## Supplementary information: The top-down design of high-performance V-based MBene anode for the Li/Naion batteries

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Fig. S1 The binary variable-composition search for V-Pb-B system with respect to V2B3 and A phases.



**Fig. S2** Crystal structures of layered  $V_xB_z$  compounds ( $V_5B_6$ ,  $V_2B_3$ ,  $VB_2$ ,  $V_3B_4$ , and VB).  $BV_6$  unit is framed in the red dotted box.

Since hex-V<sub>4</sub>PB<sub>6</sub> has thicker  $M_xB_z$  layers, its out-of-plane lattice constant c (12.244 Å) is larger than

that of Ti<sub>2</sub>InB<sub>2</sub> (7.917 Å) and Hf<sub>3</sub>InB<sub>4</sub> (11.868 Å), while its in-plane parameter *a* (3.015 Å) keeps the same level with them (3.077 Å and 3.182 Å), as shown in Table S1. Crossing bands at the Fermi energy ( $E_f$ ) for hex-V<sub>4</sub>PB<sub>6</sub> shown in Fig. S4a represent its metallic feature. The hybridization of V\_3*d* and P\_3*p* orbitals located near -1.5 eV and the hybridization of V\_3*d* and B\_2*p* orbitals observed in the area lower than -3.4 eV in the density of states (DOS) part of Fig. S4a indicate the V-P bond and stronger V-B bonds. The ELF results in Fig. S4b reveal that there is obvious electron accumulation around P atoms, and significant sharing electron density between adjacent boron atoms means the covalent B-B bonds inside the boron sheet.

Phases	Space	Energy	Lattice constants (Å)			Atom sites			
	group	(eV/atom)	а	b	С	Atom	Position	Coordinate	
V <sub>4</sub> PB <sub>6</sub>	Р <sup>д</sup> т2	-8.165	3.015	3.015	12.244	V	2h	(1/3, 2/3, 0.134)	
								(1/3, 2/3, 0.378)	
						Р	1 <i>a</i>	(0, 0, 0)	
						В	2 <i>i</i>	(2/3, 1/3, 0.253)	
							2g	(0, 0, 0.253)	
							1b	(0, 0, 0.5)	
							lf	(2/3, 1/3, 0.5)	
V <sub>4</sub> PB <sub>6</sub>	Cmmm	-8.041	2.988	20.956	3.039	V	4 <i>i</i>	(0.5, 0.431, 0)	
								(0.5, 0.188, 0)	
								(0, 0.069, 0)	
								(0, 0.312, 0)	
						Р	2 <i>c</i>	(0.5, 0, 0.5)	
						В	4 <i>j</i>	(0, 0.396, 0.5)	
								(0, 0.146, 0.5)	
								(0, 0.229, 0.5)	
								(0.5, 0.104, 0.5)	
								(0.5, 0.354, 0.5)	
								(0.5, 0.271, 0.5)	

Table S1 Crystallographic structure information of hex- $V_4PB_6(P\overline{6}m2, No. 187)$  and orth- $V_4PB_6(Cmmm, No. 65)$ .



Fig. S3 The phonon dispersion curve and crystal structure for orth-V<sub>4</sub>PB<sub>6</sub>(Cmmm).

Phases	V <sub>2</sub> PB <sub>2</sub>	V <sub>3</sub> PB <sub>4</sub>	V <sub>4</sub> PB <sub>6</sub>	V <sub>2</sub> B <sub>3</sub>	Cr <sub>4</sub> AlB <sub>6</sub>	V <sub>4</sub> AlC <sub>3</sub>
$C_{11}$	472	545	583	502	455	456
$C_{12}$	108	109	112	146	136	116
$C_{13}$	140	139	137	138	138	120
C <sub>22</sub>	-	-	-	647	572	-
$C_{23}$	-	-	-	109	139	-
$C_{33}$	459	474	478	663	516	385
$C_{44}$	226	232	230	251	191	173
$C_{55}$	-	-	-	230	154	-
$C_{66}$	182	232	235	258	192	170
В	242	260	268	287	262	223
G	193	219	222	241	182	165
Ε	457	513	521	564	443	398
v	0.19	0.17	0.18	0.17	0.22	0.20
$H^{Chen}_{V}$	30.22	35.28	34.74	37.16	24.27	25.01
$H_{V}^{Tian}$	29.41	34.36	33.95	36.48	24.08	24.39

**Table S2** Calculated bulk modulus *B* (GPa), shear modulus *G* (GPa), Young's modulus *E* (GPa), Poisson's ratio *v* and Vickers hardness  ${}^{H_{V}^{Chen}}$ ,  ${}^{H_{V}^{Tian}}$  (GPa) of hex-V<sub>2</sub>PB<sub>2</sub> ( $P\overline{6}m2$ ), hex-V<sub>3</sub>PB<sub>4</sub> ( $P\overline{6}m2$ ), hex-V<sub>4</sub>PB<sub>6</sub> ( $P\overline{6}m2$ ), binary compound orth-V<sub>2</sub>B<sub>3</sub> (*Cmcm*), hex-V<sub>4</sub>AlC<sub>3</sub> ( $P\overline{6}_{3}/mmc$ ) and orth-Cr<sub>4</sub>AlB<sub>6</sub> (*Cmmm*).



**Fig. S4** (a) The projected band structure and density of states (DOS) for hex-V<sub>4</sub>PB<sub>6</sub> ( $P^{\overline{6}}m2$ ). The Fermi energy ( $E_{f}$ ) is indicated by solid line at 0 eV. In the picture, DOS are measured in the unit of states/eV. (b) The three-dimensional electron localization function (ELF) with a value of isosurface 0.7 for hex-V<sub>4</sub>PB<sub>6</sub> ( $P^{\overline{6}}m2$ ).



**Fig. S5** (a) Calculated elastic constants for hex-V<sub>2</sub>PB<sub>2</sub> ( $P\overline{6}m2$ ), hex-V<sub>3</sub>PB<sub>4</sub> ( $P\overline{6}m2$ ), hex-V<sub>4</sub>PB<sub>6</sub> ( $P\overline{6}m2$ ). (b) Calculated elastic modulus and Poisson's ratio for hex-V<sub>2</sub>PB<sub>2</sub> ( $P\overline{6}m2$ ), hex-V<sub>3</sub>PB<sub>4</sub> ( $P\overline{6}m2$ ), hex-V<sub>4</sub>PB<sub>6</sub> ( $P\overline{6}m2$ ), binary compound orth-V<sub>2</sub>B<sub>3</sub> (*Cmcm*), hex-V<sub>4</sub>AlC<sub>3</sub> ( $P6_3/mmc$ ) and orth-Cr<sub>4</sub>AlB<sub>6</sub> (*Cmmm*).

For the M-A, M-B and A-A bonds in MAB phases, their interlayer distance at ground state is  $d_0$ . As the distance *d* increases, the energy of compounds becomes higher, and until *d* is large enough, the energy no longer changes. Taking M-A bond for example, the relative energy can be obtained from the following equation:

$$E_{relative} = -(E_{MAB} - E_{MB-A})/S$$
(S1)

where  $E_{MAB}$  is the total energy of ground-state MAB phase, and  $E_{MB-A}$  is the total energy of MAB phase with an interlayer distance *d* between MB and A layers. *S* represents the in-plane surface area of the unit cell of the MAB phase. The bonding energy of the M-A bond is the convergent value of relative energy.



**Fig. S6** Calculated energy varying with d- $d_0$  for hex-V<sub>4</sub>PB<sub>6</sub>( $P\overline{6}m2$ ), hex-Ti<sub>2</sub>InB<sub>2</sub>( $P\overline{6}m2$ ), hex-Hf<sub>2</sub>InB<sub>2</sub>( $P\overline{6}m2$ ), orth-Cr<sub>2</sub>AlB<sub>2</sub>(*Cmmm*), and orth-MoAlB (*Cmcm*).



**Fig. S7** The electron localization function (ELF) contour plots along the (110) plane for hex-V<sub>4</sub>PB<sub>6</sub> ( $P\overline{6}m2$ ) under [001] tensile strains of 1%, 11%, 20% and 23% with an isosurface value 0.85.



Fig. S8 Diagram of five magnetic configurations using  $V_4B_6$  as an example: (a) ferromagnetic (FM), and antiferromagnetic (AFM) including (b) AFM1, (c) AFM2, (d) AFM3, (e) AFM4.



Fig. S9 Calculated phonon dispersion spectra of 2D  $V_4B_6$  and  $V_4B_6T_2$  (T = Cl, O, and S).



**Fig. S10** Top and side views of equilibrium structures for 2D  $V_4B_6$  and  $V_4B_6T_2$  (T = Cl, O, and S) after undergoing a 5 ps AIMD simulation at 900 K.



Fig. S11 Calculated band structures using PBE and PBE+U with U = 2 eV, 3 eV, 4 eV for (a)  $V_4B_6$ , (b)  $V_4B_6Cl_2$ , (c)  $V_4B_6O_2$ , and (d)  $V_4B_6S_2$ .

**Table S3** The space group, lattice constants a, and elastic constants  $(C_{ij})$  of 2D V<sub>4</sub>B<sub>6</sub>, V<sub>4</sub>B<sub>6</sub>Cl<sub>2</sub>, V<sub>4</sub>B<sub>6</sub>O<sub>2</sub>, and V<sub>4</sub>B<sub>6</sub>S<sub>2</sub>.

	$V_4B_6$	$V_4B_6Cl_2$	$V_4B_6O_2$	$V_4B_6S_2$
Space group	P6/mmm	$P\overline{6}_{m2}$	$P\bar{3}m1$	$P\overline{6}_{m2}$
<i>a</i> (Å)	2.976	3.030	2.936	3.011
$C_{11}$ (N/m)	640	650	733	721
$C_{12}$ (N/m)	82	96	128	116
$C_{66} ({ m N/m})$	279	277	303	302



**Fig. S12** The average adsorption energy ( $^{E}ave$ ) with increasing adsorbed Li/Na layers on 2D V<sub>4</sub>B<sub>6</sub>, V<sub>4</sub>B<sub>6</sub>Cl<sub>2</sub>, V<sub>4</sub>B<sub>6</sub>O<sub>2</sub> and V<sub>4</sub>B<sub>6</sub>S<sub>2</sub> monolayers. The inset is the side view of structures of Li<sub>2</sub>V<sub>4</sub>B<sub>6</sub>, Li<sub>0.25</sub>V<sub>4</sub>B<sub>6</sub>Cl<sub>2</sub>, Li<sub>2</sub>V<sub>4</sub>B<sub>6</sub>O<sub>2</sub>, Na<sub>2</sub>V<sub>4</sub>B<sub>6</sub>, Na<sub>2</sub>V<sub>4</sub>B<sub>6</sub>O<sub>2</sub> and Na<sub>0.25</sub>V<sub>4</sub>B<sub>6</sub>S<sub>2</sub> with the largest number of Li/Na layer.



Fig. S13 The elementary density of states (DOS) of the  $V_4B_6S_2$ ,  $Li_{0.25}V_4B_6S_2$ , and  $Li_4V_4B_6S_2$ .



Fig. S14 Calculated phonon dispersion spectra of 2D  $h-M_4B_6$  (M = Sc, Ti, Zr, Nb, Hf, and Ta).



Fig. S15 Top and side views of equilibrium structures for 2D  $h-M_4B_6$  (M = Sc, Ti, Zr, Nb, Hf, and Ta) after undergoing a 5 ps AIMD simulation at 900 K.