

Electronic Supplementary Information for: Mechanochemical Decomposition of Tricresyl Phosphate  
between Sliding Ferrous Surfaces

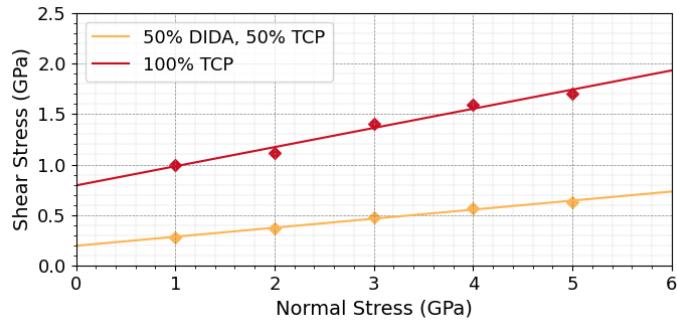
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**Figure S1** Change in shear stress with normal stress for 100% TCP and 50% TCP, 50% DIDA at 600K and 1-5 GPa. Friction coefficient,  $\mu = 0.18$  (100% TCP),  $\mu = 0.17$  (50% TCP, 50% DIDA) and Derjaguin offset,  $\tau_0 = 0.79$  GPa (100% TCP),  $\tau_0 = 0.20$  GPa (50% TCP, 50% DIDA).

Temperature	Stiffness (GPa)						
	300 K	400 K	500 K	600 K	700 K	800 K	900 K
Iron	206	206	202	200	200	199	196
Iron Carbide	200	200	201	198	200	196	181
Iron Oxide	220	216	210	209	206	199	196

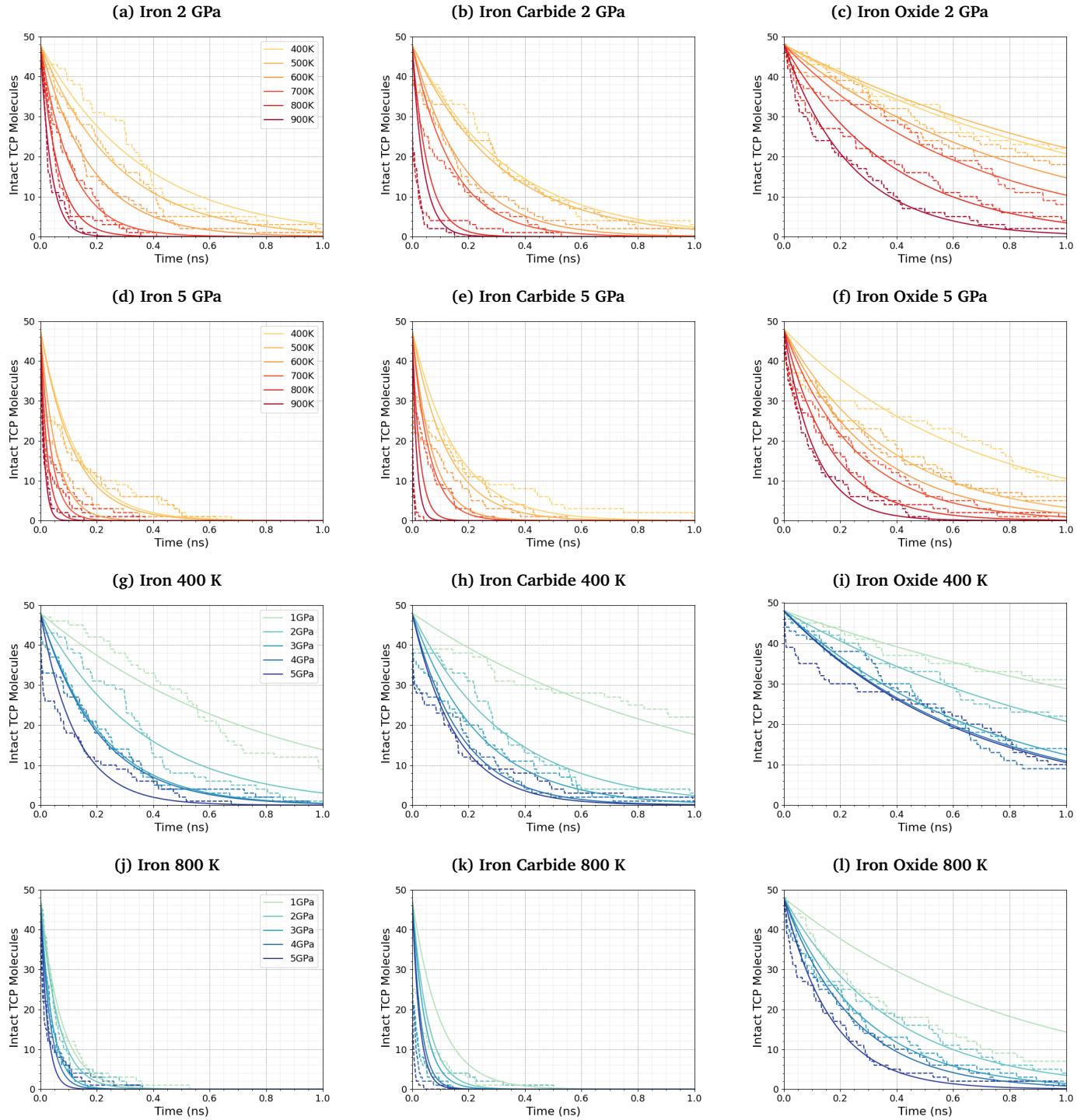
**Table S1** Elastic constant ( $C_{33}$ ) values for  $\alpha$ -Fe(110),  $\text{Fe}_3\text{C}(010)$ , and  $\text{Fe}_3\text{O}_4(001)$  slabs with ReaxFF.

Temperature	Friction Coefficient ( $\mu$ )					
	400 K	500 K	600 K	700 K	800 K	900 K
Iron	0.35	0.22	0.18	0.25	0.17	0.16
Iron Carbide	0.28	0.31	0.23	0.24	0.19	0.20
Iron Oxide	0.24	0.24	0.23	0.23	0.26	0.27

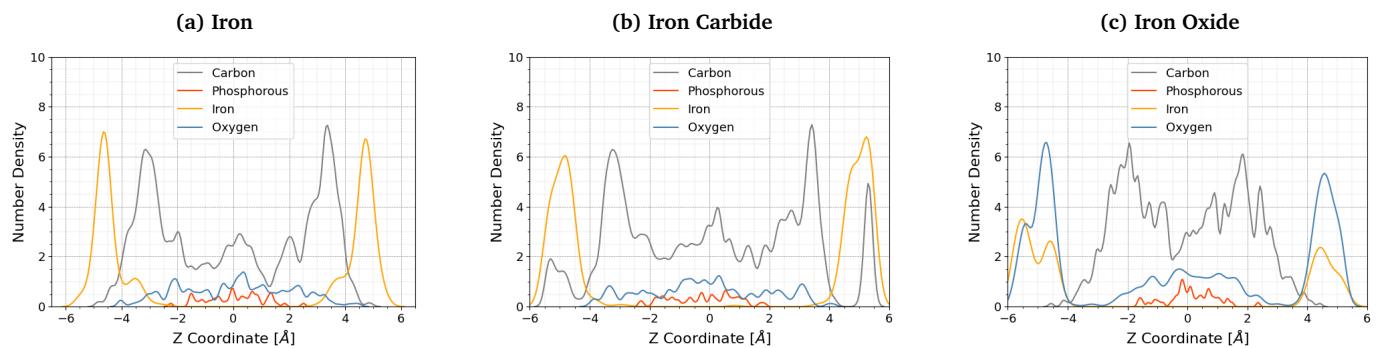
**Table S2** Friction Coefficient,  $\mu$ , for TCP between  $\alpha$ -Fe(110),  $\text{Fe}_3\text{C}(010)$ , and  $\text{Fe}_3\text{O}_4(001)$  surfaces at different temperatures.

Temperature	Derjaguin Offset					
	400 K	500 K	600 K	700 K	800 K	900 K
Iron	0.28	0.72	0.79	0.53	0.75	0.79
Iron Carbide	0.38	0.31	0.66	0.57	0.73	0.77
Iron Oxide	0.26	0.27	0.31	0.32	0.34	0.34

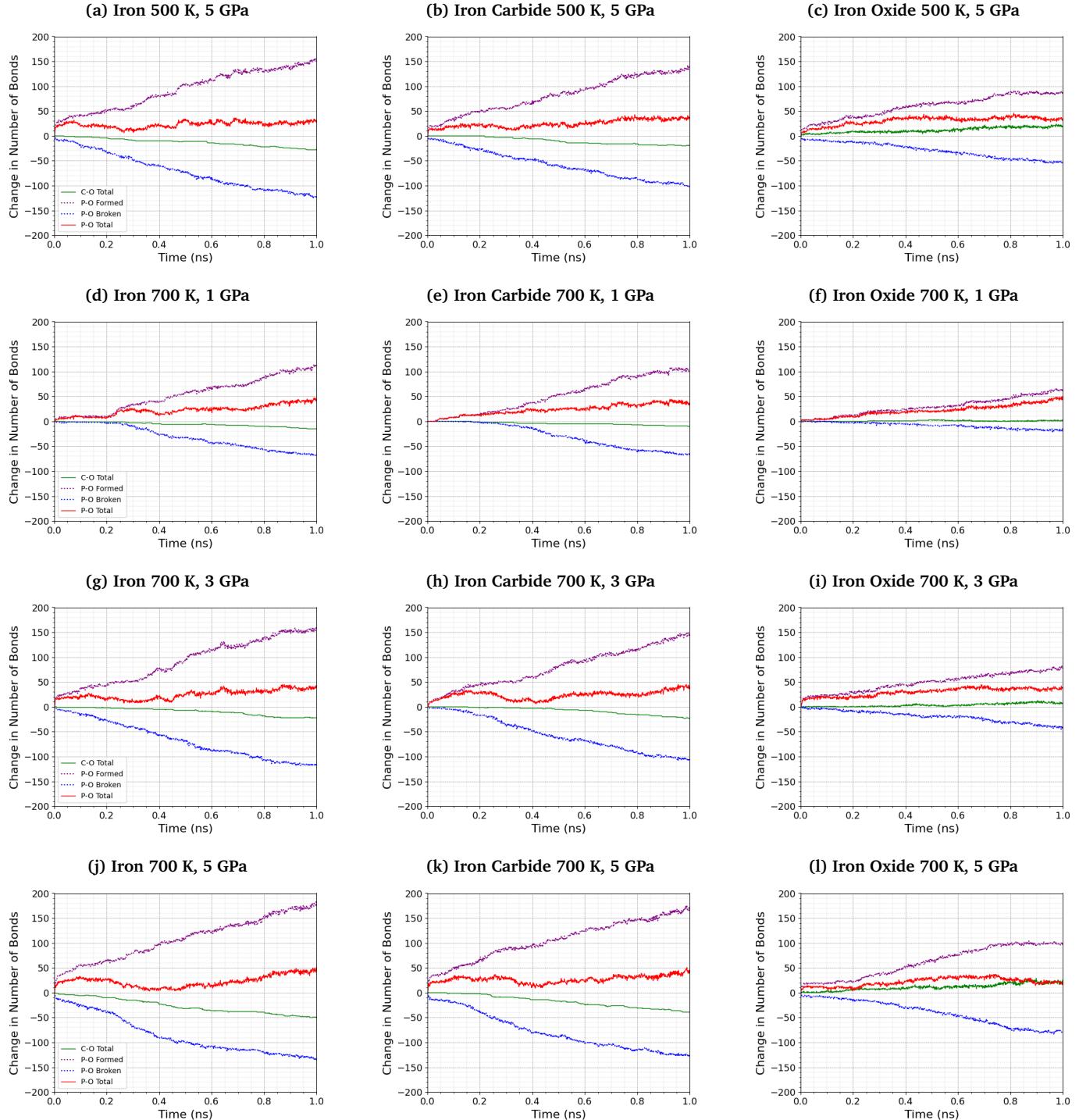
**Table S3** Derjaguin offset for TCP between  $\alpha$ -Fe(110),  $\text{Fe}_3\text{C}(010)$ , and  $\text{Fe}_3\text{O}_4(001)$  surfaces at different temperatures.



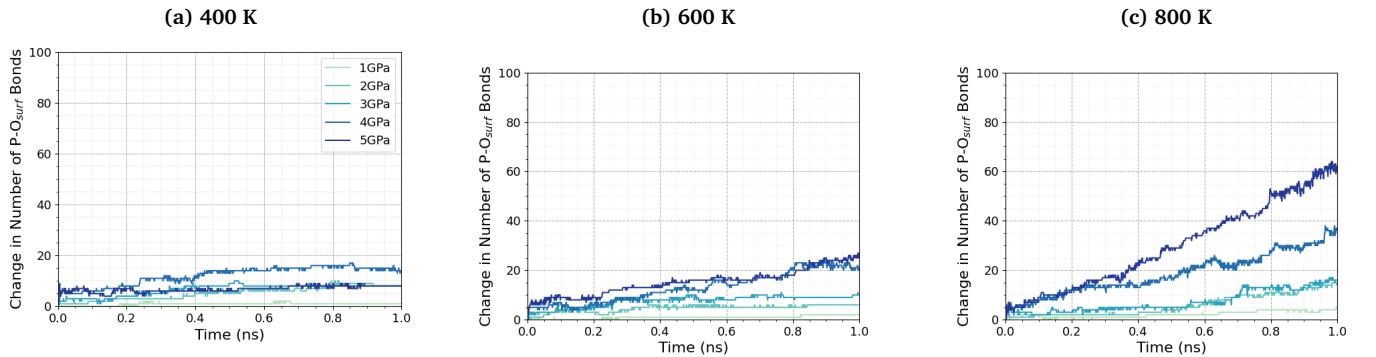
**Figure S2** Variation in TCP dissociation rates for  $\alpha\text{-Fe}(110)$ ,  $\text{Fe}_3\text{C}(010)$ , and  $\text{Fe}_3\text{O}_4(001)$  under various conditions.



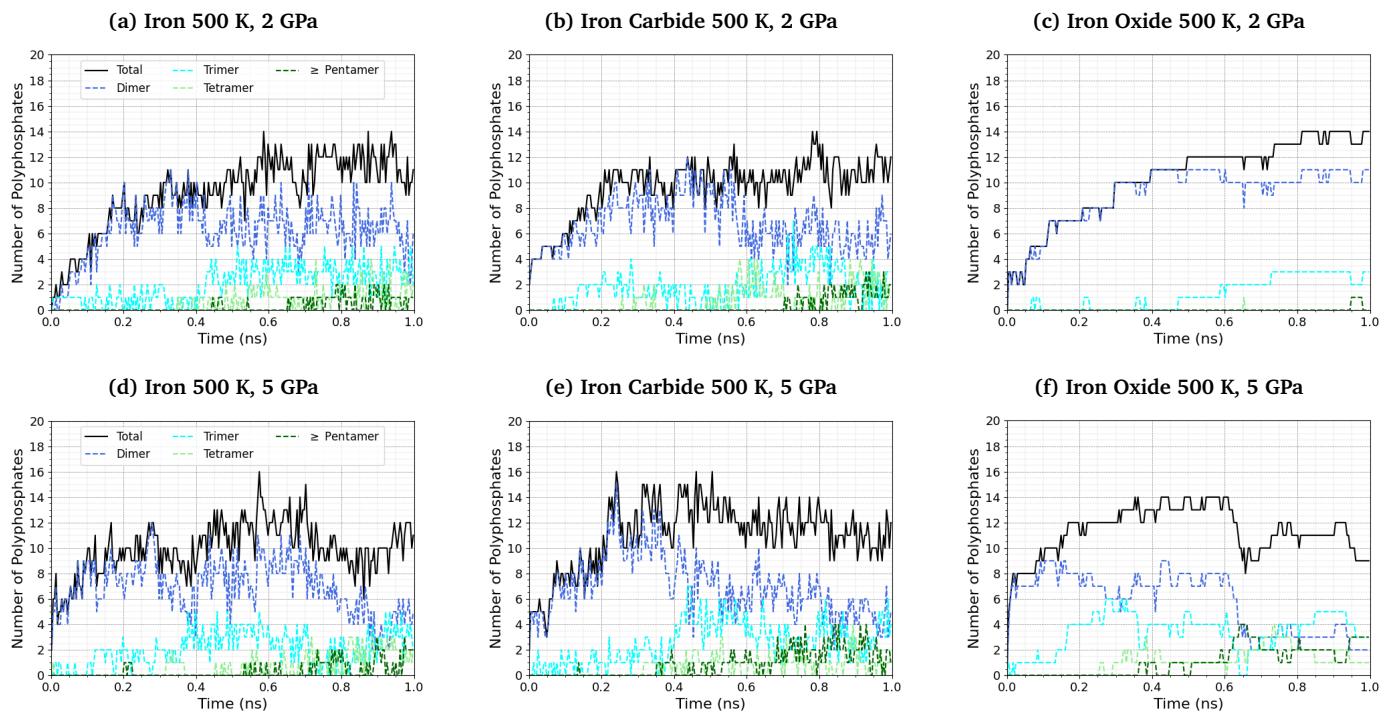
**Figure S3** Number density profiles for the Fe, C, O, and P atoms in the a)  $\alpha$ -Fe(110), b)  $\text{Fe}_3\text{C}(010)$ , and c)  $\text{Fe}_3\text{O}_4(001)$  systems at 500 K and 3 GPa



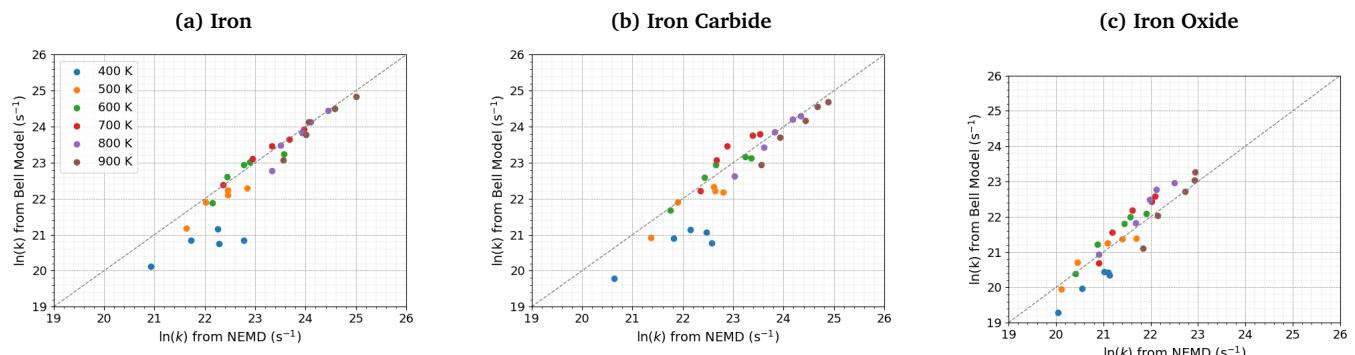
**Figure S4** Change in the number of P-O and C-O bonds compared to TCP, for  $\alpha$ -Fe(110),  $\text{Fe}_3\text{C}(010)$ , and  $\text{Fe}_3\text{O}_4(001)$  under various conditions. Solid lines represent total change, dashed lines show bond formation and dotted lines show bond cleavage.



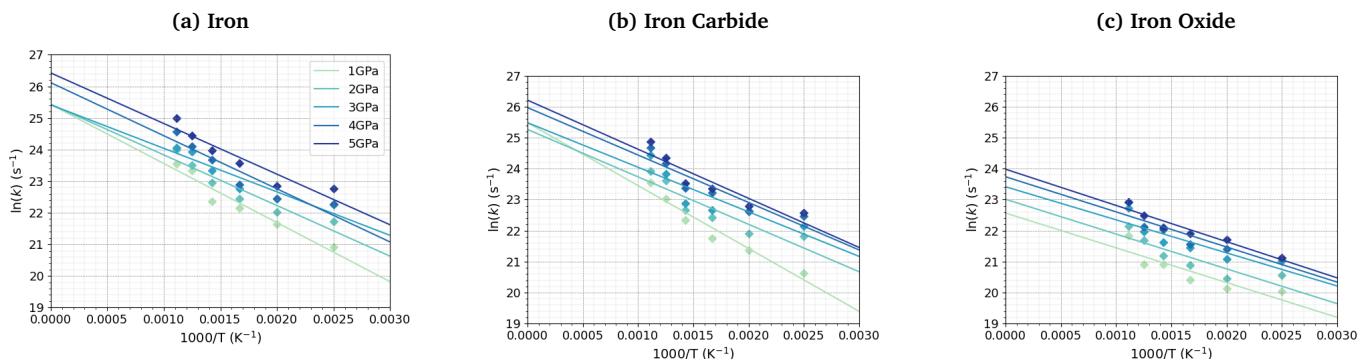
**Figure S5** Comparison of P-O<sub>surf</sub> bonds formed in the Fe<sub>3</sub>O<sub>4</sub>(100) system at a) 400 K, b) 600 K, and c) 800 K



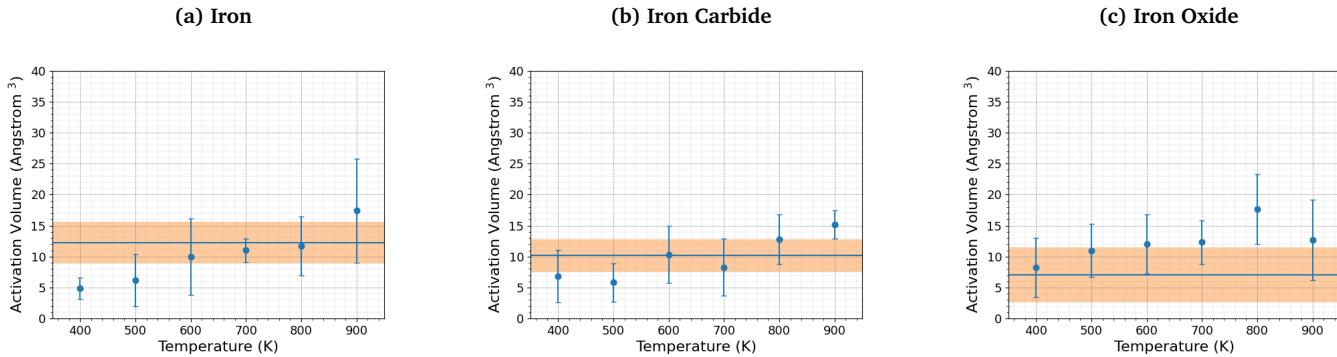
**Figure S6** Change in the number of polyphosphate chains with sliding time for a)  $\alpha$ -Fe(110), b) Fe<sub>3</sub>C(010) and c) Fe<sub>3</sub>O<sub>4</sub>(100) systems.



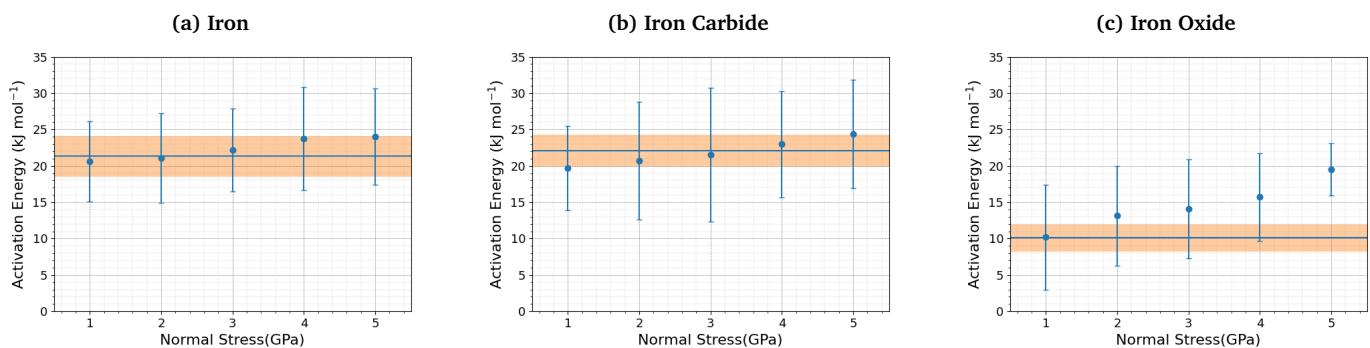
**Figure S7** Comparison of the logarithm of the reaction rates calculated from the NEMD simulations and those predicted using Equation 1. R<sup>2</sup> = 0.91 for  $\alpha$ -Fe(110), R<sup>2</sup> = 0.88 for Fe<sub>3</sub>C(010), and R<sup>2</sup> = 0.87 for Fe<sub>3</sub>O<sub>4</sub>(001).



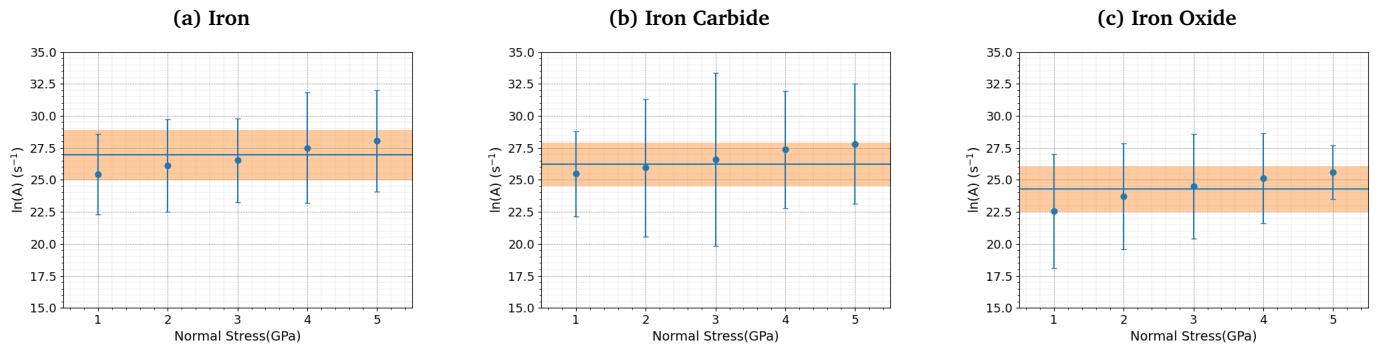
**Figure S8** Arrhenius plots of  $\ln(k)$  versus  $1000/T$  for a)  $\alpha\text{-Fe}(110)$ , b)  $\text{Fe}_3\text{C}(010)$ , and c)  $\text{Fe}_3\text{O}_4(001)$  obtained at different pressures. From Equation 1, the intercept of the lines gives  $A$  and gradient gives  $E_a$ .



**Figure S9** Temperature dependence of  $\Delta V^*$  for a)  $\alpha\text{-Fe}(110)$ , b)  $\text{Fe}_3\text{C}(010)$ , and c)  $\text{Fe}_3\text{O}_4(001)$ . Blue points are from the 2D fits, orange are from the 3D fits. Vertical bars indicate 95 % confidence intervals.



**Figure S10** Pressure dependence of  $E_a$  for a)  $\alpha\text{-Fe}(110)$ , b)  $\text{Fe}_3\text{C}(010)$ , and c)  $\text{Fe}_3\text{O}_4(001)$ . Blue points are from the 2D fits, orange are from the 3D fits. Vertical bars indicate 95 % confidence intervals.



**Figure S11** Pressure dependence of  $\ln(A)$  for a)  $\alpha\text{-Fe}(110)$ , b)  $\text{Fe}_3\text{C}(010)$ , and c)  $\text{Fe}_3\text{O}_4(001)$ . Blue points are from the 2D fits, orange are from the 3D fits. Vertical bars indicate 95 % confidence intervals.

	$\Delta V^*$ ( $\text{\AA}^3$ )					
	400 K	500 K	600 K	700 K	800 K	900 K
Iron	$4.9 \pm 1.7$	$6.2 \pm 4.2$	$10.0 \pm 6.2$	$11.0 \pm 1.9$	$11.7 \pm 4.7$	$17.4 \pm 8.4$
Iron Carbide	$6.8 \pm 4.2$	$5.5 \pm 3.1$	$10.3 \pm 4.6$	$8.3 \pm 4.6$	$12.8 \pm 4.0$	$15.2 \pm 2.3$
Iron Oxide	$8.3 \pm 4.5$	$10.9 \pm 4.3$	$12.0 \pm 4.8$	$12.6 \pm 3.8$	$17.7 \pm 5.6$	$12.8 \pm 6.6$

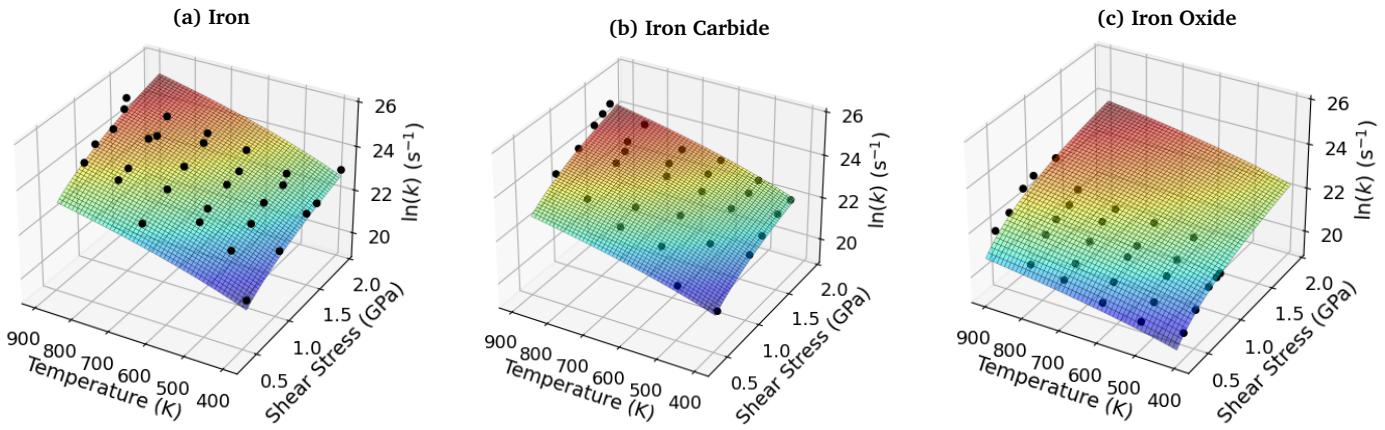
**Table S4**  $\Delta V^*$  for TCP between  $\alpha\text{-Fe}(110)$ ,  $\text{Fe}_3\text{C}(010)$ , and  $\text{Fe}_3\text{O}_4(001)$  surfaces.

	$E_a$ (kJ mol $^{-1}$ )				
	1 GPa	2 GPa	3 GPa	4 GPa	5 GPa
Iron	$20.6 \pm 5.5$	$21.1 \pm 6.2$	$22.2 \pm 5.7$	$23.7 \pm 7.1$	$24.0 \pm 6.6$
Iron Carbide	$19.7 \pm 5.8$	$20.7 \pm 8.1$	$21.5 \pm 9.2$	$23.0 \pm 7.3$	$24.3 \pm 7.4$
Iron Oxide	$10.4 \pm 7.2$	$13.2 \pm 6.9$	$14.1 \pm 6.8$	$15.8 \pm 6.1$	$19.4 \pm 3.6$

**Table S5**  $E_a$  for TCP between  $\alpha\text{-Fe}(110)$ ,  $\text{Fe}_3\text{C}(010)$ , and  $\text{Fe}_3\text{O}_4(001)$  surfaces.

	$\ln(A)$ (s $^{-1}$ )				
	1 GPa	2 GPa	3 GPa	4 GPa	5 GPa
Iron	$25.4 \pm 3.1$	$26.1 \pm 3.6$	$26.5 \pm 3.3$	$27.5 \pm 4.3$	$28.0 \pm 3.9$
Iron Carbide	$25.4 \pm 3.3$	$26.0 \pm 5.4$	$26.6 \pm 6.7$	$27.4 \pm 4.6$	$27.8 \pm 4.7$
Iron Oxide	$22.6 \pm 4.5$	$23.7 \pm 4.1$	$24.5 \pm 4.1$	$25.1 \pm 3.5$	$25.6 \pm 2.1$

**Table S6**  $\ln(A)$  for TCP between  $\alpha\text{-Fe}(110)$ ,  $\text{Fe}_3\text{C}(010)$ , and  $\text{Fe}_3\text{O}_4(001)$  surfaces.



**Figure S12** 3D fits of TCP decomposition  $\ln(k)$  as a function of temperature and shear stress for a)  $\alpha\text{-Fe}(110)$ , b)  $\text{Fe}_3\text{C}(010)$  and c)  $\text{Fe}_3\text{O}_4(100)$  surfaces.