

**F and rare V⁴⁺ doped cobalt hydroxide hybrid nanostructures: Excellent OER activity
with ultralow overpotential**

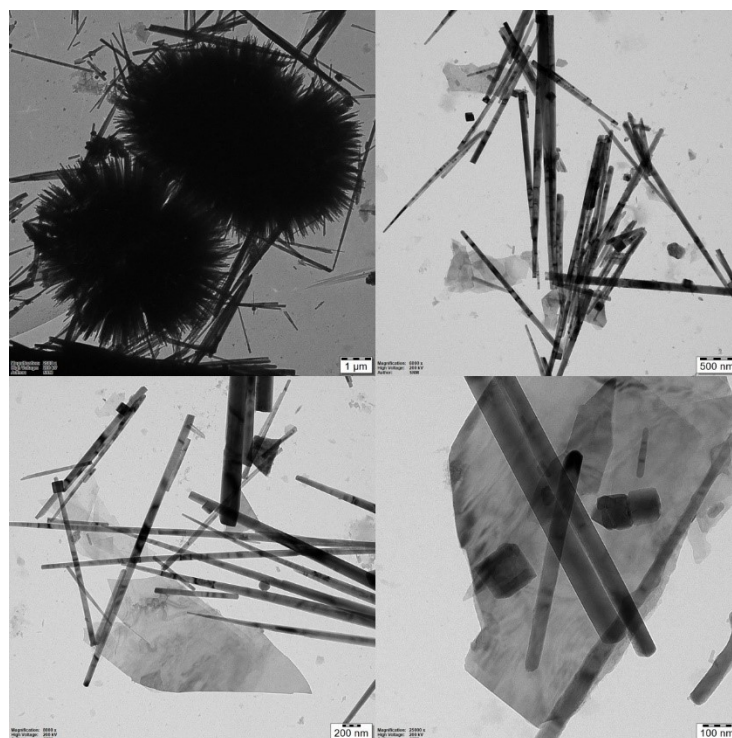


Figure S1. HR-TEM images of Co(OH)₂.

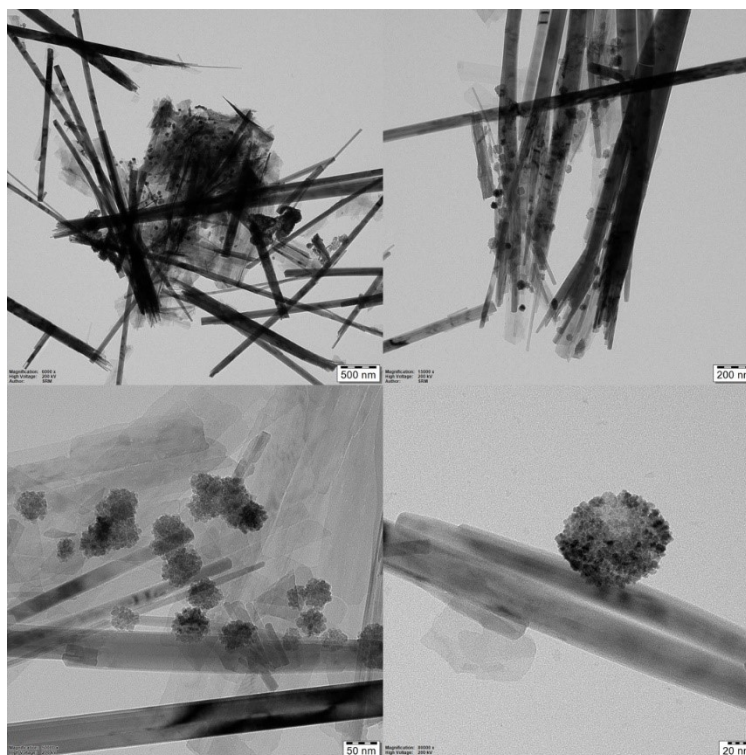


Figure S2. HR-TEM images of V-Co(OH)₂.

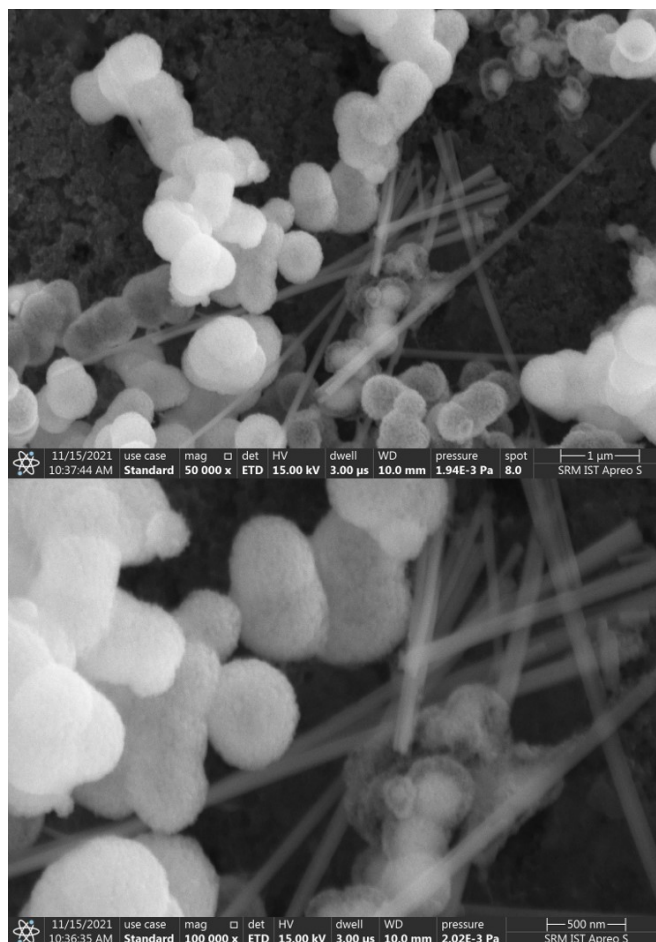


Figure S3. FE-SEM images of V-Co(OH)₂.

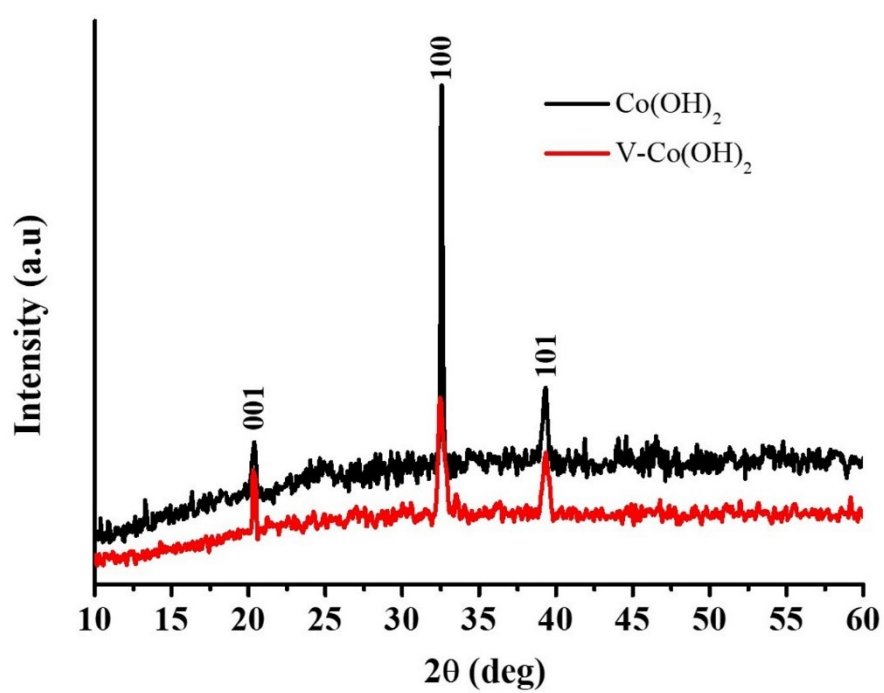


Figure S4. PXRD patterns of Co(OH)₂ and V-Co(OH)₂.

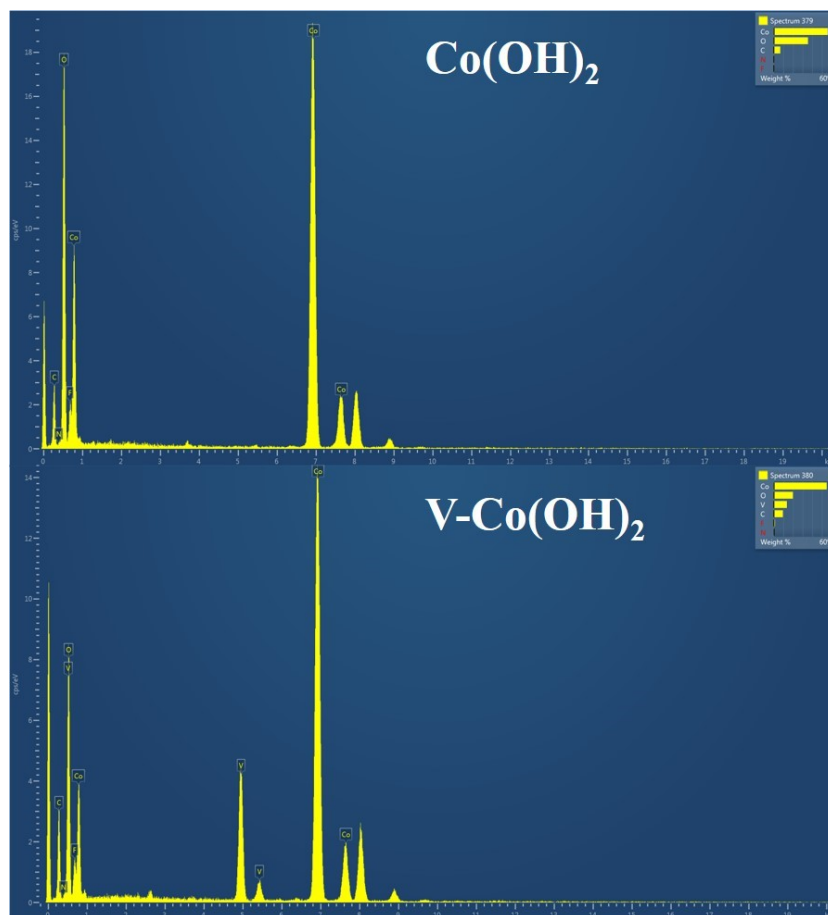


Figure S5. EDX graph of Co(OH)_2 and V-Co(OH)_2 .

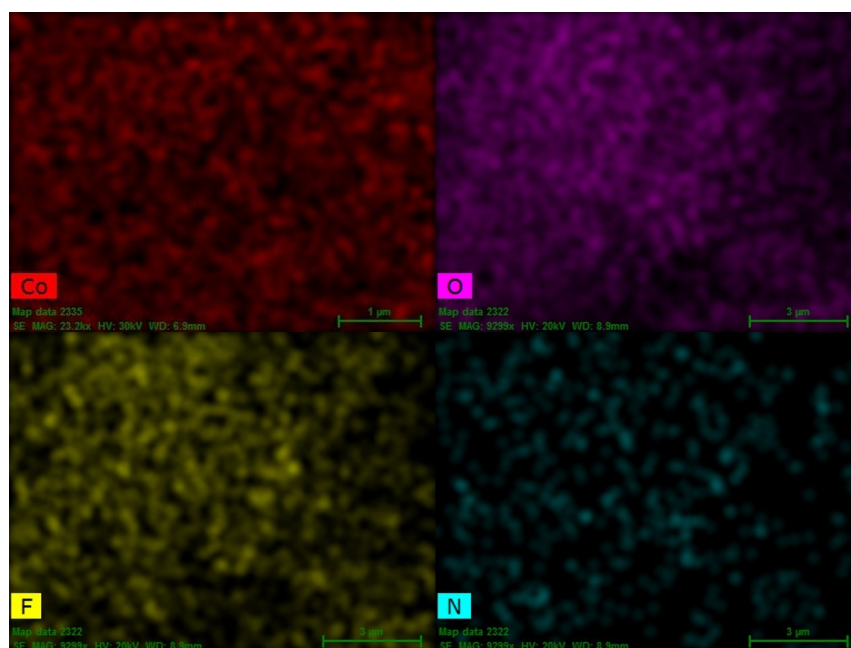


Figure S6. Elemental mapping of Co(OH)_2 .

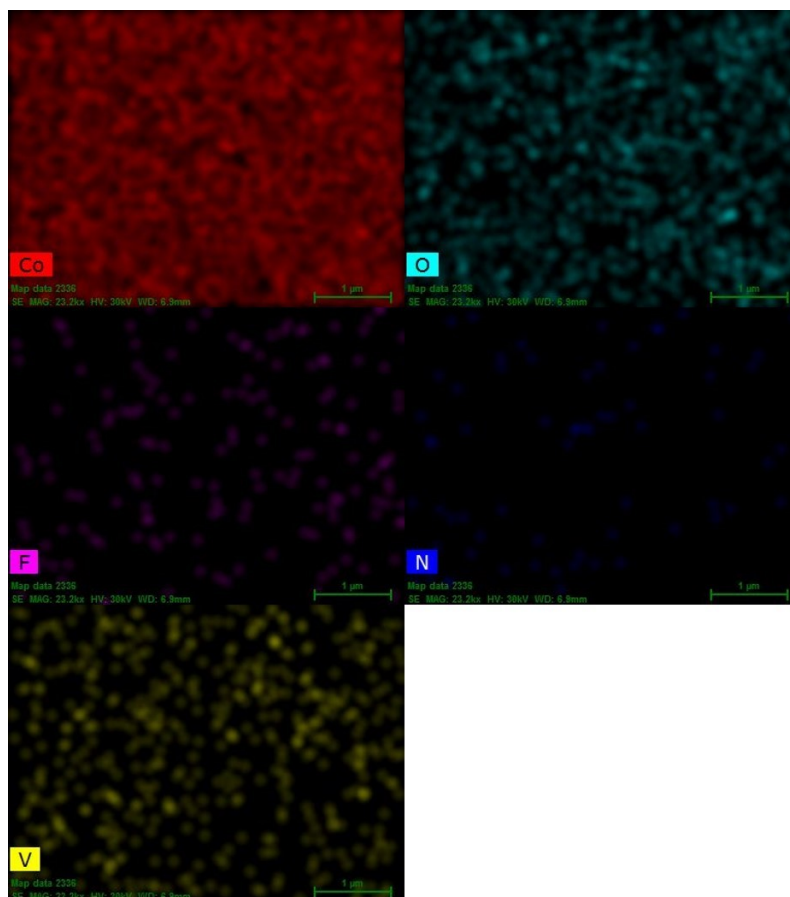


Figure S7. Elemental mapping of V-Co(OH)₂.

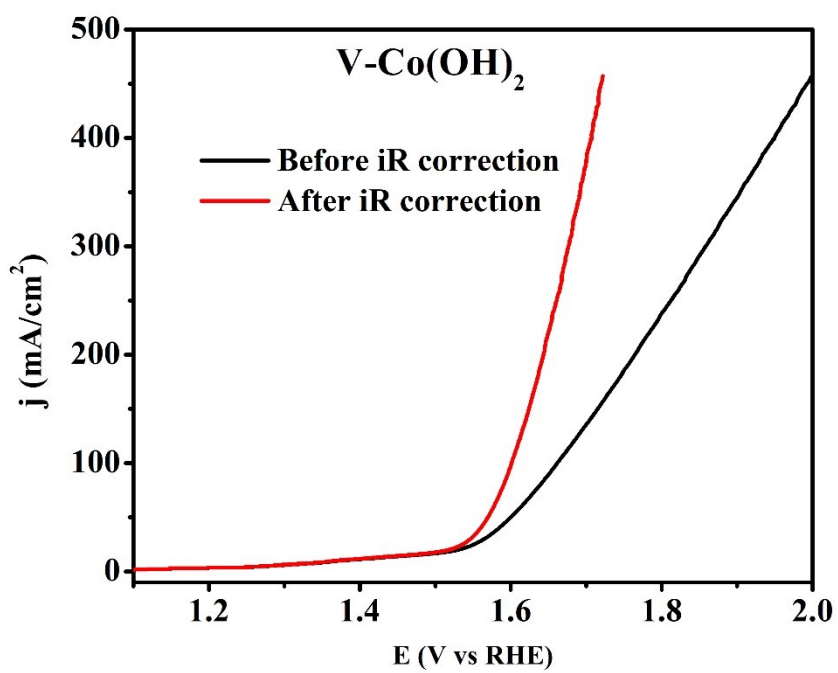


Figure S8. OER polarization curves of V-Co(OH)₂ before and after iR correction.

Table S1- Comparison table to comparing the OER activity of V-Co(OH)₂ with recently reported catalysts.

Catalyst	Overpotential at 10 mA cm ⁻² (mV)	Tafel Slope (mV/dec)	Reference
V-Co(OH)₂	136	47.2	Our work
V-doped CoP	254	35	1
V-Doped Co ₃ O ₄	293.6	53.3	2
CoFeV LDH/NF	242	57	3
V-Ni ₃ S ₂ /CC	180	56	4
CoVOOH	190	39.6	5
CoV-HNNs	268	80	6
CoFeV LDH	242	41.4	7
NiFeV LDHs	195 (20 j)	42	8
V-Co/CoO@C	320	143	9
V-Ni ₃ S ₂	148	70	10
VO _x /Ni ₃ S ₂ @NF	358 (100 j)	82	11
V-doped CoP	340	95.7	12
Ru/CoFe LDH	198	39	13
V ³⁺ -β-Co(OH) ₂	198	60	14
CoVP@CC	290	55.9	15
V-NiFe-LDH/NF	195	31.3	16
V-Ni ₃ S ₂ @NiO/NF	170	98.8	17
Ni ₁ V ₁ P NSs/NF	250 (50 j)	80	18
Ru doped Co(OH) ₂	305 (50 j)	64.8	19
Fe doped MOF CoV@CoO nanoflakes	220 (25j)	59	20
NiCoV-LDH	280	67	21
Co/VN	320	50.4	22

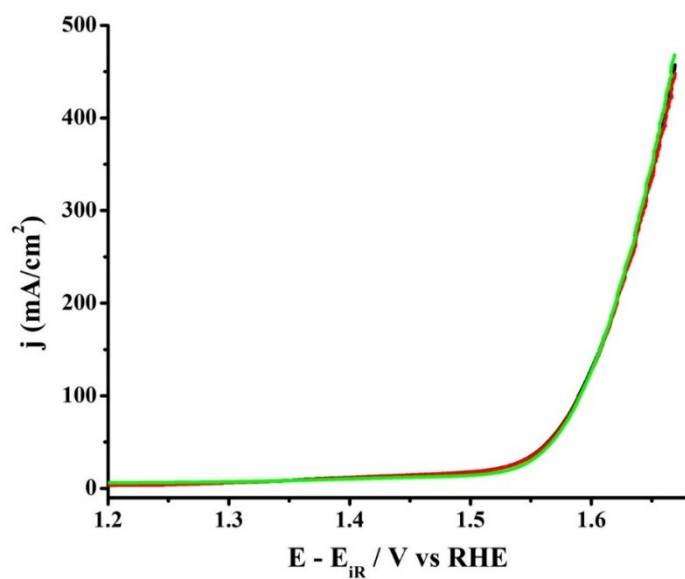


Figure S9. OER polarization curves of V-Co(OH)₂ fabricated at three different batches.

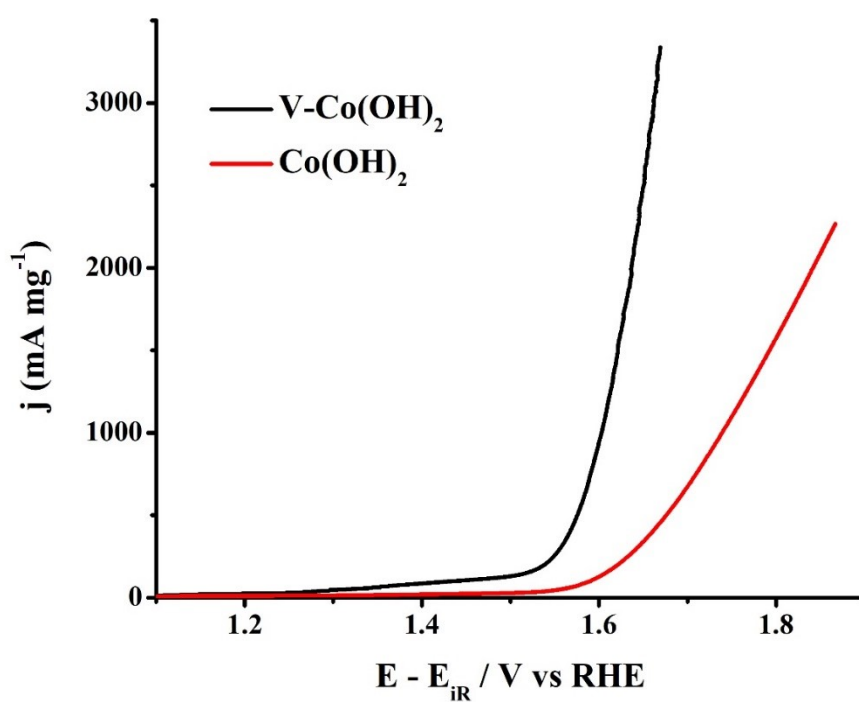


Figure S10. Mass activity of undoped Co(OH)₂ and V-Co(OH)₂ catalysts.

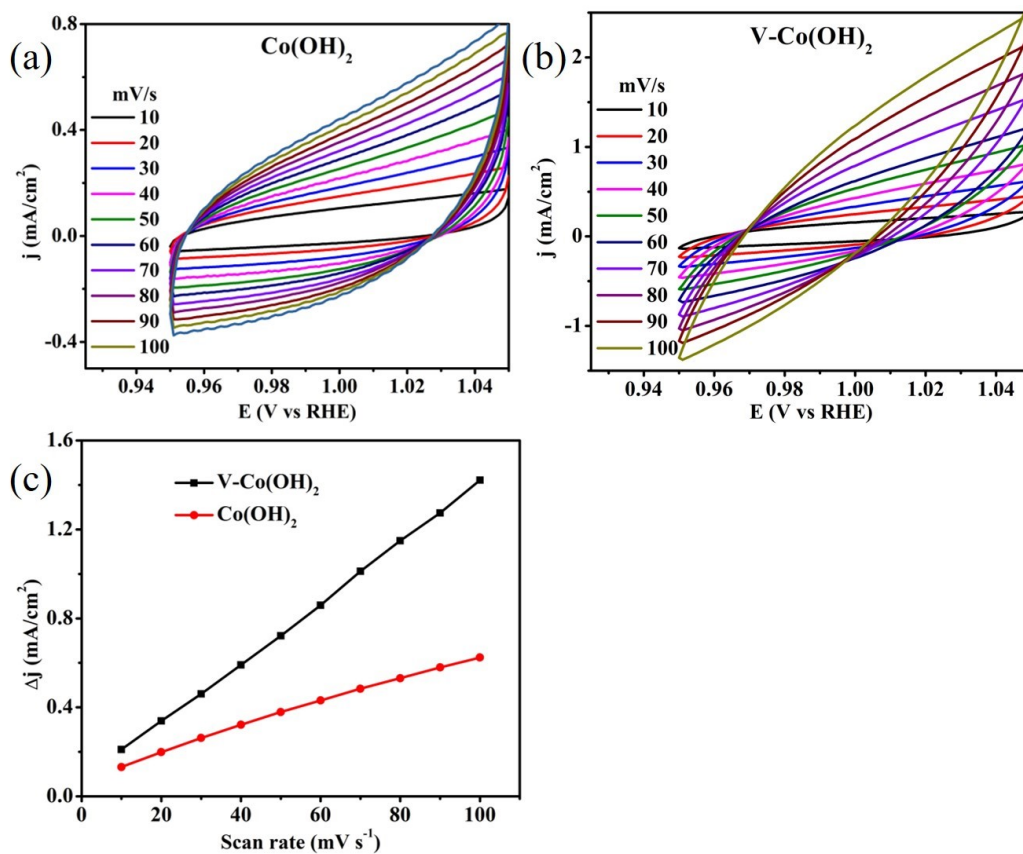


Figure S11. (a, b) Double layer capacitance and (c) capacitive currents as a functional of scan rate.

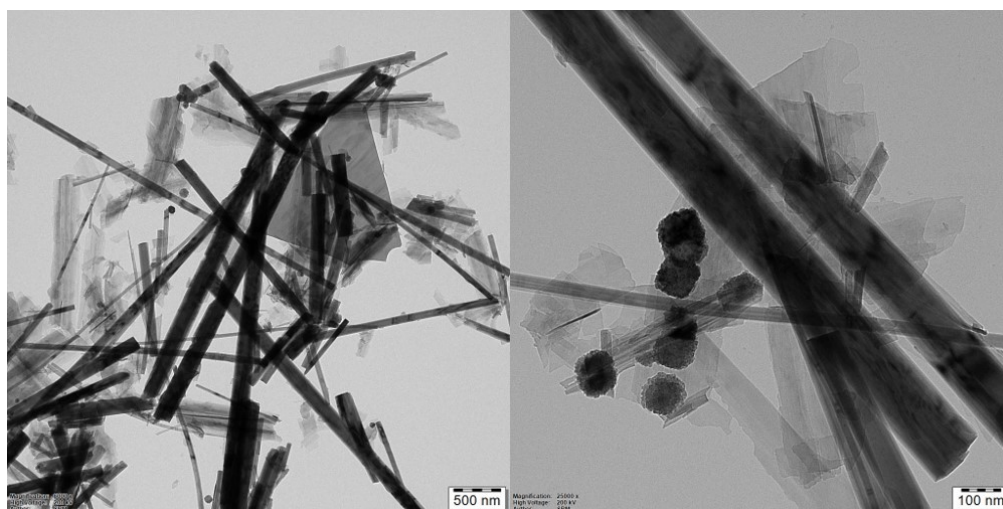


Figure S12. HR-TEM, SAED pattern and lattice fringe pattern images of $\text{V-Co(OH)}_2\text{-8}$.

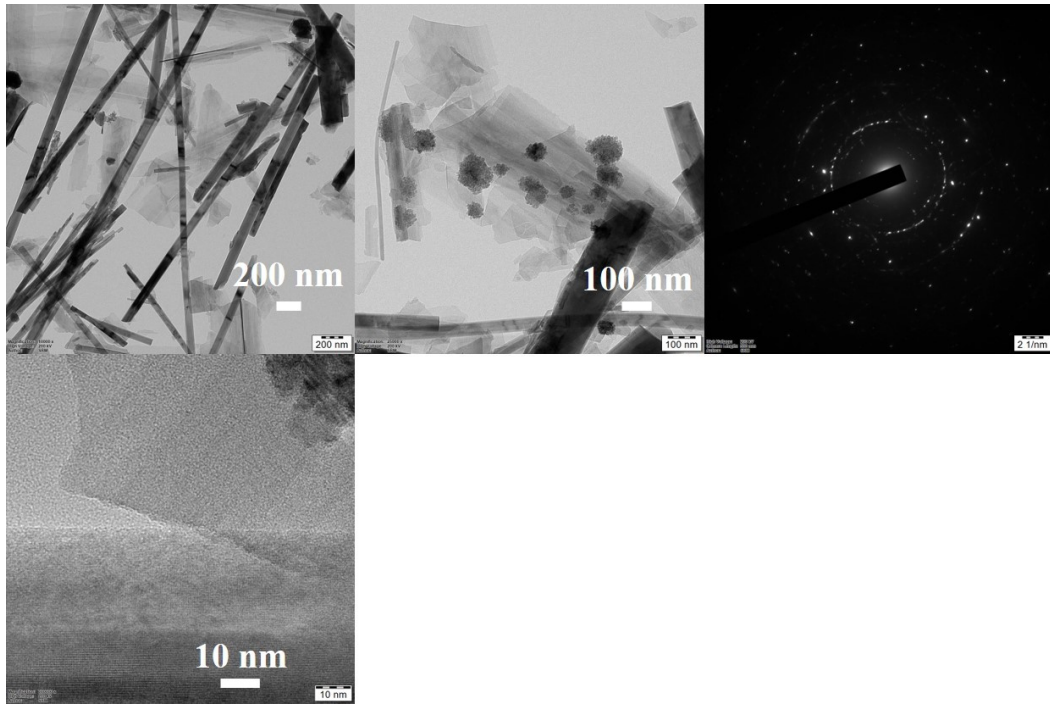


Figure S13. HR-TEM, SAED pattern and lattice fringe pattern images of V-Co(OH)₂-12.

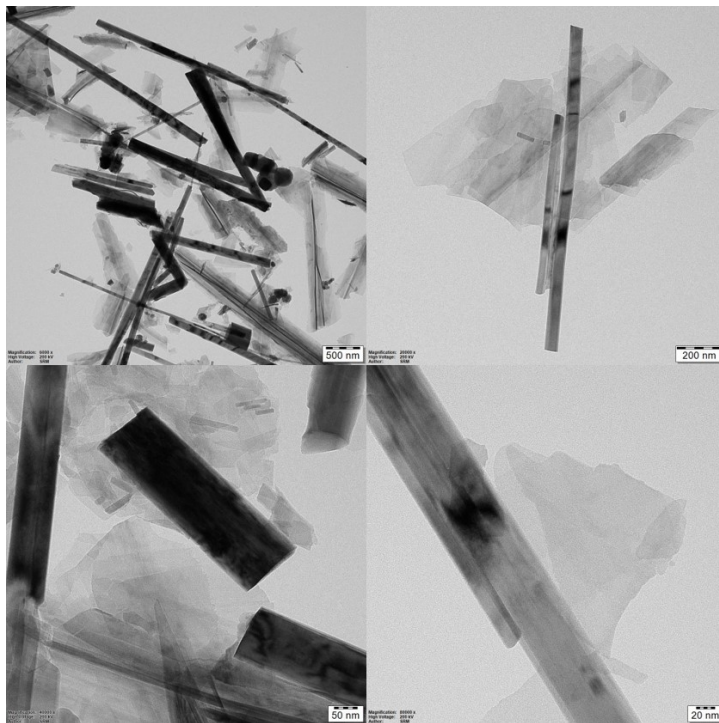


Figure S14. HR-TEM images of V-Co(OH)₂-12 at various magnification.



Figure S15. FE-SEM images of $V-Co(OH)_2-8$.

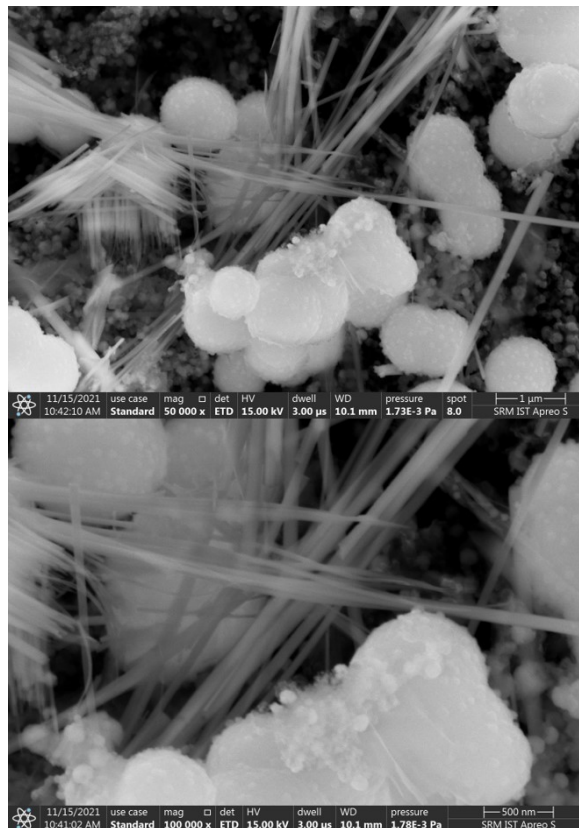


Figure S16. FE-SEM images of $V-Co(OH)_2-12$.

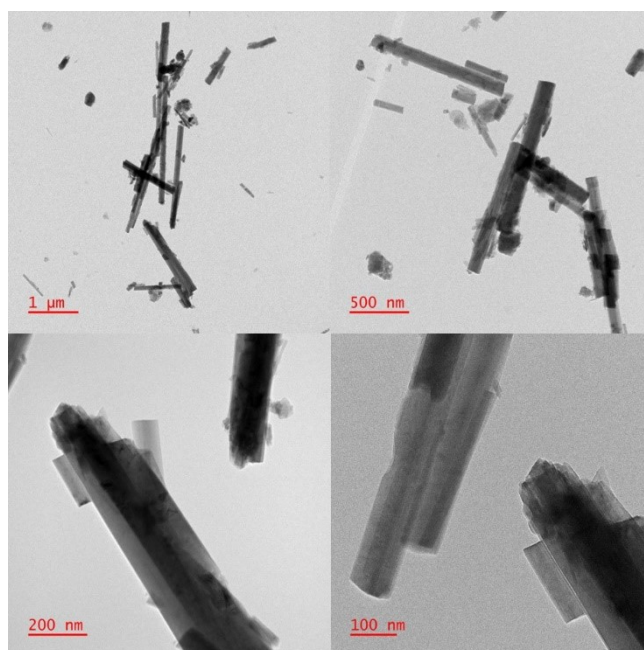


Figure S17. HR-TEM images of V-Co(OH)₂-2h.

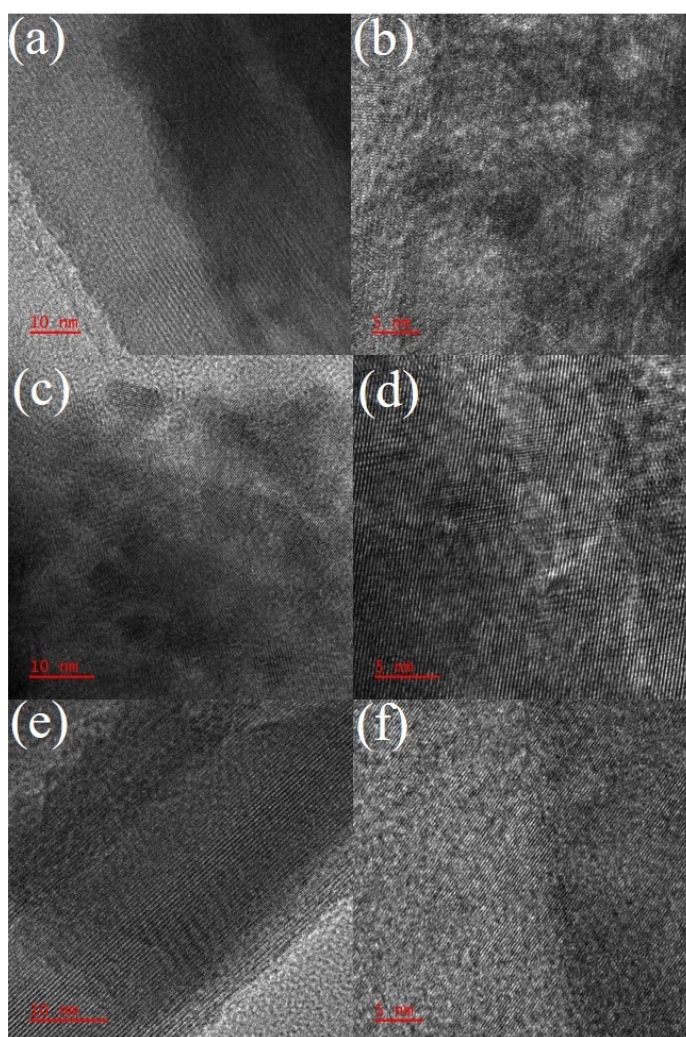


Figure 18. Lattice fringe pattern images of (a, b) V-Co(OH)₂-2h, (c, d) V-Co(OH)₂-4h and (e, f) V-Co(OH)₂ prepared without using NH₄F.

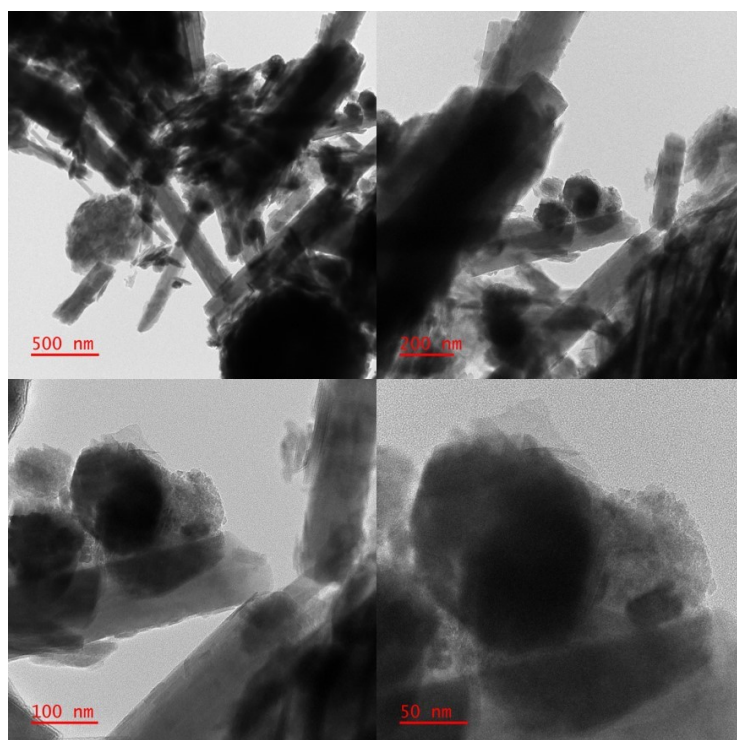


Figure S19. HR-TEM images of V-Co(OH)₂-4h.

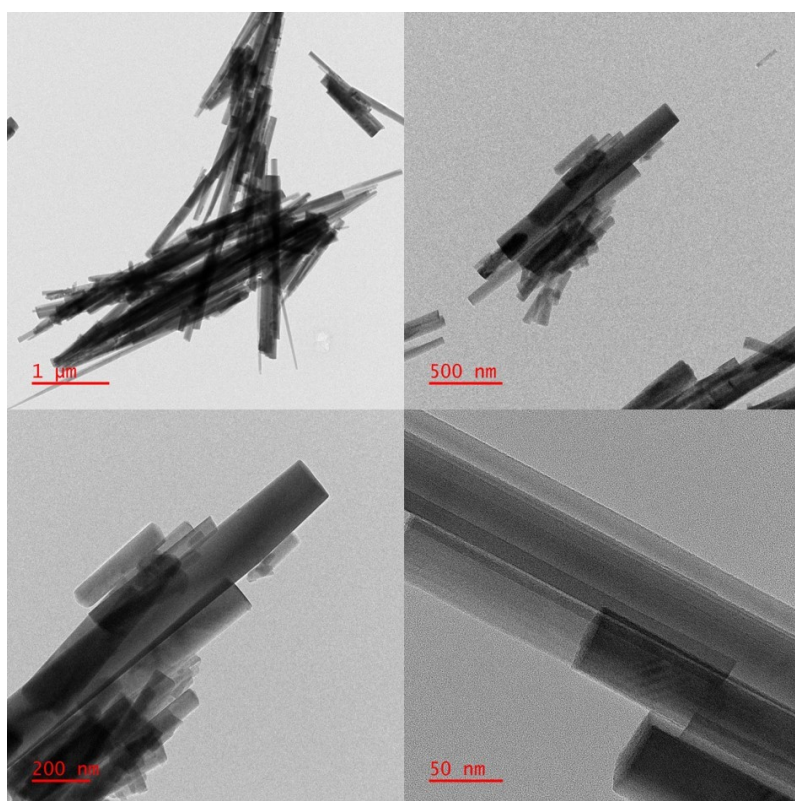


Figure S20. HR-TEM images of V-Co(OH)₂ without NH₄F.

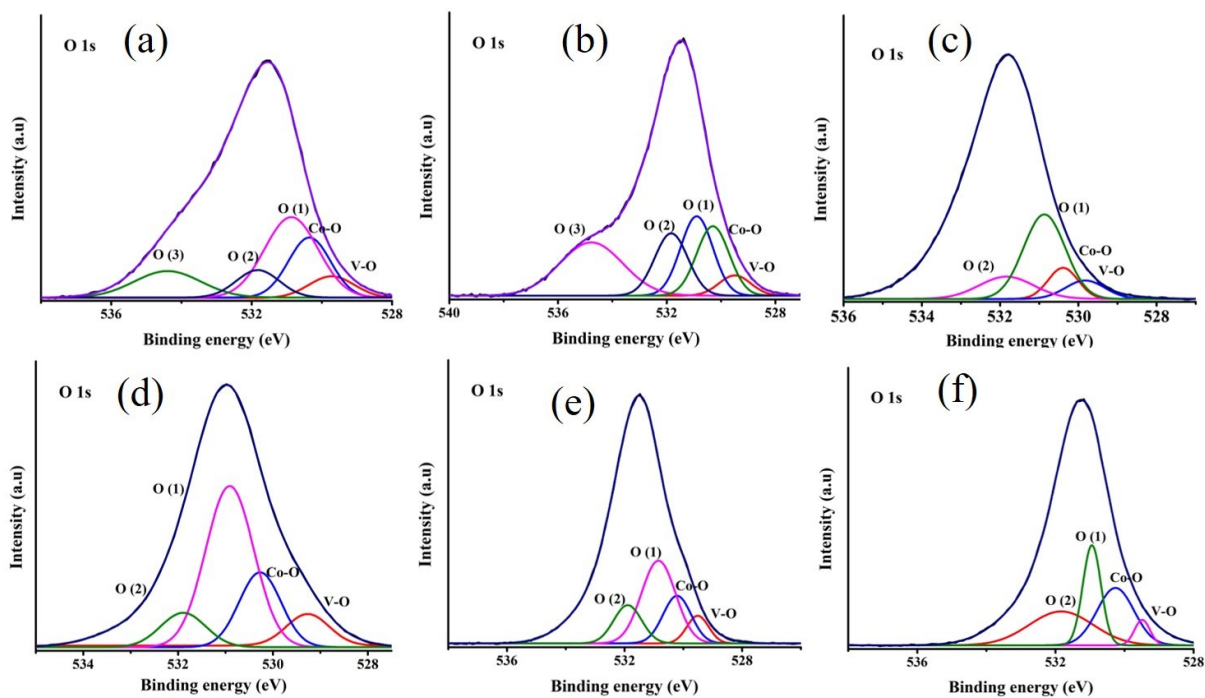


Figure S21. The high-resolution XPS spectra of O 1s in (a) V-Co(OH)₂-8, (b) V-Co(OH)₂-12, (c) V-Co(OH)₂-2h, (d) V-Co(OH)₂-4h, (e) V-Co(OH)₂ without NH₄F and (f) V-Co(OH)₂ with NaF.

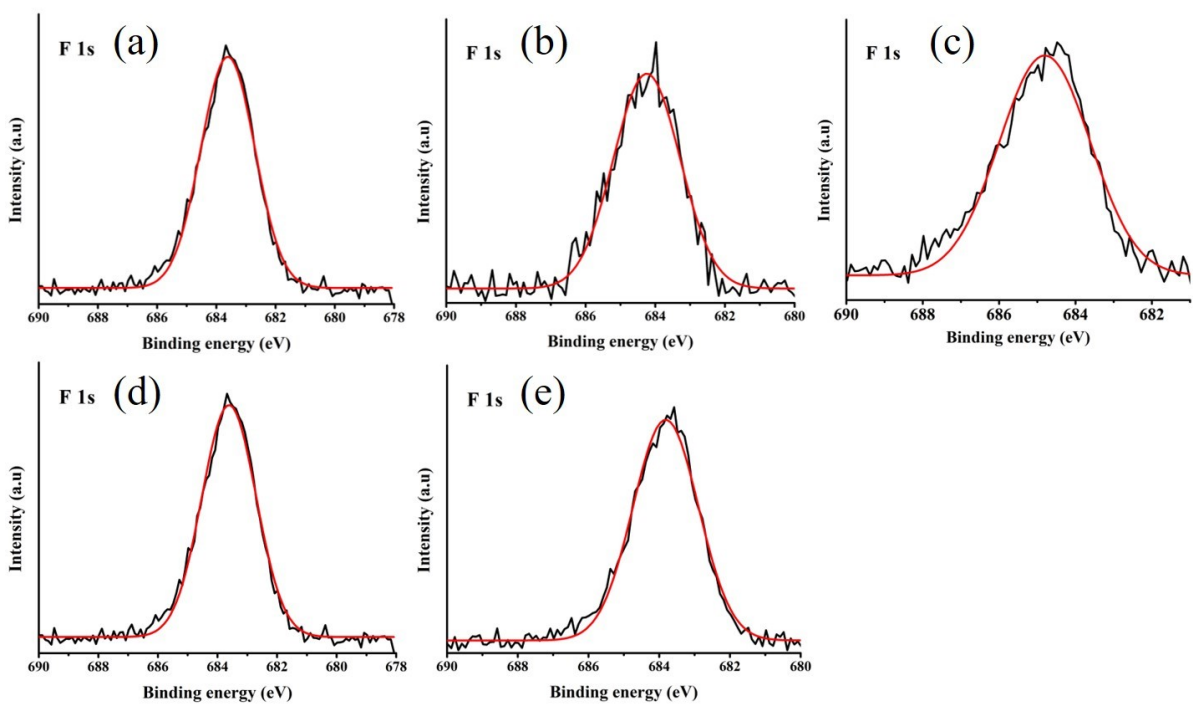


Figure S22. The high-resolution XPS spectra of F 1s in (a) V-Co(OH)₂-8, (b) V-Co(OH)₂-12, (c) V-Co(OH)₂-2h, (d) V-Co(OH)₂-4h, (e) with NaF.

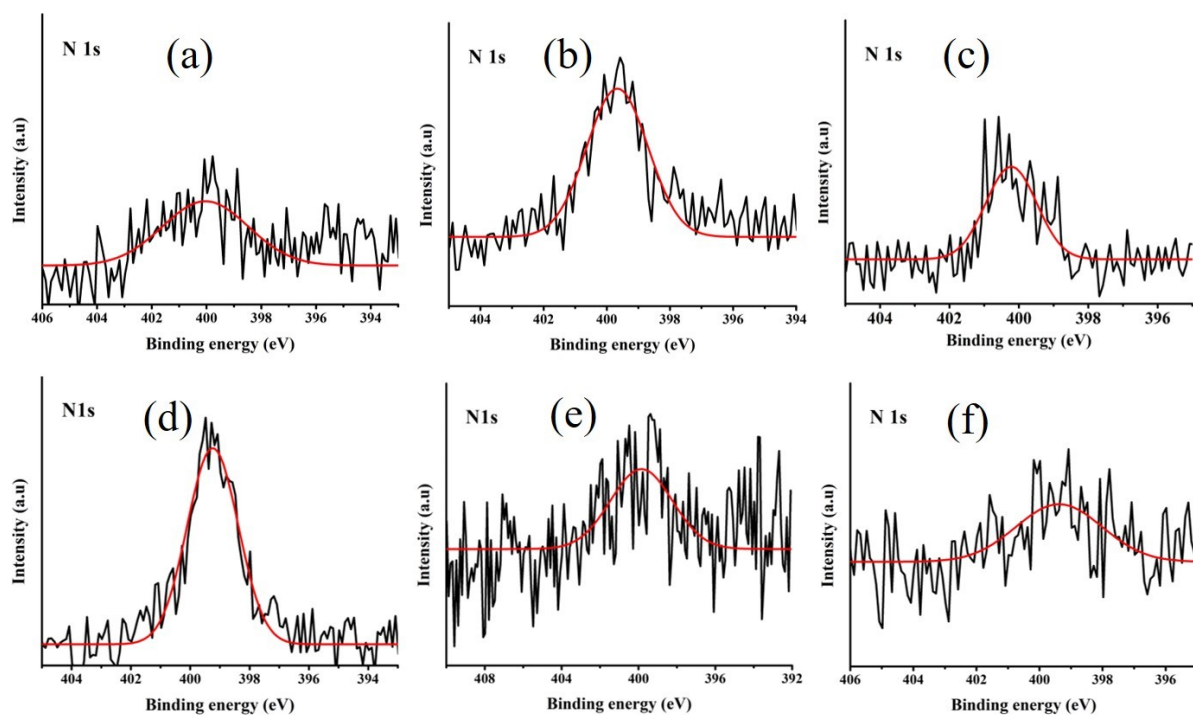


Figure S23. The high-resolution XPS spectra of N 1s in (a) V-Co(OH)₂-8, (b) V-Co(OH)₂-12, (c) V-Co(OH)₂-2h, (d) V-Co(OH)₂-4h, (e) V-Co(OH)₂ without NH₄F and (f) V-Co(OH)₂ with NaF.

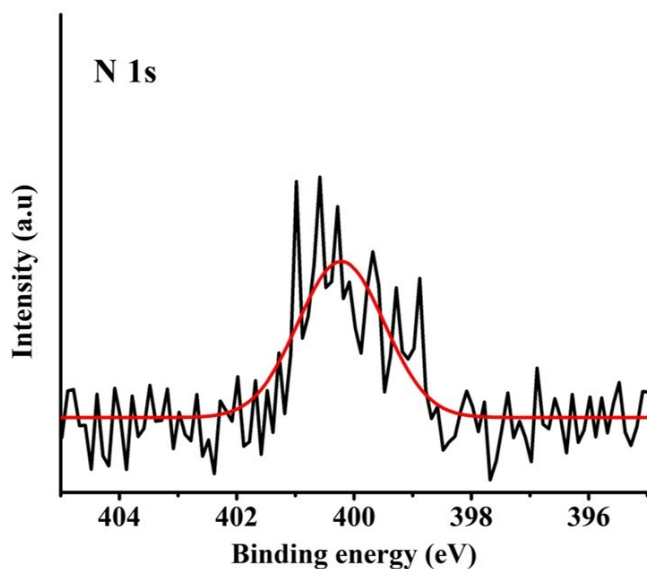


Figure S24. The high-resolution XPS spectra of N 1s in V-Co(OH)₂ after OER studies.

Table S2. Binding energy comparison of Co(OH)₂ and V, F doped catalysts.

Catalyst	Binding energy (eV)		Intensity ratio (Co ²⁺ /Co ³⁺)	Binding energy (eV)		Intensity ratio (V ⁴⁺ /V ⁵⁺)
	Co ²⁺	Co ³⁺		V ⁴⁺	V ⁵⁺	
Co(OH) ₂	797.68	795.72	7.5>	-	-	-
	781.83	780.68	9.5>			
V-Co(OH) ₂ -8	798.15	795.72	3.2>	516.75	517.85	1.8>
	782.13	780.2	1.4>			
V-Co(OH) ₂ -10	798.01	795.71	2.0>	516.57	517.75	3.4>
	781.97	780.2	2.1>			
V-Co(OH) ₂ -12	798.3	796.5	2.9<	516.38	517.29	1.5>
	782.6	780.7	1.2<			
V-Co(OH) ₂ -2h	798.06	795.66	8>	516.12	517.87	5.0>
	782.32	780.15	4.9>			
V-Co(OH) ₂ -4h	798.04	796.3	2.4<	516.17	517.16	3.1>
	781.6	780.2	1.3<			
V-Co(OH) ₂ (without NH ₄ F)	798.02	796.4	2<	516.62	517.94	3.0>
	781.71	779.86	1.3<			
V-Co(OH) ₂ -NaF	798.2	796.5	5.2<	516.17	517.15	2.0>
	781.72	780.2	1.2<			
V-Co(OH) ₂ -after OER	797.72	796.14	1.8<	516.5	517.36	1.3>
	781.67	780.37	1.3<			

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