## Electronic Supplementary Information

## Prediction of 2D group-11 chalcogenides: Insights into novel auxetic M<sub>2</sub>X (M =

## Cu, Ag, Au; X = S, Se, Te) monolayers

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**Figure S1.** (a)-(i) Relative energies of each  $M_2X$  monolayer with different lattice parameters. Red represents the most energetically favorable structure.



**Figure S2.** (a)-(i) Strain energies of  $\alpha$ -Cu<sub>2</sub>S,  $\alpha$ -Cu<sub>2</sub>Se,  $\alpha$ -Cu<sub>2</sub>Te,  $\beta$ -Ag<sub>2</sub>S,  $\beta$ -Ag<sub>2</sub>Se,  $\alpha$ -Ag<sub>2</sub>Te,  $\beta$ -Au<sub>2</sub>S,  $\beta$ -Au<sub>2</sub>Se, and  $\alpha$ -Au<sub>2</sub>Te monolayers with respect to various uniaxial strains along the x direction.



**Figure S3.** (a)-(i) Strain energies of  $\alpha$ -Cu<sub>2</sub>S,  $\alpha$ -Cu<sub>2</sub>Se,  $\alpha$ -Cu<sub>2</sub>Te,  $\beta$ -Ag<sub>2</sub>S,  $\beta$ -Ag<sub>2</sub>Se,  $\alpha$ -Ag<sub>2</sub>Te,  $\beta$ -Au<sub>2</sub>S,  $\beta$ -Au<sub>2</sub>Se, and  $\alpha$ -Au<sub>2</sub>Te monolayers with respect to various equi-biaxial strains along the x and y directions.



**Figure S4.** (a)-(i) Strain energies of  $\alpha$ -Cu<sub>2</sub>S,  $\alpha$ -Cu<sub>2</sub>Se,  $\alpha$ -Cu<sub>2</sub>Te,  $\beta$ -Ag<sub>2</sub>S,  $\beta$ -Ag<sub>2</sub>Se,  $\alpha$ -Ag<sub>2</sub>Te,  $\beta$ -Au<sub>2</sub>S,  $\beta$ -Au<sub>2</sub>Se, and  $\alpha$ -Au<sub>2</sub>Te monolayers with respect to various shear strains along the xy directions.



**Figure S5.** (a)-(i) Calculated orientation-dependent Poisson's ratio v( $\theta$ ) of  $\alpha$ -Cu<sub>2</sub>S,  $\alpha$ -Cu<sub>2</sub>Se,  $\alpha$ -Cu<sub>2</sub>Te,  $\beta$ -Ag<sub>2</sub>S,  $\beta$ -Ag<sub>2</sub>Se,  $\alpha$ -Ag<sub>2</sub>Te,  $\beta$ -Au<sub>2</sub>S,  $\beta$ -Au<sub>2</sub>Se, and  $\alpha$ -Au<sub>2</sub>Te monolayers, respectively.



**Figure S6.** (a)-(i) Calculated orientation-dependent Young's modulus Y( $\theta$ ) of  $\alpha$ -Cu<sub>2</sub>S,  $\alpha$ -Cu<sub>2</sub>Se,  $\alpha$ -Cu<sub>2</sub>Te,  $\beta$ -Ag<sub>2</sub>S,  $\beta$ -Ag<sub>2</sub>Se,  $\alpha$ -Ag<sub>2</sub>Te,  $\beta$ -Au<sub>2</sub>S,  $\beta$ -Au<sub>2</sub>Se, and  $\alpha$ -Au<sub>2</sub>Te monolayers, respectively.



**ure S7.** (a)-(i) Variation of transverse strains of  $\alpha$ -Cu<sub>2</sub>S,  $\alpha$ -Cu<sub>2</sub>Se,  $\alpha$ -Cu<sub>2</sub>Te,  $\beta$ -Ag<sub>2</sub>S,  $\beta$ -Ag<sub>2</sub>Se,  $\alpha$ -Ag<sub>2</sub>Te,  $\beta$ -Au<sub>2</sub>S,  $\beta$ -Au<sub>2</sub>Se, and  $\alpha$ -Au<sub>2</sub>Te monolayers with respect to tensile strains along x direction. The pink and blue shaded regions represent positive Poisson's function (+PF) and negative Poisson's function (-PF), respectively.



**Figure S8.** (a) Top and side views of the  $\alpha$ -Cu<sub>2</sub>S monolayers. (b) The transverse strains of  $\alpha$ -Cu<sub>2</sub>S monolayers with S vacancy in tensile strains along the x direction. The pink shaded region corresponds to the positive Poisson's function (+PF).



**Figure S9.** Top and side views of the  $\alpha$ -Cu<sub>2</sub>S and  $\alpha$ -Ag<sub>2</sub>Te monolayers taken from ab initio molecular dynamic simulations carried out at 300 K for 2 ps.



**Figure S10.** Bond lengths ( $d_1$  and  $d_2$ ) and bond angle ( $\theta$ ) are indicated by red, dark blue, and light blue, respectively.



**Figure S11.** Variation of (a)-(b) bond lengths ( $d_1$  and  $d_2$ ), (c) bond angle ( $\theta$ ) and (d) thickness (h) of  $\alpha$ -Cu<sub>2</sub>Te monolayer. Bond lengths ( $d_1$  and  $d_2$ ) and bond angle ( $\theta$ ) are indicated in Fig. S10.



**Figure S12.** (a)-(f) Projected density of states of  $\alpha$ -Cu<sub>2</sub>S monolayer with +1%, +2%, +3%, +4%, +5%, and +6% tensile strains, respectively. The d/p orbitals of Cu/S are indicated by red/blue.



Figure S13. (a)Top view of  $\alpha$ -Cu<sub>2</sub>S monolayer. (b) Electron localization functions of  $\alpha$ -Cu<sub>2</sub>S with +1%,+3% and +6% tensile strains, respectively.



**Figure S14.** (a)-(f) Projected density of states of  $\alpha$ -Cu<sub>2</sub>Te monolayer with +1%, +2%, +3%, +4%, +5%, and +6% tensile strains, respectively. The d/p orbitals of Cu/Te are indicated by red/blue.



Figure S15. Mechanism of PPF for  $\alpha$ -phase monolayers. The solid and dashed M-X bonds are initial and final configurations at each relaxation step, respectively. The dashed circles represent the movements of M and X atoms.

**Table S1** Structural parameters (*a*, *b* and *h*) of the  $\alpha$ -Cu<sub>2</sub>S,  $\alpha$ -Cu<sub>2</sub>Se,  $\alpha$ -Cu<sub>2</sub>Te,  $\beta$ -Ag<sub>2</sub>S,  $\beta$ -Ag<sub>2</sub>Se,  $\alpha$ -Ag<sub>2</sub>Te,  $\beta$ -Au<sub>2</sub>S,  $\beta$ -Au<sub>2</sub>Se, and  $\alpha$ -Au<sub>2</sub>Te monolayers. *a* (*b*) and *h* are the lattice parameters, thickness of  $\alpha$ -phase and  $\beta$ -phase monolayers, respectively.

| Material                     | <i>a</i> (Å) | <i>b</i> (Å) | <i>h</i> (Å) |
|------------------------------|--------------|--------------|--------------|
| $\alpha$ -Cu <sub>2</sub> S  | 5.02         | 5.02         | 2.55         |
| $\alpha$ -Cu <sub>2</sub> Se | 4.99         | 4.99         | 3.00         |
| $\alpha$ -Cu <sub>2</sub> Te | 4.99         | 4.99         | 3.52         |
| $\beta$ -Ag <sub>2</sub> S   | 5.88         | 5.88         | 2.48         |
| β-Ag <sub>2</sub> Se         | 5.90         | 5.90         | 2.93         |
| α-Ag <sub>2</sub> Te         | 5.71         | 5.71         | 3.54         |
| $\beta$ -Au <sub>2</sub> S   | 5.81         | 5.81         | 2.49         |
| $\beta$ -Au <sub>2</sub> Se  | 5.82         | 5.82         | 2.95         |
| $\alpha$ -Au <sub>2</sub> Te | 5.62         | 5.62         | 3.55         |

| Strains | S1      | S2      | Cu              |
|---------|---------|---------|-----------------|
| +1%     | +0.621e | +0.621e | -0.312e         |
| +3%     | +0.628e | +0.628e | -0.318e         |
| +6%     | +0.645e | +0.645e | -0.323 <i>e</i> |

**Table S2** Bader analysis of  $\alpha$ -Cu<sub>2</sub>S monolayer with +1%, +3% and +6% tensile strains, respectively. S1, S2 and Cu atoms correspond to those of Fig. S13 (a).