmittance T , reflectivity R , optical conductivity σ and energy-loss function L of α , β and γ configurations are plotted in Fig. S1.

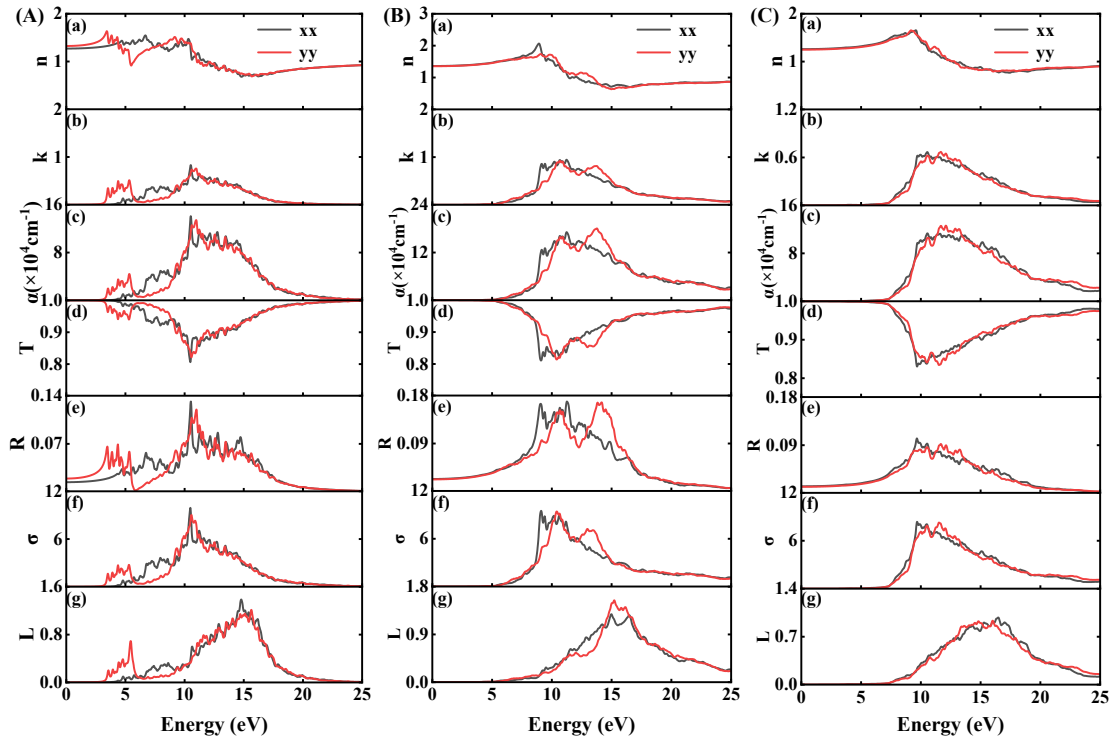


Fig. S1 The calculated optical properties of α (left panel), β (middle panel) and γ (right panel) configurations. There are two different directions of light polarization, xx and yy directions. (a) Refractive index n , (b) extinction coefficient k , (c) absorption coefficient α , (d) transmittance T , (e) reflectivity R , (f) optical conductivity σ , and (g) energy-loss function L .