

## ***Supplementary Information***

### **Luminescent and magnetic [TbEu] 2D Metal-Organic Frameworks**

E. Bartolomé<sup>a\*</sup>, A. Arauzo<sup>b</sup>, S. Fuertes<sup>c</sup>, L. Navarro-Spreafico<sup>d</sup>, P. Sevilla<sup>a</sup>, H. Fernández Cortés,<sup>d</sup> N. Settineri,<sup>e</sup> S. J. Teat,<sup>e</sup> E. C. Sañudo<sup>d,f\*</sup>

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<sup>a</sup> Escola Universitària Salesiana de Sarrià (EUSS), Passeig Sant Joan Bosco 74, 08017 Barcelona, Spain

<sup>b</sup> Instituto de Nanociencia y Materiales de Aragón (INMA), CSIC-Universidad de Zaragoza, and Departamento de Física de la Materia Condensada, 50009 Zaragoza, Spain

<sup>c</sup> Departamento de Química Inorgánica, Facultad de Ciencias, Instituto de Síntesis Química y Catalisis, Homogénea (ISQCH), CSIC-Universidad de Zaragoza, 50009 Zaragoza, Spain

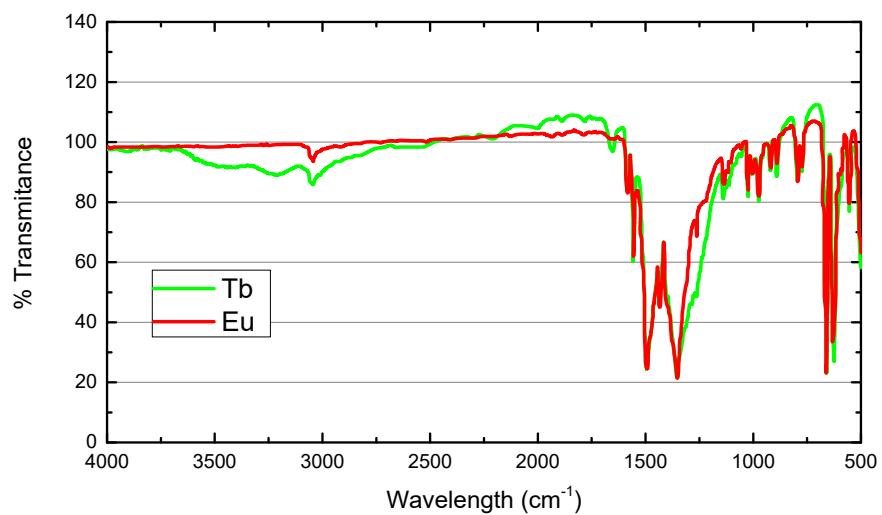
<sup>d</sup> Secció de Química Inorgànica, Departament de Química Inorgànica i Orgànica Universitat de Barcelona, C/Martí i Franquès, 1-11, 08028 Barcelona, Spain

<sup>e</sup> ALS Berkeley USA

<sup>f</sup> Institut de Nanociència i Nanotecnologia, Universitat de Barcelona, IN2UB. C/Martí i Franquès, 1-11, 08028 Barcelona, Spain

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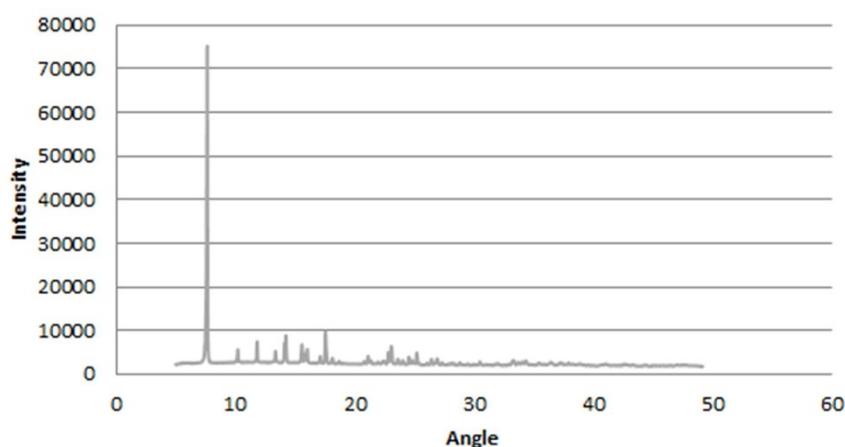
**S1. Infrared spectra of Ln 2D MOFs**



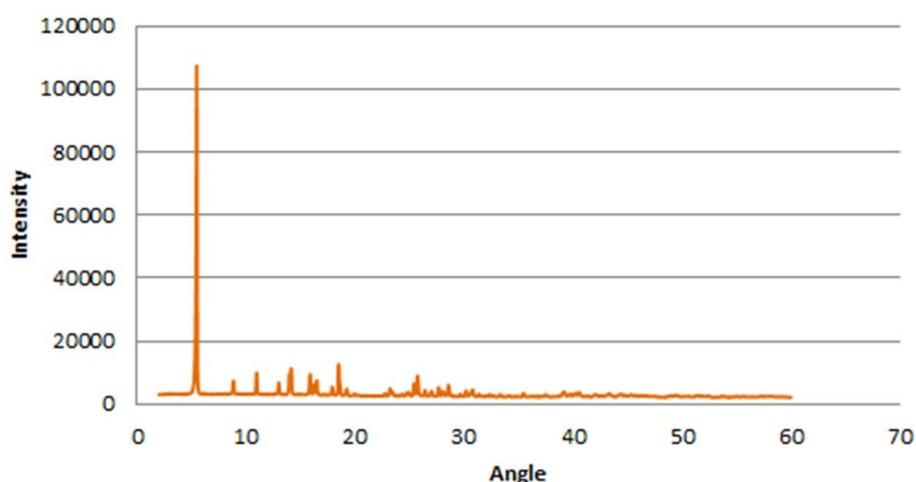
**Figure S1.** Infrared spectra of homometallic Ln 2DMOFs (Ln= Tb, Eu).

## S2. PXRD

Dy



Tb<sub>0.3</sub>Eu<sub>0.7</sub>



Tb<sub>0.2</sub>Eu<sub>0.8</sub>

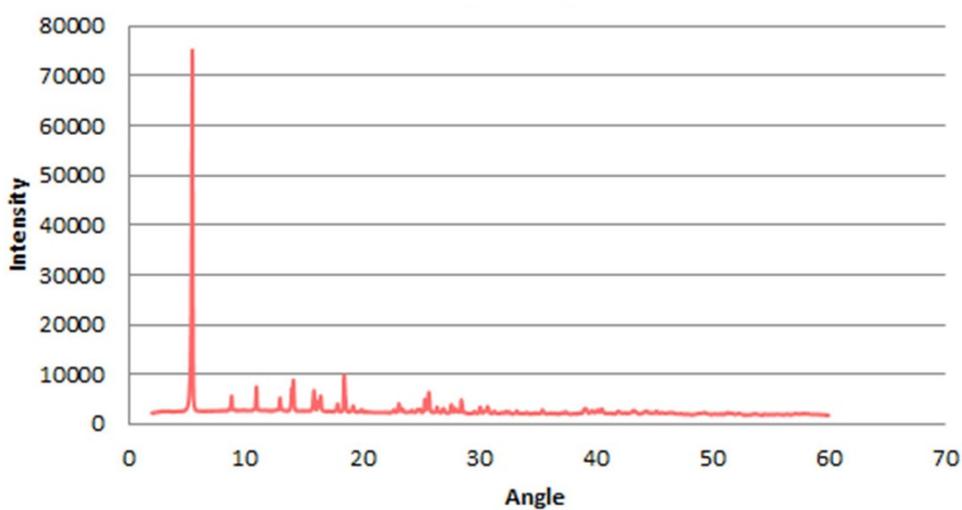
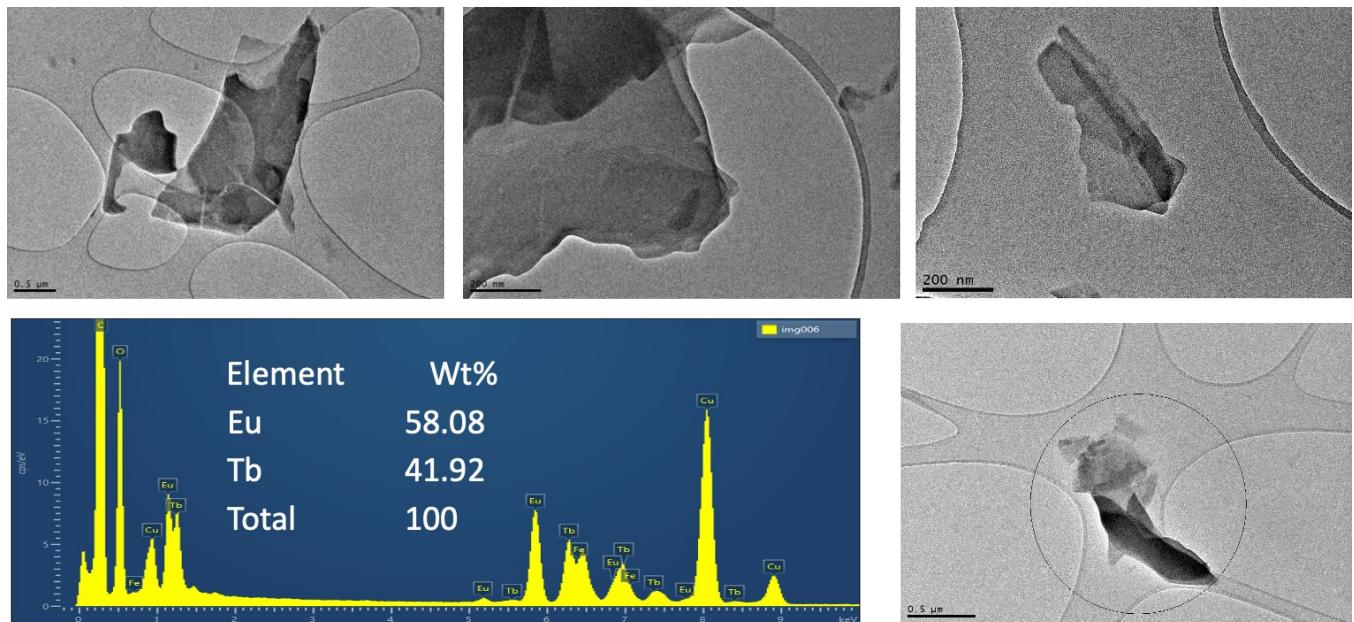


Figure S2. PXRD spectra of the studied mixed compounds (the rest are shown in Fig. 2 of the main paper).

### S3. Tb/Eu ratio analysis

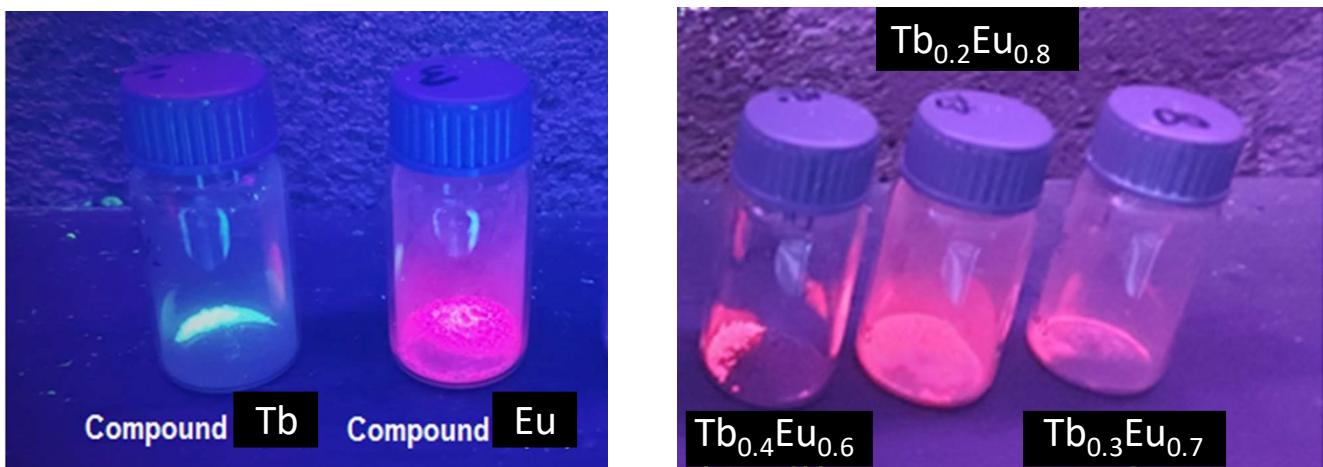
Sample name	ICP		SEM-EDS		TEM-EDX		Fluorescence	
	Tb	Eu	Tb	Eu			Tb	Eu
$\text{Tb}_x\text{Eu}_{1-x}$								
$\text{Tb}_{0.2}\text{Eu}_{0.8}$	0.17	0.83	0.25	0.75				
$\text{Tb}_{0.3}\text{Eu}_{0.7}$	0.28	0.72	0.36	0.64				
$\text{Tb}_{0.4}\text{Eu}_{0.6}$	0.42	0.58			0.42	0.58		
$\text{Tb}_{0.7}\text{Eu}_{0.3}$	0.71	0.29					0.73	0.27
$\text{Tb}_{0.9}\text{Eu}_{0.1}$	0.86	0.14					0.88	0.13

**Table S3** Tb/Eu ratios determined for the heteronuclear samples by ICP (standard error <2%), SEM-EDS, TEM-EDX and Fluorescence.



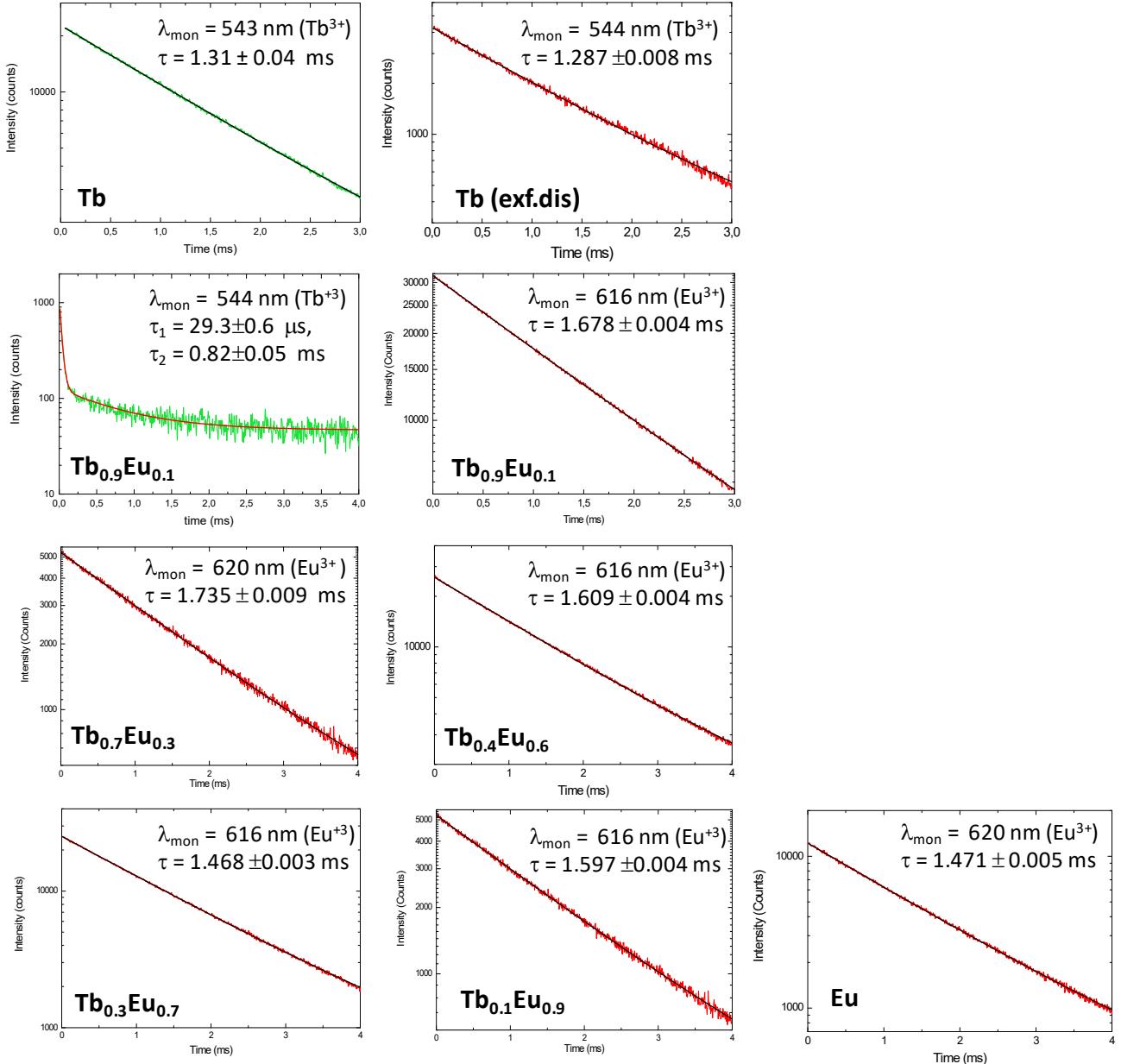
**Figure S3.** TEM images showing exfoliated flakes of  $\text{Tb}_{0.4}\text{Eu}_{0.6}$  and EDX analysis.

### S4. Color emission under UV light

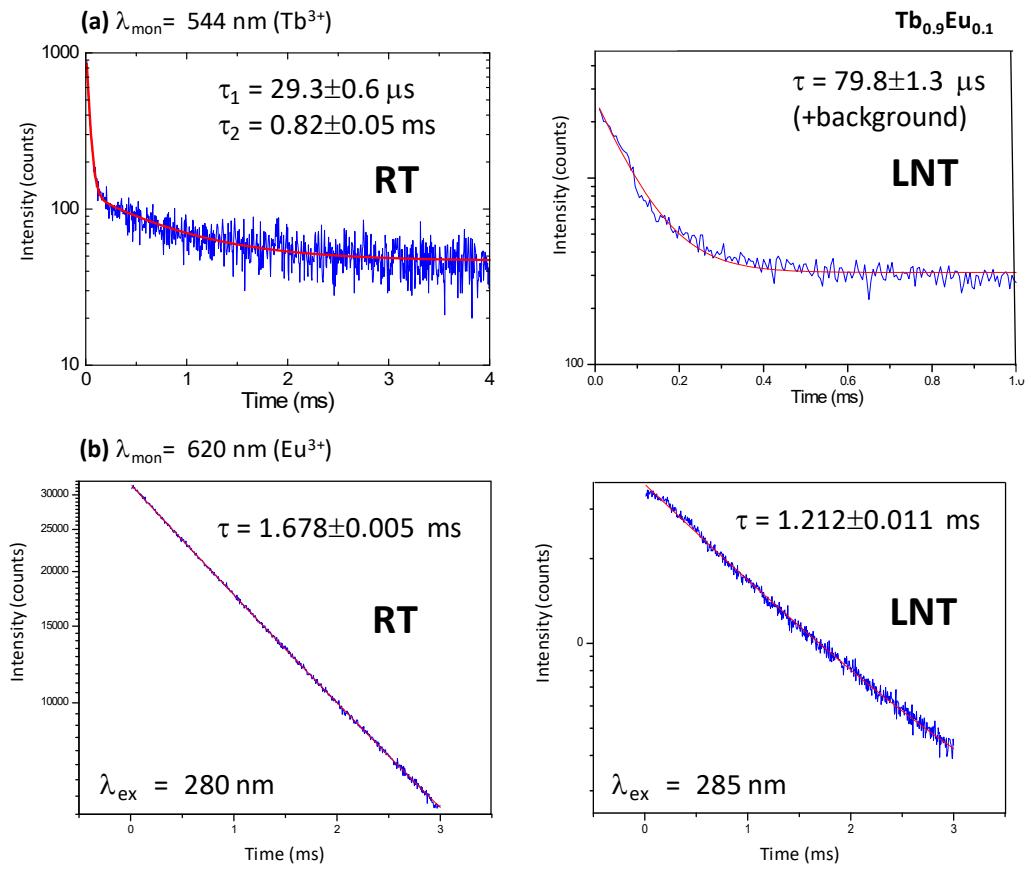


**Figure S4.** Visible color emission of (left) homonuclear **Tb**, **Eu** compounds, and (right) heteronuclear [TbEu] compounds, under UV light.

## S5. Luminescence lifetime measurements

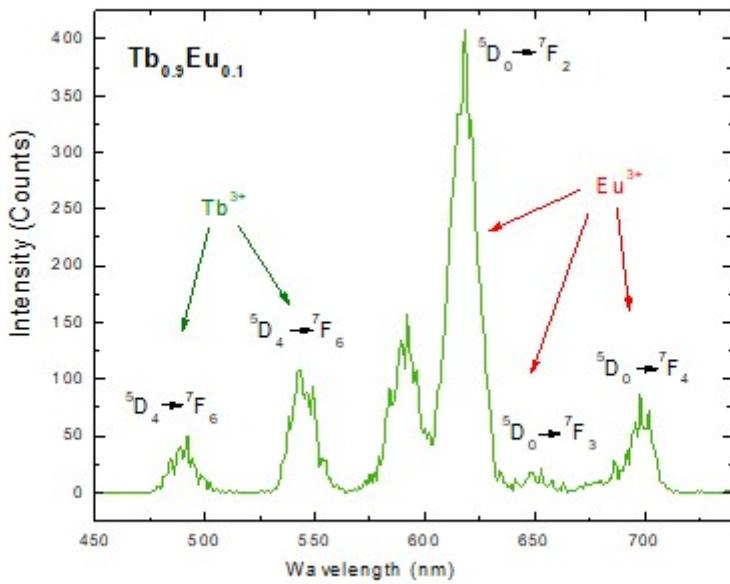


**Figure S5.** Lifetime measurements for homonuclear Tb and Eu compounds, and heterodinuclear Tb<sub>x</sub>Eu<sub>1-x</sub> compounds, excited at  $\lambda_{\text{exc}}=280$  nm. The decay of either the Eu<sup>3+</sup> main peak at 620 nm or the Tb<sup>3+</sup> peak at 544 nm were monitored. For Tb<sub>0.9</sub>Eu<sub>0.1</sub> the lifetime data were fit to a biexponential law with two time constants ( $\tau_1$  and  $\tau_2$ ):  $I(t) = A_1 \exp(-t/\tau_1) + A_2 \exp(-t/\tau_2) + I_0$ , with  $A_1 = 1022 \pm 16$ ,  $\tau_1 = 29.3 \pm 0.6$  μs,  $A_2 = 80 \pm 2$ ,  $\tau_2 = 0.82 \pm 0.05$  μs,  $I_0 = 46.5 \pm 0.8$ . For all other compounds the data were fit to an exponential decay law with a single time constant ( $\tau$ ).



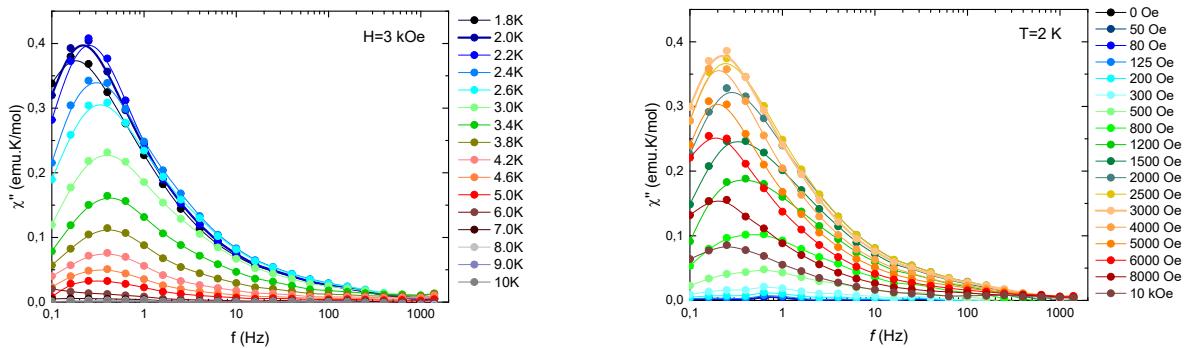
**Figure S6.** Lifetime measurements for heterodinuclear **Tb<sub>0.9</sub>Eu<sub>0.1</sub>** compound, monitored at (a) the Tb<sup>3+</sup> main peak at 544 nm and (b) the Eu<sup>3+</sup> main peak at 620 nm peak at, at (left) room temperature (RT), and (right) liquid nitrogen temperature (LNT).

## S6. Emission of mixed compound $\text{Tb}_{0.9}\text{Eu}_{0.1}$



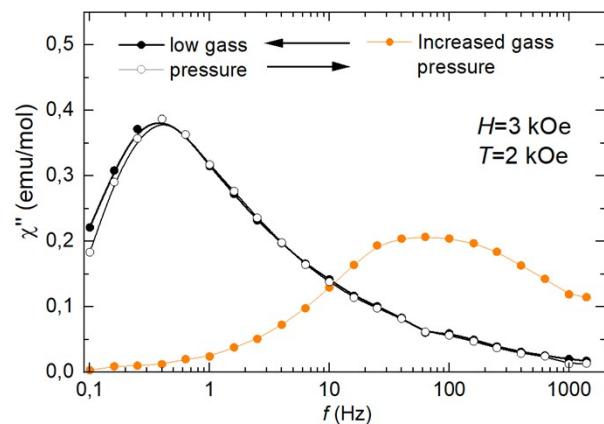
**Figure S7.** Emission spectra of complex  $\text{Tb}_{0.9}\text{Eu}_{0.1}$  excited at  $\lambda_{\text{exc}}=280$  nm, measured in. Fluorolog FL-1057, Jobin Yvon HORIBA. The characteristic emission bands for  $\text{Tb}^{3+}$  and  $\text{Eu}^{3+}$  are visible.

## S7. Ac susceptibility ox mixed compound $\text{Tb}_{0.9}\text{Eu}_{0.1}$



**Figure S8.** Ac susceptibility results. (Left)  $\chi''(f, T)$  at constant magnetic field  $H=3$  kOe and (Right)  $\chi''(f, H)$  at constant  $T=2$  K for mixed compound  $\text{Tb}_{0.9}\text{Eu}_{0.1}$ .

## S8. Bottleneck effect



**Fig. S9**  $\chi''(f)$  measurements on pure Tb compound at  $H = 3$  kOe and  $T = 2$  K at different experimental SQUID pressure conditions, showing the influence of the bottleneck effect.