## Dual-mode sensing of hydrogen peroxide on self-assembled Ag

## nanoparticles anchored on polydopamine wrapping 2D Cu-MOF

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Fig. S1 EDS spectrum of Cu-MOF@PDA-Ag composite.



Fig. S2 (A) XRD patterns and (B)  $N_2$  adsorption-desorption isotherms of Cu-MOF and Cu-MOF@PDA-Ag.



Fig. S3 The scan rate study of Cu-MOF, Cu-MOF@PDA and Cu-MOF@PDA-Ag in 0.1 M KCl solution containing 5 mM K<sub>3</sub>[Fe(CN)<sub>6</sub>].



Fig. S4 The effect of the applied potential on amperometric responses of Cu-MOF@PDA-Ag/ITO toward the successive addition of 1 mM H<sub>2</sub>O<sub>2</sub>.

<b>Table S1.</b> The comparison of seven $H_2O_2$ sensors.				
Sensors	Sensitivity	<b>Detection limit</b>	Linear	Ref.
	(µA mM <sup>-1</sup> cm <sup>-2</sup> )	(µM)	range (mM)	
FeOOH/Ag/PGE	8.07	22.8	0.03–15	1
FeOOH@/PDA-Ag/GCE	11.8	2.5	0.0075 - 18.8	2
Cu <sub>2</sub> O/GNs/GCE	-	20.8	0.3-7.8	3
Cu <sub>2</sub> O-rGO <sub>pa</sub> /GCE	20.7	21.7	0.03-12.80	4
Cu-MOF@PDA-Ag/ITO	103.7	2.3	0.001-35	This
				work



Fig. S5 Amperometric responses of Cu-MOF@PDA-Ag/ITO toward the addition of 2 mM H<sub>2</sub>O<sub>2</sub> and interferences of UA, AA, DA, KCl, NaCl and Glu.



Fig. S6 The reproducibility of Cu-MOF@PDA-Ag/ITO.



Fig. S7 (A) TEM image and (B) XRD pattern of Cu-MOF@PDA-Ag after electrocatalysis.



**Fig. S8** The influences of (A) Cu-MOF@PDA-Ag concentrations, (B) pH values and (C) temperatures on the absorbance of oxTMB at 652 nm.



Fig. S9 The Selectivity of Cu-MOF@PDA-Ag for 0.5 mM  $H_2O_2$  in comparison with 5 mM of UA, DA, AA, Glu, K<sup>+</sup> and Na<sup>+</sup>, respectively.

## Reference

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