

**Reversing Mg suppression effect on Co-Site water oxidation of MgCo_2O_4
based on vanadium-atom electronic affinity synergy with Mg sites toward
electronic redistribution**

Hui Zhang,^{a,†} Hui Han,^{a,†} Xuan Yang,^a Hongyu Ma,^a Zhifei Song^{a,b} and Xuqiang Ji^{a,*}

^a College of Materials Science and Engineering, Institute for Graphene Applied
Technology Innovation, Qingdao University, Qingdao 266071, China

^b School of Electromechanic Engineering, Qingdao University, Qingdao 266071,
China

*Corresponding Author, E-mail: xuqianglucky@163.com.

† These workers contribute equally to this paper.

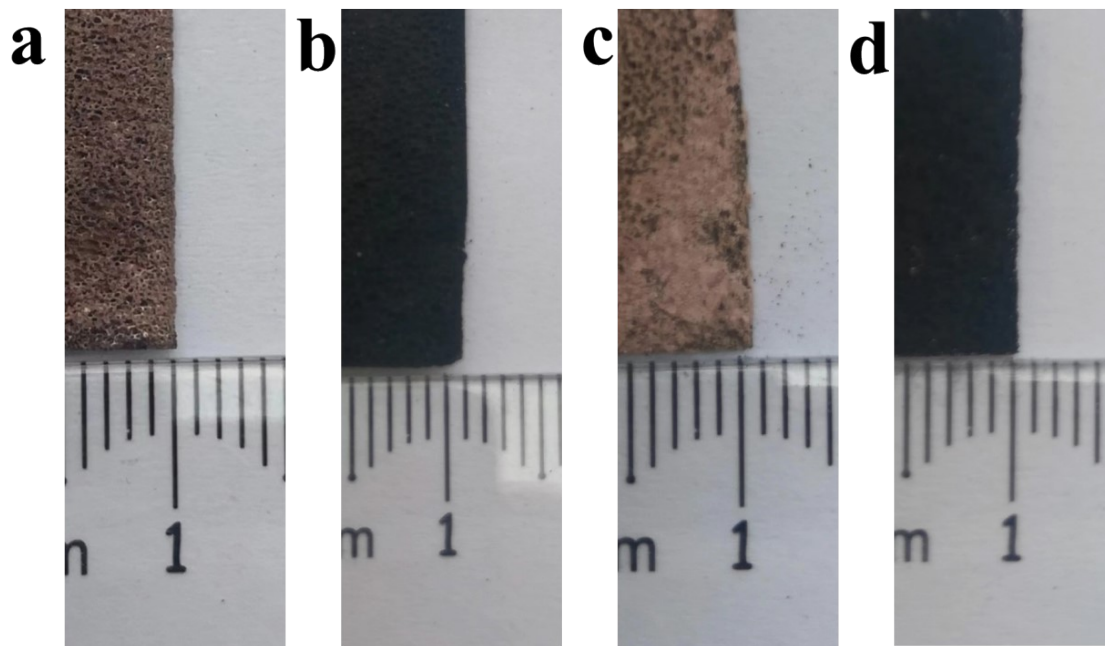


Fig. S1 The optical images of (a) V-CoOOH@MgCO₃, (b) V-MgCo₂O₄@MgO, (c) CoOOH@MgCO₃ (d) MgCo₂O₄@MgO.

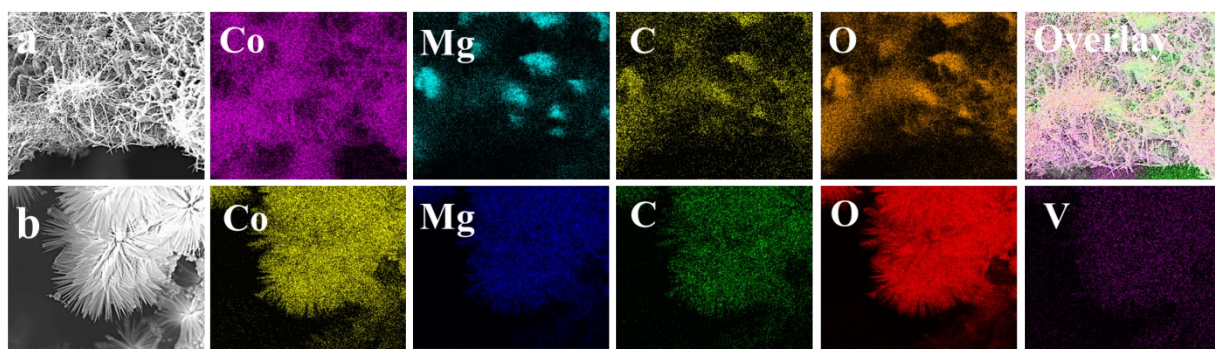


Fig. S2 The SEM images of (a) MgCo₂O₄@MgO (b) V-CoOOH@MgCO₃ and their corresponding EDX element mapping images.

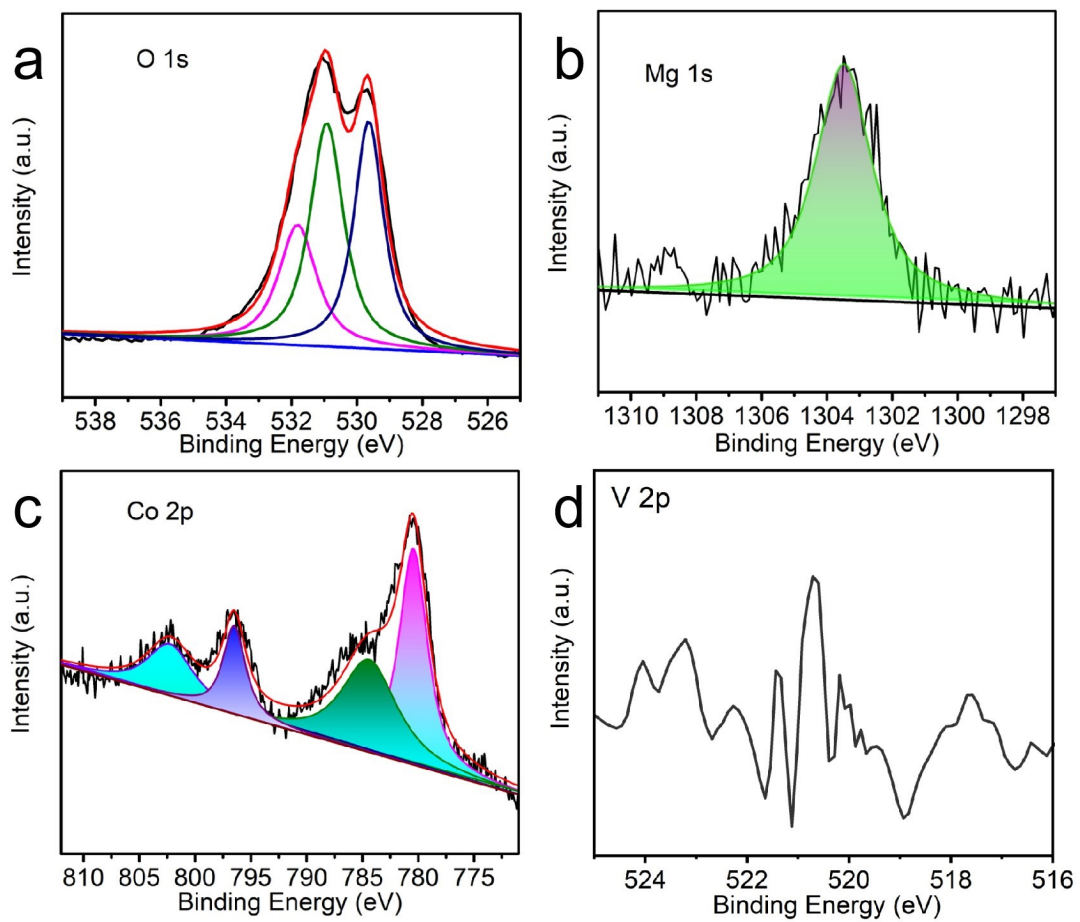


Fig. S3 XPS spectra of (a) O 1s, (b) Mg 1s, (c) Co 2p, (d) V 2p for MgCo₂O₄@MgO.

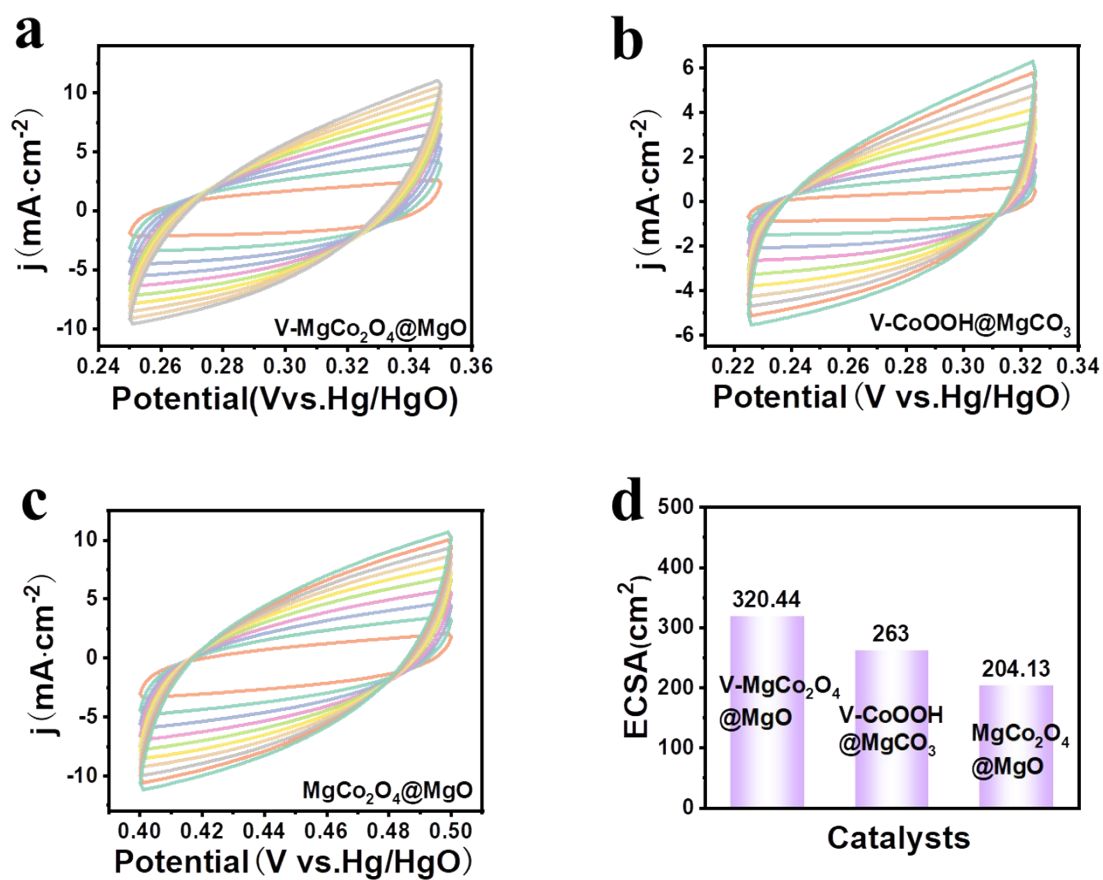


Fig. S4 CVs for (a) V-MgCo₂O₄@MgO, (b) V-CoOOH@MgCO₃, (c) MgCo₂O₄@MgO at scan rates from 10 to 100 mV·s⁻¹ in 1.0 M KOH. (d) ECSA values of V-MgCo₂O₄@MgO, V-CoOOH@MgCO₃ and MgCo₂O₄@MgO.

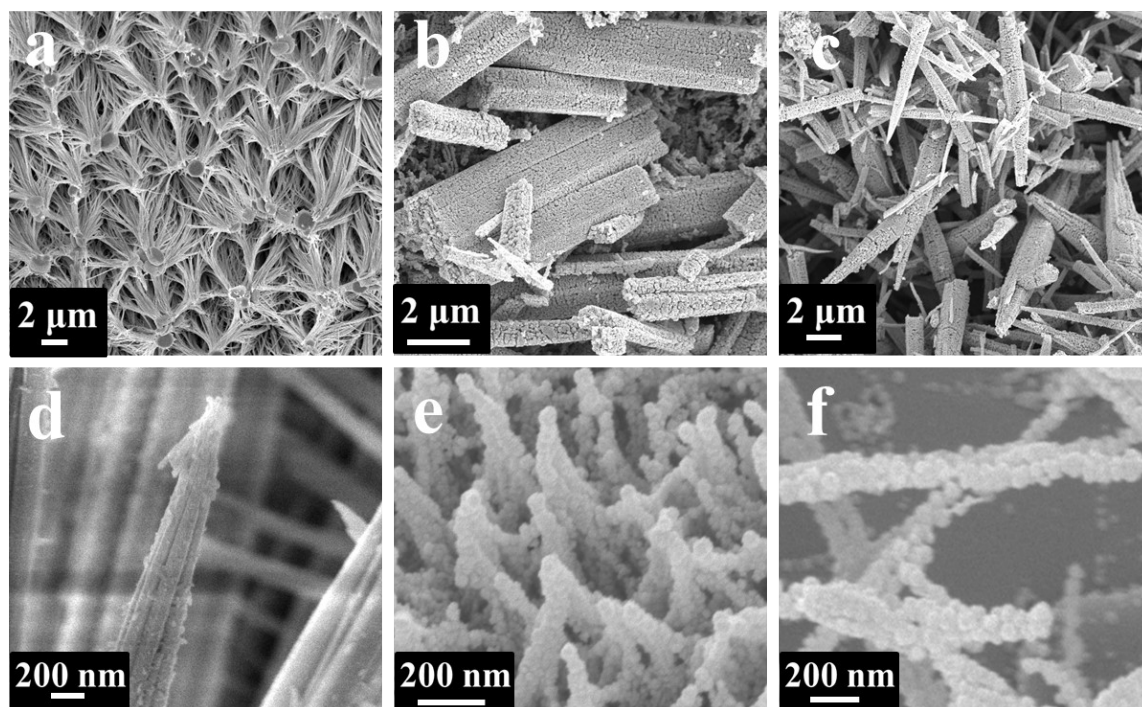


Fig. S5 The SEM images of (a and d) V-CoOOH@MgCO₃, (b and e) V-MgCo₂O₄@MgO and (c and f) MgCo₂O₄@MgO after electrolysis for 24 hours at different magnifications.

Table S1. Comparison of OER performances for V-MgCo₂O₄@MgO with other electrocatalysts recently reported in 1.0 M KOH.

Catalyst	$\eta@100\text{mA cm}^{-2}$ (mV)	Ref.
V-MgCo ₂ O ₄ @MgO	240	This work
Ni ₅ P ₄ /NiP ₂ /NiFe LDH	243	<i>J. Mater. Chem. A</i> 2018 , <i>6</i> , 13619–13623.
Fe(PO ₃) ₂	221	<i>P NATL A SCI.</i> 2017 , <i>114</i> , 5607–5611.
CoNiP@LDH	249	<i>ACS Appl. Energy Mater.</i> 2018 , <i>1</i> , 623–631.
CoFe@CoNi	262	<i>Chem. Cat. Chem.</i> 2022 , <i>14</i> , e202200347.
Ni@NiFe LDH	269	<i>J. Mater. Chem. A</i> 2019 , <i>7</i> , 21722–21729.
NiFe LDHs	280	<i>Adv. Energy Mater.</i> 2018 , <i>8</i> , 1703341.
NiMoN/NiF	294	<i>ACS Appl. Mater. Interfaces</i> 2018 , <i>10</i> , 30400–30408.
CoSe ₂ /CoP	299	<i>Part. Part. Syst. Charact.</i> 2018 , <i>35</i> , 1800135.
Cu@CoFe LDH	300	<i>Nano Energy</i> 2017 , <i>41</i> , 327–336.
Ni ₃ FeN	300	<i>ACS Catal.</i> 2018 , <i>8</i> , 9686–9696.
FeOOH/Co/FeOOH	309	<i>Angew. Chem.-Int. Edit.</i> 2016 , <i>55</i> , 3694–3698.
W _{0.5} Co _{0.4} Fe _{0.1} /NF	310	<i>Angew. Chem.-Int. Edit.</i> 2017 , <i>56</i> , 4502–4506.
Oxygenated-CoSe ₂ -MoSe ₂	310	<i>ACS Catal.</i> 2018 , <i>8</i> , 4612–4621.
NiFeCr/NF	318	<i>Energy Environ. Sci.</i> 2020 , <i>13</i> , 4225–4237.

$\text{Fe}_{0.09}\text{Co}_{0.13}\text{-NiSe}_2$	330	<i>Adv. Mater.</i> 2018 , <i>30</i> , 1802121.
$\text{Ni}_{0.94}\text{Fe}_{0.06}\text{Se}_2$	332	<i>J. Mater. Chem. A</i> 2020 , <i>8</i> , 8113–8120.
CoFe-PBA	339	<i>J. Mater. Chem. A</i> 2021 , <i>9</i> , 2135–2144.
$\text{Fe}_{0.96}\text{S/Fe(OH)}_3$	341	<i>Appl. Catal. B-Environ.</i> 2019 , <i>246</i> , 337–348.
CoOOH/Cu	358	<i>ACS Sustain. Chem. Eng.</i> 2021 , <i>9</i> , 12300–12310.
Ni-B/NF	360	<i>Nanotechnology</i> 2016 , <i>27</i> , 12LT01.
Fe-CoO	370	<i>Inorg. Chem. Front.</i> 2020 , <i>7</i> , 3327–3339.
CoFe(OH) _x	371	<i>Catal. Sci. Technol.</i> 2020 , <i>10</i> , 215–221.
NiMoN-CF	389	<i>ChemSusChem</i> 2018 , <i>11</i> , 3198–3207.
NiO/Co ₃ O ₄	394	<i>ACS Catal.</i> 2021 , <i>10</i> , 12376–12384.
two-tiered NiSe	410	<i>Adv. Energy Mater.</i> 2018 , <i>8</i> , 1702704.
Ni ₃ N	420	<i>J. Am. Chem. Soc.</i> 2015 , <i>137</i> , 4119–4125.
NiCoP/C	430	<i>Angew. Chem.-Int. Edit.</i> 2017 , <i>56</i> , 3897–3900.
NiO _x /Ni	469	<i>Nanoscale</i> 2020 , <i>11</i> , 22261–22269.
Ni ₃ FeN/CC	494	<i>ACS Appl. Mater. Interfaces</i> 2018 , <i>10</i> , 3699–3706.

Table S2. Comparison of HER performances for V-MgCo₂O₄@MgO with other electrocatalysts recently reported in 1.0 M KOH.

Catalyst	$\eta@10\text{mA cm}^{-2}$ (mV)	Ref.
V-MgCo ₂ O ₄ @MgO	105	This work
NiCo ₂ P _x	58	<i>Adv. Mater.</i> 2017 , <i>29</i> , 1605502.
Ni _{0.89} Co _{0.11} Se ₂	85	<i>Adv. Mater.</i> 2017 , <i>29</i> , 1606521.
NiP ₂ NS/CC	102	<i>Nanoscale</i> 2014 , <i>6</i> , 13440–13445.
CoMnP/Ni ₂ P/NF	108	<i>J. Mater. Chem. A</i> 2021 , <i>9</i> , 22129–22139.
NiMoN film	109	<i>Adv. Energy Mater.</i> 2016 , <i>6</i> , 1600221.
NiFeP/SG	115	<i>Nano Energy</i> 2019 , <i>58</i> , 870–876.
ReS ₂ /CFP	116	<i>Nano Energy</i> 2018 , <i>46</i> , 305–313.
Co ₄ N-VN _{1-x} O _x /CC	118	<i>Appl. Catal. B-Environ.</i> 2018 , <i>241</i> , 521–527.
NiCo ₂ O ₄ @CoMoO ₄ /NF-7	121	<i>J. Mater. Chem. A</i> 2018 , <i>6</i> , 16950–16958.
Co _x Ni _y P	129	<i>Adv. Funct. Mater.</i> 2017 , <i>27</i> , 1703455.
Ni ₂ P	136	<i>Nano Res.</i> 2020 , <i>13</i> , 2098–2105.
FeNi-LDH/CoP/CC	138.6	<i>Angew. Chem.-Int. Edit.</i> 2019 , <i>58</i> , 11903–11909.
3D-MoS ₂ /G	143	<i>Nano Energy</i> 2019 , <i>61</i> , 611–616.

FeCoNi	149	<i>ACS Catal.</i> 2017 , 7, 469–479.
NiCoFeP/C	149	<i>ChemComm</i> 2019 , 55, 10896–10899.
CoNiP@NF	155	<i>J. Mater. Chem. A</i> 2016 , 4, 10195–10202.
(Ni _{0.33} Co _{0.67})S ₂ NWs/CC	156	<i>ACS Appl. Mater. Interfaces</i> 2018 , 10, 27723–27733.
Co ₂ P/Co foil	157	<i>J. Mater. Chem. A</i> 2017 , 5, 10561–10566.
FeP ₂ –NiP ₂ @PC	179	<i>ACS Appl. Mater. Interfaces</i> 2020 , 12, 727–733.
Co ₁ Mn ₁ CH/NF	180	<i>J. Am. Chem. Soc.</i> 2017 , 139, 8320–8328.
Co _x P/N-doped C	187	<i>Carbon</i> 2019 , 145, 694–700.
CoP/Ni ₂ P	200	<i>RSC Adv.</i> 2021 , 11, 22467–22472.
Ni ₃ Se ₂	203	<i>Nano Energy</i> 2016 , 24, 103–116.
Co ₂ P/CoNPC	208	<i>Adv. Mater.</i> 2020 , 32, 2003649.
CoP	209	<i>J. Am. Chem. Soc.</i> 2018 , 140, 2610–2618.
CoNi ₂ Se ₄	220	<i>ChemComm</i> 2017 , 53, 5412–5415.
bulk MoB	225	<i>Angew. Chem.-Int. Edit.</i> 2012 , 51, 12703–12706.
FeCo/Co ₂ P@NPCF	260	<i>Adv. Energy Mater.</i> 2020 , 10, 1903854.
Co ₂ P@N, P-PCN	360	<i>J. Mater. Chem. A</i> 2016 , 4, 15501–15510.