

Electronic Supplementary Material

Not the usual suspect – an unexpected organometallic product during the synthesis of a cytotoxic platinum(II) complex

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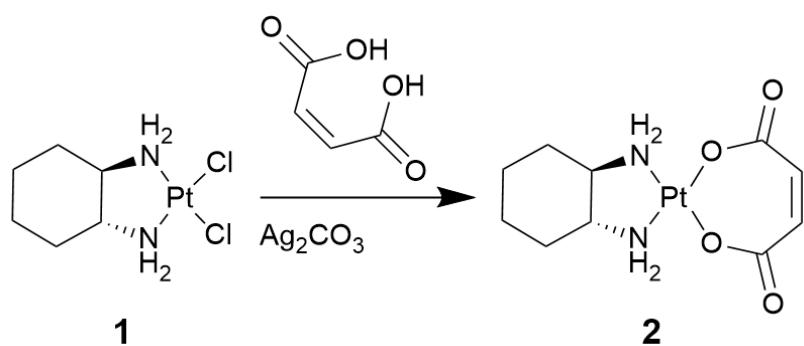
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Contents

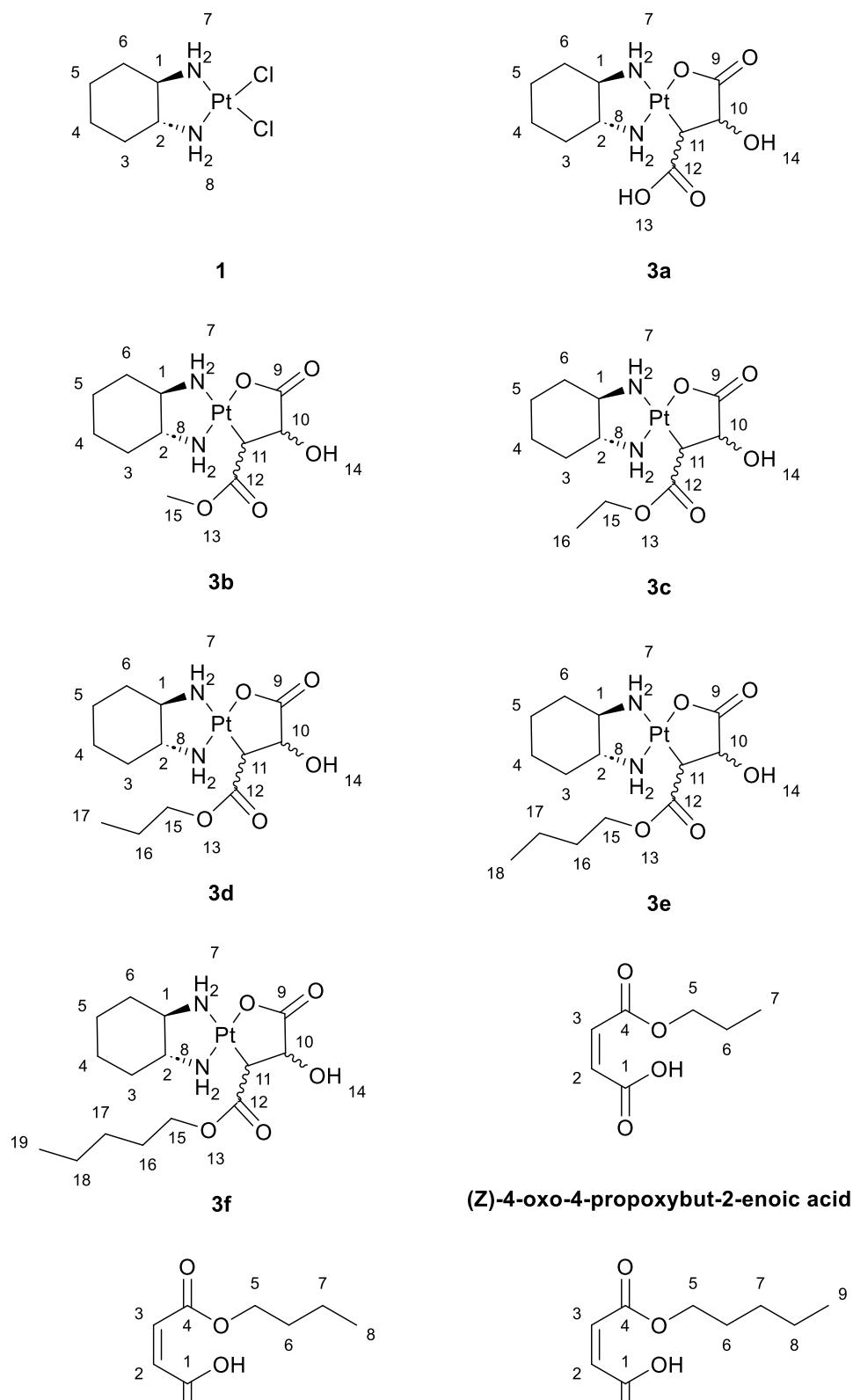
Contents	2
Planned reaction	3
NMR Numbering Scheme	4
Cytotoxicity data	5
DNA-interaction assay	6
Stability assessment via RP-HPLC	7
¹ H NMR spectrum of (Z)-4-oxo-4-propoxybut-2-enoic acid	10
¹³ C NMR spectrum of (Z)-4-oxo-4-propoxybut-2-enoic acid	11
¹ H NMR spectrum of (Z)-4-butoxy-4-oxobut-2-enoic acid	12
¹³ C NMR spectrum of (Z)-4-butoxy-4-oxobut-2-enoic acid	13
¹ H NMR spectrum of (Z)-4-oxo-4-(pentyloxy)but-2-enoic acid	14
¹³ C NMR spectrum of (Z)-4-oxo-4-(pentyloxy)but-2-enoic acid	15
¹ H NMR spectrum of compound 3a	16
¹³ C NMR spectrum of compound 3a	17
¹⁹⁵ Pt NMR spectrum of compound 3a	18
¹ H NMR spectrum of compound 3b	19
¹³ C NMR spectrum of compound 3b	20
¹⁹⁵ Pt NMR spectrum of compound 3b	21
¹ H NMR spectrum of compound 3c	22
¹³ C NMR spectrum of compound 3c	23
¹⁹⁵ Pt NMR spectrum of compound 3c	24
¹ H NMR spectrum of compound 3d	25
¹³ C NMR spectrum of compound 3d	26
¹⁹⁵ Pt NMR spectrum of compound 3d	27
¹ H NMR spectrum of compound 3e	28
¹³ C NMR spectrum of compound 3e	29
¹⁹⁵ Pt NMR spectrum of compound 3e	30
¹ H NMR spectrum of compound 3f	31
¹³ C NMR spectrum of compound 3f	32
¹⁹⁵ Pt NMR spectrum of compound 3f	33
Comparison of calculated geometries of diastereoisomers A and B of complex 3a with experimental geometries	34
Comparison of calculated geometries of diastereoisomers A and B of complex 3a with either carboxyl or carboxylate group	35
X-Ray Diffraction Analysis of complex 3a	36

Planned reaction



Scheme S1 Planned reaction to the target complex **2**, featuring a chelating maleato complex.

NMR Numbering Scheme



(Z)-4-butoxy-4-oxobut-2-enoic acid (Z)-4-oxo-4-(pentyloxy)but-2-enoic acid

Fig. S1 Atom labeling as used for all descriptions. Both diastereomers are numbered in the same fashion.

Cytotoxicity data

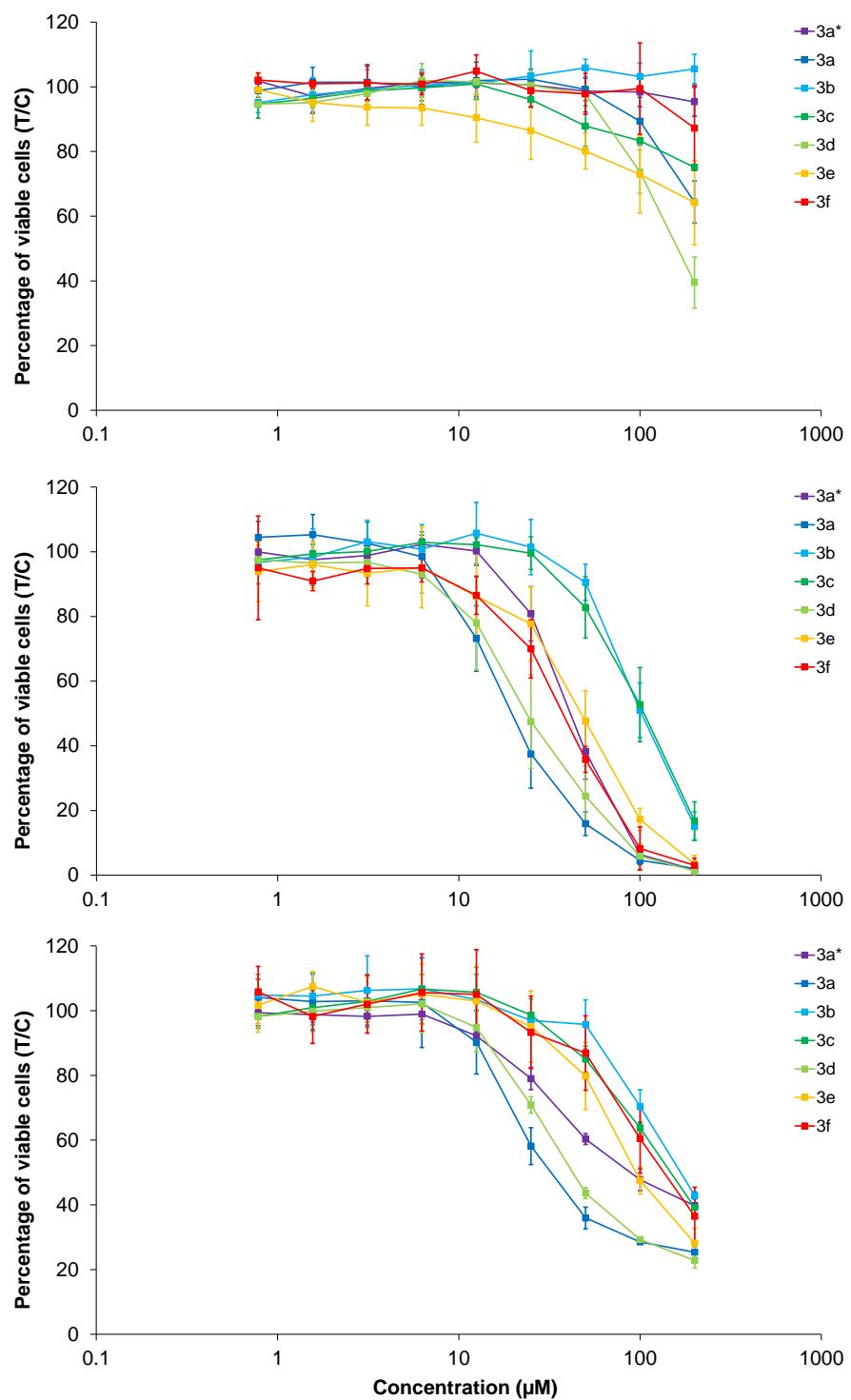


Fig. S2 Concentration-effect curves of **3a**–**3f** in A549 (top), CH1/PA-1 (middle) and SW480 (bottom) cells, obtained by MTT assays with 96 h exposure time. Values are means \pm standard deviations from at least three independent experiments.

DNA-interaction assay

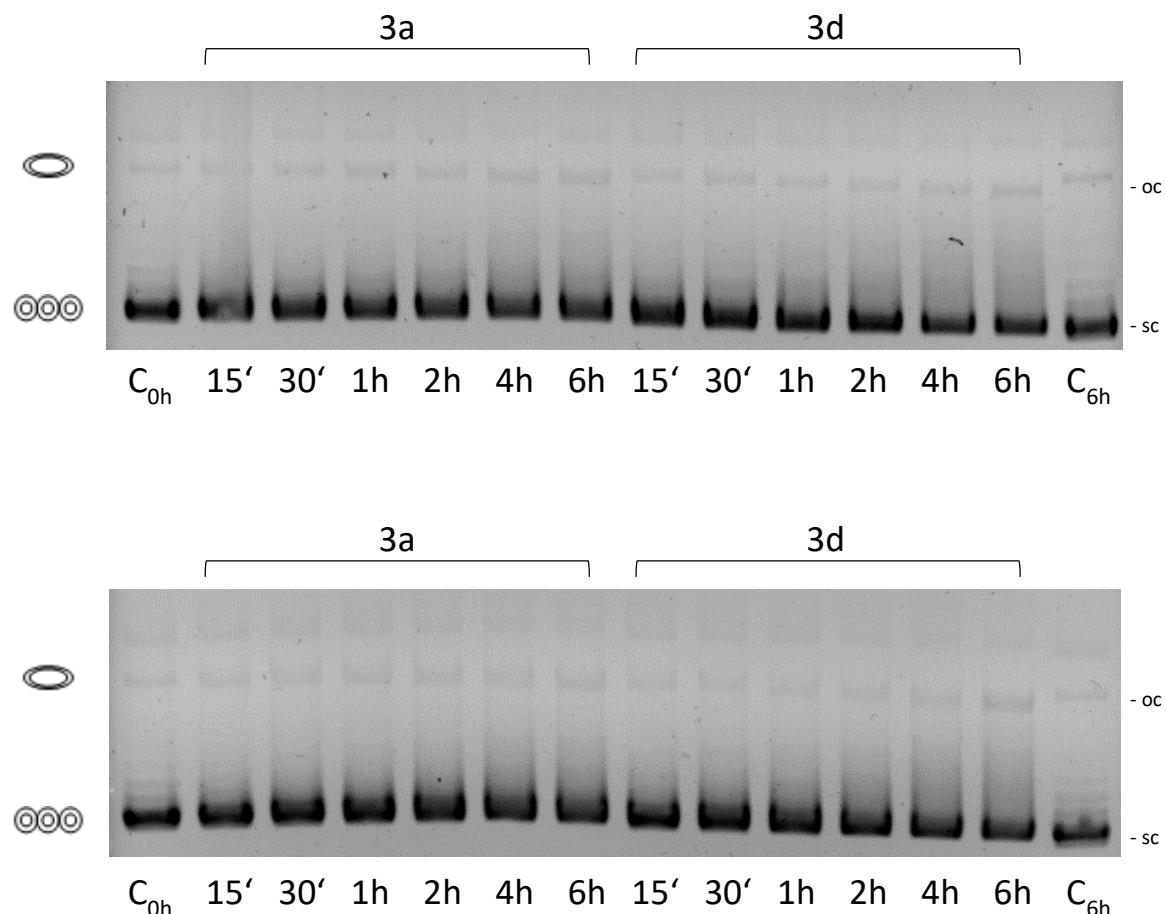
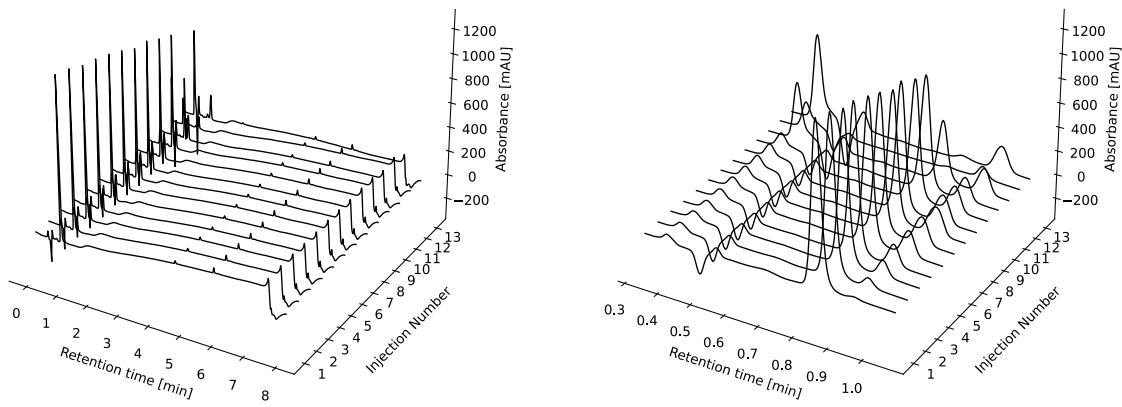


Fig. S3 Electropherogram of plasmid dsDNA upon treatment with 50 μ M of compounds **3a** and **3d**. The pUC19 plasmid was incubated with the complexes for 6 different time periods (from 15 min to 6 h) at 37 °C and then subjected to agarose gel electrophoresis. 'C_{0h}' corresponds to an untreated, non-preincubated control at the time point of gel loading, while 'C_{6h}' corresponds to an untreated, 6 h incubated control; oc, open circular; sc, supercoiled. (As a reference for analyzing the bands of the gel, Figure S80 from ref. doi:10.1021/acs.inorgchem.1c01083 was used.)

Stability assessment via RP-HPLC

Oxaliplatin without 5'-GMP



Oxaliplatin with 5'-GMP

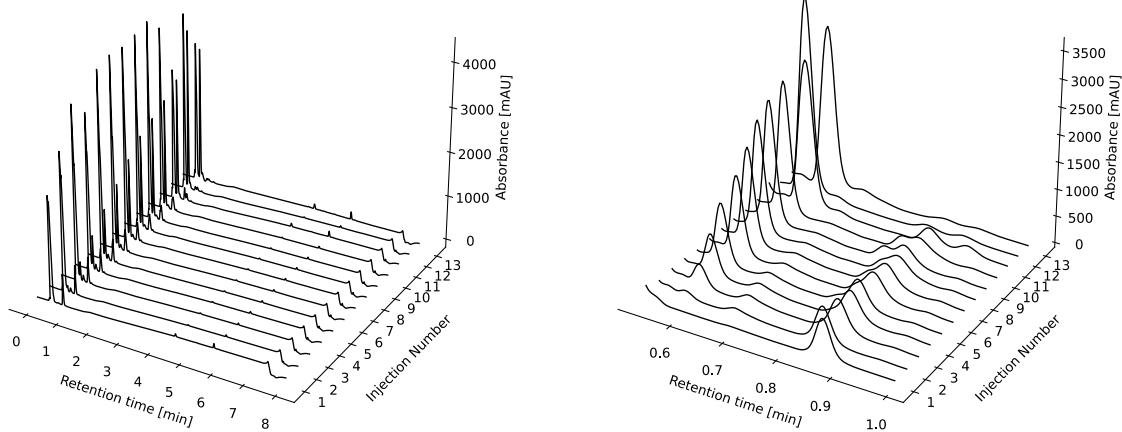
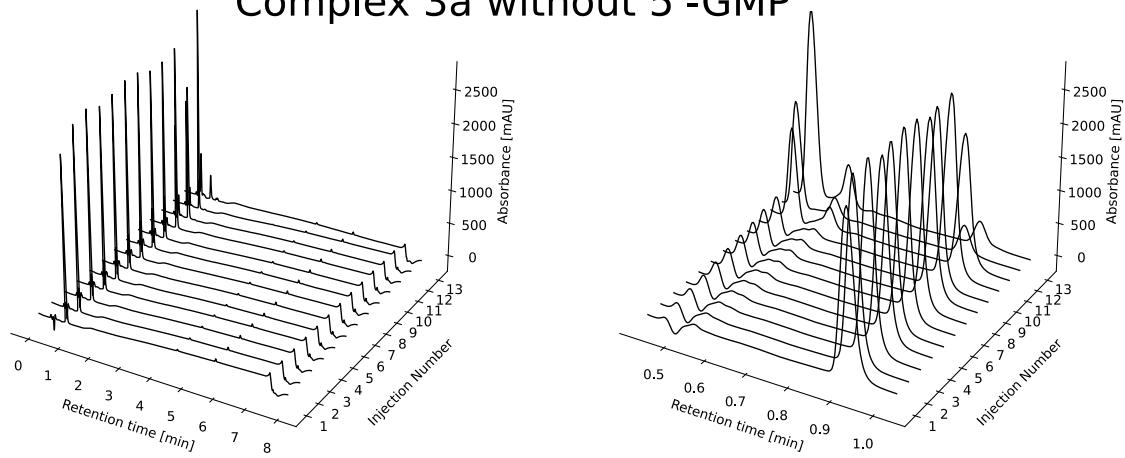


Fig. S4 HPLC traces indicating the reactivity of oxaliplatin towards 5'-GMP under physiologically relevant conditions as described in the experimental section. The left image shows the complete HPLC run, whereas the right image is zoomed to the retention time window showing the relevant peaks. Retention time of oxaliplatin is 0.8 min and of GMP is 0.45 min respectively.

Complex 3a without 5'-GMP



Complex 3a with 5'-GMP

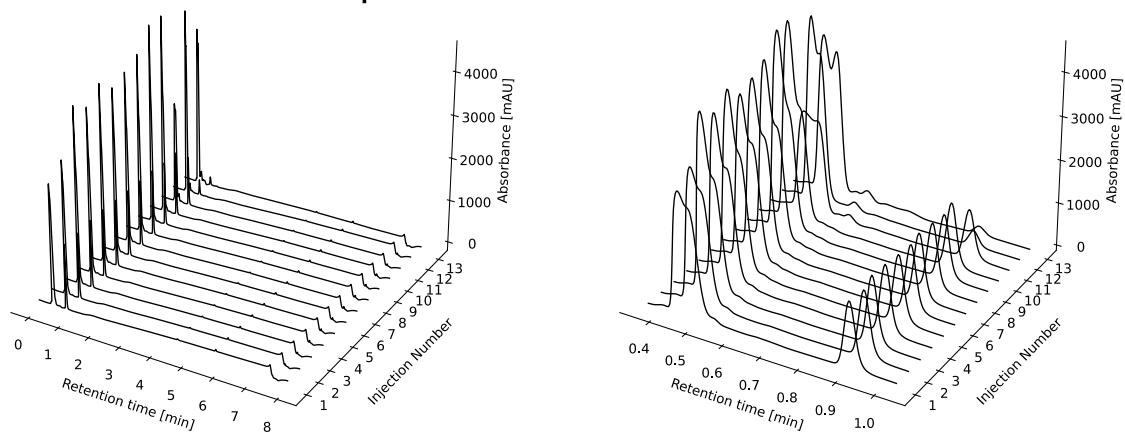


Fig. S5 HPLC traces indicating the reactivity of complex **3a** towards 5'-GMP under physiologically relevant conditions as described in the experimental section. The left image shows the complete HPLC run, whereas the right image is zoomed to the retention time window showing the relevant peaks. Retention time of complex **3a** is 0.9 min and of GMP is 0.45 min, respectively.

5'-GMP

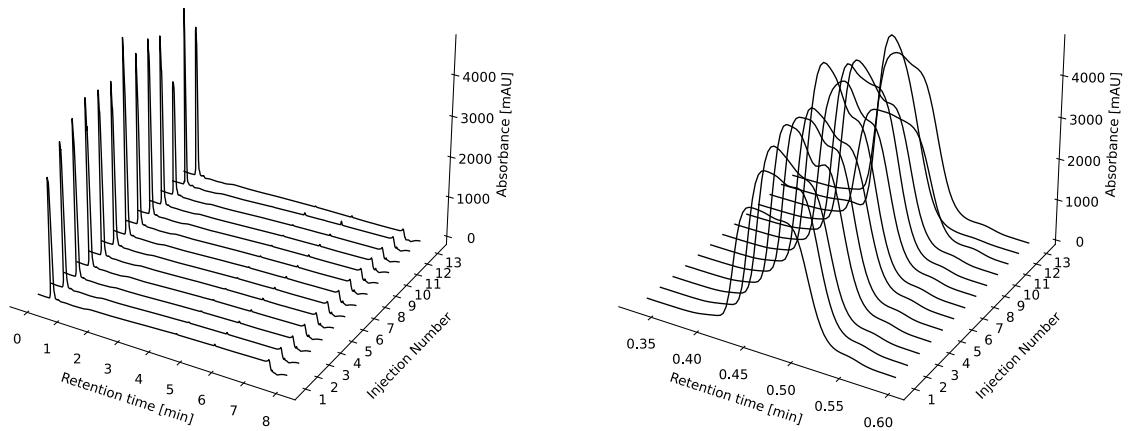
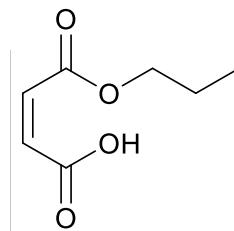
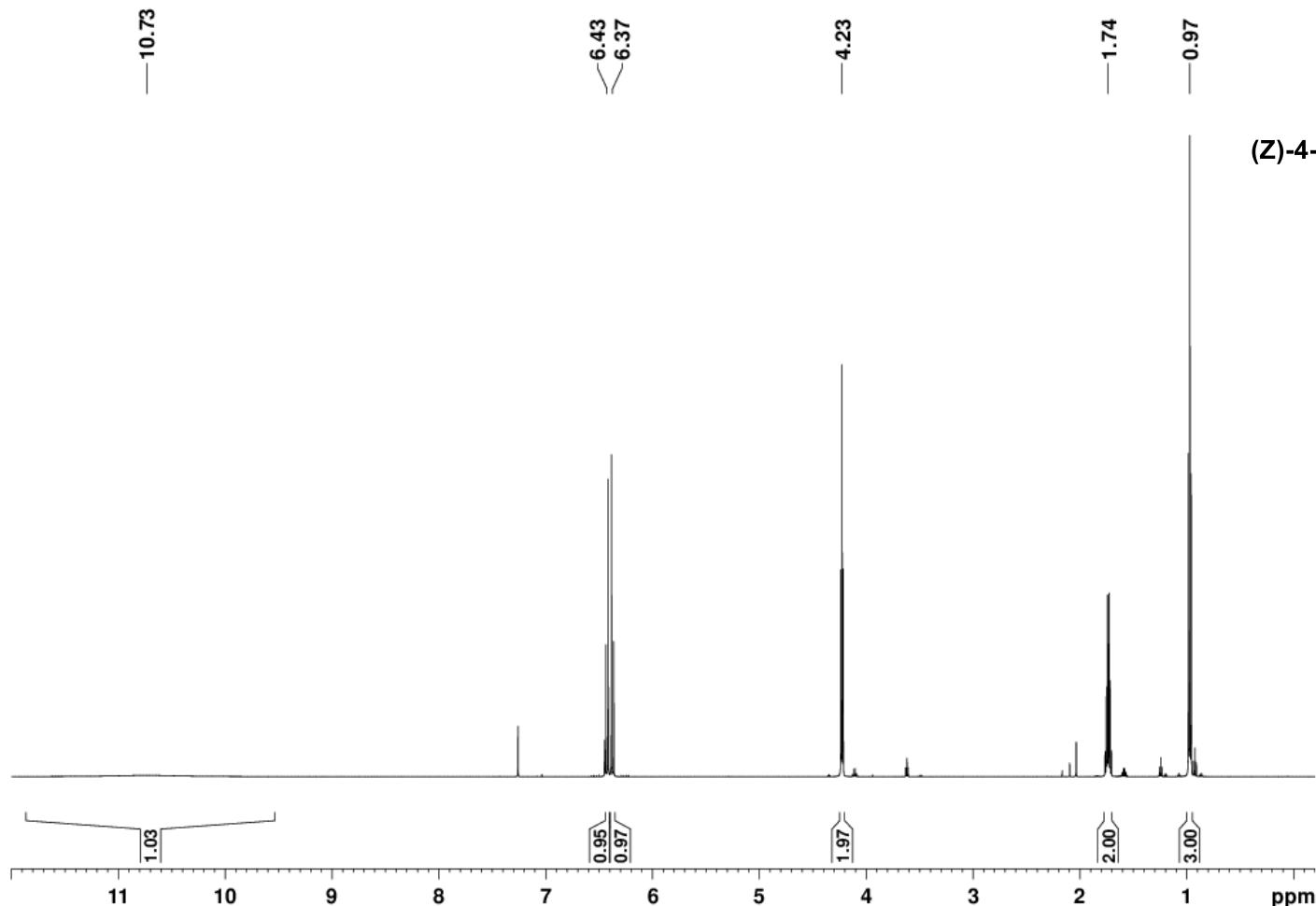


Fig. S6 HPLC traces indicating the stability of 5'-GMP under physiologically relevant conditions as described in the experimental section. The left image shows the complete HPLC run, whereas the right image is zoomed to the retention time window showing the relevant peaks. Retention time of GMP is 0.45 min.

^1H NMR spectrum of (Z)-4-oxo-4-propoxybut-2-enoic acid



(Z)-4-oxo-4-propoxybut-2-enoic acid

Fig. S4 ^1H NMR spectrum of (Z)-4-oxo-4-propoxybut-2-enoic acid.

^{13}C NMR spectrum of (Z)-4-oxo-4-propoxybut-2-enoic acid

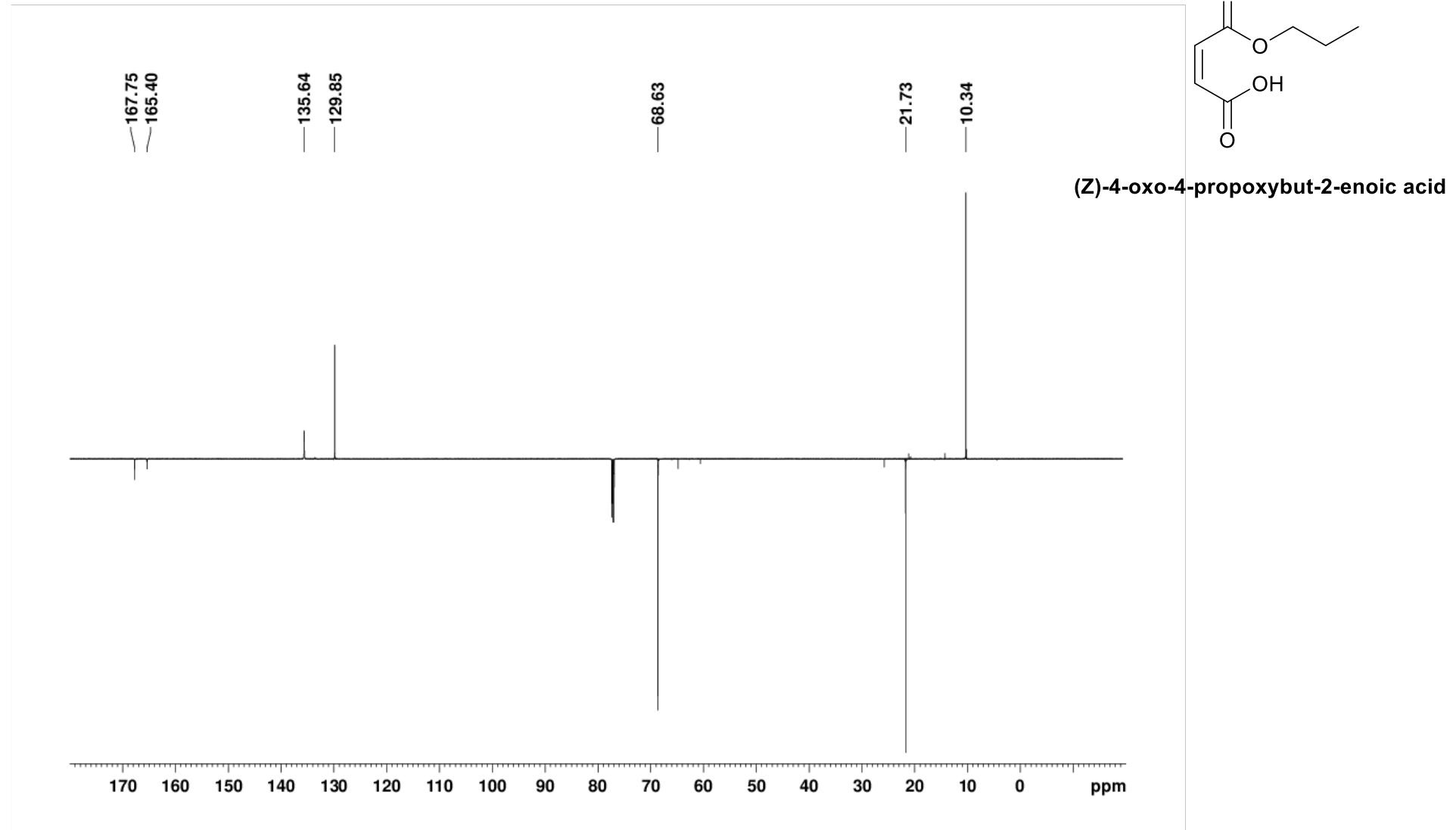


Fig. S5 ^{13}C NMR spectrum of (Z)-4-oxo-4-propoxybut-2-enoic acid.

^1H NMR spectrum of (Z)-4-butoxy-4-oxobut-2-enoic acid

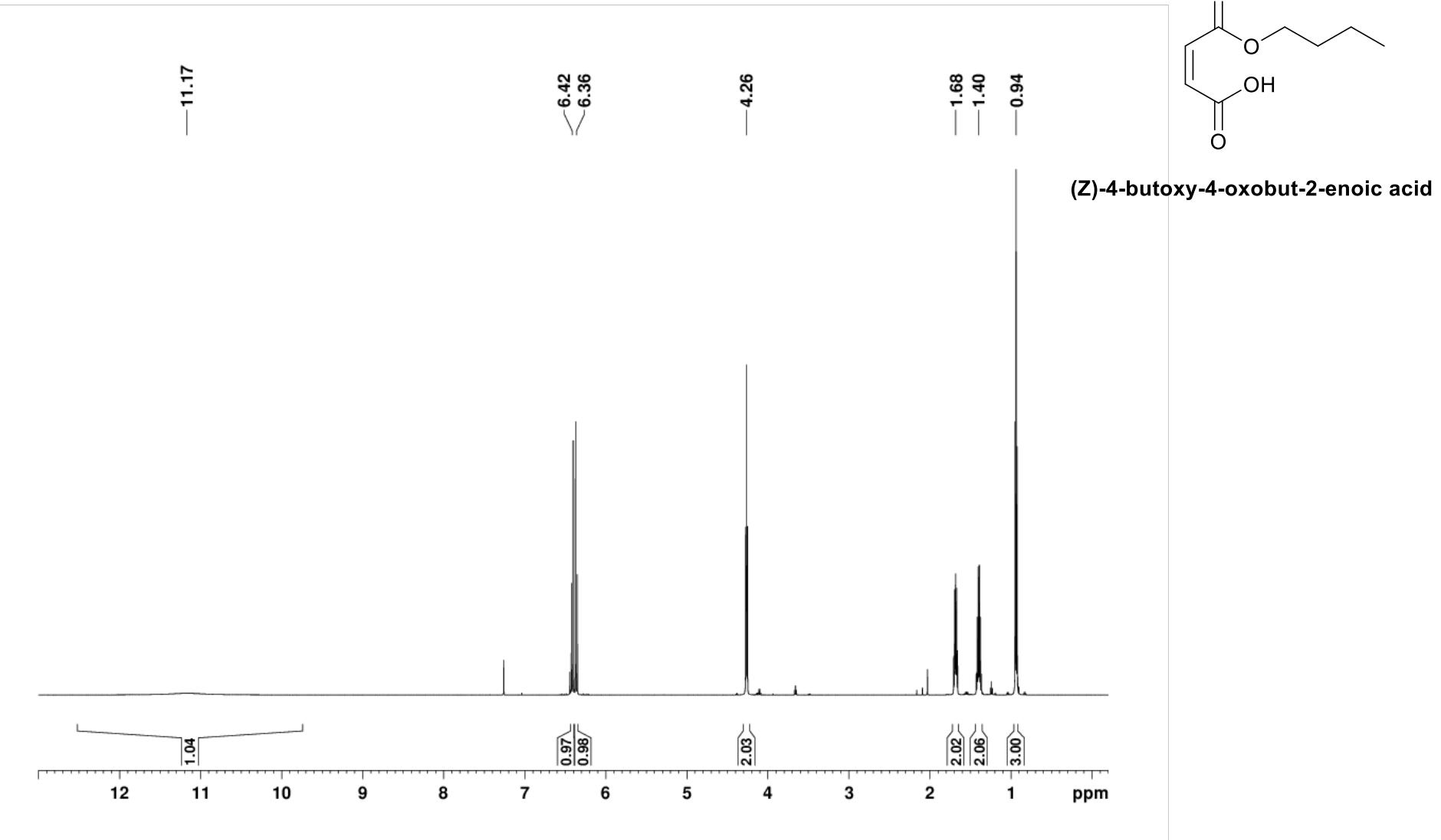


Fig. S6 ^1H NMR spectrum of (Z)-4-butoxy-4-oxobut-2-enoic acid.

^{13}C NMR spectrum of (Z)-4-butoxy-4-oxobut-2-enoic acid

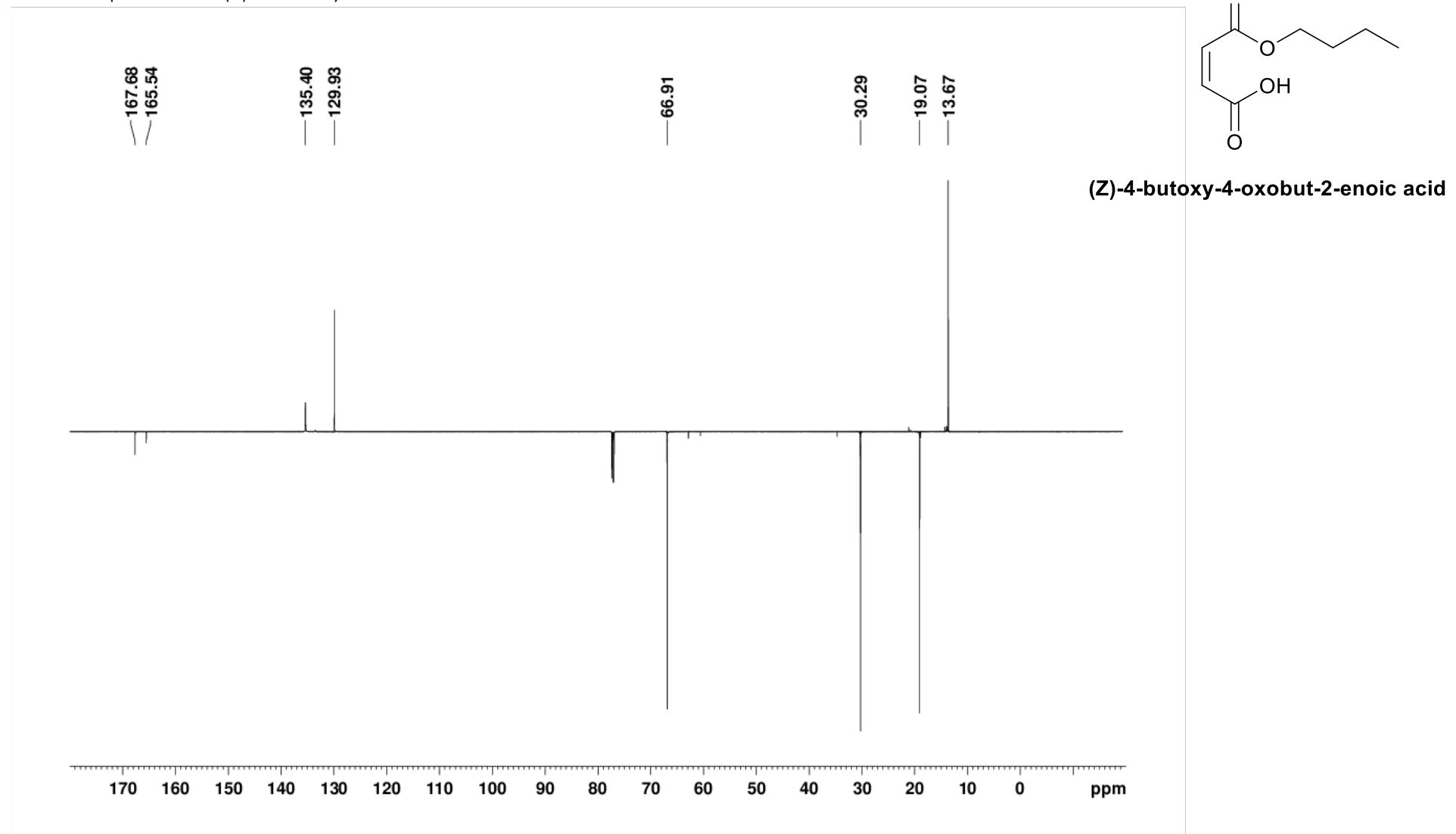


Fig. S7 ^{13}C NMR spectrum of (Z)-4-butoxy-4-oxobut-2-enoic acid

^1H NMR spectrum of (Z)-4-oxo-4-(pentyloxy)but-2-enoic acid

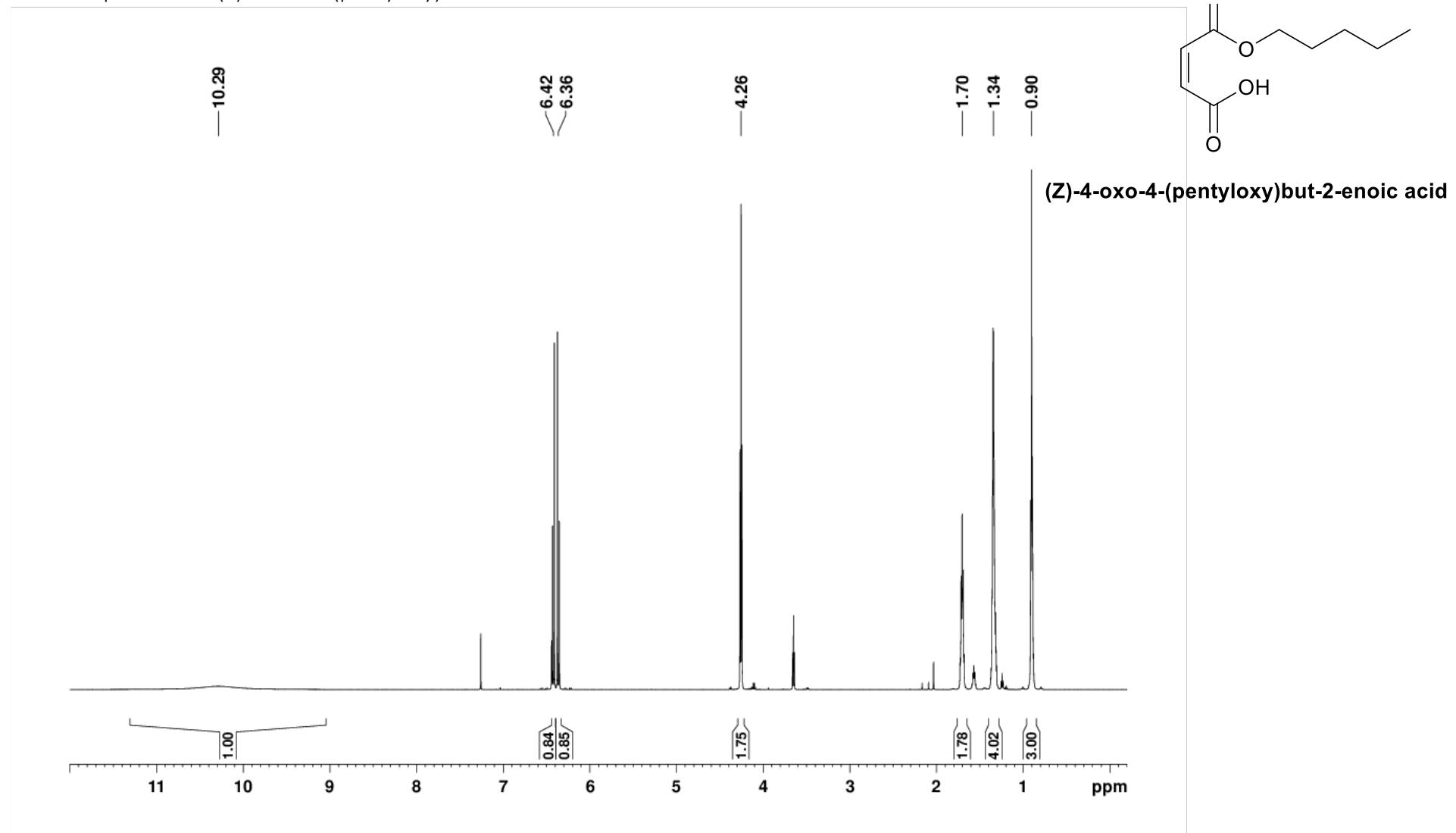


Fig. S8 ^1H NMR spectrum of (Z)-4-oxo-4-(pentyloxy)but-2-enoic acid.

^{13}C NMR spectrum of (Z)-4-oxo-4-(pentyloxy)but-2-enoic acid

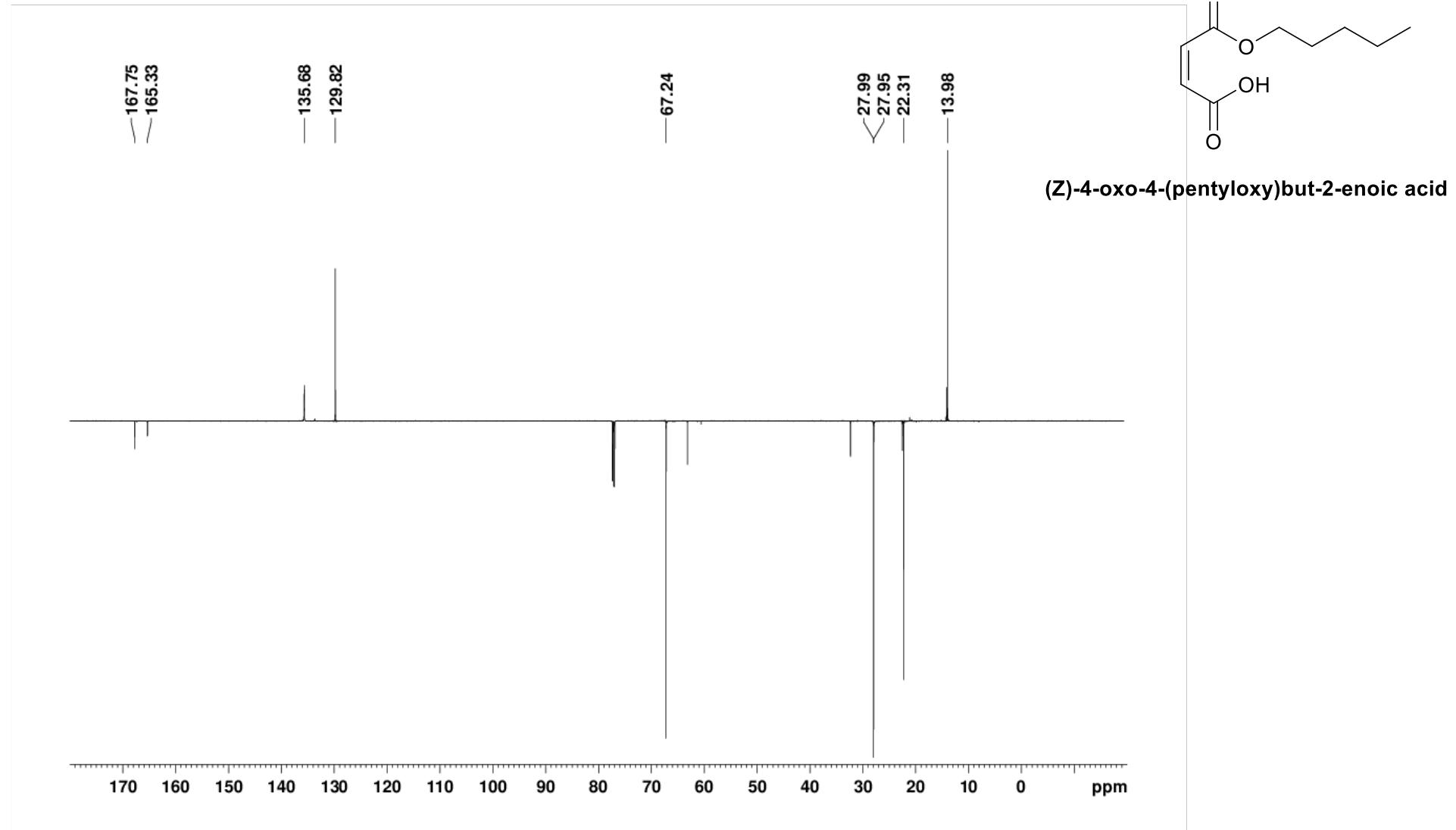


Fig. S9 ^{13}C NMR spectrum of (Z)-4-oxo-4-(pentyloxy)but-2-enoic acid.

¹H NMR spectrum of compound 3a

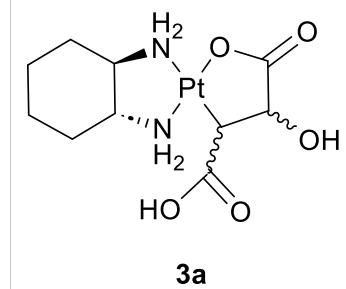
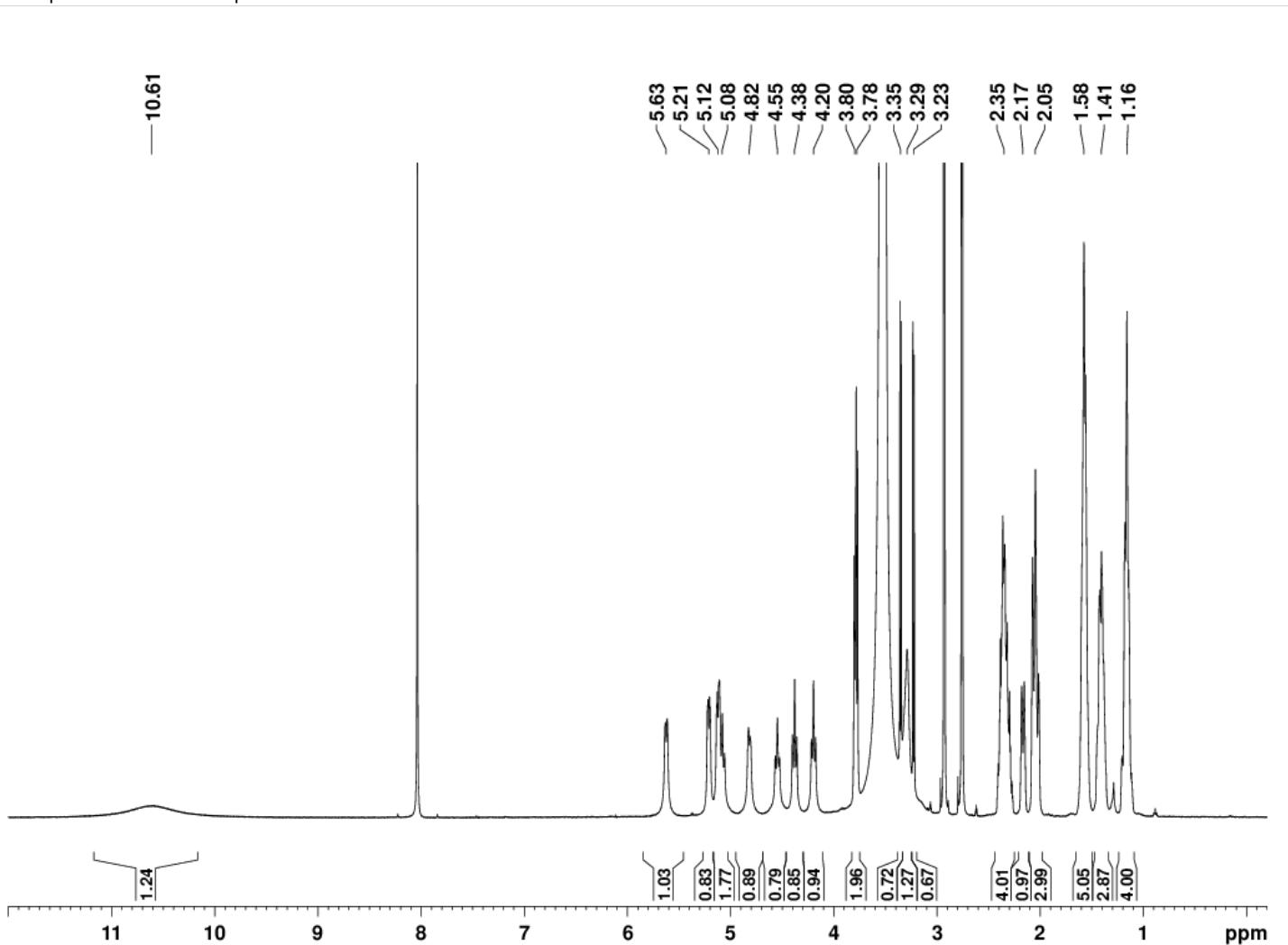


Fig. S10 ¹H NMR spectrum of compound 3a.

^{13}C NMR spectrum of compound 3a

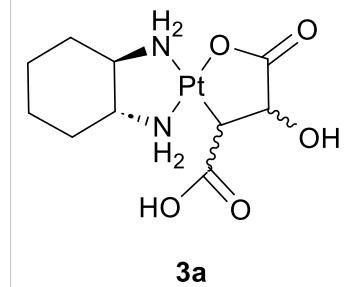
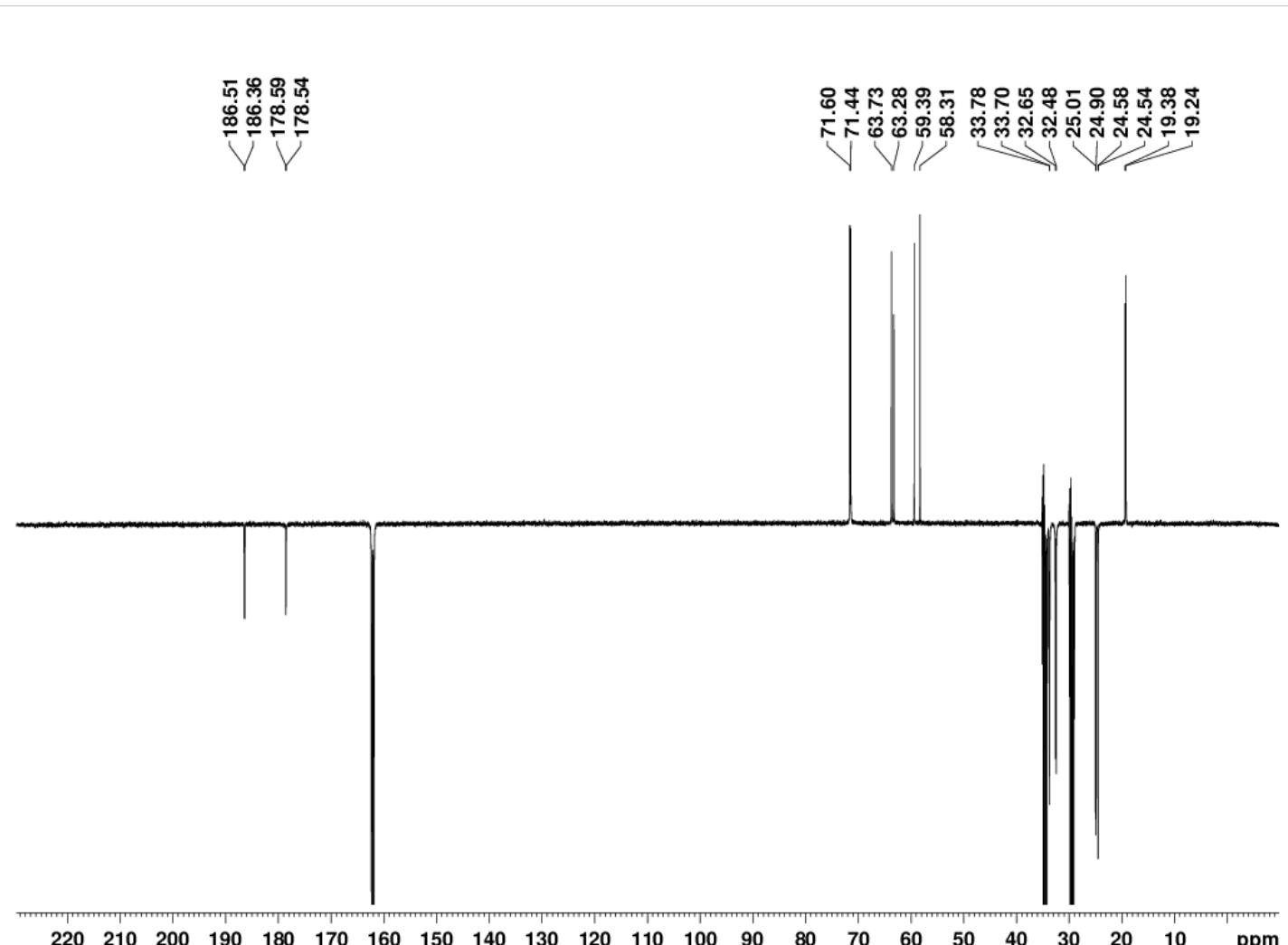


Fig. S11 ^{13}C NMR spectrum of compound 3a.

^{195}Pt NMR spectrum of compound **3a**

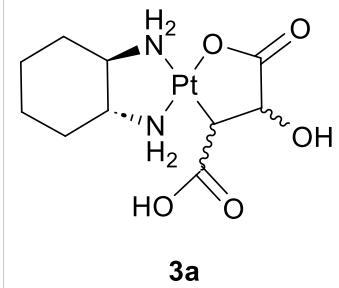
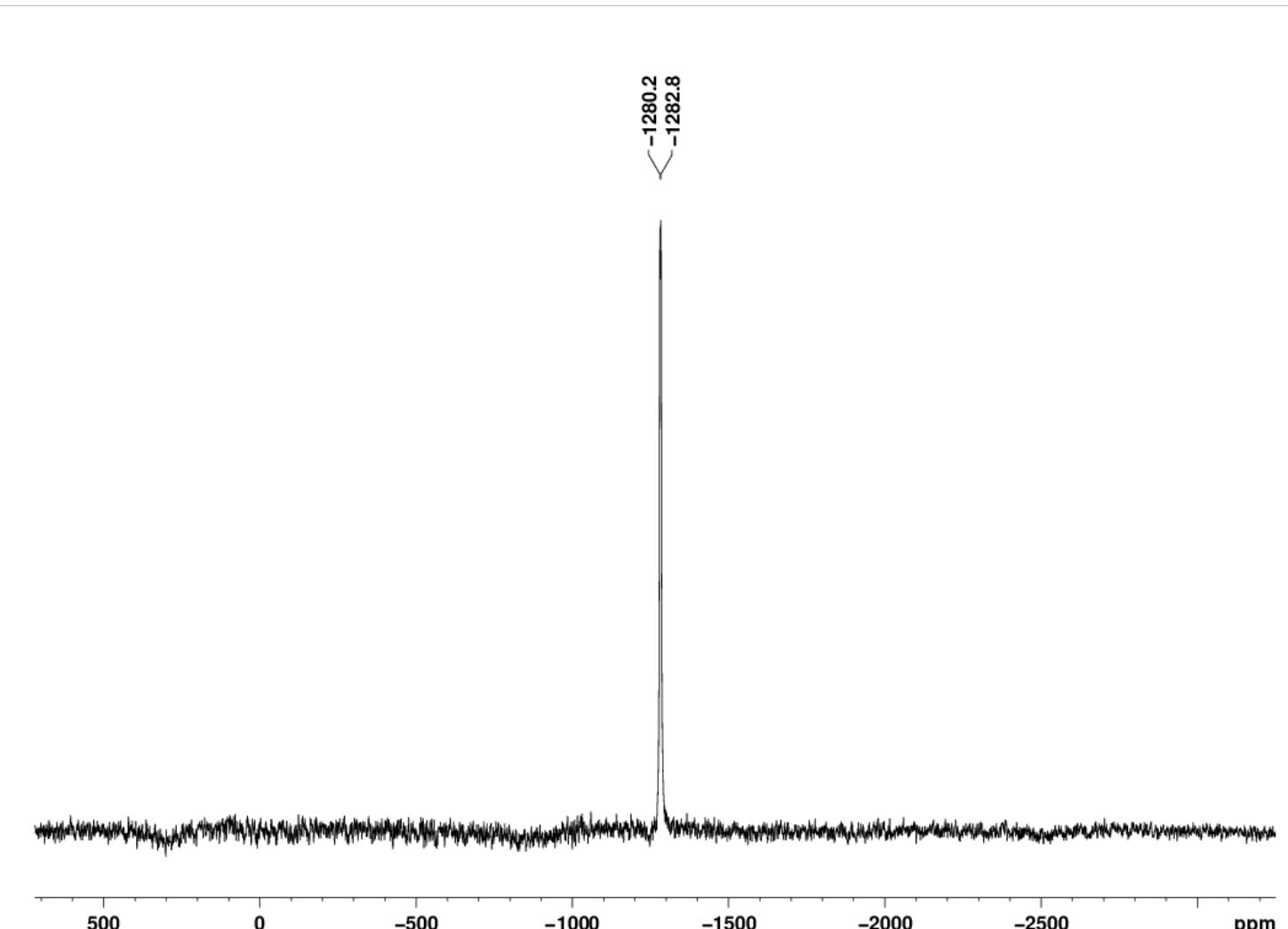


Fig. S12 ^{195}Pt NMR spectrum of compound **3a**.

¹H NMR spectrum of compound 3b

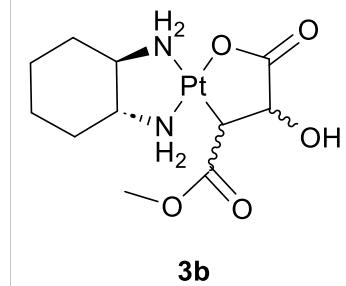
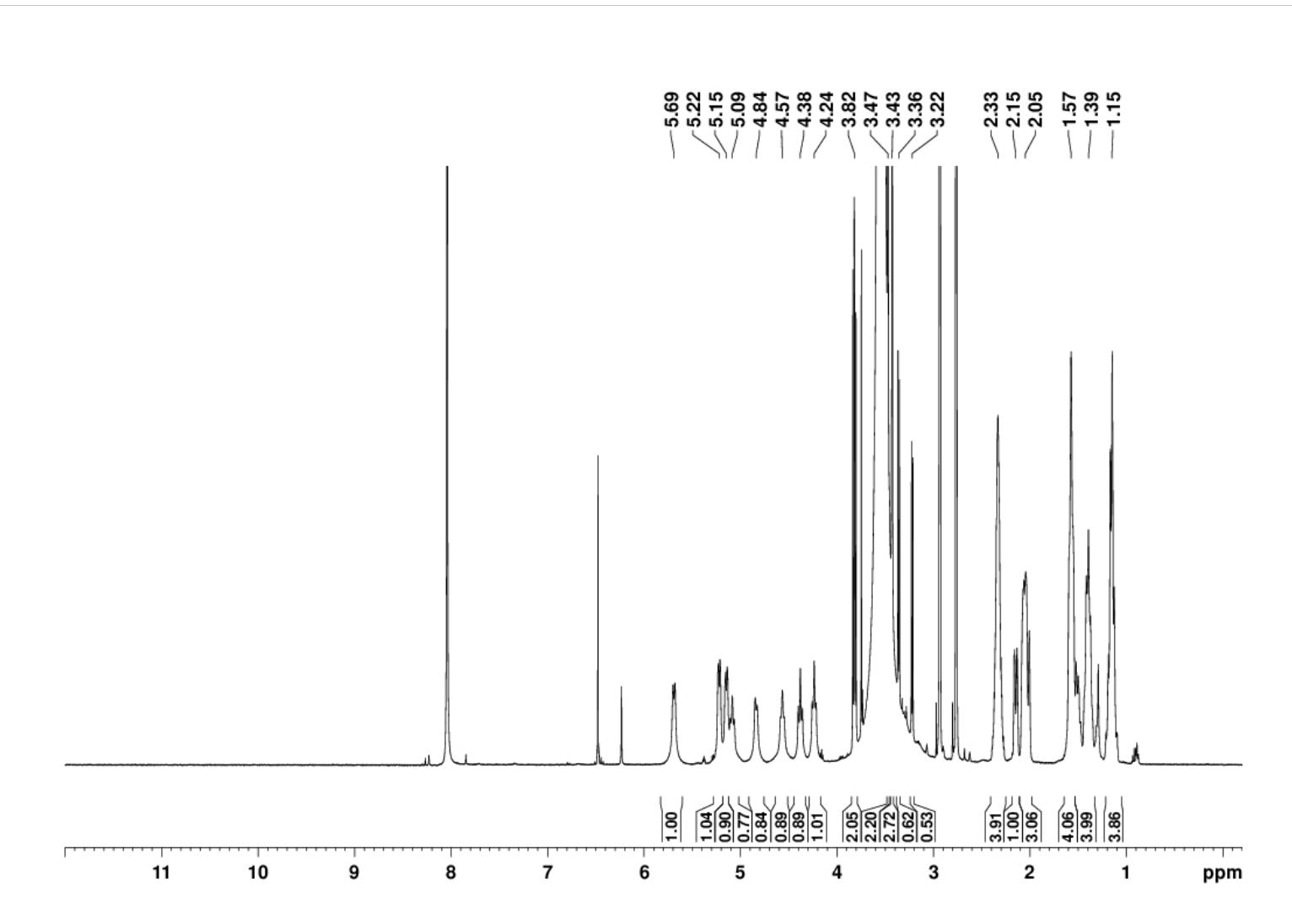


Fig. S13 ¹H NMR spectrum of compound 3b.

¹³C NMR spectrum of compound 3b

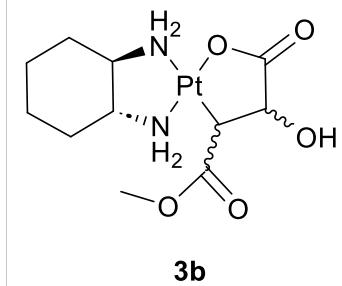
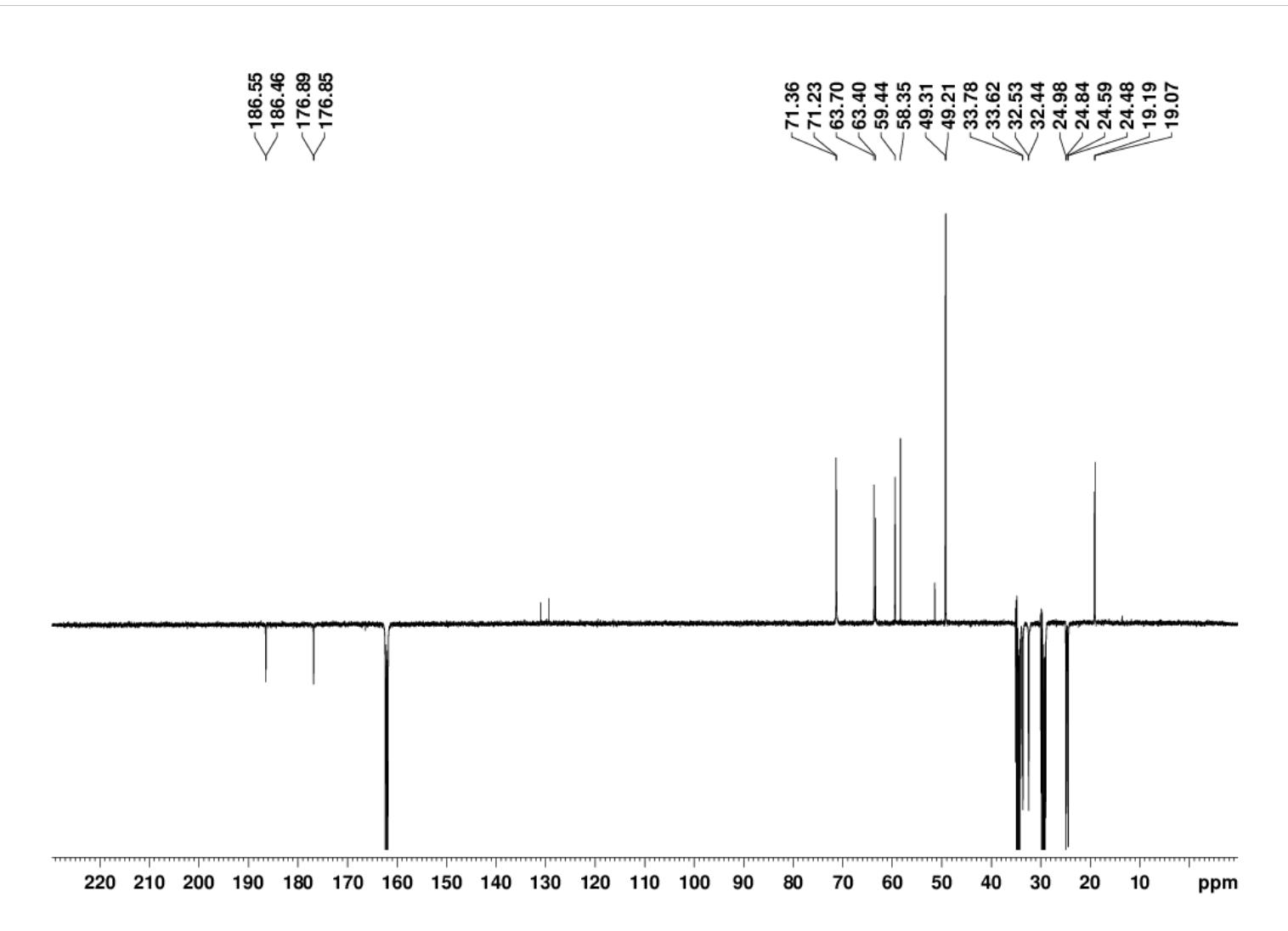


Fig. S14 ¹³C NMR spectrum of compound 3b.

^{195}Pt NMR spectrum of compound **3b**

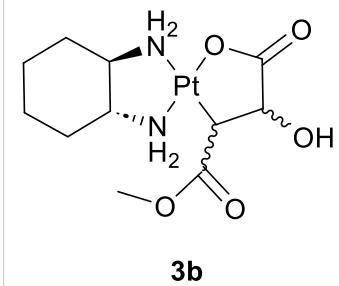
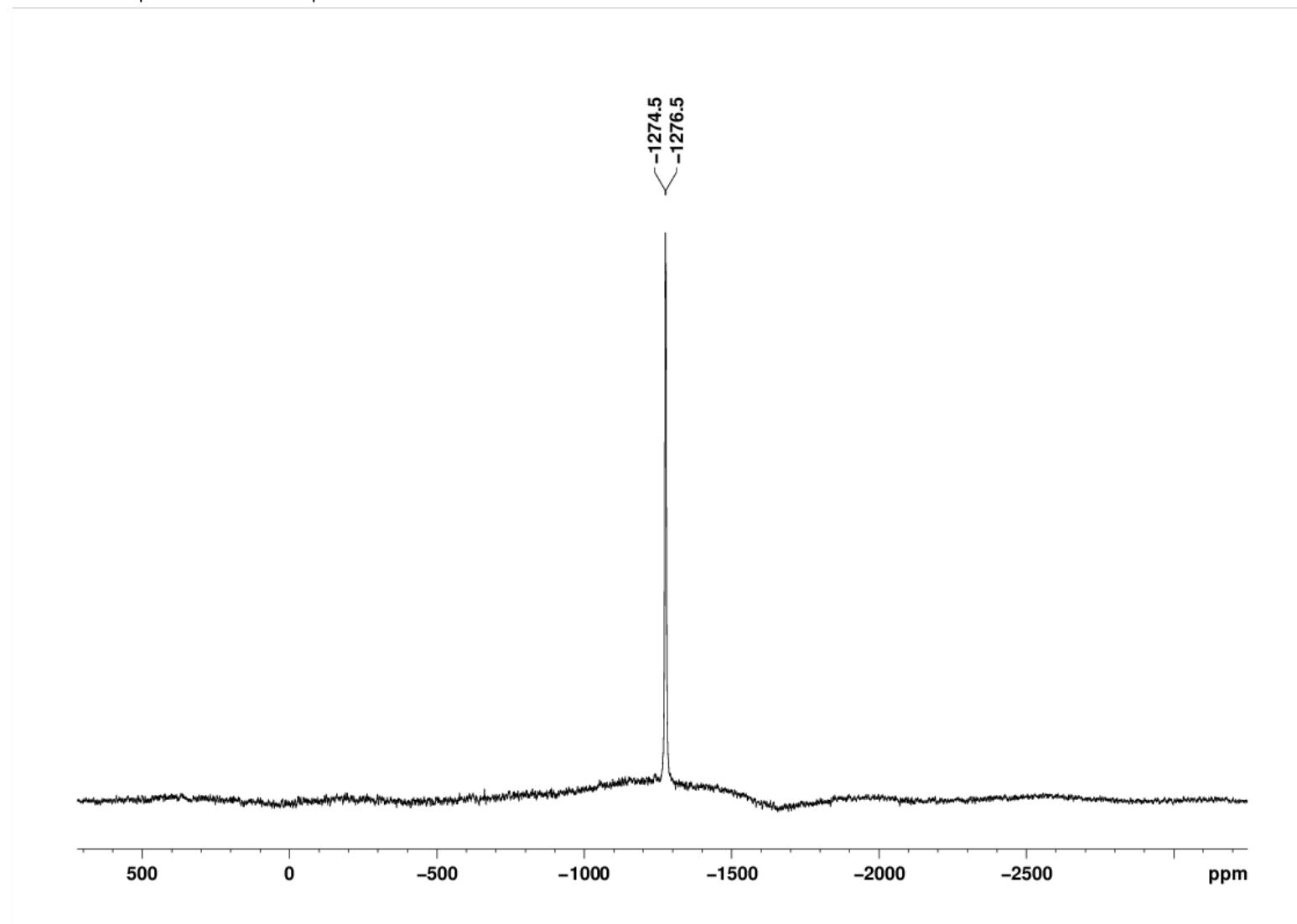


Fig. S15 ^{195}Pt NMR spectrum of compound **3b**.

¹H NMR spectrum of compound 3c

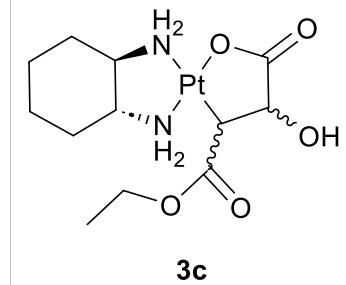
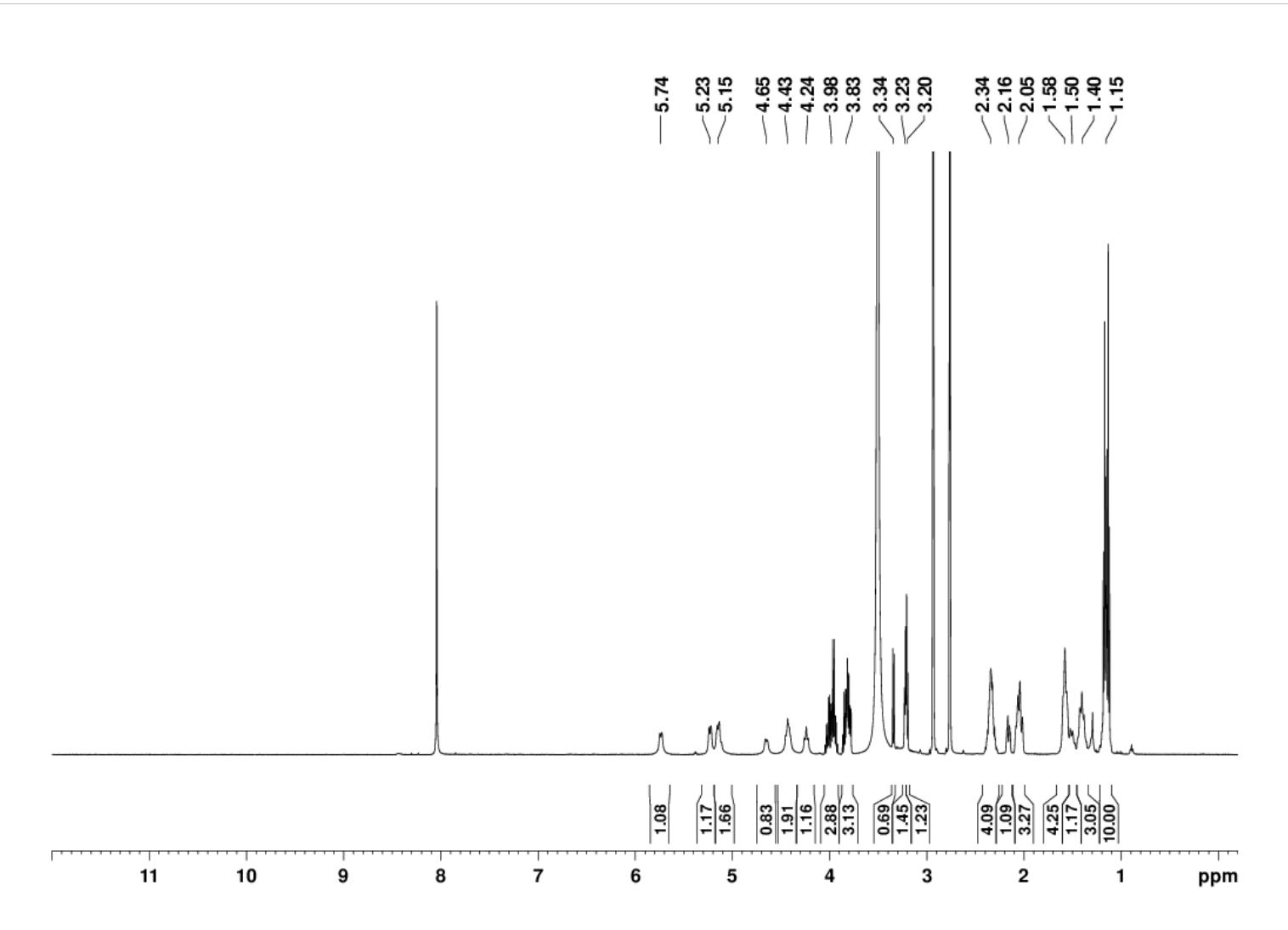


Fig. S16 ¹H NMR spectrum of compound 3c.

¹³C NMR spectrum of compound 3c

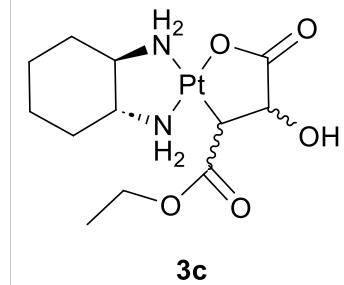
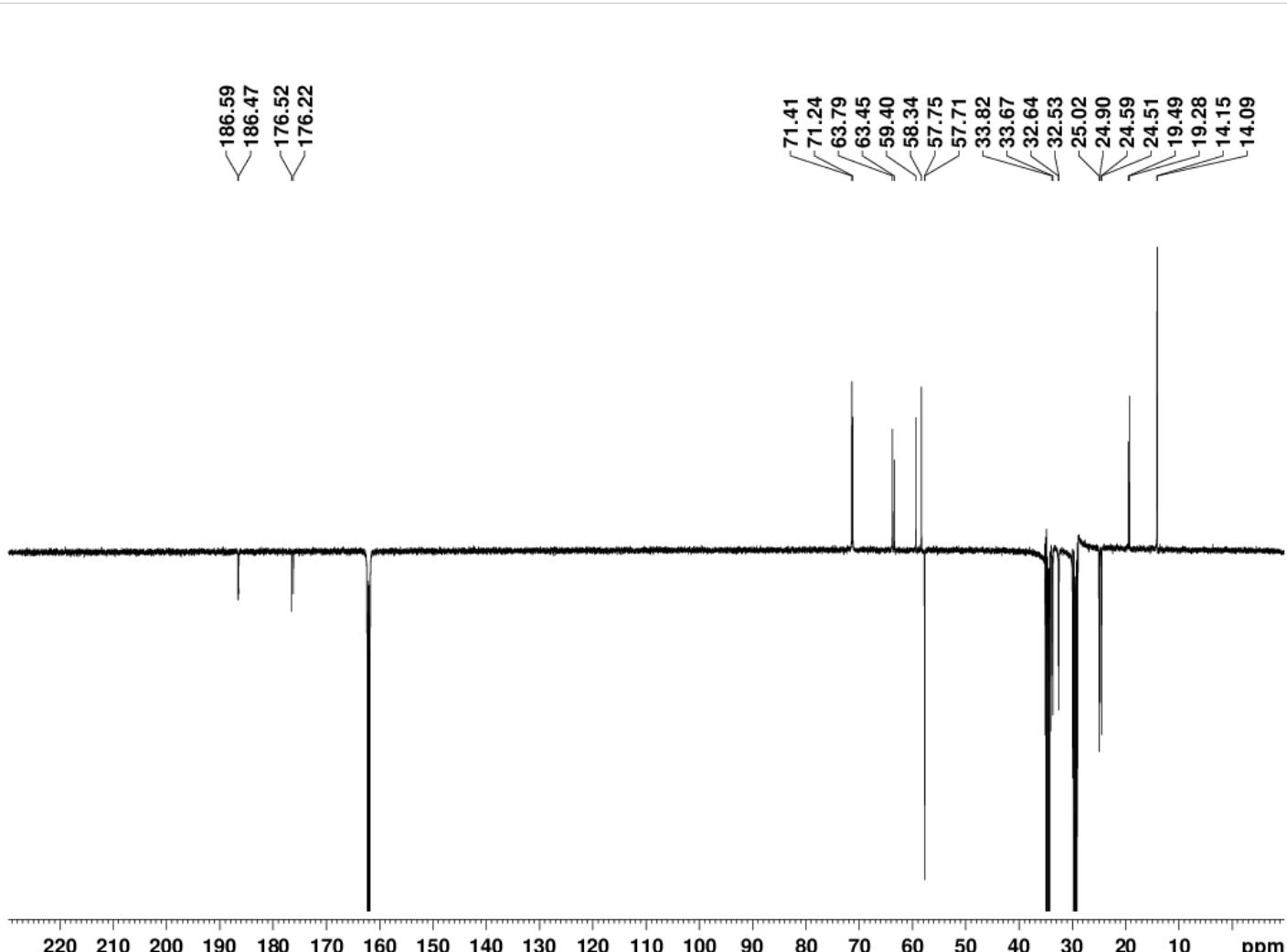


Fig. S17 ¹³C NMR spectrum of compound 3c.

^{195}Pt NMR spectrum of compound **3c**

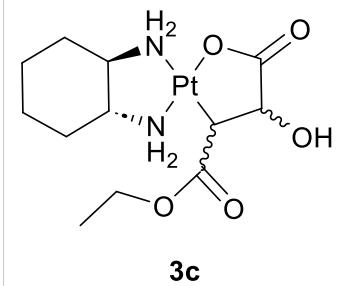
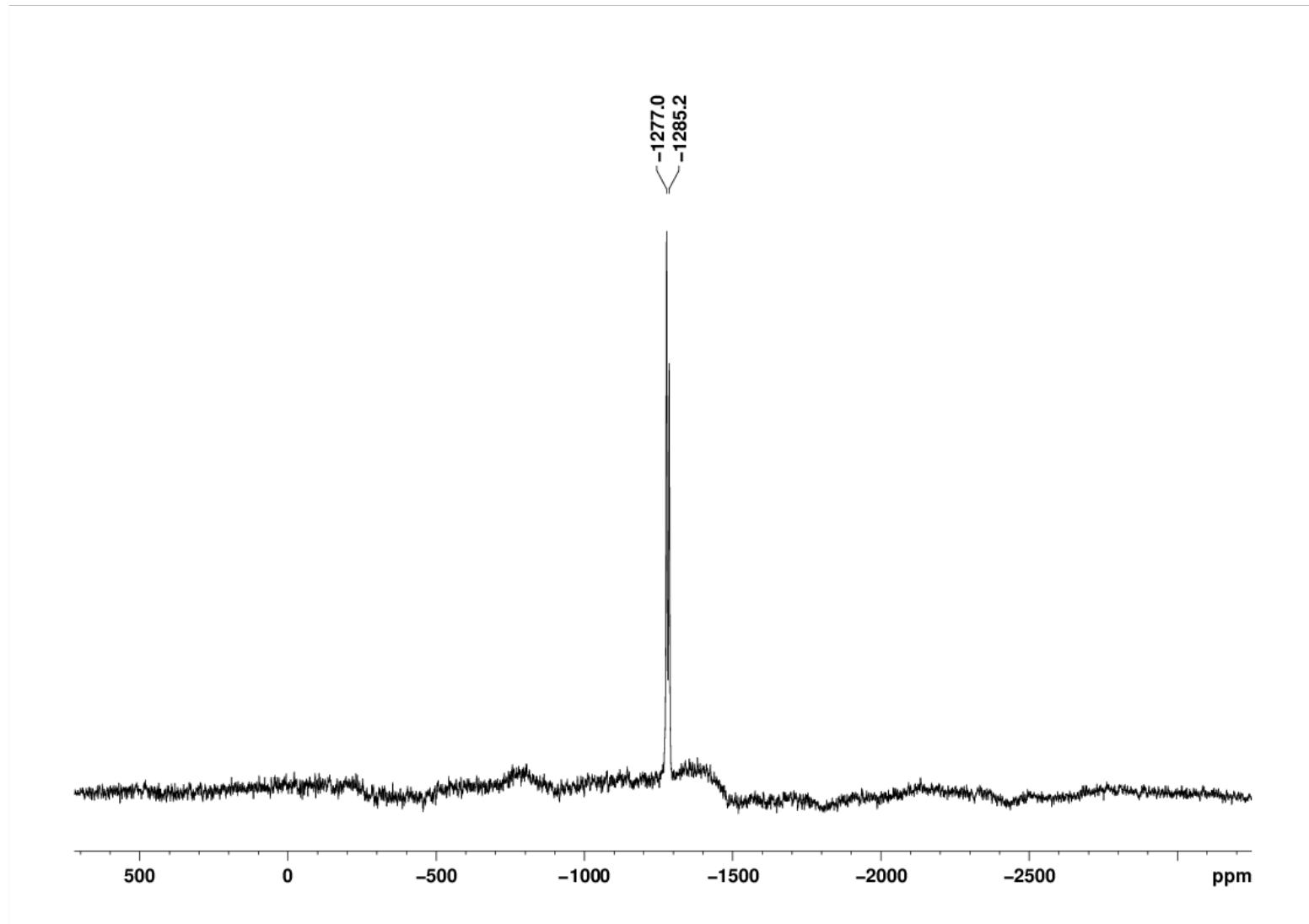


Fig. S18 ^{195}Pt NMR spectrum of compound **3c**.

¹H NMR spectrum of compound 3d

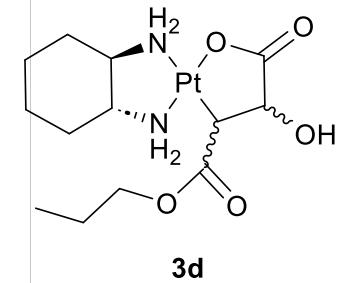
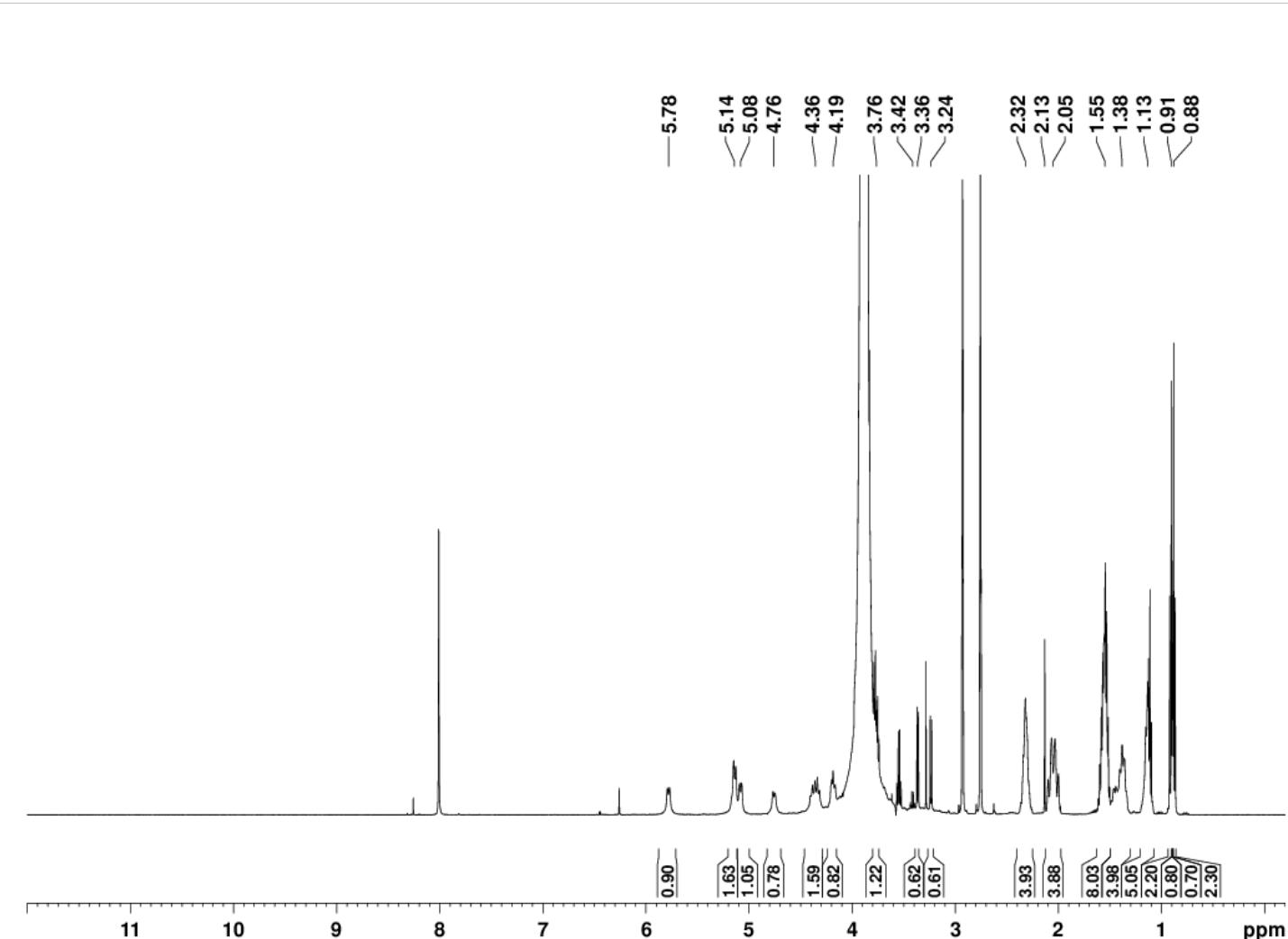


Fig. S19 ¹H NMR spectrum of compound 3d.

^{13}C NMR spectrum of compound 3d

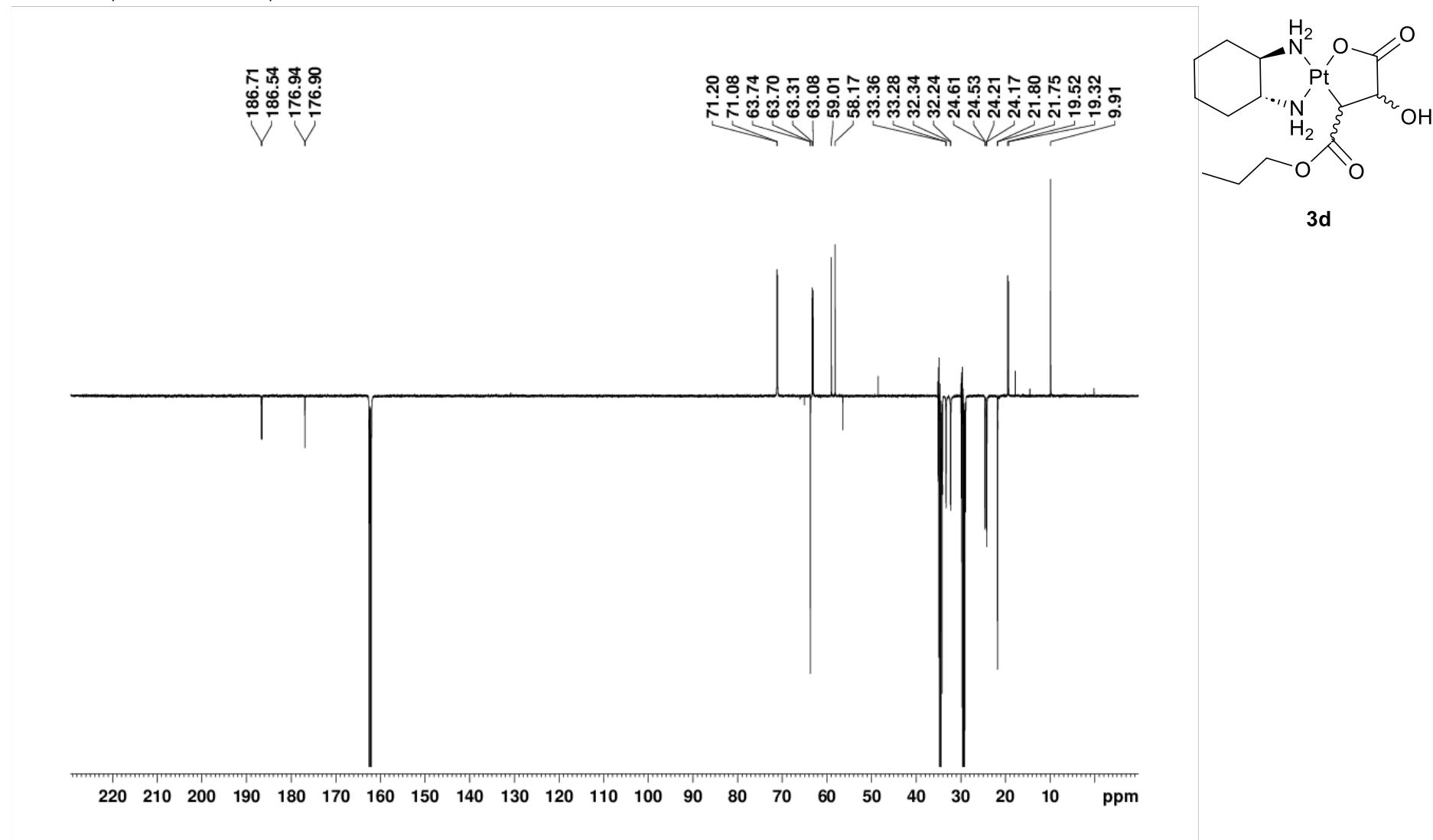


Fig. S20 ^{13}C NMR spectrum of compound 3d.

^{195}Pt NMR spectrum of compound **3d**

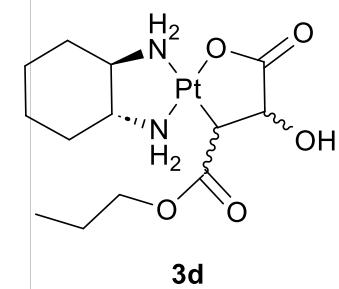
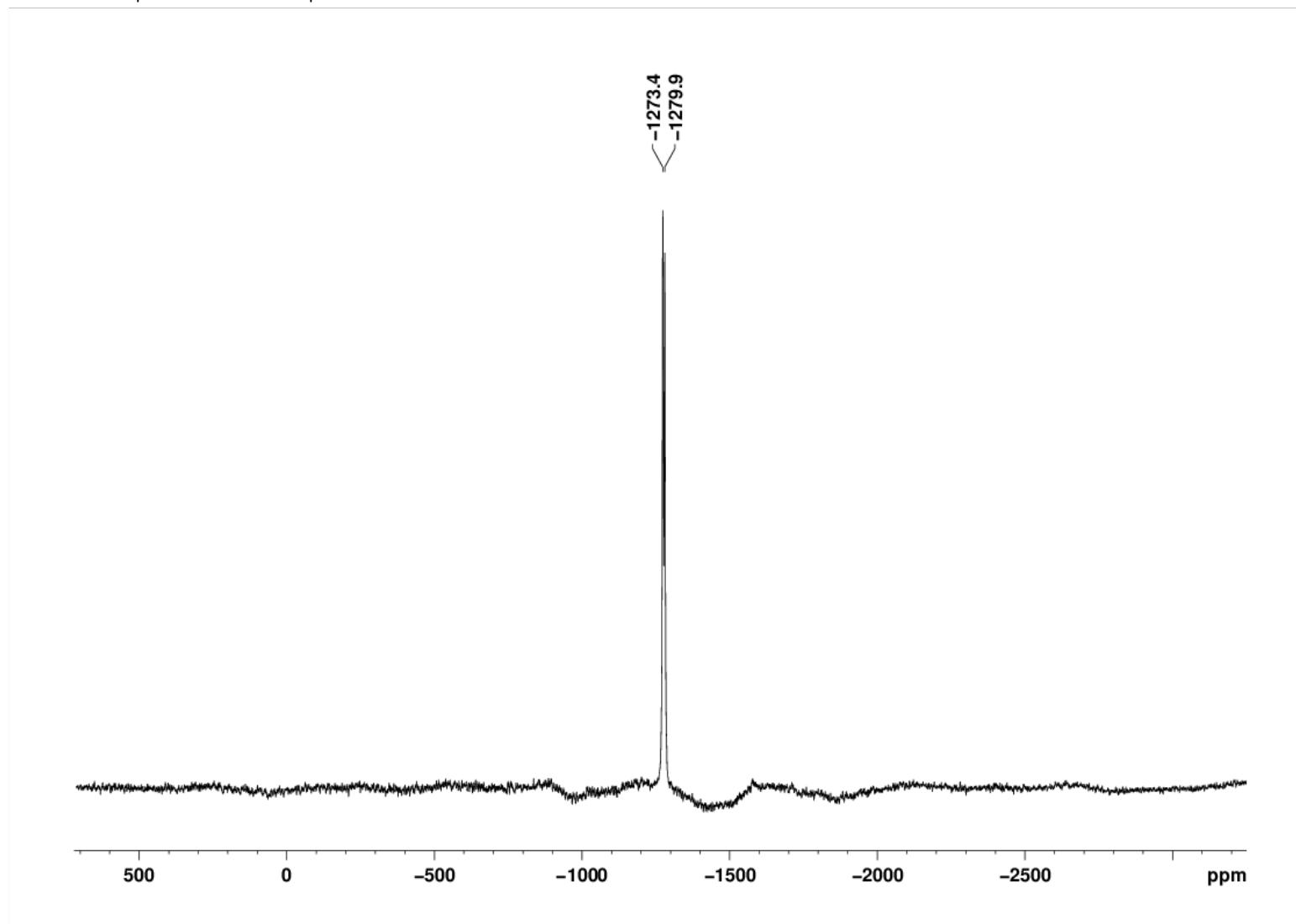


Fig. S21 ^{195}Pt NMR spectrum of compound **3d**.

^1H NMR spectrum of compound 3e

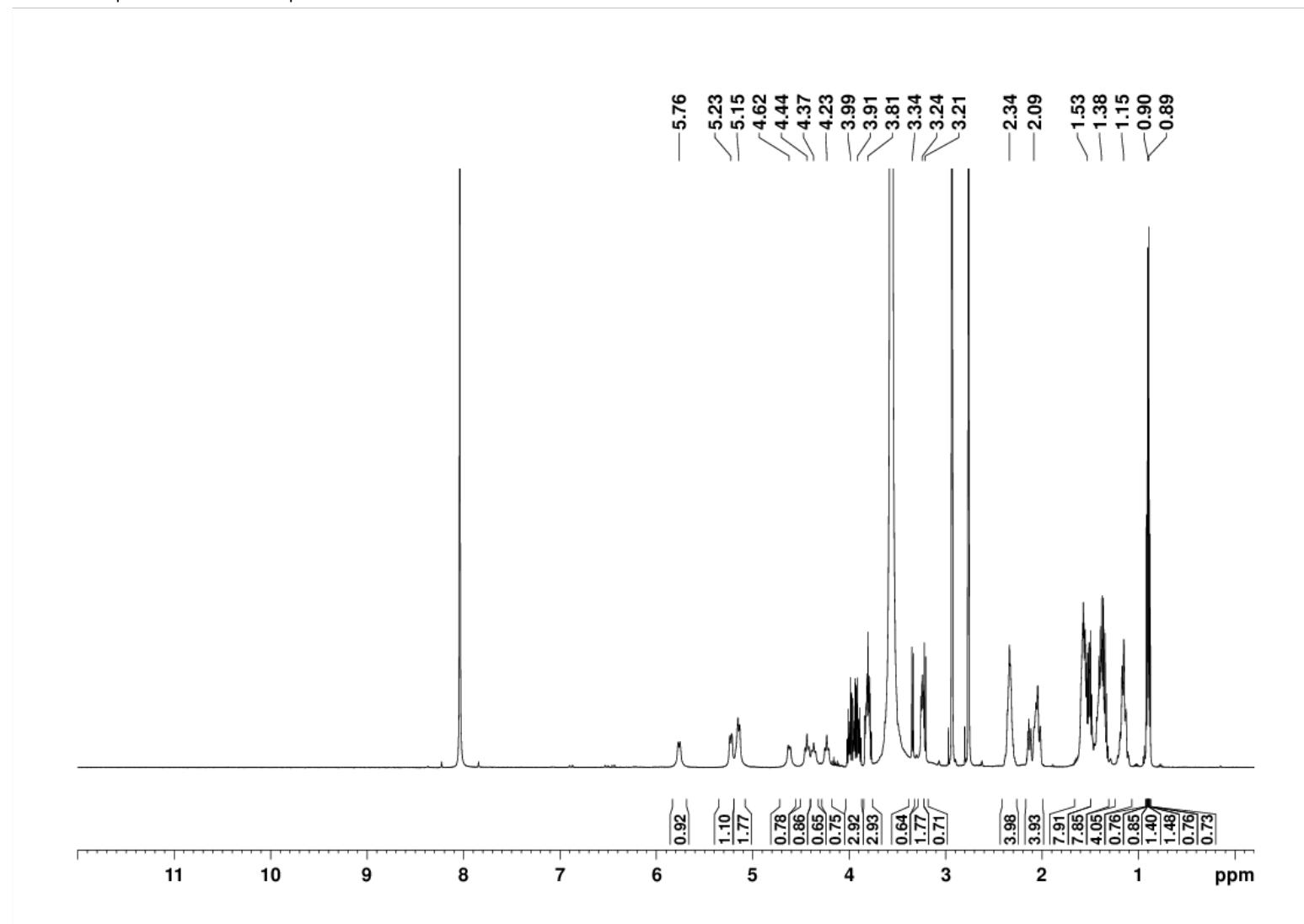
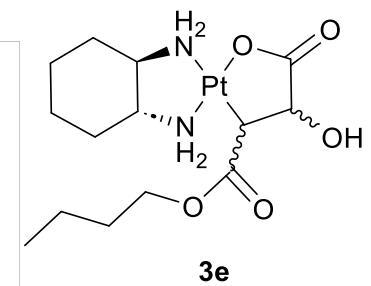


Fig. S22 ^1H NMR spectrum of compound 3e.



^{13}C NMR spectrum of compound 3e

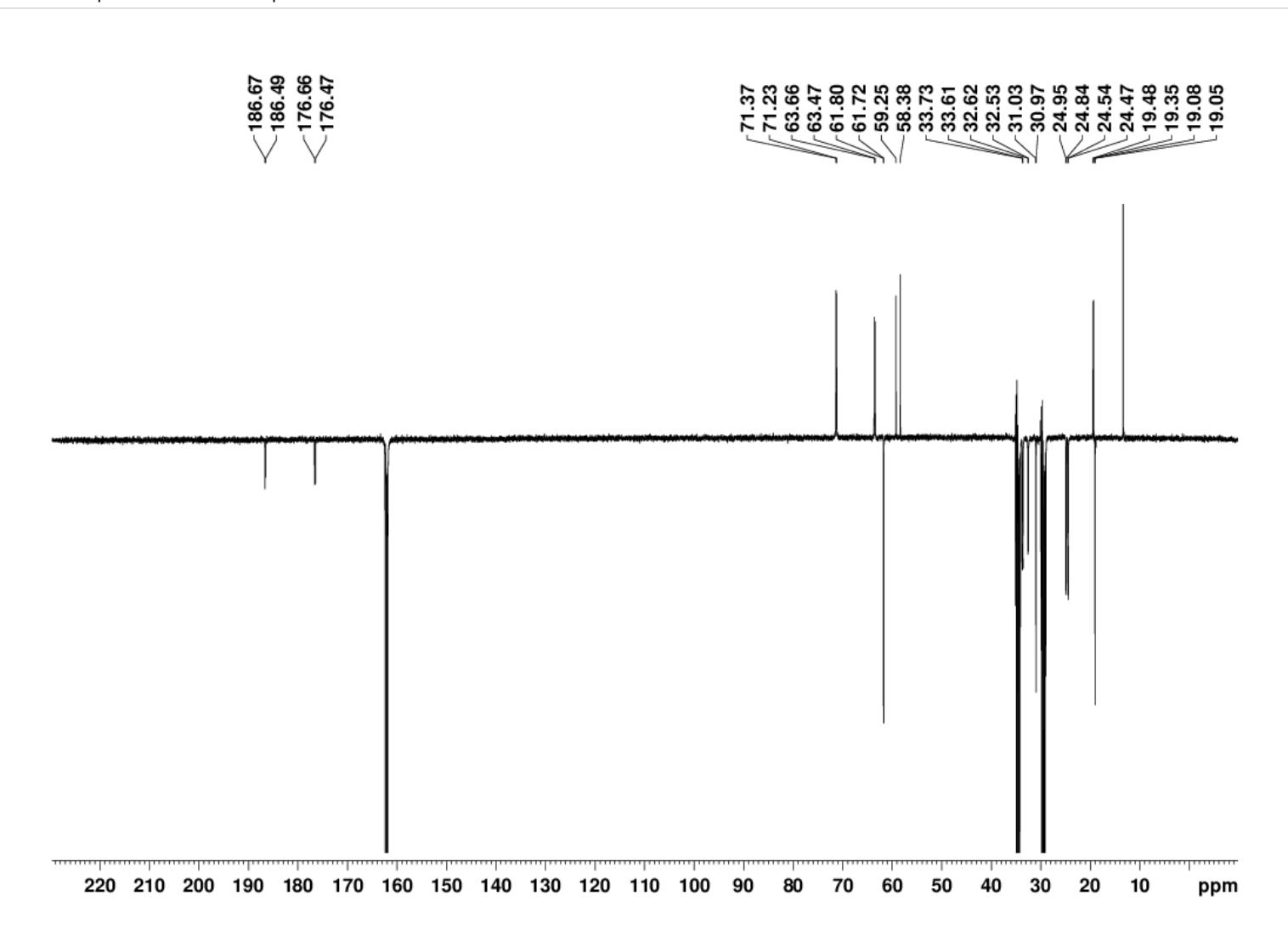
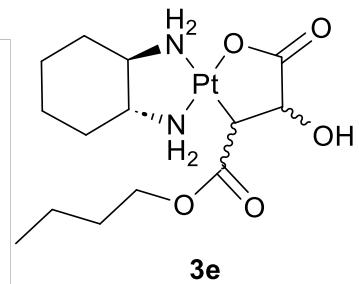


Fig. S23 ^{13}C NMR spectrum of compound 3e.



^{195}Pt NMR spectrum of compound **3e**

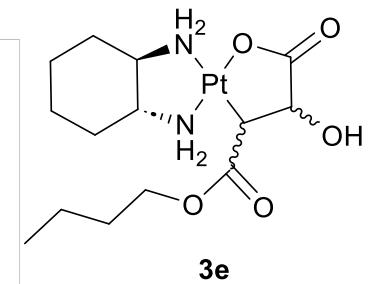
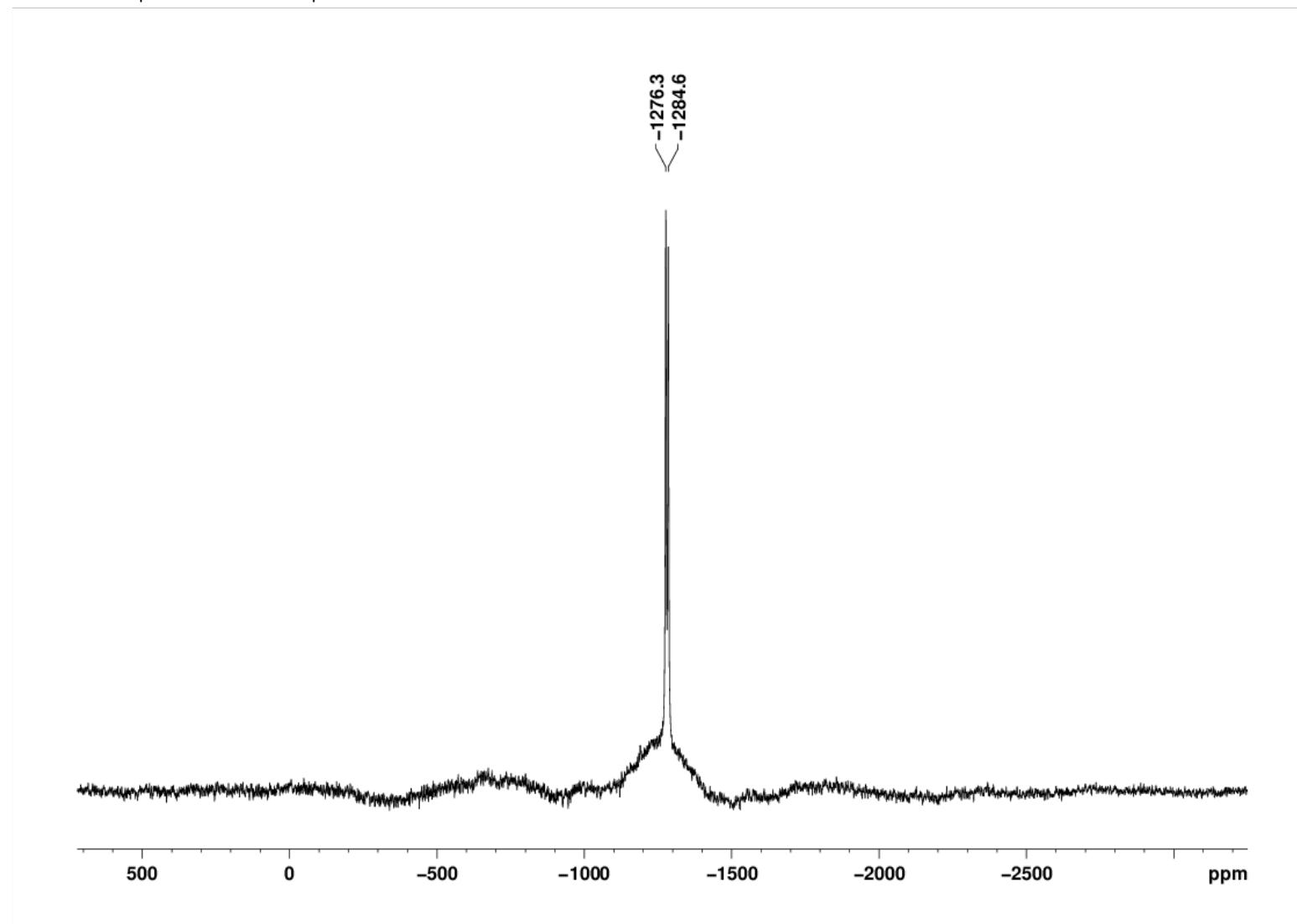


Fig. S24 ^{195}Pt NMR spectrum of compound **3e**.

^1H NMR spectrum of compound 3f

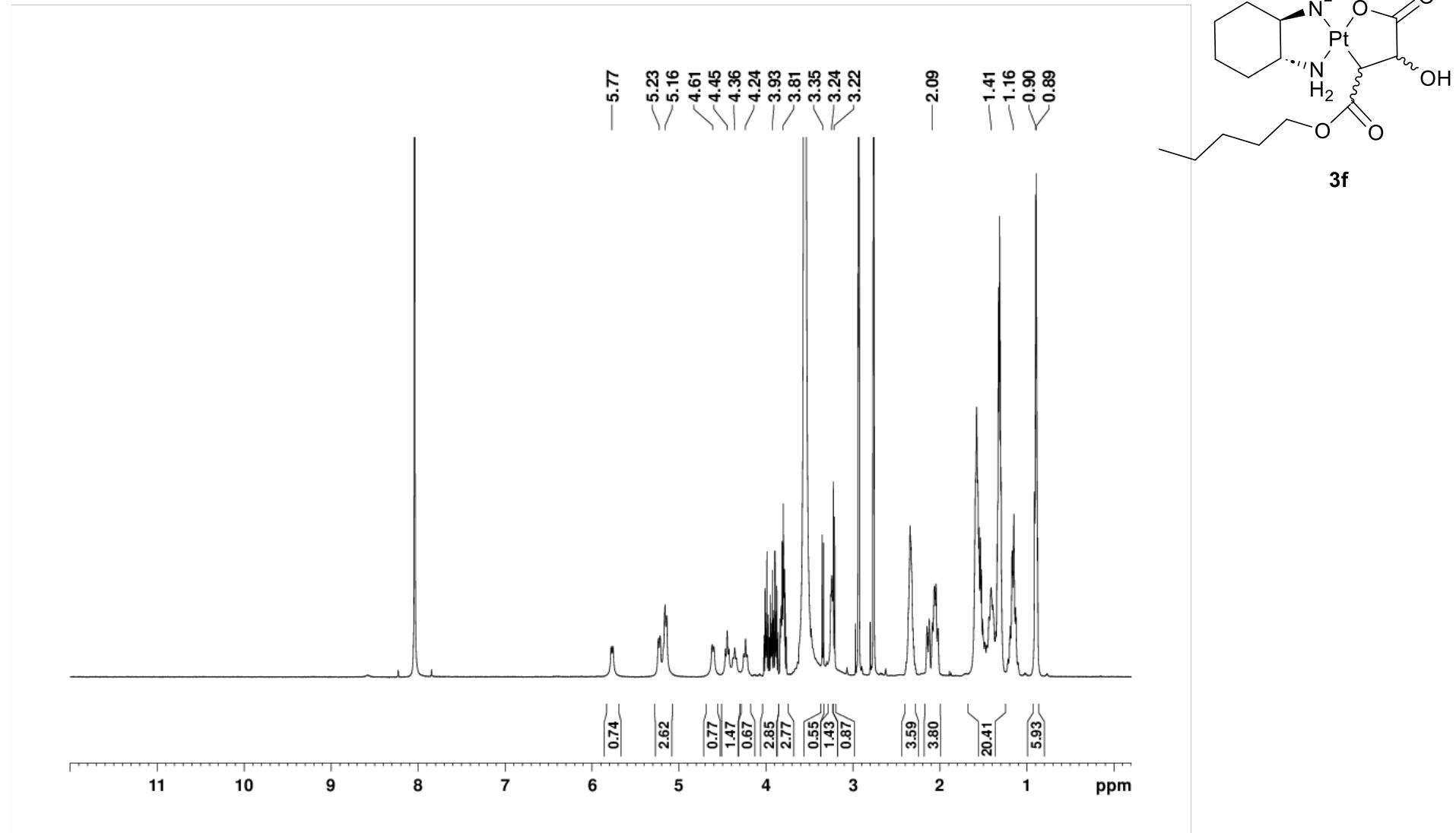


Fig. S25 ^1H NMR spectrum of compound 3f.

¹³C NMR spectrum of compound 3f

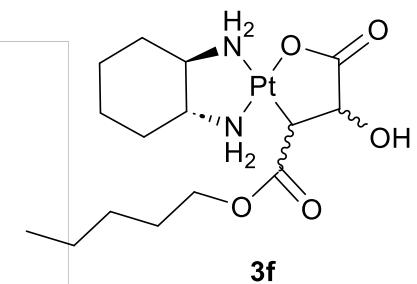
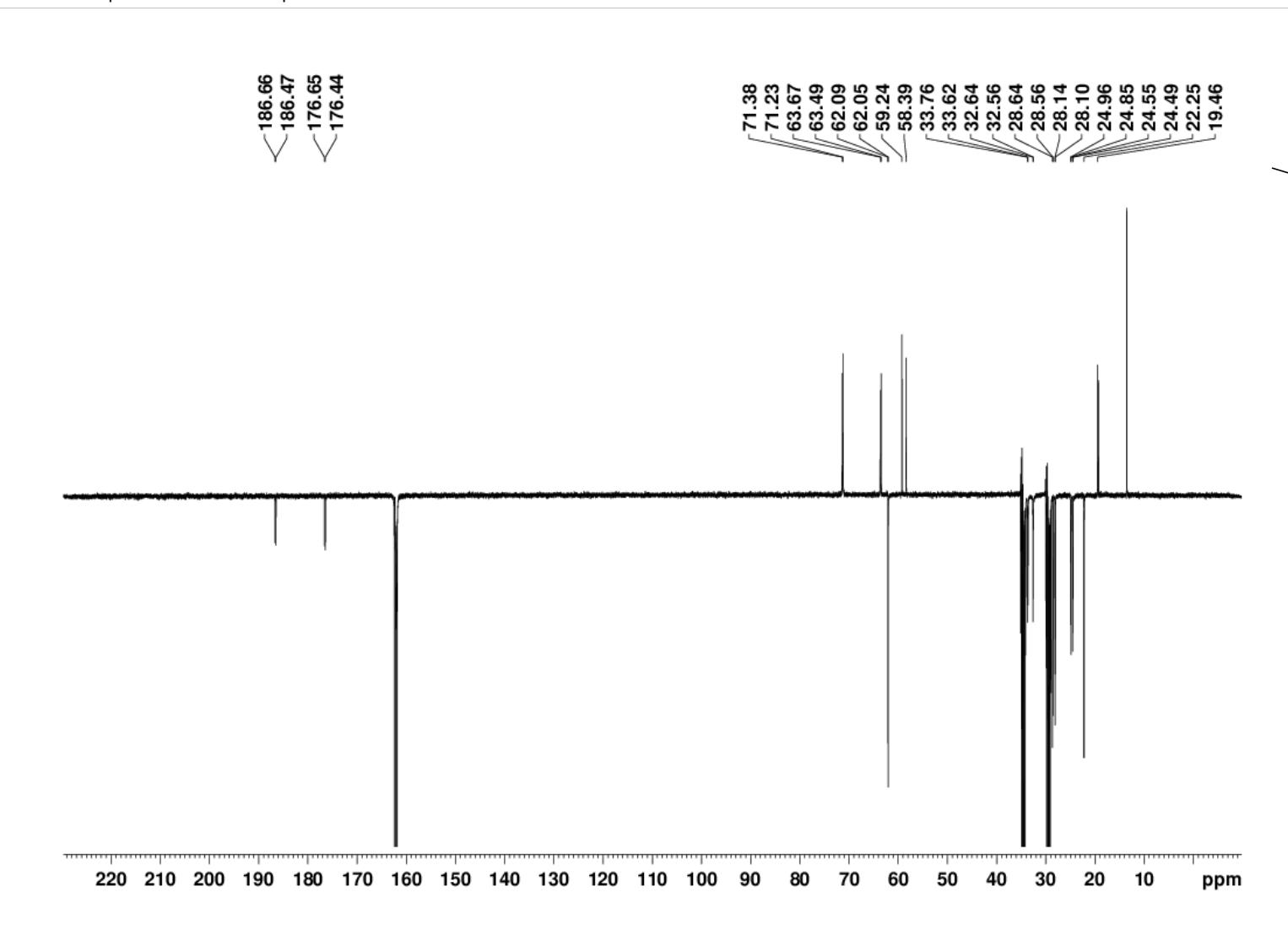


Fig. S26 ¹³C NMR spectrum of compound 3f.

^{195}Pt NMR spectrum of compound **3f**

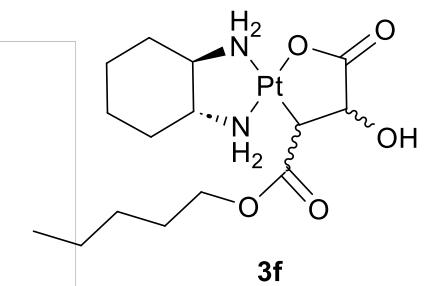
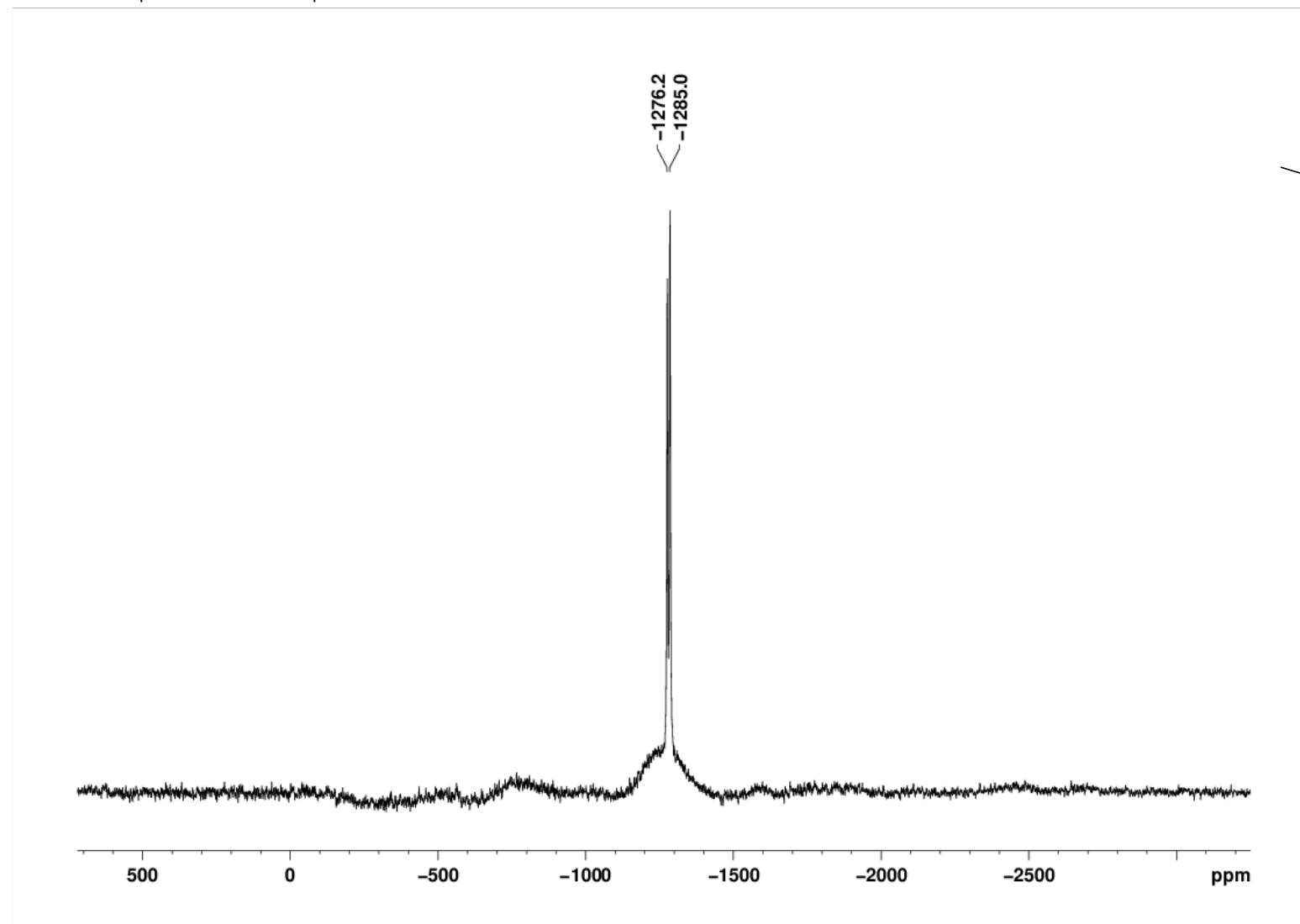


Fig. S27 ^{195}Pt NMR spectrum of compound **3f**.

Comparison of calculated geometries of diastereoisomers A and B of complex **3a** with experimental geometries

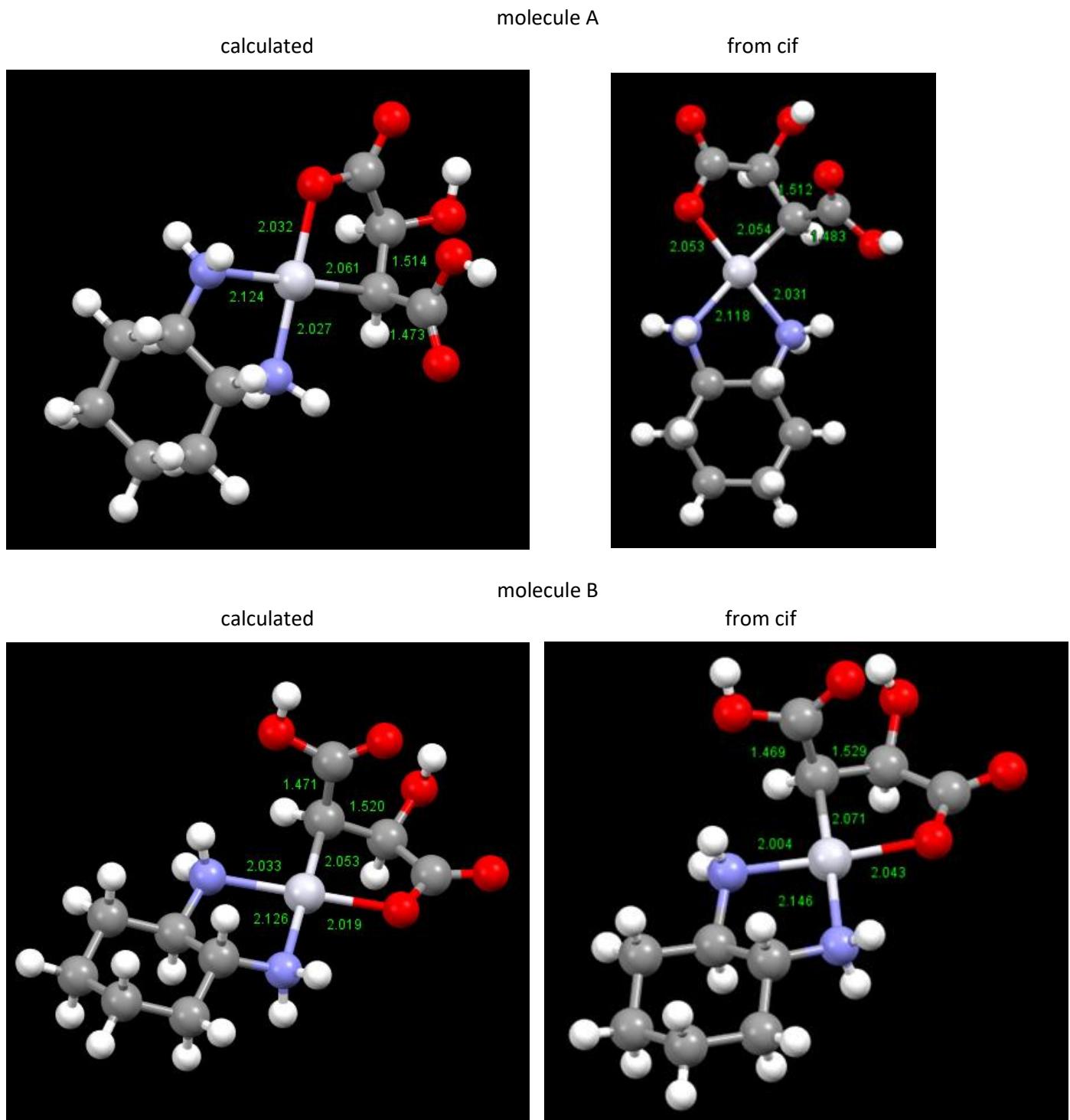


Fig. S28 Comparison of calculated geometries of diastereoisomers A and B of complex **3a** with experimental geometries.

Comparison of calculated geometries of diastereoisomers A and B of complex **3a** with either carboxyl or carboxylate group

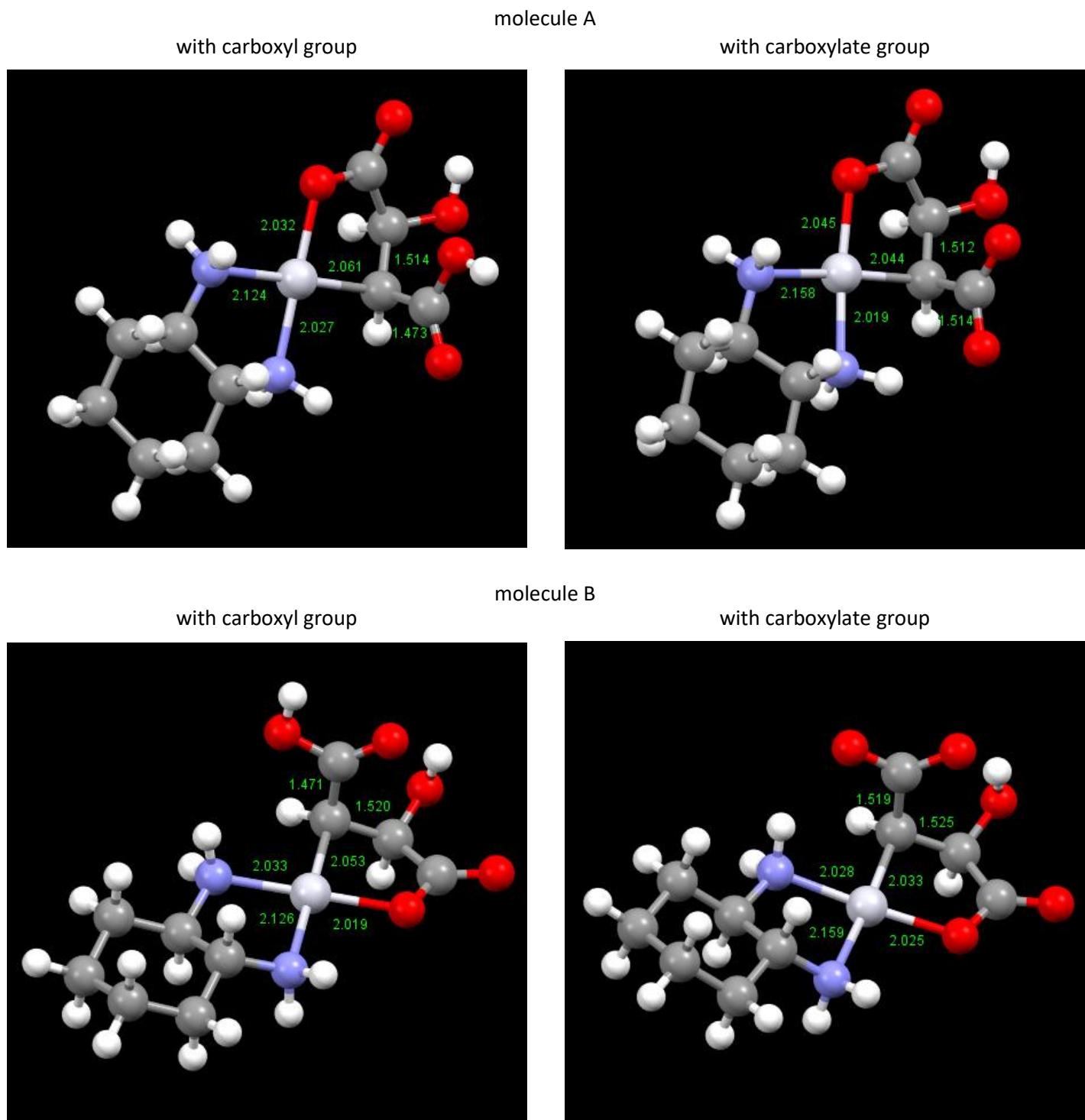


Fig. S29 Comparison of calculated geometries of diastereoisomers A and B of complex **3a** with either carboxyl or carboxylate group.

X-Ray Diffraction Analysis of complex 3a

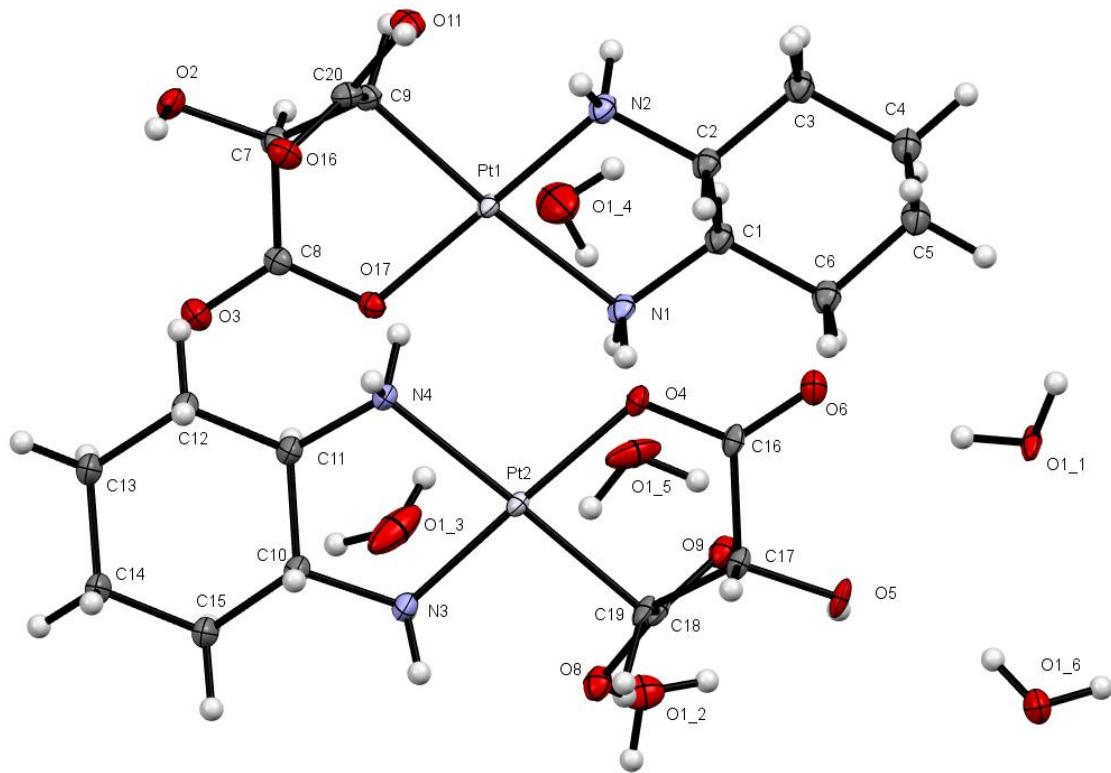


Fig. S30 ORTEP view of complex 3a drawn with 50% displacement ellipsoids.

Table S1. Overview of the sample and crystal data, data collection and structure refinement for complex **3a**.

Identification code	mawa001_b	
Empirical formula	C ₁₀ H ₂₄ N ₂ O ₈ Pt	
Formula weight	495.40 g/mol	
Temperature	100 K	
Wavelength	1.54186 Å	
Crystal system	Triclinic	
Space group	P1	
Unit cell dimensions	a = 8.2831(17) Å b = 9.6777(19) Å c = 10.192(2) Å	α= 73.15(3)° β= 76.80(3)° γ = 80.70(3)°
Volume	757.23(4) Å ³	
Z	2	
Density (calculated)	2.173 g/cm ³	
Absorption coefficient	17.721 mm ⁻¹	
F(000)	480	
Crystal size	0.130 x 0.070 x 0.050 mm ³	
Theta range for data collection	5.515° to 70.870°	
Index ranges	-6<=h<=10, -11<=k<=10, -10<=l<=12	
Reflections collected	15268	
Independent reflections	3380 [R(int) = 0.0164]	
Completeness to theta = 67.679°	94.9 %	
Absorption correction	Sphere	
Max. and min. transmission	0.4261 and 0.1088	
Refinement method	Full-matrix least-squares on F2	
Data / restraints / parameters	3380 / 501 / 420	
Goodness-of-fit on F2	1.048	
Final R indices [I>2sigma(I)]	R1 = 0.0178, wR2 = 0.0459	
R indices (all data)	R1 = 0.0185, wR2 = 0.0462	
Absolute structure parameter	0.087(15)	
Extinction coefficient	0.00154(9)	
Largest diff. peak and hole	0.960 and -1.462 e·Å ³	

Table S2. Overview of bond lengths [Å] for complex **3a**.

Pt(1)-N(2)	2.032(12)		N(3)-C(10)	1.506(13)
Pt(1)-O(17)	2.053(10)		N(4)-C(11)	1.492(14)
Pt(1)-C(9)	2.054(14)		C(1)-C(6)	1.527(14)
Pt(1)-N(1)	2.118(13)		C(1)-C(2)	1.535(10)
Pt(2)-N(3)	2.005(12)		C(2)-C(3)	1.517(13)
Pt(2)-O(4)	2.044(10)		C(3)-C(4)	1.534(13)
Pt(2)-C(19)	2.071(13)		C(4)-C(5)	1.518(10)
Pt(2)-N(4)	2.145(13)		C(5)-C(6)	1.500(15)
O(2)-C(7)	1.428(15)		C(7)-C(9)	1.512(14)
O(3)-C(8)	1.237(14)		C(7)-C(8)	1.518(13)
O(4)-C(16)	1.302(15)		C(9)-C(20)	1.483(19)
O(5)-C(17)	1.419(14)		C(10)-C(11)	1.525(10)
O(6)-C(16)	1.220(15)		C(10)-C(15)	1.529(13)
O(8)-C(18)	1.344(15)		C(11)-C(12)	1.505(14)
O(9)-C(18)	1.247(16)		C(12)-C(13)	1.560(15)
O(11)-C(20)	1.324(15)		C(13)-C(14)	1.523(10)
O(16)-C(20)	1.212(16)		C(14)-C(15)	1.529(14)
O(17)-C(8)	1.298(15)		C(16)-C(17)	1.547(14)
N(1)-C(1)	1.510(15)		C(17)-C(19)	1.529(14)
N(2)-C(2)	1.481(13)		C(18)-C(19)	1.469(19)

Table S3. Overview of angles [°] for complex **3a**.

N(2)-Pt(1)-O(17)	178.5(5)		O(3)-C(8)-O(17)	120.7(10)
N(2)-Pt(1)-C(9)	94.1(5)		O(3)-C(8)-C(7)	122.2(11)
O(17)-Pt(1)-C(9)	85.5(4)		O(17)-C(8)-C(7)	116.9(10)
N(2)-Pt(1)-N(1)	82.1(5)		C(20)-C(9)-C(7)	113.4(13)
O(17)-Pt(1)-N(1)	98.4(4)		C(20)-C(9)-Pt(1)	105.5(9)
C(9)-Pt(1)-N(1)	175.9(6)		C(7)-C(9)-Pt(1)	105.2(8)
N(3)-Pt(2)-O(4)	176.5(5)		N(3)-C(10)-C(11)	107.2(7)
N(3)-Pt(2)-C(19)	95.2(5)		N(3)-C(10)-C(15)	112.5(8)
O(4)-Pt(2)-C(19)	83.0(5)		C(11)-C(10)-C(15)	112.2(7)
N(3)-Pt(2)-N(4)	83.2(5)		N(4)-C(11)-C(12)	113.0(8)
O(4)-Pt(2)-N(4)	98.6(4)		N(4)-C(11)-C(10)	108.4(7)
C(19)-Pt(2)-N(4)	178.4(6)		C(12)-C(11)-C(10)	110.1(7)
C(16)-O(4)-Pt(2)	117.1(8)		C(11)-C(12)-C(13)	110.4(9)
C(8)-O(17)-Pt(1)	112.1(7)		C(14)-C(13)-C(12)	111.6(7)
C(1)-N(1)-Pt(1)	106.5(7)		C(13)-C(14)-C(15)	111.2(7)
C(2)-N(2)-Pt(1)	112.0(7)		C(10)-C(15)-C(14)	109.8(8)
C(10)-N(3)-Pt(2)	109.5(7)		O(6)-C(16)-O(4)	124.6(10)
C(11)-N(4)-Pt(2)	107.7(7)		O(6)-C(16)-C(17)	120.9(11)
N(1)-C(1)-C(6)	115.5(8)		O(4)-C(16)-C(17)	114.4(10)
N(1)-C(1)-C(2)	105.2(7)		O(5)-C(17)-C(19)	115.2(10)
C(6)-C(1)-C(2)	111.3(7)		O(5)-C(17)-C(16)	110.8(10)
N(2)-C(2)-C(3)	112.3(8)		C(19)-C(17)-C(16)	112.1(10)
N(2)-C(2)-C(1)	107.1(7)		O(9)-C(18)-O(8)	122.0(11)
C(3)-C(2)-C(1)	111.7(7)		O(9)-C(18)-C(19)	123.5(11)
C(2)-C(3)-C(4)	111.0(8)		O(8)-C(18)-C(19)	114.4(12)
C(5)-C(4)-C(3)	109.5(7)		C(18)-C(19)-C(17)	113.7(13)
C(6)-C(5)-C(4)	112.9(7)		C(18)-C(19)-Pt(2)	101.4(8)
C(5)-C(6)-C(1)	109.6(8)		C(17)-C(19)-Pt(2)	106.9(8)
O(2)-C(7)-C(9)	115.6(10)		O(16)-C(20)-O(11)	123.9(12)
O(2)-C(7)-C(8)	112.1(9)		O(16)-C(20)-C(9)	123.8(11)
C(9)-C(7)-C(8)	113.0(10)		O(11)-C(20)-C(9)	112.3(12)