

## Two Novel Metal-Organic Frameworks functionalised with pentamethylcyclopentadienyl iridium(III) chloride for catalytic conversion of carbon dioxide to formate

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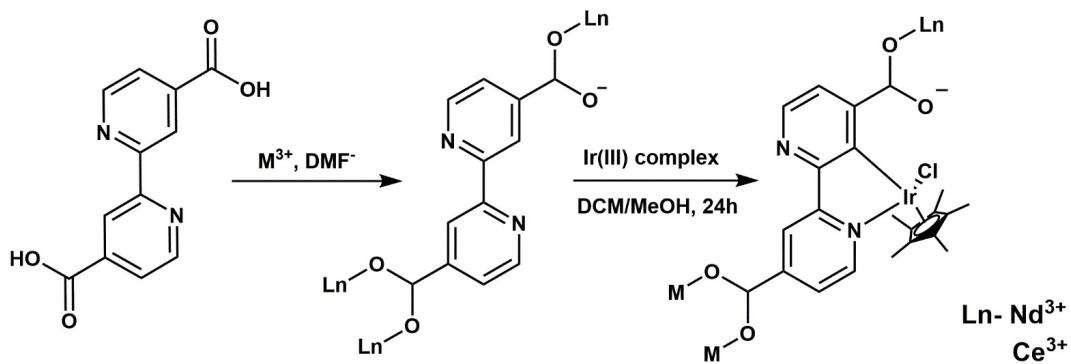
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**Table S1:** Iridium supported on different materials for CO<sub>2</sub> hydrogenation.

	<b>Product</b>	<b>Activity (TON)</b>	<b>Reference</b>
Ir supported on mesoporous silica	formate	-	<sup>1</sup>
Ir supported on zeolite imidazolate framework	formate	30000	<sup>2</sup>
Ir on UiO-66 MOF	formate	6149	<sup>3</sup>
1,5-cyclooctadiene-iridium(I) chloride supported on polymerized cyclic (alkyl)(amino)carbenes	formate	195,600	<sup>4</sup>
Ir supported on Covalent Triazine Framework	formate	5000	<sup>5</sup>
Iridium supported on spinal cubic cobalt oxide	formate	3809	<sup>6</sup>
Ir supported on Covalent Triazine Framework	formate	-	<sup>7</sup>
Ir supported on Covalent organic Framework	formate	6400	<sup>8</sup>
Ir supported on mesoporous silica	formate	14200	<sup>9</sup>

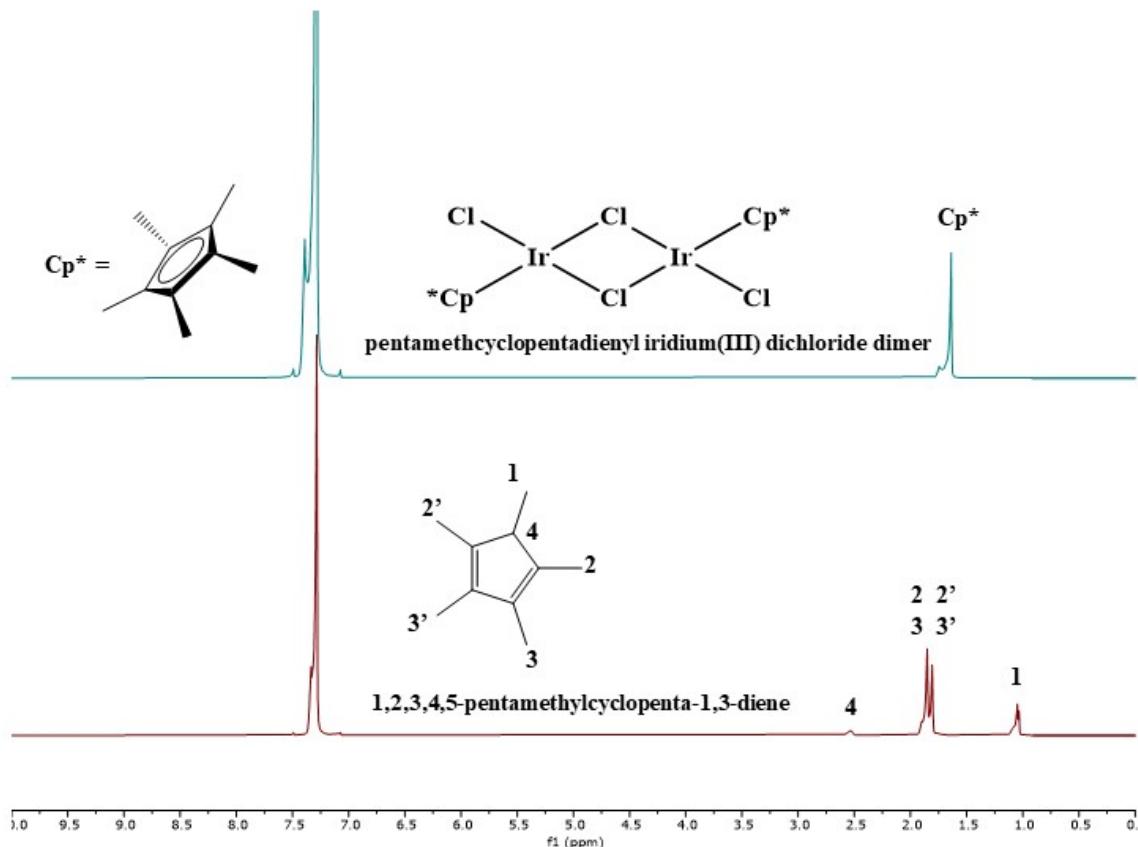


**Scheme S1:** Synthesis and functionalization scheme of JMS-6a and JMS-7a. Due to steric hinderance some sites will remain unfunctionalized. This is in line with the observed catalyst load as determined by ICP-OES

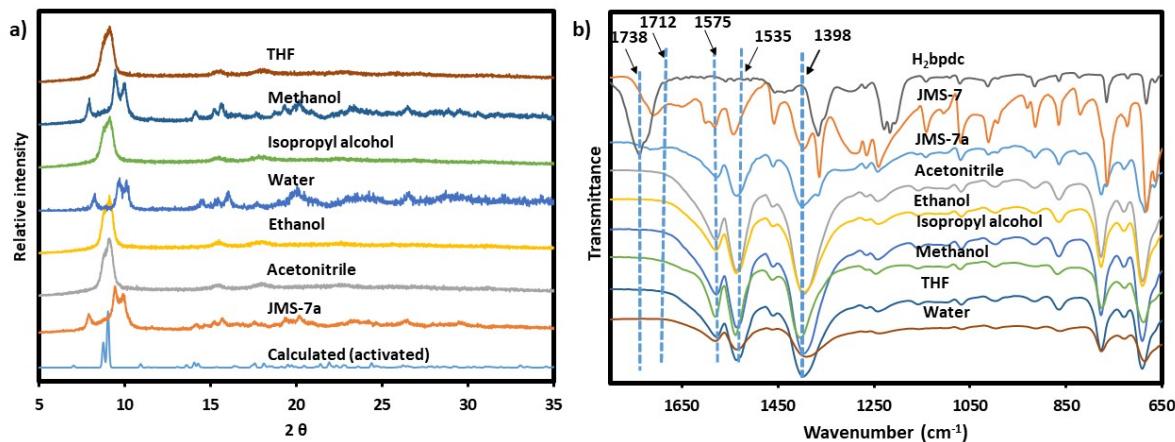
**Table S2:** JMS-6 and JMS-7 crystallographic information.

	JMS-6	JMS-7
<b>Empirical formula</b>	$\text{C}_{24}\text{H}_{26}\text{N}_5\text{Nd}_1\text{O}_{10}$	$\text{C}_{24}\text{H}_{26}\text{N}_5\text{Ce}_1\text{O}_{10}$
<b>Formula weight (gmol<math>^{-1}</math>)</b>	684.41	687.64
<b>Temperature/K</b>	100(2)	100(2)
<b>Crystal system</b>	Triclinic	Triclinic
<b>Space group</b>	$P-1$	$P-1$
<b>a/Å</b>	10.0085(4)	10.0554(4)
<b>b/Å</b>	11.2398(4)	11.1554(4)
<b>c/Å</b>	13.5602(6)	13.6202(5)
<b><math>\alpha/^\circ</math></b>	68.4370(12)	68.7124(12)
<b><math>\beta/^\circ</math></b>	86.0626(14)	86.0229(14)
<b><math>\gamma/^\circ</math></b>	74.5010(13)	74.3419(12)
<b>Volume/Å<math>^3</math></b>	1366.33(15)	1370.04 (14)
<b>Z</b>	2	2
<b>Calculated density (g/cm<math>^3</math>)</b>	1.6763	1.6667
<b><math>\mu(\text{Mo-K}\alpha)/\text{mm}^{-1}</math></b>	1.627	1.725
<b>F(000)</b>	686	692
<b>Crystal size/mm<math>^3</math></b>	0.050 x 0.120 x 0.140	0.020 x 0.040x 0.090
<b>Radiation</b>	0.71073Å	0.71073Å
<b>2θ range for data collection/°</b>	55.11°	56.57°
<b>Index ranges</b>	13,15,18	13,15,18
<b>Reflections collected</b>	6808	6850

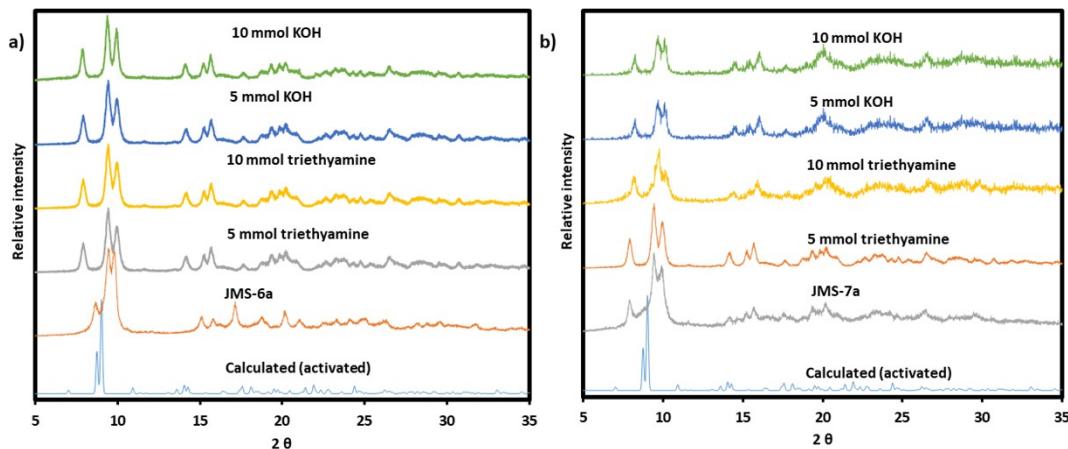
No. unique data	6452	6167
Goodness of fit on S	1.14	1.056
Final R indexes [ $I >= 2\sigma(I)$ ]	0.0265	0.0279
Final wR <sub>2</sub> indexes [all data]	0.0685	0.0581
Deepest hole/Highest peak	-0.99, 1.14	-0.96, 1



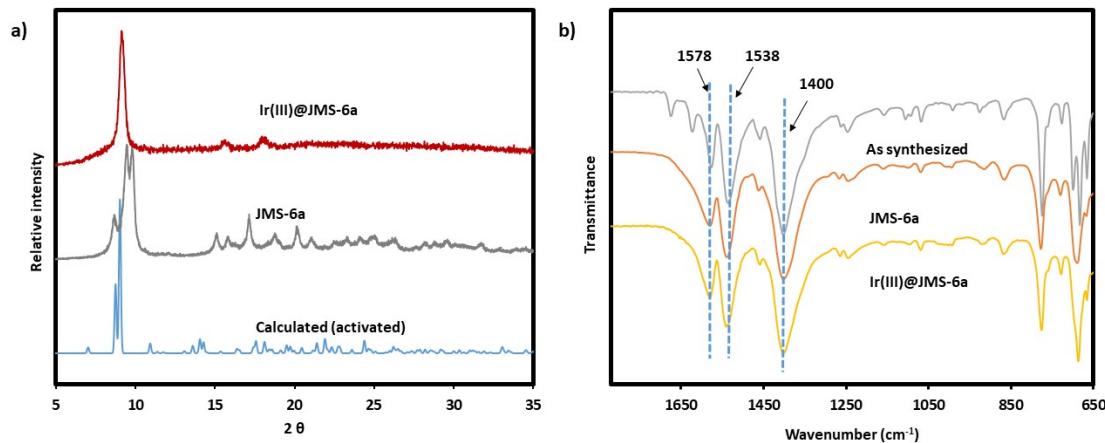
**Fig S1:** <sup>1</sup>H NMR (Bottom) Stacked NMR of precursor (1,2,3,4,5-pentamethylcyclopenta-1,3-diene) and (top) pentamethylcyclopentadienyl iridium(III) dichloride dimer.



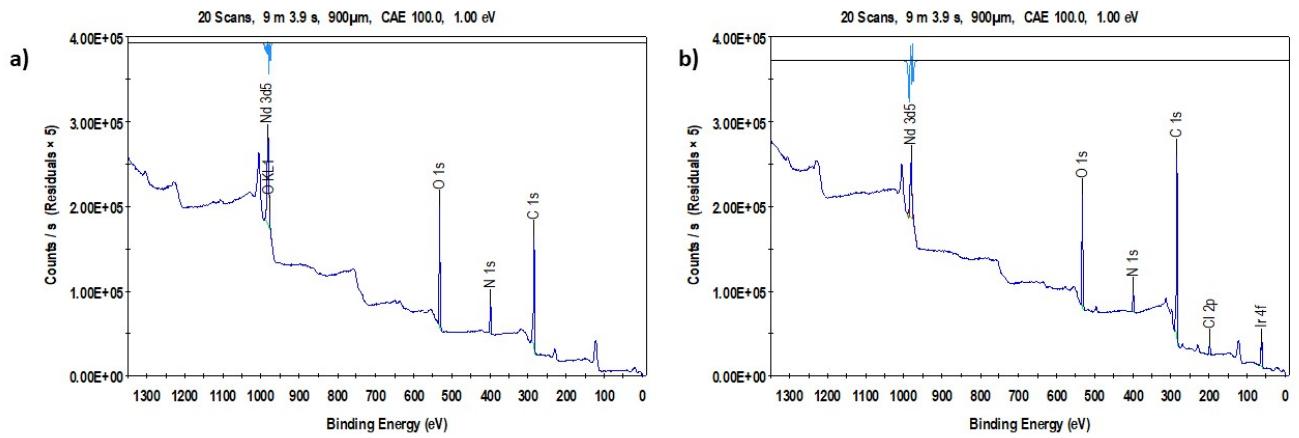
**Fig S2:** Solvent stability of JMS-7a, (a) PXRD (b) FTIR.



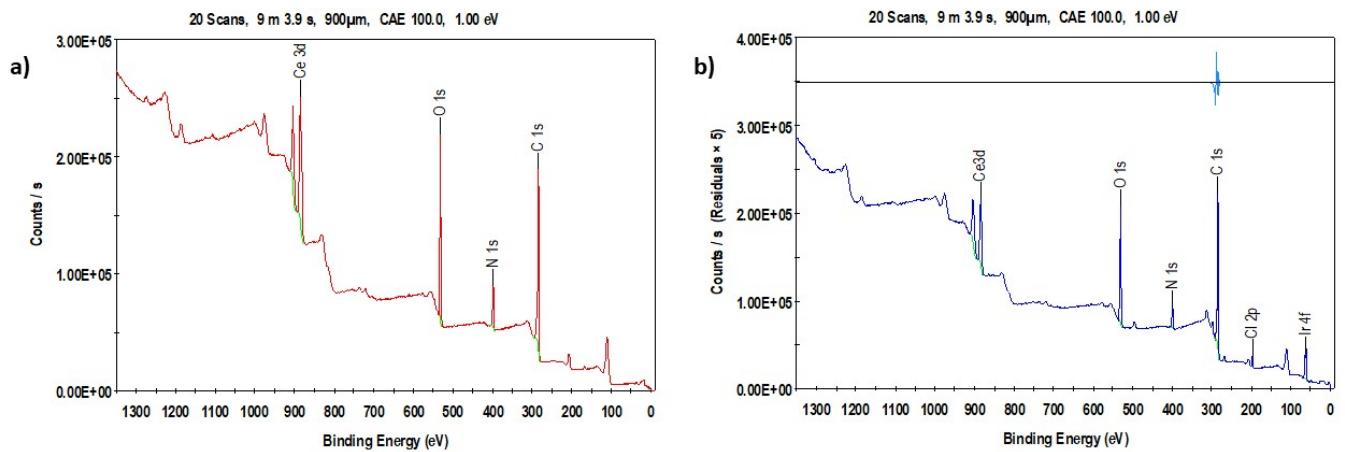
**Fig S3:** PXRD of base stability of JMS-6a and JMS-7a.



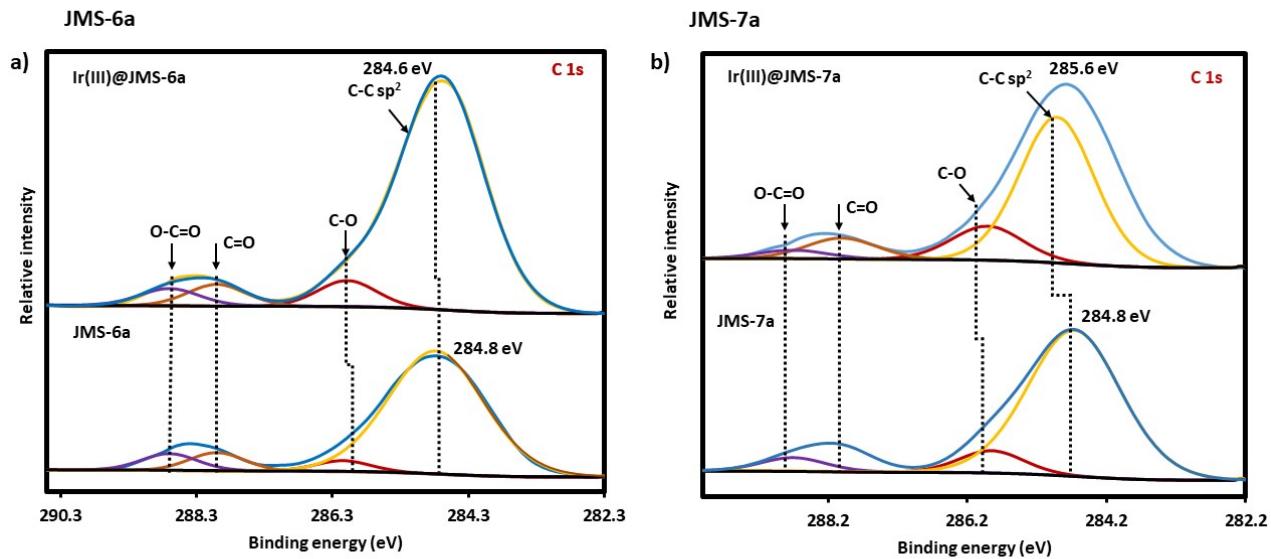
**Fig S4:** Functionalization of JMS-6a, (a) PXRD and (b) FTIR.



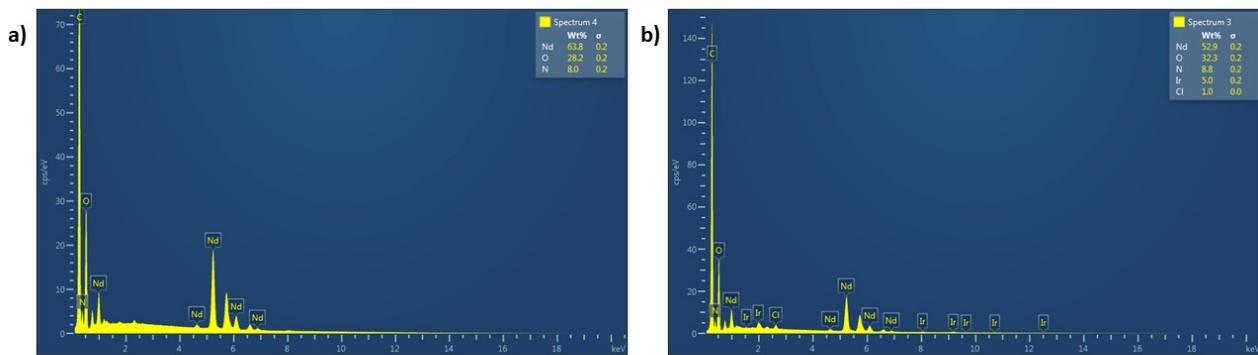
**Fig S5:** XPS spectra of JMS-6a and Ir(III)@JMS-6a.



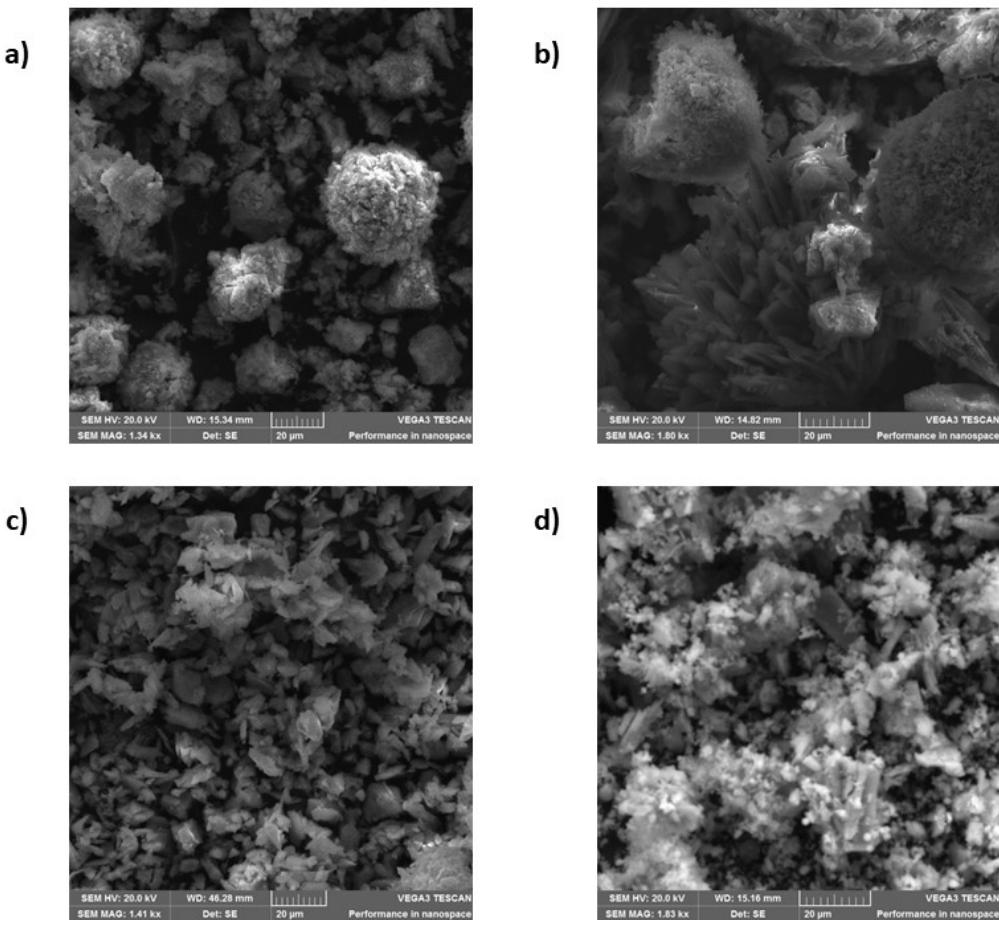
**Fig S6:** XPS spectra of JMS-7a and Ir(III)@JMS-7a.



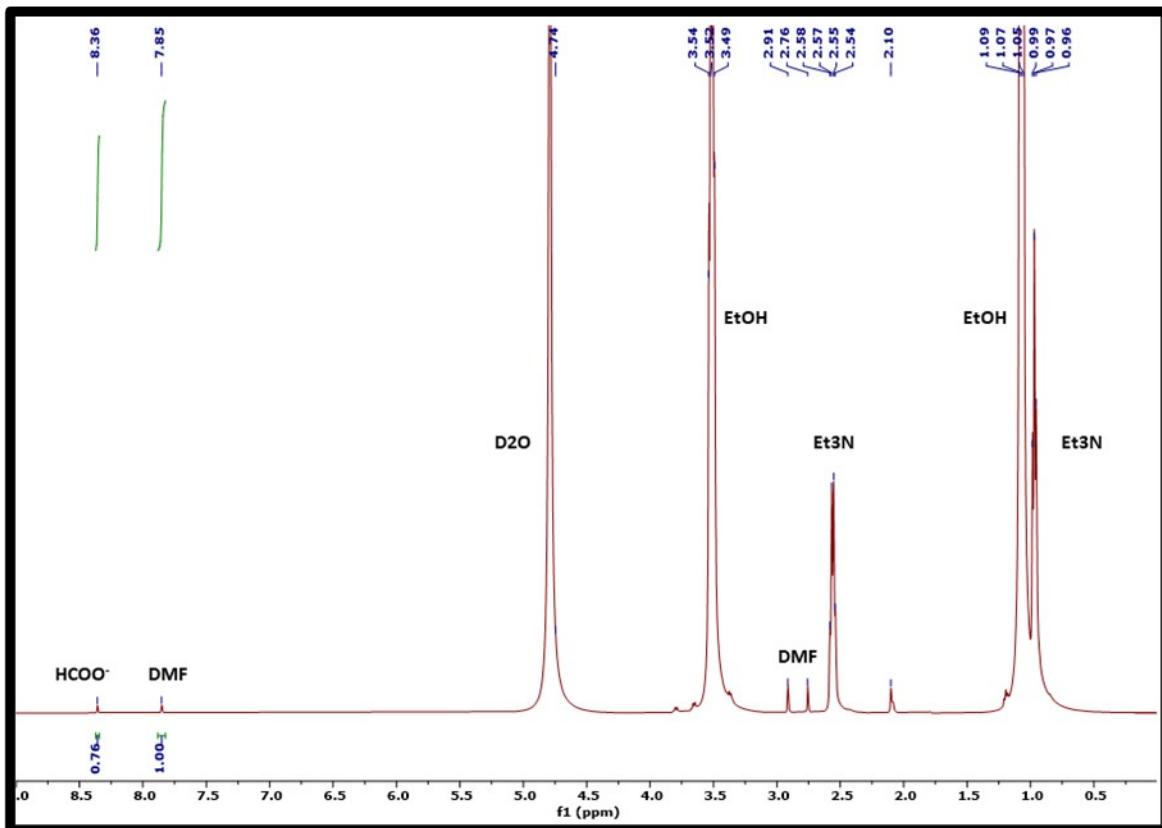
**Fig S7:** C 1s XPS analysis of Ir(III)@JMS-6a and Ir(III)@JMS-7a.



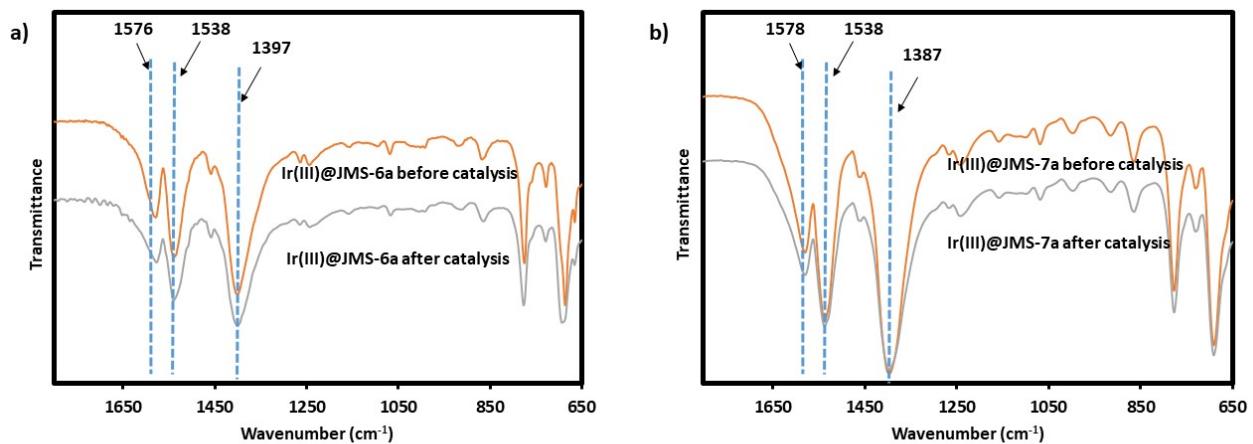
**Fig S8:** EDX spectra of JMS-6a and Ir(III)@JMS-6a.



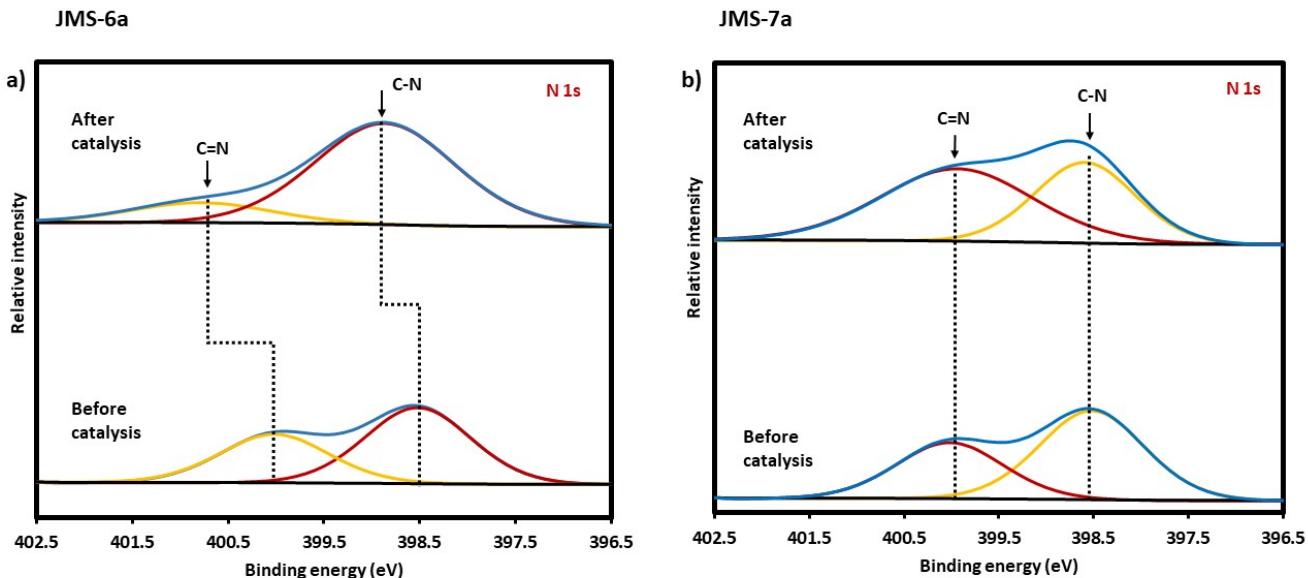
**Fig S9:** SEM images of, **(a)** JMS-6a, **(b)** JMS-7a, **(c)** Ir(III)@JMS-6a, **(d)** Ir(III)@JMS-7a.



**Fig S10:**  $^1\text{H}$  NMR post catalysis (Ir(III)@JMS-7a, used to detect and quantify formate.



**Fig S11:** FTIR analysis of Ir(III)@JMS-6a and Ir(III)@JMS-7a before and after catalysis.



**Fig S12:** N 1s XPS analysis of Ir(III)@JMS-6a and Ir(III)@JMS-7a before and after catalysis.

## References

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