Designing N-doped graphene-like supported highly dispersed bimetallic NiCoP NPs as an efficient electrocatalyst for water oxidation

Jiabo Wang^{a, *}, Yalin Fu^a, Peng Zhang^a, Jie Zhang^a, Xusen Ma^{a, c}, Jibo Zhang^{a, *}, Li Chen^{b, *}

^a Engineering Research Center of Jilin Provincial Higher Education University of Chemical Separation Technology, School of Petrochemical Technology, Jilin Institute of Chemical Technology, Jilin, 132022, P.R. China. E-mail: wangjb794@nenu.edu.cn

^b Information Materials and Intelligent Sensing Laboratory of Anhui Province, Institutes of Physical Science and Information Technology, Anhui University, Hefei 230601, PR China. E-mail: lichen@ahu.edu.cn

^c Wanhua Chemical Group Co., Ltd, Shandong, 264006, P.R. China

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Figure S1 The EDS spectrum of NiCoP-3@GL.



Figure S2 (a-d) FFT patterns and lattice spacings from the locations of (c) NiCoP and (d) Ni_{2.55}P phases in Figure 3.



Figure S3 The CV curve of (a) NiCoP-3@GL, (b) NiP@GL, (c) CoP@GL, (d) CC.



Figure S4 SEM image of NiCoP-3@GL sample after 5000 CV cycles.



Figure S5 TEM image of NiCoP-3@GL sample after 5000 CV cycles.





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Figure S7 The comparison of catalytic performance of different cobalt-based and nickel-based electrocatalysts.

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Table S1. NiCoP-x @GL, CoP@GL, NiP@GL, CC and RuO₂ overpotential and Tafel.

Samples	η ₁₀ (mV vs. RHE)	b (mV·dec⁻¹)
NiCoP-2@GL	355	68.66
NiCoP-3@GL	324	97.28
NiCoP-4@GL	353	157.59
CoP@GL	507	75.01
NiP@GL	473	76.67
CC	525	180.10
RuO ₂	292	52.19

Table S2 Results of C_{dl} , j_{ECSA} and j_{mass} of NiCoP-3@GL, CoP@GL, NiP@GL and CC in 1.0 M KOH solution.

Samples	<i>C_{dl}</i> (mF cm ⁻²)	ECSA (cm²)	η@j _{εCSA=0.002} (mV vs. RHE)	η@ j _{mass=2} (mV vs. RHE)
NiCoP-3@GL	18.19	452.75	340	370
CoP@GL	6.70	167.50	459	590
NiP@GL	8.23	205.75	444	528
CC	5.63	140.75	444	

Table S3 The electrochemical impedance fitting parameters for NiCoP-3@GL recorded in 1.0 M KOH solution.

R _s (Ω)	8.006
$Q_1 - Y_1 (\mu S \cdot S^{-1})$	9.312*10-4
<i>n</i> ₁ (0< <i>n</i> ₁ <1)	8.616*10-1
R ₁ (Ω)	1.00*10-2
W(Ω·cm²)	3.059*10 ⁻¹²
$Q_2 - Y_2 (\mu S \cdot S^{-1})$	2.152*10-3
n ₂ (0 <n<sub>2<1)</n<sub>	3.994*10-1
R ₂ (Ω)	3.139*10 ³

Table S4. Overview of OER performance of recently reported Co-based Ni-based catalysts.

Samples	η ₁₀ (mV vs. RHE)
Co/Co ₉ S ₈ /MnS	330
Co-ZIF _{1.5} /10CNF ₂	390
Ag-CeO ₂ -Co ₃ O ₄	340
Co ₂ P/CoP@NPGC	340
NiCoP-3@GL	324
CdN₄Co-gra	480
Co-Ni/NCNTs	337
Ni/ZIF8-900A	340