## **Electronic Supplementary Information**

## Impact of the Zinc complexation of polytopic polyaza ligands on the interaction with double and single stranded DNA/RNA and antimicrobial activity

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Formula	C66H106Cl9N22NaO30Zn4		
Formula weight	2291.33		
Crystal system	Orthorhombic		
Space group	Pbcn		
Cell <i>a</i> / Å <i>b</i> / Å <i>c</i> / Å <i>V</i> / Å <sup>3</sup>	13.1143(2) 18.5283(2) 38.6732(5) 9397.2(2)		
Ζ	8		
Т / К	293(2)		
size / mm	$0.80 \times 0.34 \times 0.06$		
F <sub>000</sub>	4720		
density / g.cm <sup>-3</sup>	1.620		
R(int)	0.1013		
$ \begin{array}{c} \theta_{max} \ / \ deg. \\ \theta_{min} \ / \ deg. \end{array} $	27.498 3.045		
reflections - collected - unique	58315 10672		
$R1 - all - F2 > 2\sigma_{F2}$	0.1386 0.0669		
$wR2$ - all - $F^2 > 2\sigma_{F2}$	0.2313 0.1978		
GoF - all - $F^2 > 2\sigma_{F2}$	1.045 1.047		
<ul><li> parameters</li><li> constraints</li><li> restraints</li></ul>	596 0 50		

**Table S1.-** Crystallographic data of the "supercomplex"  $\{[Zn_2PYPODCl]_2Na(ClO_4)\}^{6+}$ .

distances	Zn1 - N1	2.090(5)	Zn2 - N10	2.232(4)
	Zn1 - N2	2.285(5)	Zn2 - N6	2.244(4)
	Zn1 - N3	2.182(5)	Zn2 - N7	2.065(4)
	Zn1 - N4	2.209(5)	Zn2 - N8	2.257(4)
	Zn1 – N5	2.196(4)	Zn2 - N9	2.237(4)
	Zn1 - Cl5	2.356(2)	Zn2-N11	2.086(4)
angles	N1 - Zn1 - N2	76.2(2)	N10 - Zn2 - N6	159.19(15)
	N1 - Zn1 - N3	78.1(2)	N10 - Zn2 - N8	80.20(14)
	N1 - Zn1 - N4	91.00(19)	N10 - Zn2 - N9	81.45(16)
	N1 - Zn1 - N5	172.07(17)	N6 - Zn2 - N8	100.08(14)
	N1 - Zn1 - Cl5	93.89(15)	N7 - Zn2 - N10	92.69(14)
	N2 - Zn1 - Cl5	107.79(17)	N7 - Zn2 - N6	107.76(14)
	N3 - Zn1 - N2	146.5(2)	N7 - Zn2 - N8	77.65(15)
	N3 - Zn1 - N4	81.0(2)	N7 - Zn2 - N9	77.82(16)
	N3 - Zn1 - N5	101.2(2)	N7 - Zn2 - N11	173.84(15)
	N3 - Zn1 - Cl5	94.95(17)	N9 - Zn2 - N6	106.09(15)
	N4 - Zn1 - N2	78.4(2)	N9 - Zn2 - N8	148.46(15)
	N4 - Zn1 - Cl5	172.90(14)	N11 - Zn2 - N10	81.35(15)
	N5 - Zn1 - N2	101.39(18)	N11 - Zn2 - N6	78.09(15)
	N5 - Zn1 - N4	81.09(17)	N11 - Zn2 - N8	99.68(15)
	N5 - Zn1 - Cl5	94.05(12)	N11 - Zn2 - N9	102.65(16)
1	1		1	

Table S2.- Selected distances and angles of the complex  $[Zn_2PYPODC1]^{3+}$ 



**Figure S1.-** Fluorescence titrations of (a) Zn**PHENPOD**, (b)  $Zn_2$ **PHENPOD** and (c)  $Zn_3$ **PHENPOD** with Calf Thymus DNA (ctDNA) at pH 7.4 in cacodylate buffer 0.05 M. (d) Plot of the fluorescence intensity normalized *vs.* metal complex [ctDNA]/[Zn<sub>x</sub>**PHENPOD**].



**Figure S2.-** Fluorescence titrations of (a) Zn**PHENPOD**, (b)  $Zn_2$ **PHENPOD** and (c)  $Zn_3$ **PHENPOD** with poly rA-poly rU at pH 7.4 in cacodylate buffer 0.05 M. (d) Plot of the fluorescence intensity normalized *vs.* metal complex [poly rA-poly rU]/[Zn<sub>x</sub>**PHENPOD**].



**Figure S3.-** Fluorescence titrations of (a)  $Zn_2$ **PHENPOD** and (b)  $Zn_3$ **PHENPOD** with poly dA-poly dT at pH 7.4 in cacodylate buffer 0.05 M. (c) Plot of the fluorescence intensity normalized *vs.* metal complex [poly dA-poly dT]/[Zn<sub>x</sub>**PHENPOD**].



**Figure S4.-** Fluorescence titrations of (a) Zn**PHENPOD**, (b)  $Zn_2$ **PHENPOD** and (c)  $Zn_3$ **PHENPOD** with poly dG-poly dC at pH 7.4 in cacodylate buffer 0.05 M. (d) Plot of the fluorescence intensity normalized *vs.* metal complex [poly dG-poly dC]/[Zn<sub>x</sub>**PHENPOD**].



**Figure S5.-** Fluorescence titrations of (a) ZnPHENPOD, (b)  $Zn_2PHENPOD$  and (c)  $Zn_3PHENPOD$  with poly dA at pH 7.4 in cacodylate buffer 0.05 M with poly dA. (d) Plot of the fluorescence intensity normalized *vs.* metal complex [poly A]/[Zn<sub>x</sub>PHENPOD].



**Figure S6.-** Fluorescence titrations of (a) Zn**PHENPOD**, (b)  $Zn_2$ **PHENPOD** and (c)  $Zn_3$ **PHENPOD** with poly dA at pH 7.4 in cacodylate buffer 0.05 M with poly dC. (d) Plot of the fluorescence intensity normalized *vs.* metal complex [poly dC]/[Zn<sub>x</sub>**PHENPOD**].



**Figure S7.-** Fluorescence titrations of (a) Zn**PHENPOD**, (b)  $Zn_2$ **PHENPOD** and (c)  $Zn_3$ **PHENPOD** with poly dA at pH 7.4 in cacodylate buffer 0.05 M with poly dG. (d) Plot of the fluorescence intensity normalized *vs.* metal complex [poly dG]/[Zn<sub>x</sub>**PHENPOD**].



**Figure S8.-** Fluorescence titrations of (a) ZnPHENPOD, (b)  $Zn_2PHENPOD$  and (c)  $Zn_3PHENPOD$  with poly U at pH 7.4 in cacodylate buffer 0.05 M. (d) Plot of the fluorescence intensity normalized *vs*. metal complex [poly U]/[ $Zn_xPHENPOD$ ].



**Figure S9.** CD titration of *ctDNA* ( $c = 1.0 \times 10^{-5}$  mol dm<sup>-3</sup>) with (a) **PHENPOD** and (b) Zn<sub>2</sub>**PHENPOD** at molar ratios r = [compound] / [polynucleotide] (buffer sodium cacodylate, I = 0.05 mol dm<sup>-3</sup>) at pH 7.4.



**Figure S10.** CD titration of poly rA-poly rU ( $c = 1.0 \times 10^{-5}$  M) with (a) **PHENPOD** and (b) Zn<sub>2</sub>**PHENPOD** at molar ratios r = [compound] / [polynucleotide] (buffer sodium cacodylate, I = 0.05 M) at pH 7.4.



**Figure S11.** CD titration of poly dA-poly dT ( $c = 1.0 \times 10^{-5}$  M) with (a) **PHENPOD** and (b) Zn<sub>2</sub>**PHENPOD** at molar ratios r = [compound] / [polynucleotide] (buffer sodium cacodylate, I = 0.05 M) at pH 7.4.



**Figure S12.** CD titration of poly dG-poly dC ( $c = 1.0 \times 10^{-5}$  M) with (a) **PHENPOD** and (b) Zn<sub>2</sub>**PHENPOD** at molar ratios r = [compound] / [polynucleotide] (buffer sodium cacodylate, I = 0.05 M) at pH 7.4.



**Figure S13.** CD titration of poly d(G-C)<sub>2</sub> ( $c = 1.0 \times 10^{-5}$  M) with (a) **PHENPOD** and (b) Zn<sub>2</sub>**PHENPOD** at molar ratios r = [compound] / [polynucleotide] (buffer sodium cacodylate, I = 0.05 M) at pH 7.4.



**Figure S14.** CD titration of poly dU ( $c = 1.0 \times 10^{-5}$  M) with (a) **PHENPOD** and (b) Zn<sub>2</sub>**PHENPOD** at molar ratios r = [compound] / [polynucleotide] (buffer sodium cacodylate, I = 0.05 M) at pH 7.4.



**Figure S15.** CD titration of poly dC ( $c = 1.0 \times 10^{-5}$  M) with (a) **PHENPOD** and (b) Zn<sub>2</sub>**PHENPOD** at molar ratios r = [compound] / [polynucleotide] (buffer sodium cacodylate, I = 0.05 M) at pH 7.4.



**Figure S16.** CD titration of poly dC ( $c = 1.0 \times 10^{-5}$  M) with (a) **PHENPOD** and (b) Zn<sub>2</sub>**PHENPOD** at molar ratios r = [compound] / [polynucleotide] (buffer sodium cacodylate, I = 0.05 M) at pH 7.4.



**Figure S17.** CD titration of poly dA ( $c = 1.0 \times 10^{-5}$  M) with (a) **PHENPOD** and (b) Zn<sub>2</sub>**PHENPOD** at molar ratios r = [compound] / [polynucleotide] (buffer sodium cacodylate, I = 0.05 M) at pH 7.4.



**Figure S18.** Cell viability analysis based on the optical density at 600 nm for **PYPOD**, Zn**PYPOD**, Zn**PYPOD** and Zn<sub>3</sub>**PYPOD** in *S. aureus*. Bottom legend represents the concentration in  $\mu$ M.



**Figure S19.** Cell viability analysis based on the optical density at 600 nm for **PHENPOD**, Zn**PHENPOD**, Zn<sub>2</sub>**PHENPOD** and Zn<sub>3</sub>**PHENPOD** in *S. aureus*. Bottom legend represents the concentration in  $\mu$ M.