Novel two-dimensional Janus TiSiGeN $_4$ monolayer with N vacancies

for efficient photocatalytic nitrogen reduction

Zhe Sun, ac Rongfeng Guan, b Huimin Li, ac Shilong Feng, ac Lin Ma, ac Qianqian Shen, *ac

Lixia Ling,^d Husheng Jia^{ace} and Jinbo Xue^{*ac}

^aKey Laboratory of Interface Science and Engineering in Advanced Materials (Taiyuan University

of Technology), Ministry of Education, Taiyuan 030024, P. R. China. E-mail address:

xuejinbo@tyut.edu.cn, shenqianqian@tyut.edu.cn

^bKey Laboratory for Advanced Technology in Environmental Protection of Jiangsu Province,

Yancheng Institute of Technology, Yancheng 224051, P. R. China

^cCollege of Materials Science and Engineering, Taiyuan University of Technology, Taiyuan 030024, P. R. China

^dCollege of Chemistry and Chemical Engineering, Taiyuan University of Technology, Taiyuan 030024, P. R. China

^eShanxi-Zheda Institute of Advanced Materials and Chemical Engineering, Taiyuan 030032, P. R. China



Fig. S1 Phonon dispersion curves for (a) $TiSi_2N_4$ and (b) $TiSiGeN_4$.



Fig. S2 Profile of electrostatic potential of (a) $TiSi_2N_4$ and (b) $TiSiGeN_4$. The different colors in the picture represent different electron densities, as shown in the scale on the right.



Fig. S3 Variations of the total energy versus time for AIMD simulations of Janus TiSiGeN₄ with different N vacancies. The AIMD runs under 500 K for 5 ps with a time step of 1 fs.



Fig. S4 Optical absorption spectra of TiSiGeN₄-p, TiSiGeN₄- V_N -t, TiSiGeN₄- V_N -m, and TiSiGeN₄- V_N -mt.



Fig. S5 Optimized structures of various intermediates along the reaction paths of the NRR proceeding on the (a) $TiSiGeN_4-V_N-t$ and (b) $TiSiGeN_4-V_N-mt$ surfaces through distal, alternating and enzymatic pathways.



Reaction coordinate

Fig. S6 Free energy diagrams for HER on TiSiGeN₄-V_N-t and TiSiGeN₄-V_N-mt.

Species	E _{ZPE} (eV)	<i>TS</i> (eV)	$E_{\rm ZPE} - TS$ (eV)
*NN	0.23	0.08	0.15
*NNH	0.55	0.09	0.46
*NNH ₂	0.84	0.11	0.73
*NNH ₃	1.01	0.13	0.88
*NH	0.41	0.03	0.38
*NH ₂	0.69	0.09	0.60
*NH ₃	0.96	0.12	0.84
*NHNH	0.80	0.11	0.69
*NHNH ₂	0.99	0.14	0.85
*NH ₂ NH ₂	1.27	0.16	1.11
*N*N	0.21	0.06	0.15
*N*NH	0.53	0.08	0.45
*NH*NH	0.79	0.10	0.69
*NH*NH ₂	0.91	0.16	0.75
*NH ₂ *NH ₂	1.24	0.18	1.06
*NH ₂	0.68	0.08	0.60
*NH3	0.96	0.12	0.84
*Н	0.18	0.16	0.02
H ₂	0.27	0.41	-0.14
N ₂	0.15	0.60	-0.45

Table S1 Free energy corrections: E_{ZPE} and S represent the zero-point energy change and theentropy change of intermediate for NRR on TiSiGeN₄-V_N-t, where * denoted the adsorption site.

NH ₃	0.94	0.60	0.34

Table S2. Free energy corrections: E_{ZPE} and S represent the zero-point energy change and the entropy change of intermediate for NRR on TiSiGeN₄-V_N-mt, where * denoted the adsorption site.

Species	E_{ZPE} (eV)	<i>TS</i> (eV)	$E_{\rm ZPE} - TS$ (eV)
*NN	0.25	0.10	0.15
*NNH	0.57	0.10	0.47
*NNH ₂	0.86	0.11	0.75
*NNH ₃	1.09	0.12	0.97
*NH	0.45	0.03	0.42
*NH ₂	0.69	0.09	0.60
*NH ₃	0.92	0.16	0.76
*NHNH	0.82	0.09	0.73
*NHNH ₂	1.00	0.14	0.86
*NH ₂ NH ₂	1.30	0.18	1.12
*N*N	0.28	0.04	0.15
*N*NH	0.56	0.04	0.45
*NH*NH	0.84	0.06	0.78
*NH*NH ₂	0.99	0.10	0.89
*NH ₂ *NH ₂	1.25	0.17	1.08
*NH ₂	0.70	0.09	0.61
*NH ₃	0.93	0.17	0.76
*Н	0.18	0.17	0.01

H ₂	0.27	0.41	-0.14
N ₂	0.15	0.60	-0.45
NH ₃	0.94	0.60	0.34