

Tunable bandgap in self-assembled transition metal-incorporated heterometallic M_2Sb_4 ($M = V, Mn, Co, Ni, \text{ and } Cu$) oxo clusters

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Table S1: crystallographic information of compounds **1-5**.

	1	2	3
Formula	C ₄₄ H ₆₄ Cl ₄ O ₂₂ P ₄ Sb ₄ V ₂	C ₆₈ H ₁₁₀ Mn ₂ N ₂ O ₂₂ P ₄ Sb ₄	C ₆₇ Co ₂ H ₁₀₆ N ₂ O ₂₁ P ₄ Sb ₄
Formula weight	1799.51	2028.33	2004.27
Temperature/K	109(3)	100(2)	100(2)
Crystal system	monoclinic	monoclinic	monoclinic
Space group	<i>P2₁/n</i>	<i>C2/c</i>	<i>C2/c</i>
a/Å	9.9319(2)	25.252(3)	25.233(8)
b/Å	24.0942(5)	20.379(3)	20.100(8)
c/Å	13.5380(2)	16.0757(18)	16.085(6)
α/°	90	90	90
β/°	99.005(2)	91.123(2)	91.279(10)
γ/°	90	90	90
Volume/Å ³	3199.73(11)	8271.1(16)	8156(5)
Z	2	4	4
ρ _{calc} /g/cm ³	1.868	1.629	1.632
μ/mm ⁻¹	2.281	1.730	1.849
F(000)	1764.0	4088.0	4032.0
Crystal size/mm ³	0.23 × 0.19 × 0.14	0.16 × 0.14 × 0.13	0.23 × 0.17 × 0.13
Radiation	MoKα (λ = 0.71073)	MoKα (λ = 0.71073)	MoKα (λ = 0.71073)
2θ range (°)	4.484 to 50.698	2.568 to 52.838	2.59 to 53.728
Index ranges	-11 ≤ h ≤ 11, -29 ≤ k ≤ 28, -16 ≤ l ≤ 15	-31 ≤ h ≤ 31, -25 ≤ k ≤ 25, -20 ≤ l ≤ 20	-31 ≤ h ≤ 31, -25 ≤ k ≤ 25, -20 ≤ l ≤ 20
Reflections collected	29157	43796	43123
Independent reflections	5855	8474	8521
Goodness-of-fit on F ²	1.089	1.114	1.021
R ₁ (F) (I > 2σ(I))	0.0351	0.0356	0.0471
wR ₂ (F ²) (all data)	0.0908	0.0837	0.1058
Largest diff. peak/hole / e Å ⁻³	0.92/-0.87	1.17/-0.61	1.11/-0.69
completeness to θ _{max} , %	100.0	99.5	96.8

	4	5
Formula	C ₆₈ H ₁₁₀ N ₂ Ni ₂ O ₂₂ P ₄ Sb ₄	C ₆₈ H ₁₁₀ Cu ₂ N ₂ O ₂₂ P ₄ Sb ₄
Formula weight	2035.87	2045.53
Temperature/K	100(2)	100(2)
Crystal system	monoclinic	triclinic
Space group	<i>C2/c</i>	<i>P-1</i>
<i>a</i> /Å	25.249(3)	12.8207(14)
<i>b</i> /Å	20.009(3)	13.3637(15)
<i>c</i> /Å	16.163(2)	13.7013(15)
α /°	90	99.588(2)
β /°	91.427(2)	93.106(2)
γ /°	90	114.547(2)
Volume/Å ³	8163.4(18)	2085.5(4)
<i>Z</i>	4	1
$\rho_{\text{calc}}/\text{cm}^3$	1.657	1.629
μ/mm^{-1}	1.904	1.922
F(000)	4112.0	1030.0
Crystal size/mm ³	0.16 × 0.14 × 0.12	0.23 × 0.18 × 0.16
Radiation	MoK α ($\lambda = 0.71073$)	MoK α ($\lambda = 0.71073$)
2 θ range (°)	2.598 to 52.776	3.044 to 52.866
Index ranges	-31 ≤ <i>h</i> ≤ 31, -24 ≤ <i>k</i> ≤ 24, -20 ≤ <i>l</i> ≤ 20	-15 ≤ <i>h</i> ≤ 16, -16 ≤ <i>k</i> ≤ 16, -17 ≤ <i>l</i> ≤ 16
Reflections collected	43172	16745
Independent reflections	8334	8416
Goodness-of-fit on F ²	1.081	1.027
R ₁ (F) (<i>I</i> > 2 σ (<i>I</i>))	0.0365	0.0311
wR ₂ (F ²) (all data)	0.0873	0.0757
Largest diff. peak/hole / e Å ⁻³	1.14/-0.64	0.84/-0.52
completeness to θ_{max} , %	99.7	98.2

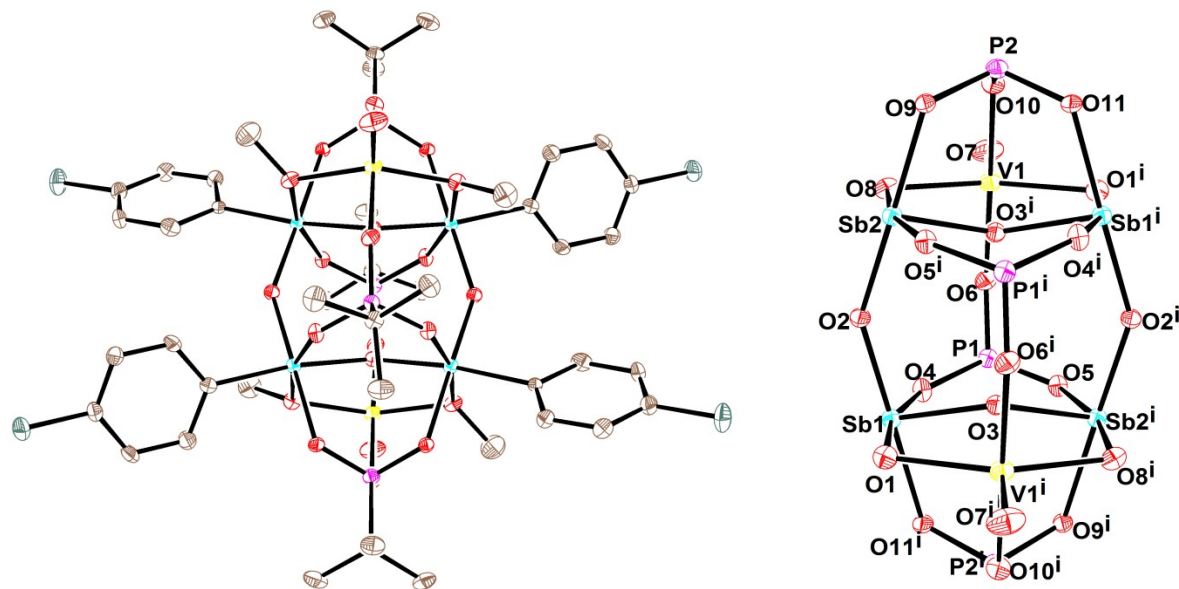


Figure S1: ORTEP view of **1** with thermal ellipsoids shown at 30% probability.

Table S2: Selected bond lengths (Å) and bond angles (deg) parameters of compound **1**.

Sb1-O4	2.005(3)	O3-Sb1-O11 ¹	86.59(11)	O6-V1-O1 ¹	93.74(11)
Sb1-O3	1.967(3)	O3-Sb1-O1	76.46(11)	O7-V1-O3 ¹	178.88(13)
Sb1-O2	1.921(3)	O2-Sb1-O4	92.25(11)	O7-V1-O8	109.20(15)
Sb1-O11 ¹	2.052(3)	O2-Sb1-O3	94.59(11)	O7-V1-O10	93.31(14)
Sb1-O1	2.048(3)	O2-Sb1-O11 ¹	177.78(11)	O7-V1-O1 ¹	108.97(15)
Sb2-O3 ¹	1.965(3)	O1-Sb1-O11 ¹	88.20(11)	O7-V1-O6	92.61(14)
Sb2-O8	2.047(3)	O31-Sb2-O8	77.12(11)	O4-P1-O5	111.24(16)
Sb2-O9	2.060(3)	O31-Sb2-O9	86.38(10)	O6-P1-O4	112.36(16)
Sb2-O2	1.927(3)	O31-Sb2-O5 ¹	90.11(11)	O6-P1-O5	111.89(16)
Sb2-O5 ¹	2.004(3)	O8-Sb2-O9	87.73(11)	O11-P2-O9	109.51(16)
V1-O3 ¹	2.230(3)	O2-Sb2-O3 ¹	94.17(11)	O10-P2-O9	110.98(16)
V1-O8	2.051(3)	O2-Sb2-O8	91.46(11)	O10-P2-O11	110.59(16)
V1-O10	2.000(3)	O2-Sb2-O9	178.89(10)	P1-O4-Sb1	136.01(16)
V1-O11	2.065(3)	O2-Sb2-O5 ¹	93.72(11)	Sb1-O3-V11	103.55(12)
V1-O6	1.987(3)	O51-Sb2-O8	166.54(11)	Sb2 ¹ -O3-Sb1	138.61(14)
V1-O7	1.602(3)	O51-Sb2-O9	87.24(11)	Sb2 ¹ -O3-V1 ¹	102.89(10)
P1-O4	1.542(3)	O8-V1-O3 ¹	71.37(10)	Sb2-O8-V1	106.52(12)
P1-O5	1.543(3)	O8-V1-O1 ¹	141.02(11)	P2-O9-Sb2	122.87(15)

P1-O6	1.490(3)	O10-V1-O3 ¹	87.69(10)	Sb1-O2-Sb2	146.09(15)
P2-O9	1.554(3)	O10-V1-O8	84.97(11)	P2-O11-Sb1 ¹	123.16(16)
P2-O11	1.548(3)	O10-V1-O1 ¹	85.30(11)	P1-O5-Sb2 ¹	137.08(17)
P2-O10	1.510(3)	O1 ¹ -V1-O3 ¹	70.60(10)	P2-O10-V1	123.13(16)
O4-Sb1-O11 ¹	85.85(11)	O6-V1-O3 ¹	86.40(11)	Sb1-O1-V1 ¹	106.77(13)
O4-Sb1-O1	166.00(11)	O6-V1-O8	92.12(12)	P1-O6-V1	153.53(18)
O3-Sb1-O4	90.54(11)	O6-V1-O10	173.99(12)		

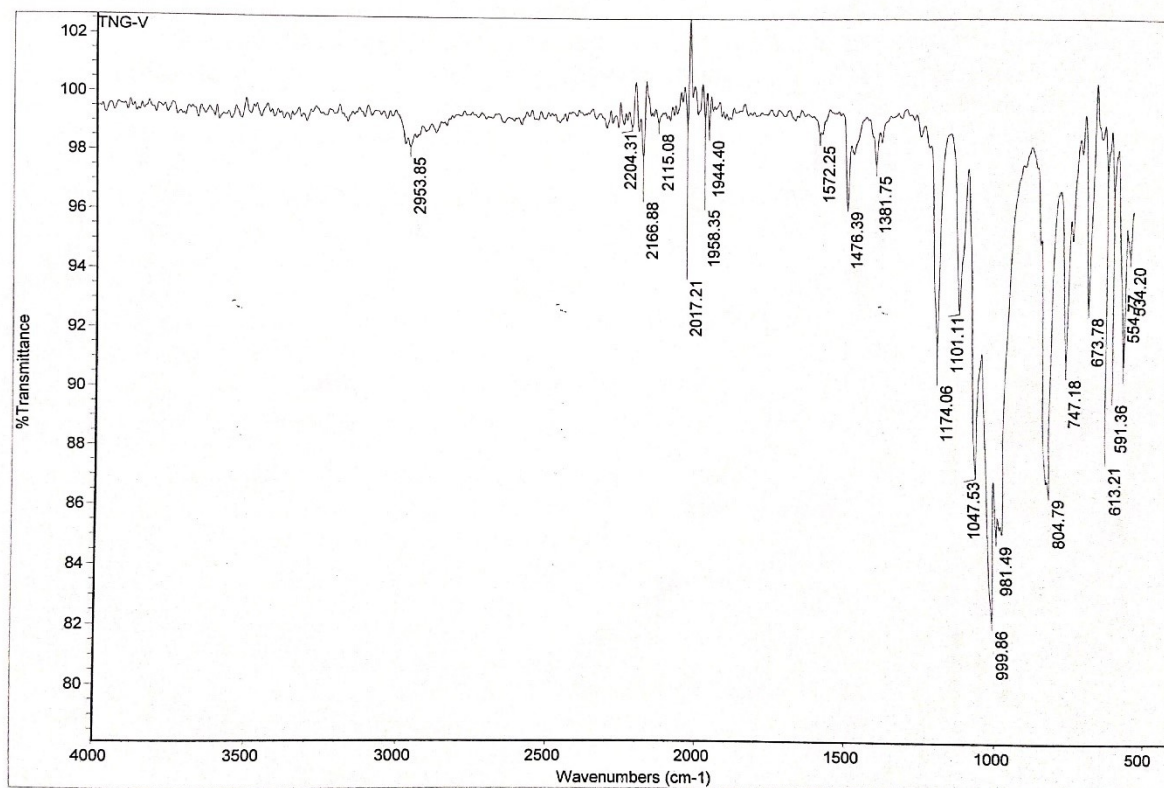


Figure S2: IR spectrum of compound **1**.

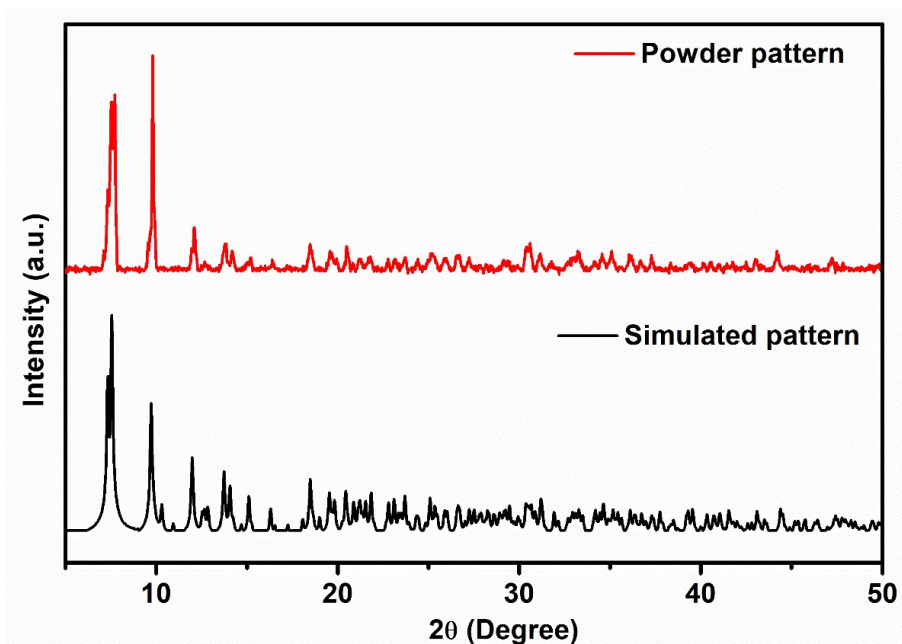
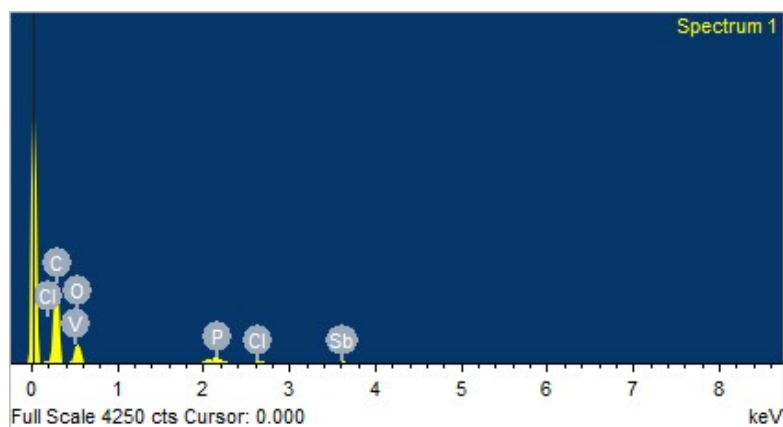


Figure S3: Powder X-ray diffraction pattern of a bulk sample of **1** compared to the simulated powder pattern extracted from single-crystal diffraction data.



Quantitative results

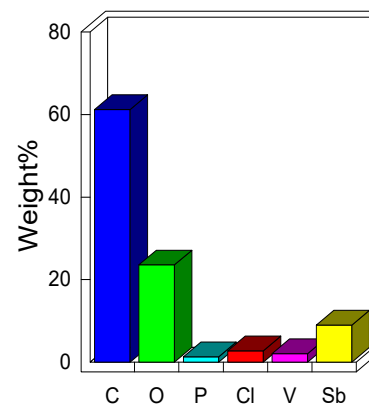


Figure S4: EDAX spectrum of compound **1**.

Table S3: EDAX analysis of **1**.

Element	Calculated weight%	Measured weight% from EDAX analysis
C	29.37	28.87
Cl	7.88	7.33
O	19.56	21.75
P	6.88	5.68
V	5.66	4.82
Sb	27.06	31.55
H	3.58	Not detected

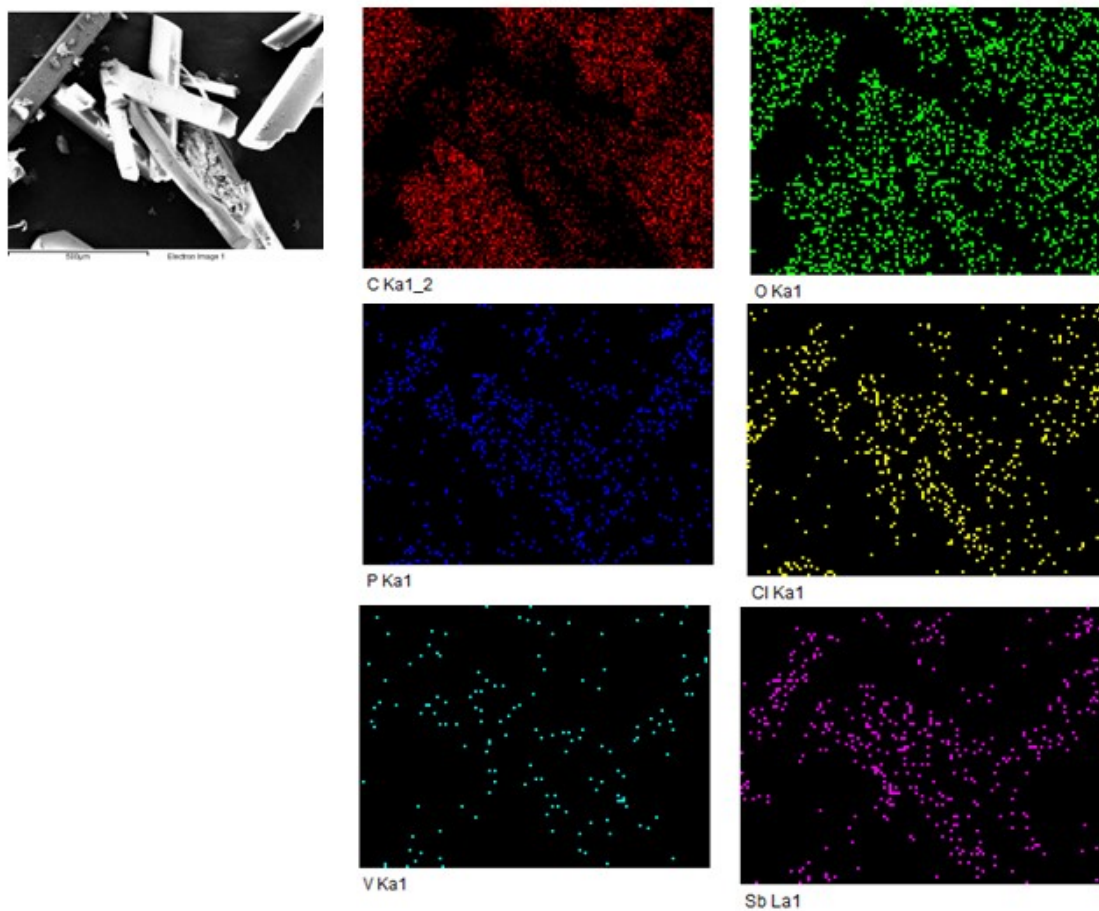


Figure S5: Elemental mapping of compound 1.

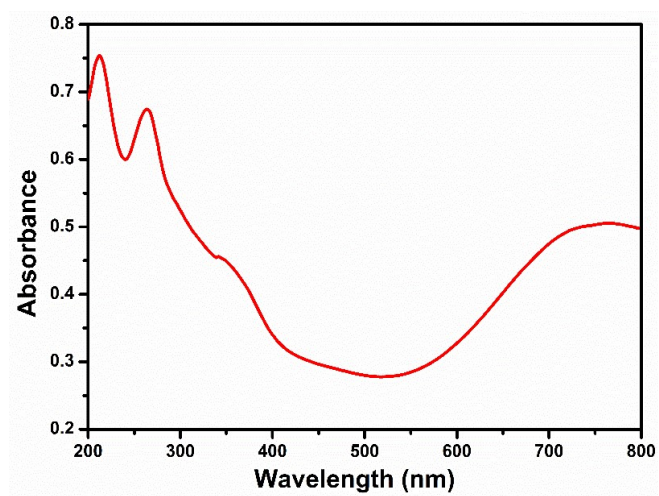
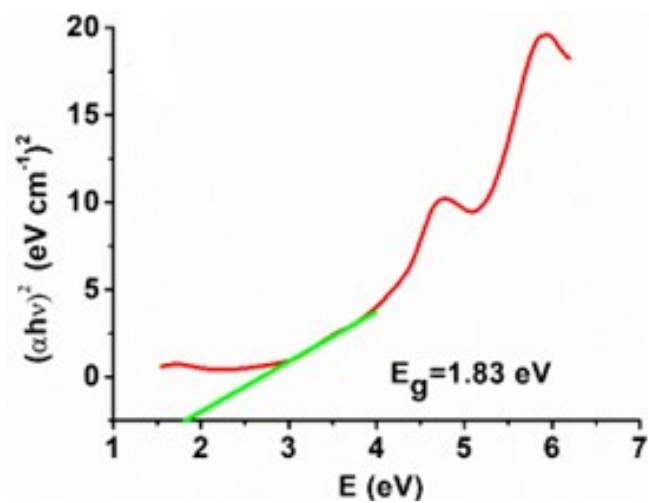
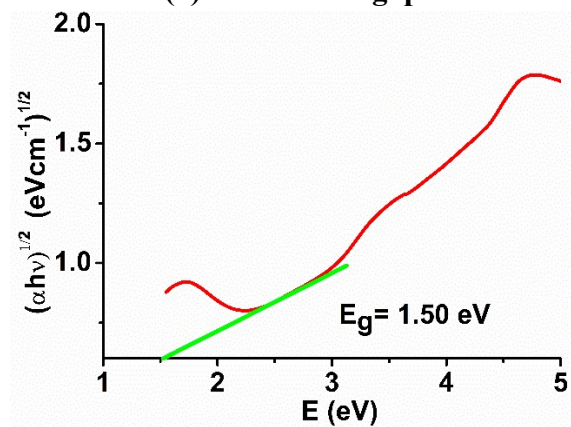


Figure S6: Solid-state UV-vis absorption spectrum of compound 1.



(a) Direct bandgap



(b) Indirect bandgap

Figure S7: Tauc plot of compound 1.

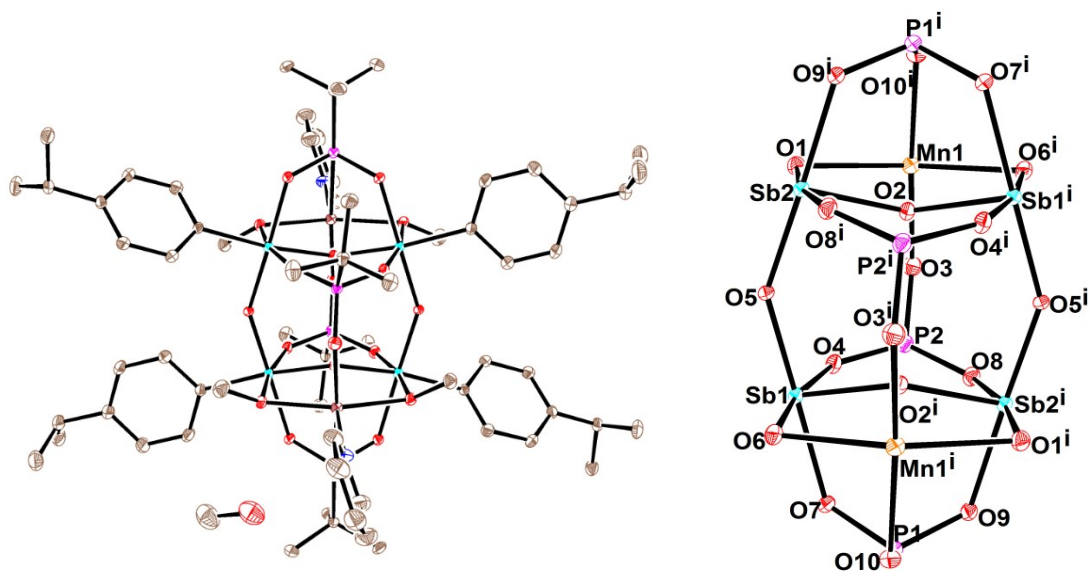


Figure S8: ORTEP view of **2** with thermal ellipsoids shown at 30% probability.

Table S4: Selected bond lengths (Å) and bond angles (deg) parameters of compound **2**.

Sb2-O5	1.926(2)	O5-Sb2-O9 ¹	178.57(9)	O10 ¹ -Mn1-N1	88.81(11)
Sb2-O1	2.024(2)	O1-Sb2-O9 ¹	86.86(10)	O3-Mn1-O1	97.67(10)
Sb2-O8 ¹	2.017(2)	O8 ¹ -Sb2-O1	169.68(10)	O3-Mn1-O6 ¹	95.34(9)
Sb2-O2	1.962(2)	O8 ¹ -Sb2-O9 ¹	87.70(10)	O3-Mn1-O2	92.07(9)
Sb2-O9 ¹	2.053(2)	O2-Sb2-O1	80.63(10)	O3-Mn1-O10 ¹	176.85(10)
Sb1-O5	1.927(2)	O2-Sb2-O8 ¹	90.31(10)	O3-Mn1-N1	90.97(11)
Sb1-O7	2.056(2)	O2-Sb2-O9 ¹	86.71(9)	N1-Mn1-O2	175.65(11)
Sb1-O6	2.015(2)	O5-Sb1-O7	178.64(10)	O9-P1-O7	109.16(14)
Sb1-O4	2.010(2)	O5-Sb1-O6	91.74(10)	O10-P1-O7	112.01(14)
Sb1-O2 ¹	1.962(2)	O5-Sb1-O4	92.02(10)	O10-P1-O9	112.30(14)
Mn1-O1	2.204(2)	O5-Sb1-O2 ¹	93.90(10)	O8-P2-O4	110.58(13)
Mn1-O6 ¹	2.199(2)	O6-Sb1-O7	86.90(10)	O3-P2-O4	112.80(14)
Mn1-O2	2.264(2)	O4-Sb1-O7	89.33(10)	O3-P2-O8	113.10(14)
Mn1-O10 ¹	2.181(3)	O4-Sb1-O6	170.39(9)	Sb2-O5-Sb1	145.94(13)
Mn1-O3	2.069(3)	O2 ¹ -Sb1-O7	85.96(9)	Sb2-O1-Mn1	103.44(10)
Mn1-N1	2.219(3)	O2 ¹ -Sb1-O6	80.86(10)	P1-O7-Sb1	124.73(14)
P1-O7	1.558(2)	O2 ¹ -Sb1-O4	90.06(10)	Sb1-O6-Mn1 ¹	103.87(10)
P1-O9	1.550(3)	O1-Mn1-O2	70.50(8)	P2-O4-Sb1	138.57(14)
P1-O10	1.506(3)	O1-Mn1-N1	106.01(12)	P2-O8-Sb2 ¹	137.75(14)
P2-O4	1.554(2)	O6 ¹ -Mn1-O1	139.26(9)	Sb2-O2-Sb1 ¹	138.73(13)
P2-O8	1.549(2)	O6 ¹ -Mn1-O2	70.60(9)	Sb2-O2-Mn1	103.39(9)

P2-O3	1.497(3)	O6 ¹ -Mn1-N1	112.20(12)	Sb11-O2-Mn1	103.33(10)
O5-Sb2-O1	93.21(10)	O10 ¹ -Mn1-O1	85.41(9)	P1-O9-Sb2 ¹	124.81(14)
O5-Sb2-O8 ¹	92.44(10)	O10 ¹ -Mn1-O6 ¹	81.84(9)	P1-O10-Mn1 ¹	116.50(14)
O5-Sb2-O2	94.72(9)	O10 ¹ -Mn1-O2	88.32(9)	P2-O3-Mn1	142.66(16)

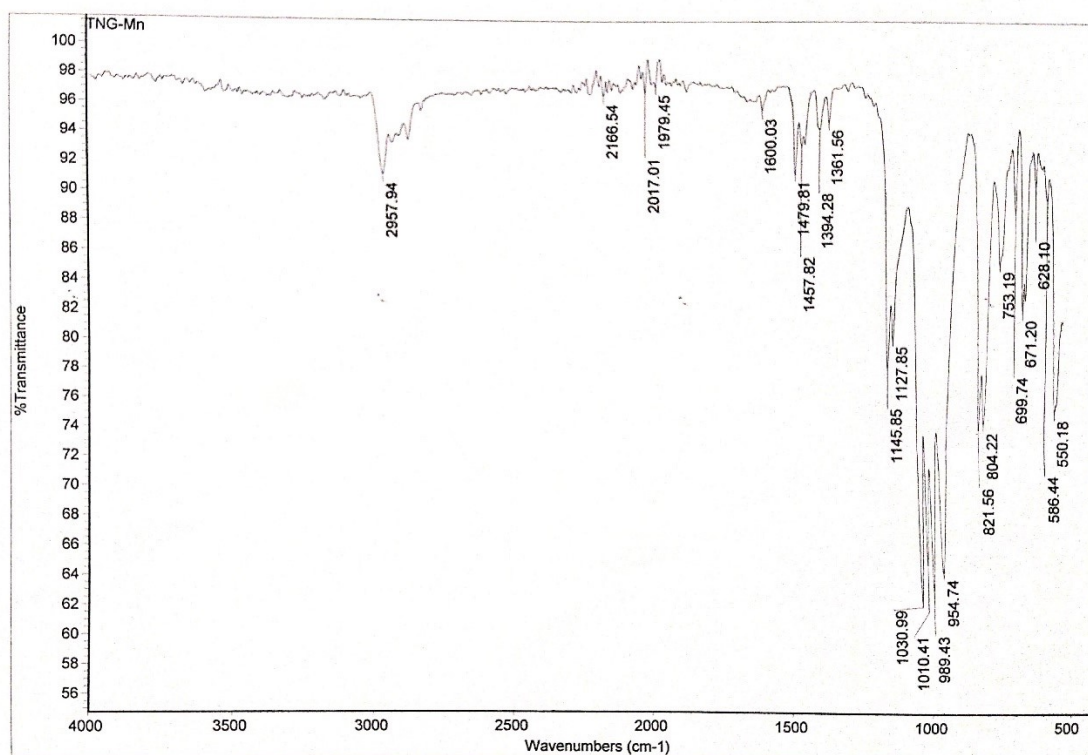


Figure S9: IR spectrum of compound 2.

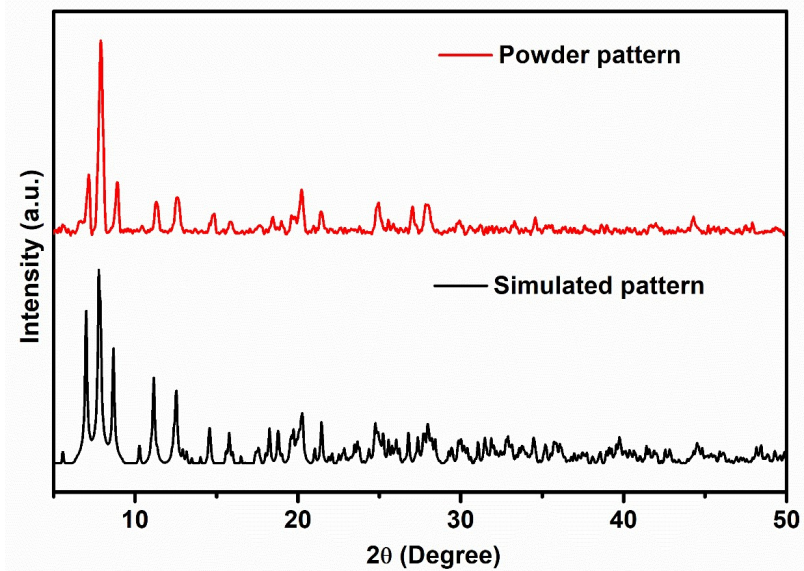


Figure S10: Powder X-ray diffraction pattern of a bulk sample of **2**, compared to the simulated powder pattern extracted from single-crystal diffraction data.

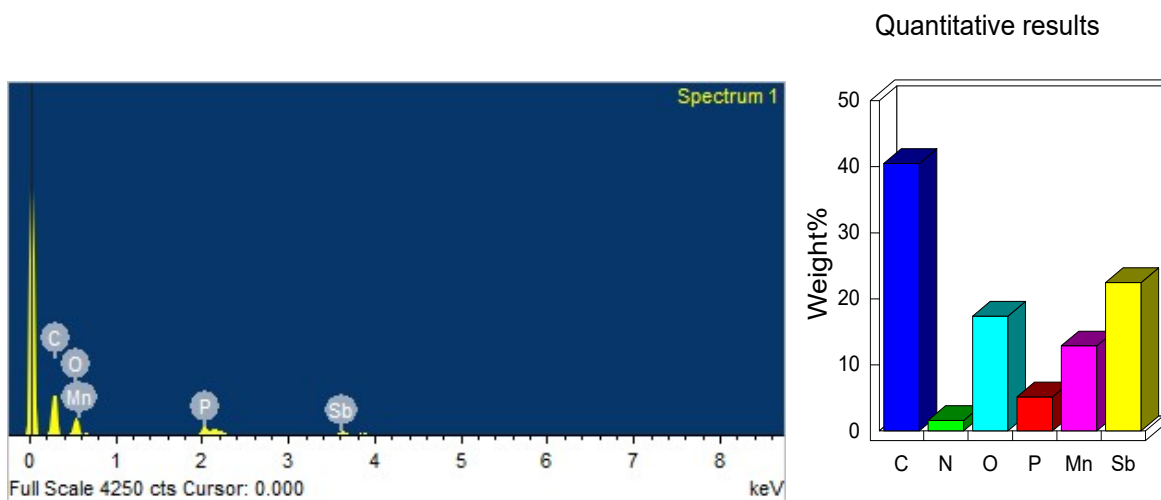


Figure S11: EDAX spectrum of compound **2**.

Table S5: EDAX analysis of **2**.

Element	Calculated weight%	Measured weight% from EDAX analysis
C	40.36	40.49
N	1.43	1.60
O	16.29	17.39
P	6.31	7.15
Mn	5.59	10.88
Sb	24.79	22.49
H	5.23	Not detected

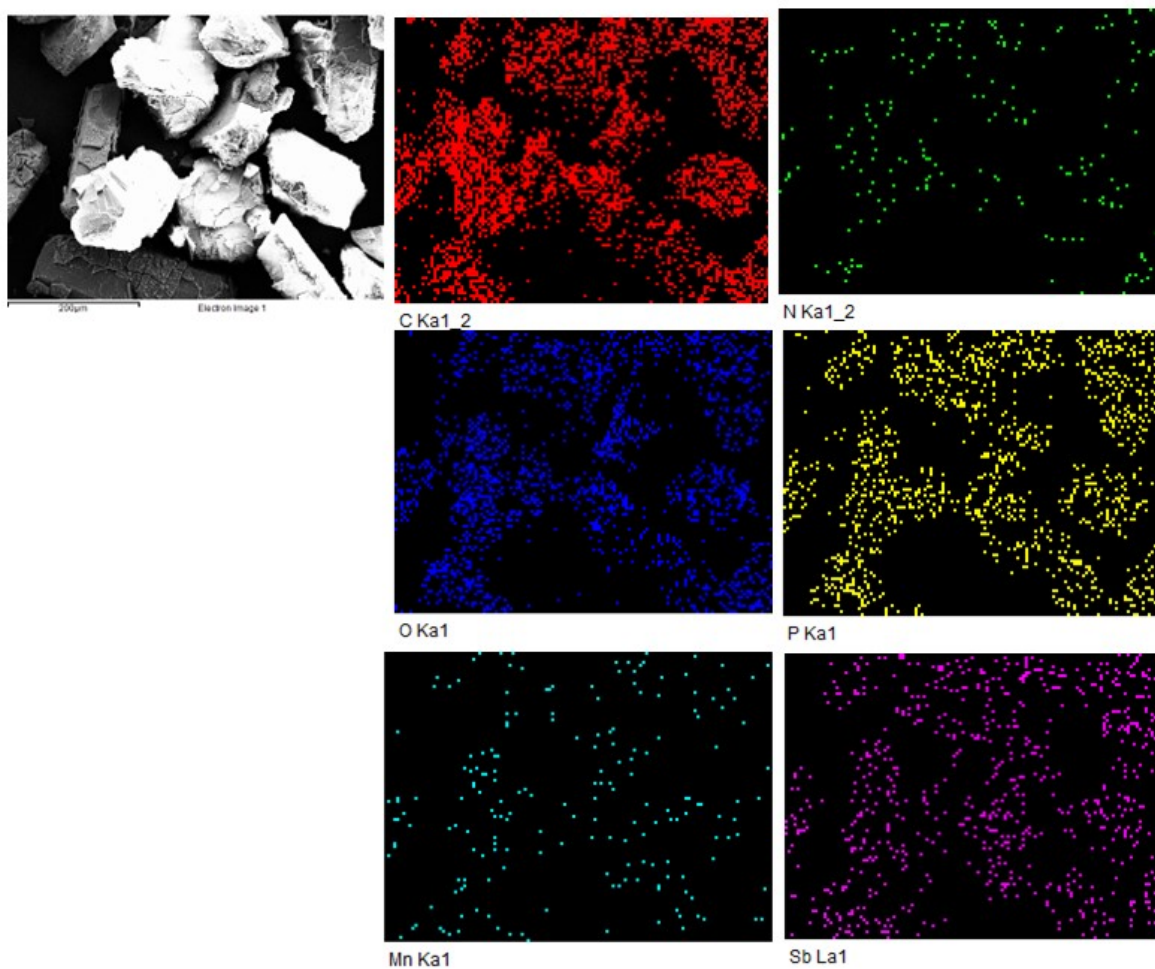


Figure S12: Elemental mapping of compound **2**.

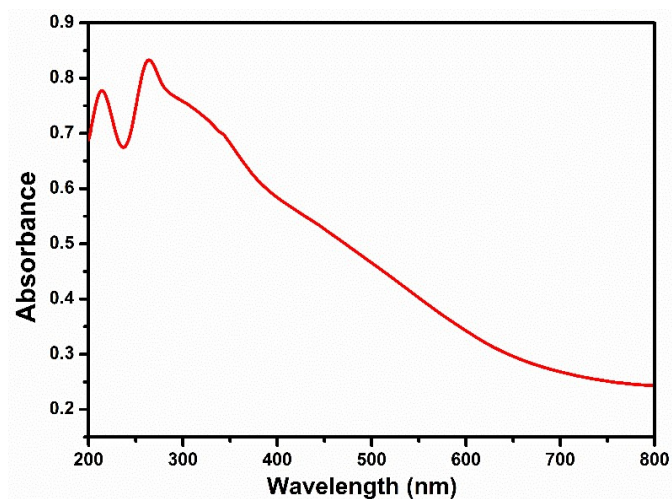
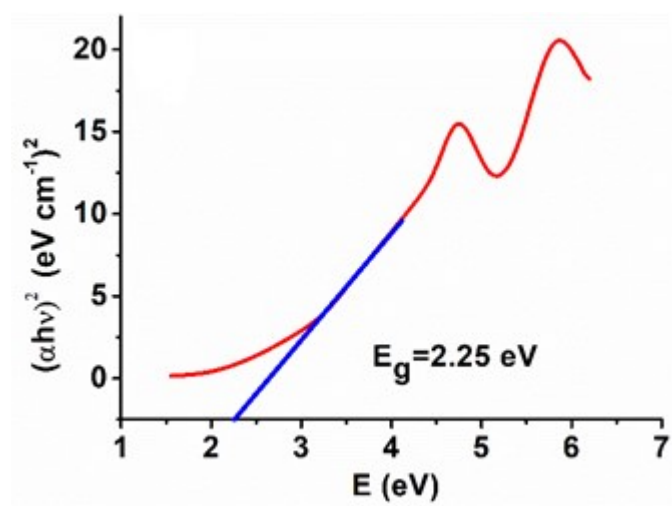
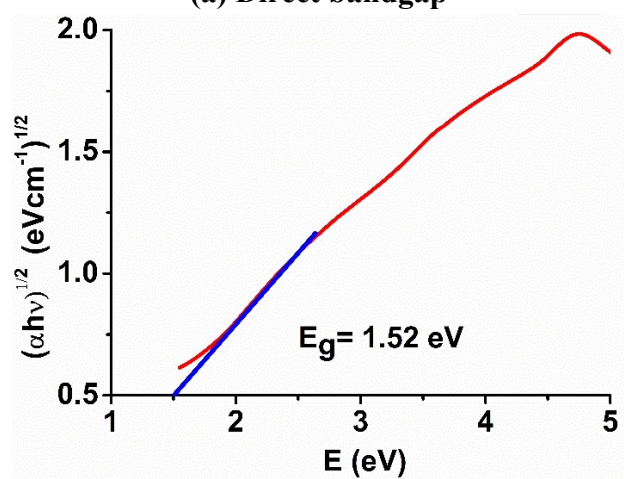


Figure S13: Solid-state UV-vis absorption spectrum of compound 2.



(a) Direct bandgap



(b) Indirect bandgap

Figure S14: Tauc plot of compound 2.

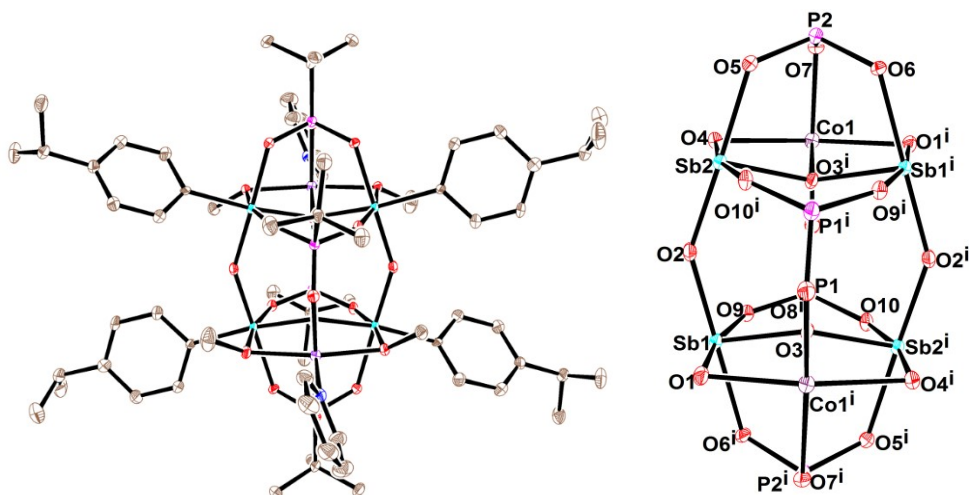


Figure S15: ORTEP view of **3** with thermal ellipsoids shown at 30% probability.

Table S6: Selected bond lengths (Å) and bond angles (deg) parameters of compound **3**.

Sb2-O10 ¹	2.017(3)	O3 ¹ -Sb2-O4	79.74(13)	O8-Co1-O4	97.01(13)
Sb2-O3 ¹	1.967(3)	O3 ¹ -Sb2-O5	86.95(13)	O8-Co1-O1 ¹	94.72(13)
Sb2-O2	1.918(3)	O2-Sb2-O10 ¹	91.66(14)	O8-Co1-O7	176.28(13)
Sb2-O4	2.031(3)	O2-Sb2-O3 ¹	94.88(13)	O8-Co1-N1	89.32(16)
Sb2-O5	2.050(3)	O2-Sb2-O4	94.30(14)	N1-Co1-O3 ¹	177.95(14)
Sb1-O3	1.960(3)	O2-Sb2-O5	177.67(14)	N1-Co1-O4	104.32(15)
Sb1-O6 ¹	2.059(3)	O4-Sb2-O5	87.43(13)	N1-Co1-O1 ¹	108.30(14)
Sb1-O2	1.921(3)	O3-Sb1-O6 ¹	86.08(13)	O5-P2-O6	109.53(19)
Sb1-O1	2.016(3)	O3-Sb1-O1	79.81(13)	O7-P2-O6	112.03(19)
Sb1-O9	2.007(3)	O3-Sb1-O9	89.74(13)	O7-P2-O5	111.99(19)
Co1-O3 ¹	2.145(3)	O2-Sb1-O3	94.24(13)	O9-P1-O10	110.62(19)
Co1-O4	2.122(3)	O2-Sb1-O6 ¹	179.51(14)	O8-P1-O10	113.30(19)
Co1-O1 ¹	2.127(3)	O2-Sb1-O1	92.91(14)	O8-P1-O9	112.84(19)
Co1-O7	2.088(3)	O2-Sb1-O9	91.18(14)	P1-O10-Sb2 ¹	137.0(2)
Co1-O8	2.003(3)	O1-Sb1-O6 ¹	87.51(13)	Sb2 ¹ -O3-Co1 ¹	102.84(13)
Co1-N1	2.118(4)	O9-Sb1-O6 ¹	88.45(13)	Sb1-O3-Sb2 ¹	138.87(18)
P2-O6	1.550(3)	O9-Sb1-O1	169.02(12)	Sb1-O3-Co1 ¹	103.25(14)
P2-O5	1.549(3)	O4-Co1-O3 ¹	73.84(12)	P2-O6-Sb1 ¹	123.50(19)
P2-O7	1.501(4)	O4-Co1-O1 ¹	145.36(13)	Sb2-O2-Sb1	145.98(17)
P1-O10	1.548(3)	O1 ¹ -Co1-O3 ¹	73.33(12)	Sb2-O4-Co1	101.47(14)
P1-O9	1.545(3)	O7-Co1-O3 ¹	91.04(13)	Sb1-O1-Co1 ¹	101.95(14)
P1-O8	1.500(3)	O7-Co1-O4	86.10(13)	P1-O9-Sb1	138.5(2)
O10 ¹ -Sb2-O4	168.60(13)	O7-Co1-O1 ¹	83.78(13)	P2-O5-Sb2	123.25(19)
O10 ¹ -Sb2-O5	86.90(13)	O7-Co1-N1	87.93(16)	P2-O7-Co1	116.45(19)

O3 ¹ -Sb2-O10 ¹	90.09(13)	O8-Co1-O3 ¹	91.79(13)	P1-O8-Co1	144.3(2)
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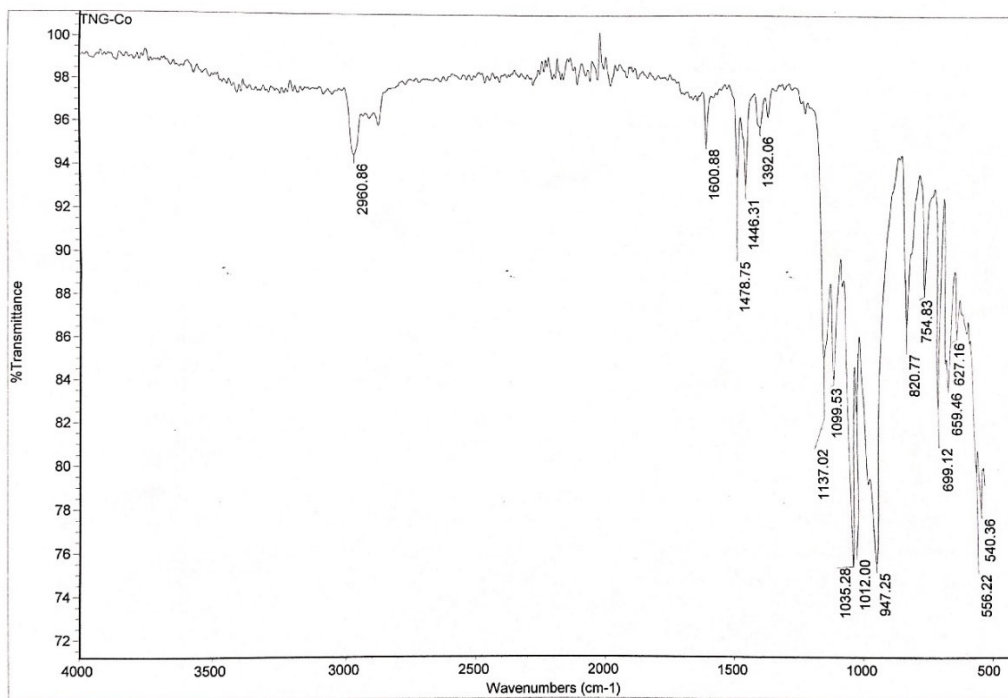


Figure S16: IR spectrum of compound 3.

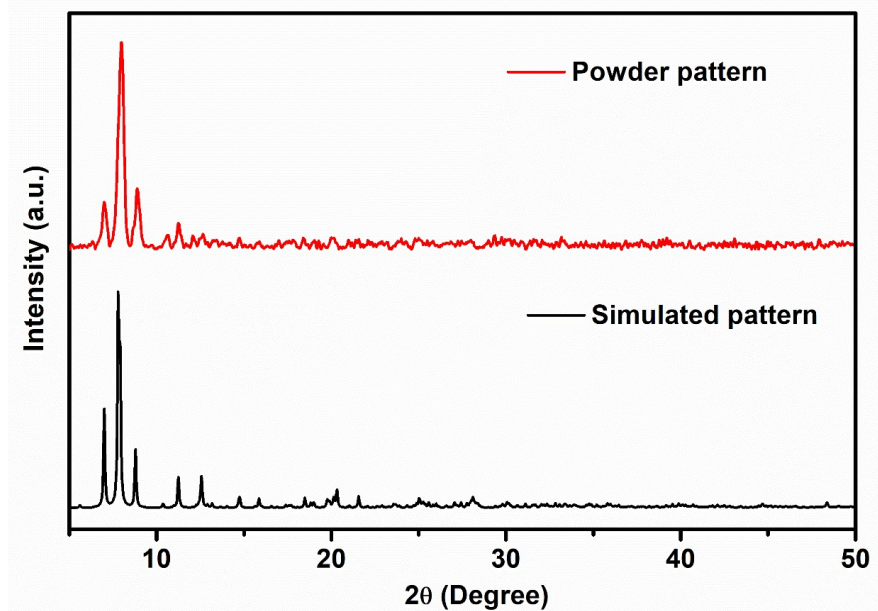


Figure S17: Powder X-ray diffraction pattern of a bulk sample of **3** compared to the simulated powder pattern extracted from single-crystal diffraction data.

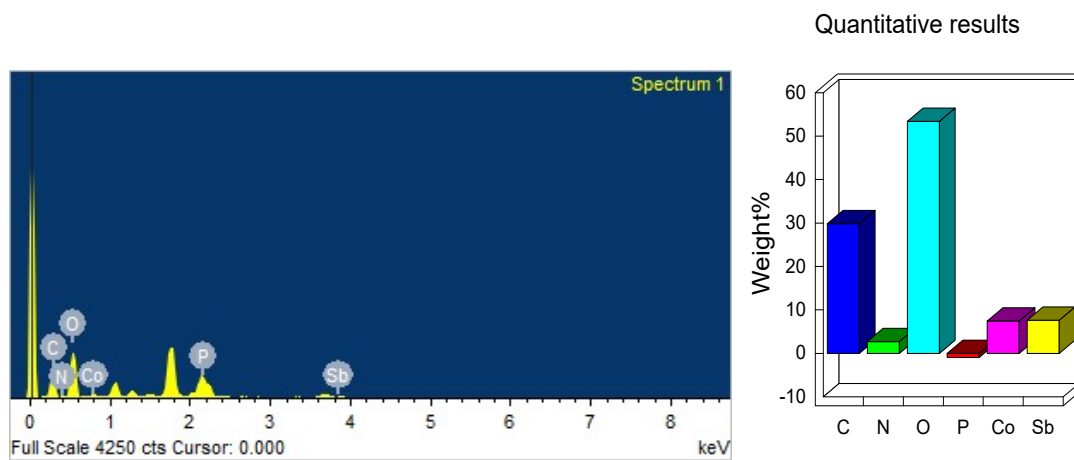


Figure S18: EDAX spectrum of compound **3**.

Table S7: EDAX analysis of **3**.

Element	Calculated weight%	Measured weight% from EDAX analysis
C	40.19	41.79
N	1.42	2.05
O	16.22	18.07
P	6.28	6.09
Co	5.98	8.11
Sb	24.69	23.89
H	5.21	Not detected

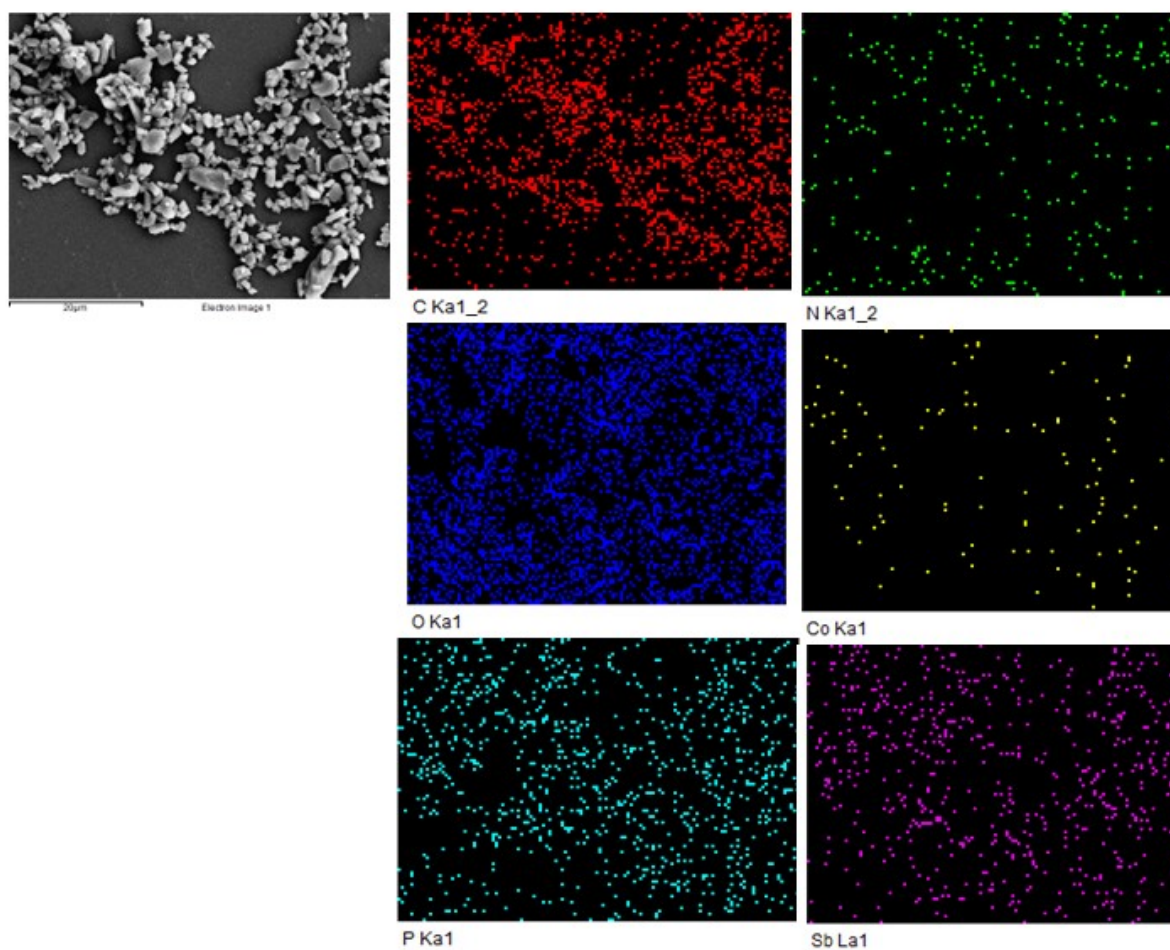


Figure S19: Elemental mapping of compound **3**.

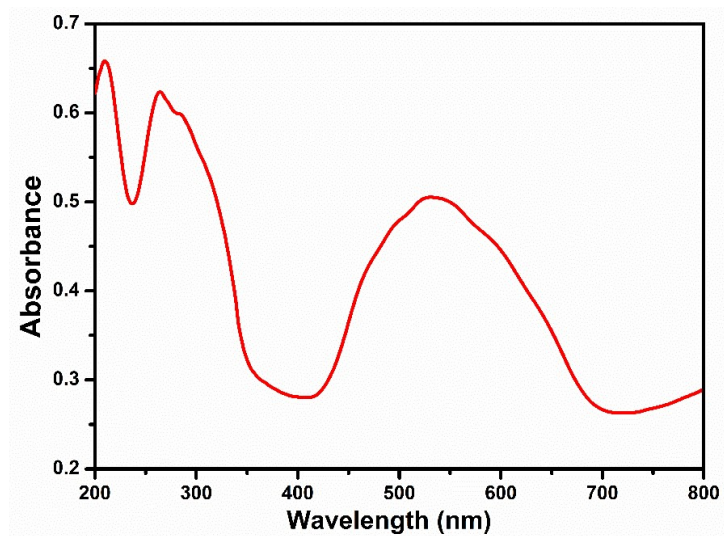
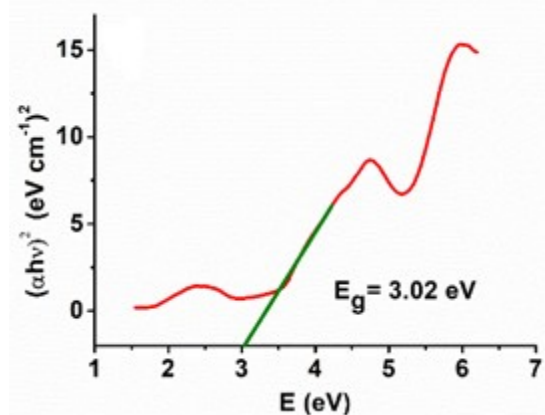
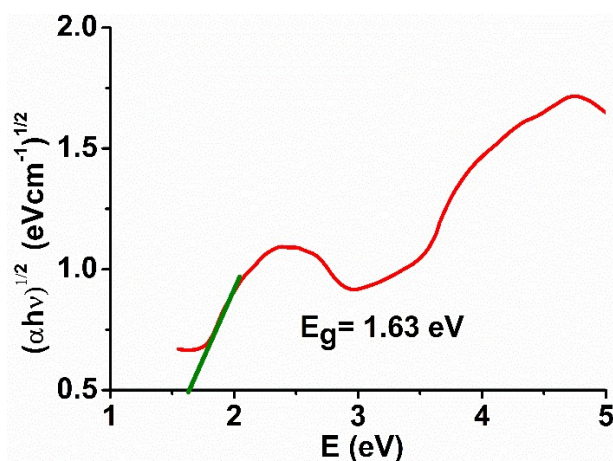


Figure S20: Solid-state UV-vis absorption spectrum of compound 3.



(a) Direct bandgap



(b) Indirect bandgap

Figure S21: Tauc plot of compound 3.

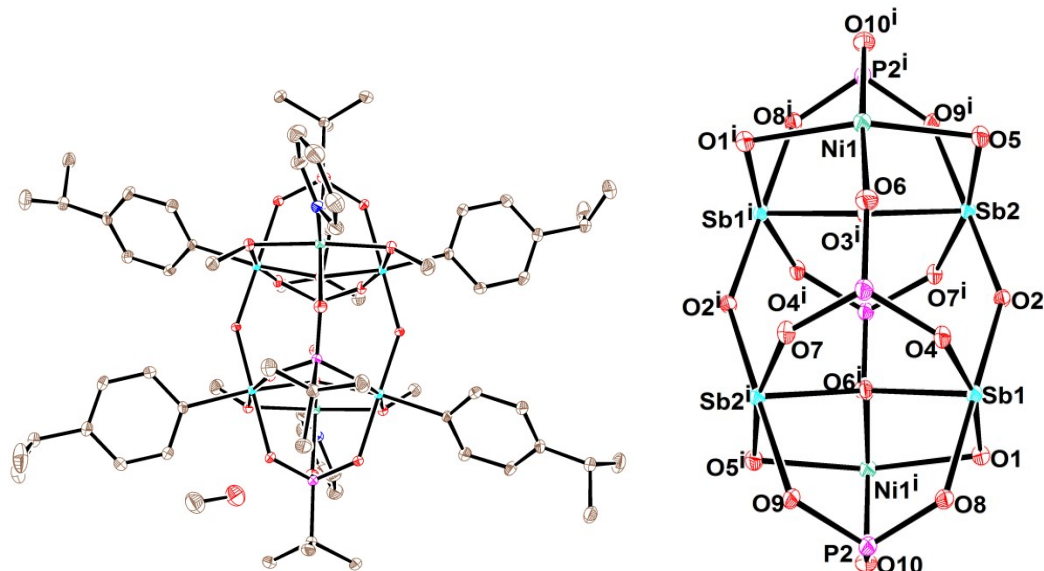


Figure S22: ORTEP view of **4** with thermal ellipsoids shown at 30% probability.

Table S8: Selected bond lengths (Å) and bond angles (deg) parameters of compound **4**.

Sb2-O2	1.926(2)	O2-Sb2-O7 ¹	91.03(11)	O6-Ni1-O1 ¹	96.73(10)
Sb2-O9 ¹	2.053(2)	O31-Sb2-O91	86.24(9)	O6-Ni1-O10 ¹	174.90(10)
Sb2-O3 ¹	1.974(2)	O3 ¹ -Sb2-O5	79.18(10)	O6-Ni1-N1	89.03(11)
Sb2-O5	2.025(3)	O31-Sb2-O7 ¹	89.84(10)	N1-Ni1-O3 ¹	177.87(11)
Sb2-O7 ¹	2.003(2)	O5-Sb2-O9 ¹	87.87(10)	N1-Ni1-O5	106.82(11)
Sb1-O2	1.921(2)	O7 ¹ -Sb2-O9 ¹	87.65(10)	N1-Ni1-O1 ¹	102.33(11)
Sb1-O3	1.972(2)	O7 ¹ -Sb2-O5	168.39(10)	N1-Ni1-O10 ¹	87.08(12)
Sb1-O4	2.010(2)	O2-Sb1-O3	94.72(10)	O7-P1-O4	110.50(14)
Sb1-O1	2.039(3)	O2-Sb1-O4	91.57(10)	O6-P1-O4	113.70(14)
Sb1-O8	2.050(2)	O2-Sb1-O1	95.00(11)	O6-P1-O7	113.15(14)
Ni1-O3 ¹	2.076(3)	O2-Sb1-O8	177.23(11)	O8-P2-O9	109.56(14)
Ni1-O5	2.100(3)	O3-Sb1-O4	90.21(10)	O10-P2-O9	111.51(15)
Ni1-O1 ¹	2.090(2)	O3-Sb1-O1	78.98(10)	O10-P2-O8	112.00(15)
Ni1-O10 ¹	2.076(3)	O3-Sb1-O8	86.72(10)	Sb1-O2-Sb2	146.83(13)
Ni1-O6	2.002(2)	O4-Sb1-O1	167.76(10)	P2-O9-Sb2 ¹	123.17(14)
Ni1-N1	2.069(3)	O4-Sb1-O8	86.06(10)	Sb21-O3-Ni1 ¹	103.27(10)
P1-O4	1.555(3)	O1-Sb1-O8	87.59(10)	Sb1-O3-Sb2 ¹	138.56(13)
P1-O7	1.549(2)	O3 ¹ -Ni1-O5	75.24(9)	Sb1-O3-Ni1 ¹	103.00(10)

P1-O6	1.491(3)	O3 ¹ -Ni1-O1 ¹	75.54(9)	P1-O4-Sb1	136.55(15)
P2-O9	1.558(2)	O3 ¹ -Ni1-O10 ¹	92.63(10)	Sb2-O5-Ni1	100.64(10)
P2-O8	1.547(3)	O1 ¹ -Ni1-O5	148.87(10)	Sb1-O1-Ni1 ¹	100.23(11)
P2-O10	1.505(3)	O10 ¹ -Ni1-O5	83.52(10)	P2-O8-Sb1	123.60(14)
O2-Sb2-O9 ¹	178.68(11)	O10 ¹ -Ni1-O1 ¹	87.32(10)	P1-O7-Sb2 ¹	138.10(15)
O2-Sb2-O3 ¹	93.78(10)	O6-Ni1-O3 ¹	91.39(10)	P2-O10-Ni1 ¹	115.46(15)
O2-Sb2-O5	93.44(10)	O6-Ni1-O5	94.47(10)	P1-O6-Ni1	145.11(17)

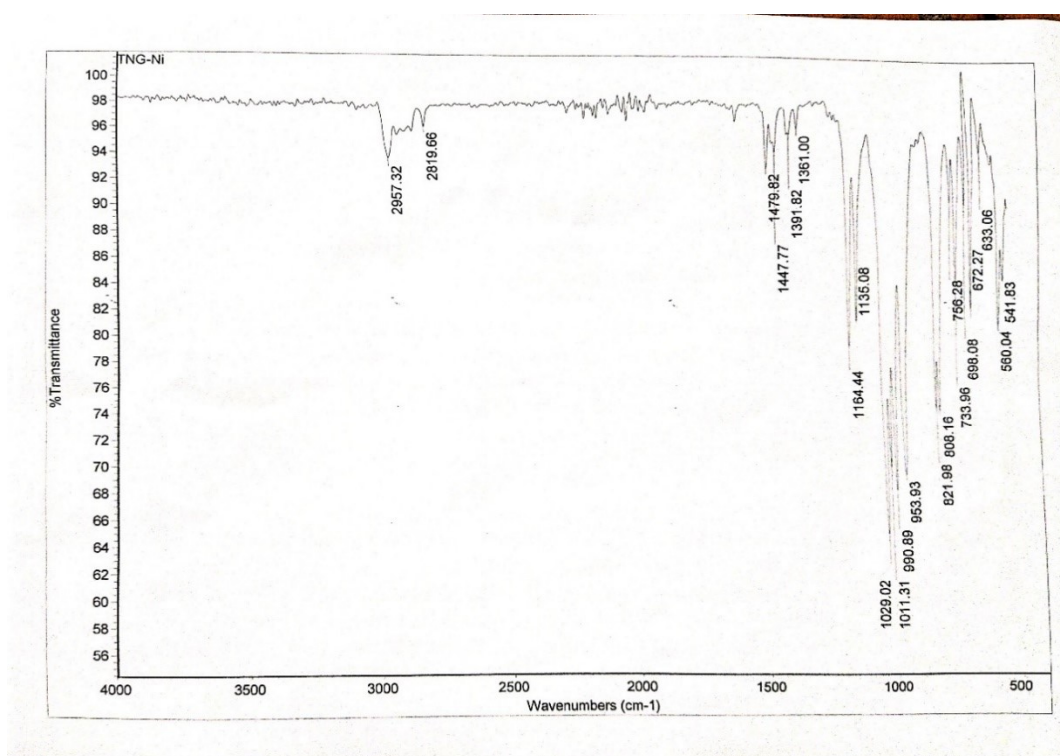


Figure S23: IR spectrum of compound 4.

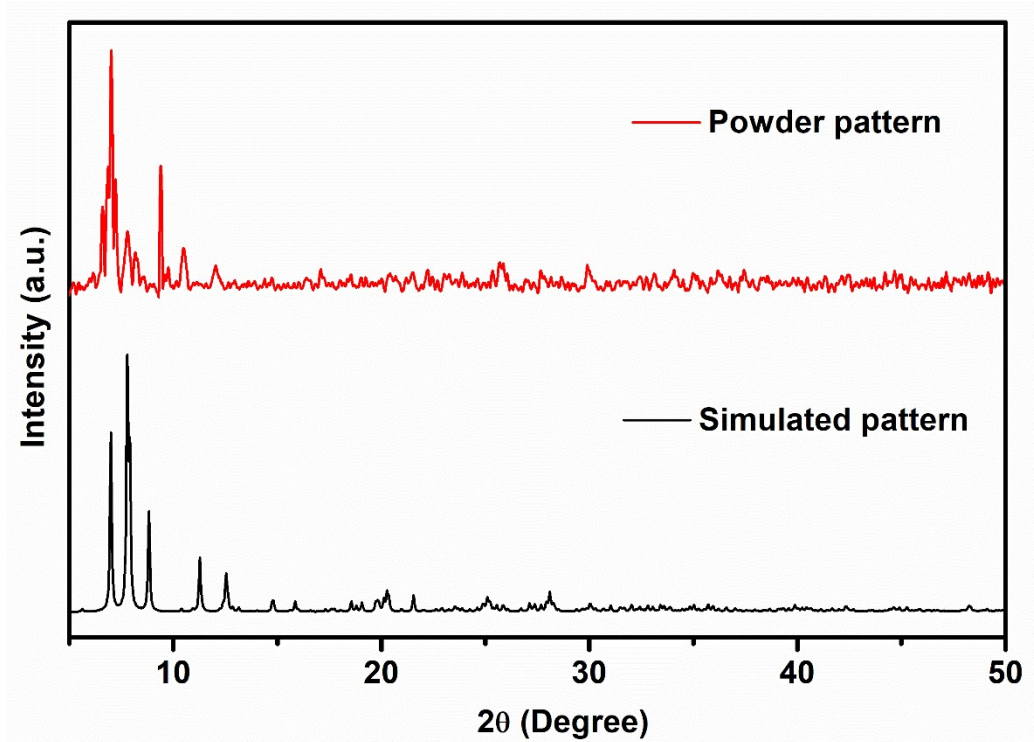


Figure S24: Powder X-ray diffraction pattern of a bulk sample of **4** compared to the simulated powder pattern extracted from single-crystal diffraction data.

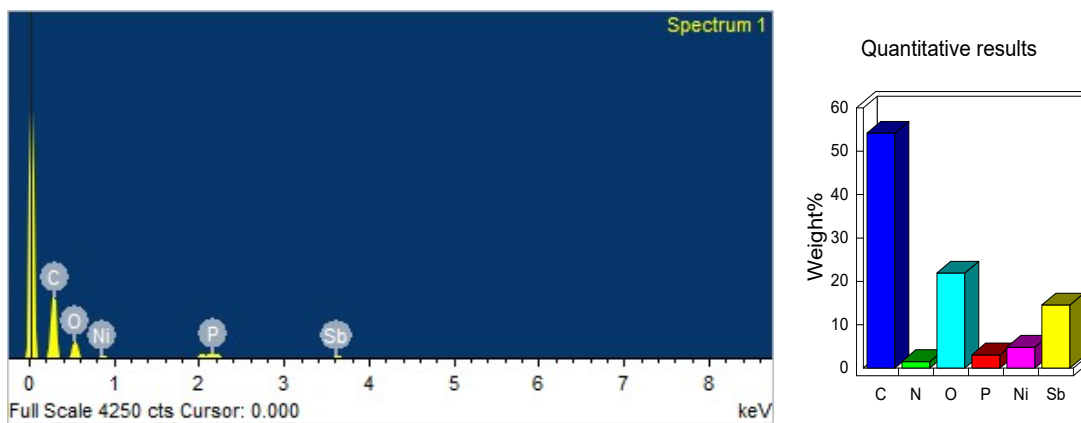


Figure S25: EDAX spectrum of compound **4**.

Table S9: EDAX analysis of 4.

Element	Calculated weight%	Measured weight% from EDAX analysis
C	40.20	43.92
N	1.42	1.79
O	16.23	18.47
P	6.28	6.56
Ni	5.95	7.34
Sb	24.70	21.92
H	5.21	Not detected

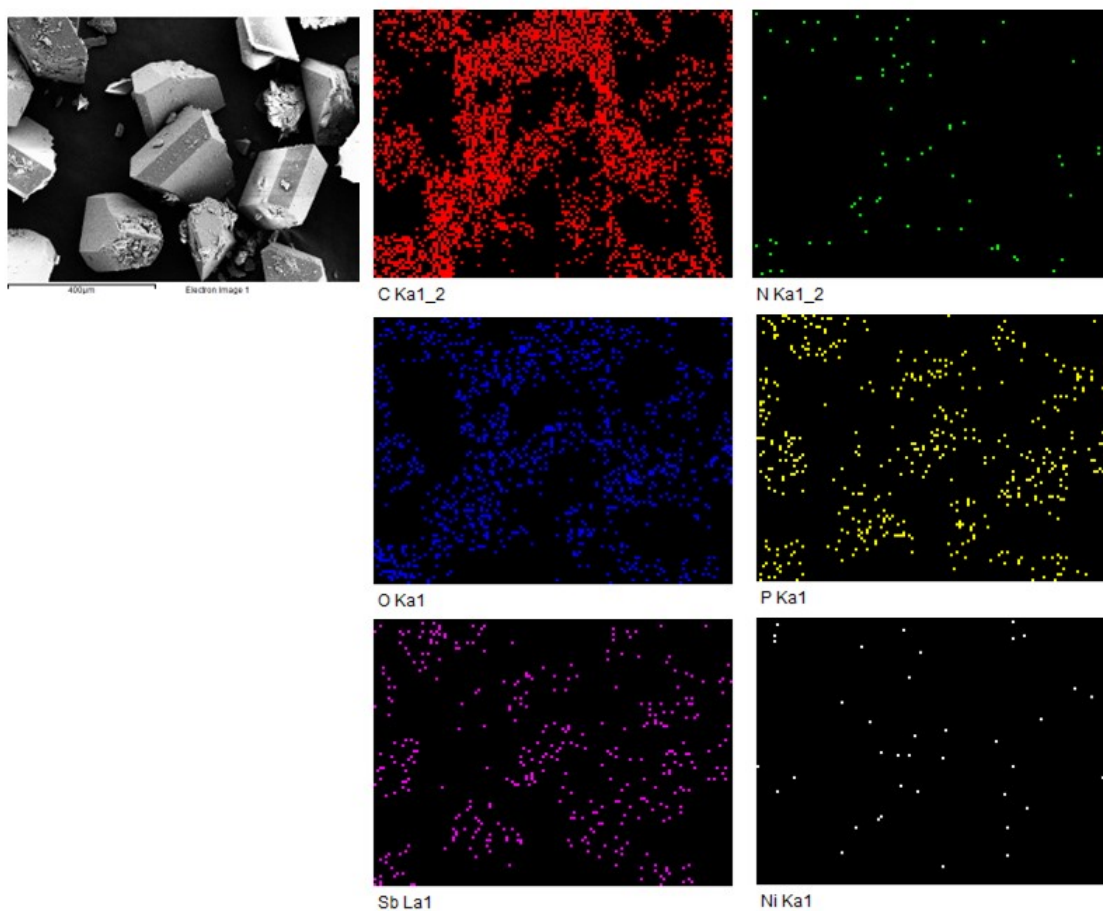


Figure S26: Elemental mapping of compound 4.

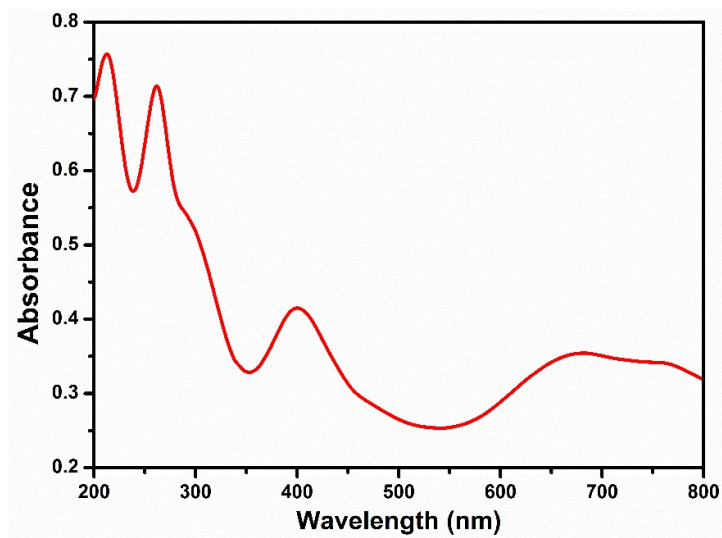
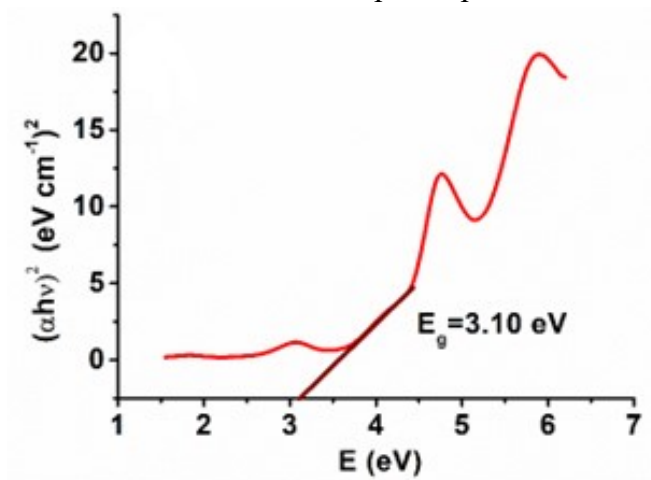
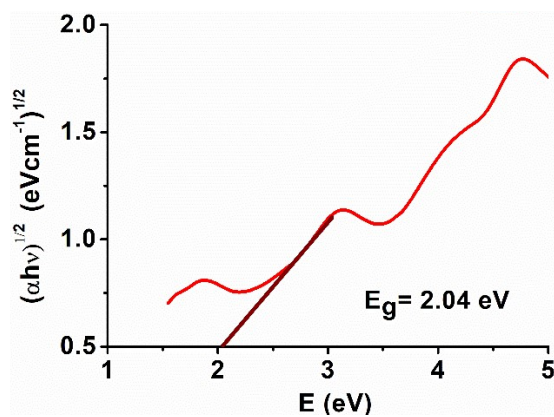


Figure S27: Solid-state UV-vis absorption spectrum of compound 4.



(a) Direct bandgap



(b) Indirect bandgap

Figure S28: Tauc plot of compound 4.

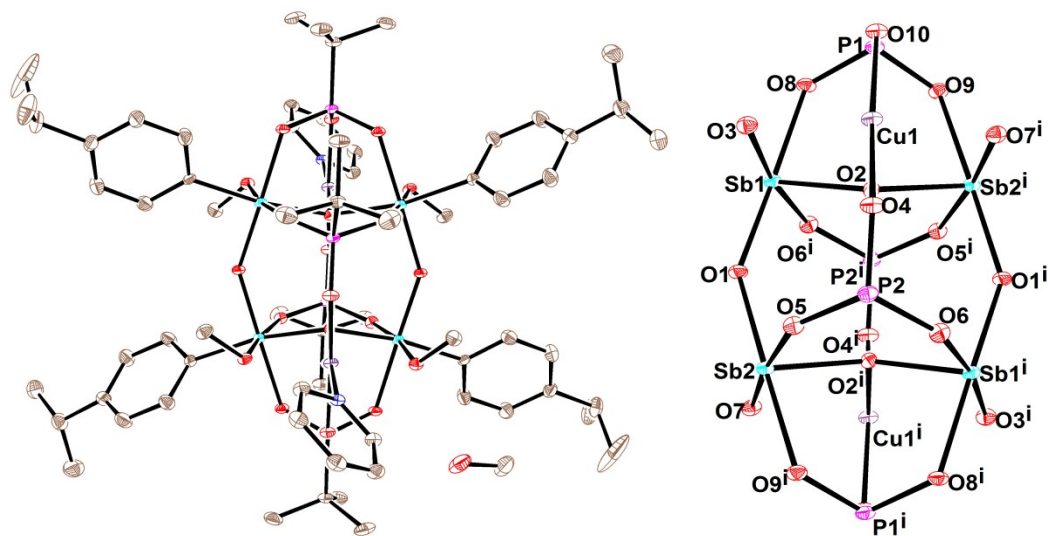


Figure S29: ORTEP view of **5** with thermal ellipsoids shown at 30% probability.

Table S10: Selected bond lengths (Å) and bond angles (deg) parameters of compound **5**.

Sb1-O2	2.0074(19)	O3-Sb1-O8	84.55(8)	O4-Cu1-O3	99.63(8)
Sb1-O3	2.015(2)	O3-Sb1-O6 ¹	171.53(8)	O4-Cu1-O7 ¹	94.91(8)
Sb1-O1	1.915(2)	O1-Sb1-O2	94.80(8)	O4-Cu1-O10	175.12(8)
Sb1-O8	2.046(2)	O1-Sb1-O3	93.10(9)	O4-Cu1-N1	88.70(10)
Sb1-O6 ¹	2.016(2)	O1-Sb1-O8	177.63(9)	N1-Cu1-O2	176.02(9)
Sb2-O2 ¹	2.013(2)	O1-Sb1-O6 ¹	93.03(9)	N1-Cu1-O3	101.78(9)
Sb2-O9 ¹	2.043(2)	O6 ¹ -Sb1-O8	89.34(9)	N1-Cu1-O7 ¹	109.94(9)
Sb2-O1	1.926(2)	O2 ¹ -Sb2-O9 ¹	84.21(8)	O8-P1-O9	109.15(12)
Sb2-O7	1.998(2)	O1-Sb2-O2 ¹	95.64(8)	O10-P1-O9	112.38(12)
Sb2-O5	2.011(2)	O1-Sb2-O9 ¹	177.85(8)	O10-P1-O8	112.13(12)
Cu1-O2	2.076(2)	O1-Sb2-O7	92.71(8)	O5-P2-O6	110.10(12)
Cu1-O3	2.348(2)	O1-Sb2-O5	88.92(9)	O4-P2-O5	113.52(12)
Cu1-O7 ¹	2.326(2)	O7-Sb2-O2 ¹	82.51(8)	O4-P2-O6	113.14(13)
Cu1-O10	1.985(2)	O7-Sb2-O9 ¹	89.39(8)	Sb1-O2-Sb2 ¹	136.38(10)
Cu1-O4	1.940(2)	O7-Sb2-O5	173.52(8)	Sb1-O2-Cu1	105.12(9)
Cu1-N1	2.004(3)	O5-Sb2-O2 ¹	91.09(8)	Sb2 ¹ -O2-Cu1	105.06(8)
P1-O9	1.549(2)	O5-Sb2-O9 ¹	88.94(8)	P1-O9-Sb2 ¹	123.54(12)
P1-O8	1.547(2)	O2-Cu1-O3	74.34(7)	Sb1-O3-Cu1	95.71(8)
P1-O10	1.509(2)	O2-Cu1-O7 ¹	73.60(7)	Sb1-O1-Sb2	145.46(12)
P2-O5	1.543(2)	O7 ¹ -Cu1-O3	145.29(7)	P1-O8-Sb1	123.23(12)
P2-O4	1.506(2)	O10-Cu1-O2	91.20(8)	Sb2-O7-Cu1 ¹	96.99(8)
P2-O6	1.543(2)	O10-Cu1-O3	84.12(8)	P2-O5-Sb2	138.77(13)

O2-Sb1-O3	83.63(8)	O10-Cu1-O7 ¹	83.58(8)	P1-O10-Cu1	118.23(12)
O2-Sb1-O8	85.30(8)	O10-Cu1-N1	87.46(10)	P2-O4-Cu1	143.44(13)
O2-Sb1-O6 ¹	90.04(8)	O4-Cu1-O2	92.82(8)	P2-O6-Sb1 ¹	139.86(13)

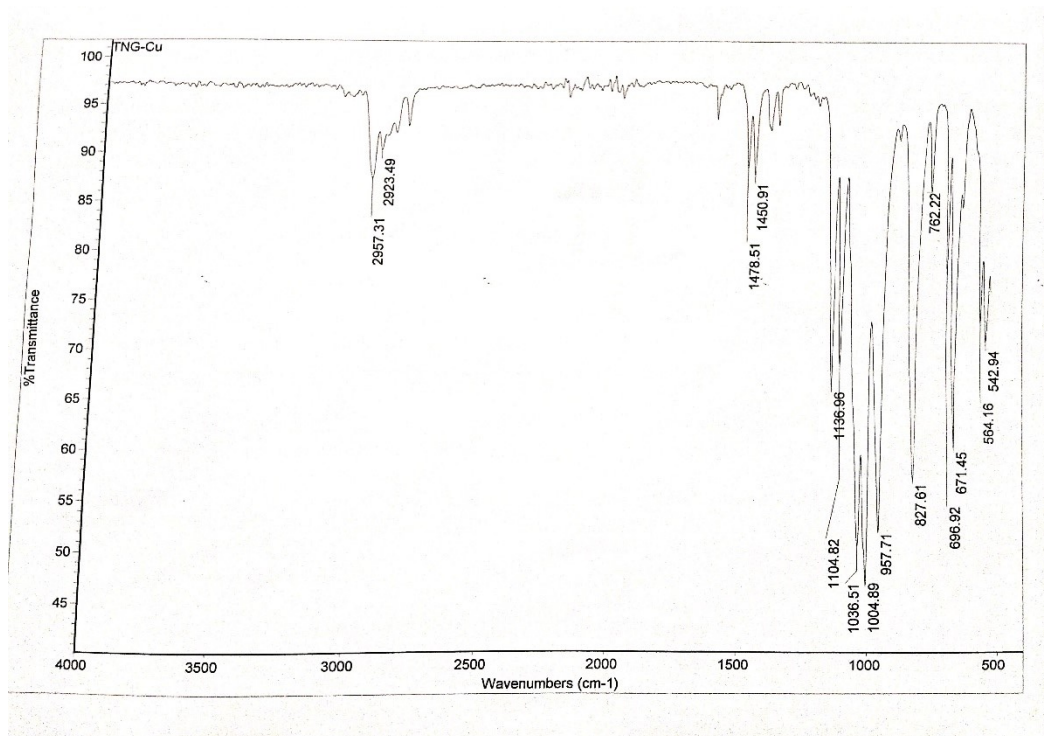


Figure S30: IR spectrum of compound 5.

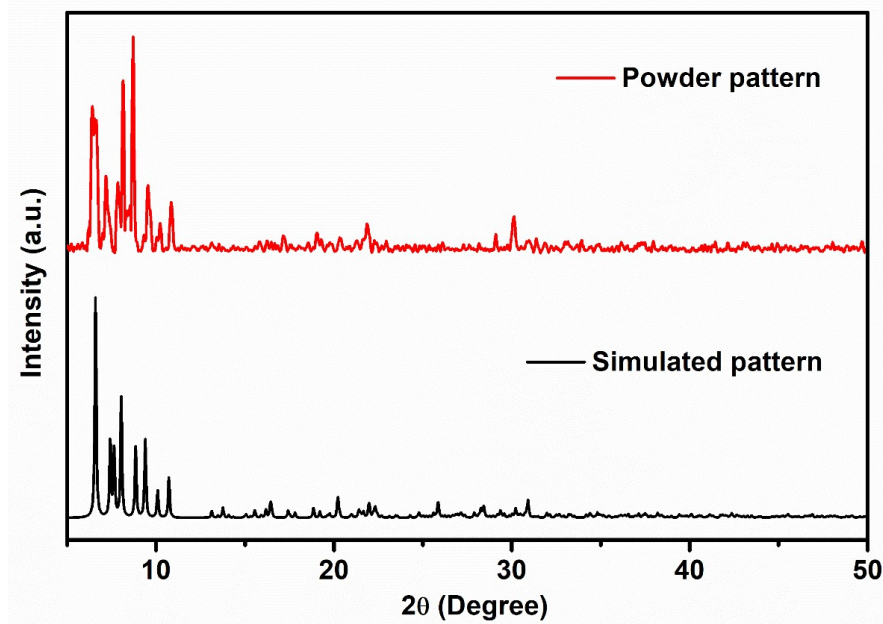


Figure S31: Powder X-ray diffraction pattern of a bulk sample of **5** compared to the simulated powder pattern extracted from single-crystal diffraction data.

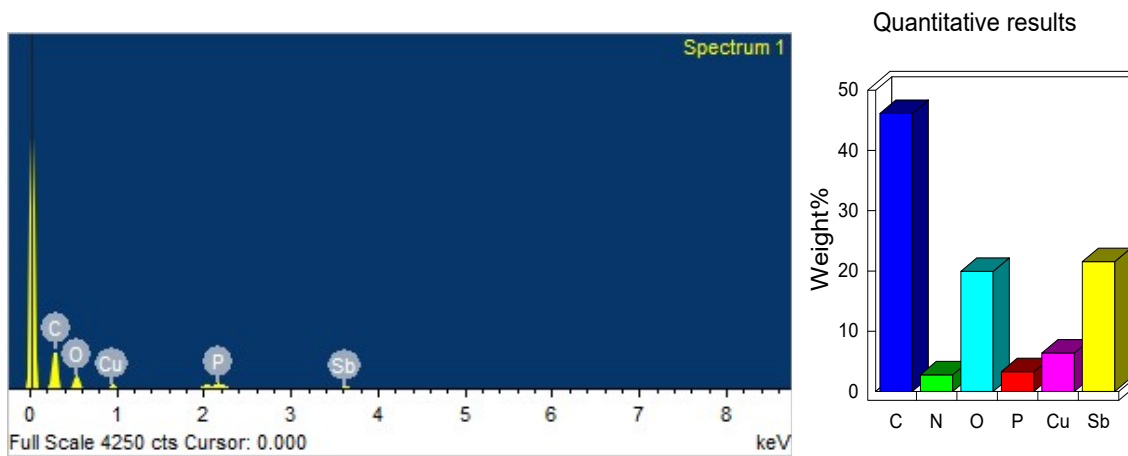


Figure S32: EDAX spectrum of compound **5**.

Table S11: EDAX analysis of **5**.

Element	Calculated weight%	Measured weight% from EDAX analysis
C	40.01	41.54
N	1.41	1.98
O	16.15	18.86
P	6.25	6.89
Cu	6.41	7.77
Sb	24.58	22.97
H	5.19	Not detected

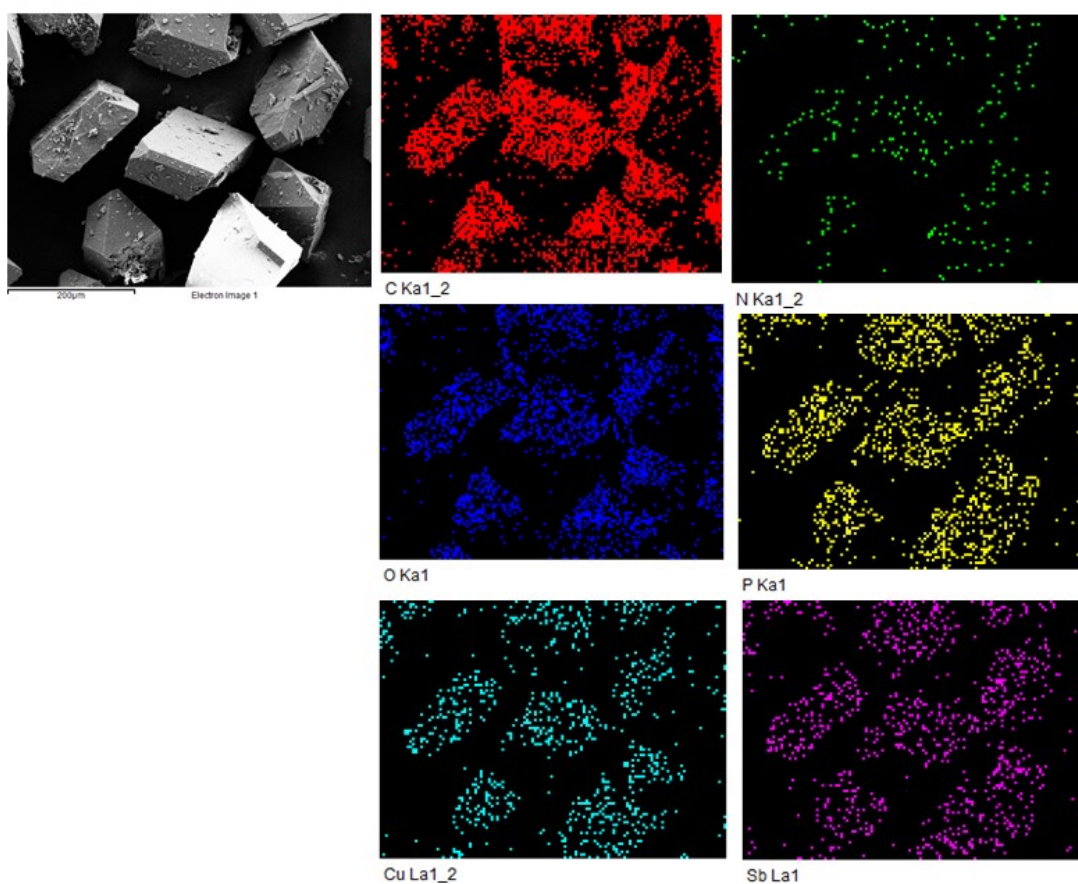


Figure S33: Elemental mapping of compound **5**.

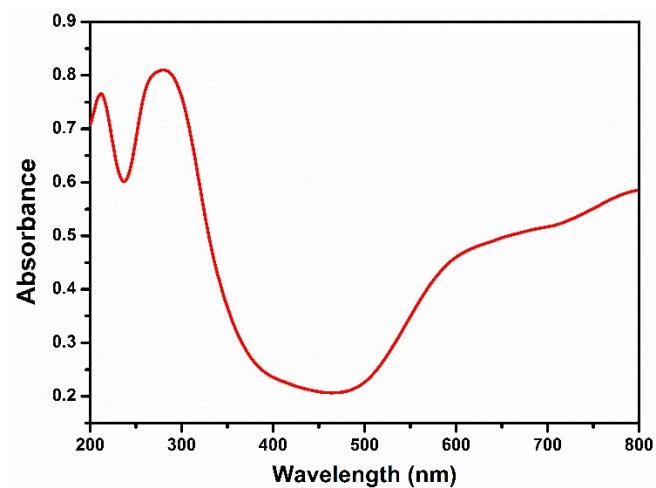
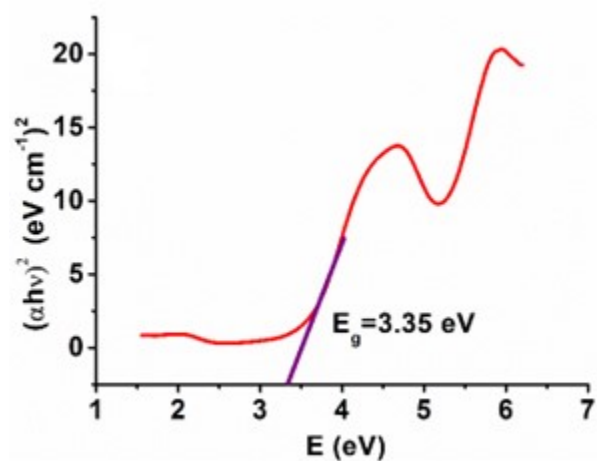
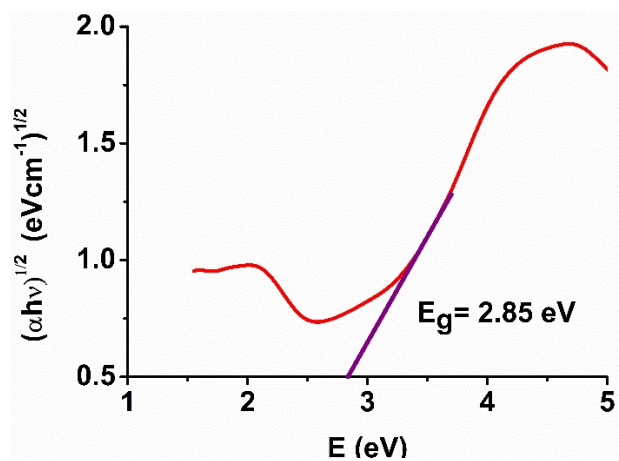


Figure S34: Solid-state UV-vis absorption spectrum of compound 5.



(a) Direct bandgap



(b) Indirect bandgap

Figure S35: Tauc plot of compound 5.

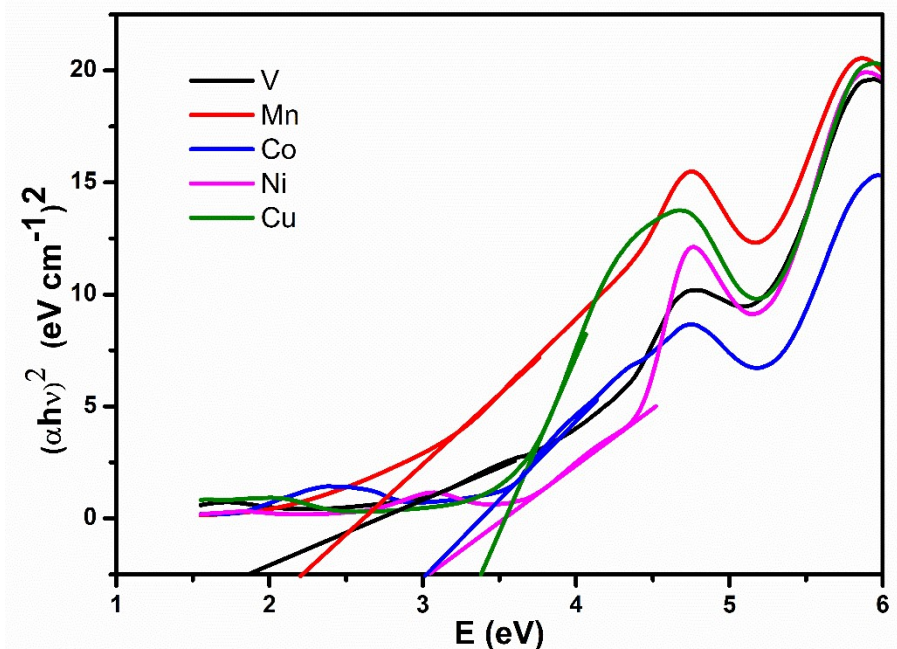


Figure S36: Tauc plots of compounds **1-5** (combined spectra). The direct bandgap (E_g) values are 1.83 eV for **1**, 2.25 eV for **2**, 3.02 eV for **3**, 3.10 eV for **4**, and 3.35 eV for **5**.

Table S12: Shape calculation of compounds **1-5**.

a) Summary of SHAPE analysis for compounds **1-5** around metal ions:

S.No	Geometry	CShM value for V1	CShM value for Mn1	CShM value for Co1	CShM value for Ni1	CShM value for Cu1
1	Hexagon (D6h)	34.395	34.710	34.342	33.692	34.010
2	Pentagonal pyramid (C5v)	23.097	21.247	21.876	22.046	22.472
3	Octahedron (Oh)	2.070	2.993	2.317	1.988	3.043
4	Trigonal prism (D3h)	12.110	11.258	11.348	11.167	10.521
5	Johnson pentagonal pyramid J2 (C5v)	27.043	25.050	25.80	25.994	26.240

b) Summary of SHAPE analysis for Sb1 in compound **1-5**:

S.No	Geometry	CShM value for Sb1 in 1	CShM value for Sb1 in 2	CShM value for Sb1 in 3	CShM value for Sb1 in 4	CShM value for Sb1 in 4
1	Hexagon (D6h)	31.737	32.292	32.201	31.871	31.871

2	Pentagonal pyramid (C5v)	26.785	27.564	27.295	28.048	28.048
3	Octahedron (Oh)	0.636	0.332	0.396	0.284	0.284
4	Trigonal prism (D3h)	14.617	15.231	15.066	14.675	14.675
5	Johnson pentagonal pyramid J2 (C5v)	30.427	31.346	31.031	31.571	31.571

Table S13: Bond Valence Sum (BVS) calculations:

Compound 1			Compound 2		
Metal Site	BVS	Assigned Oxidation state	Metal Site	BVS	Assigned Oxidation state
V1	3.98	4	Mn1	1.97	2

Compound 3			Compound 4		
Metal Site	BVS	Assigned Oxidation state	Metal Site	BVS	Assigned Oxidation state
Co1	2.03	2	Ni1	2.05	2

Compound 5		
Metal Site	BVS	Assigned Oxidation state
Cu1	1.99	2

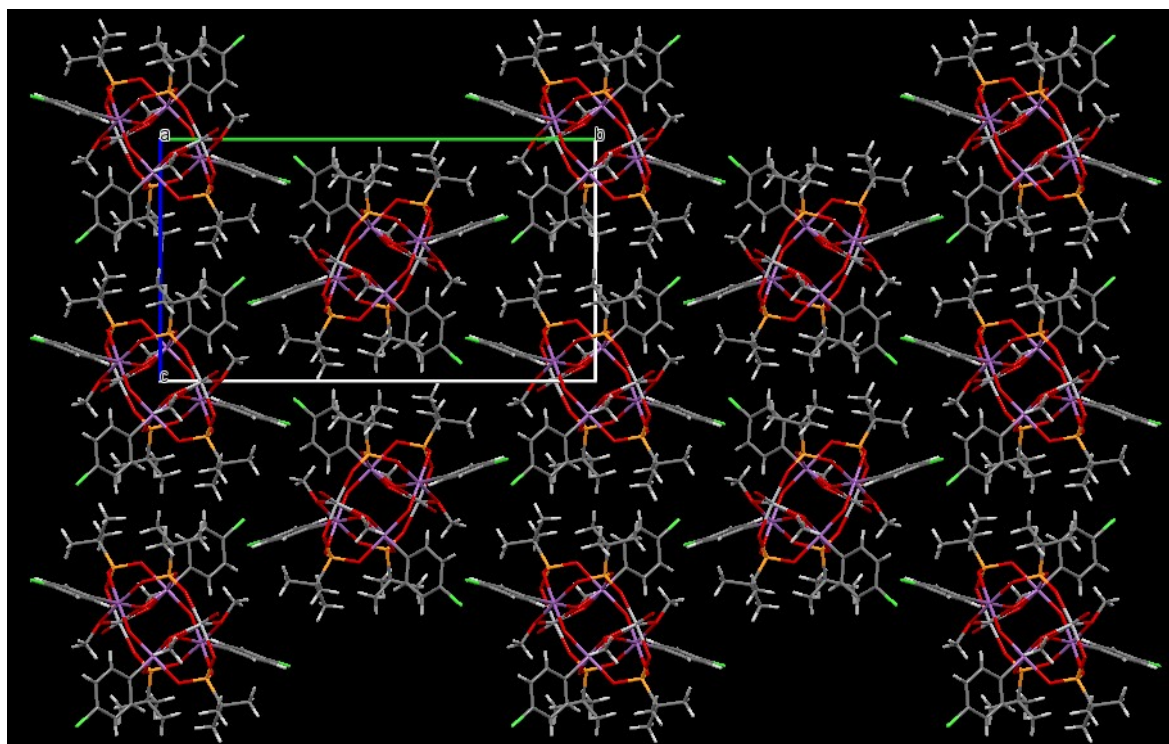


Figure S37: Packing diagram of compound **1** along a-axis

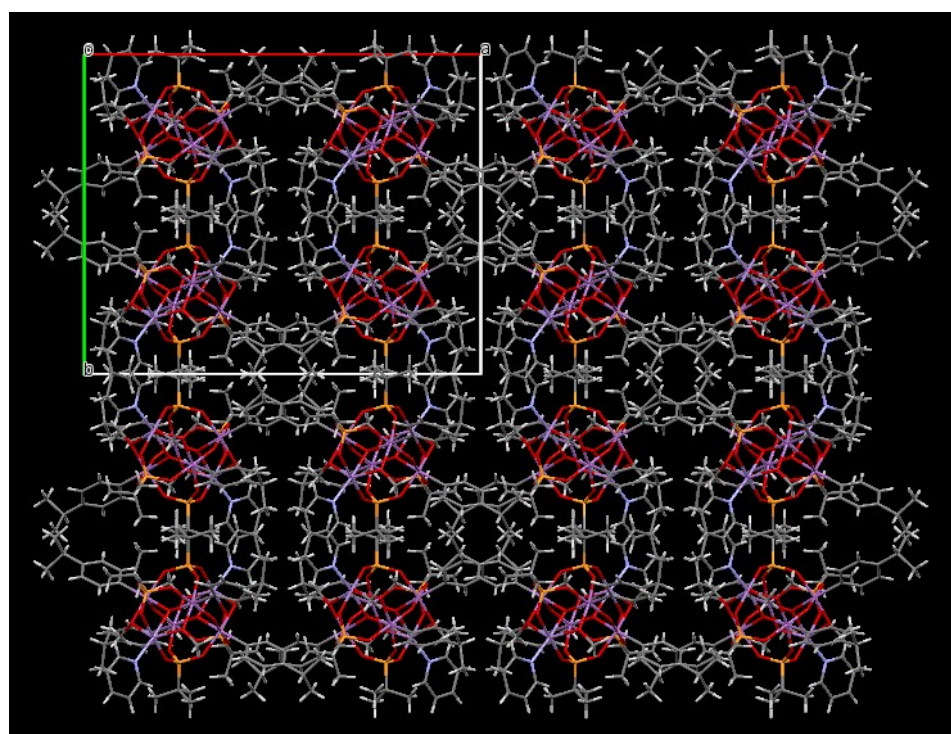


Figure 38: Packing diagram of compound **2** along the c-axis

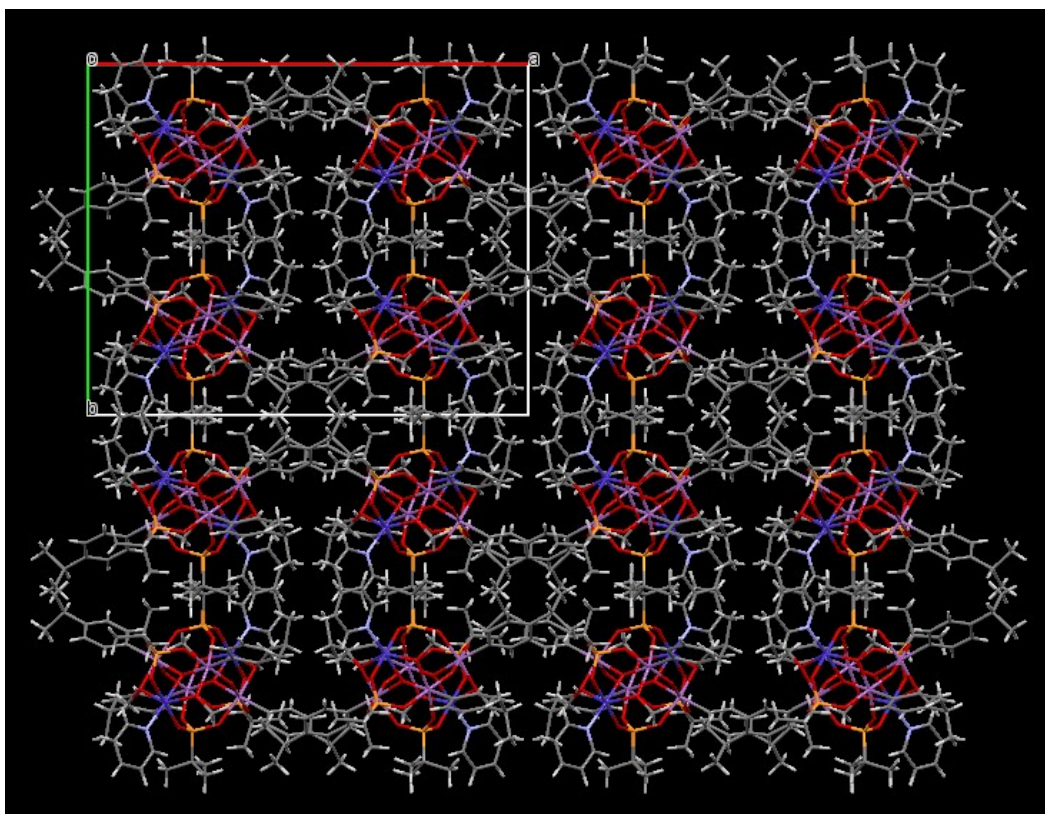


Figure S39: Packing diagram of compound **3** along the c-axis

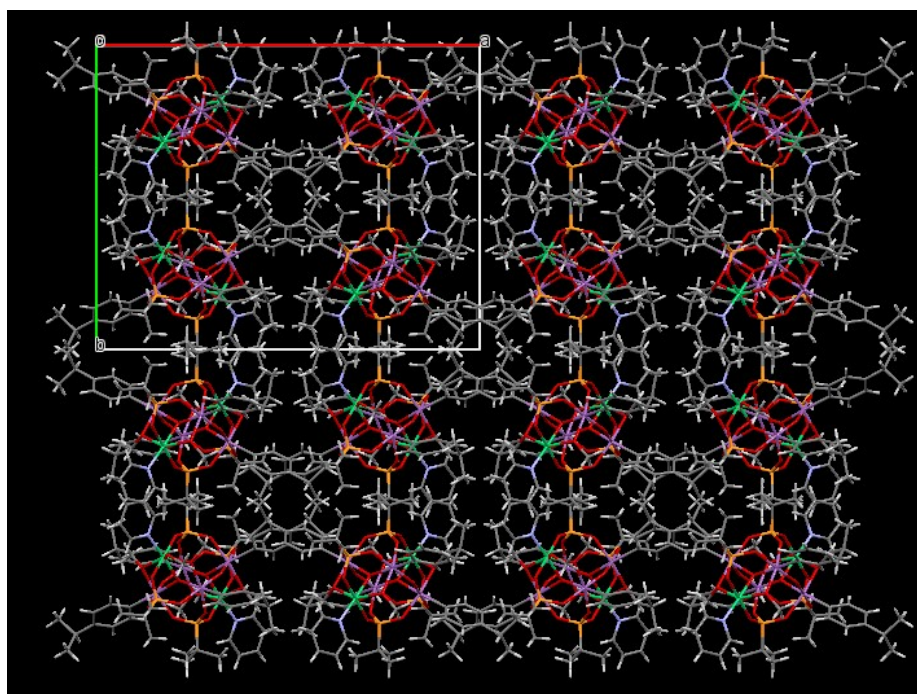


Figure S40: Packing diagram of compound **4** along the c-axis

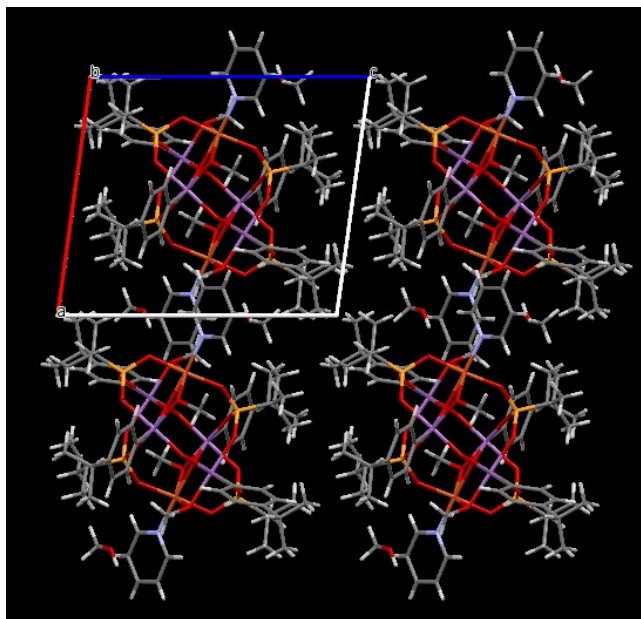


Figure S41: Packing diagram of compound **5** along b-axis

Solvent treatment of compound 3:

A solvent mask was calculated and 88 electrons were found in a volume of 740 Å³ in 3 voids. This is consistent with the presence of 1[CH₃OH] per formula unit, which account for 72.0 electrons.