

Supporting information

Amorphous Phosphorus Oxynitride as a Robust Catalyst for Steam-free Direct Dehydrogenation of Ethylbenzene to Styrene: Effect of Calcination Temperature

Lukai Luo,^a Yuan Ma,^a Yuwei Liu,^a Baining Lin,^b Chaojun Guo,^a Jun Gong,^a Yating Xie^a and
Yonghua Zhou^{a*}

a. College of Chemistry and Chemical Engineering, Central South University, Changsha 410083, China.

b. State Key Laboratory of Utilization of Woody Oil Resource, Hunan Academy of Forestry, Changsha 410004, China.

Corresponding author: Yonghua Zhou, E-mail: zhouyonghua@csu.edu.cn

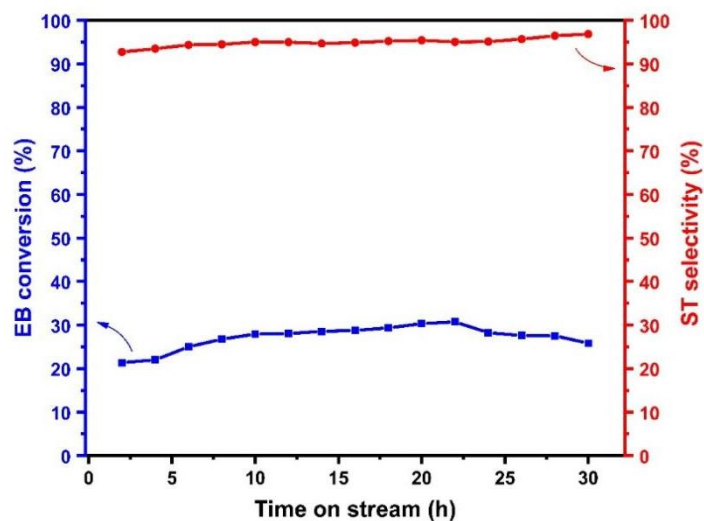


Fig. S1 The plots of x_{EB} and S_{ST} vs. time on stream of PNO-700 for the DDH reaction of ethylbenzene to styrene. Reaction conditions: 50 mg catalyst, 600 °C, 2.8 vol% ethylbenzene in N_2 , 20 mL min⁻¹.

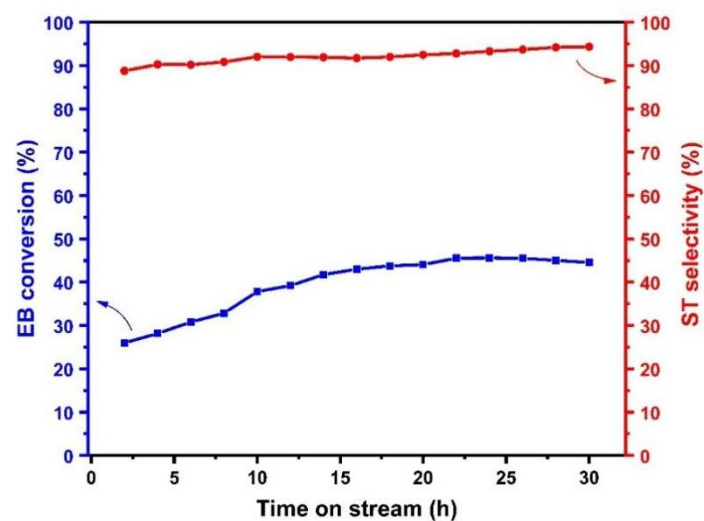


Fig. S2 The plots of x_{EB} and S_{ST} vs. time on stream of PNO-900 for the DDH reaction of ethylbenzene to styrene. Reaction conditions: 50 mg catalyst, 600 °C, 2.8 vol% ethylbenzene in N_2 , 20 mL min⁻¹.

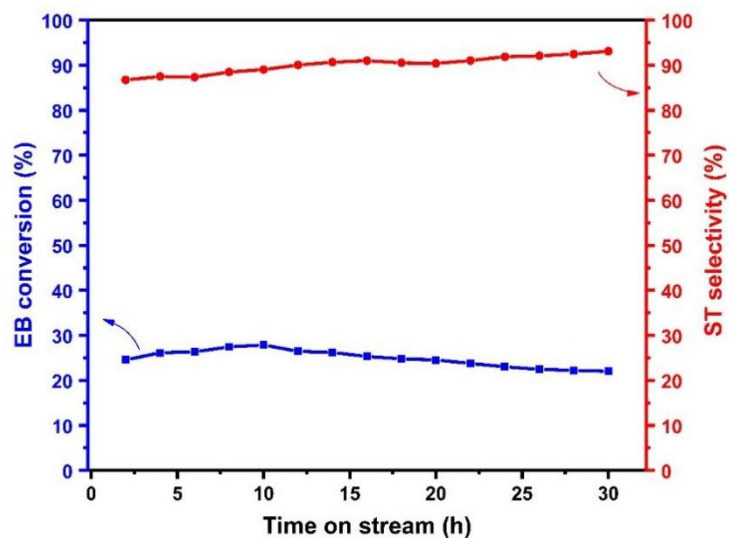


Fig. S3 The plots of x_{EB} and S_{ST} vs. time on stream of PNO-1000 for the DDH reaction of ethylbenzene to styrene. Reaction conditions: 50 mg catalyst, 600 °C, 2.8 vol% ethylbenzene in N_2 , 20 mL min⁻¹.

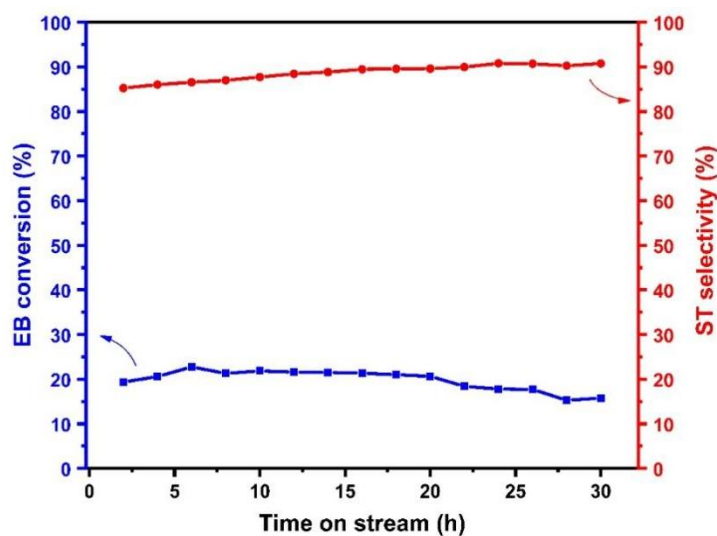


Fig. S4 The plots of x_{EB} and S_{ST} vs. time on stream of the commercial P_3N_5 for the DDH reaction of ethylbenzene to styrene. Reaction conditions: 50 mg catalyst, 600 °C, 2.8 vol% ethylbenzene in N_2 , 20 mL min⁻¹.

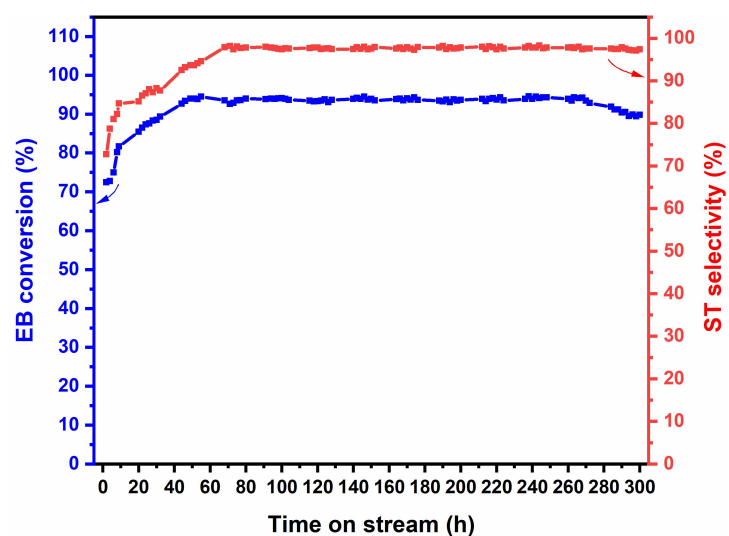


Fig. S5 The plots of x_{EB} and S_{ST} vs. time on stream of PNO-800/ Al_2O_3 catalyst. Reaction conditions: 2.0g catalyst (40-60 mesh A.S.T.M), 600 °C, 2.8 vol% ethylbenzene in N_2 , 20 mL min⁻¹.

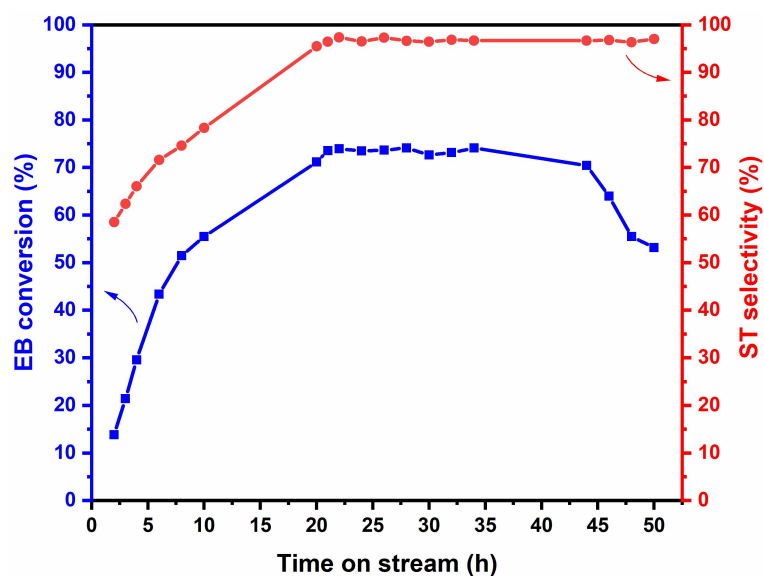


Fig. S6 The plots of x_{EB} and S_{ST} vs. time on stream of PNO-800/ SiO_2 catalyst. Reaction conditions: 2.0g catalyst (40-60 mesh A.S.T.M), 600 °C, 2.8 vol% ethylbenzene in N_2 , 20 mL min⁻¹.

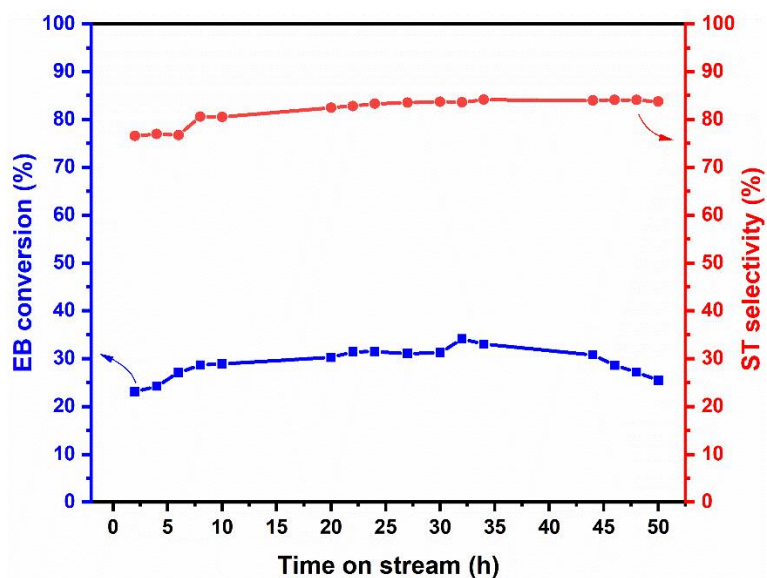


Fig. S7 The plots of x_{EB} and S_{ST} vs. time on stream of PNO-800/ $AlPO_4$ catalyst. Reaction conditions: 2.0g catalyst (40-60 mesh A.S.T.M), 600 °C, 2.8 vol% ethylbenzene in N_2 , 20 mL min⁻¹.

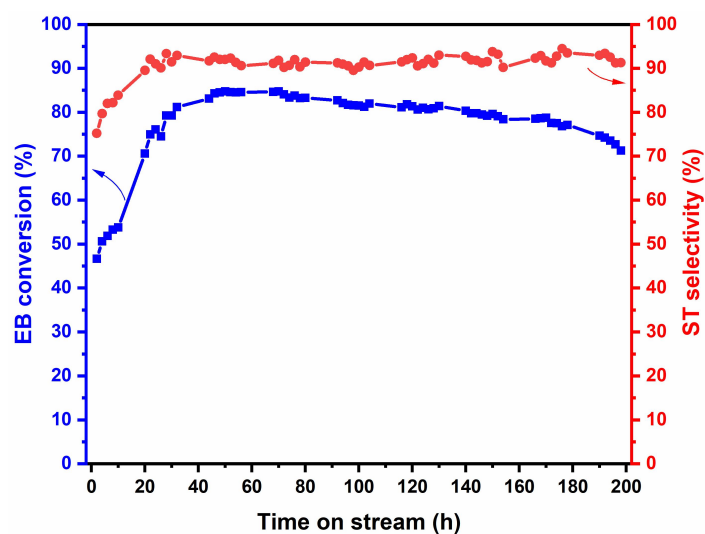


Fig. S8 The plots of x_{EB} and S_{ST} vs. time on stream of PNO-800/AlN catalyst. Reaction conditions: 2.0g catalyst (40-60 mesh A.S.T.M), 600 °C, 2.8 vol% ethylbenzene in N_2 , 20 mL min⁻¹.

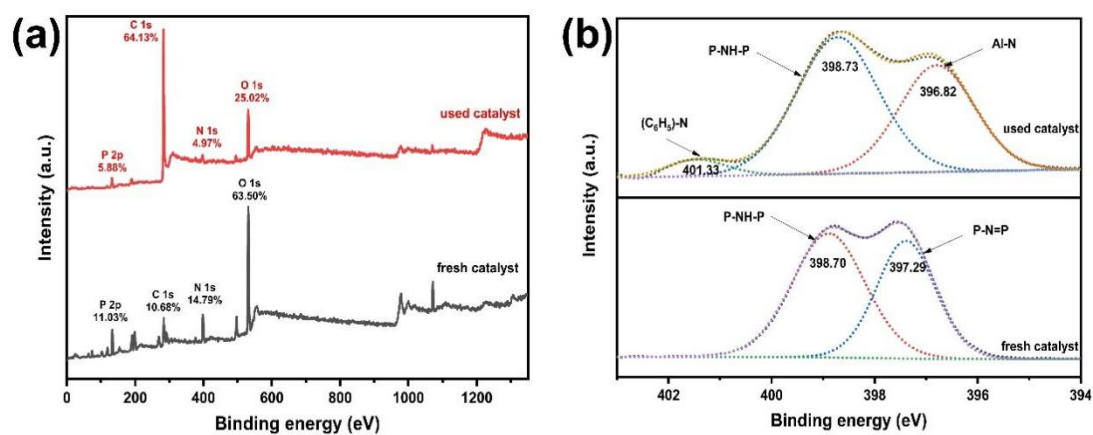


Fig. S9 XPS survey spectra (a), N 1s spectra (b) of PNO-800/Al₂O₃ before and after 300 h reaction.

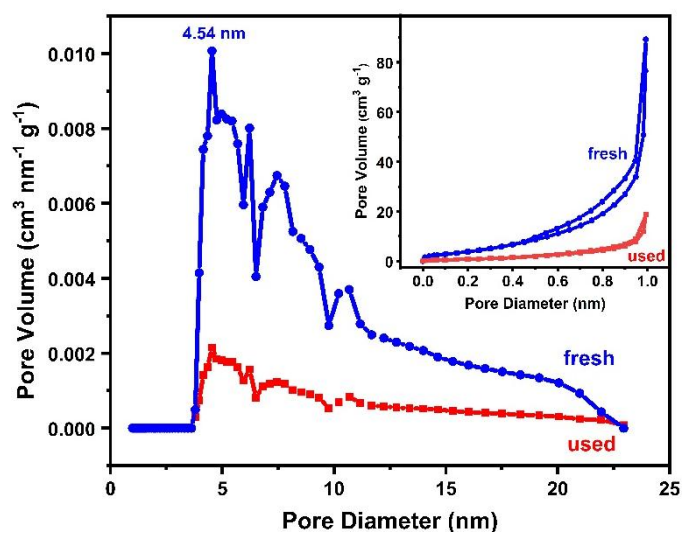


Fig. S10 N₂ adsorption-desorption isotherms and BJH pore-size distribution curves of PNO-800/Al₂O₃ before and after 300 h reaction.

Table S1 Comparison of various metal-free catalysts for DDH of ethylbenzene.

Catalyst	Reaction Condition	Catalytic Performance			Reference
		X_{EB} (%)	S_{ST} (%)	ST Yield (mmol g ⁻¹ h ⁻¹)	
PNO-800	50 mg cat, 600 °C, 2.8 vol% EB in 20 mL min ⁻¹ N ₂	52.86	98.24	15.59	---
PBN-1.2	50 mg cat, 600 °C, 2.8 vol% EB in 20 mL min ⁻¹ N ₂	61.85	93.5	17.45	Ref.11
N-CNT-AC	25 mg cat, 550 °C, 2.8 vol% EB in 10 mL min ⁻¹ Ar	---	94.7	2.46	Ref.29
H-ND	25 mg cat, 550 °C, 2.8 vol% EB in 10 mL min ⁻¹ Ar	---	95.8	4.8	Ref.30
ND-CN	25 mg cat, 550 °C, 2.8 vol% EB in 10 mL min ⁻¹ Ar	---	99.3	5.61	Ref.6
ND@NMC-700	150 mg cat, 550 °C, 2.8 vol% EB in 30 mL min ⁻¹ He	37.7	99.6	5.8	Ref.31
ND/CNT-SiC-ms-HN	30 mg cat, 550 °C, 2.8 vol% EB in 10 mL min ⁻¹ Ar	---	98.4	5.49	Ref.32
ND-CNT-SDS/SiC	25 mg cat, 550 °C, 2.8 vol% EB in 10 mL min ⁻¹ Ar	---	99.5	6.25	Ref.33
ND/CN-ms-o	25 mg cat, 550 °C, 2.8 vol% EB in 10 mL min ⁻¹ He	---	>98.2	7.06	Ref.34
ND@NMC/SiC	300 mg cat, 600 °C, 10.0 vol% EB in 30 mL min ⁻¹ He	38.3	96.7	9.9	Ref.35
NMCS-4-800	100 mg cat, 550 °C, 2.8 vol% EB in 10 mL min ⁻¹ He	---	90	2.0	Ref.36

Reference for Table S1:

[6] G. Ge, Z. Zhao. *Appl. Catal. A Gen.*, 2019, **571**, 82-88.[11] B. Lin, F. Xu, Y. Mei, Y. Liu, Y. Zou, Z. Liang, Y. Zhou, J. Diao, Y. Mao, H. Liu. *Catal. Sci. Technol.*, 2021, **11**, 5590-5597.[29] Z. Zhao, Y. Dai, G. Ge. *Catal. Sci. Technol.*, 2015, **5**, 1548-1557.[30] Z. Zhao, W. Li, Y. Dai, G. Ge, X. Guo, G. Wang. *ACS Sustain. Chem. Eng.*, 2015, **3**, 3355-3364.[31] Y. Liu, H. Ba, J. Luo, K. Wu, J. Nhut, D. Su, C. Pham-Huu. *Catal. Today.*, 2018, **301**, 38-47.[32] G. Ge, X. Wei, H. Guo, Z. Zhao. *Catal. Chin. Chem. Lett.*, 2023, **34**, 107808.[33] G. Ge, X. Wei, H. Guo, Z. Zhao. *Eur. J. Inorg. Chem.*, 2022, **26**, e202200232.[34] G. Ge, X. Guo, C. Song, Z. Zhao. *Catal. Sci. Technol.*, 2020, **10**, 1048-1055.[35] H. Ba, J. Luo, Y. Liu, C. Duong-Viet, G. Tuci, G. Giambastiani, J. Nhut, L. Nguyen-Dinh, O. Ersen, D. Su, C. Pham-Huu. *Appl. Catal. B Environ.*, 2017, **200**, 343-350.[36] J. Wang, H. Liu, J. Diao, X. Gu, H. Wang, J. Rong, B. Zong, D. Su, J. Mater. Chem. A, 2015, **3**, 2305-2313.**Table. S2** S_{BET} and pore structure analysis of fresh and used PNO-800/Al₂O₃.

Catalysts	S_{BET} (m ² g ⁻¹)	Micropore volume (cm ³ g ⁻¹)	Mesoporous volume (cm ³ g ⁻¹)	Total pore volume (cm ³ g ⁻¹)
fresh PNO-800/Al ₂ O ₃	15.139	0	0.142	0.142
used PNO-800/Al ₂ O ₃	3.291	0	0.030	0.030