

Fabrication of metal-organic salts with heterogeneous conformations of ligand as dual-functional of urease and nitrification inhibitors

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Table S1 Crystallographic data for MOFs **1–4**.

MOS	1	2	3	4
Empirical formula	C ₃₀ H ₃₀ Cl ₄ CuF ₄ N ₂	C ₃₀ H ₃₀ Cl ₄ F ₄ N ₂ Zn	C ₃₀ H ₃₀ CdCl ₄ F ₄ N ₂	C ₃₀ H ₃₀ Cl ₄ CoF ₄ N ₂
Formula weight	699.90	701.73	748.76	695.29
Temperature (K)	298	298	298	298
Crystal size (mm)	0.14 × 0.13 × 0.10	0.14 × 0.12 × 0.10	0.14 × 0.12 × 0.09	0.14 × 0.12 × 0.10
Crystal system	Triclinic	Triclinic	Triclinic	Triclinic
Space group	P-1	P-1	P-1	P-1
Z	2	2	2	2
a (Å)	9.9316(5)	10.0080(5)	9.9952(6)	10.0083(5)
b (Å)	12.8000(6)	12.4928(6)	12.5695(8)	12.4895(6)
c (Å)	14.2699(7)	14.2889(7)	14.5331(9)	14.2962(7)
α (deg)	96.338(2)	93.920(2)	93.582(2)	93.778(2)
β (deg)	106.813(2)	107.938(2)	108.067(2)	108.137(2)
γ (deg)	112.103(2)	110.797(2)	111.162(2)	110.772(2)
V (Å ³)	1558.96(13)	1556.88(13)	1587.13(17)	1556.27(13)
D _x (mg cm ⁻³)	1.491	1.497	1.567	1.484
μ (mm ⁻¹)	1.091	1.179	1.071	0.942
F(000)	714	716	752	710
R _{int}	0.0255	0.027	0.028	0.028
No. of collected data (unique)	28615	26072	18084	39078
No. of data with [I>=2σ (I)]	7744	7564	5360	12424
No. of parameters	370	358	358	370
S	1.022	1.107	1.052	1.055
R _f /wR _f	0.0408/0.1064	0.0521/0.1349	0.0534/0.1320	0.0582/0.1570
All data R _f /wR _f	0.0552/0.1173	0.0663/0.1470	0.0586/0.1367	0.0880/0.1807

^a $R_1 = \sum |F_o| - |F_c| / \sum |F_o|$, ^b $wR_2 = \sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]^{1/2}$.

Table S2 The hydrogen bonds (\AA) and angles ($^\circ$) for MOS 1.

D–H \cdots A	D–H (\AA)	H \cdots A (\AA)	D \cdots A (\AA)	D–H \cdots A ($^\circ$)
N1–H1 \cdots Cl1(i)	0.980	2.154	3.079	156.61
N2–H2A \cdots Cl3(ii)	0.980	2.232	3.165	158.62
C23–H23B \cdots Cl4(iii)	0.987	2.652	3.563	156.51
C9–H9B \cdots Cl2(iv)	0.971	2.719	3.676	169.00
C8–H8A \cdots Cl2(v)	0.969	2.704	3.499	139.54
C16–H16 \cdots Cl4(vi)	0.930	2.924	3.598	130.44
C20–H20 \cdots Cl3(vii)	0.930	2.774	3.682	165.58
C13–H13 \cdots Cl3(viii)	0.931	2.708	3.571	154.54

Table S3 The hydrogen bonds (\AA) and angles ($^\circ$) for MOS 2.

D–H \cdots A	D–H (\AA)	H \cdots A (\AA)	D \cdots A (\AA)	D–H \cdots A ($^\circ$)
N1–H1 \cdots Cl4(i)	0.980	2.118	3.076	165.28
N2–H2 \cdots Cl2(ii)	0.980	2.252	3.159	152.82
C15–H15A \cdots Cl3(iii)	0.970	2.775	3.601	143.53
C16–H16B \cdots Cl3(iv)	0.969	2.850	3.606	135.44
C30–H30B \cdots Cl1(v)	0.968	2.823	3.725	155.28
C1–H1B \cdots Cl3(vi)	0.970	2.925	3.800	150.61
C19–H19 \cdots Cl2(vii)	0.930	2.879	3.758	158.14
C8–H8B \cdots Cl3(viii)	0.970	2.858	3.822	172.78

Table S4 The hydrogen bonds (\AA) and angles ($^\circ$) for MOS 3.

D–H \cdots A	D–H (\AA)	H \cdots A (\AA)	D \cdots A (\AA)	D–H \cdots A ($^\circ$)
N1–H1 \cdots Cl3(i)	0.981	2.253	3.164	154.13
N2–H2 \cdots Cl1(ii)	0.980	2.114	3.077	167.18
C23–H23A \cdots Cl2(iii)	0.970	2.898	3.726	143.93
C30–H30B \cdots Cl2(iv)	0.970	2.784	3.648	148.80
C11–H11 \cdots Cl3(v)	0.930	2.833	3.733	163.75
C22–H22A \cdots Cl2(vi)	0.970	2.884	3.851	175.30
C15–H15A \cdots Cl2(vii)	0.970	2.905	3.787	151.72
C1–H1B \cdots Cl4(viii)	0.969	2.841	3.750	156.68

Table S5 The hydrogen bonds (\AA) and angles ($^\circ$) for MOS 4.

D–H \cdots A	D–H (\AA)	H \cdots A (\AA)	D \cdots A (\AA)	D–H \cdots A ($^\circ$)
N1–H1 \cdots Cl1(i)	0.980	2.251	3.157	153.25
N2–H2A \cdots Cl4(ii)	0.980	2.115	3.072	165.14
C7–H7A \cdots Cl2(iii)	0.970	2.932	3.803	149.91
C9–H9A \cdots Cl2(iv)	0.970	2.848	3.607	135.72
C24–H24A \cdots Cl2(v)	0.970	2.832	3.663	144.32
C23–H23B \cdots Cl2(vi)	0.970	2.769	3.600	144.07
C8–H8A \cdots Cl3(vii)	0.970	2.825	3.729	155.36
C12–H12 \cdots Cl1(viii)	0.930	2.871	3.749	158.05

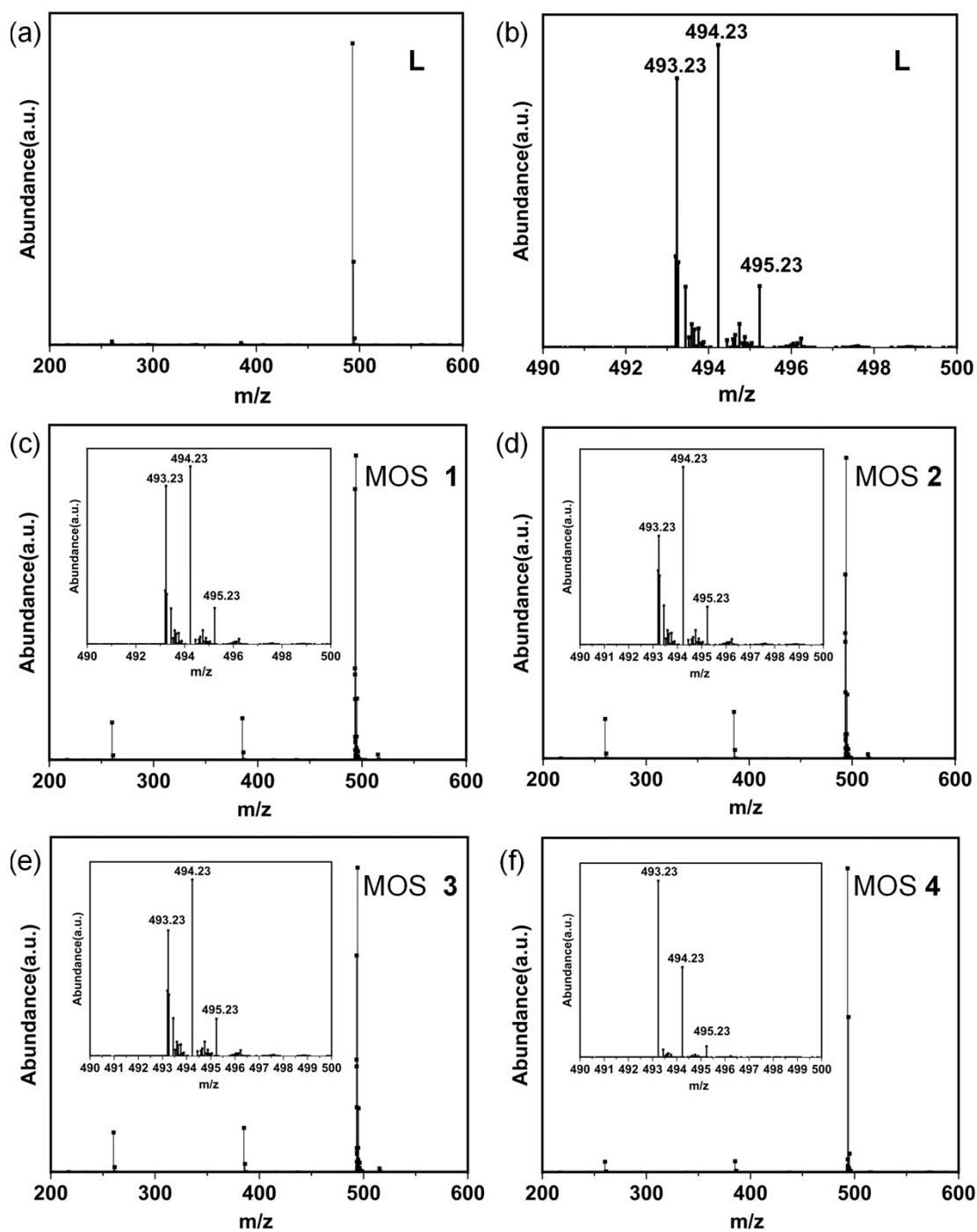


Fig. S1 The mass spectra of the L (a and b) and MOS 1 (c), 2 (d), 3 (e) and 4 (f).

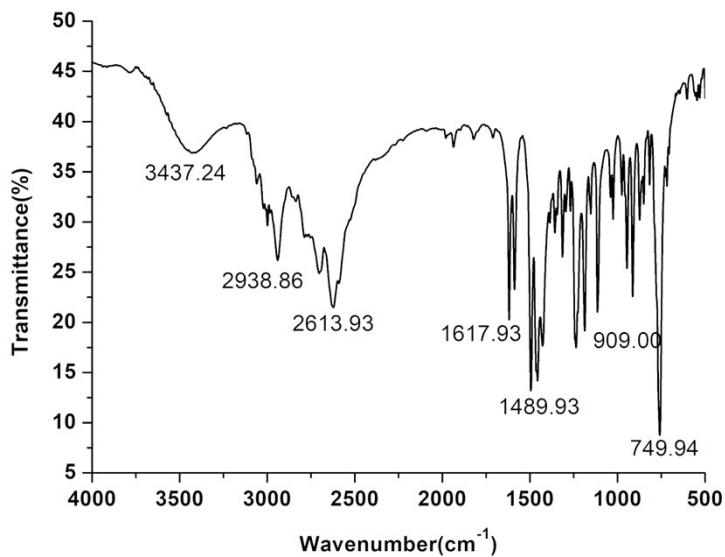


Fig. S2 IR spectrum of MOS 1.

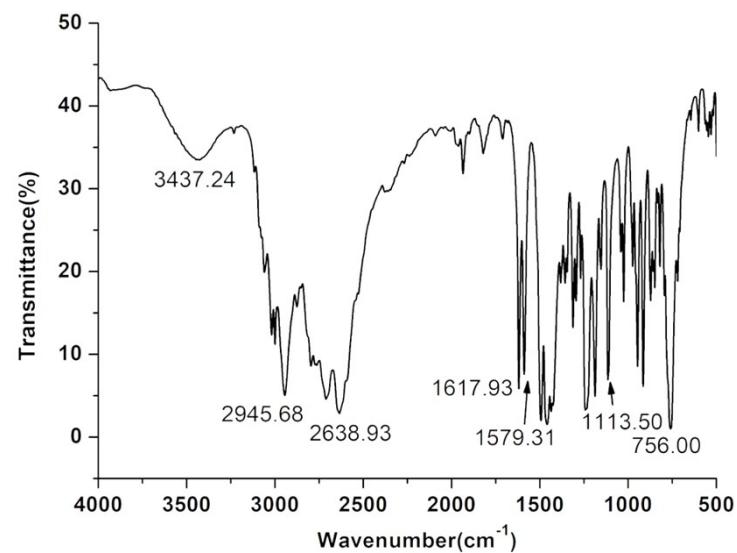


Fig. S3 IR spectrum of MOS 2.

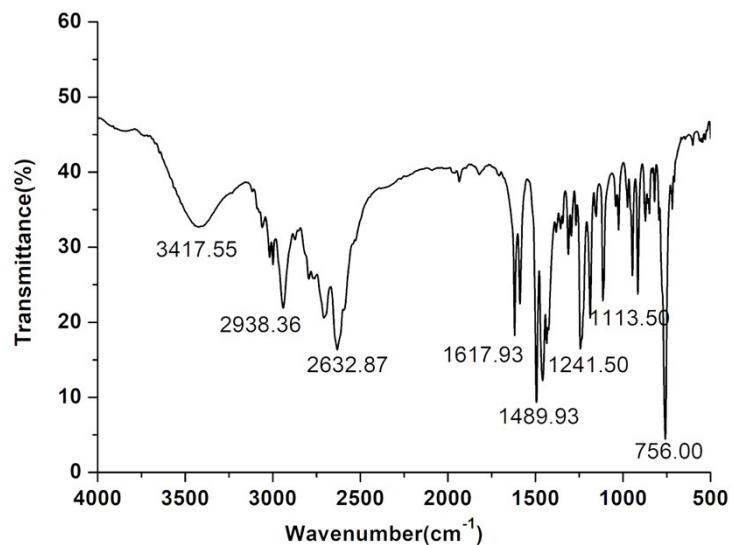


Fig. S4 IR spectrum of MOS 3

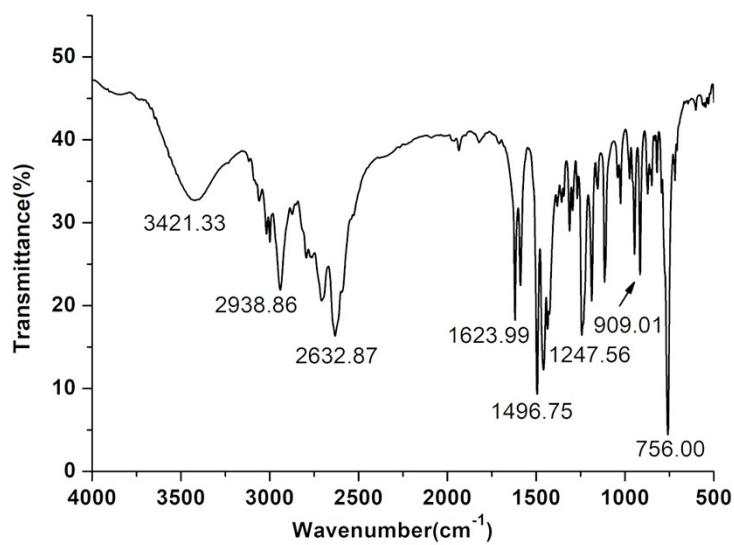


Fig. S5 IR spectrum of MOS 4.

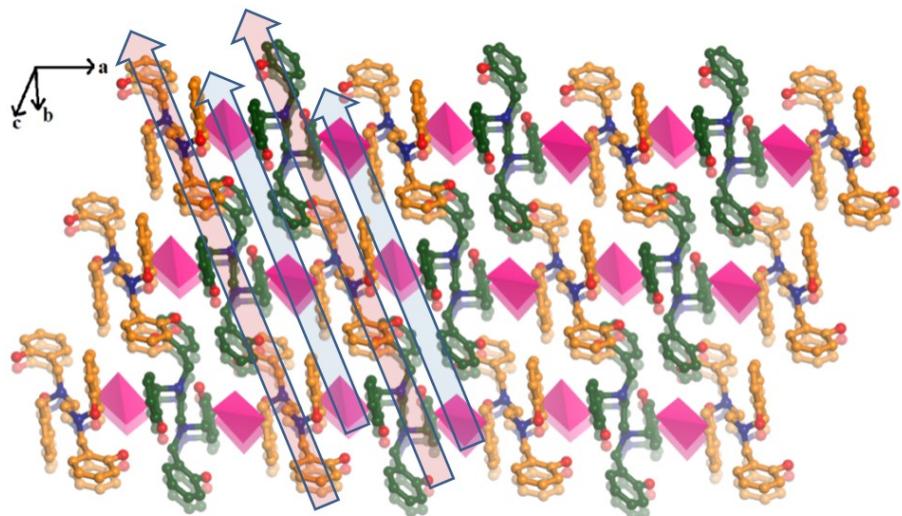


Fig. S6 3D packing of MOS 1.

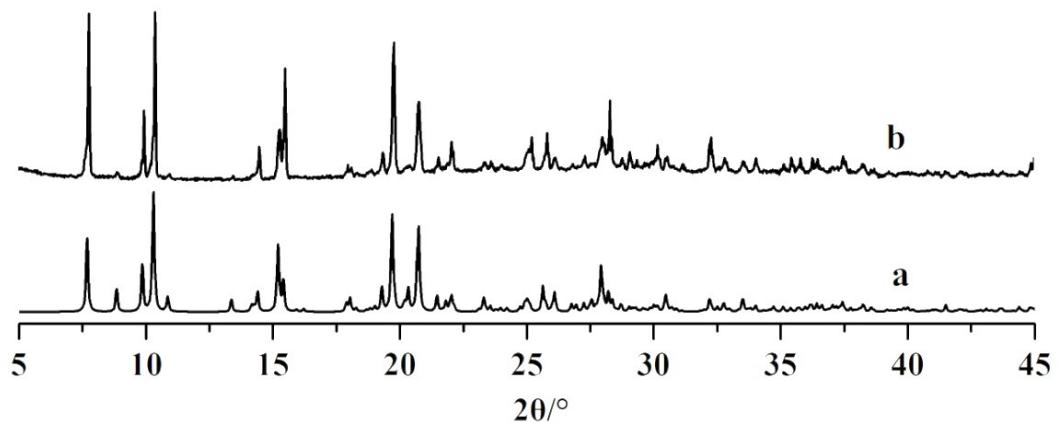


Fig. S7 Experimental (b) and simulated (a) PXRD patterns of MOS 1.

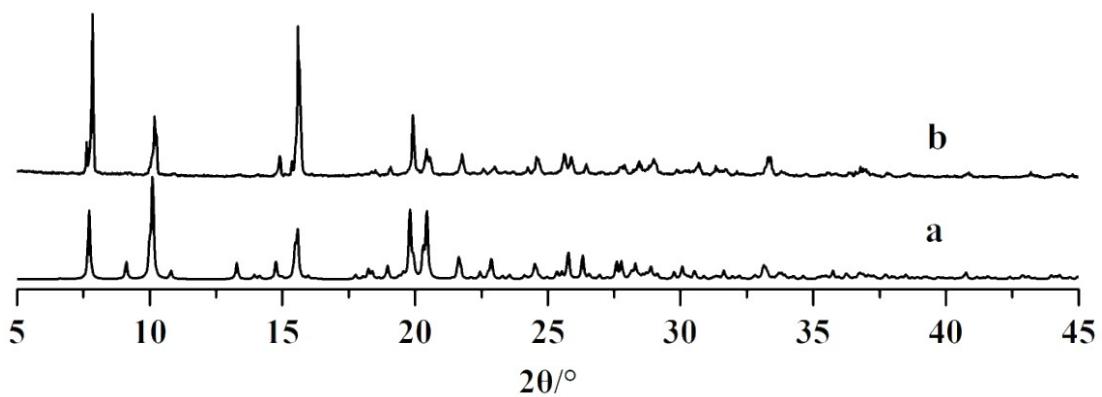


Fig. S8 Experimental (b) and simulated (a) PXRD patterns of MOS 2.

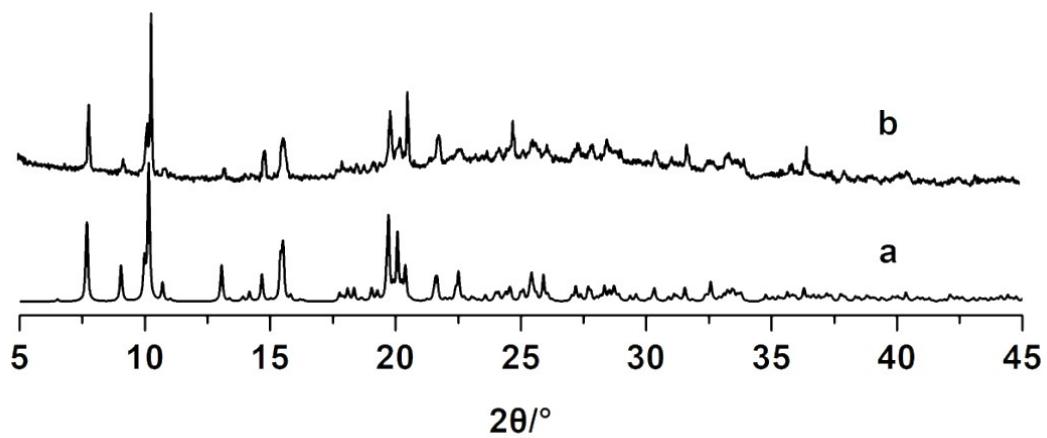


Fig. S9 Experimental (b) and simulated (a) PXRD patterns of MOS 3.

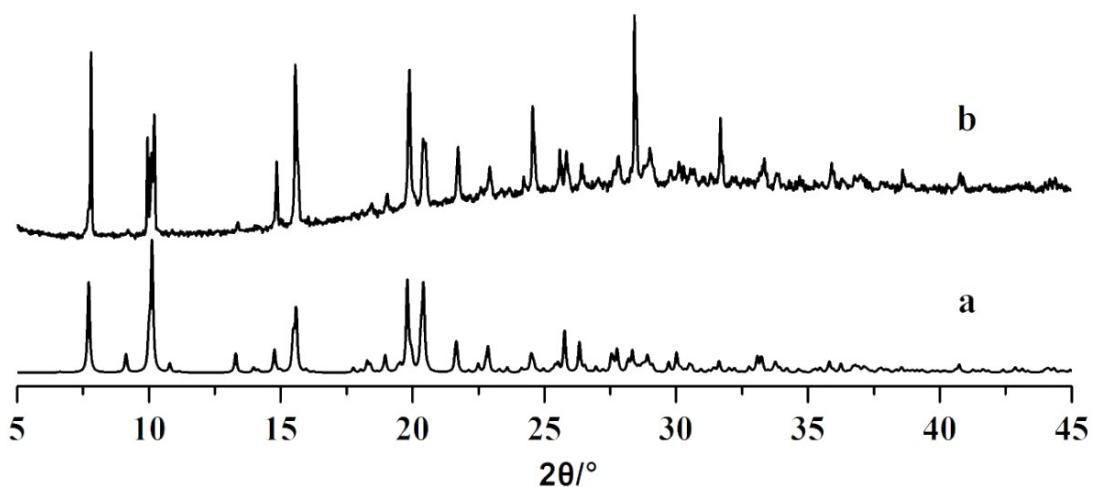


Fig. S10 Experimental (b) and simulated (a) PXRD patterns of MOS 4.

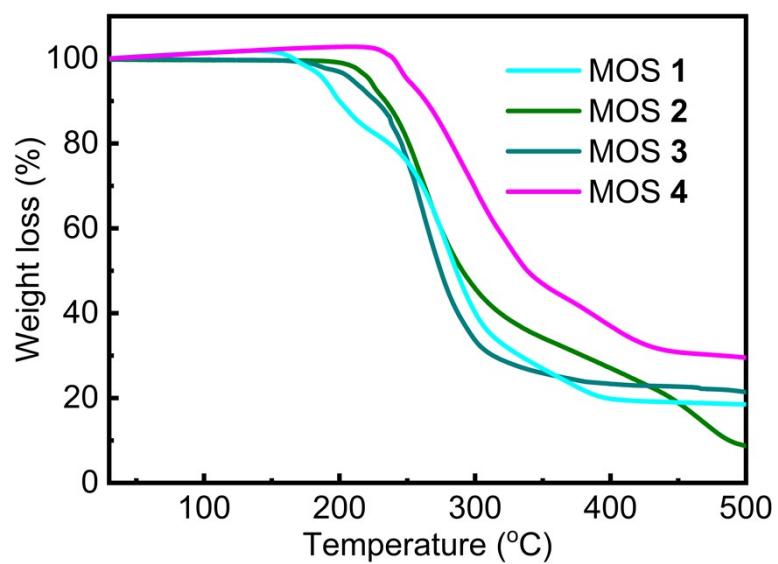


Fig. S11 TGA curves of MOSs 1–4.

Table S6 Nitrification inhibitory of the tested materials.

Material	Day	Nitrification inhibition rate (%)
DCD	14	68.30%
L	14	8.12%
MOS 1	14	65.29%
MOS 2	14	48.68%
MOS 3	14	26.12%
MOS 4	14	20.76%
DCD	28	73.18%
L	28	34.03%
MOS 1	28	70.75%
MOS 2	28	13.29%
MOS 3	28	—
MOS 4	28	6.31%
DCD	37	74.80%
L	37	1.63%
MOS 1	37	36.52%
MOS 2	37	8.54%
MOS 3	37	—
MOS 4	37	4.91%
DCD	56	61.69%
L	56	—
MOS 1	56	—
MOS 2	56	11.01%
MOS 3	56	—
MOS 4	56	—
DCD	87	14.42%
L	87	—
MOS 1	87	11.49%
MOS 2	87	—
MOS 3	87	8.89%
MOS 4	87	—