

Supplementary Information

**Anodic voltage performance of conducting polymer-functionalized boron
nitride nanosheets: A DFT assessment**

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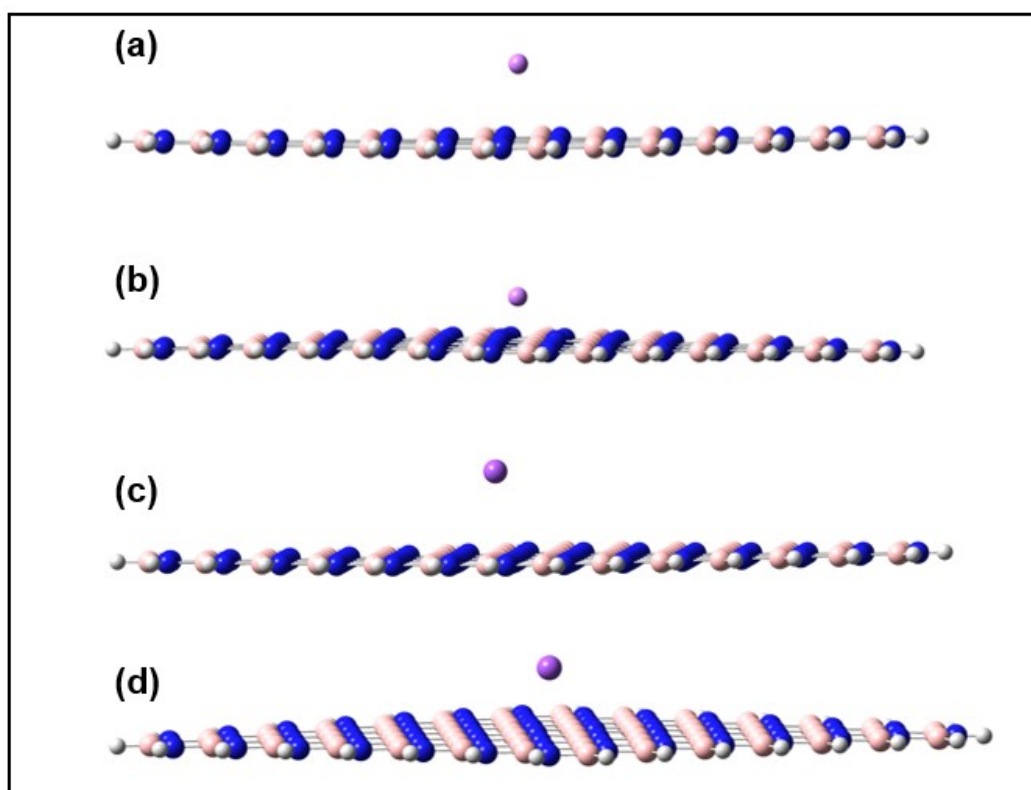


Figure S1. Optimized cross-sectional structures of (a) Li@BNNS (b) Li⁺@BNNS, (c) Na@BNNS, and (d) Na⁺@BNNS. The pristine boron nitride nanosheet (BNNS) comprises 63 atoms of boron and nitrogen each with dangling atoms at the structural boundaries saturated with hydrogen atoms to minimize the boundary effects [37].

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Table S1. Adsorption distances of metals and metal-ions on BNNS complexes.

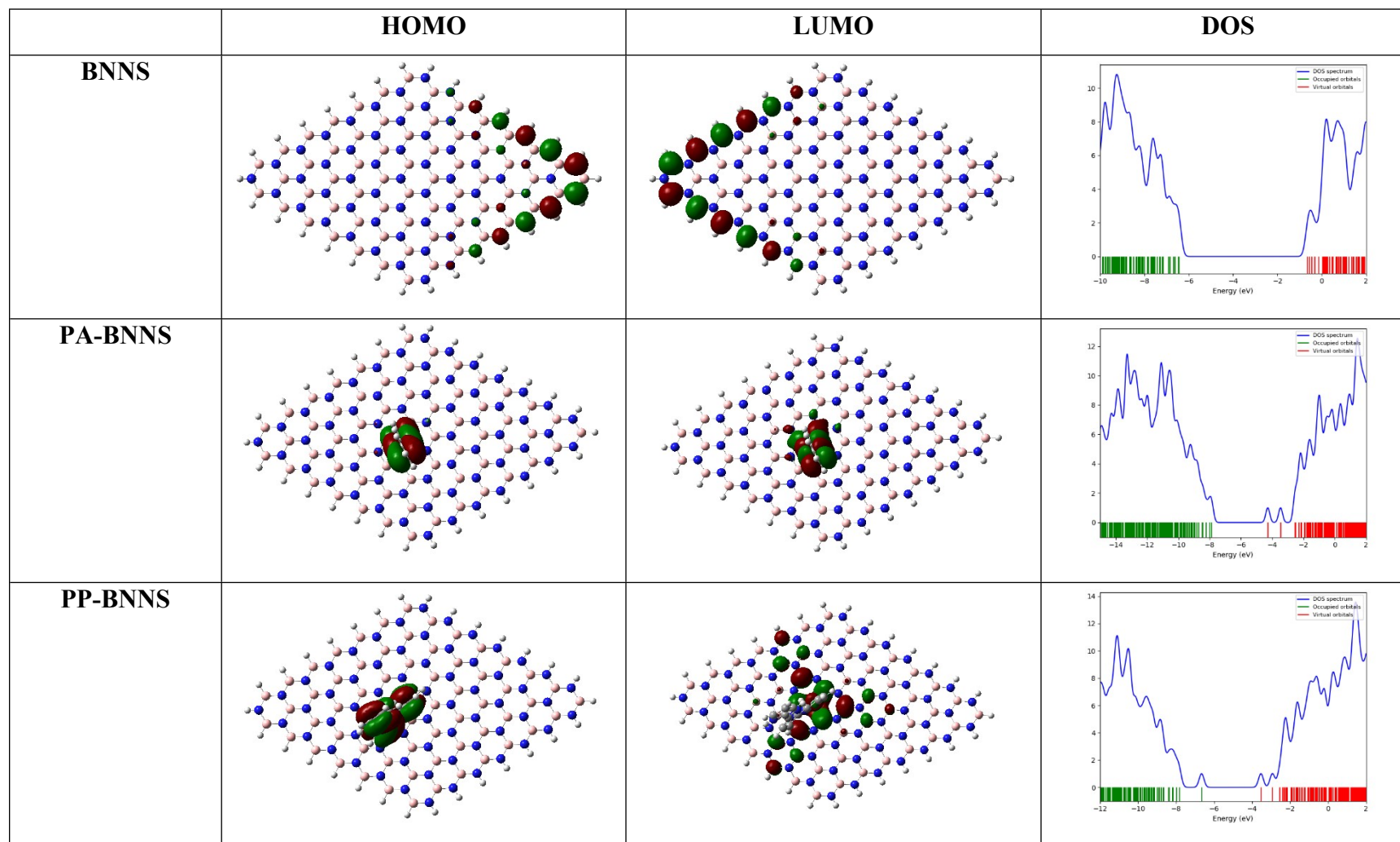
BNNS	Li@BNNS	Li⁺@BNNS	Na@BNNS	Na⁺@BNNS
B-N Bond Distance (Å)				
1.45	1.45	1.46	1.45	1.46
M/Mⁿ⁺-N Bond Distance (Å)				
0.00	3.31	2.23	3.66	2.73
M/Mⁿ⁺-B Bond Distance (Å)				
0.00	3.33	2.33	3.91	2.80

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Table S2. Adsorption distances of conducting polymers, metals, and metal-ions on BNNS and CP-BNNS complexes.

B-N Bond Distance (Å)											
PA-BNNS	PA-BNNS	Li@PA-BNNS	Li⁺@PA-BNNS	Na@PA-BNNS	Na⁺@PA-BNNS	PP-BNNS	PP-BNNS	Li@PP-BNNS	Li⁺@PP-BNNS	Na@PP-BNNS	Na⁺@PP-BNNS
		1.55	1.48	1.48	1.48		1.48		1.56	1.45 (float)	1.48
PPPh-BNNS	PPPh-BNNS	Li@PPPh-BNNS	Li⁺@PPPh-BNNS	Na@PPPh-BNNS	Na⁺@PPPh-BNNS	PT-BNNS	PT-BNNS	Li@PT-BNNS	Li⁺@PT-BNNS	Na@PT-BNNS	Na⁺@PT-BNNS
		1.56	1.46(float)	1.48	1.45(float)		1.48		1.56	1.46(float)	1.49
M/Mⁿ⁺-N Bond Distance (Å)											
PA-BNNS	PA-BNNS	Li@PA-BNNS	Li⁺@PA-BNNS	Na@PA-BNNS	Na⁺@PA-BNNS	PP-BNNS	PP-BNNS	Li@PP-BNNS	Li⁺@PP-BNNS	Na@PP-BNNS	Na⁺@PP-BNNS
		-	2.97	2.36	2.85		2.71		-	2.83	2.60
PPPh-BNNS	PPPh-BNNS	Li@PPPh-BNNS	Li⁺@PPPh-BNNS	Na@PPPh-BNNS	Na⁺@PPPh-BNNS	PT-BNNS	PT-BNNS	Li@PT-BNNS	Li⁺@PT-BNNS	Na@PT-BNNS	Na⁺@PT-BNNS
		-	2.32	2.28	2.74		2.92		-	3.15	2.22
M/Mⁿ⁺-B Bond Distance (Å)											
PA-BNNS	PA-BNNS	Li@PA-BNNS	Li⁺@PA-BNNS	Na@PA-BNNS	Na⁺@PA-BNNS	PP-BNNS	PP-BNNS	Li@PP-BNNS	Li⁺@PP-BNNS	Na@PP-BNNS	Na⁺@PP-BNNS
		-	2.58	2.46	2.91		2.83		-	2.68	2.45
PPPh-BNNS	PPPh-BNNS	Li@PPPh-BNNS	Li⁺@PPPh-BNNS	Na@PPPh-BNNS	Na⁺@PPPh-BNNS	PT-BNNS	PT-BNNS	Li@PT-BNNS	Li⁺@PT-BNNS	Na@PT-BNNS	Na⁺@PT-BNNS
		-	2.40	2.39	2.67		3.02		-	2.69	2.68

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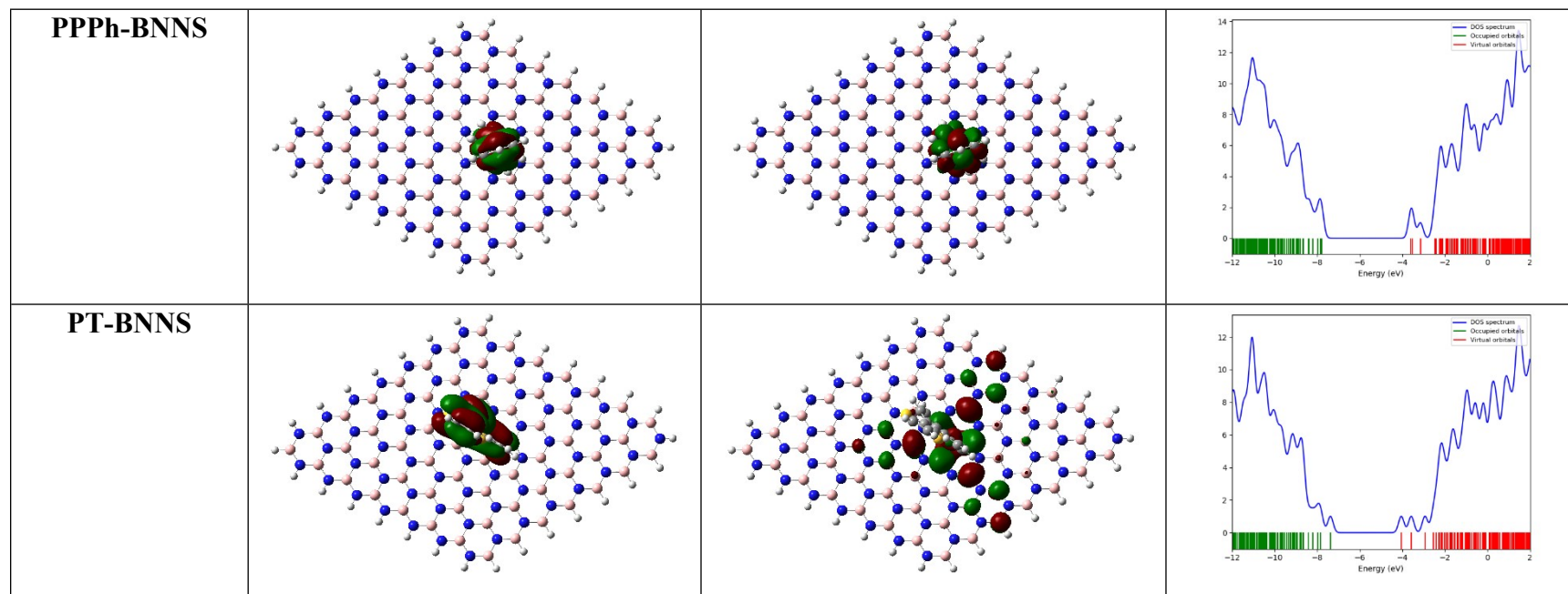


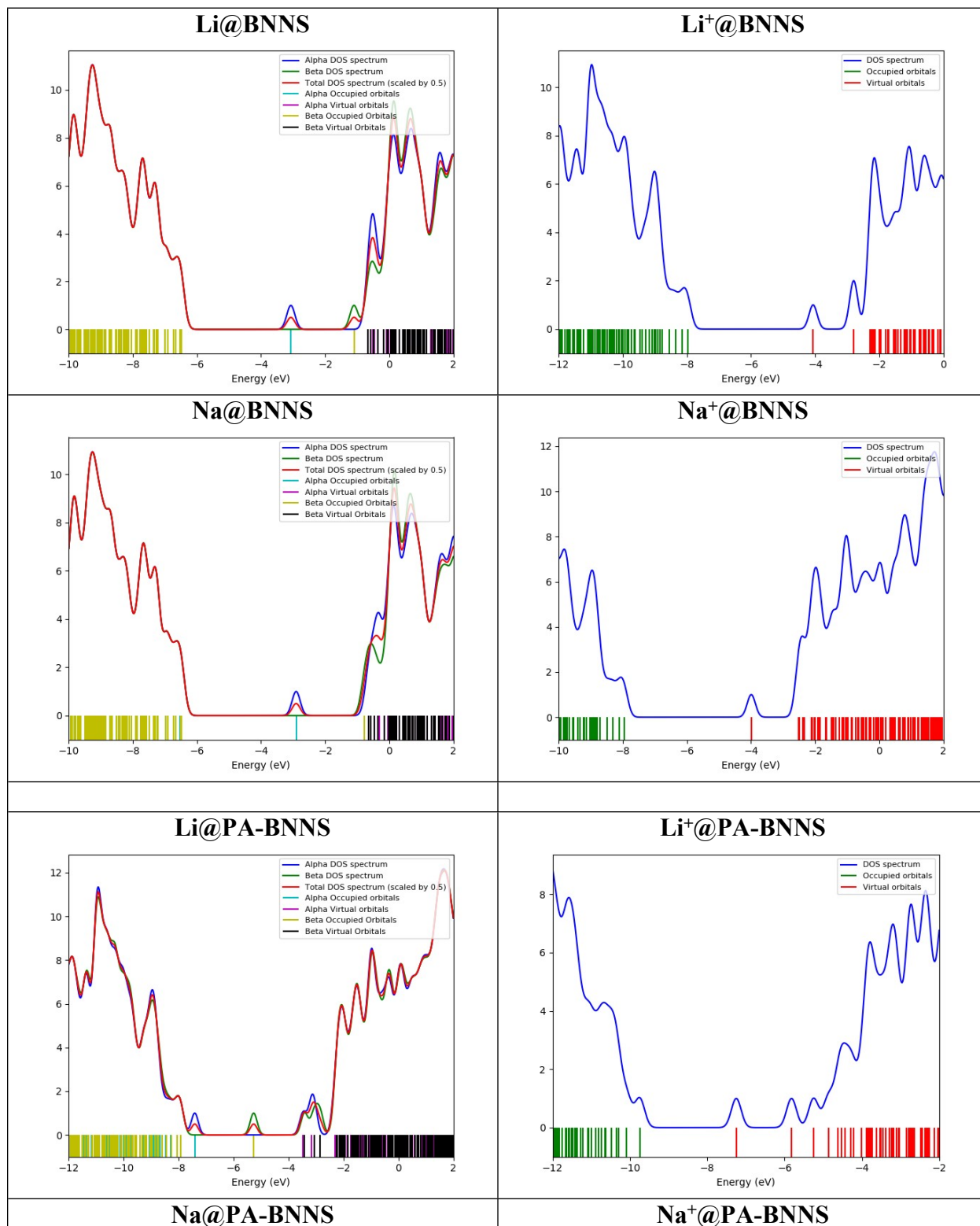
Figure S2. The HOMO and LUMO iso-surfaces of the conducting polymers adsorbed onto the BNNS (M/Mn+@BNNS).

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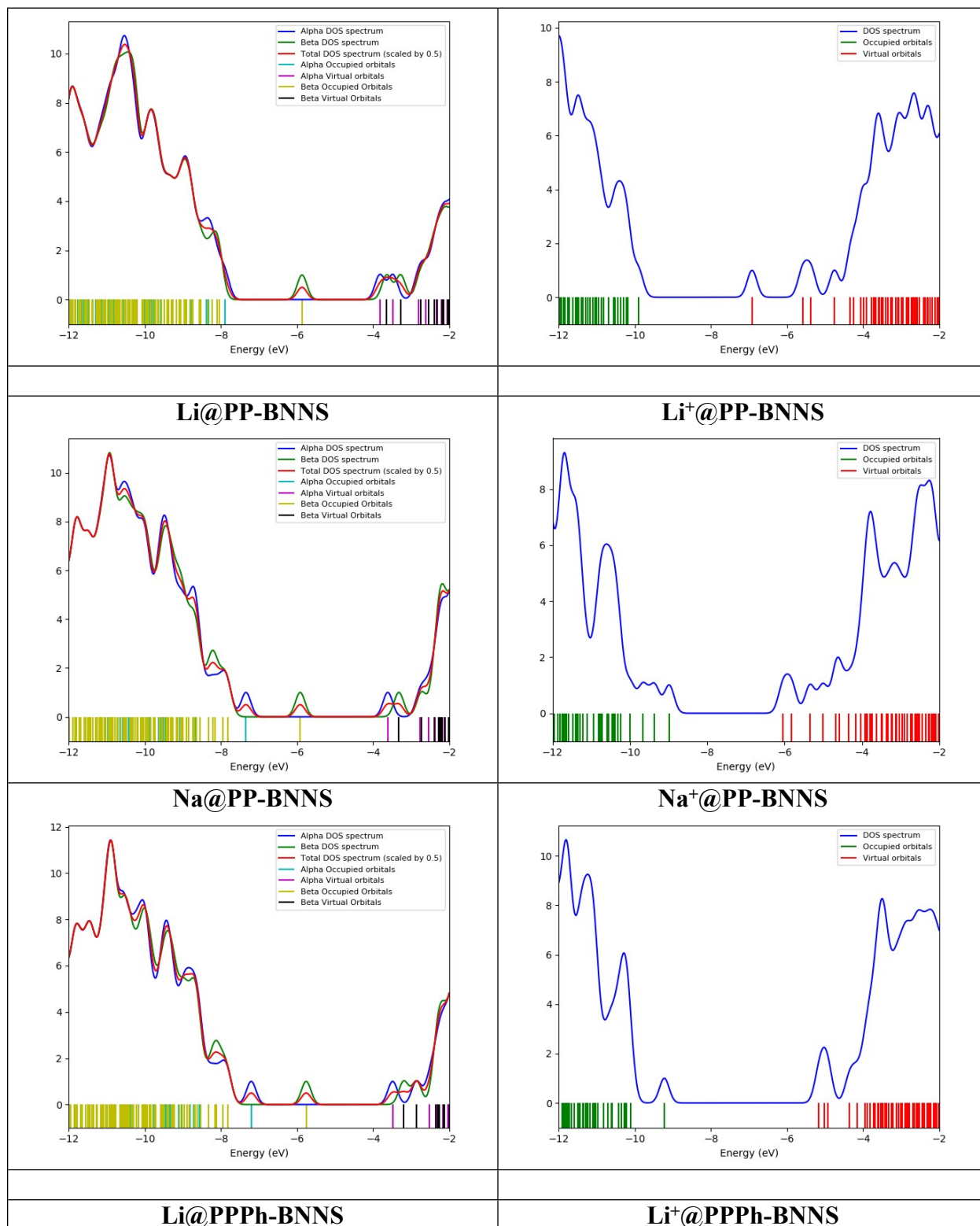
Table S3. HOMO, LUMO, band gap and adsorption energies (all in eV) of metal atoms and ions on BNNS and CP-BNNS complexes compared to bare and functionalized BNNS structures.

	HOMO	LUMO	E_g	E_d
BNNS	-6.45	-0.64	5.81	
Li@BNNS	-3.07	-0.66	2.41	-11.85
Li⁺@BNNS	-7.98	-4.07	3.91	-49.97
Na@BNNS	-2.91	-0.66	2.25	-7.50
Na⁺@BNNS	-7.96	-3.99	3.96	-32.35
PA-BNNS				
PA-BNNS	-7.86	-4.28	3.58	536.18
Li@PA-BNNS	-7.42	-3.47	3.94	-33.30
Li⁺@PA-BNNS	-9.75	-7.25	2.50	-6.97
Na@PA-BNNS	-7.90	-3.83	4.07	-23.16
Na⁺@PA-BNNS	-9.91	-6.92	2.99	7.84
PP-BNNS				
PP-BNNS	-6.66	-3.53	3.13	535.87
Li@PP-BNNS	-7.35	-3.62	3.73	-92.94
Li⁺@PP-BNNS	-8.99	-6.04	2.95	-8.46
Na@PP-BNNS	-7.21	-3.49	3.72	-81.89
Na⁺@PP-BNNS	-9.23	-5.17	4.06	-16.52
PPPh-BNNS				
PPPh-BNNS	-7.81	-3.60	4.21	533.27
Li@PPPh-BNNS	-8.05	-4.28	3.78	-73.10
Li⁺@PPPh-BNNS	-9.41	-6.18	3.23	-5.10
Na@PPPh-BNNS	-8.18	-4.12	4.06	-64.48
Na⁺@PPPh-BNNS	-9.36	-6.31	3.05	6.49
PT-BNNS				
PT-BNNS	-7.38	-4.05	3.33	539.47
Li@PT-BNNS	-7.90	-4.89	3.01	-94.64
Li⁺@PT-BNNS	-9.22	-7.03	2.19	-5.48
Na@PT-BNNS	-7.77	-4.72	3.05	-82.41
Na⁺@PT-BNNS	-10.07	-6.26	3.82	-9.62

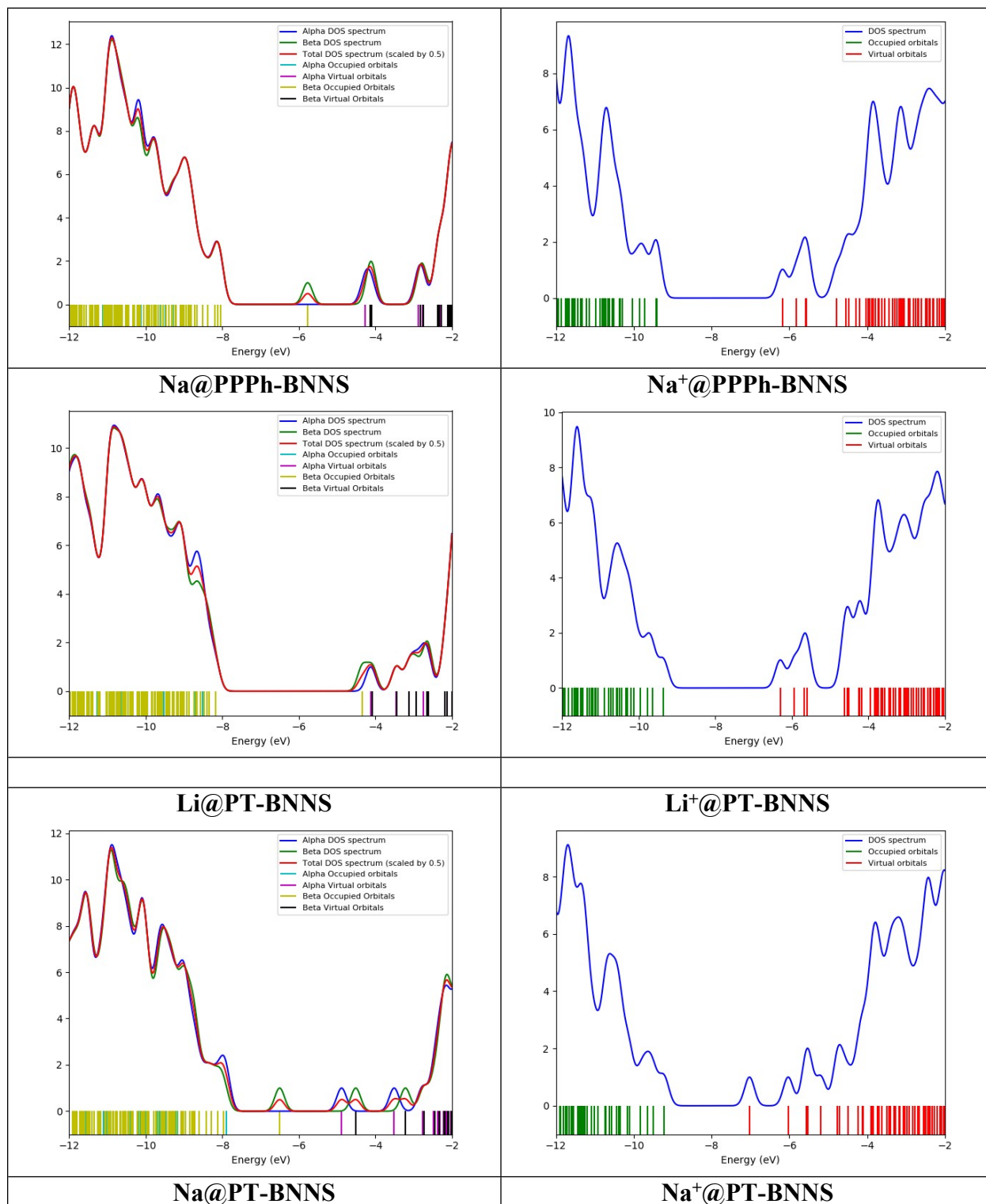
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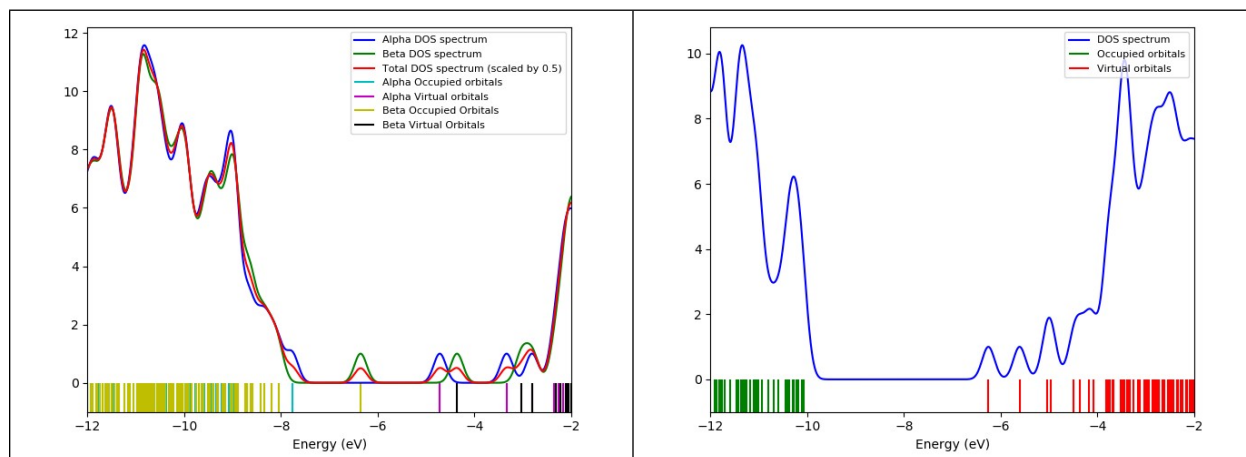
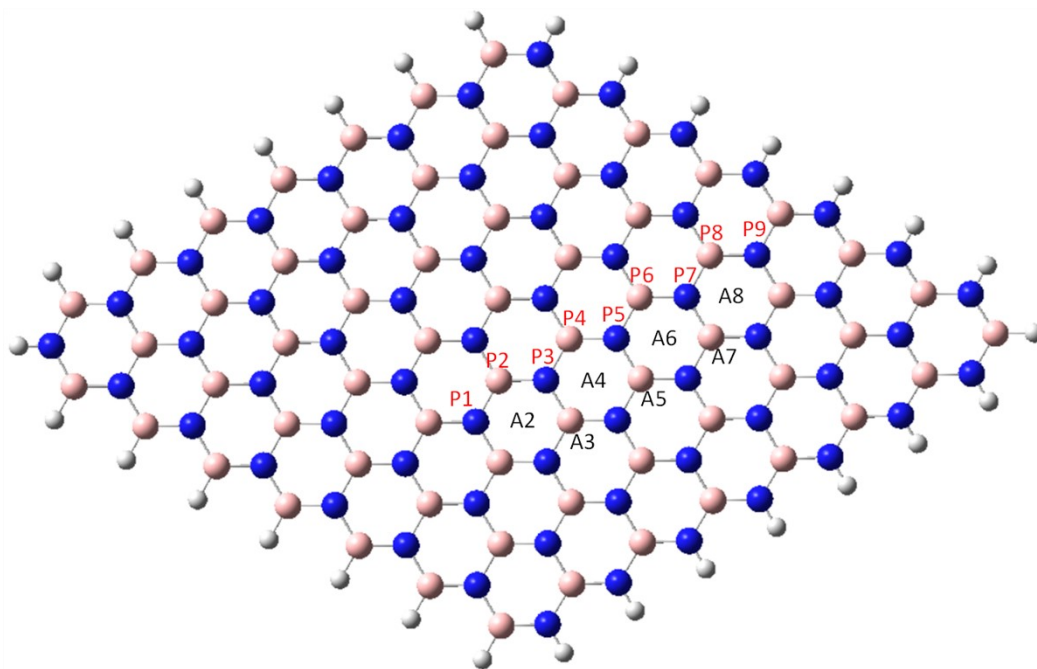


Figure S3. The density of States of the metal atoms and ions adsorbed onto the CP-BNNS.

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Path 1 (via B-N bonds): P1-P2-P3-P4-P5-P6-P7-P8-P9

Path 2 (via ring-B): P1-A2-A3-A4-A5-A6-A7-A8-P9

Path 3 (via ring-N): P1-A2-P3-A4-P5-A6-P7-A8-P9

Figure S4. The pathway of ion diffusion on the sheets

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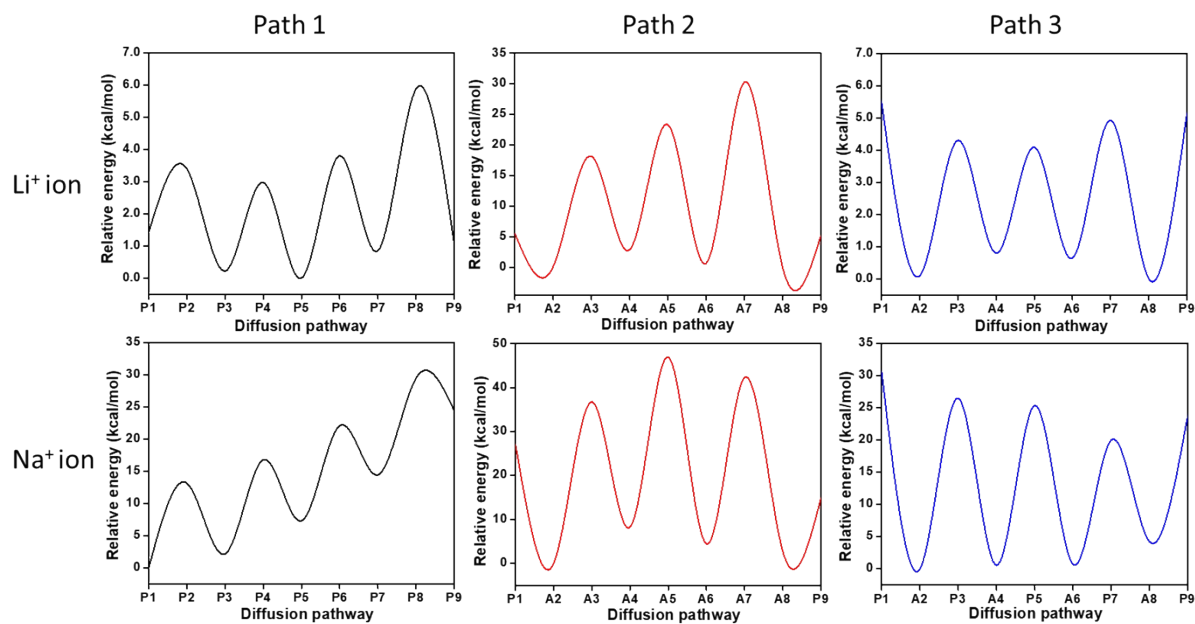


Figure S5. The diffusion profile of Li^+ and Na^+ ions via path 1, path 2 and path 3 on the pristine BNNS sheet

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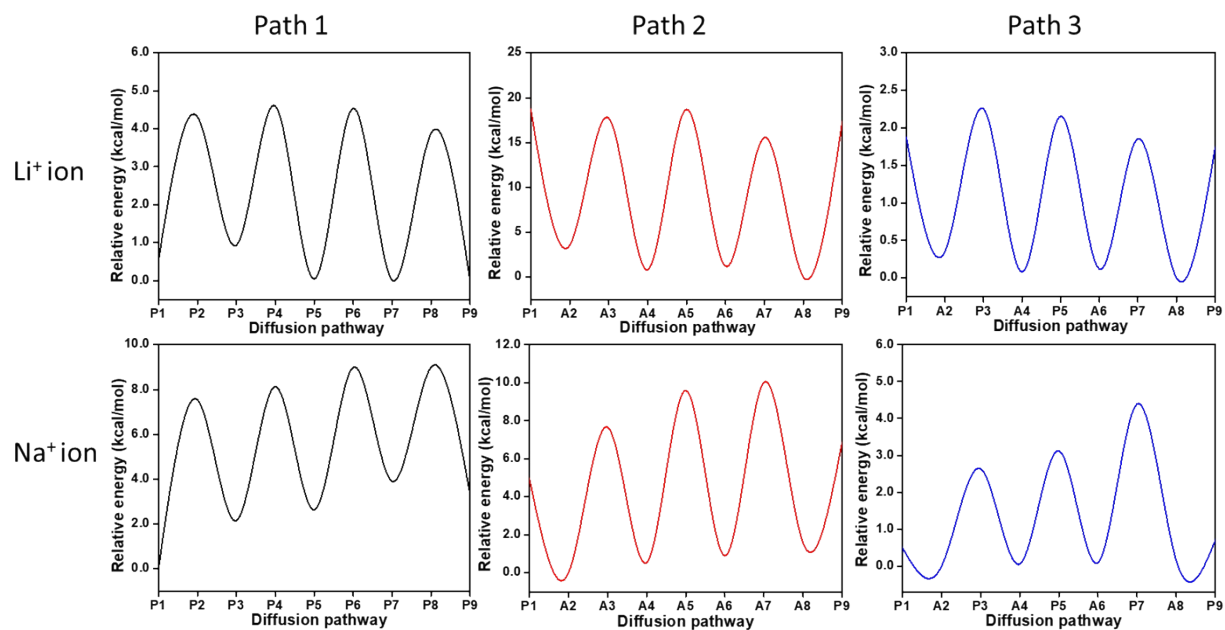


Figure S6. The diffusion profile of Li^+ and Na^+ ions via path 1, path 2 and path 3 on the PT functionalized BNNS sheet

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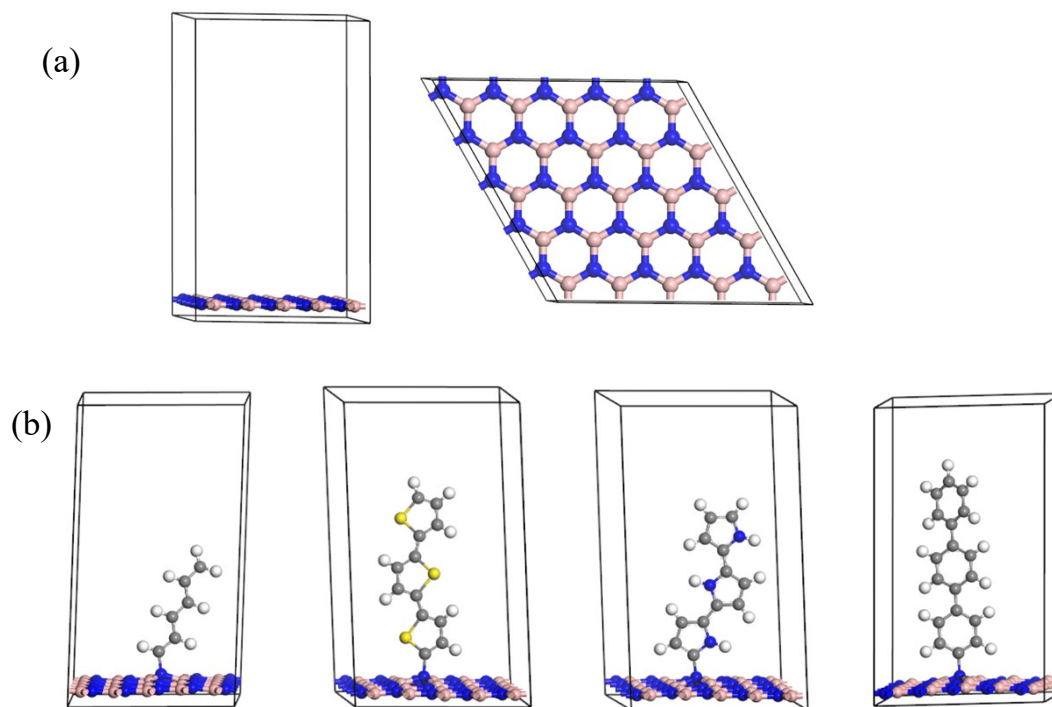


Figure S7 The geometrically optimized supercell with lattice dimensions of $12 \times 12 \times 20 \text{ \AA}$ of (a) BNNS, and the conducting polymer functionalized BNNS (b) from left to right: PA-BNNS, PT-BNNS, PP-BNNS and PPh-BNNS

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Table S4. The electronic properties of the pristine and the CP functionalized BNNS sheets

System	Cluster model			Periodic system		
	HOMO	LUMO	E_g	HOMO	LUMO	E_g
BNNS	-6.45	-0.64	5.81	-6.89	-1.02	5.87
PA-BNNS	-7.86	-4.28	3.58	-5.55	-3.14	2.41
Li@PA-BNNS	-7.42	-3.47	3.94	-5.75	-2.50	3.25
Li ⁺ @PA-BNNS	-9.75	-7.25	2.50	-5.36	-3.19	2.18
Na@PA-BNNS	-7.90	-3.83	4.07	-7.01	-3.28	3.73
Na ⁺ @PA-BNNS	-9.91	-6.92	2.99	-5.65	-3.25	2.39
PP-BNNS	-6.66	-3.53	3.13	-4.96	-2.69	2.27
Li@PP-BNNS	-7.35	-3.62	3.73	-4.88	-2.55	2.34
Li ⁺ @PP-BNNS	-8.99	-6.04	2.95	-5.03	-2.92	2.11
Na@PP-BNNS	-7.21	-3.49	3.72	-4.87	-2.15	2.99
Na ⁺ @PP-BNNS	-9.23	-5.17	4.06	-4.88	-1.11	3.77
PPPh-BNNS	-7.81	-3.60	4.21	-5.80	-2.41	3.38
Li@PPPh-BNNS	-8.05	-4.28	3.78	-5.64	-2.87	2.77
Li ⁺ @PPPh-BNNS	-9.41	-6.18	3.23	-5.58	-2.89	2.69
Na@PPPh-BNNS	-8.18	-4.12	4.06	-5.81	-2.96	2.85
Na ⁺ @PPPh-BNNS	-9.36	-6.31	3.05	-5.78	-3.20	2.58
PT-BNNS	-7.38	-4.05	3.33	-5.04	-2.56	2.48
Li@PT-BNNS	-7.90	-4.89	3.01	-5.14	-3.50	1.63
Li ⁺ @PT-BNNS	-9.22	-7.03	2.19	-4.92	-3.44	1.48
Na@PT-BNNS	-7.77	-4.72	3.05	-5.05	-2.87	2.18
Na ⁺ @PT-BNNS	-10.07	-6.26	3.82	-5.02	-1.76	3.27