Orthometallated Pd(II) C^N^S Pincer Complexes Catalyzed Sustainable Synthesis of Bis(indolyl)methanes via Acceptorless Dehydrogenative Coupling of Alcohols

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1. Materials and methods

Chemically pure and analar grade reagents were used for all the reactions. Commercially available [PdCl₂(PPh₃)₂], Phenyl isothiocyanate, benzhydrazide derivatives, CDCl₃, DMSO-d₆, Indole, 5-methoxy indole and various alcohols was used as supplied from sigma aldrich. The solvents werefreshly distilled before use following the standard procedures.¹ Melting point was recorded in the Boetiesmicro heating table and is uncorrected. The elemental analysis of carbon, hydrogen, nitrogen and sulphurwere performed at Sophisticated Test and Instrumentation Centre (STIC), Cochin University of Scienceand Technology, Kochi. Infrared spectra of complexes were recorded in KBr pellets with a Perkin–Elmer 597 spectrophotometer in the range of 4000-400cm⁻¹.The ¹H and ¹³C{¹H} NMR spectra were recorded in CDCl₃ and DMSO-d₆ with Bruker 400 MHz instrument using TMS as internal reference.

X-ray crystallographic data collection

Single crystals of complex 1 was grown by slow evaporation of a chloroform in methanol (2:1) solution at room temperature. The data collection was carried out using a Bruker AXS Kappa APEX II single crystal X-ray diffractometer using monochromated Mo–K α radiation (kI = 0.71073 A°). Data was collected at 296 K. The absorption corrections were performed by the multi-scan method using SADABS software.³ Corrections were made for Lorentz and polarization effects. The structures were solved by direct methods (SHELXS 97) and refined by full-matrix least squares on F2 using SHELXL 97.⁴ All non-hydrogen atoms were refined anisotropically and the hydrogen atoms in these structures were located from the difference Fourier map and constrained to the ideal positions in the refinement procedure. The unit cell parameters were determined by the method of difference vectors using reflections scanned from three different zones of the reciprocal lattice. The intensity data were measured using ω and φ scan with a frame width of 0.5°. Frame integration and data reduction were performed using the Bruker SAINT-Plus (Version 7.06a) software.⁵ Figure 1 was drawn with ORTEP and the structural data have been deposited at the Cambridge Crystallographic Data Centre: CCDC **2066887**.

Crystaldata	Pd1
Empirical formula	C ₃₂ H ₂₆ N ₃ OPPdS
Formula weight	637.99
Colour	yellow
CCDC number	2066887
Temperature(K)	295(2)
Crystal system	triclinic
Space group	P-1
a/Å	9.8480(4)
b/Å	11.6412(5)
c/Å	14.0807(6)
α/°	71.591(4)
β/°	74.372(4)
γ/°	75.139(4)
Volume/Å ³	1448.84(12)
Ζ	2
Crystal_density (Mg m ⁻³)	1.462
Absorption coefficient(μ)(mm ⁻¹)	0.798
F(000)	648.0
Radiations	Mo Ka ($\lambda = 0.71073$)
Crystal size(mm)	0.14 imes 0.06 imes 0.05
Theta range(°)	6.84 to 58.99
Limiting indices	$-13 \le h \le 13, -13 \le k \le 16, -19 \le l \le 18$
Reflections collected	17484
Data/restraints/parameters	6991/0/360
Goodness-of-fit on F ²	1.044
FinalRindices[I>=2 σ (I)]	
Rindices (alldata)	$R_1 = 0.0387, wR_2 = 0.0761$ $R_1 = 0.0387, wR_2 = 0.0761$ 1.13/-0.65
Largestdiff.Peak andhole(e °A ⁻³)	

2. Table S1.Crystal data and structure refinement for complex Pd1

Pd1			
Bond lengths (Å)			
S(1)Pd(1)	2.3547(9)		
P(1)Pd(1)	2.2444(7)		
N(1)Pd(1)	1.997(2)		
C(3)Pd(1)	2.040(3)		
N(1)N(2)	1.377(3)		
C(1)O(1)	1.255(3)		
C(8)N(3)	1.347(4)		
Bond angles (°)			
P(1)Pd(1) S(1)	97.75(3)		
N(1)Pd(1) S(1)	82.64(7)		
N(1)Pd(1) P(1)	175.74(7)		
N(1)Pd(1)C(3)	80.59(10)		
C(8)S(1)Pd(1)	96.72(11)		
C(3)Pd(1)P(1)	98.89(8)		
C(1)N(1)Pd(1)	120.02(19)		
N(2)C(8)N(3)	115.7(2)		
O(1)C(1)N(1)	124.4(3)		
C(1)N(1)Pd(1)	120.029(19)		
C(1)N(1)N(2)	119.2(2)		

Table S2. Selected bond lengths (Å) and bond angles (°) for the complex Pd1



Figure 1. ¹H NMR spectrum of ligand L1in DMSO- d₆ (400 MHz, 293 K)



Figure 2. ¹³C{¹H} NMR spectrum of ligand L1 in DMSO- $d_6(100 \text{ MHz}, 293 \text{ K})$.



Figure 3. ¹H NMR spectrum of ligand L2 in DMSO- d₆ (400 MHz, 293 K)



Figure 4. ¹³C{¹H} NMR spectrum of ligand L2 in DMSO- $d_6(100 \text{ MHz}, 293 \text{ K})$.



Figure 5. ¹H NMR spectrum of ligand L3 in DMSO- d₆ (400 MHz, 293 K).



Figure 6. ¹³C $\{^{1}H\}$ NMR spectrum of ligand L3 in DMSO-d₆ (100 MHz, 293 K).



Figure 7. ¹H NMR spectrum of complex Pd1 in CDCl₃ (400 MHz, 293 K)



Figure 8. ¹³C{¹H} NMR spectrum of complex Pd1 in CDCl₃ (100 MHz, 293 K)



Figure 9. ¹H NMR spectrum of complex Pd2 in CDCl₃ (400 MHz, 293 K)



Figure 10. ¹³C{1H} NMR spectrum of complex Pd2 in CDCl₃ (100 MHz, 293 K)



Figure 11. ¹H NMR spectrum of complex Pd3 in CDCl₃ (400 MHz, 293 K)



Figure 12. ¹³C{1H} NMR spectrum of complex Pd3 in CDCl₃ (100 MHz, 293 K)

5. HRMS spectra of the complexes Pd1-Pd3







Figure 14. HR-MS spectrum of complex Pd2



Figure 15. HR-MS spectrum of complex Pd3



6. HRMS spectra of the Pd alkoxy intermediate

Figure 16. HR-MS spectrum of palladium alkoxy intermediate

General Procedure for Synthesis of Bis(Indolyl)methanes

To an oven dried 50 mL round-bottom flask, indole (2 mmol), alcohol (1 mmol), Pd2 catalyst (0.5 mol%), KO'Bu (0.50 mmol) and toluene (3 mL) were added under a gentle stream of nitrogen. The reaction mixture was kept for stirring at 110 °C (oil-bath temperature) for 18 h. Then, the reaction was quenched with water and extracted with ethyl acetate. The combined organic layer was dried over anhydrous Na₂SO₄. The crude mixture was purified by silica gel column chromatography using hexane/ethyl acetate as an eluting system to yield the desired bis(indolyl)methane product.

7. Characterization of the catalytic isolated product of bis(indolyl)methane product.

3,3'-(phenylmethylene)bis(1H-indole) (3a).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 8.11 (d, ³*J*= 8Hz, 1H), 7.59 (s, 1H), 7.46 (d, ³*J*= 8Hz, 1H), 7.38-7.13 (m, 10H), 6.99 (t, ³*J*=4Hz, ³*J*=8Hz, 2H), 6.58 (s, 2H), 5.86 (s, 1H).¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 144.07, 136.71, 133.73, 130.25, 128.77, 128.55, 128.28, 127.10, 126.19, 123.71, 121.95, 119.97, 119.68, 119.25, 111.12, 40.22.

3,3'-(p-tolylmethylene)bis(1H-indole) (3b).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.81 (s, 1H), 7.37 (d, ³J= 8Hz, 2H), 7.25 (d, ³J= 8Hz, 2H), 7.14-7.10 (m, 5H), 6.96 (t, ³J=8Hz, ³J=8Hz, 4H), 6.52 (s, 2H), 5.80 (s, 1H), 2.24 (s, 3H).¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 144.10, 137.71, 136.74, 129.54, 128.14, 127.17, 126.98, 125.82, 123.74, 121.88, 119.96, 119.74, 119.21, 111.16, 40.20, 21.59.

3,3'-(m-tolylmethylene)bis(1H-indole) (3c).⁵



Red solid. 1H NMR (400 MHz, CDCl3) δ (ppm) 7.73 (s, 1H), 7.36 (d, ${}^{3}J$ = 8Hz, 2H), 7.25 (d, ${}^{3}J$ = 8Hz, 2H), 7.19 (d, ${}^{3}J$ = 8Hz, 2H), 7.12 (t, ${}^{3}J$ =8Hz, ${}^{3}J$ =8Hz, 2H), 7.04 (d, ${}^{3}J$ = 8Hz, 2H), 6.97 (t, ${}^{3}J$ =8Hz, 2H), 6.53 (s, 2H), 5.81 (s, 1H), 2.29 (s, 3H). ${}^{13}C{}^{1H}NMR$ (100 MHz, CDCl₃) δ (ppm) 141.10, 136.73, 135.55, 128.99, 128.64, 127.15, 123.66, 121.90, 120.01, 119.86, 119.22, 111.13, 39.84, 21.13.

3,3'-((4-methoxyphenyl)methylene)bis(1H-indole) (3d).⁵



Orange solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.82 (s, 2H), 7.37 (d, ³*J*= 8Hz, 2H), 7.30 (d, ³*J*= 8Hz, 2H), 7.23 (d, ³*J*= 8Hz, 2H), 7.14 (t, ³*J*=8Hz, ³*J*=4Hz, 2H), 6.99 (t, ³*J*=4Hz, ³*J*=8Hz, 2H), 6.80 (d, ³*J*= 8Hz, 2H), 6.59 (s, 2H), 5.82 (s, 1H), 3.76 (s, 3H).¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 157.94, 136.75, 136.30, 129.65, 127.11, 123.57, 121.92, 120.06, 120.01, 119.22, 113.62, 111.07, 55.25, 39.37.

3,3'-((3-methoxyphenyl)methylene)bis(1H-indole) (3e).⁵



Orange solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.81 (s, 2H), 7.39 (d, ³*J*= 8Hz, 2H), 7.30 (d, ³*J*= 8Hz, 2H), 7.23-7.12 (m, 3H), 7.01-6.90 (m, 4H), 6.75 (d, ³*J*= 8Hz, 1H), 6.60 (s, 2H), 5.83 (s, 1H), 3.71 (s, 3H).¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 159.59, 145.81, 136.70, 129.16, 127.12, 123.63, 121.92, 121.36, 119.92, 119.54, 119.24, 114.83, 111.27, 111.06, 55.14, 40.25.

3,3'-((4-(methylthio)phenyl)methylene)bis(1H-indole) (3f).⁵



Orange solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.67 (s, 2H), 7.34 (d, ³*J*= 8Hz, 2H), 7.25-7.19 (m, 4H), 7.13 (t, ³*J*=8Hz, ³*J*=8Hz, 4H), 6.97 (t, ³*J*=8Hz, ³*J*=8Hz, 2H), 6.49 (s, 2H), 5.79 (s, 1H), 2.40 (s, 3H). ¹³C{¹H}NMR (100 MHz, CDCl3) δ (ppm) 141.23, 136.72, 135.60, 129.31, 127.04, 126.73, 123.70, 122.03, 119.97, 119.49, 119.32, 111.19, 39.72, 16.04.

3,3'-((4-isopropylphenyl)methylene)bis(1H-indole) (3g).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.49 (s, 2H), 7.35 (d, ³*J*= 8Hz, 2H), 7.18 (d, ³*J*= 8Hz, 4H), 7.13-7.07 (m, 4H), 6.95 (t, ³*J*=8Hz, ³*J*=8Hz, 2H), 6.44 (s, 2H), 5.79 (s, 1H), 2.89-2.82 (m, 1H), 1.20 (d, ³*J*=8Hz, 6H).¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 146.59, 141.42, 136.72, 128.65, 127.22, 126.32, 123.71, 121.93, 120.06, 119.93, 119.24, 111.18, 39.80, 33.76, 24.18.

4-(di(1H-indol-3-yl)methyl)-N,N-dimethylaniline (3h).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.83 (s, 2H), 7.31 (d, ³*J*= 8Hz, 2H), 7.20-7.12 (m, 4H), 6.98 (t, ³*J*=8Hz, ³*J*=8Hz, 2H), 6.97 (d, ³*J*=8Hz, 2H), 6.62 (s, 2H), 5.79 (s, 1H), 2.90 (s, 6H). ¹³C{¹H}NMR (100 MHz, CDCl3) δ (ppm) 149.11, 136.74, 132.45, 129.29, 127.23, 123.53, 121.78, 120.46, 120.12, 119.11, 112.76, 110.97, 40.86, 39.18.

3,3'-((2,5-dimethoxyphenyl)methylene)bis(1H-indole) (3i).⁵



Orange solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.70 (s, 2H), 7.39 (d, ³*J*= 8Hz, 2H), 7.22 (d, ³*J*= 8Hz, 2H), 7.11 (t, ³*J*=8Hz, ³*J*=8Hz, 2H), 6.97 (t, ³*J*=4Hz, ³*J*=8Hz, 2H), 6.83 (d, ³*J*= 12Hz, 1H), 6.70 (t, ³*J*=12Hz, ³*J*=8Hz, 2H),6.46 (s, 2H), 6.29 (s, 1H), 3.72 (s, 3H), 3.60 (s, 3H).¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 153.47, 151.49, 136.75, 134.19, 127.25, 123.59, 121.75, 119.97, 119.30, 119.04, 116.95, 111.91, 111.03, 110.63, 56.65, 55.54, 32.19.

3,3'-((3,4,5-trimethoxyphenyl)methylene)bis(1H-indole) (3j).⁵



Orange solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.97 (s, 2H), 7.41 (d, ³*J*= 8Hz, 2H), 7.35 (d, ³*J*= 8Hz, 2H), 7.16 (t, ³*J*=8Hz, ³*J*=8Hz, 2H), 7.01 (t, ³*J*=8Hz, ³*J*=8Hz, 2H), 6.67 (s, 1H), 6.59 (s, 2H), 5.81 (s, 1H), 3.83 (s, 3H), 3.70 (s, 6H).¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 152.99, 139.88, 136.73, 127.08, 123.59, 121.96, 119.91, 119.58, 119.25, 111.08, 105.88, 60.88, 60.54, 56.04, 40.59.

3,3'-((4-chlorophenyl)methylene)bis(1H-indole) (3k).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.80 (s, 2H), 7.35-7.28 (m, 4H), 7.24-7.13 (m, 6H), 6.99 (t, ³*J*=8Hz, ³*J*=8Hz, 2H), 6.54 (s, 2H), 5.82 (s, 1H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 142.63, 136.73, 131.83, 130.83, 130.13, 128.42, 126.93, 123.68, 122.12, 119.86, 119.40, 119.19, 111.21, 39.66.

3,3'-((3-bromophenyl)methylene)bis(1H-indole) (3I).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.80 (s, 2H), 7.47 (s, 1H), 7.36-7.29 (m, 5H), 7.24 (d, ³*J*= 4Hz, 1H), 7.18-7.08 (m, 3H), 7.00 (t, ³*J*=8Hz, ³*J*=8Hz, 2H), 6.55 (s, 2H), 5.82 (s, 1H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 146.56, 136.71, 131.74, 129.89, 129.41, 127.43, 126.90, 123.73, 122.45, 122.13, 119.79, 119.43, 118.94, 111.20, 39.98.

3,3'-((2-bromophenyl)methylene)bis(1H-indole) (3m).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.56 (s, 2H), 7.50 (d, 3J= 8Hz, 1H), 7.29 (d, 3J= 8Hz, 2H), 7.17 (d, 3J= 8Hz, 2H), 7.09-6.89 (m, 7H), 6.37 (s, 2H), 6.19 (s, 1H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 143.05, 136.74, 132.92, 130.55, 127.92, 127.38, 127.06, 124.89, 123.97, 122.09, 119.93, 119.39, 118.41, 111.21, 39.59.

3,3'-((4-fluorophenyl)methylene)bis(1H-indole) (3n).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.70 (s, 2H), 7.33 (d, 3J= 8Hz, 2H), 7.26-7.20 (m, 4H), 7.13 (t, ³*J*=8Hz, ³*J*=8Hz, 2H), 6.98 (t, ³*J*=4Hz, ³*J*=8Hz, 2H), 6.89 (t, ³*J*=8Hz, 2H), 6.89 (t, ³*J*=8Hz, 2H), 6.48 (s, 2H), 5.81 (s, 1H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 162.68, 160.26, 139.78, 136.76, 130.18, 130.10, 126.99, 123.69, 122.09, 119.92, 119.54, 119.37, 115.14, 114.92, 111.25, 39.52.

3,3'-(thiophen-2-ylmethylene)bis(1H-indole) (30).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.87 (s, 2H), 7.46 (d, 3J= 8Hz, 3H), 7.34 (d, 3J= 8Hz, 2H), 7.18-7.13 (m, 2H), 7.02 (t, ³J=8Hz, ³J=8Hz, 2H), 6.91 (d, ³J=4Hz, 2H), 6.80 (d, ³J=8Hz, 2H), 6.15 (s, 1H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 172.83, 148.68, 136.61, 126.79, 126.45, 125.17, 123.63, 123.21, 122.05, 119.79, 119.71, 119.39, 111.15, 35.35.

3,3'-(furan-2-ylmethylene)bis(1H-indole) (3p).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.80 (s, 2H), 7.45 (d, ³J= 8Hz, 2H), 7.32-7.26 (m, 3H), 7.14 (t, ³J=8Hz, ³J=8Hz, 2H), 7.01 (t, ³J=8Hz, ³J=8Hz, 2H), 6.75 (s, 2H), 6.27 (s, 1H), 6.03 (s, 1H), 5.90 (s, 1H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 157.15, 141.27, 136.55, 126.80, 123.15, 121.99, 119.71, 119.39, 117.14, 111.21, 110.21, 106.66, 34.14.

3,3'-(naphthalen-1-ylmethylene)bis(1H-indole) (3q).⁵



Red solid. ¹H NMR (400 MHz, dmso-d₆) δ (ppm) 10.83 (s, 2H), 8.28 (d, 3J= 8Hz, 2H), 7.92 (d, 3J= 8Hz, 1H), 7.78 (d, 3J= 8Hz, 1H), 7.44-7.28 (m, 7H), 7.05 (t, ³J=8Hz, ³J=8Hz, 2H), 6.86 (t, ³J=8Hz, ³J=8Hz, 2H), 6.76 (s, 2H), 6.66 (s, 1H). ¹³C{¹H}NMR (100 MHz, dmso-d₆) δ (ppm) 140.27, 136.64, 133.58, 131.30, 128.53, 126.60, 125.78, 125.50, 125.31, 124.24, 123.95, 120.87, 118.94, 118.24, 117.67, 111.50, 35.33.

3,3'-([1,1'-biphenyl]-4-ylmethylene)bis(1H-indole) (3r).⁵



Red solid. ¹H NMR (400 MHz, dmso-d₆) δ (ppm) 10.87 (s, 2H), 7.64 (d, 3J= 8Hz, 2H), 7.58 (d, 3J= 8Hz, 2H), 7.47-7.30 (m, 10H), 7.06 (t, ³J=8Hz, ³J=8Hz, 2H), 6.90 (d, ³J=8Hz, 3H), 5.91 (s, 1H). ¹³C{¹H}NMR (100 MHz, dmso-d₆) δ (ppm) 144.30, 140.05, 137.58, 136.60, 128.82, 127.08, 126.61, 126.45, 126.32, 123.56, 120.88, 119.09, 118.19, 117.89, 111.45, 28.96.

3,3'-(phenylmethylene)bis(5-methoxy-1H-indole) (3s).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.82 (s, 2H), 7.34 (d, ³J= 8Hz, 2H), 7.29-7.21 (m, 5H), 6.83-6.80 (m, 4H), 6.64 (s, 2H), 5.76 (s, 1H), 3.68 (s, 6H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 153.72, 143.96, 131.91, 128.75, 128.24, 127.54, 126.13, 124.46, 119.32, 111.92, 111.69, 102.02, 55.88, 40.33.

3,3'-(p-tolylmethylene)bis(5-methoxy-1H-indole) (3t).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.78 (s, 2H), 7.23-7.17 (m, 4H), 7.06 (d, ³*J* = 8 Hz, 2H), 6.80 (d, ³*J* = 8 Hz, 4H), 6.61 (s, 2H), 5.71 (s, 1H), 3.68 (s, 6H), 2.30 (s, 3H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 153.66, 140.96, 135.49, 131.93, 128.95, 128.60, 127.57, 124.45, 119.49, 111.83, 111.71, 102.08, 55.91, 39.89, 21.10.

3,3'-((4-methoxyphenyl)methylene)bis(5-methoxy-1H-indole)(3u).⁵



orange solid.¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.83 (s, 2H), 7.23 (t, ${}^{3}J = 8$ Hz, ${}^{3}J = 8$ Hz, 4H), 6.81 (d, ${}^{3}J = 8$ Hz, 6H), 6.62 (s, 2H), 5.71 (s, 1H), 3.77 (s, 4H), 3.69 (s, 5H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 172.06, 153.67, 136.22, 131.94, 129.64, 127.52, 124.42, 123.07, 119.63, 118.61, 113.60, 111.85, 111.70, 102.07, 55.91, 39.44.

3,3'-((4-(tert-butyl)phenyl)methylene)bis(5-methoxy-1H-indole)(3v).⁵



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.77 (s, 2H), 7.28-7.23 (m, 4H), 7.18 (d, ³*J* = 8 Hz, 2H), 6.80 (d, ³*J* = 8 Hz, 4H), 6.63 (s, 2H), 5.72 (s, 1H), 3.66 (s, 6H), 1.28 (s, 9H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 153.63, 140.90, 131.90, 128.35, 127.60, 125.09, 124.43, 119.53, 111.85, 111.72, 102.07, 55.87, 39.82, 34.42, 31.49.

3,3'-(hexane-1,1-diyl)bis(1H-indole) (3w).



Red solid. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 7.59 (s, 2H), 7.49 (d, ³*J* = 8 Hz, 2H), 7.12 (d, ³*J* = 8 Hz, 2H), 7.02 (t, ³*J* = 8 Hz, ³*J* = 8 Hz, 2H), 6.93 (t, ³*J* = 8 Hz, ³*J* = 8 Hz, 2H), 6.73 (s, 1H), 4.34 (t, ³*J* = 8 Hz, ³*J* = 4 Hz, 1H), 2.08 (dd, 2H), 1.30-1.12 (m, 6H), 0.75 (t, ³*J* = 8 Hz, ³*J* = 8 Hz, 3H). ¹³C{¹H}NMR (100 MHz, CDCl₃) δ (ppm) 135.48, 126.09, 120.58, 120.41, 119.40, 118.57, 117.87, 110.06, 34.81, 32.91, 30.96, 26.97, 21.61, 13.10, 12.98.

8a. NMR spectrum of the isolated benzaldehyde intermediate



Figure 17. ¹H NMR spectrum of Benzaldehyde in CDCl₃ (400 MHz, 293 K)



Figure 18. ¹³C{¹H} NMR spectrum of Benzaldehyde in CDCl₃ (100 MHz, 293 K) **8b. NMR spectrum of the Bisindole products**



Figure 19. ¹H NMR spectrum of 3a in CDCl₃ (400 MHz, 293 K)





Figure 20. ¹³C{¹H} NMR spectrum of **3a** in CDCl₃(100 MHz, 293 K)

Figure 21. ¹H NMR spectrum of 3b in CDCl₃ (400 MHz, 293 K)





Figure 22. ¹³C{¹H} NMR spectrum of **3b** in CDCl₃(100 MHz, 293 K)

Figure 23. ¹H NMR spectrum of 3c in CDCl₃ (400 MHz, 293 K)



Figure 24. ¹³C{¹H} NMR spectrum of 3c in CDCl₃ (100 MHz, 293 K)



Figure 25. ¹H NMR spectrum of 3d in CDCl₃ (400 MHz, 293 K)



Figure 26. ¹³C{¹H} NMR spectrum of 3d in $CDCl_3(100 \text{ MHz}, 293 \text{ K})$



Figure 27. ¹H NMR spectrum of 3e in CDCl₃ (400 MHz, 293 K)



Figure 28. ¹³C{¹H} NMR spectrum of **3e** in CDCl₃ (100 MHz, 293 K)



Figure 29. ¹H NMR spectrum of 3f in CDCl₃ (400 MHz, 293 K)



Figure 30. ¹³C{¹H} NMR spectrum of **3f** in CDCl₃ (100 MHz, 293 K)



Figure 31. ¹H NMR spectrum of 3g in CDCl₃ (400 MHz, 293 K)



Figure 32. ¹³C{¹H} NMR spectrum of **3g** in CDCl₃ (100 MHz, 293 K)



Figure 33. ¹H NMR spectrum of 3h in CDCl₃ (400 MHz, 293 K)



Figure 34. ¹³C{¹H} NMR spectrum of **3h** in CDCl₃ (100 MHz, 293 K)



Figure 35. ¹H NMR spectrum of 3i in CDCl₃ (400 MHz, 293 K)



Figure 36. ¹³C{¹H} NMR spectrum of **3i** in CDCl₃(100 MHz, 293 K)



Figure 37. ¹H NMR spectrum of 3j in CDCl₃ (400 MHz, 293 K)



Figure 38. ¹³C{¹H} NMR spectrum of **3j** in CDCl₃(100 MHz, 293 K)



Figure 39. ¹H NMR spectrum of 3k in CDCl₃ (400 MHz, 293 K)



Figure 40. ¹³C{¹H} NMR spectrum of **3k** in CDCl₃ (100 MHz, 293 K)



Figure 41. ¹H NMR spectrum of 3l in CDCl₃ (400 MHz, 293 K)



Figure 42. ¹³C{¹H} NMR spectrum of **3**I in CDCl₃ (100 MHz, 293 K)



Figure 43. ¹H NMR spectrum of 3m in CDCl₃ (400 MHz, 293 K)



Figure 44. ¹³C{¹H} NMR spectrum of **3m** in CDCl₃ (100 MHz, 293 K)



Figure 45. ¹H NMR spectrum of **3n** in CDCl₃ (400 MHz, 293 K).



Figure 46. ${}^{13}C{}^{1}H$ NMR spectrum of **3n** in CDCl₃ (100 MHz, 293 K



Figure 47. ¹H NMR spectrum of **30** in CDCl₃ (400 MHz, 293 K)



Figure 48. ¹³C{¹H} NMR spectrum of **30** in CDCl₃(100 MHz, 293 K)



Figure 49. ¹H NMR spectrum of 3p in CDCl₃ (400 MHz, 293 K)



Figure 50. ¹³C{¹H} NMR spectrum of **3p** in CDCl₃ (100 MHz, 293 K)



Figure 51. ¹H NMR spectrum of 3q in DMSO-d6(400 MHz, 293 K)



Figure 52. ¹³C{¹H} NMR spectrum of **3q** in DMSO-d6 (100 MHz, 293 K)



Figure 53. ¹H NMR spectrum of 3r in DMSO-d6 (400 MHz, 293 K)



Figure 54. ¹³C{¹H} NMR spectrum of **3r** in DMSO-d6 (100 MHz, 293 K)



Figure 55. ¹H NMR spectrum of 3s in CDCl₃ (400 MHz, 293 K)



Figure 56. ¹³C{¹H} NMR spectrum of 3s in CDCl₃ (100 MHz, 293 K)



Figure 57. ¹H NMR spectrum of 3t in CDCl₃ (400 MHz, 293 K)



Figure 58. ¹³C{¹H} NMR spectrum of **3t** in CDCl₃ (100 MHz, 293 K)



Figure 59. ¹H NMR spectrum of 3u in CDCl₃ (400 MHz, 293 K)



Figure 60. ¹³C{¹H} NMR spectrum of **3u** in CDCl₃(100 MHz, 293 K)



Figure 61. ¹H NMR spectrum of **3v** in CDCl₃ (400 MHz, 293 K)



Figure 62. ¹³C{¹H} NMR spectrum of **3v** in CDCl₃ (100 MHz, 293 K)



Figure 63. ¹H NMR spectrum of 3w in CDCl₃ (400 MHz, 293 K)



Figure 64. ¹³C{¹H} NMR spectrum of **3w** in CDCl₃(100 MHz, 293 K)

9. Experimental confirmation for the H₂ liberation using GC-TCD detector

Under N_2 atmosphere, a mixture of Pd2 catalyst (0.5 mol%), 'BuOK (0.50 mmol), benzyl alcohol (1.0 mmol), and indole (2.0 mmol) was dissolved in 3 mL of toluene, and the mixture was placed in an oil bath and heated at 110 °C for 4 h. Then, the gaseous reaction mixture has been analyzed on GC-TCD detector to witness the liberation of hydrogen gas.



Figure 65. chromatogram for the evaluation of H₂

10. References

- 1. A. I. Vogel, Test book of practical organic chemistry, Longman, London, 5th edn,1989, 395.
- 2. F. H. Allen, D. Motherwell, R. Raithby, G. P. Shields and R. Taylor, New J. Chem. **1999**, *23*, 25.
- 3. G. M. Sheldrick, ActaCrystallogr., Sect. A: Found. Crystallogr. 2008, 64, 112.
- 4. Bruker-Nonius (2004), APEX-II and SAINT-Plus (Version 7.06a), Bruker AXS Inc., Madison, Wisconsin, USA.
- 5. V. Yadav, E. Balaraman and S. B. Mhaske. Adv. Synth. Catal., 2021, 363, 4430-4439.