# Synthesis of π-extended and bowl-shaped sumanene-ferrocene conjugates and their application in highly selective and sensitive cesium cations electrochemical sensors

# ELECTRONIC SUPPLEMENTARY INFORMATION (ESI)

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# 1. Experimental section

# 1.1. Materials and methods

Chemical reagents and solvents for the synthesis were commercially purchased and purified according to the standard methods, if necessary. Thin layer chromatography (TLC) and preparative thin layer chromatography (PTLC) were performed using Merck Silica gel 60 F254 plates.

The NMR experiments were carried out using a Varian VNMRS 500 MHz spectrometer (<sup>1</sup>H NMR at 500 MHz, <sup>13</sup>C{<sup>1</sup>H} NMR at 125 MHz) equipped with a multinuclear z-gradient inverse probe head. The spectra were recorded at 25 °C and standard 5 mm NMR tubes were used. <sup>1</sup>H chemical shifts ( $\delta$ ) were reported in parts per million (ppm) relative to the solvent signal, *i.e.*, CDCl<sub>3</sub>:  $\delta_{H}$  (residual CHCl<sub>3</sub>) 7.26 ppm,  $\delta_{C}$  (residual CHCl<sub>3</sub>) 77.2 ppm, THF-*d*<sub>8</sub>:  $\delta_{H}$  (residual THF) 3.58 ppm, C<sub>6</sub>D<sub>6</sub>:  $\delta_{H}$  (residual benzene) 7.16 ppm, DMSO-*d*<sub>6</sub>:  $\delta_{H}$  (residual DMSO) 2.50 ppm,  $\delta_{C}$  (residual DMSO) 39.5 ppm. NMR spectra were analyzed with the MestReNova v12.0 software (Mestrelab Research S.L).

ESI-HRMS (TOF) measurements were performed with a Q-Exactive ThermoScientific spectrometer.

UV-vis measurements were performed with the PerkinElmer VWR 1600-PC spectrometer, at room temperature in quartz cuvette of 1 cm length of optical window. For the UV-Vis measurements, the wavelengths for the absorption maxima  $\lambda_{max}$  were reported in nm. Spectrofluorimetric analyzes were performed with a Hitachi F-7100 fluorescence spectrophotometer with the spectral resolution of 1 nm, the wavelengths for the emission maxima were reported in nm.

# 1.2. Synthesis of sumanene (1)

Sumanene (1) was synthesized following a literature procedure.<sup>1</sup>

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm),  $\delta_{\rm H}$  7.10 (s, 6H), 4.72 (d, <sup>2</sup>*J*<sub>H-H</sub> = 18.1 Hz, 3H), 3.42 (d, <sup>2</sup>*J*<sub>H-H</sub> = 18.1 Hz, 3H).

# 1.3. Synthesis of ferrocenemethylazide (9)

Ferrocenemethylazide (9) was synthesized from ferrocenemethylalcohol following a literature procedure.<sup>2</sup>

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm),  $\delta_{H}$  4.24-4.23 (t-like m, <sup>3</sup>*J*<sub>H-H</sub> = 1.8 Hz, 2H), 4.21-4.20 (t-like m, <sup>3</sup>*J*<sub>H-H</sub> = 1.8 Hz, 2H), 4.18 (s, 5H), 4.11 (s, 2H).

# 1.4. Synthesis of 1,3,5-tris(4-iodophenyl)benzene (13)

1,3,5-Tris(4-iodophenyl)benzene (**13**) was synthesized by means of the thionyl chloride catalyzed trimerization of 4-iodoacetophenone following a literature procedure.<sup>3</sup>

 $^1H$  NMR (CDCl\_3, 500 MHz, ppm),  $\delta_{\rm H}$  7.82-7.80 (m, 6H), 7.68 (s, 3H), 7.41-7.39 (m, 6H).

# 1.5. Synthesis of 1,3,5-tris(4-ethynylphenyl)benzene (15)

1,3,5-Tris(4-ethynylphenyl)benzene (**14**) was synthesized in the two-step process following a literature procedure.<sup>4,5</sup> 1,3,5-Tris[4'-[(trimethylsilyl)ethynyl]phenyl]benzene was firstly synthesized from 1,3,5-tris(4-iodophenyl)benzene (**13**) by Sonogashira

cross-coupling with the use of trimethylsillylacetylene, and the subsequent basic hydrolysis of this intermediate product provided target 1,3,5-tris(4-ethynylphenyl)benzene (14).

1,3,5-tris[4'-[(trimethylsilyl)ethynyl]phenyl]benzene:

 $^1H$  NMR (CDCl\_3, 500 MHz, ppm),  $\delta_{\rm H}$  7.75 (s, 3H), 7.64-7.62 (m, 6 H), 7.58-7.56 (m, 6H), 0.28 (s, 27H).

1,3,5-tris(4-ethynylphenyl)benzene (15):

 $^1H$  NMR (CDCl\_3, 500 MHz, ppm),  $\delta_{\rm H}$  7.76 (s, 13H), 7.67-7.64 (m, 6 H), 7.62-7.60 (m, 6H), 3.16 (s, 3H).

#### 1.6. Synthesis of sumanene-ferrocene conjugate sumFc-1

#### Synthesis of 4-(formylphenyl)ferrocene<sup>6</sup> (4)



To a solution of ferroceneboronic acid (**2**; 200.0 mg, 0.87 mmol, 1 eq), 4-bromobenzaldehyde (**3**; 643.9 mg, 3.5 mmol, 4 eq), and tetrakis(triphenylphosphine)palladium(0) (Pd(PPh<sub>3</sub>)<sub>4</sub>; 100.5 mg, 0.09 mmol, 0.1 eq) in toluene (12 mL) and THF (5 mL), 2M K<sub>2</sub>CO<sub>3 aq</sub>

(2 mL) was added. The reaction mixture was refluxed for 48 hours. Distilled water (6 mL) was added, and the crude product was extracted with  $CH_2CI_2$  (3x20 mL). Organic layers were combined, washed with water, and brine. After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using a column chromatography (SiO<sub>2</sub>, 10% AcOEt/hex) to provide 4-(formylphenyl)ferrocene (**4**) as a red solid (192.5 mg, 63%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm),  $\delta_{H}$  9.97 (s, 1H), 7.81-7.78 (m, 2H), 7.61-7.59 (m, 2H), 4.75-4.74 (t-like m, <sup>3</sup>*J*<sub>H-H</sub> = 1.9 Hz, 2H), 4.44-4.43 (t-like m, <sup>3</sup>*J*<sub>H-H</sub> = 1.9 Hz, 2H), 4.05 (s, 5H); R<sub>f</sub> (10% AcOEt/hex) = 0.62. *The NMR data are consistent with the literature.*<sup>6</sup>

#### Synthesis of sumanene-ferrocene conjugate sumFc-1



Sumanene (1; 10.0 mg, 0.038 mmol, 1 eq) was placed in test tube. а Tetrabutylammonium bromide (TBAB; 6.1 mg, 0.019 mmol, 0.5 eq) was added, followed by the addition of THF (0.2 mL) and NaOH<sub>ag</sub> (30%; 2 mL). The reaction mixture was stirred for 5 min at 27°C. Solid 4-(formylphenyl)ferrocene (4; 69.0 mg, 0.23 mmol, 6 eq) was added in one portion, and the reaction mixture was stirred for 48 hours at 27°C. Distilled water (6 mL) was added, and the crude product was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3x20 mL). Organic layers were

combined, washed with saturated NH<sub>4</sub>Cl, water, and brine. After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using a column chromatography (SiO<sub>2</sub>, 50% CH<sub>2</sub>Cl<sub>2</sub>/cyclohexane) to provide the target sumanene-ferrocene conjugate **sumFc-1** as a deep-red solid (27.2 mg, 60%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm),  $\delta_{H}$  7.88-7.78 (m, 6H), 7.62-7.50 (m, 8H), 7.40-7.34 (m, 5.5H), 7.23-7.18 (m, 1.5H), 4.76-4.74 (m, 3H), 4.70-4.68 (m, 3H), 4.41-4.40 (m, 3H), 4.36-4.35 (m, 3H), 4.12-4.06 (4xs, 15H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 125 MHz, ppm),  $\delta_{C}$  147.9, 147.7, 147.5, 147.5, 147.3x2, 146.1, 145.9, 145.6, 145.5, 145.3, 145.2, 143.4, 143.2, 141.6, 141.3, 140.7, 140.6, 140.5, 139.9, 139.8x2, 139.7, 134.0, 133.9, 133.9, 133.8, 132.0x2, 131.3x2, 130.0x4, 128.7, 128.5, 128.4, 128.2, 126.3x3, 123.8, 123.6, 123.5, 123.4x2, 121.9, 120.8, 120.7, 120.6, 84.7x3, 84.6, 70.0., 69.6, 69.5, 66.8, 66.7x2; ESI-HRMS (TOF): calcd. for C<sub>72</sub>H<sub>48</sub>Fe<sub>3</sub> [M]<sup>+</sup> = 1080.1799, found: m/z 1080.1791; UV-Vis,  $\lambda_{max}$  (CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 *v*/*v*) 275, 365, 510 nm; R<sub>f</sub> (50% CH<sub>2</sub>Cl<sub>2</sub>/cyclohexane) = 0.47.

#### 1.7. Synthesis of sumanene-ferrocene conjugate sumFc-2

#### Synthesis of 4-(ferrocenylethynyl)benzaldehyde (7)



reaction flask. The content of the tube was evacuated and filled with argon. Triethylamine (TEA; 30 mL) was added, and the mixture was stirred for 15 min at 60°C under argon atmosphere. A solution of ethynylferrocene (5; 205.0 mg; 0.98 mmol, 1.5 eg) in TEA (15 mL) was added via syringe. The reaction mixture was stirred for 24 hours at 60°C under argon atmosphere. Distilled water (40 mL) was added, and the crude product was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3x20 mL). Organic layers were combined, washed with 1M HCl (5x10 mL), and water. After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using column chromatography  $(SiO_2,$ 30% hex/CHCl<sub>3</sub>) to provide а 4-(ferrocenylethynyl)benzaldehyde (7) as a red solid (173.8 mg, 84%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm)  $\delta_{H}$  10.01 (s, 1H), 7.85-7.82 (m, 2H), 7.63-7.61 (m, 2H), 4.54-4.54 (t-like m, <sup>3</sup>*J*<sub>H-H</sub> = 1.8 Hz, 2H), 4.30-4.30 (t-like m, <sup>3</sup>*J*<sub>H-H</sub> = 1.8 Hz, 2H), 4,26 (s, 5H); R<sub>f</sub> (30% hex/CHCl<sub>3</sub>) = 0.46. *The NMR data are consistent with the literature*.<sup>7</sup>

# Synthesis of sumanene-ferrocene conjugate sumFc-2 by the reaction between sumanene (1) and 4-(ferrocenylethynyl)benzaldehyde (7)



Sumanene (1; 10.0 mg, 0.038 mmol, 1 eq) was placed in a test tube. TBAB (6.1 mg, 0.019 mmol, 0.5 eq) was added, followed by the addition of THF (0.2 mL) and NaOH<sub>aq</sub> (30%; 2 mL). The reaction mixture was stirred for 5 min at 27°C. Solid 4-(ferrocenylethynyl)benzaldehyde (7: 71.5 mg, 0.23 mmol, 6 eq) was added in one portion, and the reaction mixture was stirred for 48 hours at 27°C. Distilled water (6 mL) was added, and the crude product was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3x20 mL). Organic layers were combined,

washed with saturated NH<sub>4</sub>Cl, water, and brine. After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using a column chromatography (SiO<sub>2</sub>, 50% CH<sub>2</sub>Cl<sub>2</sub>/hexane) to provide the target sumanene-ferrocene conjugate **sumFc-2** as a deep-red solid (39.4 mg, 86%).

<sup>1</sup>H NMR (THF- $d_8$ , 500 MHz, ppm)  $\delta_H$  7.95-7.88 (m, 6H), 7.58-7.51 (m, 10H), 7.47-7.47 (m, 2H), 7.30-7.29 (m, 3H), 4.54-4.53 (m, 3H), 4.50-4.50 (m, 3H), 4.30-4.28 (m, 3H), 4.27-4.25 (m, 11H), 4.23-4.22 (2xs, 7H); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm)  $\delta_H$  7.90-7.80 (m, 6H), 7.61-7.52 (m, 6H), 7.50-7.47 (m, 1.5H), 7.38-7.34 (m, 5H), 7.26 (s, 1H), 7.18-7.15 (m, 1.5H), 4.56-4.55

(m, 3H), 4.53-4.52 (m, 3H), 4.30-4.28 (m, 11H), 4.27-4.25 (m, 10H);  ${}^{13}C{}^{1}H{}$  NMR (CDCl<sub>3</sub>, 125MHz, ppm)  $\delta_{C}$  147.7, 147,6, 147.4, 147.3, 146.0, 145.7, 145.4, 143.4, 143.1, 141.5, 141.3, 140.9, 140.8x2, 140.7, 135.7, 135.6x2, 135.5, 131.7x4, 129.8x4, 128.2, 128.0, 127.9, 127.6, 124.5x2, 124.4x2, 123.9, 123.7, 123.6, 123.5, 123.4, 121.1x2, 121.0, 121.8, 90.7, 90.6x3, 86.1x2, 86.0x2, 71.7x2, 70.2x2, 69.2, 69.1, 65.2x3, 61.1; ESI-HRMS (TOF): calcd. for C<sub>78</sub>H<sub>48</sub>Fe<sub>3</sub>[M]<sup>+</sup> = 1152.1799, found: m/z 1152.1787; UV-Vis,  $\lambda_{max}$  (CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 *v*/*v*) 260, 370, 500 nm; R<sub>f</sub> (50% CH<sub>2</sub>Cl<sub>2</sub>/hex) = 0.55.

#### Synthesis of tris[(4-iodophenyl)methidene]sumanene (8a)



Sumanene (1; 10.0 mg, 0.038 mmol, 1 eq) was placed in a test tube. TBAB (6.1 mg, 0.019 mmol, 0.5 eq) was added, followed by the addition of THF (0.2 mL) and NaOH<sub>aq</sub> (30%; 2 mL). The reaction mixture was stirred for 5 min at 27°C. Solid 4-iodobenzaldehyde (**6a**; 54.3 mg, 0.23 mmol, 6 eq) was added in one portion, and the reaction mixture was stirred for 48 hours at 27°C. Distilled water (6 mL) was added, and the crude product was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3x20 mL). Organic layers were combined, washed with saturated NH<sub>4</sub>Cl, water,

and brine. After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using a column chromatography (SiO<sub>2</sub>, 50% CH<sub>2</sub>Cl<sub>2</sub>/hexane) to provide tris[(4-iodophenyl)methidene]sumanene (**8a**) as an orange solid (21.4 mg, 62%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm)  $\delta_{H}$  7.81-7.74 (m, 6H), 7.63-7.53 (m, 6H), 7.38-7.35 (m, 2H), 7.31-7.26 (m, 3H), 7.24 (s, 1H), 7.13-7.10 (m, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 125MHz, ppm)  $\delta_{C}$  149.3, 149.2; 149.1, 148.9, 147.7x2, 147.6, 147.4x2, 147.3x2, 145.9, 145.7, 145.5, 145.4x2, 143.2, 143.0, 141.5, 141.2, 141.1, 141.0, 138.0x2, 135.9, 135.8, 135.7, 135.6, 131.4x2, 127.6, 127.3x2, 127.0, 123.8, 123.6x2, 123.4, 121.2, 121.1, 121.0, 120.9, 94.9, 94.8, 94.7x2; ESI-HRMS (TOF): calcd. for C<sub>42</sub>H<sub>21</sub>I<sub>3</sub> [M]<sup>+</sup> = 905.8772, found: m/z 905.8760; R<sub>f</sub> (50% CH<sub>2</sub>Cl<sub>2</sub>/hex) = 0.51.

Synthesis of sumanene-ferrocene conjugate sumFc-2 by the Sonogashira crosscoupling between tris[(4-iodophenyl)methidene]sumanene (8a) and ethynylferrocene (5)



Tris[(4-iodophenyl)methidene]sumanene (8; 15.0 mg, 0.02 mmol, 1 eq),  $Pd(PPh_3)_2Cl_2$  (2.2 mg; 0.003) mmol, 0.15 eq), and Cul (1.2 mg; 0.006 mmol, 0.3 eq) were added to a reaction flask. The content of the tube was evacuated and filled with argon. TEA (5 mL) was added, and the mixture was stirred for 15 min at 60°C under argon atmosphere. A solution of ethynylferrocene (5; 17.0 mg; 0.08 mmol, 4 eq) in TEA (2.5 mL) was added via syringe. The reaction mixture was stirred for 24 hours at 60°C under argon atmosphere.

Distilled water (10 mL) was added, and the crude product was extracted with  $CH_2CI_2$  (3x20 mL). Organic layers were combined, washed with 1M HCl (3x10 mL), and water. After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using a column chromatography (SiO<sub>2</sub>, 50% hex/CHCl<sub>3</sub>) to provide the target sumanene-ferrocene conjugate **sumFc-2** as a deep-red solid (21.5 mg, 95%).

The spectral data for **sumFc-2** obtained with this method conformed to that presented above.

# Synthesis of tris[(4-bromophenyl)methidene]sumanene (8b)



Sumanene (1; 10.0 mg, 0.038 mmol, 1 eq) was placed in a test tube. TBAB (6.1 mg, 0.019 mmol, 0.5 eq) was added, followed by the addition of THF (0.2 mL) and NaOH<sub>aq</sub> (30%; 2 mL). The reaction mixture was stirred for 5 min at 27°C. Solid 4-bromobenzaldehyde (**6b**; 42.6 mg, 0.23 mmol, 6 eq) was added in one portion, and the reaction mixture was stirred for 48 hours at 27°C. Distilled water (6 mL) was added, and the crude product was extracted with  $CH_2Cl_2$  (3x20 mL). Organic layers were combined, washed with saturated NH<sub>4</sub>Cl,

water, and brine. After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using a column chromatography (SiO<sub>2</sub>, 50% CH<sub>2</sub>Cl<sub>2</sub>/hexane) to provide tris[(4-bromophenyl)methidene]sumanene (**8b**) as an orange solid 17.4 mg, 60%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm)  $\delta_{H}$  7.78-7.68 (m, 6H), 7.62-7.54 (m, 6H), 7.39-7.32 (m, 6H), 7.16-7.13 (m, 3H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 125MHz, ppm)  $\delta_{C}$  147.7x2, 147.5, 147.4x2, 146.0, 145.8, 145.6, 145.5x2, 143.3, 143.0, 141.5, 141.3, 141.1, 141.0, 135.3, 135.2x2, 135.1, 132.1, 132.0, 131.3x2, 127.5, 127.3, 127.2, 127.0, 123.8, 123.7, 123.6, 123.4, 123.1, 123.0x2, 121.2,

121.2, 121.1, 121.0; ESI-HRMS (TOF): calcd. for  $C_{42}H_{21}Br_3$  [M]<sup>+</sup> = 763.9167, found: m/z 763.9155;  $R_f$  (50% CH<sub>2</sub>Cl<sub>2</sub>/hex) = 0.68.

The **sumFc-2** synthesis yield starting from tris[(4-bromophenyl)methidene]sumanene (**8b**) and ethynylferrocene (**5**) was 52%.

#### 1.8. Synthesis of sumanene-ferrocene conjugate sumFc-3

# Synthesis of 4-(1-(ferrocenylmethyl)-1H-1,2,3-triazol-4-yl)benzaldehyde (11)



Ferrocenemethylazide (9; 241.0 mg, 1.0 mmol, 1 eq), 4ethynylbenzaldehyde (10; 130.1 mg, 1.0 mmol, 1 eq), copper(II) sulphate pentahydrate (CuSO<sub>4</sub>·5H<sub>2</sub>O; 62.4 mg, 0.25 mmol, 0.25 eq) and sodium ascorbate (148.6 mg, 0.75 mmol, 0.75 eq) were placed in a reaction flask. DMF (40 mL) was added. The reaction mixture was stirred for 24 hours at

50°C. Distilled water (100 mL) was added, and the formed precipitate was filtrated on a nylon membrane (0.45  $\mu$ m). The resultant solid was dissolved in CHCl<sub>3</sub> (50 mL). After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using a column chromatography (SiO<sub>2</sub>, CHCl<sub>3</sub>) to provide 4-(1-(ferrocenylmethyl)-1*H*-1,2,3-triazol-4-yl)benzaldehyde (**11**) as a yellow solid (280.0 mg, 75%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm),  $\delta_{H}$  10.01 (s, 1H), 7.97-7.95 (m, 2H), 7.92-7.90 (m, 2H), 7.74 (s, 1H), 5.37 (s, 1H), 4.33-4.32 (t-like m,  ${}^{3}J_{H-H} = 1.9$  H, 2H), 4.26–4.25 (t-like m,  ${}^{3}J_{H-H} = 1.9$  H, 2H), 4.22 (s, 5H);  ${}^{13}C{}^{1}H$  NMR (CDCl<sub>3</sub>, 125 MHz, ppm),  $\delta_{C}$  191.8, 146.5, 136.6, 135.9, 130.5, 126.1, 120.2, 80.5, 69.5, 69.2x2, 50.5; ESI-HRMS (TOF): calcd. for C<sub>20</sub>H<sub>17</sub>FeN<sub>3</sub>O [M]<sup>+</sup> = 371.0716, found: m/z 371.0711; R<sub>f</sub> (CHCl<sub>3</sub>) = 0.40.

# Synthesis of tris[(4-ethynylphenyl)methidene]sumanene (12)



Sumanene (1; 20.0 mg, 0.076 mmol, 1 eq) was placed in a test tube. TBAB (12.2 mg, 0.038 mmol, 0.5 eq) was added, followed by the addition of THF (0.4 mL) and NaOH<sub>aq</sub> (30%; 4 mL). The reaction mixture was stirred for 5 min at 27°C. Solid 4-ethynylbenzaldehyde (**10**; 59.1 mg, 0.456 mmol, 6 eq) was added in one portion, and the reaction mixture was stirred for 48 hours at 27°C. Distilled water (6 mL) was added, and the crude product was extracted with  $CH_2CI_2$  (3x20 mL). Organic layers were combined, washed with saturated NH<sub>4</sub>CI, water, and

brine. After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using a column chromatography (SiO<sub>2</sub>, 50% CH<sub>2</sub>Cl<sub>2</sub>/hexane) to provide tris[(4-ethynylphenyl)methidene]sumanene (**12**) as a red solid (22.9 mg, 50%).

 $^1\text{H}$  NMR (CDCl<sub>3</sub>, 500 MHz, ppm)  $\delta_{\text{H}}$  7.88-7.78 (m, 6H), 7.61-7.53 (m, 6H), 7.41-7.33 (m, 6H), 7.16-7.15 (m, 3H), 3.21-3.17 (4xs, 3H);  $^{13}\text{C}\{^1\text{H}\}$  NMR (CDCl<sub>3</sub>, 125MHz, ppm)  $\delta_{\text{C}}$  147.8,147.7, 147.6, 147.4x2, 147.3, 146.0, 145.7x2, 145.6, 145.4x2, 143.3, 143.0, 141.5, 141.3x2, 141.2x2, 136.9, 136.8x2, 136.7, 132.6, 132.5x3, 129.7x4, 127.8, 127.6, 127.5, 127.3, 123.9, 123.7, 123.6, 123.5, 122.5x2, 122.4x2, 121.2, 121.1, 121.0, 120.9, 83.7x3, 83.6, 78.9, 78.8x3; ESI-HRMS (TOF): calcd. for C\_{48}H\_{24} [M]<sup>+</sup> = 600.1873, found: m/z 600.1872; R<sub>f</sub> (3% CH<sub>3</sub>OH/CHCl<sub>3</sub>) = 0.50.

Synthesis of sumanene-ferrocene conjugate sumFc-3 by the click chemistry between tris[(4-ethynylphenyl)methidene]sumanene (12) and ferrocenemethylazide (9)



Ferrocenemethylazide (9; 12.5 mg, 0.052 mmol. 3 ea). tris[(4ethynylphenyl)methidene]sumanene (12; 10.0 mg, 0.017 mmol, 1 eq), CuSO<sub>4</sub>·5H<sub>2</sub>O (6.2 mg, 0.025 mmol, 1.5 eq) and sodium ascorbate (14.9 mg, 0.075 mmol, 4 eq) were placed in a reaction flask. DMF (4 mL) was added. The reaction mixture was stirred for 48 hours at 50°C. Distilled water (15 mL) was added, and the formed precipitate was filtrated on a nylon membrane (0.45 µm). The resultant solid was dissolved in CHCl<sub>3</sub> (30 mL). After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator.

Finally, the product was purified using a column chromatography (SiO<sub>2</sub>, 5% CH<sub>3</sub>OH/CHCl<sub>3</sub>) to provide the target sumanene-ferrocene conjugate **sumFc-3** as a deep-red solid (16.6 mg, 74%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm), δ<sub>H</sub> 7.91-7.80 (m, 12H), 7.79 (s, 1H), 7.75 (s, 1H), 7.70-7.69 (2xs, 1H), 7.44-7.38 (m, 1.5H), 7.32-7.27 (m, 4H), 7.23 (s, 1H), 7.17-7.17 (m, 1H), 7.08-7.04 (m, 1.5H), 5.37-5.34 (4xs, 6H), 4.36-4.35 (m, 3H), 4.33-4.32 (m, 3H), 4.26-4.24 (m, 6H), 4.23-4.21 (4xs, 15H); <sup>13</sup>C{<sup>1</sup>H} NMR (CDCl<sub>3</sub>, 125 MHz, ppm),  $\delta_{C}$  147.6, 147.4x3, 147.3, 147.1, 147.0x2, 145.8, 145.4x2, 145.3, 145.1, 144.0, 143.2, 142.9, 141.3, 140.7, 140.6x3, 140.4, 136.1x2, 136.0x2, 131.0x2, 130.9x2, 130.3x2, 128.1, 127.9, 127.7, 127.6, 125.9x3, 123.7, 123.6, 123.5, 123.2, 120.9, 120.8x2, 120.7, 119.8, 119.6x2, 119.5, 81.1, 81.0x2, 80.9, 69.3, 69.1, 50.4x2, 50.3; ESI-HRMS (TOF): calcd. for C<sub>81</sub>H<sub>57</sub>Fe<sub>3</sub>N<sub>9</sub>[M]<sup>+</sup> = 1323.2780, found: m/z 1323.2780; UV-Vis, λ<sub>max</sub> (CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 *v*/*v*) 278, 380, 480 nm; R<sub>f</sub> (50% CH<sub>2</sub>Cl<sub>2</sub>/hex) = 0.60.

#### 1.9. Synthesis of 1,3,5-triphenylbenzene-ferrocene conjugate 14



1,3,5-Tris(4-iodophenyl)benzene (**13**; 50.0 mg; 0.079 mmol, 1 eq), Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (7.7 mg; 0.011 mmol, 0.15 eq), and Cul (4.2 mg; 0.024 mmol, 0.30 eq) were added to a reaction flask. The content of the tube was evacuated and filled with argon. Triethylamine (TEA; 10 mL) was added, and the mixture was stirred for 15 min at 60°C under argon atmosphere. A solution of ethynylferrocene (**5**; 61.3 mg; 0.316 mmol, 4.0 eq) in TEA (5 mL) was added via syringe. The reaction mixture was stirred for 24 hours at 60°C under argon atmosphere. Distilled water (6 mL) was added, and the crude product was extracted with CH<sub>2</sub>Cl<sub>2</sub>

(3x20 mL). Organic layers were combined, washed with 1M HCl (4x15 mL), and water. After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using a column chromatography (SiO<sub>2</sub>, 50% hex/CH<sub>2</sub>Cl<sub>2</sub>), and further with preparative thin layer chromatography (PTLC; SiO<sub>2</sub>, 50% hex/CH<sub>2</sub>Cl<sub>2</sub>) to provide 1,3,5-triphenylbenzene-ferrocene conjugate (**14**) as a red solid (58.1 mg, 80%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm),  $\delta_{H}$  7.80 (s, 3H), 7.69-7.67 (m, 6H), 7.62-7.60 (m, 6H), 4.54-4.54 (t-like m,  ${}^{3}J_{H-H} = 1.8$  H, 6H), 4.27-4.25 (m, 21H); <sup>1</sup>H NMR (C<sub>6</sub>D<sub>6</sub>, 500 MHz, ppm),  $\delta_{H}$  7.70-7.69 (m, 6H), 7.60 (s, 3H), 7.37-7.36 (m, 6H), 4.56-4.45 (t-like m,  ${}^{3}J_{H-H} = 1.8$  H, 6H), 4.12 (s, 15H), 3.98-3.97 (t-like m,  ${}^{3}J_{H-H} = 1.8$  H, 6H); 142.0, 140.2, 132.0, 127.3, 125.1, 123.5, 89.6, 85.7, 71.6, 70.2, 69.1, 65.4; ESI-HRMS (TOF): calcd. for C<sub>60</sub>H<sub>42</sub>Fe<sub>3</sub> [M]<sup>+</sup> = 930.1329, found: m/z 930.1332; UV-Vis,  $\lambda_{max}$  (CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 *v*/*v*) 245, 312, 368 nm; R<sub>f</sub> (50% hex/CH<sub>2</sub>Cl<sub>2</sub>) = 0.45.

#### 1.10. Synthesis of 1,3,5-triphenylbenzene-ferrocene conjugate 16



Ferrocenemethylazide (9; 71.0 mg, 0.246 mmol, 3.1 eq), 1,3,5-tris(4ethynylphenyl)benzene (15; 30.0 mg, 0.079 mmol, 1 eq), CuSO<sub>4</sub>·5H<sub>2</sub>O (29.6 mg, 0.119 mmol, 1.5 eq) and sodium ascorbate (62.8 mg, 0.317 mmol, 4 eg) were placed in a reaction flask. DMF (8 mL) was added. The reaction mixture was stirred for 24 hours at 50°C. Distilled water (60 mL) was added, and the formed precipitate was filtrated on a nylon membrane (0.45 µm). The resultant solid was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (60 mL). After drying with MgSO<sub>4</sub> followed by filtration, volatiles were distilled off on a rotary evaporator. Finally, the product was purified using a column chromatography

 $(SiO_2, 3\% CH_3OH/CHCI_3)$  to provide 1,3,5-triphenylbenzene-ferrocene conjugate **16** as a yellow solid (56.5 mg, 65%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz, ppm),  $\delta_{H}$  7.91-7.90 (m, 6H), 7.81 (s, 3H), 7.75-7.73 (m, 6H), 7.69 (s, 3H), 5.37 (s, 6H), 4.35-4.44 (t-like bm, <sup>3</sup>*J*<sub>H-H</sub> = 1.8 H, 6H), 4.26-4.25 (t-like bm, <sup>3</sup>*J*<sub>H-H</sub> = 1.8 H, 6H), 4.22 (s, 15H); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 500 MHz, ppm),  $\delta_{H}$  8.67 (s, 3H), 7.99-7.98 (m, 15H), 5.38 (s, 6H), 4.43-4.42 (t-like m, <sup>3</sup>*J*<sub>H-H</sub> = 1.9 H, 6H), 4.22-4.21 (m, 21H); <sup>13</sup>C{<sup>1</sup>H} NMR (DMSO-*d*<sub>6</sub>, 125 MHz, ppm),  $\delta_{C}$  146.0, 141.1, 139.3, 130.2, 127.7, 125.6, 124.1, 121.2, 82.4, 68.8, 68.7, 68.4, 49.2; ESI-HRMS (TOF): calcd. for C<sub>63</sub>H<sub>51</sub>Fe<sub>3</sub>N<sub>9</sub>[M]<sup>+</sup> = 1101.2310, found: m/z 1101.2316; UV-Vis,  $\lambda_{max}$  (CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 *v*/*v*) 238, 292 nm; R<sub>f</sub> (3% CH<sub>3</sub>OH/CHCl<sub>3</sub>) = 0.56.

#### 1.11. Electrochemical studies

**Materials.** Dichlorometane (DCM, Sigma-Aldrich), dimethyl sulfoxide (DMSO, Merck) tetrabutylammonium hexafluorophosphate (TBAPF<sub>6</sub>, Sigma-Aldrich), tetrabutylammonium tetrafluoroborate (TBABF<sub>4</sub>, Sigma-Aldrich), cesium nitrate (CsNO<sub>3</sub>, Sigma-Aldrich), potassium nitrate (KNO<sub>3</sub>, Sigma-Aldrich), sodium nitrate (NaNO<sub>3</sub>, Sigma-Aldrich), barium nitrate (Ba(NO<sub>3</sub>)<sub>2</sub>, Sigma-Aldrich) and perfluorinated resin solution containing Nafion<sup>TM</sup> (nafion<sup>TM</sup>, Sigma-Aldrich) were used without additional purification.

**Voltammetry.** The voltammetric measurements (cyclic voltammetry and differential pulse voltammetry) were performed with using an Autolab, model PGSTAT 12 potentiostat in the three-electrode system consisting of: (i) working electrode (glassy carbon disc electrode, GC; 3.0 mm in diameter, BAS, UK), (ii) reference electrode (Ag/AgCl/3 M KCl) and (iii) counter electrode (Pt plate with surface area at least 1 cm<sup>2</sup>). Each time, before the measurements or electrode modification, the surface of the working electrode was mechanically cleaned by polishing on a wet pad with addition a 1  $\mu$ m Al<sub>2</sub>O<sub>3</sub> powder. Then, the electrode was rinsed with a direct stream of ultrapure water (Hydrolab, conductivity of ~ 0.056  $\mu$ S·cm<sup>-1</sup>) to remove residue of alumina. In all experiments, the electrochemical cell was kept in a Faraday cage to minimize the electrical noise.

**Preparation of recognition layers (GC/sumFc-TBAPF**<sub>6</sub>-nafion<sup>TM</sup>). Before each modification the GC electrode was oxidized in 0.1 M H<sub>2</sub>SO<sub>4</sub> by cycling in the potential range -0.35 - 1.3 - 0.35 V with scan rate 100 mV·s<sup>-1</sup>. The oxidation process was aimed at generating carboxyl groups involved in the immobilization of the receptor layer. Next, the surface was dried and the 10-µL droplet of 20 µM sumFc-1 – sumFc-3 solution (CH<sub>2</sub>Cl<sub>2</sub> : DMSO (1:1 v/v)) with addition of 100 mM TBAPF<sub>6</sub> and 5% nafion<sup>TM</sup> was placed onto the electrode surface. Then, the electrode was left to dry in the desiccator. Before the cesium cations (Cs<sup>+</sup>) titration the recognition layers (GC/sumFc-TBAPF<sub>6</sub>-nafion<sup>TM</sup>) were first cycled in the 50 mM TBABF<sub>4</sub> aqueous solution in the potential range 0.0 - 1.1 - 0. V with scan rate 100 mV·s<sup>-1</sup> until a stable voltammogram was obtained.

#### 1.12. Laser Ablation Inductively Coupled Plasma Mass Spectrometry (LA-ICP-MS)

Laser Ablation Inductively Coupled Plasma Mass Spectrometry (LA-ICP-MS). The LSX-213 laser ablation system (Nd:YAG,  $\lambda$ = 213nm) by CETAC (USA) was coupled to ICP-MS NexION 300D spectrometer by Perkin Elmer (USA). The energy (E=3mJ) of the laser beam was focused on the surface of the analysed samples on the area with a diameter of 100 µm. Transient signals were registered for selected isotopes (<sup>23</sup>Na, <sup>39</sup>K, <sup>57</sup>Fe, <sup>133</sup>Cs, <sup>137</sup>Ba) during a multi-line (n=3) ablation with the scan rate SR=100 µm/s; 20Hz repetition rate and time delay (10 s) between ablation of the subsequent lines. The operating conditions of the used ICP-MS system are: plasma power: 1400 W; carrier gas flow (Ar): 0.88 L·min<sup>-1</sup>; sweeps: 1; readings: 24321; repeats: 1; dwell time: 1 ms.

# 2. NMR spectra

2.1. Spectra



*Figure S1.* <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of sumanene (1).



Figure S2. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of 4-(formylphenyl)ferrocene (4).



*Figure S3.* <sup>1</sup>H NMR (500 MHz, CDCI<sub>3</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-1**.



*Figure S4.* <sup>1</sup>H-<sup>1</sup>H COSY NMR (500 MHz, CDCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-1**.







*Figure S7.* <sup>1</sup>H-<sup>13</sup>C HSQC NMR (CDCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-1**.



*Figure S8.* <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of 4-(ferrocenylethynyl)benzaldehyde (7).



Figure S9. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate sumFc-2.



*Figure S10.* <sup>1</sup>H NMR (500 MHz, THF-*d*<sub>8</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-2**.



*Figure S11.* <sup>1</sup>H-<sup>1</sup>H COSY NMR (500 MHz, THF-*d*<sub>8</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-2**.



*Figure S12.* <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-2**.



*Figure S13.* 148-120 ppm inset of the <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-2**.



*Figure S14.* <sup>1</sup>H-<sup>13</sup>C HSQC NMR (THF-*d*<sub>8</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-2**.



Figure S15. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of tris[(4-iodophenyl)methidene]sumanene (8a).



*Figure S16.* <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>) spectrum tris[(4-iodophenyl)methidene]sumanene (8a).



*Figure S17.* <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of tris[(4-bromophenyl)methidene]sumanene (8b).



*Figure S18.* <sup>1</sup>H-<sup>1</sup>H COSY NMR (500 MHz, CDCl<sub>3</sub>) spectrum of tris[(4-bromophenyl)methidene]sumanene (8b).



*Figure S19.* <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCI<sub>3</sub>) spectrum tris[(4-bromophenyl)methidene]sumanene (**8b**).



*Figure S20.* <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of ferrocenemethylazide (9).



*Figure S21.* <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of 4-(1-(ferrocenylmethyl)-1*H*-1,2,3-triazol-4-yl)benzaldehyde (**11**).



*Figure S22.* <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>) spectrum of 4-(1-(ferrocenylmethyl)-1*H*-1,2,3-triazol-4-yl)benzaldehyde (**11**).


*Figure S23.* <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of tris[(4-ethynylphenyl)methidene]sumanene (12).



Figure S24. <sup>1</sup>H-<sup>1</sup>H COSY NMR (500 MHz, CDCl<sub>3</sub>) spectrum of tris[(4-ethynylphenyl)methidene]sumanene (12).



*Figure S25.* <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>) spectrum of tris[(4-ethynylphenyl)methidene]sumanene (**12**).



Figure S26. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate sumFc-3.



Figure S27. <sup>1</sup>H-<sup>1</sup>H COSY NMR (500 MHz, CDCI<sub>3</sub>) spectrum of sumanene-ferrocene conjugate sumFc-3.





*Figure S29.* <sup>1</sup>H-<sup>13</sup>C HSQC NMR (CDCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-3**.





*Figure S31.* <sup>1</sup>H NMR (500 MHz, CDCI<sub>3</sub>) spectrum of 1,3,5-tris[4'-[(trimethylsilyl)ethynyl]phenyl]benzene.





Figure S33. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of 1,3,5-triphenylbenzene-ferrocene conjugate 14.



*Figure S34.* <sup>1</sup>H NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>) spectrum of 1,3,5-triphenylbenzene-ferrocene conjugate **14**.



*Figure S35.* <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>) spectrum of 1,3,5-triphenylbenzene-ferrocene conjugate **14**.



Figure S36. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of 1,3,5-triphenylbenzene-ferrocene conjugate 16.



*Figure* **S37.** <sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>) spectrum of 1,3,5-triphenylbenzene-ferrocene conjugate **16**.



*Figure S38.* <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, DMSO-*d*<sub>6</sub>) spectrum of 1,3,5-triphenylbenzene-ferrocene conjugate **16**.

### 2.2. Additional comments on the NMR spectra of sumanene-ferrocene conjugates

#### 2.2.1. General comment

The NMR spectra of sumanene-ferrocene conjugates **sumFc-1-sumFc-3** consisted of the signals originating from the presence of  $C_3$  symmetrical and unsymmetrical diastereomer (for the respective structures of those isomers, see Figure S39), similarly to NMR spectra of other sumanene derivatives synthesized with this method.<sup>8–10</sup>



symmetrical diastereoisomer

unsymmetrical diastereoisomer

*Figure S39.* The structures of  $C_3$  symmetrical and unsymmetrical diastereomer of the discussed sumanene derivatives. For clarity, the magnetically inequivalent protons are labeled with grey circles. The same color does not correspond to the same chemical shift in the NMR spectrum.

# 2.2.2. Comments regarding the NMR spectra of sumanene-ferrocene conjugate sumFc-1.

Several groups of multiplets and 4 singlets can be found in the <sup>1</sup>H NMR spectrum of sumanene-ferrocene conjugate **sumFc-1**. Those groups of signals are graphically presented in Figure S40 and are labeled in <sup>1</sup>H NMR spectrum presented in Figure S41.

Seven or twenty-four signals shall be found in the <sup>1</sup>H NMR spectrum of symmetrical and unsymmetrical stereoisomer of **sumFc-1**, respectively. Those signals overlap what results in the presence of several groups of multiplets in the spectrum.

The signals originating from the sumanene skeleton, ferrocene unit and phenylene linkage can be found in the spectrum. The total number of protons (48H), as well as the number of protons within given groups of signals, are consistent with the anticipated values. Noteworthy, the lack of signals coming from the benzylic protons of sumanene also confirms the successful modification of sumanene skeleton (for the reference <sup>1</sup>H NMR spectrum of sumanene, see Figure S1).

Four singlets located at 4.12, 4.11, 4.07 and 4.06 ppm were ascribed to the 15 protons of the unsubstituted cyclopentadienyl ( $C_p$ ) ring of ferrocene (symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s). <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis confirmed that those signals are indeed four singlets, not two dublets or dublet of dublets (see graphical representation in Figure S42 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S4).

Four multiplets located at 4.76-4.74, 4.70-4.68, 4.41-4.40 and 4.36-4.35 ppm were ascribed to the 12 protons of the substituted  $C_p$  ring of ferrocene (symmetrical diastereoisomer = 2 signals = 2 x t-like m, unsymmetrical diastereoisomer = 6 signals = 6 x t-like m). The presence of multiplets in the spectrum for those signals originates from the fact that characteristic triplet-like multiplets coming from the substituted  $C_p$  ring of ferrocene overlap.

Multiplets located at 7.54-7.21 ppm were ascribed to the 9 protons of the sumanene skeleton (6H; symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 6 signals = 6 x d) and methidene moieties (3H; symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s). <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis further supported this hypothesis (see graphical representation in Figure S42 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S4). Multiplet located at *ca*. 7.54-7.50 ppm overlap with the multiplet located at *ca*. 7.60-7.54 ppm coming from the sumanene skeleton. This hypothesis has been clearly confirmed with the <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis (see graphical representation in Figure S42 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S42 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis (see graphical representation in Figure S42 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis (see graphical representation in Figure S42 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis (see graphical representation in Figure S42 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis (see graphical representation in Figure S42 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S4).

Two multiplets located at 7.88-7.78 and 7.60-7.54 ppm were ascribed to the 12 protons of the phenylene linkage (symmetrical diastereoisomer = 2 signals = 2 x m, unsymmetrical diastereoisomer = 6 signals = 6 x m). This hypothesis has been confirmed with the <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis (see Figure S4).



sumFc-1 symmetrical diastereoisomer



sumFc-1 unsymmetrical diastereoisomer

24 signals in <sup>1</sup>H NMR

#### 7 signals in <sup>1</sup>H NMR



*Figure S40.* Graphical representation of the number and types of signals observed in the <sup>1</sup>H NMR spectra of sumanene-ferrocene conjugate **sumFc-1**. The same color does not correspond to the same chemical shift in the NMR spectrum.







*Figure S42.* Graphical representation of the types of signals observed in the <sup>1</sup>H NMR spectrum of sumanene-ferrocene conjugate **sumFc-1** and the graphical representation of the cross-correlations in the <sup>1</sup>H-<sup>1</sup>H COSY NMR spectra together with the crucial insets of this spectra. For the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum, see Figure S4.

Sixteen groups of signals can be found in the <sup>13</sup>C{<sup>1</sup>H} NMR spectrum of sumaneneferrocene conjugate **sumFc-1**, what is consistent with the anticipated value. Those groups of signals are graphically presented in Figure S43 and are labeled in <sup>13</sup>C{<sup>1</sup>H} NMR spectrum presented in Figure S44.

Sixteen or forty-eight signals shall be found in the  ${}^{13}C{}^{1}H$  NMR spectrum of symmetrical and unsymmetrical stereoisomer of **sumFc-1**, respectively. Noteworthy, within most groups of signals in the  ${}^{13}C{}^{1}H$  NMR spectrum of **sumFc-1**, four singlets can be observed, what is highly consisted with the anticipated value (symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s), see, *e.g.*, the signal located at 84.7-84.6 ppm (characteristic *ipso*  ${}^{13}C$  nucleus of substituted Cp ring of ferrocene).

Signals located at 148.0-120.6 ppm were ascribed to the presence of various <sup>13</sup>C nucleus coming from the sumanene skeleton, phenylene linkage and methidene moieties.

Signals located at 84.7-66.7 ppm were ascribed to the presence of ferrocene units. Signals located at 70.0-69.9 ppm were ascribed to the <sup>13</sup>C nucleus of unsubstituted  $C_p$  ring of ferrocene. Signals located at 69.6-69.5 and 66.8-66.7 ppm were ascribed to the <sup>13</sup>C nucleus of substituted  $C_p$  ring of ferrocene. Signals located at 84.7-84.6 ppm were ascribed to the characteristic *ipso* <sup>13</sup>C nucleus of substituted Cp ring of ferrocene.



sumFc-1 symmetrical diastereoisomer







sumFc-1 unsymmetrical diastereoisomer

#### 48 signals in <sup>13</sup>C{<sup>1</sup>H} NMR

signal label	number of signals
$\bigcirc$	21
$\bigcirc$	3 x 1 = 3
$\bigcirc$	3 x 4 = 12
$\bigcirc$	3 x 3 = 9
$\bigcirc$	3 x 1 = 3
	total = 48 signals

*Figure S43.* Graphical representation of the number of signals observed in the <sup>13</sup>C{<sup>1</sup>H} NMR spectra of sumanene-ferrocene conjugate **sumFc-1**. The same color does not correspond to the same chemical shift in the NMR spectrum.



*Figure S44.* <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CHCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-1** with the signals ascribed to the given groups. For the labels, see Figure S43.

The signals originating from the presence of ferrocene in the synthesized compounds overlapped in the <sup>1</sup>H NMR and <sup>13</sup>C{<sup>1</sup>H} NMR spectra, hence, <sup>1</sup>H-<sup>13</sup>C HSQC NMR analyses were especially helpful to support the ascription of the signals originating from the presence of ferrocene in the synthesized conjugates **sumFc-1** – **sumFc-3**, see graphical representation for the representative sumanene derivative **sumFc-1** in Figure S45.



Figure S45. The graphical representation of the correlations in the <sup>1</sup>H-<sup>13</sup>C HSQC NMR spectra for the ferrocene unit in sumanene-ferrocene conjugate sumFc-1 (crucial insets of this spectra are presented). For the detailed discussion on the NMR spectra of sumFc-1 – sumFc-3, see section 2 in Supporting Information

## 2.2.3. Comments regarding the NMR spectra of sumanene-ferrocene conjugate sumFc-2.

Several groups of multiplets and 4 singlets can be found in the <sup>1</sup>H NMR spectrum of sumanene-ferrocene conjugate **sumFc-2**. Those groups of signals are graphically presented in Figure S46 and are labeled in <sup>1</sup>H NMR spectrum presented in Figure S47.

Seven or twenty-four signals shall be found in the <sup>1</sup>H NMR spectrum of symmetrical and unsymmetrical stereoisomer of **sumFc-2**, respectively. Those signals overlap what results in the presence of several groups of multiplets in the spectrum.

The signals originating from the sumanene skeleton, ferrocene unit and phenylene linkage can be found in the spectrum. The total number of protons (48H), as well as the number of protons within given groups of signals, are consistent with the anticipated values. Noteworthy, the lack of signals coming from the benzylic protons of sumanene also confirms the successful modification of sumanene skeleton (for the reference <sup>1</sup>H NMR spectrum of sumanene, see Figure S1).

Four singlets located at 4.26, 4.25, 4.23 and 4.22 ppm were ascribed to the 15 protons of the unsubstituted  $C_p$  ring of ferrocene (symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s). <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis confirmed that those signals are indeed four singlets, not two dublets or dublet of dublets (see graphical representation in Figure S48 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S11).

Four multiplets located at 4.27-4.26, 4.30-4.28, 4.51-4.50 and 4.54-4.53 ppm were ascribed to the 12 protons of the substituted  $C_p$  ring of ferrocene (symmetrical diastereoisomer = 2 signals = 2 x t-like m, unsymmetrical diastereoisomer = 6 signals = 6 x t-like m). The presence of multiplets in the spectrum for those signals originates from the fact that the characteristic triplet-like multiplets coming from the substituted  $C_p$  ring of ferrocene overlap. Noteworthy, in the 4.27-4.25 ppm area of the spectrum, two singlets (unsubstituted  $C_p$  ring of ferrocene) and one multiplet (substituted  $C_p$  ring of ferrocene) overlap. This observation has been also confirmed with the <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis (see graphical representation in Figure S48 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S11).

Multiplets located at 7.30-7.29, 7.48-7.47 and 7.54-7.51 ppm were ascribed to the 9 protons of the sumanene skeleton (6H; symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 6 signals = 6 x d) and methidene moieties (3H; symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s). <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis further supported this hypothesis (see graphical representation in Figure S48 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S11).

Multiplets located at 7.58-7.53, 7.89-7.88 and 7.95-7.94 ppm were ascribed to the 12 protons of the phenylene linkage (symmetrical diastereoisomer = 2 signals = 2 x m, unsymmetrical diastereoisomer = 6 signals = 6 x m). This hypothesis has been clearly confirmed with the <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis (see graphical representation in Figure S48 and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S11). Multiplet located at 7.58-7.53 ppm overlap with the multiplet located at 7.54-7.51 ppm coming from the sumanene skeleton.



*Figure S46.* Graphical representation of the number of signals observed in the <sup>1</sup>H NMR spectra of sumanene-ferrocene conjugate **sumFc-2**. The same color does not correspond to the same chemical shift in the NMR spectrum.



*Figure S47.* <sup>1</sup>H NMR (500 MHz, THF-*d*<sub>8</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-2** with the signals ascribed to the given groups. For the labels, see Figure S46.



*Figure S48.* Graphical representation of the types of signals observed in the <sup>1</sup>H NMR spectrum of sumanene-ferrocene conjugate **sumFc-2** and the graphical representation of the cross-correlations in the <sup>1</sup>H-<sup>1</sup>H COSY NMR spectra together with the crucial insets of this spectra. For the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum, see Figure S11.

Eighteen groups of signals can be found in the <sup>13</sup>C{<sup>1</sup>H} NMR spectrum of sumaneneferrocene conjugate **sumFc-2**, what is consistent with the anticipated value. Those groups of signals are graphically presented in Figure S49 and are labeled in <sup>13</sup>C{<sup>1</sup>H} NMR spectrum presented in Figure S50.

Eighteen or fifty-four signals shall be found in the <sup>13</sup>C{<sup>1</sup>H} NMR spectrum of symmetrical and unsymmetrical stereoisomer of **sumFc-2**, respectively. Noteworthy, within most groups of signals in the <sup>13</sup>C{<sup>1</sup>H} NMR spectrum of **sumFc-2**, four singlets can be observed, what is highly consisted with the anticipated value (symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s), see, *e.g.*, the signal located at 65.2-65.1 ppm (<sup>13</sup>C nucleus of the acetylene linkage).

Signals located at 147.7-120.8 ppm were ascribed to the presence of various <sup>13</sup>C nucleus coming from the sumanene skeleton, phenylene linkage and methidene moieties.

Signals located at 90.6-65.1 ppm were ascribed to the presence of ferrocene units and acetylene linkages. Signals located at 70.2 ppm were ascribed to the <sup>13</sup>C nucleus of unsubstituted  $C_p$  ring of ferrocene. Signals located at 71.7 and 69.2-69.1 ppm were ascribed to the <sup>13</sup>C nucleus of substituted  $C_p$  ring of ferrocene. Signals located at 86.1-86.0 ppm were ascribed to the characteristic *ipso* <sup>13</sup>C nucleus of substituted  $C_p$  ring of ferrocene. Signals located at 90.6 ppm and 65.2-65.1 ppm were ascribed to the <sup>13</sup>C nucleus of acetylene linkage.







*Figure S50.* <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CHCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-2** with the signals ascribed to the given groups. For the labels, see Figure S49.

# 2.2.4. Comments regarding the NMR spectra of sumanene-ferrocene conjugate sumFc-3.

Several groups of multiplets and three groups of singlets can be found in the <sup>1</sup>H NMR spectrum of sumanene-ferrocene conjugate **sumFc-3**. Those groups of signals are graphically presented in Figure S51 and are labeled in <sup>1</sup>H NMR spectrum presented in Figure S52.

Nine or thirty signals shall be found in the <sup>1</sup>H NMR spectrum of symmetrical and unsymmetrical stereoisomer of **sumFc-3**, respectively. Those signals overlap what results in the presence of several groups of multiplets in the spectrum.

The signals originating from the sumanene skeleton, ferrocene unit, methylene group, phenylene and 1,2,3-triazole linkage can be found in the spectrum. The total number of protons (57H), as well as the number of protons within given groups of signals, are consistent with the anticipated values. Noteworthy, the lack of signals coming from the benzylic protons of sumanene also confirms the successful modification of sumanene skeleton (for the reference <sup>1</sup>H NMR spectrum of sumanene, see Figure S1).

Singlets located at 4.23-4.21 ppm were ascribed to the 15 protons of the unsubstituted  $C_p$  ring of ferrocene (symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s). <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis confirmed that those signals are indeed singlets, not two dublets or dublet of dublets (see graphical representation in Figure S53and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S27).

Multiplets located at 4.38-4.23 ppm were ascribed to the 12 protons of the substituted  $C_p$  ring of ferrocene (symmetrical diastereoisomer = 2 signals = 2 x t-like m, unsymmetrical diastereoisomer = 6 signals = 6 x t-like m). The presence of multiplets in the spectrum for those signals originates from the fact that the characteristic triplet-like multiplets coming from the substituted  $C_p$  ring of ferrocene overlap.

Four singlets located at 5.38-5.33 ppm were ascribed to the 6 protons of the methylene linkage (symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s). <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis confirmed that those signals are indeed four singlets, not two dublets or dublet of dublets (see graphical representation in Figure S53and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S27).

Multiplets located at 7.45-7.00 were ascribed to the 9 protons of the sumanene skeleton (6H; symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 6 signals = 6 x d) and methidene moieties (3H; symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s). <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis further supported this hypothesis (see graphical representation in Figure S53and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S27).

Four singlets located at 7.80-7.68 ppm were ascribed to the 3 protons of the 1,2,3-triazole linkage (symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s). <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis confirmed that those signals are indeed four singlets, not two dublets or dublet of dublets (see graphical representation in Figure S53and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S27).

Multiplets located at 7.93-7.82 ppm were ascribed to the 12 protons of the phenylene linkage (symmetrical diastereoisomer = 2 signals = 2 x m, unsymmetrical diastereoisomer = 6 signals = 6 x m). This hypothesis has been confirmed with the <sup>1</sup>H-<sup>1</sup>H COSY NMR analysis (see graphical representation in Figure S53and the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum in Figure S27).



*Figure S51.* Graphical representation of the number and types of signals observed in the <sup>1</sup>H NMR spectra of sumanene-ferrocene conjugate **sumFc-3**. The same color does not correspond to the same chemical shift in the NMR spectrum.



*Figure S52.* <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-3** with the signals ascribed to the given groups. For the labels, see Figure S51.



*Figure S53.* Graphical representation of the types of signals observed in the <sup>1</sup>H NMR spectrum of sumanene-ferrocene conjugate **sumFc-3** and the graphical representation of the cross-correlations in the <sup>1</sup>H-<sup>1</sup>H COSY NMR spectra together with the crucial insets of this spectra. For the full <sup>1</sup>H-<sup>1</sup>H COSY NMR spectrum, see Figure S27.

The successful synthesis of **sumFc-3** was also supported by the comparison between <sup>1</sup>H NMR (500 MHz, CDCl3) spectra of ferrocenemethylazide (**9**), tris[(4-ethynylphenyl)methidene]sumanene (**12**), and sumanene-ferrocene conjugate **sumFc-3**, see Figure S54. The following essential conclusions shall be noted:

- The signals (four singlets) located at 7.80–7.68 ppm in the spectrum of **sumFc-3** originate from the presence of the 1,2,3-triazole skeletons that were formed in the *click chemistry* process.
- The signals (four singlets) located at *ca*. 5.35 ppm in the spectrum of **sumFc-3** that originate from the presence of the methylene groups, are shifted in comparison to the respective singlet in the spectrum of ferrocenemethylazide (**9**; 4.10 ppm).
- No signals originating from the presence of acetylene groups (compound **12**, 3.22-3.16 ppm, four singlets) were found in the spectrum of **sumFc-3**.



7.90 7.85 7.80 7.75 7.60 7.55 7.50 7.45 7.40 7.35 7.30 7.25 7.20 7.15 7.10 7.05 5.40 5.304.40 4.35 4.30 4.25 4.20 4.15 4.10 3.20 3.1

*Figure S54.* Comparison between <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) spectra of ferrocenemethylazide (**9**; red), tris[(4-ethynylphenyl)methidene]sumanene (**12**; green), and sumanene-ferrocene conjugate (**sumFc-3**; blue). The crucial signals were ascribed to the given groups with the labels.

Nineteen groups of signals can be found in the <sup>13</sup>C{<sup>1</sup>H} NMR spectrum of sumaneneferrocene conjugate **sumFc-3**, what is consistent with the anticipated value. Those groups of signals are graphically presented in Figure S55 and are labeled in <sup>13</sup>C{<sup>1</sup>H} NMR spectrum presented in Figure S56.

Nineteen or fifty-seven signals shall be found in the <sup>13</sup>C{<sup>1</sup>H} NMR spectrum of symmetrical and unsymmetrical stereoisomer of **sumFc-3**, respectively. Noteworthy, within most groups of signals in the <sup>13</sup>C{<sup>1</sup>H} NMR spectrum of **sumFc-3**, four singlets can be observed, what is highly consisted with the anticipated value (symmetrical diastereoisomer = 1 signal = 1 x s, unsymmetrical diastereoisomer = 3 signals = 3 x s), see, *e.g.*, the signal located at 81.1-81.0 ppm (characteristic *ipso* <sup>13</sup>C nucleus of substituted Cp ring of ferrocene).

Signals located at 147.6-119.5 ppm were ascribed to the presence of various <sup>13</sup>C nucleus coming from the sumanene skeleton, methidene moieties, phenylene and 1,2,3-triazole skeletons.

Signals located at 81.1-81.0 ppm were ascribed to the presence of ferrocene units. Signals located at 81.1-81.0 ppm were ascribed to the characteristic *ipso* <sup>13</sup>C nucleus of unsubstituted Cp ring of ferrocene. Signals located at 69.3-69.1 ppm were ascribed to the <sup>13</sup>C nucleus of substituted and unsubstituted Cp ring of ferrocene.

Direct comparison of <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CDCl<sub>3</sub>) spectra of 4-(1-(ferrocenylmethyl)-1*H*-1,2,3-triazol-4-yl)benzaldehyde (**11**; blue), and sumanene-ferrocene conjugate (**sumFc-3**; red) is presented in Figure S57. It can be concluded that the <sup>13</sup>C{<sup>1</sup>H} spectrum of the target sumanene-ferrocene conjugate **sumFc-3** comprises the similar groups of signals as the reference spectrum of **11** (excluding the **C**HO signal located at 191.8 ppm).



*Figure S55.* Graphical representation of the number and types of signals observed in the <sup>13</sup>C{<sup>1</sup>H} NMR spectra of sumanene-ferrocene conjugate **sumFc-3**. The same color does not correspond to the same chemical shift in the NMR spectrum. The same color does not correspond to the same chemical shift in the NMR spectrum.



*Figure S56.* <sup>13</sup>C{<sup>1</sup>H} NMR (125 MHz, CHCl<sub>3</sub>) spectrum of sumanene-ferrocene conjugate **sumFc-3** with the signals ascribed to the given groups. For the labels, see Figure S55.



sumanene-ferrocene conjugate (sumFc-3; red).

### 3. HRMS spectra



Figure S58. ESI-HRMS(TOF) spectrum of sumanene-ferrocene conjugate sumFc-1.



Figure S59. ESI-HRMS(TOF) spectrum of sumanene-ferrocene conjugate sumFc-2.



Figure S60. ESI-HRMS(TOF) spectrum of tris[(4-iodophenyl)methidene]sumanene (8a).



Figure S61. ESI-HRMS(TOF) spectrum of tris[(4-bromophenyl)methidene]sumanene (8b).


*Figure S63.* ESI-HRMS(TOF) spectrum of tris[(4-ethynylphenyl)methidene]sumanene (12).



Figure S64. ESI-HRMS(TOF) spectrum of sumanene-ferrocene conjugate sumFc-3.



Figure S65. ESI-HRMS(TOF) spectrum of compound 14.



Figure S66. ESI-HRMS(TOF) spectrum of compound 16.

## 4. UV-vis and emission spectra



*Figure* **S67.** UV-vis spectrum of sumanene-ferrocene conjugate **sumFc-1** (concentration: 0.02 mM, solvent: CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 v/v).



*Figure S68.* Emission spectrum of sumanene-ferrocene conjugate **sumFc-1** (concentration: 0.02 mM, solvent: CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 v/v;  $\lambda_{ex}$  = 380 nm).



*Figure S69.* UV-vis spectrum of sumanene-ferrocene conjugate **sumFc-2** (solvent: CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 v/v).



*Figure S70.* Emission spectrum of sumanene-ferrocene conjugate **sumFc-2** (concentration: 0.02 mM, solvent: CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 v/v;  $\lambda_{ex}$  = 280 nm).



*Figure S71.* UV-vis spectrum of sumanene-ferrocene conjugate **sumFc-3** (concentration: 0.02 mM, solvent: CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 v/v).



*Figure S72.* Emission spectrum of sumanene-ferrocene conjugate **sumFc-3** (concentration: 0.02 mM, solvent: CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 v/v;  $\lambda_{ex}$  = 370 nm).



*Figure S73.* UV-vis spectrum of compound **14** (concentration: 0.02 mM, solvent: CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 v/v).



*Figure* **S74.** Emission spectrum of compound **14** (concentration: 0.02 mM, solvent: CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 v/v;  $\lambda_{ex} = 290$  nm).



*Figure S75.* UV-vis spectrum of compound **16** (concentration: 0.02 mM, solvent: CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 v/v).



*Figure* **S76.** Emission spectrum of compound **16** (concentration: 0.02 mM, solvent: CHCl<sub>3</sub>:CH<sub>3</sub>OH = 1:1 v/v;  $\lambda_{ex} = 290$  nm).

## 5. Spectrofluorimetric analyzes on the interactions between sumanene-ferrocene conjugates sumFc-1 – sumFc-3 and cesium cations

In order to provide the solubility of both sumFc-1 - sumFc-3 and cesium salt, the measurements were carried out in methanol-chloroform mixture (1:1 v/v). Appropriate volumes of 1 mM CsCl solution were mixed with 1 mM solution of sumanene-ferrocene conjugates (sumFc-1 - sumFc-3) to reach given sumanene-to-metal cation molar ratio. The excitation wavelength was 380 nm, 280 nm, 370 nm for sumFc-1, sumFc-2, and sumFc-3, respectively. The concentration of sumanene-ferrocene conjugate in each sample was 0.02 mM.

Complex stoichiometries were estimated with the Job's plot (method of continuous variation).<sup>9–12</sup>

The apparent binding constants ( $K_{app}$ ) were estimated with the Benesi-Hildebrand method<sup>13,14</sup>, using the following equation:

$$\frac{1}{I-I_0} = \frac{1}{a} + \frac{1}{a \cdot K_{app} \cdot C(\mathrm{Cs}^+)}$$

where  $I_0$  and I are the fluorescence intensities of sumanene-ferrocene conjugate in the absence and presence of cesium cations, respectively, a is a constant, and  $C(Cs^+)$  is the concentration of cesium cations in solution. The association constant was determined as a ratio of intercept-to-slope of  $1/(I - I_0)$  vs.  $1/C(Cs^+)$  linear plot.

The above-discussed spectra and data are presented below.



*Figure* S77. Emission spectra ( $\lambda_{ex}$  = 380 nm) of sumanene-ferrocene conjugate **sumFc-1** in the presence of various amounts (equivalents = eq) of cesium cations.



Figure S78. Job's plot regarding the interactions between sumanene-ferrocene conjugate sumFc-1 and cesium cations (x stands for the molar fraction of sumaneneferrocene conjugate sumFc-1; l<sub>0</sub> and l are the fluorescence intensities of sumanene-ferrocene conjugate sumFc-1 in the absence and in the presence of cesium cations).



*Figure S79.* Benesi-Hildebrand plot regarding the interactions between sumaneneferrocene conjugate **sumFc-1** and cesium cations (*C* stands for the molar concentration of cesium cations in the sample; *I*<sub>0</sub> and *I* are the fluorescence intensities of sumanene-ferrocene conjugate **sumFc-1** in the absence and in the presence of cesium cations). The linear fit data and the calculated *K*<sub>app</sub> are also presented.



Figure S80. Emission spectra (λ<sub>ex</sub> = 380 nm) of sumanene-ferrocene conjugate sumFc-1 in the absence and in the presence of rubidium cations (10 eq); spectrum of sumanene-ferrocene conjugate sumFc-1 in the presence of cesium cations (10 eq) is also presented for comparison.



*Figure S81.* Emission spectra ( $\lambda_{ex}$  = 280 nm) of sumanene-ferrocene conjugate **sumFc-2** in the presence of various amounts (equivalents = eq) of cesium cations.



Figure S82. Job's plot regarding the interactions between sumanene-ferrocene conjugate sumFc-2 and cesium cations (x stands for the molar fraction of sumaneneferrocene conjugate sumFc-2; l<sub>0</sub> and l are the fluorescence intensities of sumanene-ferrocene conjugate sumFc-2 in the absence and in the presence of cesium cations).



*Figure S83.* Benesi-Hildebrand plot regarding the interactions between sumaneneferrocene conjugate **sumFc-2** and cesium cations (*C* stands for the molar concentration of cesium cations in the sample; *I*<sub>0</sub> and *I* are the fluorescence intensities of sumanene-ferrocene conjugate **sumFc-2** in the absence and in the presence of cesium cations). The linear fit data and the calculated *K*<sub>app</sub> are also presented.



Figure S84. Emission spectra (λ<sub>ex</sub> = 280 nm) of sumanene-ferrocene conjugate sumFc-2 in the absence and in the presence of rubidium cations (10 eq); spectrum of sumanene-ferrocene conjugate sumFc-2 in the presence of cesium cations (10 eq) is also presented for comparison.



*Figure S85.* Emission spectra ( $\lambda_{ex}$  = 370 nm) of sumanene-ferrocene conjugate **sumFc-3** in the presence of various amounts (equivalents = eq) of cesium cations.



*Figure S86.* Job's plot regarding the interactions between sumanene-ferrocene conjugate **sumFc-3** and cesium cations (*x* stands for the molar fraction of sumanene-ferrocene conjugate **sumFc-3**; *l*<sub>0</sub> and *l* are the fluorescence intensities of sumanene-ferrocene conjugate **sumFc-3** in the absence and in the presence of cesium cations).



*Figure S87.* Benesi-Hildebrand plot regarding the interactions between sumaneneferrocene conjugate **sumFc-3** and cesium cations (*C* stands for the molar concentration of cesium cations in the sample; *I*<sub>0</sub> and *I* are the fluorescence intensities of sumanene-ferrocene conjugate **sumFc-3** in the absence and in the presence of cesium cations). The linear fit data and the calculated *K*<sub>app</sub> are also presented.



*Figure S88.* Emission spectra (λ<sub>ex</sub> = 370 nm) of sumanene-ferrocene conjugate **sumFc-3** in the absence and in the presence of rubidium cations (10 eq); spectrum of sumanene-ferrocene conjugate **sumFc-3** in the presence of cesium cations (10 eq) is also presented for comparison.

## 6. Calculation

The calculations were performed for the representative symmetrical diastereoisomer of sumanene-ferrocene conjugate **sumFc-3** and its complex with Cs<sup>+</sup>. Theoretical calculations were performed using *Gaussian16* program.<sup>15</sup>

The structure of symmetrical diastereoisomer of sumanene-ferrocene conjugate **sumFc-3** was optimized using B3LYP functional<sup>16</sup> – Density Functional Theory (DFT) with 6-311++G(d,p) basis set<sup>17</sup>. The initial structure was generated manually in the *GaussView* program<sup>18</sup>. For the DFT-optimized structure of symmetrical diastereoisomer of sumanene-ferrocene conjugate **sumFc-3**, see Figure S89.

To illustratively visualize the potential structure of the formed sandwich complexes between two sumanene bowls within **sumFc-3** (representative symmetrical diastereoisomer) and Cs<sup>+</sup>, the optimization using Gaussian16 program was also performed. Notably, sandwich complex consists of 301 atoms in total, including a heavy, electron-rich cesium cation. Thus, the semi-empirical PM6 method<sup>19–21</sup> was used in order to illustratively visualize the approximate geometry of the molecule. For the optimized structure of two symmetrical diastereoisomers of sumanene-ferrocene conjugate **sumFc-3** and Cs<sup>+</sup>, see Figure S90.



*Figure S89.* DFT-optimized structure of the sumanene-ferrocene conjugate **sumFc-3** (symmetrical diastereoisomer) viewed from two different perspectives. Hydrogens were omitted for clarity.



*Figure S90.* PM6-optimized structures of the formed sandwich complex between the representative two symmetrical diastereoisomers of sumanene-ferrocene conjugate **sumFc-3** and Cs<sup>+</sup>. For clarity, hydrogens are omitted and the sumanene bowls are marked yellow. The views from two different perspectives are presented.

## 7. References

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