

## Electronic Supplementary Information for

# Models of the putative antimony(V)-diolate motifs in antileishmanial pentavalent antimonial drugs

*Brent Lindquist-Kleissler and Timothy C. Johnstone\**

Department of Chemistry and Biochemistry, University of California Santa Cruz, Santa Cruz,  
California 95064, United States.

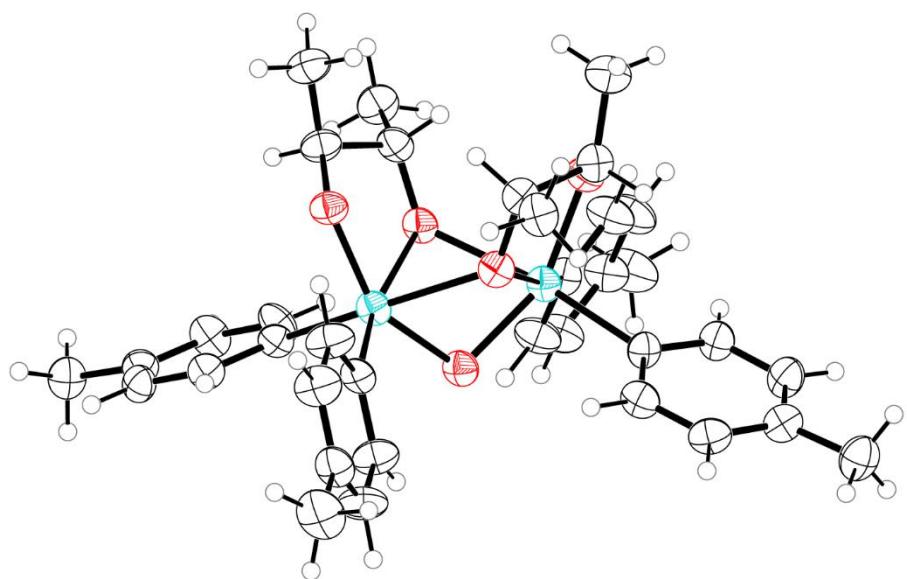
\* Correspondence to: [johnstone@ucsc.edu](mailto:johnstone@ucsc.edu)

## CONTENTS

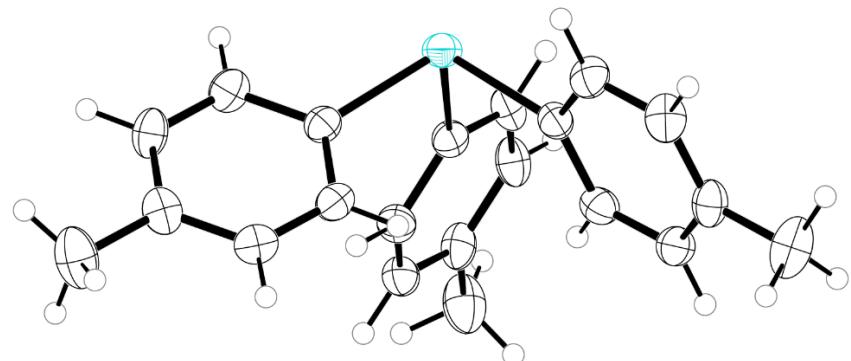
	Page
Table S1 – Refinement details for crystal structures	S3
Figure S1 – Crystal structure of hydrolysis decomposition product	S4
Figure S2 – Crystal structure of Sb(tol) <sub>3</sub>	S4
Figure S3 – Overlays of crystal structures of <b>1-6</b>	S5
Figure S4 – Stacked <sup>1</sup> H NMR spectra of free butanediols, <b>2</b> , and <b>4</b>	S5
Figure S5 – Stacked <sup>1</sup> H NMR spectra of free ( <i>S</i> )-1,2-propanediol and <b>6</b>	S6
Figure S6 – NMR spectra of <b>1</b>	S7
Figure S7 – NMR spectra of <b>2</b>	S8
Figure S8 – NMR spectra of <b>3</b>	S9
Figure S9 – NMR spectra of <b>4</b>	S10
Figure S10 – NMR spectra of <b>5</b>	S11
Figure S11 – NMR spectra of <b>6</b>	S12

**Table S1.** Refinement Details for the Crystal Structures of  $\text{Sb}(\text{tol})_3$  and  $[(\text{tol})_3\text{Sb}(\mu\text{-C}_4\text{H}_8\text{O}_2)]_2\text{O}$ .

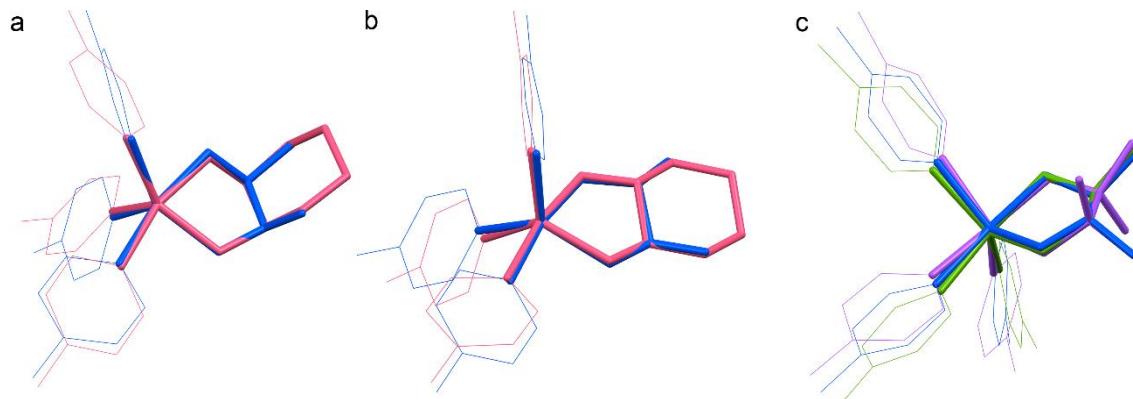
	$\text{Sb}(\text{tol})_3$	$[(\text{tol})_3\text{Sb}(\mu\text{-C}_4\text{H}_8\text{O}_2)]_2\text{O}$
Formula	$\text{C}_{21}\text{H}_{21}\text{Sb}$	$\text{C}_{36}\text{H}_{44}\text{O}_5\text{Sb}_2$
FW	395.13	800.21
T (K)	101.1(6)	102(2)
$\lambda$ (Å)	1.54184	1.54184
Crystal System	Trigonal	Orthorhombic
Space group	$R\bar{3}$	$P2_12_12_1$
$a$ (Å)	12.6758(3)	12.17550(10)
$b$ (Å)		13.5809(2)
$c$ (Å)	19.8058(4)	20.6690(3)
Volume (Å <sup>3</sup> )	2755.96(14)	3417.71(8)
Z	6	4
$\rho_{\text{calc}}$ (Mg/m <sup>3</sup> )	1.428	1.555
Size (mm <sup>3</sup> )	0.21×0.14×0.04	0.12×0.06×0.05
$\theta$ range (°)	4.605-67.036	3.895-67.684
Total data	6111	51012
Unique data	1099	7095
Parameters	68	397
Completeness	100.0%	100.0%
$R_{\text{int}}$	4.87%	5.45%
$R_1$ ( $I > 2\sigma$ )	2.11%	3.25%
$R_1$ (all data)	2.16%	3.39%
$wR_2$ ( $I > 2\sigma$ )	5.41%	8.52%
$wR_2$ (all data)	5.43%	8.64%
$S$	1.098	1.078
Flack $x$	—	-0.015(9)



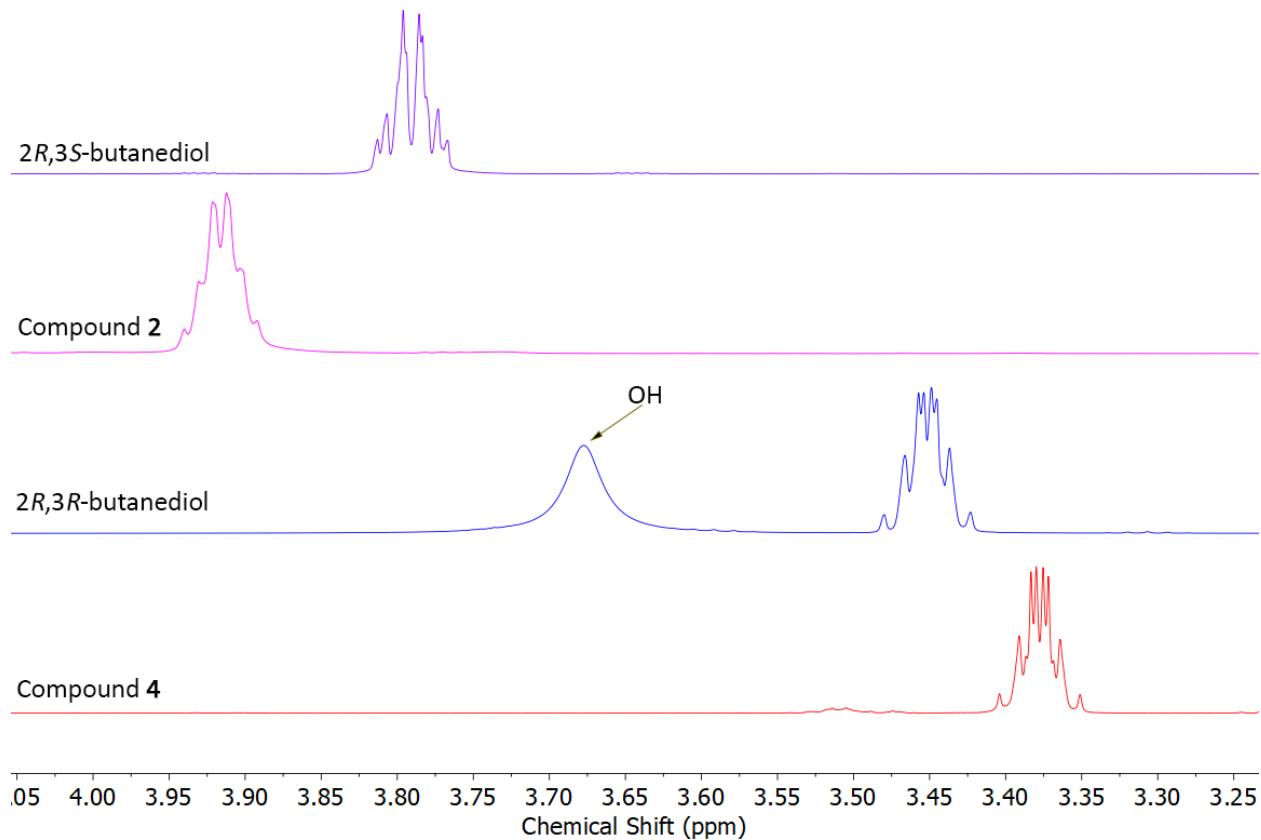
**Figure S1.** Thermal ellipsoid plot (50% probability, H atoms as spheres of arbitrary radius) of a hydrolysis decomposition product observed to form during the synthesis of butanediolate complexes. Color code: Sb teal, O red, C black, H white.



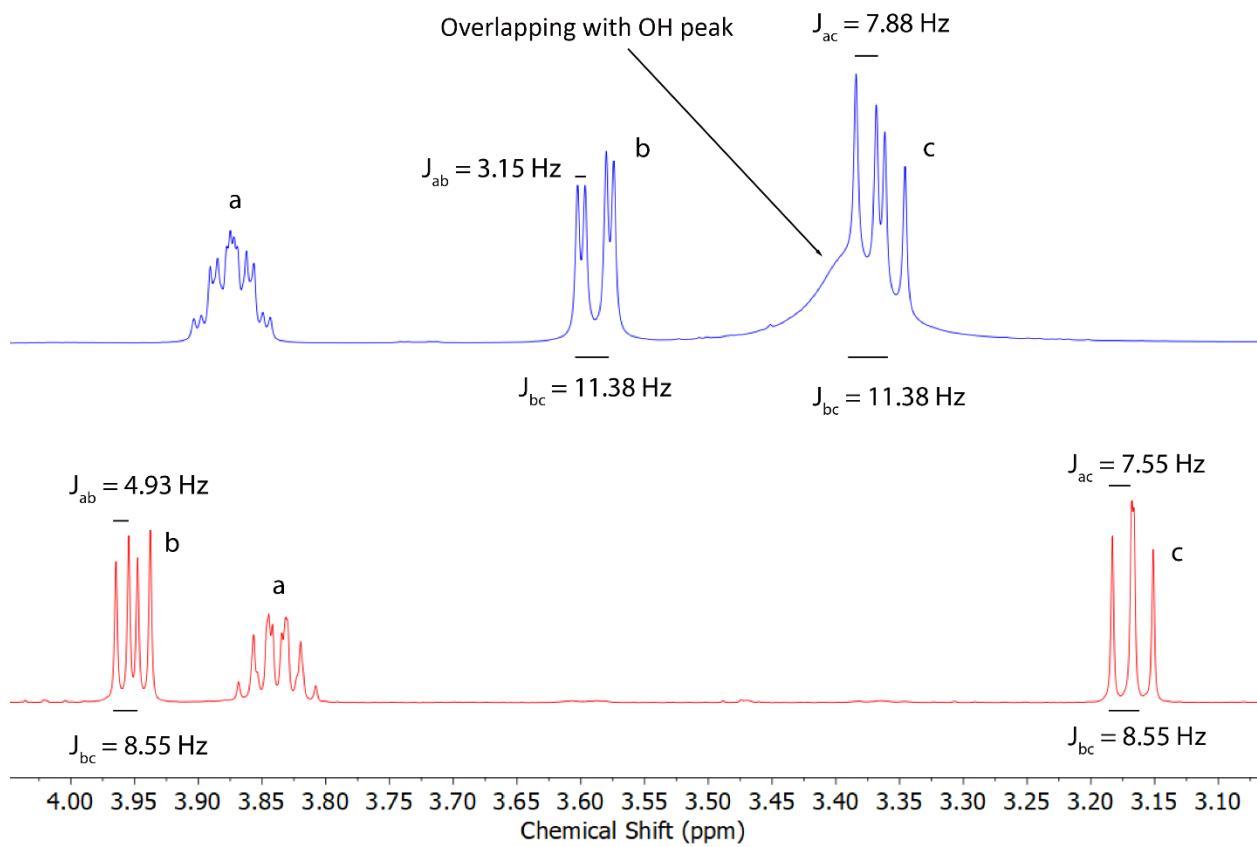
**Figure S2.** Thermal ellipsoid plot (50% probability, H atoms as spheres of arbitrary radius) of Sb(tol)<sub>3</sub>. Color code: Sb teal, C black, H white.



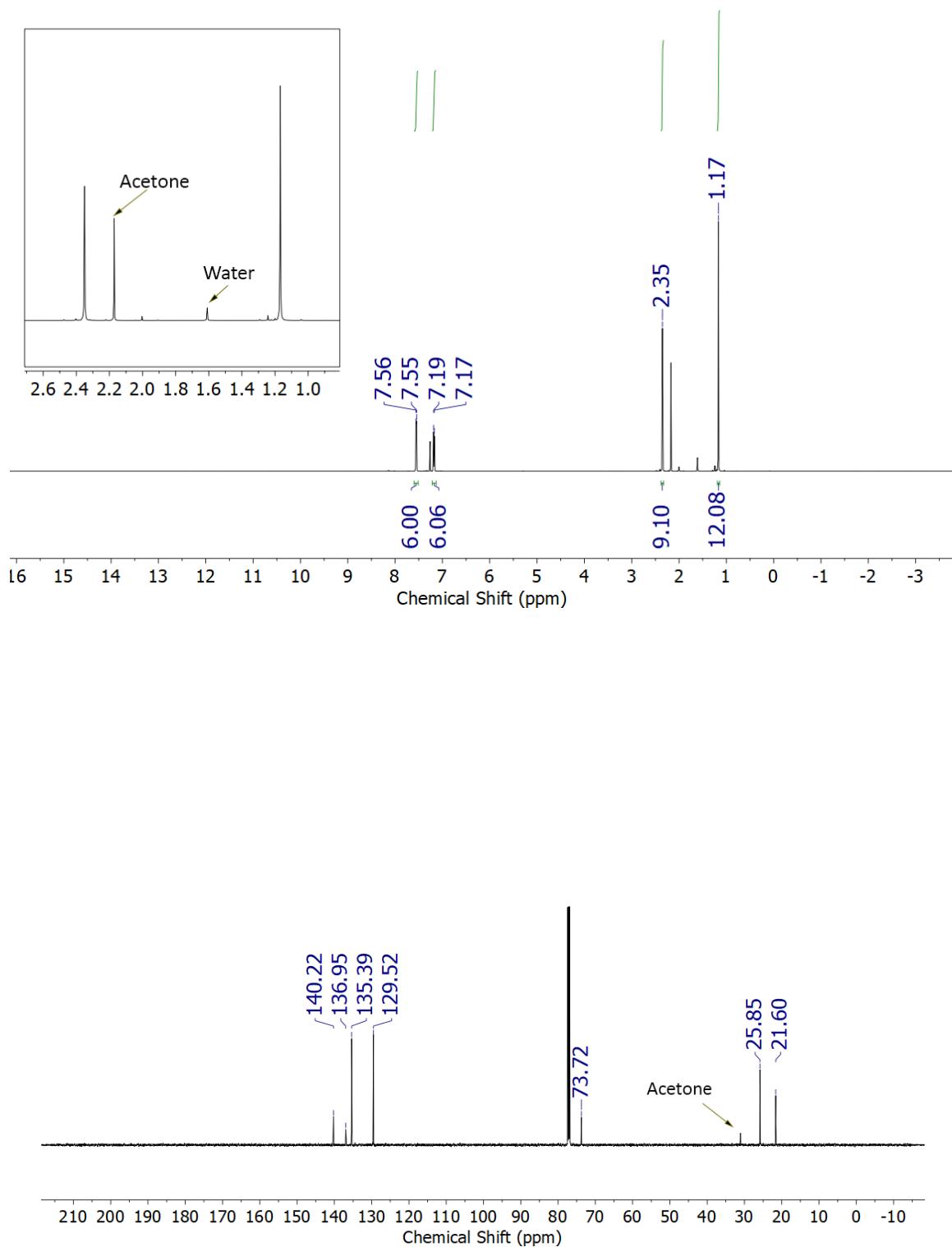
**Figure S3.** Overlays of: a) **2** (blue) and **3** (pink). b) **4** (blue) and **5** (pink). c) **1** (purple), **4** (blue), and **6** (green).



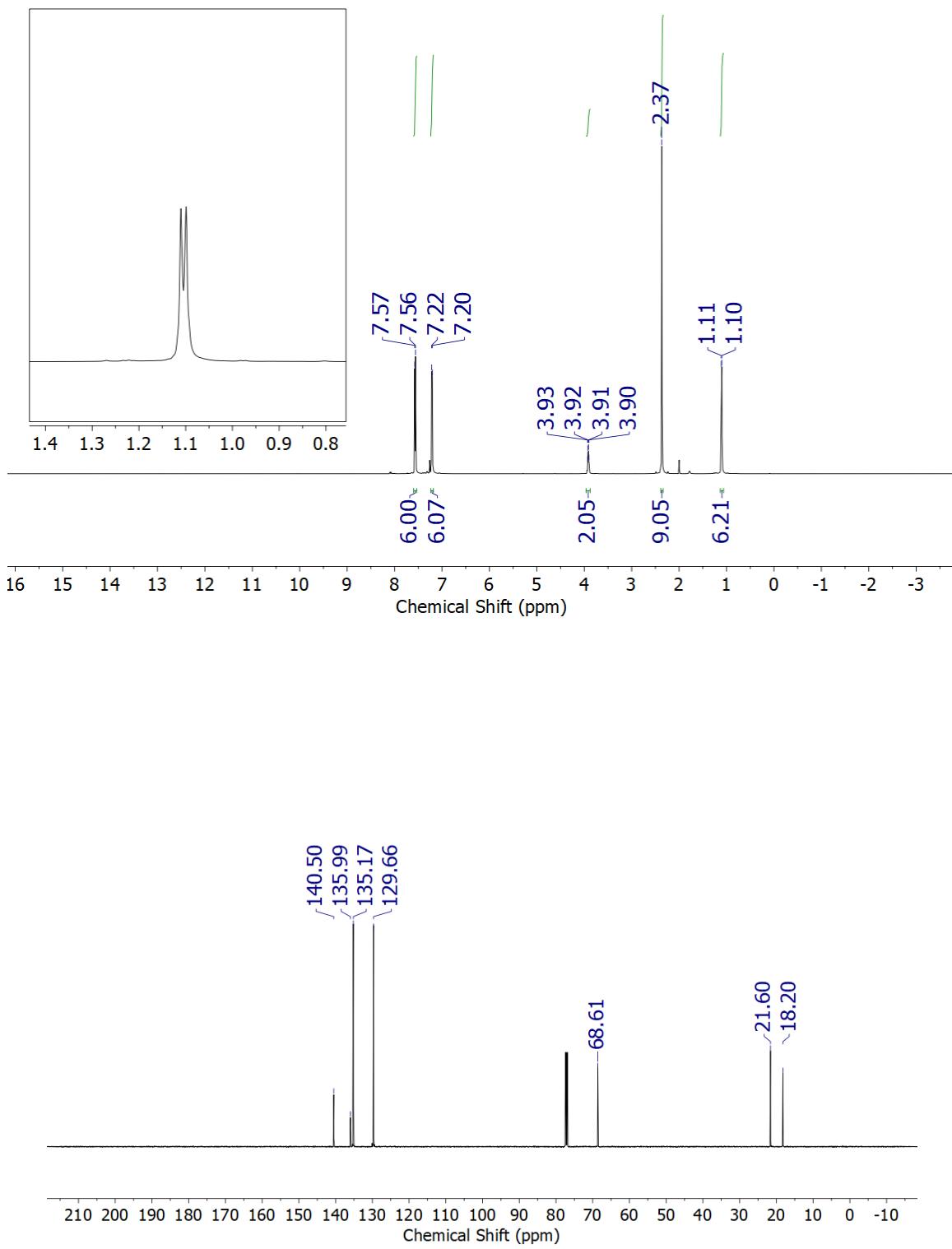
**Figure S4.** Stacked  $^1\text{H}$  NMR spectra showing shift in backbone CH resonances. Color code: unbound 2R,3S-butanediol purple, compound **2** pink, unbound 2R,3R-butanediol blue, compound **4** red.



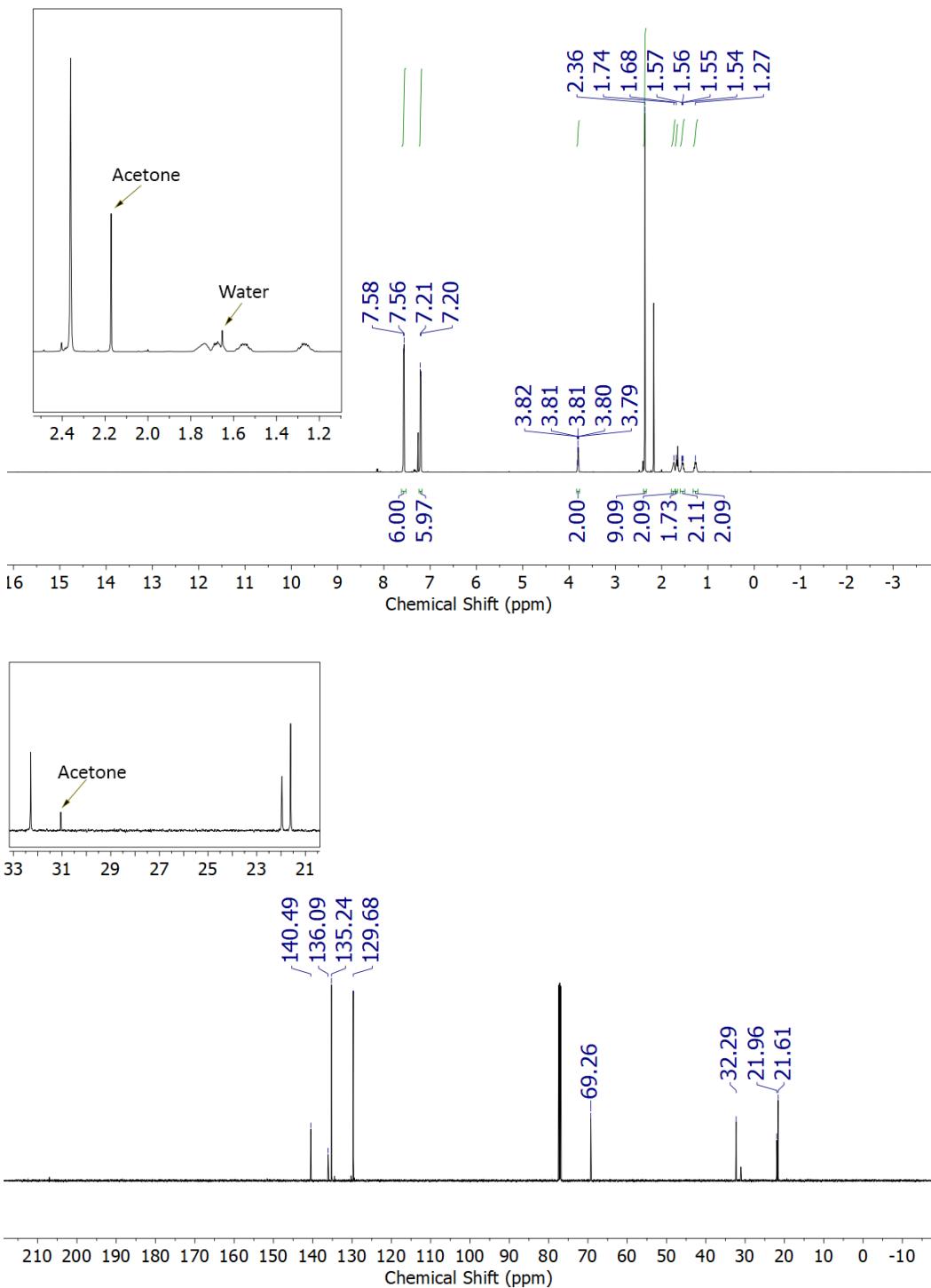
**Figure S5.** *Top (blue):*  $^1\text{H}$  NMR spectrum of free (S)-1,2-propanediol. Peaks are arbitrarily labeled based on the labelling scheme used for the Newman projections. *Middle (red):*  $^1\text{H}$  NMR spectrum of **6**. *Bottom:* Newman projections displaying conformational change of (S)-1,2-propanediol upon binding to Sb.



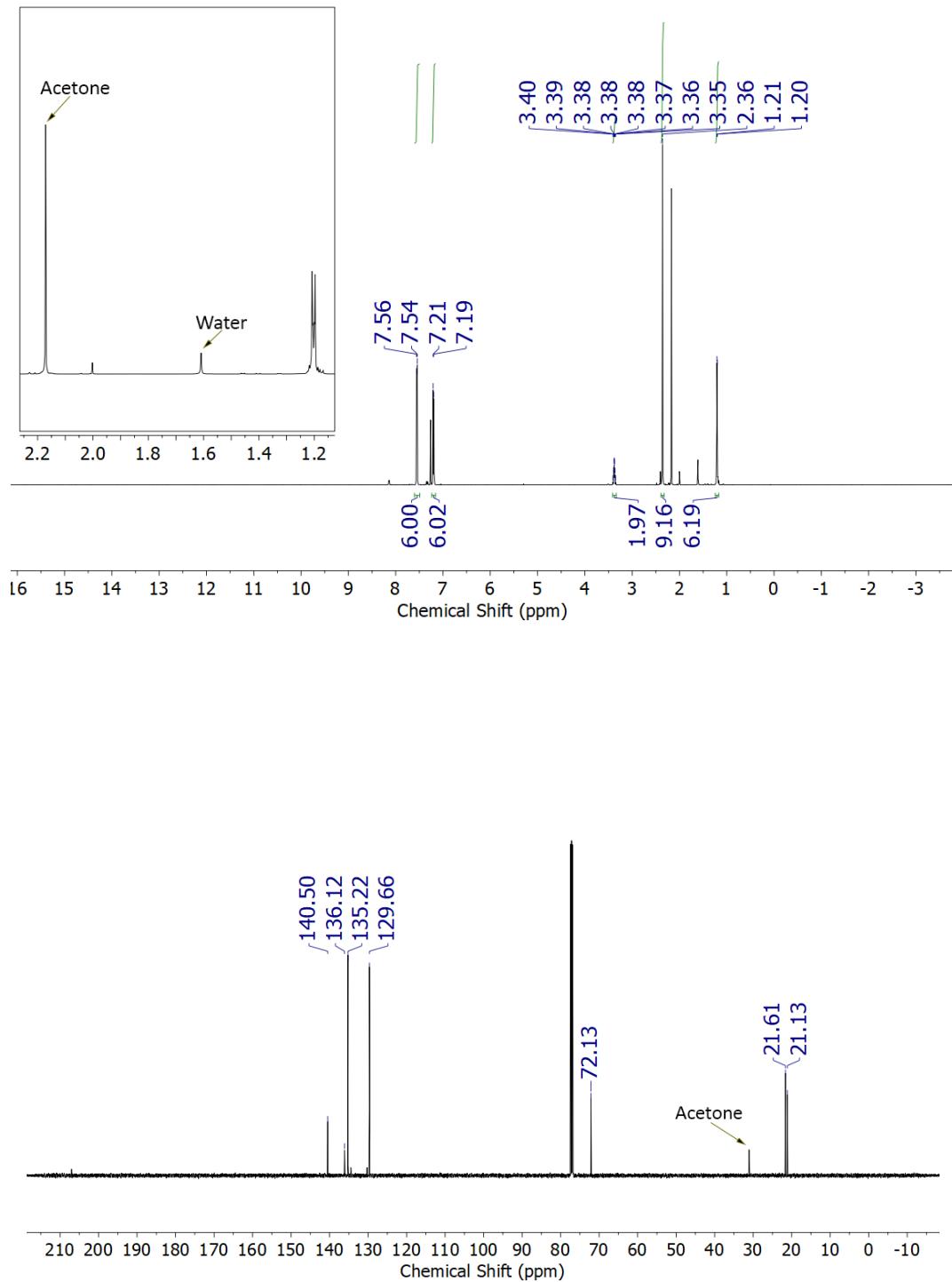
**Figure S6.** Top:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) spectrum of **1**. Bottom:  $^{13}\text{C}\{^1\text{H}\}$  NMR (126 MHz,  $\text{CDCl}_3$ ) spectrum of **1**.



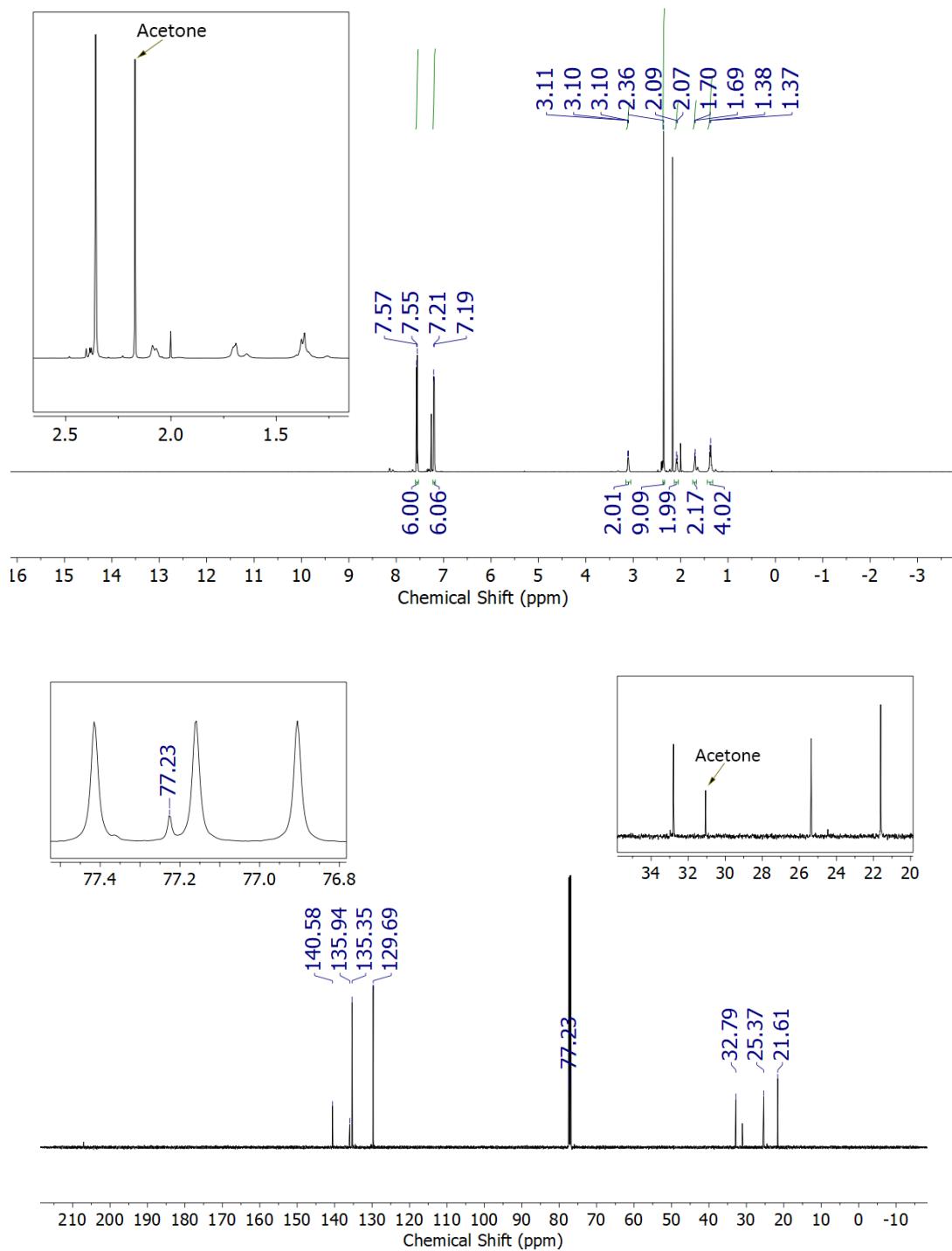
**Figure S7.** Top:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) spectrum of **2**. Bottom:  $^{13}\text{C}\{\text{H}\}$  NMR (126 MHz,  $\text{CDCl}_3$ ) spectrum of **2**.



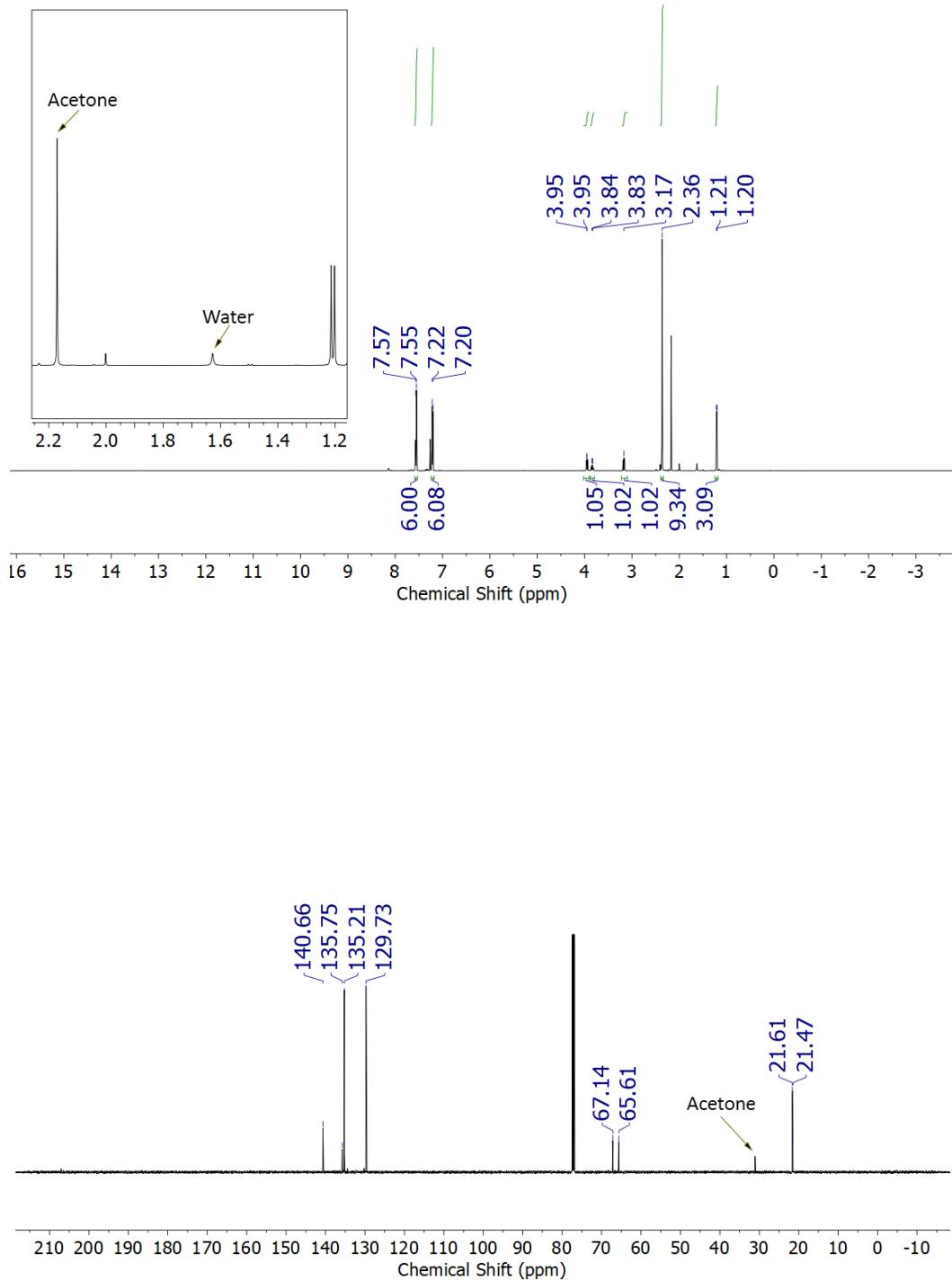
**Figure S8.** Top:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) spectrum of **3**. Bottom:  $^{13}\text{C}\{\text{H}\}$  NMR (126 MHz,  $\text{CDCl}_3$ ) spectrum of **3**.



**Figure S9.** Top:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) spectrum of **4**. Bottom:  $^{13}\text{C}\{^1\text{H}\}$  NMR (126 MHz,  $\text{CDCl}_3$ ) spectrum of **4**.



**Figure S10.** Top:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) spectrum of **5**. Bottom:  $^{13}\text{C}\{\mathbf{^1\text{H}}\}$  NMR (126 MHz,  $\text{CDCl}_3$ ) spectrum of **5**.



**Figure S11.** Top:  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) spectrum of **6**. Bottom:  $^{13}\text{C}\{\text{H}\}$  NMR (126 MHz,  $\text{CDCl}_3$ ) spectrum of **6**.