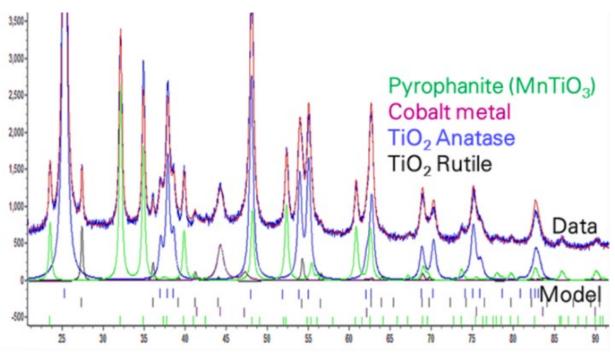
## **Supporting Information**

## **Controlling Cobalt Fischer-Tropsch Stability and Selectivity through Manganese Titanate formation**

James Paterson, David Brown, Sarah Haigh, Philip Landon, Qizhen Li, Mark Peacock, Hendrik van Rensburg, Zhuoran Xu



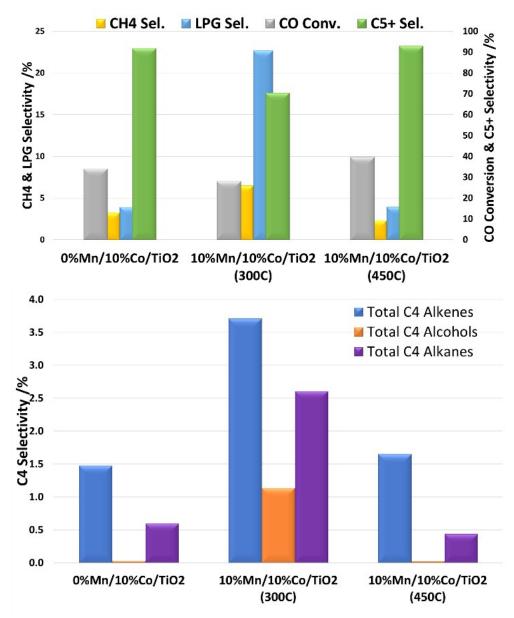
SI 1 – XRD pattern of the high temperature reduced CoMnTiO2 sample, and the XRD modelled pattern for MnTiO<sub>3</sub>

	Ru	Rutile	Ŝ	Co <sub>3</sub> O <sub>4</sub>	J	CoO	MnTiO	1i0
e, å	%, Wt/Wt	:e, Å %, Wt/Wt Cryst Size, Å %, Wt/Wt Cryst Size, Å %, Wt/Wt Cryst Size, Å %, Wt/Wt Cryst	%, Wt/Wt	Cryst Size, Å	%, Wt/Wt	Cryst Size, Å	%, Wt/Wt	Cryst
	12.5	295	16.1	73				
	12.1	300	7.1	82	7.1	48		
	12.1	314			14.3	51		
	12.6	302			8.0	57		
	13.2	314			1.2	92		
	13.2	313						
	8.2	198	20.6	41				
	8.1	209	23.3	39				
	7.5	212	20.2	32	3.9	53		
	7.1	230	6.3	25	17.1	52		
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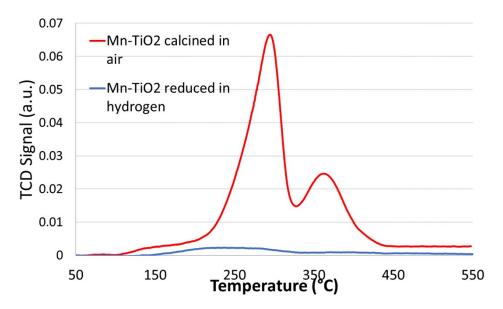
SI 2 – In situ XRD reduction data for the low and high manganese promoted samples. In particular showing the sudden onset of the MnTiO3 phase at 450C (extended data from table in manuscript)

	T-Rx	GHSV	CO Conv	Sel. CH4	Sel C5+	Sel. C2-C4	Sel. CO2	Total Alkenes	1-Butanol	Total Alkanes
	[°C]	[h-1]	[%]	[%]	[%]	[%]	[%]	[%]	[%]	[%]
0%Mn	220	8825	34.15	3.24	91.90	3.93	0.01	1.48	0.03	0.59
5%Mn	245	8823	29.60	5.46	71.24	22.86	0.04	4.77	1.01	2.19
10%Mn	243	8819	28.03	6.50	70.37	22.67	0.04	3.71	1.13	2.60
(300C)	245	0019	28.05	0.50	70.57	22.07	0.04	5.71	1.15	2.00
10%Mn	214	8821	39.62	2.30	92.92	3.96	0.00	1.65	0.03	0.44
(450C)	214	0021	39.02	2.50	92.92	5.90	0.00	1.05	0.05	0.44

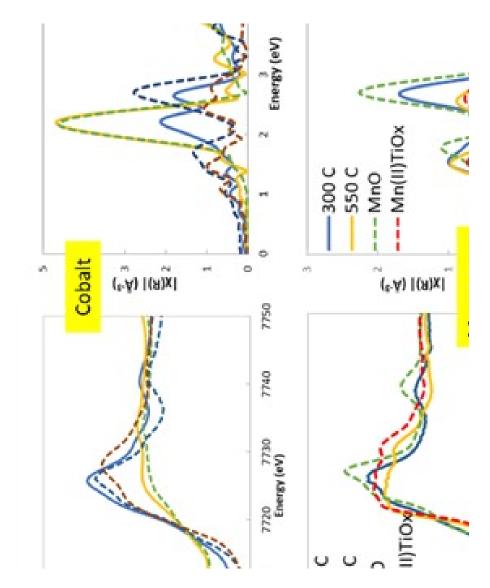
SI 3 – Summary data showing selectivity to C5+ products, alkenes and alcohols for the 10%Mn (450C) returning to a 0%Mn type level rather than matching the 10%Mn(300C) condition



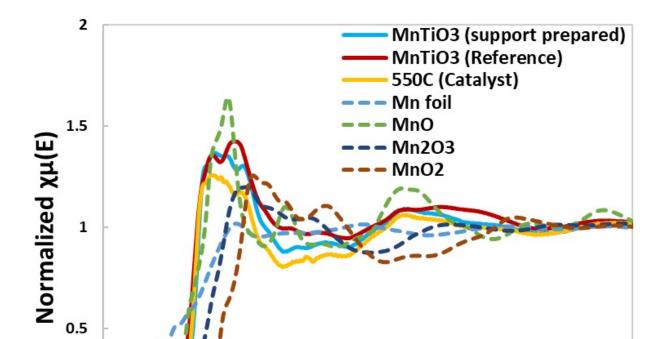
**SI 4** – Graphical comparison of the data comparing the mon-manganese and the high manganese samples at 300C and 450C



SI 5 - Temperature programmed reduction of the calcined and reduced manganese-titania ex situ support samples

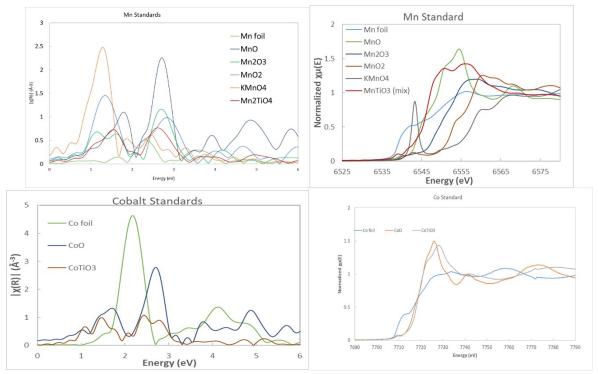


SI 6 - EXAFS and XANES images from manuscripts, expanded for clarity

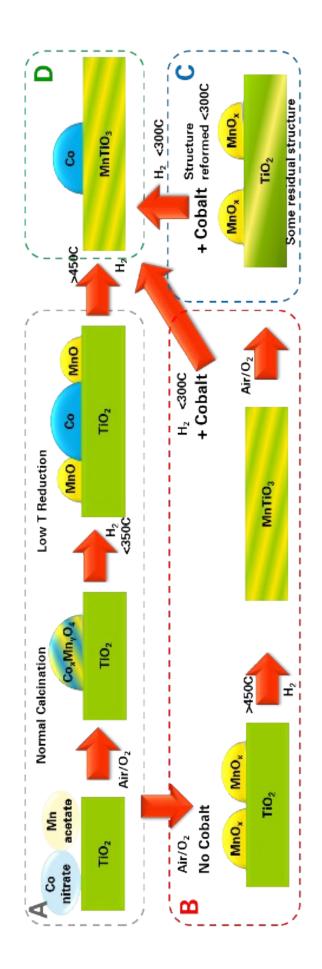


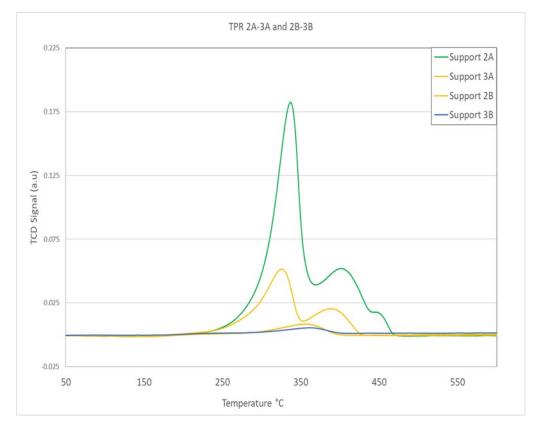
		Mn-O/Å	CN	σ <sup>2</sup>	Mn-Mn/ Å	CN	Mn-Ti/ Å	CN	σ <sup>2</sup>
100/00/50/50/040/7:00	300 °C	2.10	5.6	0.010	3.12	4.8	-	-	0.009
10%Co/5%Mn/TiO2	550 °C	2.12	5.1	0.010	3.06	3.9	3.41	2.5	0.007
10%Co/F%/Mp/T:02	300 °C	2.12	4.6	0.007	3.12	7.3	-	-	0.009
10%Co/5%Mn/TiO2	550 °C	2.12	5.7	0.011	3.05	4.5	3.43	3.0	0.008
	MnO	2.22	6	-	3.14	12	-	-	-
	Mn <sub>2</sub> O <sub>3</sub>	2.012	6	-	3.13	6	-	-	-
References	MnO <sub>2</sub>	1.883	6	-	3.418	8	-	-	-
	MnTiO <sub>3</sub>	2.11, 2.28	6	-	3.063	3	3.433	3	-
	MnTi <sub>2</sub> O <sub>4</sub>	2.041	4	-	3.724	4	-	-	-
Ex situ Support	(b) MnTiO <sub>3</sub>	2.11, 2.29	6	0.009	3.058	4.3	3.474	2.9	0.008

SI 7 – XAS analysis of samples with references, and comparison with the ex situ prepared support



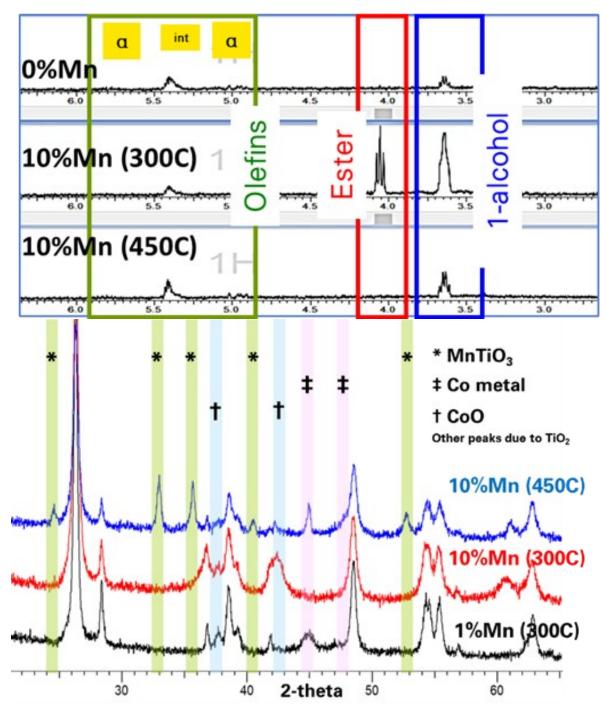
SI 9 - XAS Standards for EXAFS and XANES for manganese and cobalt edge species



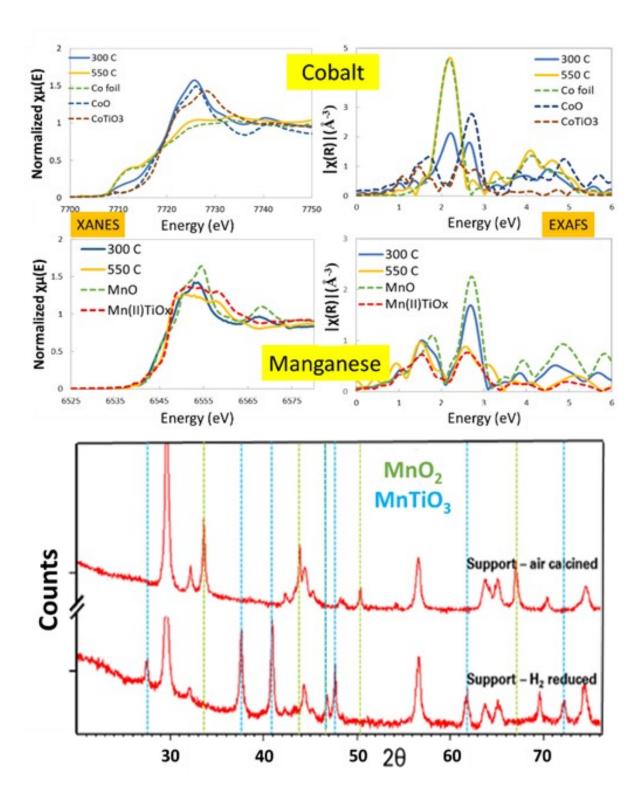


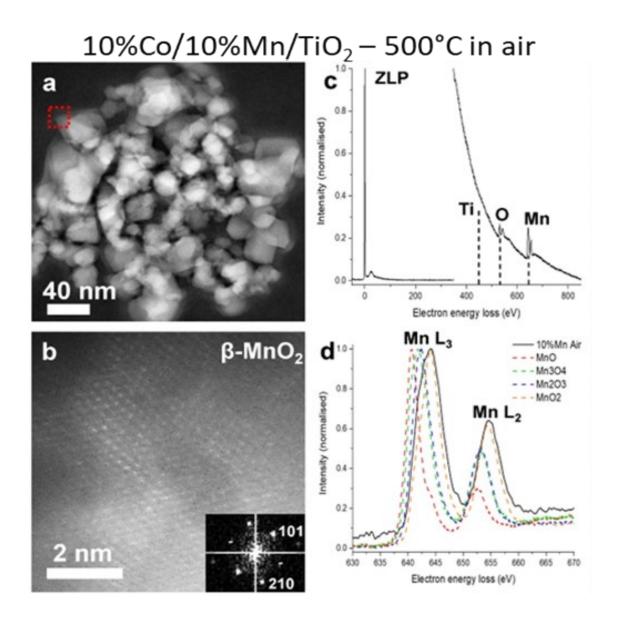
SI 10 – Overview of the catalyst transitions

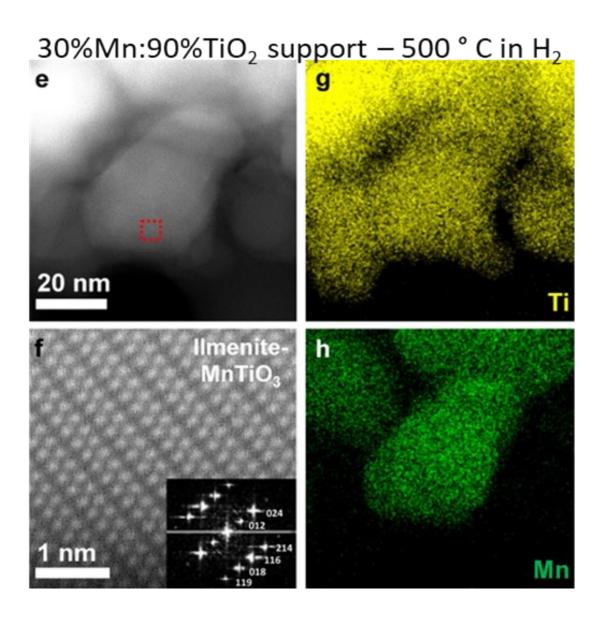
SI-11 Temperature programmed reduction of the oxidised support materials and reduced support materials. Reduced supports show very small amounts of hydrogen uptake (2B and 3B, 30%Mn:70%Ti, and 10%Mn:90%Ti)). Of the oxide forms, some reduction is seen, likely from  $MnO_2$  to MnO initially, but also a small peak at 450°C which is likely to be the  $MnTiO_3$  transition. While there is no stoichiometry mass loss between  $MnO.TiO_2$  and  $MnTiO_3$ , there is likely to be some reorganisation which forms this peak at 450°C.

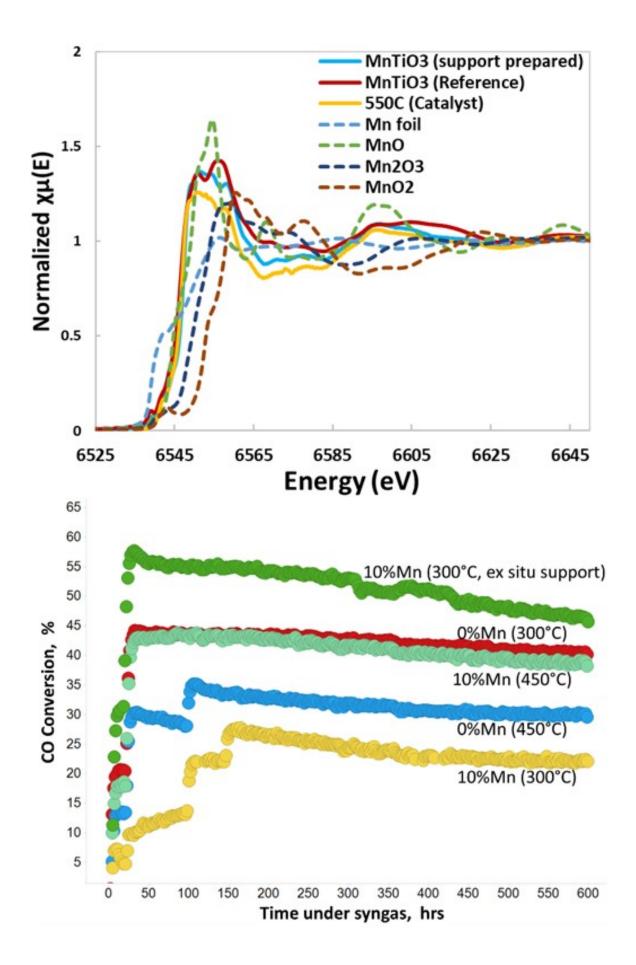


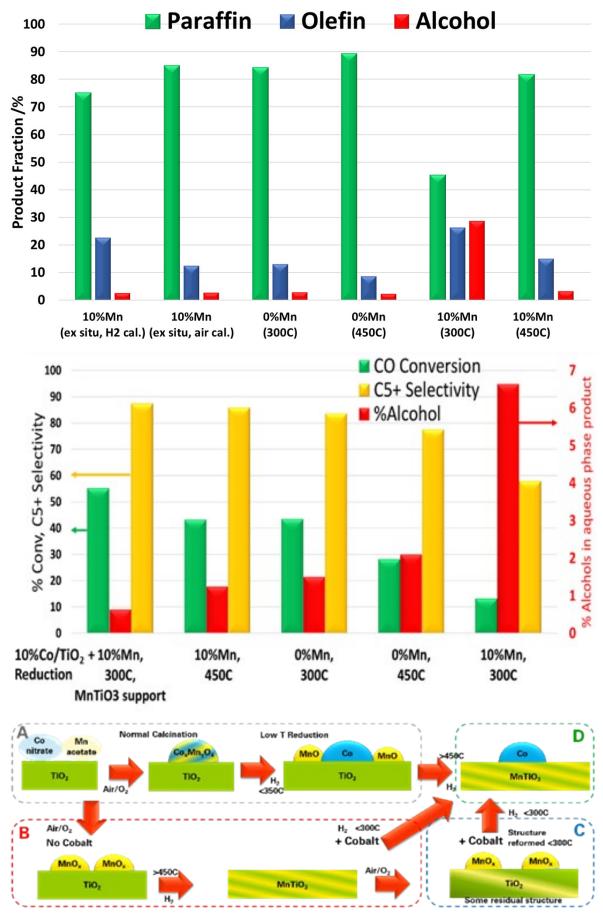
A larger version of the images from the main manuscript:











SI-12 – images from the main publication for easier viewing