Supporting Information

for

Reaction of a Nitrosyl Complex of Mn(II)-porphyrinate with Superoxide:

NOD Activity is More Favorable Over SOD Activity

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Figure S1. FT-IR spectrum of ligand TMPPH₂ in KBr.



Figure S2. ESI-mass spectrum of ligand TMPPH₂ in acetonitrile.



Figure S3. ¹H NMR spectrum of ligand TMPPH₂ in CDCl₃.



Figure S4. ¹³C NMR spectrum of ligand TMPPH₂ in CDCl₃.



Figure S5. FT-IR spectrum of complex 1a in KBr.



Figure S6. ESI-mass spectrum of complex 1a in acetonitrile. [Inset: (a) experimental and (b) simulated isotopic distribution pattern].



Figure S7. UV-visible spectrum of complex 1a in dichloromethane at RT.



Figure S8. X-band EPR spectrum of complex 1a in dichloromethane at 77K.



Figure S9. ORTEP diagram of complex 1a (30% thermal ellipsoid plot, H-atoms are omitted for clarity).



Figure S10. FT-IR spectrum of complex 1 in KBr.



Figure S11. ESI-mass spectrum of complex 1 in acetonitrile. [Inset: (a) experimental and (b) simulated isotopic distribution pattern].



Figure S12. UV-visible spectrum of complex 1 in dichloromethane at RT.



Figure S13. ¹H NMR spectrum of complex 1 in DMSO-d₆.



Figure S14. X-band EPR spectrum of complex 1 in THF at 77K.

Figure S15. Cyclic voltammograms of complex 1 in dichloromethane *vs.* Ag/AgCl, supporting electrolyte 0.1 M TBAP, Scan rate 0.1 v/s.

Figure S16: UV-visible spectrum of the reaction mixture of complex 1a and *t*BuOOH at -40 °C in THF.

Figure S17. FT-IR spectrum of complex 2 in KBr.

Figure S18. ESI-mass spectrum of complex **2** in acetonitrile. [Inset: (a) experimental and (b) simulated isotopic distribution pattern].

Figure S19. UV-visible spectrum of complex 2 in THF.

Figure S20. X-band EPR spectrum of complex 2 in THF at 77K.

Figure S21. ESI-mass spectrum of 2,4-di-tert-butyl-6-nitrophenol in acetonitrile.

Figure S22. ¹H NMR spectrum of 2,4-di-tert-butyl-6-nitrophenol in CDCl₃.

Figure S23. ¹³C NMR spectrum of 2,4-di-*tert*-butyl-6-nitrophenol in CDCl₃.

Figure S24. ESI-mass spectrum of 9-fluorenone in acetonitrile.

Figure S25. ¹H NMR spectrum of 9-fluorenone in CDCl₃.

Figure S26. ¹³C NMR spectrum of 9-fluorenone in CDCl₃.

Figure S27. ESI-mass spectrum of tmpNO₃ in acetonitrile.

Figure S28. ¹H NMR spectrum of tmpNO₃ in CDCl₃.

Figure S29. ¹³C NMR spectrum of tmpNO₃ in CDCl₃.

Figure S30. ORTEP diagram of tmpNO₃ (30% thermal ellipsoid plot, H-atoms are omitted for clarity).

 Table S1. Crystallographic data for complex 2.

	2
Formulae	C ₄₈ H ₃₈ N ₄ O ₆ Mn
Mol. wt.	821.76
Crystal system	Monoclinic
Space group	P 21/c
Temperature /K	100(2)
Wavelength /Å	0.71073
a /Å	9.8(2)
b /Å	10.4(2)
c /Å	21.2(4)
$\alpha/_{\circ}$	90
β/°	111.6(7)
$\gamma/^{\circ}$	90
V/ Å ³	2009(68)
Z	2
Density/Mgm ⁻³	1.358
Abs. Coeff. /mm ⁻¹	0.385
Abs. correction	none
F(000)	854
Total no. of reflections	3524
Reflections, $I > 2\sigma(I)$	1310
Max. 20/°	24.995
Ranges (h, k, l)	$-11 \le h \le 11$
	$-12 \le k \le 12$
	-11 ≤1 ≤ 25
Complete to 2θ (%)	0.998
Refinement method	Full-matrix least-
	squares on F^2
Goof (F^2)	0.841
R indices $[I > 2\sigma(I)]$	0.0904
R indices (all data)	0.1702

Table S2. Selected bond lengths (Å) of complex 2.

Atoms	2
Mn1-N1	1.91(3)
Mn1-N2	1.95(3)
Mn1-O1	2.13(3)
C1-C2	1.28(2)
C2-C3	1.37(3)
C4-C5	1.49(2)
C5-C6	1.32(2)
C6-C7	1.37(2)

Table S3. Selected bond angles (°) of complex 2.

Atoms	2
N1-Mn1-N2	86.8(12)
N1-Mn1-O1	90.5(14)
C1-C2-C3	105.2(11)
C5-C6-C7	119.3(14)
C8-O2-C9	119.8(7)