## Supporting information for

# Experimental Phase Diagram and Its Temporal Evolution for Submicron 2-Methylglutaric Acid and Ammonium Sulfate Aerosol Particles 

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Summary: The content of this supporting information includes the size distribution of submicron 2MGA+AS particles, example optical images of micrometer droplets at different RH values, the measured ERH of submicron inorganic aerosol particles and literature values $\left(\mathrm{KCl}, \mathrm{K}_{2} \mathrm{SO}_{4}\right)$, and the AIOMFAC prediction of the viscosity of 2MGA+AS aerosols.

## Experimental Methods

## Supplementary Figures and Tables



Figure S1. Size distribution of $2 M G A+A S$ aerosols $\left(m f_{d}(A S)=0.33\right)$. The images were collected using a diffusion dryer coupled with cryo-TEM. LLPS is inhibited for particles smaller than 30 $n m$.


Figure S2. Example optical images of micrometer droplets $\left(m f_{d}(A S)=0.5\right)$, at different $R H$ conditions. The labels under the images denote different phase states of droplets. Scale bar $=2$ $\mu m$.

Table S1. The ERH of KCl and $\mathrm{K}_{2} \mathrm{SO}_{4}$ measured in this study experimentally and measured in the literature by Morris et al. ${ }^{1}$ The temperature of this study was $297 \pm 1 \mathrm{~K}$, which is overall consistent with Morris et al with variation in temperature about 1 K .

| Composition | Experimental ERH (\%) | Literature value(\%) |
| :---: | :---: | :---: |
| $\mathbf{K C l}$ | $56 \pm 1$ | $56 \pm 2$ |
| $\mathbf{K}_{2} \mathbf{S O}_{4}$ | $59 \pm 1$ | $60 \pm 1$ |

Table S2. The estimated viscosity of $2 M G A+A S$ aerosol particles with various compositions at $\sim 78 \%$ RH and 298 K . The $78 \%$ RH is close to the SRH of micrometer-sized $2 M G A+A S$ aerosol particles.

| $\mathbf{m f}_{\mathbf{d}}(\mathbf{A S})$ | $\mathbf{R H}$ | $\boldsymbol{\operatorname { l o g }}_{\mathbf{1 0}}(\mathbf{\eta}(\mathbf{P a} \cdot \mathbf{S}) \mathbf{)}$ |
| :---: | :---: | :---: |
| 0.10 | 78.33 | -1.612 |
| 0.20 | 78.29 | -1.811 |
| 0.33 | 78.50 | -2.014 |
| 0.50 | 78.68 | -2.172 |
| 0.80 | 78.50 | -2.385 |
| 0.90 | 78.34 | -2.470 |

## References:

1 H. S. Morris, A. D. Estillore, O. Laskina, V. H. Grassian and A. V. Tivanski, Anal. Chem., 2016, 88, 3647-3654.

