

Supporting Information

Efficient Co/NSPC Catalyst for Selective Hydrogenation of Halonitrobenzenes and Mechanistic Insight

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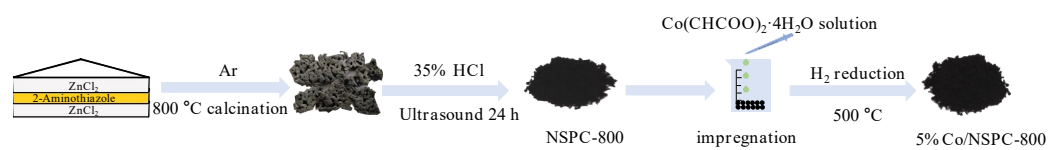
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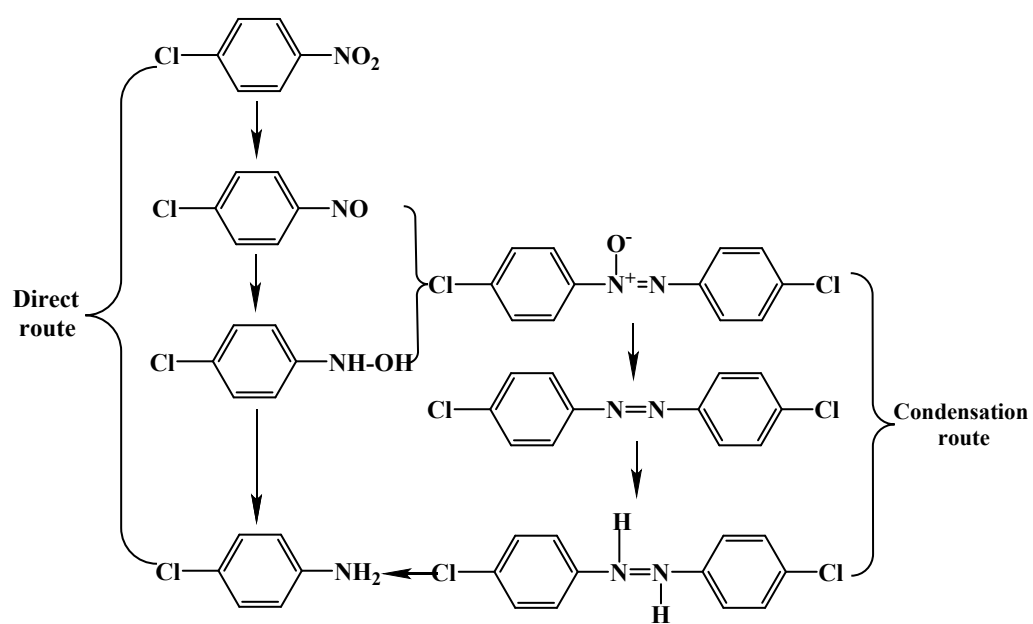
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Supporting Figures and Tables



Scheme S1. The schematic illustration of the preparation of 5%Co/NSPC-800.



Scheme S2. The possible reaction pathway for *p*-CNB hydrogenation.

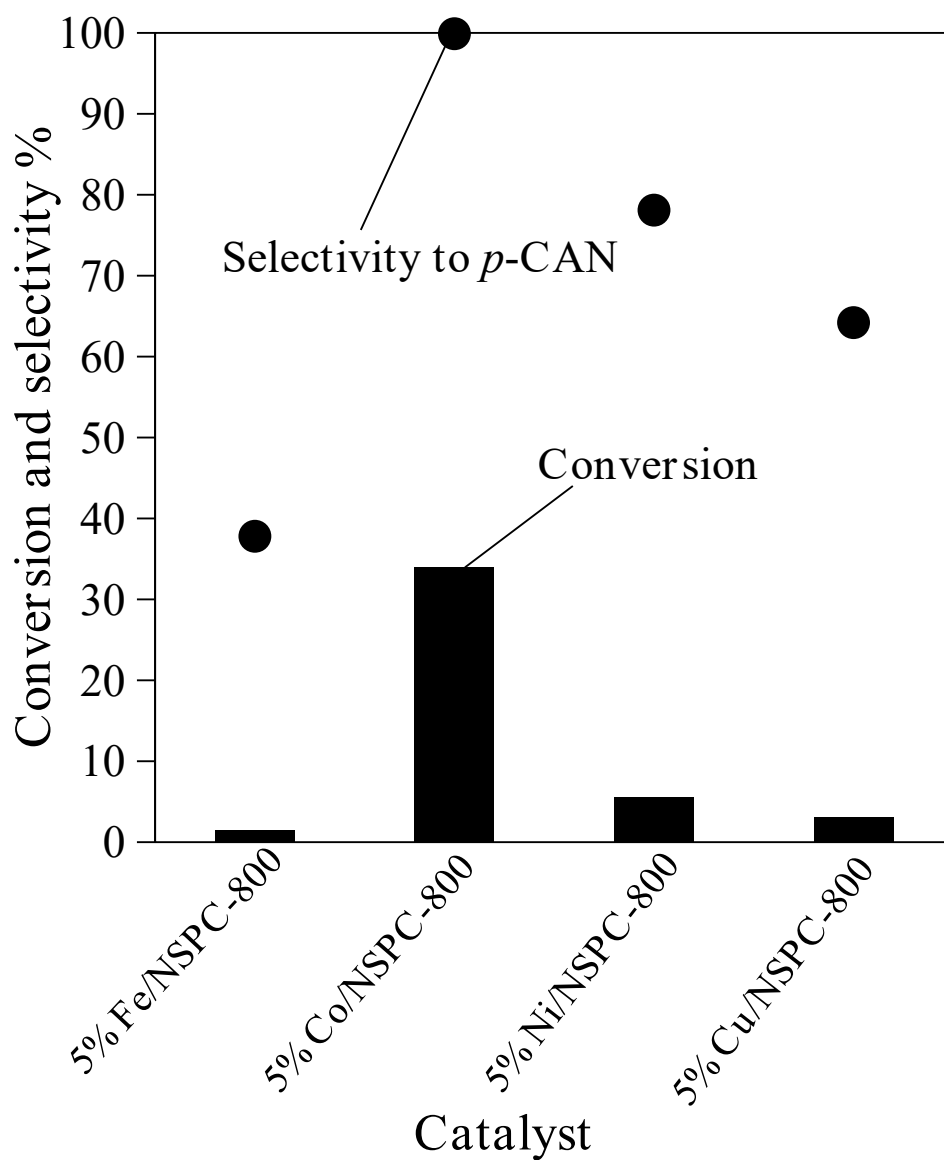


Figure S1. The effect of non-noble metal on the catalytic performance for *p*-CNB hydrogenation. Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa, temperature 353 K, reaction time 3 h.

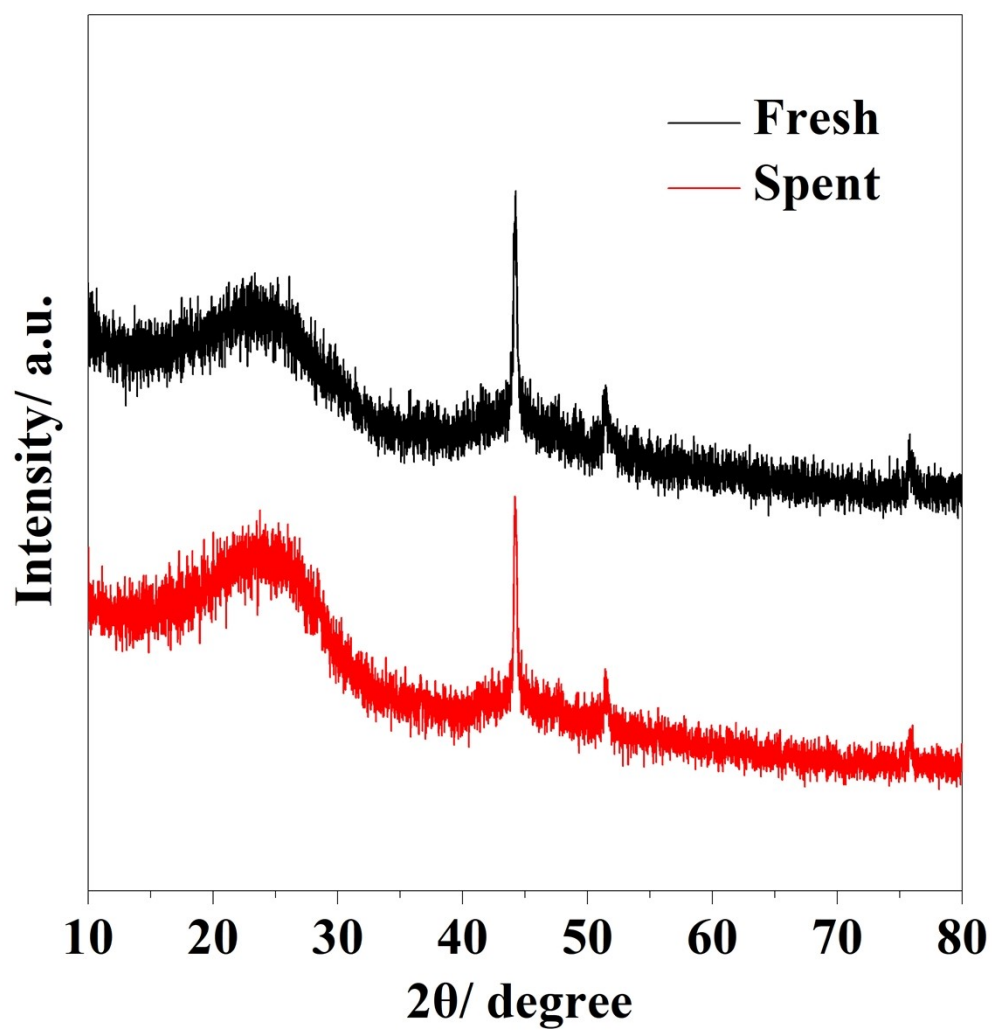


Figure S2. The XRD patterns of 5%Co/NSPC-800 before and after reaction.

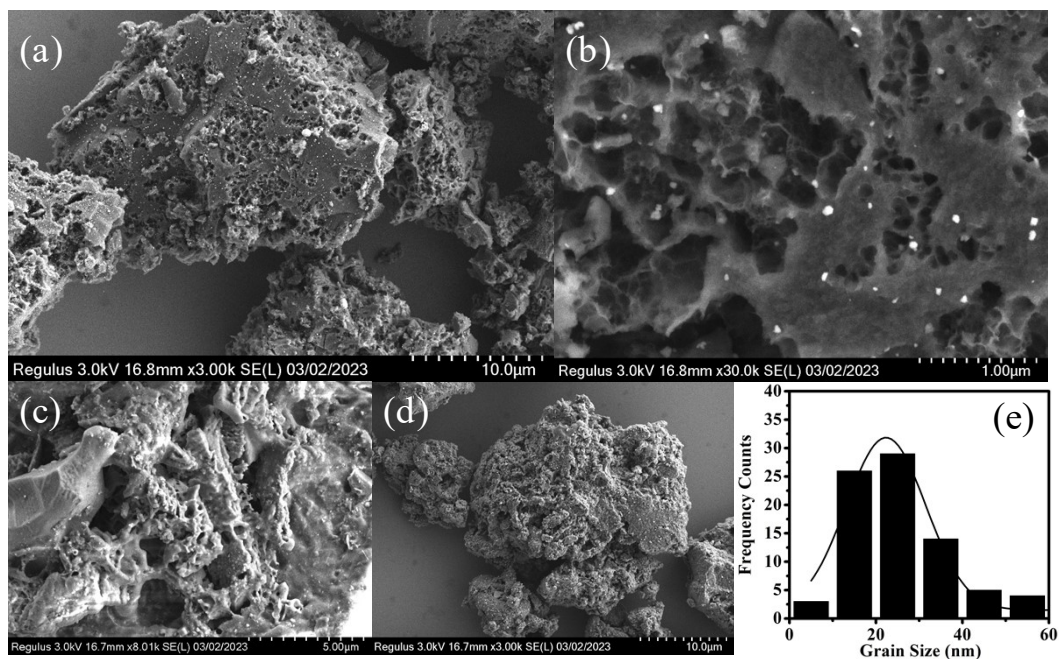


Figure S3. (a~b) The SEM images of 5%Co/NSPC-800, (c) 5%Co/NPC-800, (d) 5%Co/SPC-800 and (e) particle size distribution of 5%Co/NSPC-800.

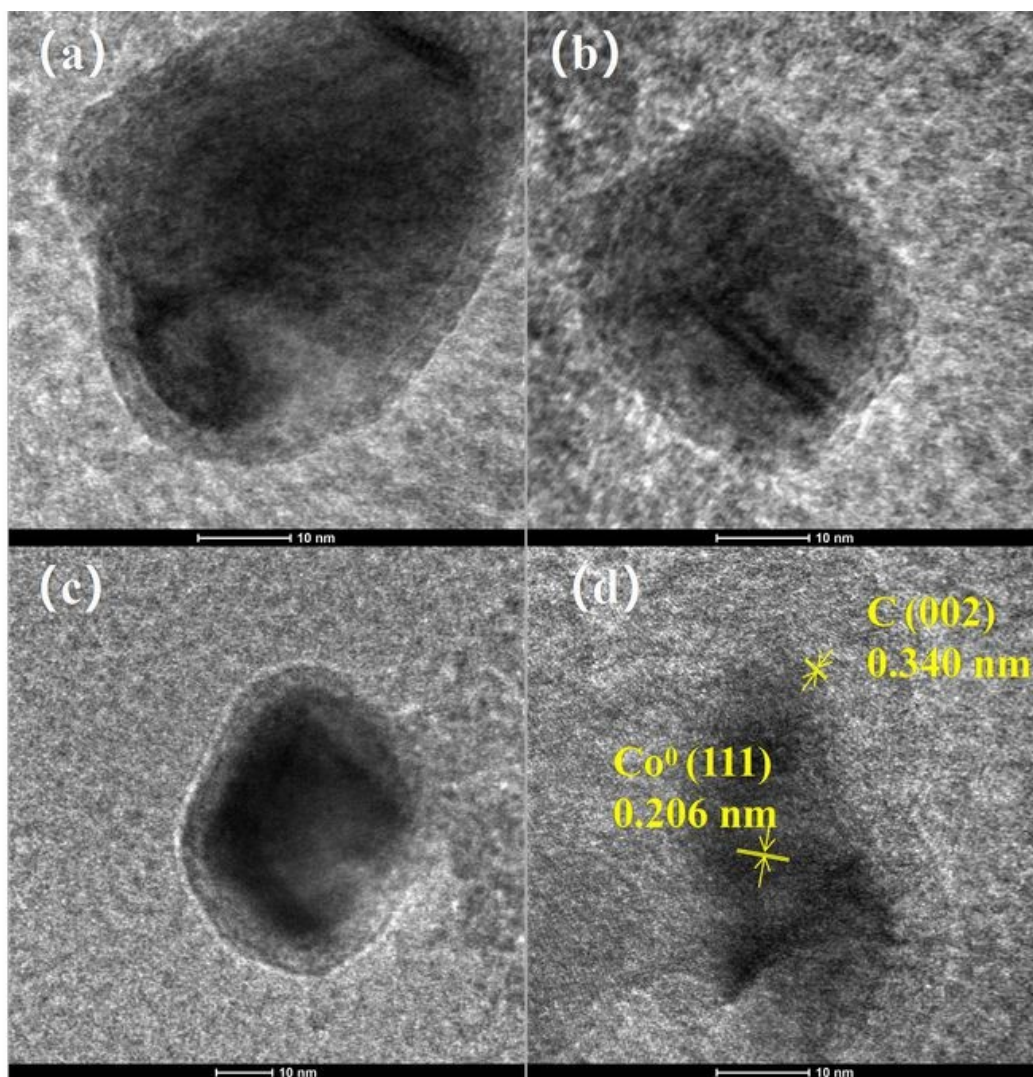


Figure S4. The HRTEM images of 5%Co/NSPC-800.

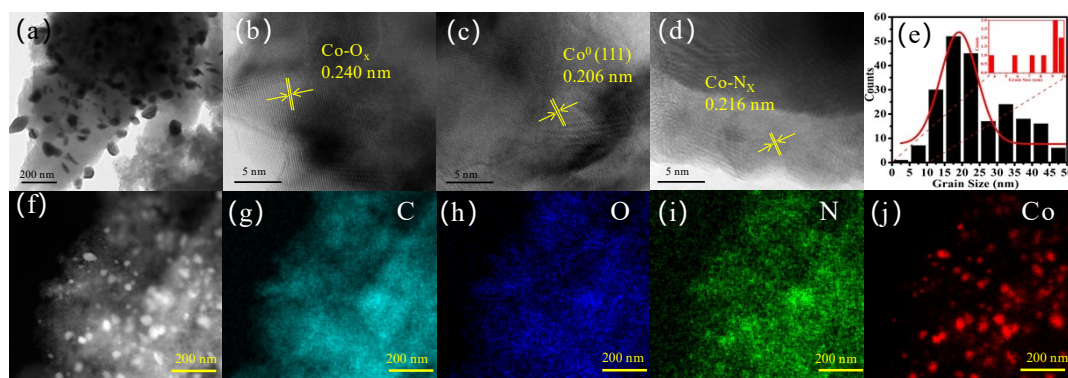


Figure S5. (a) The TEM images of 5%Co/NPC-800, (b~d) HRTEM images of 5%Co/NPC-800, (e) particle size distribution of 5%Co/NPC-800 and (f~j) EDX mapping of 5%Co/NPC-800, C, O, N and Co.

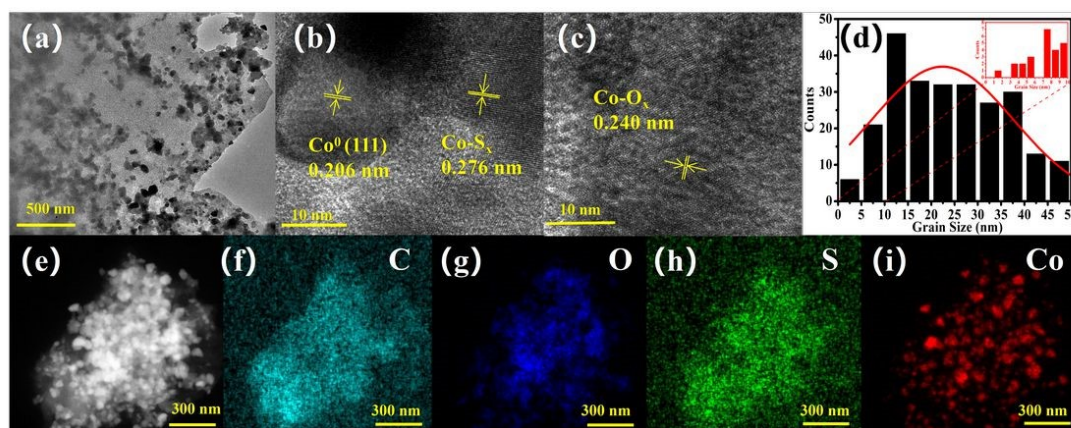


Figure S6. (a) The TEM images of 5%Co/SPC-800, (b~c) HRTEM images of 5%Co/SPC-800, (d) particle size distribution of 5%Co/SPC-800 and (e~i) EDX mapping of 5%Co/SPC-800, C, O, S and Co.

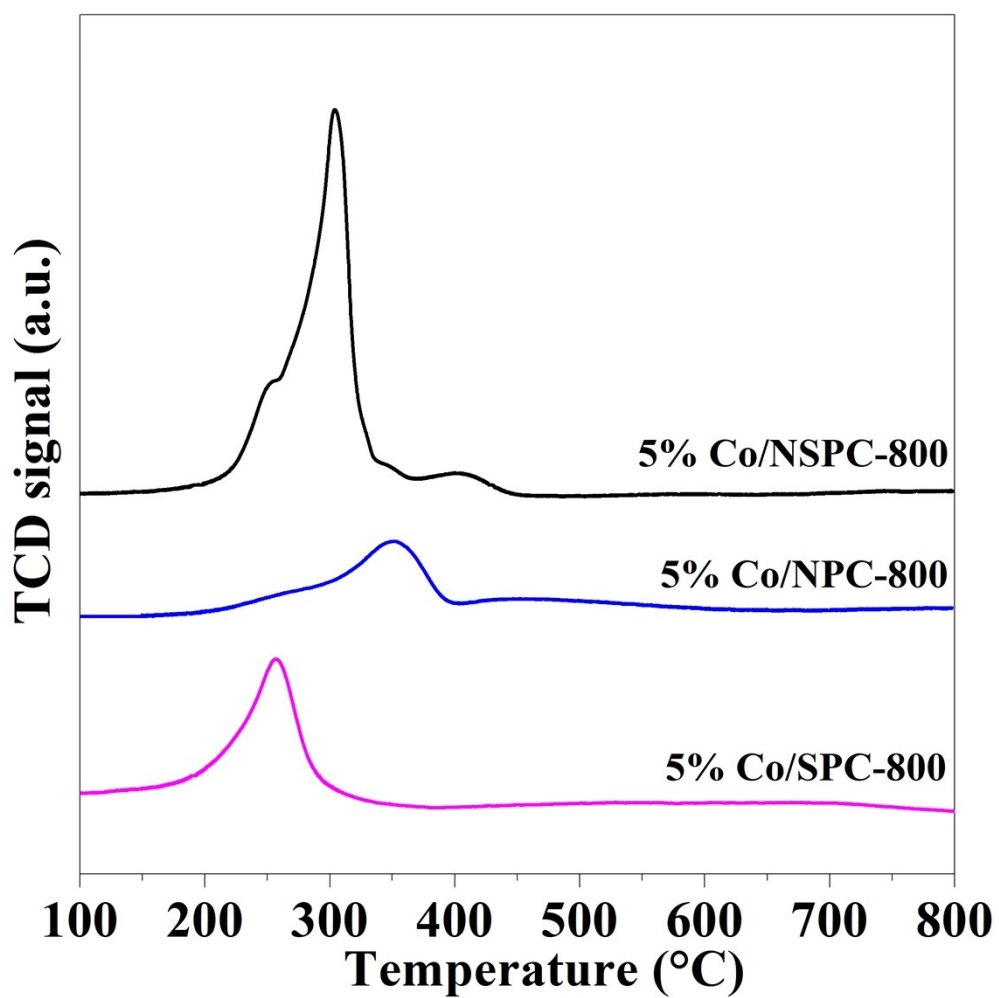


Figure S7. The H₂-TPR profile of 5%Co/NSPC-800, 5%Co/NPC-800 and 5%Co/SPC-800.

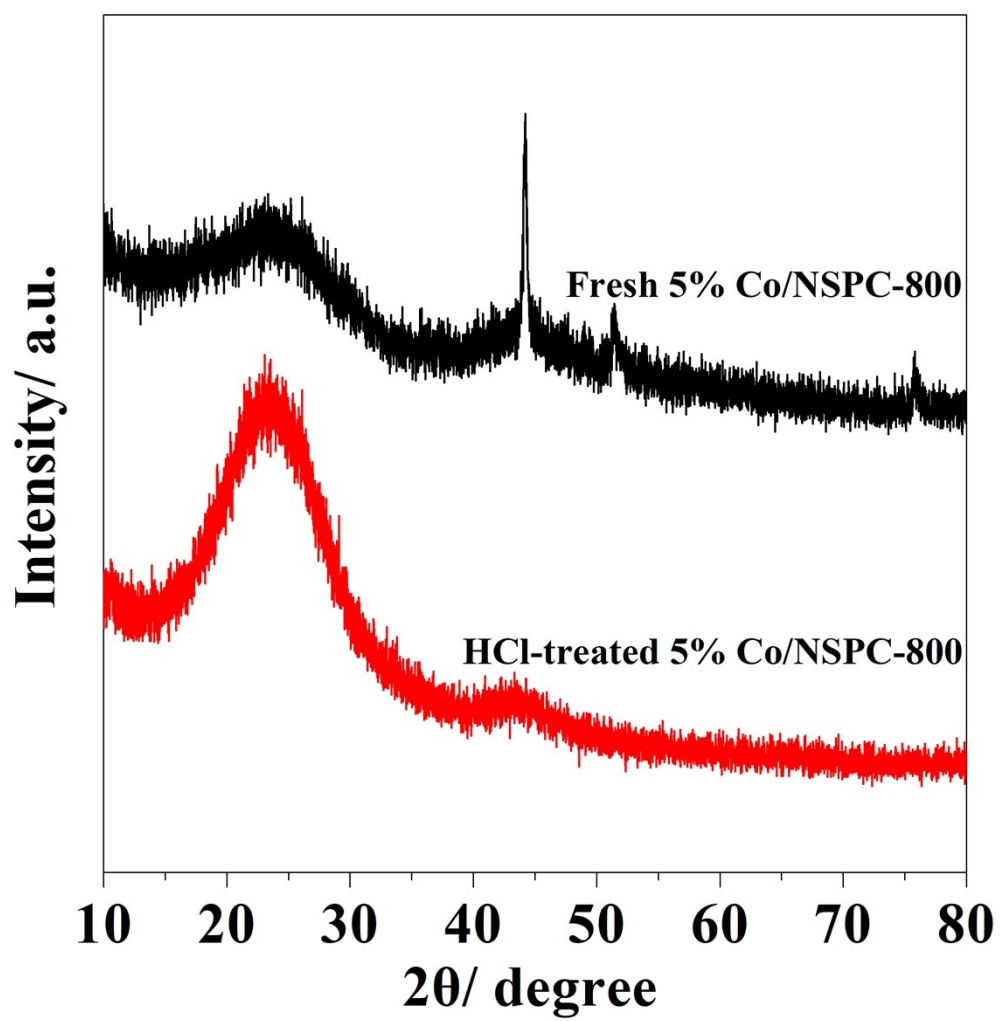


Figure S8. The XRD patterns of 5%Co/NSPC-800 before and after acid leaching.

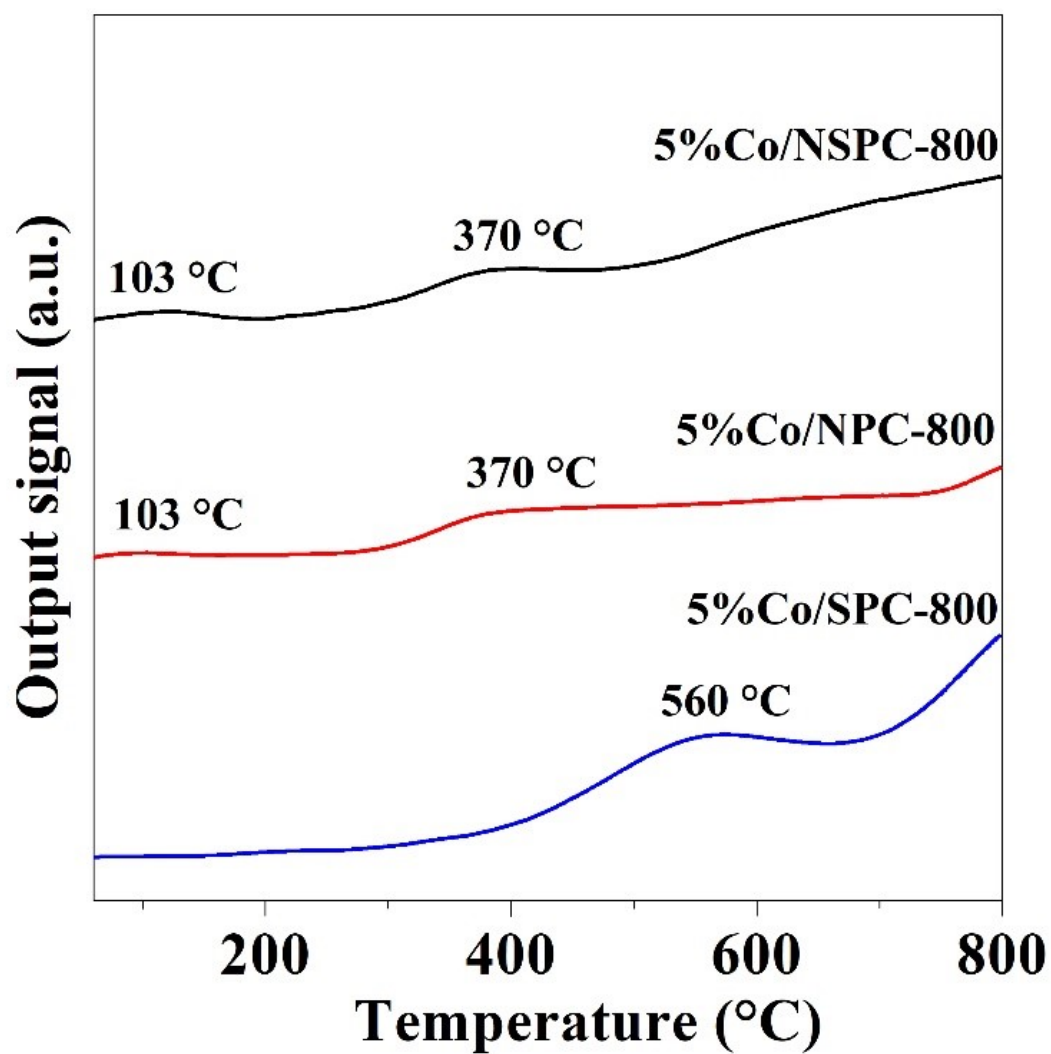


Figure S9. The H₂-TPD profile of 5%Co/NSPC-800, 5%Co/NPC-800 and 5%Co/SPC-800.

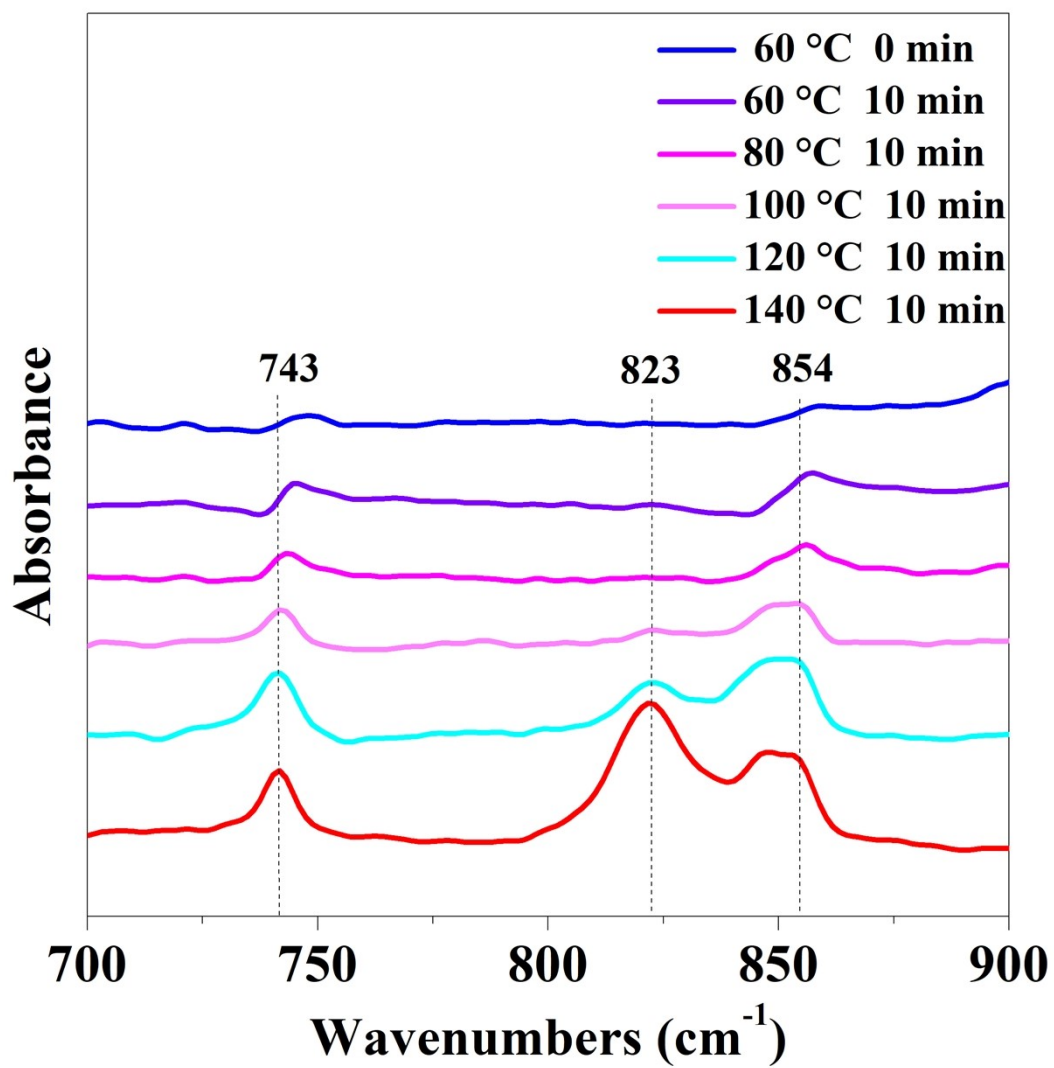


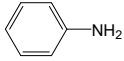
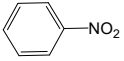
Figure S10. The *in-situ* DRIFTS of *p*-CNB over 5%/Co/NSPC-800 under 1 MPa H₂

and

60~140

°C.

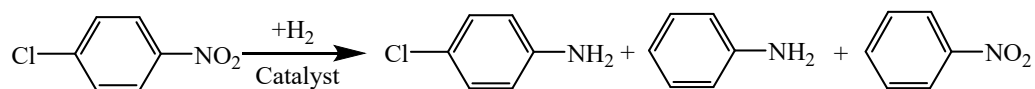
Table S1. The effect of support on the catalytic performance for *p*-CNB hydrogenation.

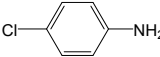
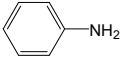
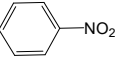
$ \begin{array}{c} \text{Cl}-\text{C}_6\text{H}_4-\text{NO}_2 \xrightarrow[\text{Catalyst}]{+\text{H}_2} \text{Cl}-\text{C}_6\text{H}_4-\text{NH}_2 + \text{C}_6\text{H}_5-\text{NH}_2 + \text{C}_6\text{H}_5-\text{NO}_2 \\ p\text{-CNB} \qquad\qquad\qquad p\text{-CAN} \qquad\qquad\qquad \text{AN} \qquad\qquad\qquad \text{NB} \end{array} $					
Entry	Catalyst	Conversion /%	Selectivity /%		
					
1	SPC-800	0.71	6.8	93.2	<0.1
2	NPC-800	0.88	74.1	<0.1	25.9
3	NSPC-800	2.2	>99.9	<0.1	<0.1
4	5%Co/NSPC-800	34.0	>99.9	<0.1	<0.1
5	5%Co/NPC-800	17.8	86.9	<0.1	13.1
6	5%Co/SPC-800	2.1	64.7	<0.1	35.3
7	5%Co/PC-800	2.7	78.5	<0.1	21.5
8	5%Co/SiO ₂	5.2	3.1	79.4	17.5

Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa,

temperature 353 K, reaction time 3 h.

Table S2. The effect of Co loading amount on catalytic performance for *p*-CNB hydrogenation.



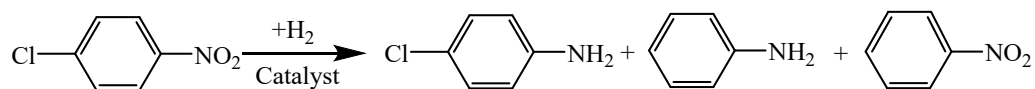
Entry	Catalyst	Conversion /%	Selectivity /%		
					
1	1%Co/NSPC-800	10.3	85.3	<0.1	14.7
2	2%Co/NSPC-800	20.7	92.7	<0.1	7.3
3	3.5%Co/NSPC-800	27.5	97.6	<0.1	2.4
4	5%Co/NSPC-800	34.0	>99.9	<0.1	<0.1
5	7.5%Co/NSPC-800	15.8	99.1	0.9	<0.1

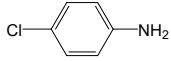
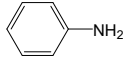
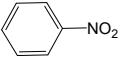
Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa,

temperature 353 K, reaction time 3 h.

Table S3. The effect of different metal on catalytic performance for *p*-CNB

hydrogenation.



Entry	Catalyst	Conversion /%	Selectivity /%		
					
1	5%Fe/NSPC-800	1.5	37.8	<0.1	61.2
2	5%Co/NSPC-800	34.0	>99.9	<0.1	<0.1
3	5%Ni/NSPC-800	5.6	78.1	0.9	21.0
4	5%Cu/NSPC-800	3.1	64.2	<0.1	35.8

Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa,

temperature 353 K, reaction time 3 h.

Table S4. The time course over 5%Co/NSPC-800 for *p*-CNB hydrogenation.

$$\text{Cl-C}_6\text{H}_4\text{-NO}_2 \xrightarrow[\text{Catalyst}]{+\text{H}_2} \text{Cl-C}_6\text{H}_4\text{-NH}_2 + \text{C}_6\text{H}_5\text{-NH}_2 + \text{C}_6\text{H}_5\text{-NO}_2$$

Entry	React time/ h	Conversion /%	Selectivity /%		
1	1.5	17.7	>99.9	<0.1	<0.1
2	3	34.0	>99.9	<0.1	<0.1
3	6	59.8	>99.9	<0.1	<0.1
4	9	82.1	>99.9	<0.1	<0.1
5	12	100	>99.9	<0.1	<0.1
6	15	100	>99.9	<0.1	<0.1

Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa,

temperature 353 K, reaction time 0-15 h.

Table S5. The time course over 5%Co/NPC-800 for *p*-CNB hydrogenation.

Entry	React time/ h	Conversion /%	Selectivity /%		
1	1.5	10.0	84.7	<0.1	15.3
2	3	17.8	86.9	<0.1	13.1
3	6	33.5	91.4	<0.1	8.6
4	9	46.8	93.4	<0.1	6.6
5	12	60.8	94.8	<0.1	5.2
6	15	74.6	96.6	<0.1	3.2
7	18	86.5	97.3	<0.1	2.7
8	21	100	97.3	<0.1	2.7
9	24	100	93.5	3.8	2.7

Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa,

temperature 353 K, reaction time 0-24 h.

Table S6. The elemental content of samples obtained by EA and ICP-OES.

Entry	Catalyst	EA				ICP-OES
		N(wt%)	C(wt%)	H(wt%)	S(wt%)	Co(wt%)
1	2%Co/NSPC-800	-	-	-	-	1.1
2	3.5%Co/NSPC-800	-	-	-	-	2.5
3	5%Co/NSPC-800	6.7	60.3	2.1	1.8	3.5
4	7.5%Co/NSPC-800	-	-	-	-	4.8
5	5%Co/NPC-800	4.7	71.3	1.9	0	3.0
6	5%Co/SPC-800	0	57.1	1.8	3.0	1.4

Table S7. The particle size and dispersion of catalysts.

Entry	Catalyst	Reduction temp./K	Particle size /nm		^b Dispersion Co /%
			XRD	TEM	
1	2%Co/NSPC-800	773	26.5	-	-
2	3.5%Co/NSPC-800	773	26.8	-	-
3	5%Co/NSPC-800	773	24.5	22.4	6.0
4	7.5%Co/NSPC-800	773	25.7	-	-
5	5%Co/NPC-800	773	25.7	25.0	5.3
6	5%Co/SPC-800	773	26.0	25.2	4.3
7	^a 5%Co/NSPC-800	773	26.9	-	-

^a After the reaction^b Determination by CO adsorption

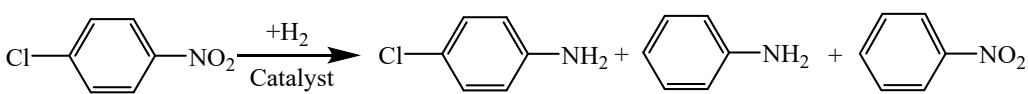
Table S8. The TOF and TOF_s values of reported effective Co-based catalysts.

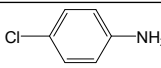
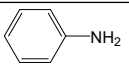
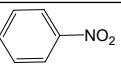
Catalyst	P _{H2} /MPa	T /K	TOF _s ^a /h ⁻¹	TOF /h ⁻¹	Ref.
Co-NSPC-N	1.5	393	123	7 ^b	[1]
Co-NSPC-C	1.5	393	180	6 ^b	[1]
Co-NSPC-S	1.5	393	154	3 ^b	[1]
Co-NSPC-Cl	1.5	393	29	2 ^b	[1]
Co ₁ Cu _{0.30} @CN/SiO ₂	2	393	160	13 ^b	[2]
Co@mesoNC	3	383	42	42	[3]
Co@CN-400	1	333	59	5	[4]
Co-N-C	3	353	36	36	[5]
Co-PC	1	353	108	10	[6]
Co-Phen	1	353	56	3	[6]
Co/N-C-600	1	373	45	5	[7]
Co-NSPC-800	3	353	84	84	[7]
Co/NMC-800	1	353	275	275	[8]
Co800NH ₃	3	353	152	152	[9]
Co-N _x /C-800-ATc	0.35	383	392	392	[10]
Co ₁ @NC (SBA)	1	413	22	22	[11]
Co SAs/NC-800	0.5	393	110	110	[12]
Co SAs/NC	3	383	77	77	[13]
Co SAs/NHPCN	2	373	44	44	[14]
5%Co/SPC-800	3	353	44	2	This work
5%Co/NPC-800	3	353	140	7	This work
5%Co/NSPC-800	3	353	203	12	This work

^a Calculated based on the amount of surface metal amount. TOF_s (h⁻¹) = (Converted *p*-CNB amount (mol))/(surface metal amount (mol))/(Time (h)). surface metal amount (mol) = (total metal amount (mol))·(1/particle size (nm)). The dispersion of single atom catalyst is considered to be 100%.

^b Calculated based on the amount of total metal amount. TOF (h⁻¹) = (Converted *p*-CNB amount (mol))/(total metal amount (mol))/(Time (h))

Table S9. The reusability experiment over 5%Co/NSPC-800 for *p*-CNB hydrogenation.

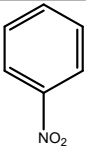
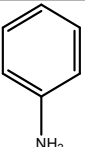
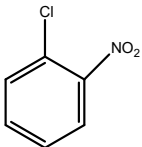
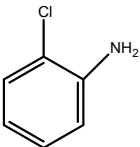
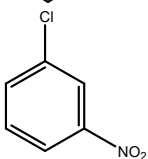
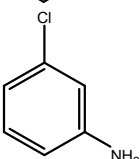
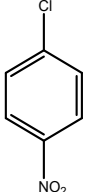
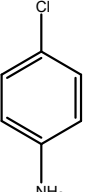
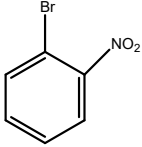
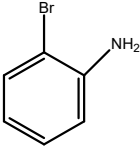
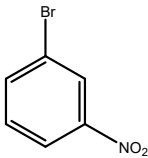
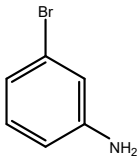
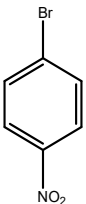
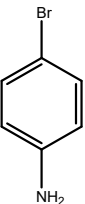
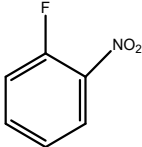
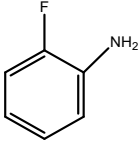
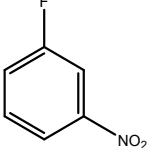
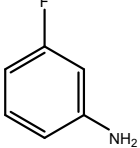


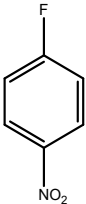
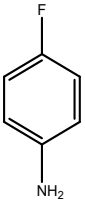
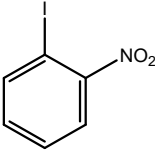
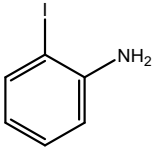
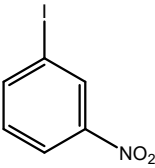
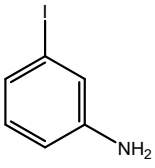
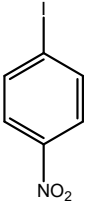
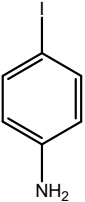
Usage time	Conversion /%	Selectivity /%		
				
1	16.6	>99.9	<0.1	<0.1
2	16.0	>99.9	<0.1	<0.1
3	14.9	>99.9	<0.1	<0.1
4	14.6	>99.9	<0.1	<0.1

Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa,

temperature 353 K, reaction time 3 h.

Table S10.The substrate scope over 5%Co/NSPC-800.

Entry	reactant	Product	t/h	Conversion /%	Selectivity /%
1			3	40.2	>99.9
			12	100	>99.9
2			3	37.8	>99.9
			12	100	>99.9
3			3	38.1	>99.9
			12	100	>99.9
4			3	34.0	>99.9
			12	100	>99.9
5			3	33.8	>99.9
			12	100	>99.9
6			3	34.9	>99.9
			12	100	>99.9
7			3	35.2	>99.9
			12	100	>99.9
8			3	38.5	>99.9
			12	100	>99.9
9			3	35.8	>99.9
			12	100	>99.9

10			3	38.6	>99.9
			12	100	>99.9
11			3	17.4	93.0
			20	100	96.2
12			3	15.8	98.7
			20	100	96.3
13			3	16.0	97.5
			20	100	97.3

Reaction conditions: HNBS 3.2 mmol, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa, temperature 353 K, reaction time 3, 12 or 20 h.

Table S11. The relationship between Co⁰ and Co-N_x XPS peak area and conversion.

Entry	Catalyst	Conversion /%	Peak area /(eV·Counts)	
			Co-N _x	Co ⁰
1	1%Co/NSPC-800	10.3	750	50
2	2%Co/NSPC-800	20.7	1370	100
3	3.5%Co/NSPC-800	27.5	1785	195
4	5%Co/NSPC-800	34.0	1900	400
5	7.5%Co/NSPC-800	15.8	1050	300
6	5%Co/NPC-800	17.8	1000	400
7	5%Co/SPC-800	2.1	0	280

Table S12. The binding energy and distribution of surface cobalt atomic chemical state based on XPS spectra.

Catalyst	Co 2p _{3/2} (eV)				Relative percentage (%)			
	Co ⁰	Co-O _x	Co-N _x	Co-S _x	Co ⁰	Co-O _x	Co-N _x	Co-S _x
1%Co/NSPC-800	778.6	780.5	782.5	778.1	2.0	66.0	31.2	0.8
2%Co/NSPC-800	778.6	780.5	782.5	778.1	2.4	64.3	32.1	1.2
3.5%Co/NSPC-800	778.6	780.5	782.5	778.1	4.0	58.8	36.2	1.0
5%Co/NSPC-800	778.6	780.5	782.5	778.1	7.5	54.9	35.7	1.9
7.5%Co/NSPC-800	778.6	780.5	782.5	778.1	7.6	64.6	26.6	1.2
5%Co/NPC-800	778.6	780.5	782.5	-	11.8	58.8	29.4	0
5%Co/SPC-800	778.6	780.5	-	778.1	12.6	80.7	0	6.7

Table S13. The Co content of 5%Co/NSPC-800, 5%Co/NPC-800 and 5%Co/SPC-800 measured by ICP-OES and XPS.

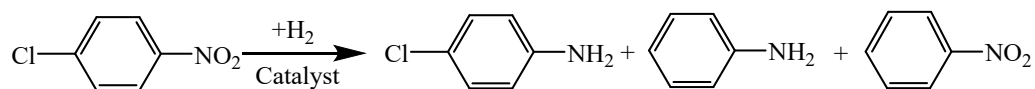
Entry	Catalyst	Co content (wt. %)	
		ICP-OES	XPS
1	5%Co/NSPC-800	3.5	0.9
2	5%Co/NPC-800	3.0	0.8
3	5%Co/SPC-800	1.4	0.5

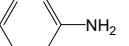
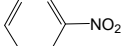
Table S14.The EXAFS fits parameters at the Co *K*-edge of various samples.

Catalyst	Shells	CN ^a	$R^b / 10^{-1} \text{ nm}$	$\sigma^c / 10^{-1} \text{ nm}$	$\Delta E_0^d / \text{eV}$	$R_f^e / \%$
5%Co/NSPC-800	Co-Co	11.0	2.47	0.06	-2.7	0.028
	Co-N/O	2.0	2.04	0.06	3.5	
	Co-S	1.8	2.35	0.06	-2.7	
5%Co/NPC-800	Co-Co	11.5	2.49	0.07	-0.8	0.026
	Co-N/O	1.2	2.00	0.08	-7.6	
5%Co/SPC-800	Co-Co	11.6	2.50	0.06	-1.4	0.028
	Co-S	3.0	2.35	0.06	-0.7	
Co foil	Co-Co	12.0	2.50	0.07	0.5	0.003
CoPc	Co-N	4.0	1.91	0.06	-2.7	0.007
CoS	Co-S	6.0	2.28	0.06	1.8	0.002

^aCoordination number. ^bBond distance. ^cDebye-Waller factor. ^dDifference in the origin of photoelectron energy between the reference and the sample. ^eResidual factor. Fourier filtering range: 0.114-0.310 nm.

Table S15. The kinetic studies on the effect of *p*-CNB concentration over 5%Co/NSPC-800.

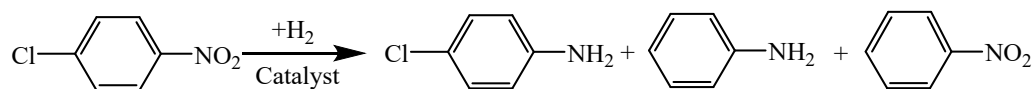


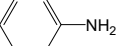
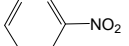
Entry	<i>p</i> -CNB mass /g	Conversion /%	Selectivity /%		
					
1	0.25	30.4	>99.9	<0.1	<0.1
2	0.5	17.0	>99.9	<0.1	<0.1
3	1	9.2	>99.9	<0.1	<0.1

Reaction conditions: *p*-CNB 0.25-1 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3

MPa, temperature 353 K, reaction time 1.5 h.

Table S16. The kinetic studies on the effect of *p*-CNB concentration over 5%Co/NPC-800.

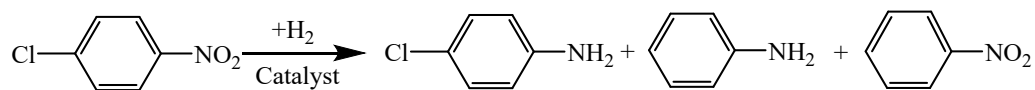


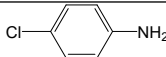
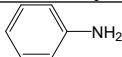
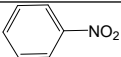
Entry	<i>p</i> -CNB mass /g	Conversion /%	Selectivity /%		
					
1	0.25	31.1	80.0	7.5	12.5
2	0.5	17.8	86.9	<0.1	13.1
3	1	10.0	84.5	5.8	9.7

Reaction conditions: *p*-CNB 0.25-1 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3

MPa, temperature 353 K, reaction time 3 h.

Table S17. The kinetic studies on the effect of H₂ pressure over 5%Co/NSPC-800.

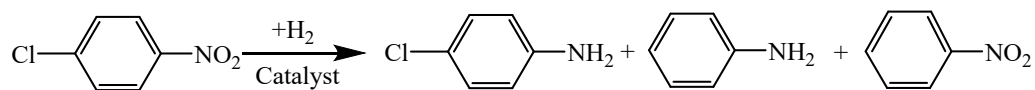


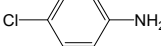
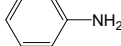
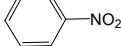
Entry	P _{H2} /MPa	Conversion /%	Selectivity /%		
					
1	0.5	4.2	>99.9	<0.1	<0.1
2	1.5	11.4	>99.9	<0.1	<0.1
3	3	34.0	>99.9	<0.1	<0.1

Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 0.5-3

MPa, temperature 353 K, reaction time 3 h.

Table S18. The kinetic studies on the effect of H₂ pressure over 5%Co/NPC-800.

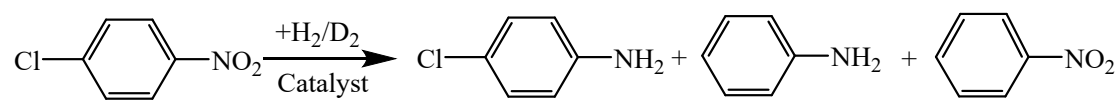


Entry	P _{H2} /MPa	Conversion /%	Selectivity /%		
					
1	0.5	2.8	>99.9	<0.1	<0.1
2	1.5	11.2	92.7	<0.1	7.3
3	3	17.8	86.9	<0.1	13.1

Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 0.5-3

MPa, temperature 353 K, reaction time 3 h.

Table S19. The H₂-D₂ exchange kinetics experiment result over 5%Co/NPC-800.

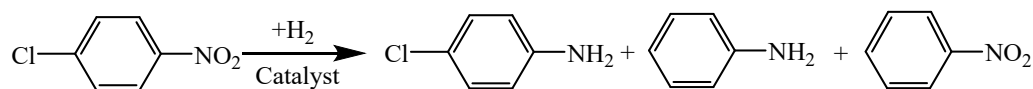


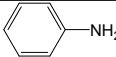
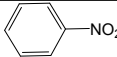
Entry	Reaction time	Conversion	
		H ₂ atmosphere	D ₂ atmosphere
1	0	0	0
2	1.5	0.055	0.012
3	3	0.117	0.025

Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, pressure 1.5 MPa,

temperature 353 K, reaction time 0-3 h.

Table S20. The effect of reaction temperature on the catalytic performance over 5%Co/NSPC-800.



Entry	Temp. /K	Conversion /%	Selectivity /%		
					
1	333	14.2	>99.9	<0.1	<0.1
2	353	34.0	>99.9	<0.1	<0.1
3 ^a	373	25.0	>99.9	<0.1	<0.1

Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa, temperature 333-373 K, reaction time 3 h.

^a reaction time 1.5 h.

Table S21. The effect of reaction temperature on the catalytic performance over 5%Co/NPC-800.

$$\text{Cl-C}_6\text{H}_4\text{-NO}_2 \xrightarrow[\text{Catalyst}]{+\text{H}_2} \text{Cl-C}_6\text{H}_4\text{-NH}_2 + \text{C}_6\text{H}_5\text{-NH}_2 + \text{C}_6\text{H}_5\text{-NO}_2$$

Entry	Temp. /K	Conversion /%	Selectivity /%		
1	333	7.5	88.1	<0.1	11.9
2	353	17.8	86.9	<0.1	13.1
3	373	49.7	48.7	51.3	<0.1

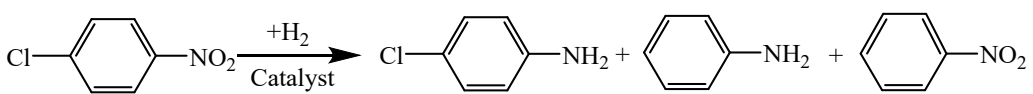
Reaction conditions: *p*-CNB 0.5 g, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa,

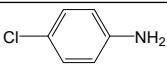
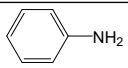
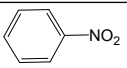
temperature 333-373 K, reaction time 3 h.

Table S22. The activation energy values of reported effective transition metal catalysts.

Catalyst	P _{H2} /MPa	T /K	E_a /KJ·mol ⁻¹	Ref.
Co-NSPC-800	3	333-373	50	[7]
Co SAs/NHPCN	2	363-383	49	[14]
Co ₁ /NC	3	363-403	51	[15]
Co ₁ /NPC	3	363-403	22	[15]
Co/NMC-800	1	343-383	28	[16]
2Co-1Zn@NC-800	1	323-353	63	[17]
5%Co/NPC-800	3	333-373	48	This work
5%Co/NSPC-800	3	333-373	33	This work

Table S23. The hydrogenation of *p*-CNB over 5%Co/NSPC-800 under different reaction conditions.



Entry	Conversion /%	Selectivity /%		
				
1	33.8	>99.9	<0.1	<0.1
2 ^a	20.8	90.6	<0.1	9.4
3 ^b	19.7	90.5	<0.1	9.5
4 ^c	15.2	84.4	<0.1	15.6
5 ^d	15.0	81.7	<0.1	18.3
6 ^e	12.4	51.1	28.7	20.2

Reaction conditions: halogenated nitrobenzene 3.2 mmol, catalyst 50 mg, methanol 30 mL, H₂ pressure 3 MPa, temperature 353 K, reaction time 3 h.

^a 0.03 mmol KSCN as catalyst inhibitor with SCN⁻ to cobalt molar ratio of 1.

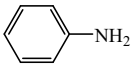
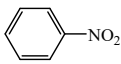
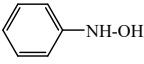
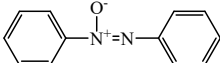
^b 0.25 mmol KSCN as catalyst inhibitor with SCN⁻ to cobalt molar ratio of 8.5.

^c 0.03 mmol EDTA as catalyst inhibitor with EDTA to cobalt molar ratio of 1.

^d 0.25 mmol EDTA as catalyst inhibitor with EDTA to cobalt molar ratio of 8.5.

^e The catalyst was treated with 35% hydrochloric acid for 12 h.

Table S24. The experimental results of hydrogenation of various intermediates.

Entry	Reactant	Conversion /%	Selectivity /%
			
1		40.2	>99.9
2		0	0
3		22.0	>99.9

Reaction conditions: reactant 3.2 mmol, 5%Co/NSPC-800 50 mg, methanol 30 mL, H₂ pressure 3 MPa, temperature 353 K, reaction time 3 h.

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