Electronic Supporting Information

Synthesis of Sm³⁺-doped YGa_{1.5}Al_{1.5}(BO₃)₄ phosphor via mechanical activation-assisted solid-state reaction

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Fig. S1 The median sizes, particle size distributions and specific surface areas of YGAB:0.03Sm³⁺ precursor mixtures ground for different time.



Fig. S2 (a) EDX analysis of YGAB:0.03Sm³⁺ precursor mixtures unground and (b) ground for 45min.



Fig. S3 (a) Amplified XRD details of YGAB:0.03Sm³⁺ phosphors synthesized from precursor mixtures ground for different time and then calcined at 1000 °C (32.5°-33.5°), and (b-c) SEM images of YGAB:0.03Sm³⁺ phosphors synthesized with precursor mixtures unground and ground for 45 min and then calcined at 1050 °C.



Fig. S4 The quantum yield (QY) of YGAB:0.03Sm³⁺ phosphor synthesized by calcination of precursor mixture ground at different time (under 405 nm excitation)



Fig. S5 Energy level transition diagram of Sm³⁺



Fig. S6 PL spectra of the white-light phosphor based on the combination of $BaMgAl_{10}O_{17}:Eu^{2+}$, $Y_3Al_5O_{12}:Ce^{3+}$, Ga^{3+} and $YGa_{1.5}Al_{1.5}(BO_3)_4$: Sm^{3+} under near-UV

excitation (inset: the fabricated warm white light with the CIE chromaticity coordinates of (0.3820, 0.3872)).

Table S1 Concentrations of Sm^{3+} in the filtrates of precursor mixtures ground for different time dissolved in nitric acid, measured by ICP-OES.

Dissolution time	Grinding time	Concentration of Sm^{3+} /mg·L ⁻¹	
/min	/min		
25	0	2.36	
	15	2.27	
	30	1.94	
40	0	2.63	
	15	2.55	
	30	2.12	
60	0	4.43	
	15	2.58	
	30	2.95	
	45	3.02	

Table S2 The crystallographic parameters of YGAB:0.03Sm³⁺ phosphors synthesized from the precursor mixtures ground for different time and calcined at 1000 °C.

Grinding Time(min)	<i>a=b</i> (Å)	c(Å)	$V(\text{\AA}^3)$
0	9.3504	7.3193	554.20
15	9.3480	7.3311	554.80
30	9.3541	7.3380	556.05
45	9.3528	7.3360	555.74
60	9.3513	7.3323	555.28